

**Electronic Supplementary Information (ESI).**

Table S1. Simultaneous determination result of POD and AsA in plant samples

Samples (Juice yield / ml g <sup>-1</sup> ) <sup>a</sup>	Dilution	Signals for POD and AsA		Reaction solution conc. <sup>b</sup>		Sample solution conc. <sup>c</sup>	
		$k_{\text{POD}} \pm \text{SD}$	$\Delta A_{\text{AsA}} \pm \text{SD}$	$E_{\text{POD}} / \text{U L}^{-1}$	$c_{\text{AsA}} / \text{mg L}^{-1}$	$E_{\text{POD}} / \text{U L}^{-1}$	$c_{\text{AsA}} / \text{mg L}^{-1}$
white radishes (0.49)	5	0.871±0.006	0.057±0.001	33.94	2.24	8486.2	59.3
Chinese cabbage (0.62)	2	0.723±0.024	0.079±0.000	28.21	3.20	2821.3	119.9
green jujube (0.38)	1	0.209±0.007	0.179±0.002	8.30	7.44	415.1	372.2

Note: *a.* Juice yield = sample (mL) / sample quality (g);

*b.* Standard curve:  $k_{\text{POD}} = 0.0258E_{\text{POD}} - 0.0047$ ,  $\Delta A_{\text{AsA}} = 0.0236c_{\text{AsA}} + 0.0034$ ;

*c.* Sample content = reaction solution conc. × reaction volume (3mL) / sample volume (0.06mL) × dilution ratio.

Table S2. Recovery tests for AsA and comparison standard curve method (SCM) and standard addition method (SAM)

samples	added AsA / mg L <sup>-1</sup>	$\square A_{\text{AsA}} (\pm \text{SD})$	$c_{\text{AsA}}^a$ / mg L <sup>-1</sup>	recovery / %	$c_{\text{AsA}} (\text{SAM})$ / mg L <sup>-1</sup>	RD <sup>b</sup> / %	<i>t</i> -test
white radishes (20%)	0	0.012 (0.001)	18.14	-	17.54	3.35	0.31
	25	0.024 (0.000)	43.54	101.59			
	50	0.036 (0.001)	68.43	100.58			
	75	0.047 (0.001)	91.63	97.99			
	100	0.058 (0.000)	115.36	97.22			
Chinese cabbage (50%)	0	0.034 (0.001)	65.51	-	63.56	2.98	0.08
	50	0.058 (0.001)	114.88	98.74			
	100	0.082 (0.002)	166.31	100.80			
	150	0.106 (0.001)	217.90	101.60			
	200	0.128 (0.001)	263.98	99.24			
green jujube (50%)	0	0.093 (0.001)	190.08	-	6.08	3.93	0.08
	50	0.116 (0.001)	239.41	98.65			
	100	0.141 (0.001)	291.10	101.02			
	200	0.187 (0.001)	389.09	99.50			
	400	0.278 (0.004)	582.63	98.14			

*a.*  $c_{\text{AsA}} = [(k_{\text{POD}} - 0.0034) / 0.0236] \times \text{reaction volume (3mL)} / \text{sample volume (0.06mL)}$ ;

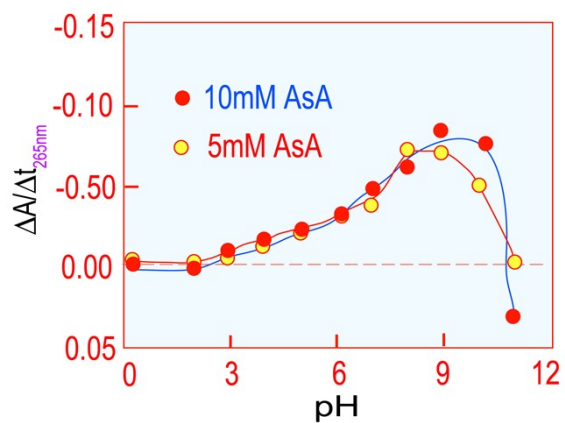
*b.*  $\text{RD} = | \text{SAM} - \text{SCM} | / \text{SCM} \times 100\%$ .

Table S3. Recovery tests for POD and comparison standard curve method (SCM) and standard addition method (SAM)

