

# A new way to enhance porosity and Y-faujasite percentage of *in-situ* crystallized FCC catalyst

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## Supplement Information

**Table S1** Chemical composition and properties of the kaolin clay

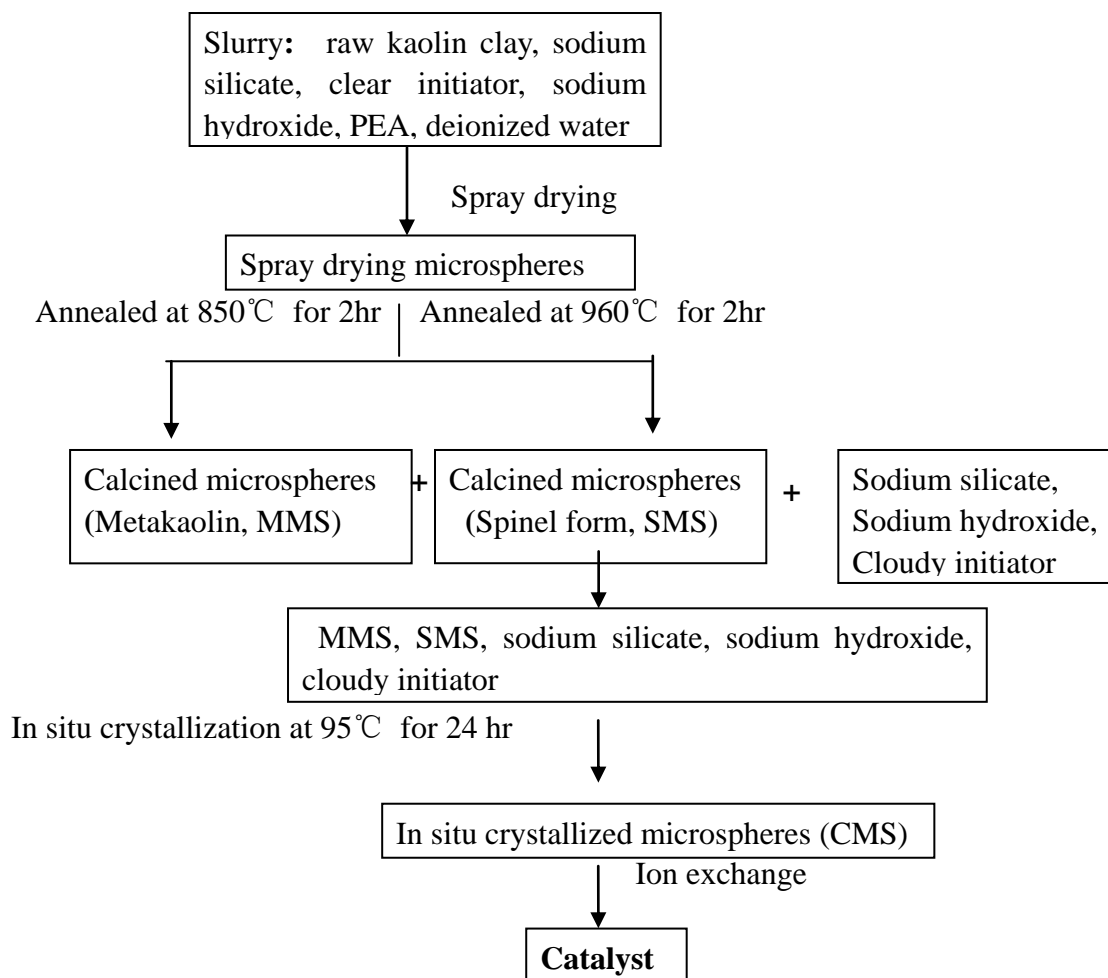
<b>Index</b>	<b>Value</b>
<b>Chemical composition:</b>	
H <sub>2</sub> O(wt%)	14.45
Al <sub>2</sub> O <sub>3</sub> (wt%)	37.36
SiO <sub>2</sub> (wt%)	46.69
Na <sub>2</sub> O(wt%)	0.16
K <sub>2</sub> O(wt%)	0.24
Fe <sub>2</sub> O <sub>3</sub> (wt%)	0.40
TiO <sub>2</sub> (wt%)	0.28
<b>XRD phases:</b>	
Kaolin (wt%)	86
Quartz (wt%)	0.5
<b>Size distributions:</b>	
D(v,10)(μm)	1.94
D(v,50)(μm)	3.33
D(v,90)(μm)	6.03
D(v,98)(μm)	9.40

