

Electronic Supplementary Information

**Supramolecular optimization of the visual contrast in a colorimetric chemosensing assay
that releases resorufin dye**

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Materials and Instrumentation

All the solvents and chemicals were purchased from Sigma-Aldrich, Alfa-Aesar, or VWR international and used without further purification unless otherwise stated. ^1H and ^{13}C spectra were recorded on Bruker AVANCE III HD 400, 500 MHz spectrometers. Mass spectrometry (MS) was performed using a Bruker microTOF II spectrometer. Synthesized compounds were purified using Biotage flash purification system with SNAP Ultra flash chromatography cartridges.

N-acetyl- β -D-glucosaminidase (NAG) Stock Solution

A 0.9 mg/mL solution of NAG enzyme (Sigma-Aldrich) from bovine kidney was prepared in 1 M PBS Buffer + 0.1 M BSA and the vendor's guarantee of enzymatic activity was confirmed using the standard chromogenic substrate p-nitrophenyl-N-acetyl- β -D-glucosaminide.

HOCl/ OCl $^-$ Stock Solution

Hypochlorite stock solution was prepared using a previously reported method.¹ To a 1500 μL solution of 154 mM NaCl, 250 μL of 10-14% w/w NaOCl was added followed by dropwise addition of 6M HCl to obtain a pH range of 3.92. The concentration of active total chlorine species in solution expressed as $[\text{HOCl}]_T$ (where $[\text{HOCl}]_T = [\text{HOCl}] + [\text{Cl}_2] + [\text{Cl}_3^-] + [\text{OCl}^-]$) in HPLC Grade water was determined by converting all the active chlorine species to OCl $^-$ with 0.1 M NaOH and measuring the concentration of OCl $^-$. The concentration of OCl $^-$ was determined spectrophotometrically at 292 nm ($\epsilon = 362 \text{ M}^{-1} \text{ cm}^{-1}$) with a UV-visible spectrophotometer. Calculation: $A = \epsilon cl$; where $l = 1 \text{ cm}$, $A = 0.6359$, $\epsilon = 362 \text{ M}^{-1} \text{ cm}^{-1}$. Thus, $c = 1.76 \text{ mM HOCl/ OCl}^-$

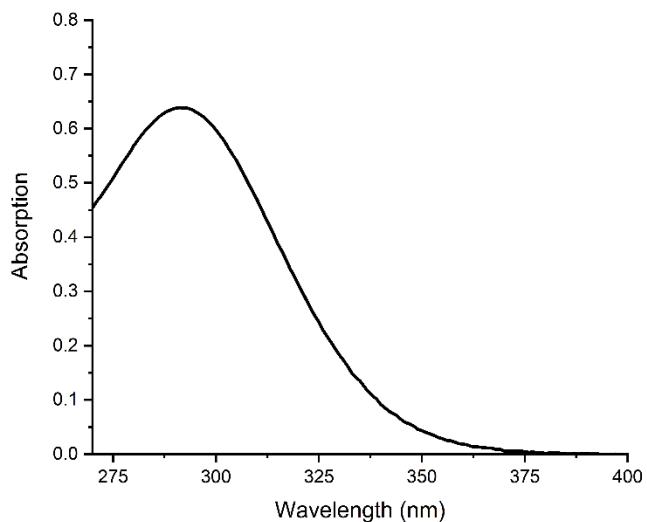


Figure S1: Absorption spectrum of HOCl/ OCl $^-$ for stock solution concentration determination.

Synthesis

The chemosensor **RT-1** and enzyme substrate **NHPO** were synthesized as previously described,^{2,3} and the following ¹H NMR and HR-MS data demonstrate high purity. Tetralactam macrocycles **M1** and **M2** were synthesized as part of previous studies^{4,5} and the purity was confirmed by ¹H NMR.

RF-TBA: Resorufin sodium salt (50 mmol, 10.6 mg) and 40 % wt tetrabutylammonium hydroxide solution (50 mmol, 33 μ L) were dissolved in 50 mL of PBS. The resulting mixture was extracted with chloroform (3×50 mL). The combined chloroform layers were dried under vacuum to obtain pure **RF.TBA** as a dark pink solid.

