

Electronic Supplementary Information

Mechanistic Insight into the Hydrazine Decomposition on Rh(111):

Effect of Reaction Intermediate on Catalytic Activity

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Fig. S1. The metastable adsorption configurations involved in hydrazine decomposition on Rh(111).

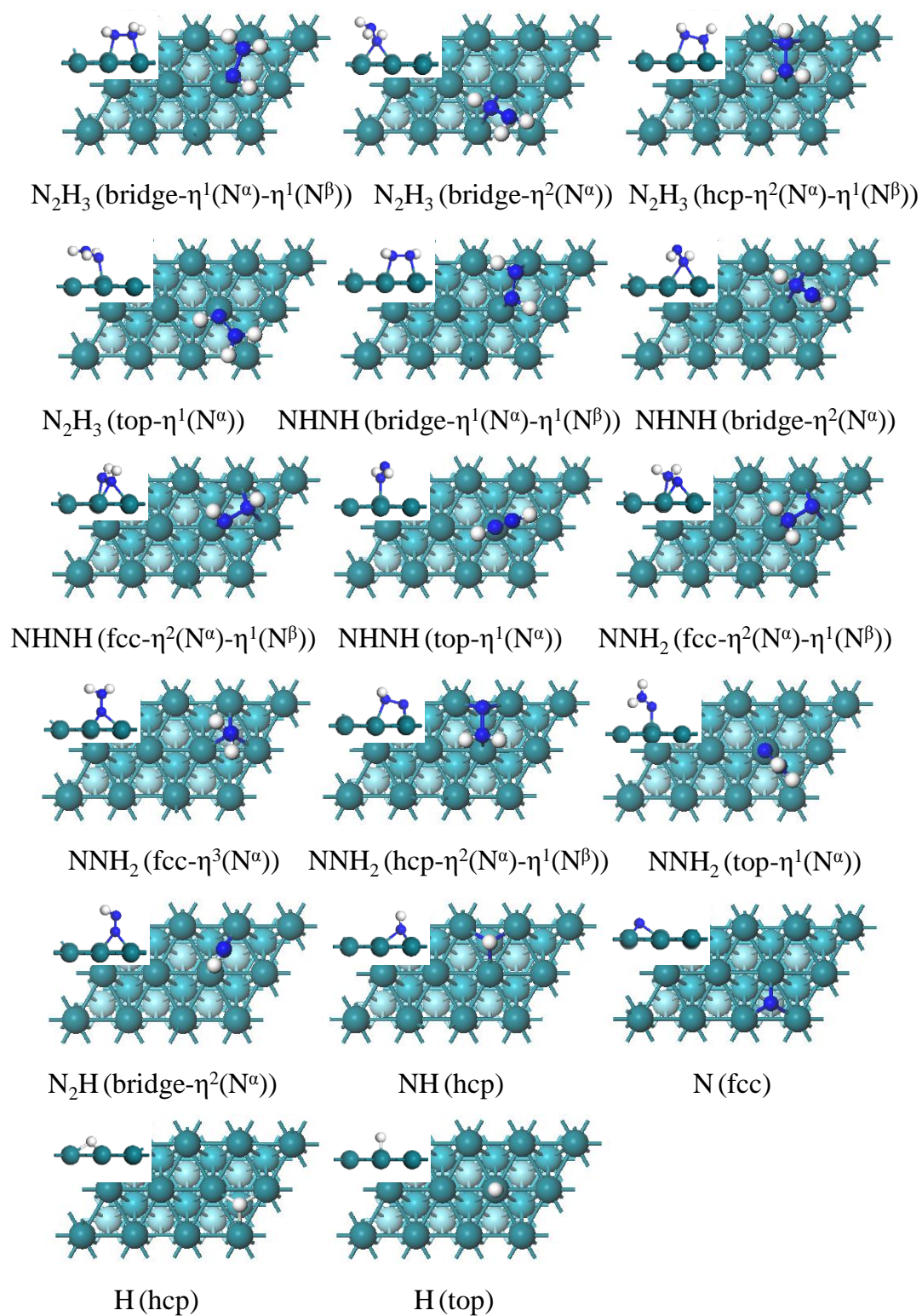


Fig S2. Elementary steps of intramolecular N_2H_4 decomposition on Rh(111) (2×2) slab.