**Table S2** Properties and composition of the mixed FCC feedstock

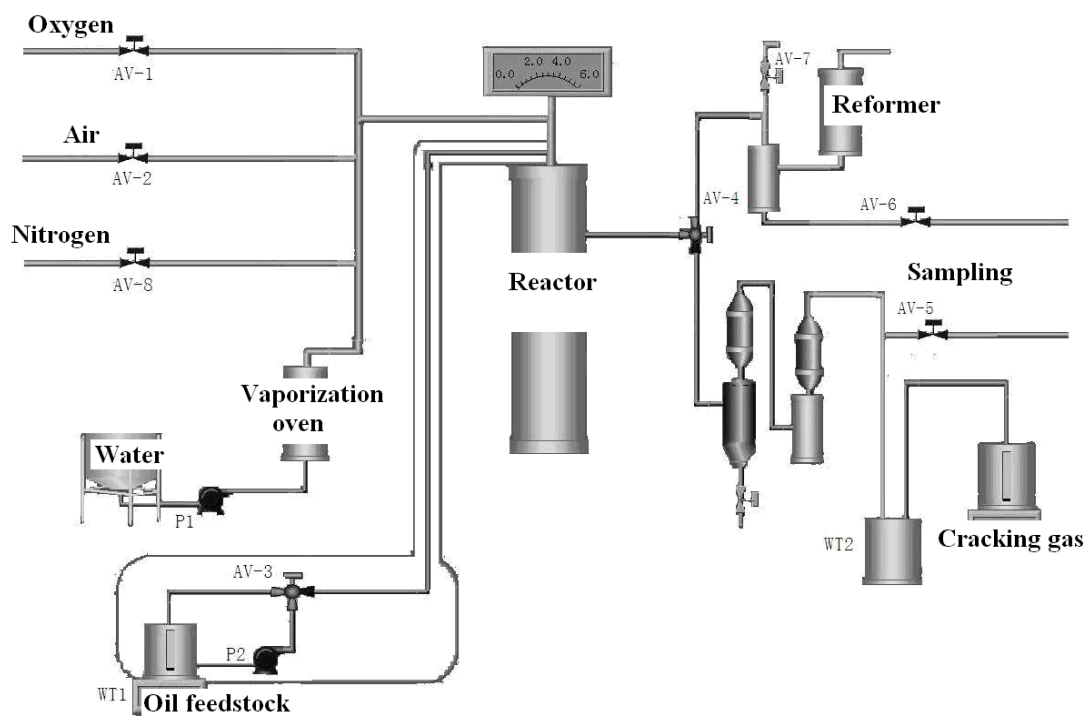
	<b>Vacuum gas oil (VGO)</b>	<b>Vacuum tower bottom (VTB)</b>
Average molecular weight	418	988
Density (g·cm <sup>-3</sup> , 20°C)	0.8726	0.9182
Solidifying point (°C)	47	–
Carbon residue		7.73
<b>Composition (%)</b>		
Saturated hydrocarbon	75.2	28.8
Aromatics	19.2	39.5
Colloidal hydrocarbons	5.5	30.7
Asphaltine	0.1	1
<b>Elements (%)</b>		
C	86.61	86.97
H	13.77	12.8
S	0.21	0.26
N	0.083	0.37
Basic nitrogen (ppm)	296	1970
<b>Metal contents (ppm)</b>		
Ni	0.2	8
V	<0.1	0.1
<b>Distillation (°C)</b>		
Initial	312	424
10%	400	
50%	454	
90%	515	