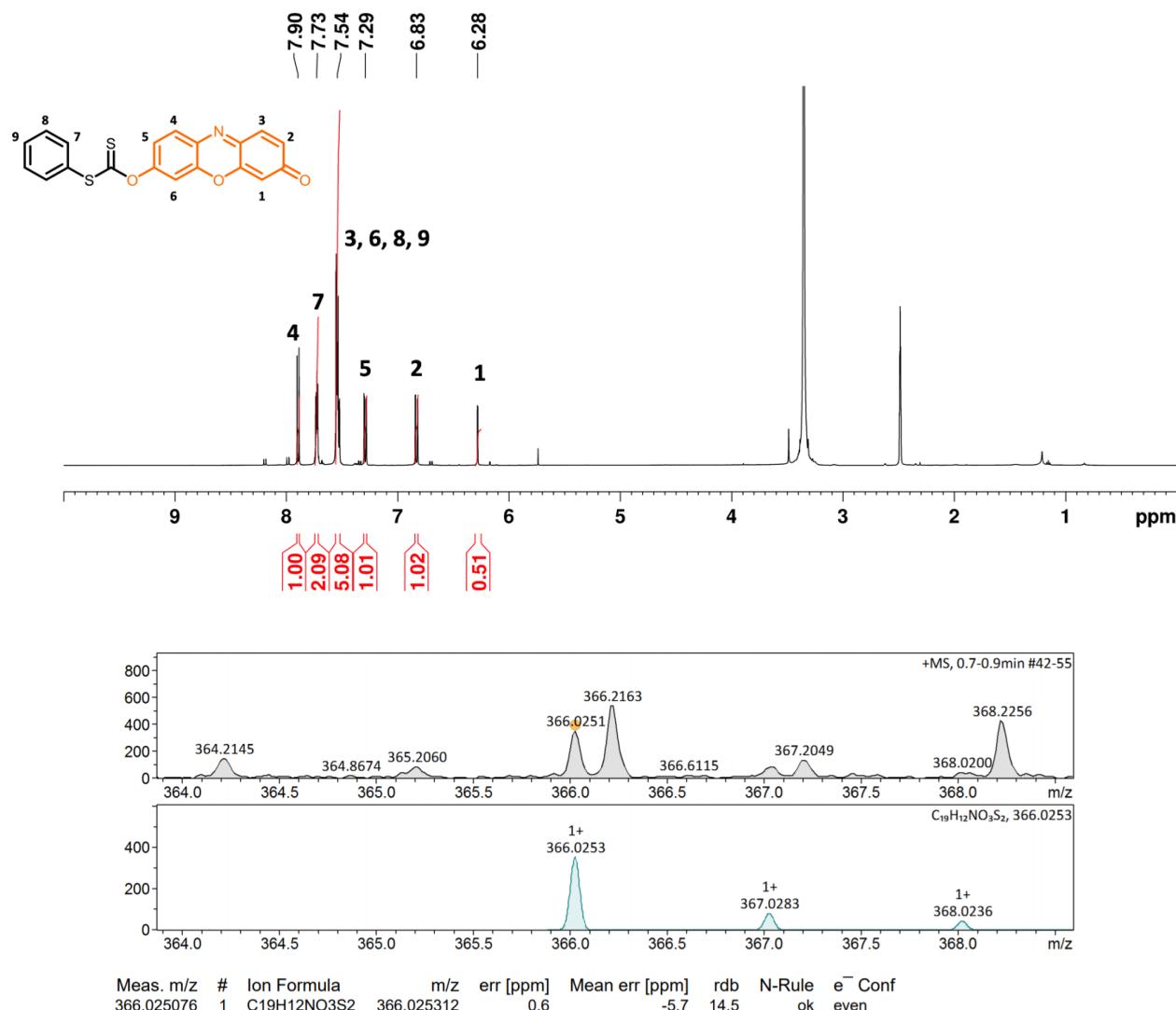


Figure S2. ¹H NMR (500 MHz; DMSO-d₆; Me₄Si) and HR-ESI mass spectrum of **RT-1**.