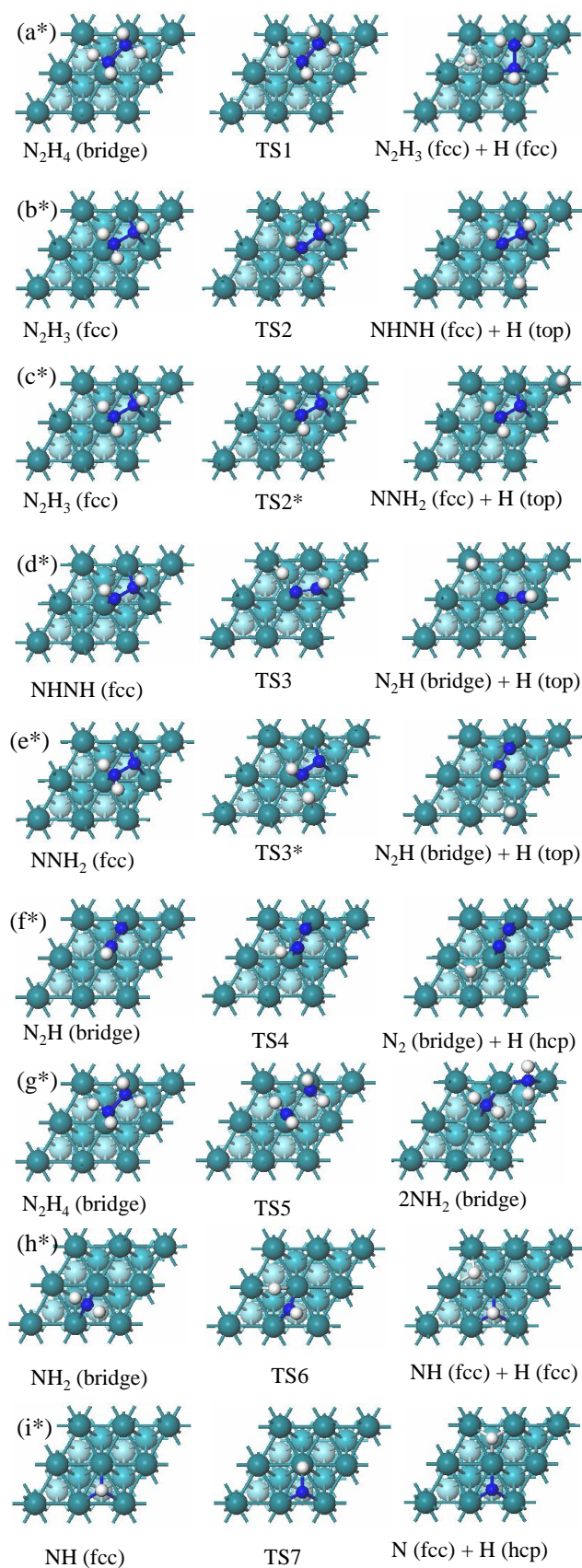
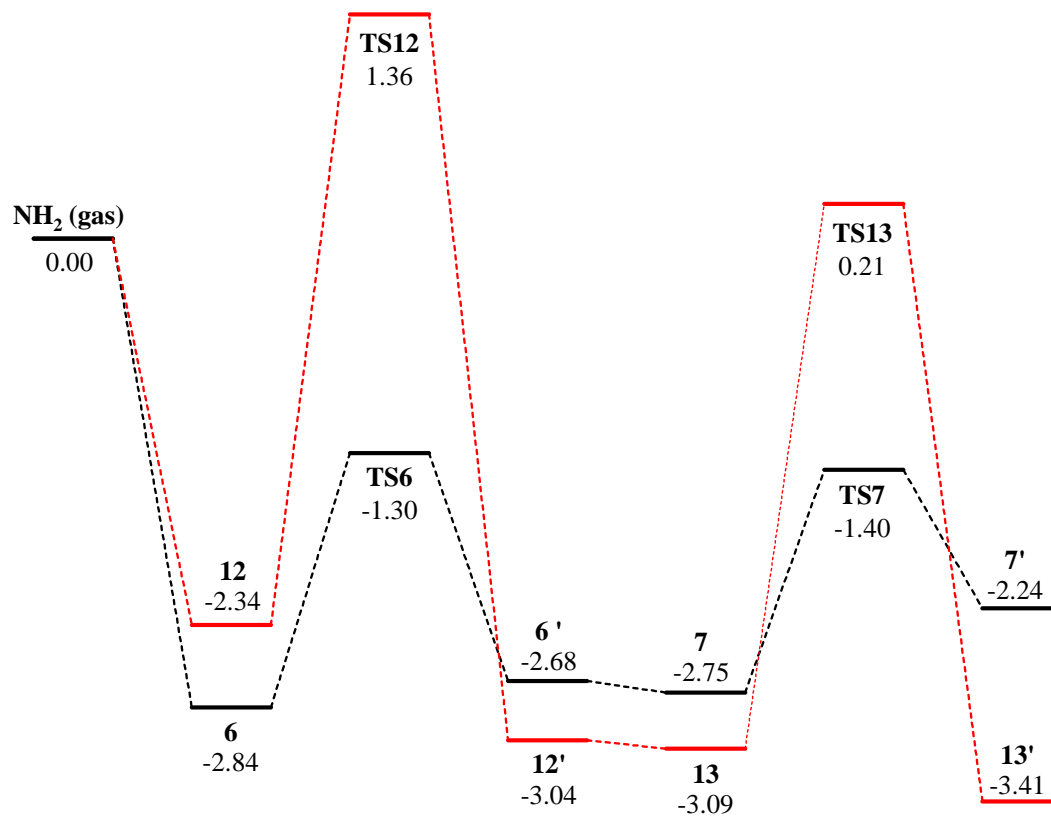