**Table S3** Chemical composition, physical properties and size distribution of the catalyst CAT-1 and CAT-0

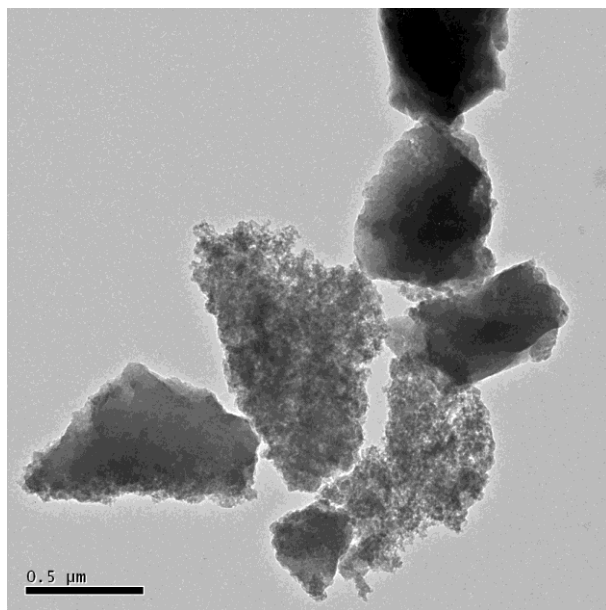
Catalysts	CAT-1	CAT-0
Al <sub>2</sub> O <sub>3</sub> (wt%)	39.75	38.56
SiO <sub>2</sub> (wt%)	53.49	53.92
Fe <sub>2</sub> O <sub>3</sub> (wt%)	0.16	0.21
La <sub>2</sub> O <sub>3</sub> (wt%)	5.84	5.57
Na <sub>2</sub> O (wt%)	0.33	0.3
Attrition Index (wt %·h <sup>-1</sup> )	2.2	2.0
P.V. (mL·g <sup>-1</sup> )	0.48	0.36
S.A.(m <sup>2</sup> ·g <sup>-1</sup> )	410	297
<b>Size distribution</b>		
<40μm (V%)	19.6	18.8
40–110μm (V%)	64.7	64.4
>110μm (V%)	15.7	16.8



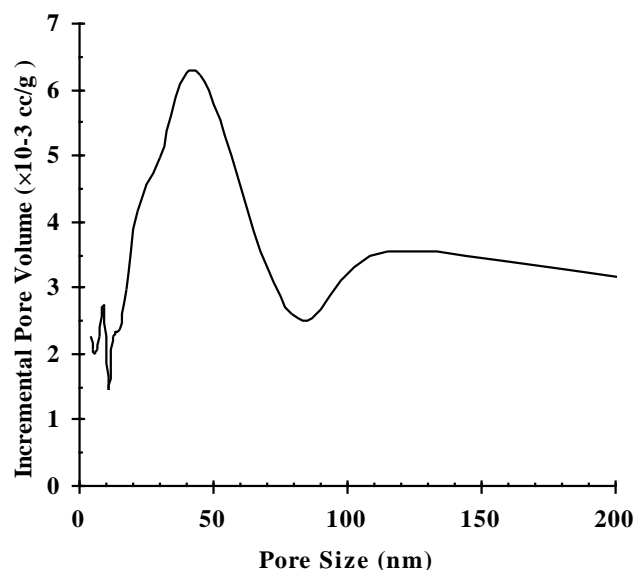
**Fig. S1** Schematic diagram of the typical synthesis procedure of the *in-situ* crystallized FCC catalyst.



**Fig. S2** Schematic diagram of the fixed fluiding bed.



**Fig. S3** TEM image of the physically and chemically treated coconut shell.



**Fig. S4** The pore size distribution of the pretreated coconut shell.

## Attrition index testing

Attrition index were determined by “Tube Attrition Tester” as shown in Figure S4. The samples (i.e. micro spheres) to be tested were dried at 350°C for 30min, then cooled down to room temperature. 10g of microspheres was placed in tube 1 of the tester. The moisture saturated air was compressed in tube 1 at a flowing velocity of  $20\text{L}\cdot\text{min}^{-1}$ , the microspheres in the tube 1 will fluidize and rub against each other or against the tube surface, producing powder due to attrition. The produced powder was blown into and then dropped down in tube 3 through tube 2. The remained sample in the tube 1 were carefully collected after 4h of running, dried at 350°C for 30min, then cooled down and weighted as  $W(\text{g})$ . The attrition index(AI) is expressed as