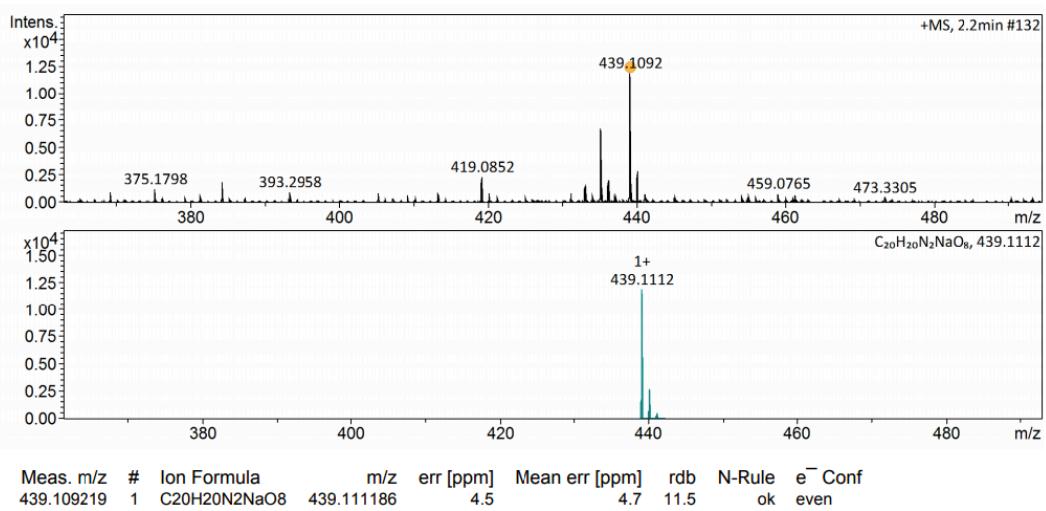
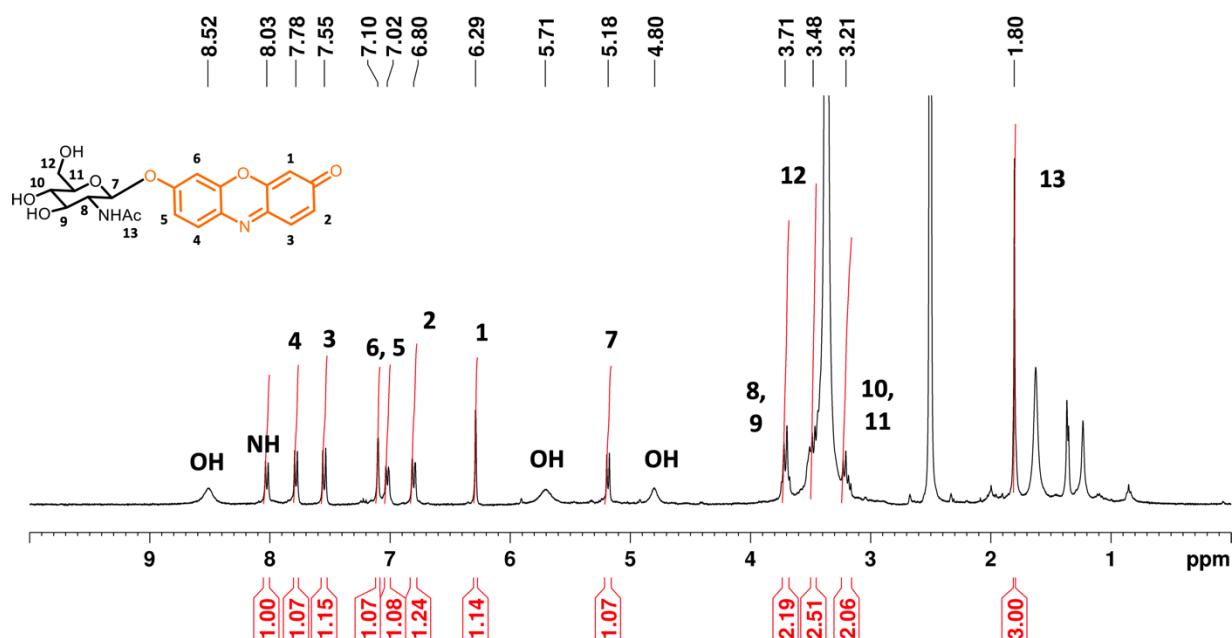