Fig. S3 PESs of NH_2 decomposition on Rh(111). The energy reference corresponds to the total energy of one gas-phase NH_2 , one clean slab, and two adsorbed NH_2 fragment.



6, $\text{NH}_2^* + 2\text{NH}_2^*$; **6'**, $[\text{NH} + \text{H}]^* + 2\text{NH}_2^*$; **7**, $\text{NH}^* + \text{H}^* + 2\text{NH}_2^*$; **7'**, $[\text{N} + \text{H}]^* + \text{H}^* + 2\text{NH}_2^*$; **12**, $[\text{NH}_2 + \text{NH}_2]^* + \text{NH}_2^*$; **12'**, $[\text{NH} + \text{NH}_3]^* + \text{NH}_2^*$; **13**, $[\text{NH} + \text{NH}_2]^* + \text{NH}_3^*$; **13'**, $[\text{N} + \text{NH}_3]^* + \text{NH}_3^*$.

Table S1 Coadsorption configurations, coadsorption energies (in eV), and structural parameters (in angstroms) for the coadsorption systems involved in NH₂-assisted hydrazine decomposition on Rh(111).

System	E_{ads}	d_{N-N}^a	d_{N-Rh}^b	d_{N-H}^a
N ₂ H ₄ (bridge) + NH ₂ (bridge)	3.67	1.450	2.248, 2.248	1.027, 1.027, 1.030, 1.031
N ₂ H ₃ (fcc) + NH ₂ (bridge)	4.80	1.439	2.118, 2.152, 2.165	1.025, 1.027, 1.027
NHNH (hcp) + NH ₂ (bridge)	4.41	1.393	2.082, 2.094, 2.123	1.027, 1.032
NNH ₂ (fcc) + NH ₂ (bridge)	6.01	1.436	2.056, 2.068, 2.126	1.026, 1.027
N ₂ H (bridge) + NH ₂ (bridge)	4.58	1.254	1.963, 2.079	1.035

^a The N–N and N–H bonds in the N₂H_x fragment. ^b The distance between the N atoms of N₂H_x and surface Rh atoms.

Table S2 Energy barriers E_a and reaction energies ΔE (in eV) for the elementary steps involved in intramolecular hydrazine decomposition on Rh(111) (2×2) slab.

Reactions	E_a	ΔE
(a*) $\text{N}_2\text{H}_4 \rightarrow \text{N}_2\text{H}_3 + \text{H}$	0.75	-0.30
(b*) $\text{N}_2\text{H}_3 \rightarrow \text{NHNH} + \text{H}$	1.07	0.63
(c*) $\text{N}_2\text{H}_3 \rightarrow \text{NNH}_2 + \text{H}$	1.50	0.46
(d*) $\text{NHNH} \rightarrow \text{N}_2\text{H} + \text{H}$	1.04	0.11
(e*) $\text{NNH}_2 \rightarrow \text{N}_2\text{H} + \text{H}$	1.12	0.20
(f*) $\text{N}_2\text{H} \rightarrow \text{N}_2 + \text{H}$	0.99	-0.28
(g*) $\text{N}_2\text{H}_4 \rightarrow 2\text{NH}_2$	0.46	-0.81
(h*) $\text{NH}_2 \rightarrow \text{NH} + \text{H}$	1.53	0.24
(i*) $\text{NH} \rightarrow \text{N} + \text{H}$	1.29	0.57

Table S3 Diffusion barrier (E_{diff}), reaction energy (ΔE) (in eV) for the diffusion of some species on Rh(111)- (2×3) slab.

Diffusion	E_{diff}	ΔE
N_2H_4 (top) \rightarrow N_2H_4 (bridge)	0.03	0.01
N_2H_4 (bridge) + NH_2 (bridge-1) \rightarrow N_2H_4 (bridge) + NH_2 (bridge-2)	0.39	0.04
N_2H_4 (top) + NH_2 (bridge-1) \rightarrow N_2H_4 (bridge) + NH_2 (bridge-2)	0.69	0.21