

Table 1'. Some observed and calculated properties of halogen-bonded complexes B...X<sub>2</sub> involving non-polar dihalogen molecules X<sub>2</sub> (including CABS singles in calculating BSSE corrections for B...Br<sub>2</sub>)<sup>a</sup>

Lewis base B	Dihalogen molecules X <sub>2</sub>											
	Difluorine F <sub>2</sub>				Dichlorine Cl <sub>2</sub>				Dibromine Br <sub>2</sub>			
	<i>k</i> <sub>σ</sub> /(N m <sup>-1</sup> )	Ref.	<i>D</i> <sub>σ</sub> /kJ mol <sup>-1</sup>	<i>r</i> (Z...X <sub>i</sub> )/Å	<i>k</i> <sub>σ</sub> /(N m <sup>-1</sup> )	Ref.	<i>D</i> <sub>σ</sub> /kJ mol <sup>-1</sup>	<i>r</i> (Z...X <sub>i</sub> )/Å	<i>k</i> <sub>σ</sub> /(N m <sup>-1</sup> )	Ref.	<i>D</i> <sub>σ</sub> /kJ mol <sup>-1</sup>	<i>r</i> (Z...X <sub>i</sub> )/Å
OC	-	-	-	-	3.68(1)	12	5.19	3.145	5.03(2)	20	7.50	3.111
C <sub>2</sub> H <sub>2</sub>	-	-	-	-	5.61(1)	13	7.45	3.146	7.80(3)	21	11.05	3.106
C <sub>2</sub> H <sub>4</sub>	-	-	-	-	5.88(2)	14	8.61	3.092	8.8(2)	22	13.49	3.004
H <sub>2</sub> S	2.34(1)	7	3.43	3.143	6.23(2)	15	8.53	3.246	9.8(2)	23	13.83	3.131
H <sub>3</sub> P	-	-	-	-	5.58(2)	16	8.34	3.222	9.79(3)	24	15.35	3.013
HCN	2.62(1)	8	4.15	2.811	6.55(2)	17	9.71	2.921	-	-	-	-
H <sub>2</sub> O	3.66(1)	9,10	4.63	2.696	7.98(3)	18	10.66	2.808	9.9(2)	25	14.96	2.804
H <sub>3</sub> N	4.61(1)	11	6.59	2.679	12.73(2)	19	17.85	2.681	18.5(4)	26	28.17	2.601

<sup>a</sup>Values of *k*<sub>σ</sub> are either taken directly from the reference having the number indicated in columns 3,7 or 11, as appropriate, or are recalculated from the centrifugal distortion constant *D*<sub>*J*</sub> or *Δ*<sub>*J*</sub> given therein by using eq.(1) or (2). The quoted error is that transmitted by the error in the distortion constant. *D*<sub>σ</sub> and *r*(Z...X<sub>i</sub>) are equilibrium values calculated *ab initio* at the CCSD(T)(F12c)/cc-pVDZ-F12 level of theory (see text). *r*(Z...X<sub>i</sub>) is the distance from the acceptor atom/centre Z in the Lewis base B to the inner halogen atom X<sub>i</sub>

Table 2'. Some observed and calculated properties of halogen-bonded complexes B $\cdots$ XY involving polar dihalogen molecules XY (including CABS singles in calculating BSSE corrections for Br- and I- containing species)<sup>a</sup>

Lewis base B	Dihalogen molecules XY											
	Chlorine monofluoride ClF				Bromine monochloride BrCl				Iodine monochloride ICl			
	$k_{\sigma}/(\text{N m}^{-1})$	Ref.	$D_{\sigma}/\text{kJ mol}^{-1}$	$r(\text{Z}\cdots\text{X})/\text{\AA}$	$k_{\sigma}/(\text{N m}^{-1})$	Ref.	$D_{\sigma}/\text{kJ mol}^{-1}$	$r(\text{Z}\cdots\text{X})/\text{\AA}$	$k_{\sigma}/(\text{N m}^{-1})$	Ref.	$D_{\sigma}/\text{kJ mol}^{-1}$	$r(\text{Z}\cdots\text{X})/\text{\AA}$
N <sub>2</sub>	5.00(3)	27	6.28	2.918	4.40(2)	35	5.75	3.106	5.37(2)	44	7.22	3.187
OC	7.04(2)	28	10.56	2.772	6.27(5)	36	9.38	3.006	8.00(3)	45	12.99	3.003
C <sub>2</sub> H <sub>2</sub>	10.01(2)	29	13.68	2.859	9.48(6)	37	13.16	3.038	12.12(8)	46	17.52	3.112
C <sub>2</sub> H <sub>4</sub>	11.01(3)	30	17.01	2.730	10.54(1)	38	16.11	2.927	14.0(1)	47	21.99	2.958
H <sub>2</sub> S	13.40(3)	31	18.13	2.835	12.07(10)	39	16.71	3.057	16.55(5)	48	22.73	3.120
H <sub>3</sub> P	-	-	-	-	11.56(7)	40	19.45	2.878	20.7(1)	49	29.13	2.898
HCN	12.33(5)	32	18.42	2.639	11.09(10)	41	17.13	2.826	14.5(1)	50	24.08	2.840
H <sub>2</sub> O	14.24(3)	33	20.14	2.554	12.08(2)	42	18.29	2.735	15.9(2)	51	24.93	2.776
H <sub>3</sub> N	34.3(5)	34	40.43	2.304	26.7(3)	43	34.58	2.532	30.4(3)	52	47.40	2.599