Figure S3. ¹H NMR (400 MHz; DMSO-d₆; Me₄Si) and HR-ESI mass spectrum of NHPO. The broad OH peaks in the ¹H NMR spectrum are due to exchange promoted by adventitious water in the DMSO-d₆.

¹H NMR Titration Data

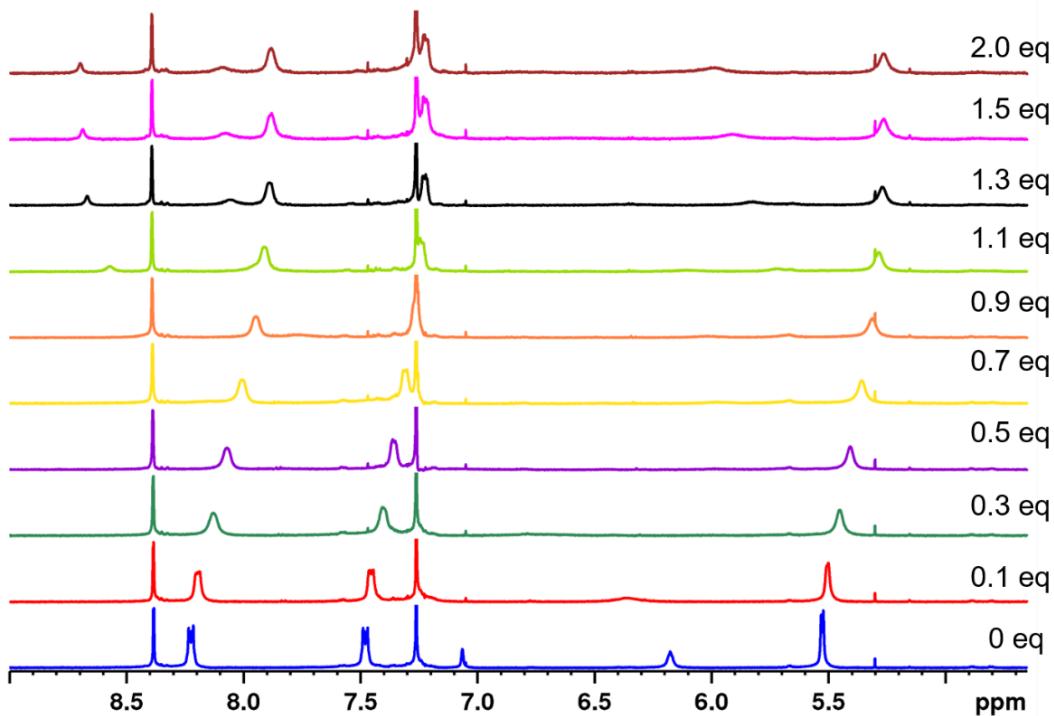


Figure S4. ¹H NMR titration (500 MHz, CDCl₃, 25° C) that added aliquots from a stock solution containing 10 mM **RF** (tetrabutylammonium salt)/ 0.5 mM **M1** to a solution of **M1** (0.5 mM).