$$\text{AI} = \frac{W(\text{g})}{4W(0)} \times 100\% (\text{h}^{-1})$$

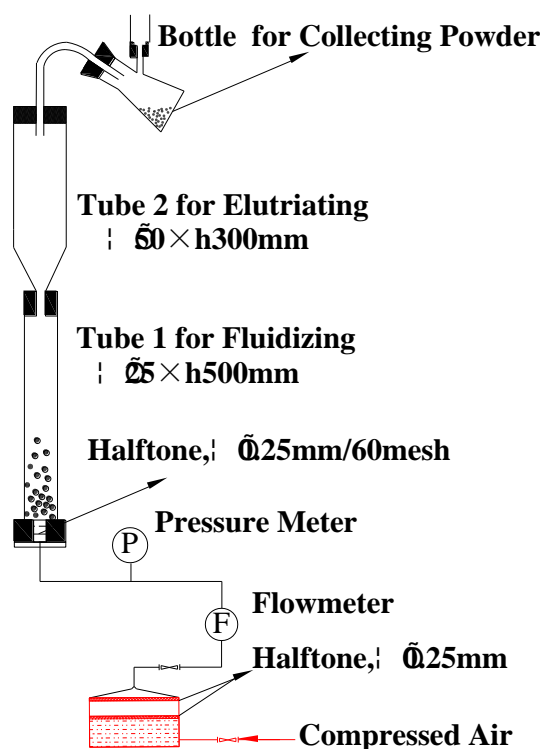
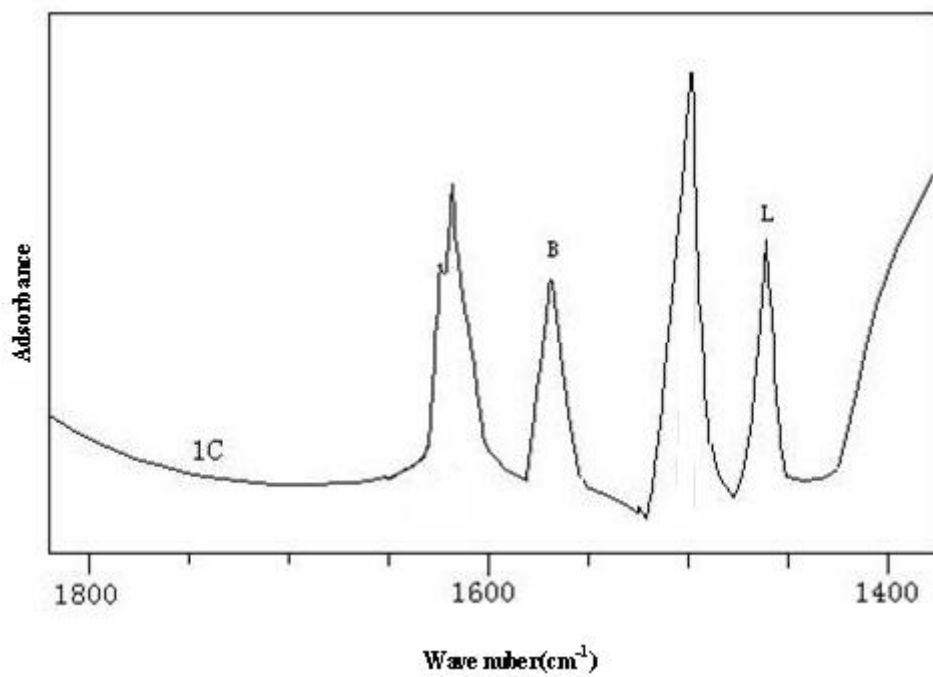


Fig. S5 Tube attrition tester





**Fig.S6** FT-IR spectra of pyridine adsorbed commercial catalyst CC-15 at 200°C.