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## **Supporting Information**

### A direct Fe-O coordination at FePc/MoO<sub>x</sub> interface investigated by XPS and

### **NEXAFS** spectroscopies

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# 1. Fitting parameters for Mo 3d

**Table S1** shows Mo 3d XPS peak fitting parameters for Fig 1a of the manuscript, including binding energy ( $E_B$ ), peak intensity (Intensity), and full width at half maximum (FWHM).

Mo species			E <sub>B</sub> (eV)	Intensity	FWHM (eV)	Amount
As-depo	+6	3d <sub>3/2</sub>	235.7	115	1.00	100%
		3d <sub>5/2</sub>	232.6	220	1.05	
250 °C	+6	3d <sub>3/2</sub>	235.7	98	1.30	51.1%
		3d <sub>5/2</sub>	232.6	163	1.30	
	+m	3d <sub>3/2</sub>	234.7	87	1.35	48.9%
		3d <sub>5/2</sub>	231.6	163	1.30	
300 °C	+6	3d <sub>3/2</sub>	235.7	165	1.31	61.2%
		3d <sub>5/2</sub>	232.6	252	1.25	
	+m	3d <sub>3/2</sub>	234.7	95	1.30	· 36.0%
		3d <sub>5/2</sub>	231.6	150	1.35	
	+4	3d <sub>3/2</sub>	232.2	8	0.55	2.8%
		3d <sub>5/2</sub>	229.1	11	0.55	
350 °C	+6	3d <sub>3/2</sub>	235.7	190	1.35	51.9%
		3d <sub>5/2</sub>	232.6	310	1.30	
	+m	3d <sub>3/2</sub>	234.7	106	1.30	26.6%
		3d <sub>5/2</sub>	231.5	150	1.30	
	+n	3d <sub>3/2</sub>	233.9	40	1.20	11.4%
		3d <sub>5/2</sub>	230.6	70	1.35	
	+4	3d <sub>3/2</sub>	232.2	39	0.6	10.1%
		3d <sub>5/2</sub>	229.1	58	0.6	
400 ℃	+6	3d <sub>3/2</sub>	235.5	150	1.20	43.7%
		3d <sub>5/2</sub>	232.6	208	1.20	
	+m	3d <sub>3/2</sub>	234.6	65	1.20	20.8%

		3d <sub>5/2</sub>	231.5	105	1.20	
	+n	3d <sub>3/2</sub>	233.7	50	1.20	15.9%
		3d <sub>5/2</sub>	230.5	80	1.20	
	+4	3d <sub>3/2</sub>	232.2	70	0.50	19.5%
		3d <sub>5/2</sub>	229.1	90	0.55	
450 °C	+6	3d <sub>3/2</sub>	235.7	95	1.30	44.1%
		3d <sub>5/2</sub>	232.6	173	1.30	
	+m	3d <sub>3/2</sub>	234.7	48	1.30	20.4%
		3d <sub>5/2</sub>	231.5	76	1.25	
	+n	3d <sub>3/2</sub>	233.8	39	1.20	16.6%
		3d <sub>5/2</sub>	230.5	62	1.25	
	+4	3d <sub>3/2</sub>	232.3	46	0.6	19.2%
		3d <sub>5/2</sub>	229.2	71	0.6	

Note: "+m" and "+n" reprent the medium states of Mo ions between +6 and +4 valence states. The Lorentz-Gaussian (GL%) ratio is fixed as 20% during the fitting process.

### 2. AFM image for MoO<sub>x</sub> adsorbed with 1 ML FePc

**Fig. S1** displays the morphology of 1 ML FePc adsorbed  $MoO_x$  sample. It is seen the FePc adsorption has less influence to the morphology of the  $MoO_x$  substrate after annealing at 400 °C. The  $MoO_x$  particles appear to be comparable to the size of the bear  $MoO_x$  substrate that shown in Figure 2b in the article. For the annealed FePc/MoO<sub>x</sub> sample, the root mean square roughness is ~5.36 nm in contrast to the 1.2 nm of the bare  $MoO_x$  due to the FePc adsorption.