K_a Determination by Fluorescence Titration

Previously described titration method was employed.⁶ Stock solutions of the guest, **RF** (1 mM) and host, **M2** (1 mM) were made in pure water. A solution of the guest was placed in a cuvette (10 μ M) and aliquots of the host (**M2**) were added fluorescence spectra were acquired (ex: 540 nm, em: 585 nm). The data was plotted and association constant for **RF** binding to **M2** was determined by non-linear squares fitting of the titration points to a model for 1:1 binding within the Origin software.⁷

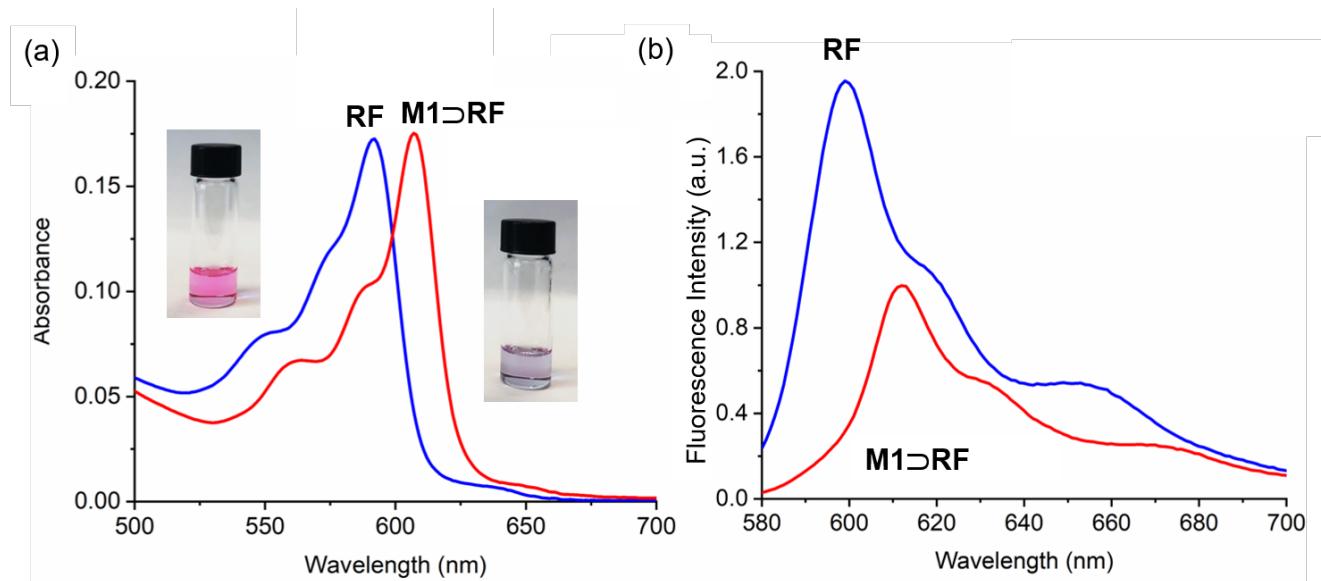


Figure S5. (a) Absorption (b) fluorescence emission of 10 μM **RF** (tetrabutylammonium salt) and **M1-DOPO-RAFT** in chloroform at 25°C.