<sup>a</sup>Values of  $k_{\sigma}$  are either taken directly from the reference having the number indicated in columns 3, 7 or 11, as appropriate, or are recalculated from the centrifugal distortion constant  $D_J$  or  $\Delta_J$  given therein by using eq.(1) or (2). The quoted error is that transmitted by the error in the distortion constant.  $D_{\sigma}$  and  $r(\text{Z}\cdots\text{X}_i)$  are equilibrium values calculated *ab initio* at the CCSD(T)(F12c)/cc-pVDZ-F12 level of theory (see text).  $r(\text{Z}\cdots\text{X})$  is the distance from the acceptor atom/centre Z in the Lewis base B to the inner halogen atom X.

Table 3'. Some observed and calculated properties of hydrogen-bonded complexes B...HX (including CABS singles in calculating BSSE corrections for Br and I containing species)<sup>a</sup>

Lewis base B	Hydrogen halide molecules HX							
	Hydrogen fluoride HF				Hydrogen chloride HCl			
	$k_{\sigma}/(\text{N m}^{-1})$	Ref.	$D_{\sigma}/\text{kJ mol}^{-1}$	$r(\text{Z}\cdots\text{H})/\text{\AA}$	$k_{\sigma}/(\text{N m}^{-1})$	Ref.	$D_{\sigma}/\text{kJ mol}^{-1}$	$r(\text{Z}\cdots\text{H})/\text{\AA}$
N <sub>2</sub>	5.13(3)	53	9.26	2.099	2.55(1)	62,63	5.12	2.400
OC	8.48(9)	54	14.21	2.103	3.88(1)	64	7.78	2.393
C <sub>2</sub> H <sub>2</sub>	-	-	-	-	6.4(3)	65	11.03	2.378
C <sub>2</sub> H <sub>4</sub>	-	-	-	-	5.88(16)	66	11.32	2.396
H <sub>2</sub> S	12.0(2)	55,56	20.36	2.284	6.81(1)	67	12.73	2.480
H <sub>3</sub> P	10.94(4)	57	19.38	2.354	6.01(2)	68	11.89	2.569
HCN	18.26(5)	58	30.33	1.859	9.25(4)	69	18.33	2.092
H <sub>2</sub> O	24.51(2)	59,60	35.34	1.721	12.72(12)	70	21.34	1.912
H <sub>3</sub> N	32.8	61	50.98	1.703	18.2(3)	71	32.72	1.820
	Hydrogen bromide HBr				Hydrogen iodide HI			
	$k_{\sigma}/(\text{N m}^{-1})$	Ref.	$D_{\sigma}/\text{kJ mol}^{-1}$	$r(\text{Z}\cdots\text{H})/\text{\AA}$	$k_{\sigma}/(\text{N m}^{-1})$	Ref.	$D_{\sigma}/\text{kJ mol}^{-1}$	$r(\text{Z}\cdots\text{H})/\text{\AA}$
N <sub>2</sub>	1.92(1)	72	4.19	2.503	-	-	-	-
OC	2.99(1)	73	6.33	2.489	1.713(1)	81	4.38	2.675
C <sub>2</sub> H <sub>2</sub>	5.39(2)	74	9.54	2.440	-	-	-	-
C <sub>2</sub> H <sub>4</sub>	5.21(2)	75	9.94	2.456	-	-	-	-
H <sub>2</sub> S	5.86(2)	76	11.12	2.526	4.02(1)	82	8.04	2.670
H <sub>3</sub> P	5.05(1)	77	10.35	2.618	3.409(2)	83	7.48	2.778
HCN	7.64(2)	78	15.28	2.161	4.44(1)	84	10.93	2.319
H <sub>2</sub> O	10.06(15)	79	17.85	1.969	6.64(1)	85	12.52	2.117
H <sub>3</sub> N	13.4(3)	80	29.05	1.800	7.18(6)	86	20.45	1.926

<sup>a</sup>Values of  $k_{\sigma}$  are either taken directly from the reference having the number indicated in columns 3 or 7, as appropriate, or are recalculated from the centrifugal distortion constant  $D_J$  or  $A_J$  given therein by using eq.(1) or (2). The quoted error is that transmitted by the error in the distortion constant.  $D_{\sigma}$  and  $r(\text{Z}\cdots\text{H})$  are equilibrium values calculated *ab initio* at the CCSD(T)(F12c)/cc-pVDZ-F12 level of theory (see text).  $r(\text{Z}\cdots\text{H})$  is the distance from the acceptor atom/centre Z of the Lewis base B to the hydrogen atom H.