Fig. S1 The AFM image of FePc film adsorbed on the MoOx surface annealed at 400  $\,\,{}^{\rm C}$ 

#### 3. N 1s core level

**Figure S2** shows the N 1s XPS spectra for the multilayer (top) and monolayer (bottom) FePc films adsorbed on  $MoO_x$ . For the multilayer FePc films, the dominant peak at 398.6 eV are assigned to the two types of N atoms of the pyrrole (N<sub>P</sub>) and the bridge nitrogen between the pyrrole ring (N<sub>B</sub>). These two types of N atoms are difficult to resolve because of a small energy difference of ~0.2 eV. In addition, a small shoulder appears at the higher binding energy in both spectra, which are ascribed to the shake-ups of nitrogen. As to the one monolayer FePc, there is only one main peak at 399.6 eV, while a small shoulder can still be deconvoluted with a sharply declined intensity. There is a large energetic shift of about 1 eV between the one monolayer FePc and the multilayer FePc.



Figure S2 N 1s XPS spectra of the multilayer (top) and monolayer (bottom) FePc

films adsorbed on MoO<sub>x</sub>.

### 4. Fitting parameters for O 1s core level

**Table S2** shows the fitting parameters and results for O 1s XPS spectra in Fig. 5 of the manuscript.

O 1s sp	ecies	E <sub>B</sub> (eV)	Intensity	FWHM (eV)	Amount
	O-Mo <sup>6+</sup>	530.2	81000	1.10	42.4%
MoO <sub>x</sub>	O-Mo <sup>m,n+</sup>	530.8	69042	1.25	36.1%
	O-Mo <sup>4+</sup>	531.6	41000	1.50	21.5%
	O-Mo <sup>6+</sup>	530.2	20000	1.00	19.0%
1 ML FePc/	O-Mo <sup>m,n+</sup>	530.7	50500	1.30	48.0%
MoOx	O-Mo <sup>4+</sup>	531.6	25000	1.45	33.8%
	O <sub>new</sub>	532.3	9600	1.31	9.2%

Note: "+m" and "+n" represent the oxygen species bonding to the Mo ions with +m and +n valence states, respectively. The two types of O-Mo bonds were fitted with one oxygen component due to the limited resolution. The Lorentz-Gaussian (GL%) is fixed as 20% during the fitting process.

#### 5. C K-edge XAS spectra

**Fig. S3** shows the dependence of C K-edge XAS spectra on the x-ray incidence angle (35°, 50° and 90°) for 1 ML FePc films adsorbed on the MoO<sub>x</sub> substrate. The XAS spectra are characterized by multiple absorption peaks, basically associated with transitions to the empty states located on the benzene and pyrrole macrocycles of  $\pi^*$  and  $\sigma^*$  characters. It is seen the angular dependence of C K-edge is very small, suggesting a lack of uniform orientation in the film.



Fig. S3 C K-edge XAS spectra for 1 ML FePc film at different x-ray incidence angle of 35 °, 50 ° and 90 °. The inset shows the incident angle  $\theta$  of the incident light with

respect to the sample plane.

### 6. Fe 2p XPS spectra

**Fig. S4** shows the Fe 2p XPS spectra for the 1 ML and multilayer FePc films that adsorbed on the  $MoO_x$ . Both spectra are dominated by two main peaks. The one at the BE of 721.8 eV is related to the Fe  $2p_{1/2}$  level, and the other at BE of 708.4 eV is the  $2p_{3/2}$  level. The Fe 2p of the 1 ML FePc sample shift ~0.7 eV towards the higher BE with respect to the multilayer.



Figure S4 Fe 2p XPS spectra of the multilayer (top) and monolayer (bottom) FePc

films adsorbed on MoO<sub>x</sub>.