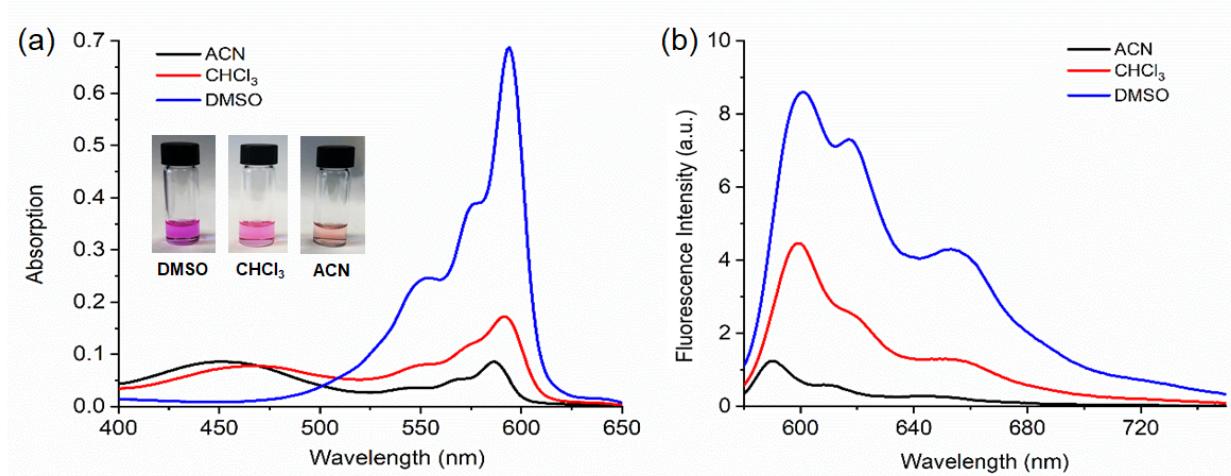


Figure S6. Absorption and emission spectra of 10 μM **RF** (tetrabutylammonium salt) in different organic solvents at 25°C, along with photographs of the solutions.

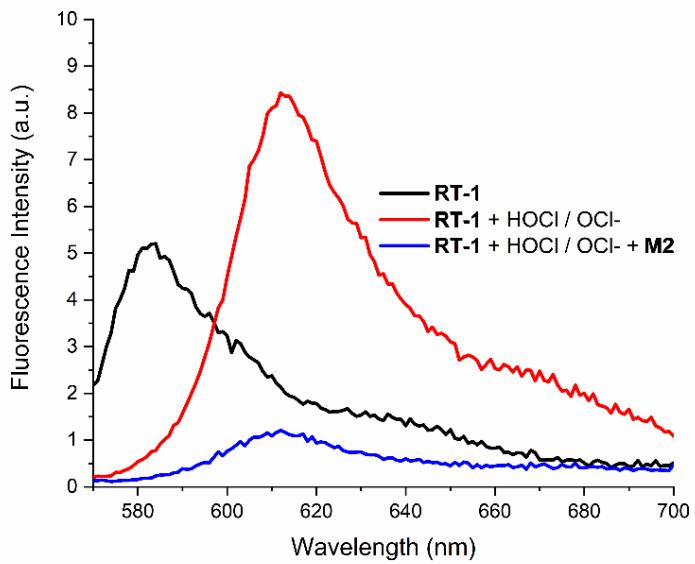


Figure S7. Fluorescence spectra ($\lambda_{\text{ex}} = 550 \text{ nm}$) of a sample, initially containing **RT-1** (50 μM , black line), and 3 minutes after addition of HOCl/OCl⁻ (5 μM , red line), or 3 minutes after a two-step addition sequence of HOCl/OCl⁻ (5 μM) and then **M2** (500 μM) (blue line). In 200 mM PBS, pH 7.4 at 25°C.

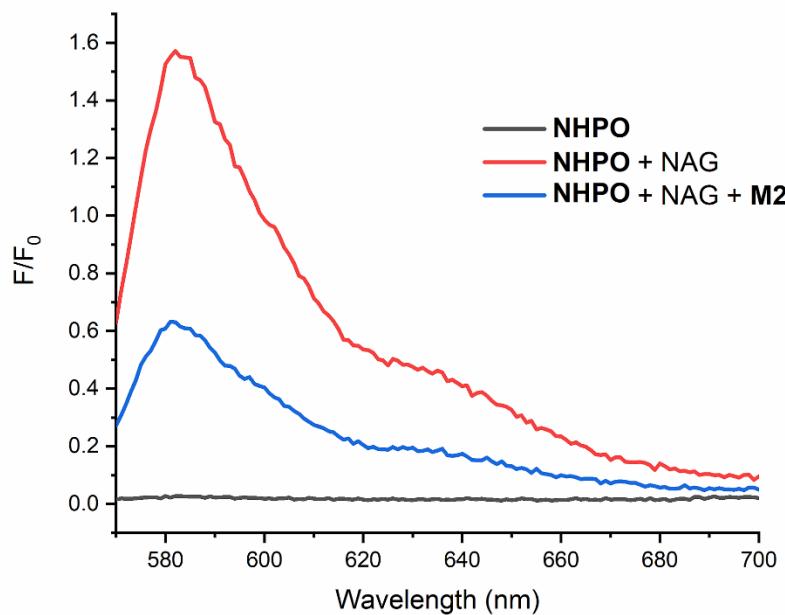


Figure S8. Fluorescence spectra ($\lambda_{\text{ex}} = 550 \text{ nm}$) of a sample initially containing **NHPO** (50 μM , black line), 30 minutes after addition of 0.9 $\mu\text{g/mL}$ NAG (red line), or 45 minutes after a two-step addition sequence of 0.9 $\mu\text{g/mL}$ NAG and then **M2** (500 μM) (blue line). In 100 mM PBS + 100 μM BSA, pH 7.4 at 25°C.

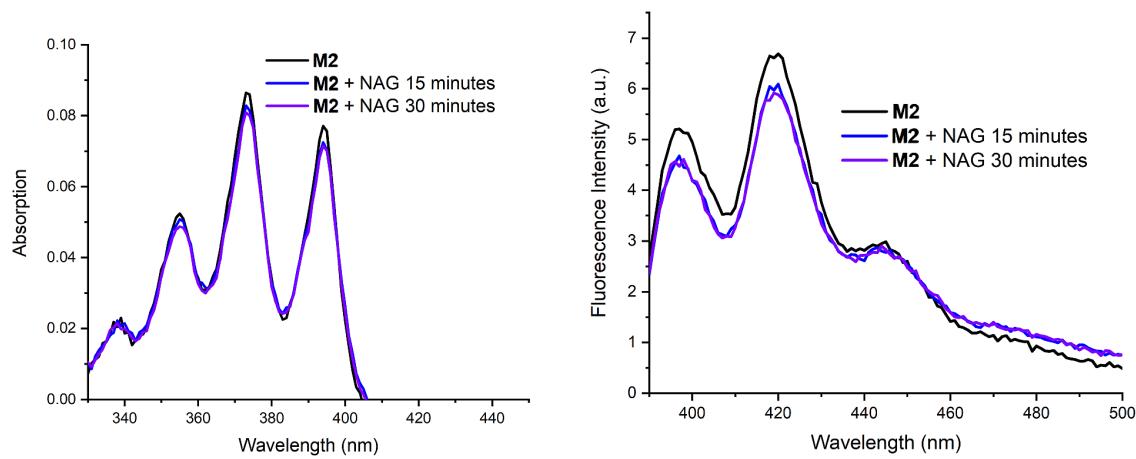


Figure S9. Absorption and fluorescence emission (ex: 370 nm, em: 390 nm) of a solution containing 15 μM **M2** in the presence and absence of (0.1 $\mu\text{g}/\text{mL}$ NAG enzyme plus $\sim 10 \mu\text{M}$ BSA), in water and 25°C. The very small intensity decrease upon protein addition is due to sample dilution, and it appears there is negligible interaction of NAG or BSA with **M2**.

Molecular Modeling

The semiempirical PM7 method was employed within the MOPAC program. (J. J. P. Stewart, MOPAC; Stewart Computational Chemistry: Colorado Springs, CO, 2008.) The dielectric constant of the solvent was set at 78.4 for water and 25 °C. Solubilizing groups are shortened to hydrogens.

Cartesian Coordinates at the PM7 Level

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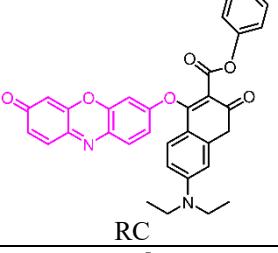
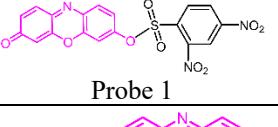
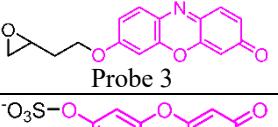
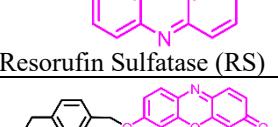
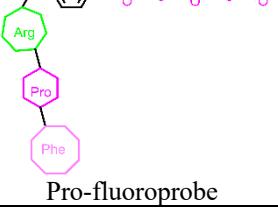
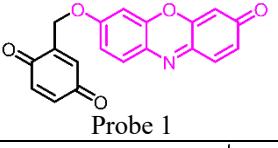
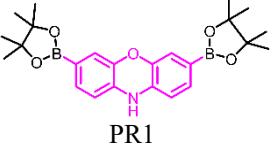
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C      4.52278289 +1   -0.98838362 +1    3.30059312 +1
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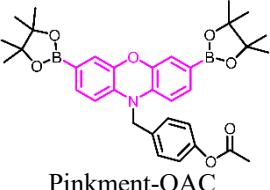
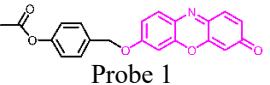
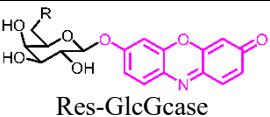
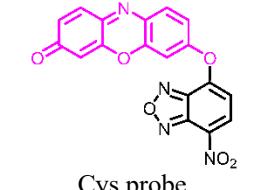
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C	7.84407762	+1	-0.50023230	+1	1.60726775	+1
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H	4.55184649	+1	-2.04663149	+1	3.03732730	+1
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H	4.88030841 +1	3.94571460 +1	1.48305548 +1
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H	10.96703491 +1	4.54264697 +1	0.48862585 +1
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H	2.34140368 +1	-2.23234920 +1	0.23727619 +1
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H	4.51120938 +1	1.26814086 +1	-4.23614250 +1
H	5.39113145 +1	4.35269899 +1	-1.31209333 +1

Table S1: Abridged collection of enzyme substrates and chemosensors that release resorufin (RF).

Probe	Analyte	Method of Detection	Reference Number
Resorufin Thionocarbonate (RT)	Mercury Hg ²⁺	Chromogenic	8
Resorufin β-D-glucuronide (REG)	E. coli	Chromogenic	9
Novel Probe 1	Alkaline Phosphatase (ALP)	Fluorescence	10
Resorufin turn on Probe (RTP-1)	Hydrazine (N ₂ H ₄)	Fluorescence	11
Resorufin-β-D-Galactopyranoside	Biotinylated DNA	Fluorescence	12
F-Chemodosimeter 1	Fluorine (F)	Chromogenic/ Fluorescence	13
Sulfite Selective Probe	Sulfite (SO ₃ ²⁻)	Chromogenic/ Fluorescence	14
R1	Perborate (BO ₃ ³⁻) / Hydrazine (N ₂ H ₄)	Chromogenic/ Fluorescence	15,16
Re-SS	Polysulfides	Fluorescence	17
Ozone Probe 1	Ozone (O ₃)	Chromogenic/ Fluorescence	18
PC1	Hydrogen Peroxide (H ₂ O ₂)	Fluorescence	19

	Hydrogen Sulfide (H ₂ S)	Fluorescence	20
	Mercury (Hg ²⁺)	Chromogenic/Fluorescence	21
	Hydrogen sulfide (H ₂ S), Cysteine (Cys), Homocysteine (Hcy), Glutathione (GSH)	Fluorescence	22
	Acetyl-cholinesterase	Fluorescence	23
	γ -glutamyl cyclotransferase	Fluorescence	24
	Epoxy-hydrolase	Fluorescence	25
	Sulfatase	Fluorescence	26
	Thrombin protease	Fluorescence	27
	Glucose	Fluorescence	28
	Reactive Oxygen, Nitrogen (ONOO ⁻)	Fluorescence and Colorimetric	29

 Pinkment-OTBS	(ONOO ⁻ and F ⁻)	Fluorescence and Colorimetric	29
 Pinkment-OAC	Esterase and H ₂ O ₂	Fluorescence and Colorimetric	29
 Probe 1	Carboxylesterase	Fluorescence	30
 NBCD	H ₂ O ₂	Fluorescence	31
 Res-GlcGcase	β -Glucocerebrosidase	Fluorescence	32
 Cys probe	Cysteine	Fluorescence	33
 AR	Phosphite and Nickel oxide	Fluorescence	34

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