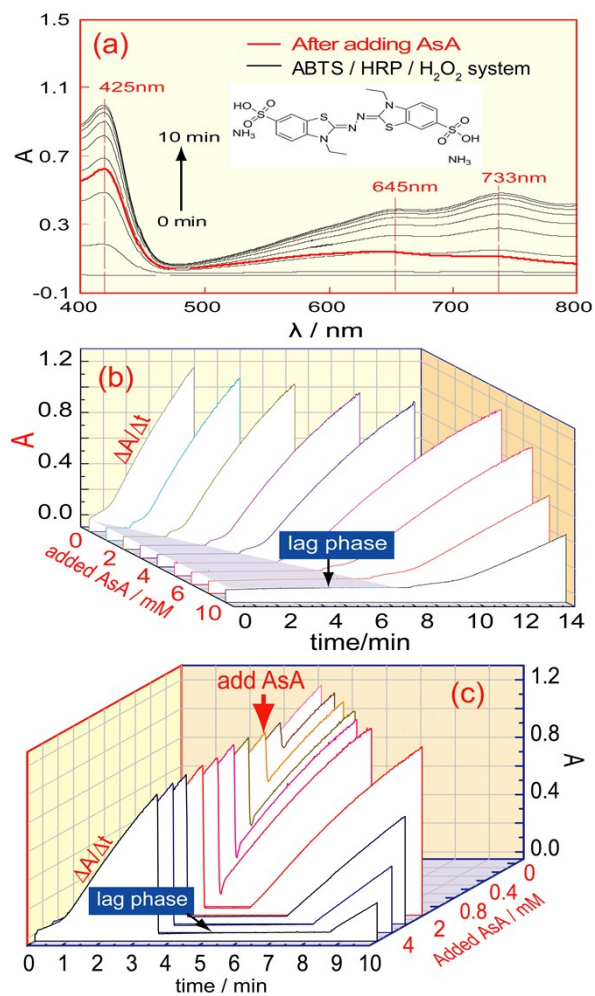
samples	Added POD / U L <sup>-1</sup>	$k_{\text{POD}} (\pm \text{SD})$	$E_{\text{POD}}^a$ /U L <sup>-1</sup>	recovery / %	$E_{\text{POD}} (\text{SAM})$ / U L <sup>-1</sup>	RD <sup>b</sup> / %	<i>t</i> -test
white radishes (5%)	0	0.214 (0.002)	424.76	-	416.48	1.95	0.37
	250	0.347 (0.002)	681.78	102.81			
	500	0.469 (0.009)	918.02	98.65			
	750	0.606 (0.002)	1183.62	101.18			
	1000	0.740 (0.002)	1442.25	101.75			
Chinese cabbage(20%)	0	0.279 (0.002)	549.39	-	553.15	0.69	0.42
	250	0.406 (0.001)	796.09	98.68			
	500	0.529 (0.002)	1034.46	97.01			
	750	0.666 (0.008)	1298.99	99.95			
	1000	0.788 (0.003)	1535.37	98.60			
green jujube (50%)	0	0.102 (0.002)	206.78	-	211.12	2.10	0.46
	100	0.154 (0.002)	307.95	101.17			
	200	0.202 (0.003)	401.36	97.29			
	300	0.258 (0.003)	509.30	100.84			
	400	0.304 (0.004)	597.87	97.77			

*a.*  $E_{\text{POD}} = [(k_{\text{POD}} + 0.0047) / 0.0258] \times \text{reaction volume (3mL)} / \text{sample volume (0.06mL)}$ ;

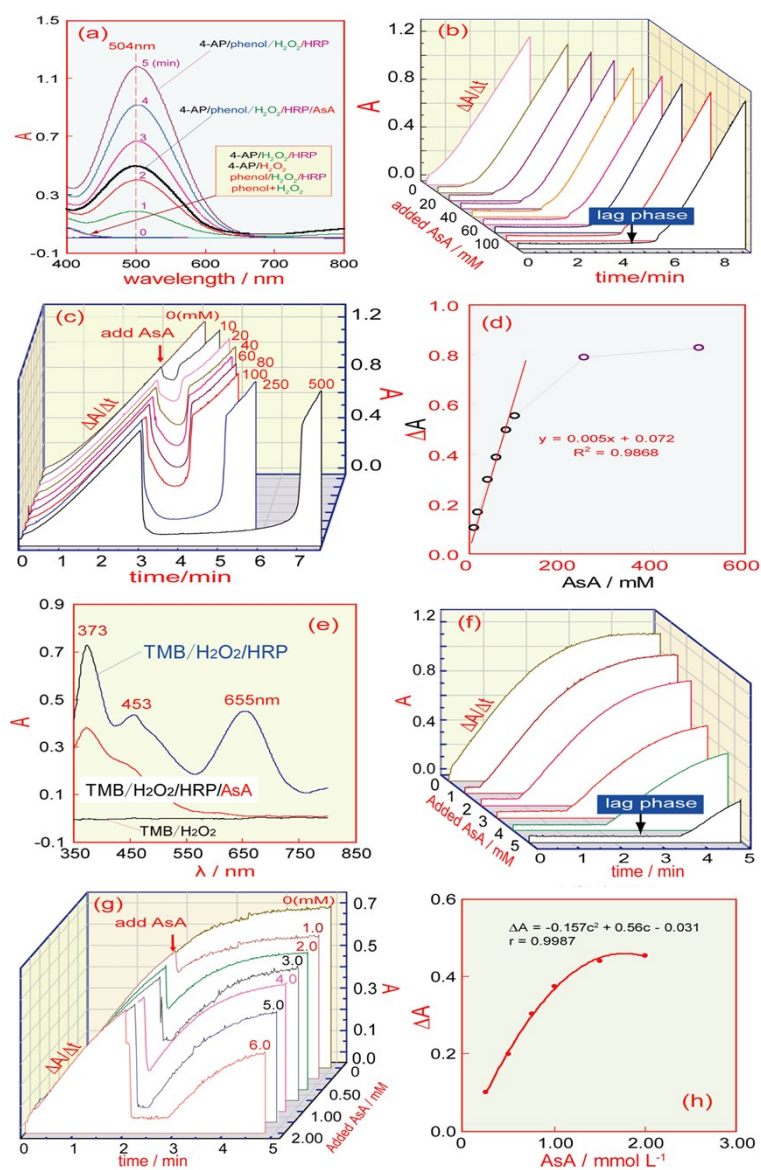
*b.*  $\text{RD} = |\text{SAM} - \text{SCM}| / \text{SCM} \times 100\%$ .



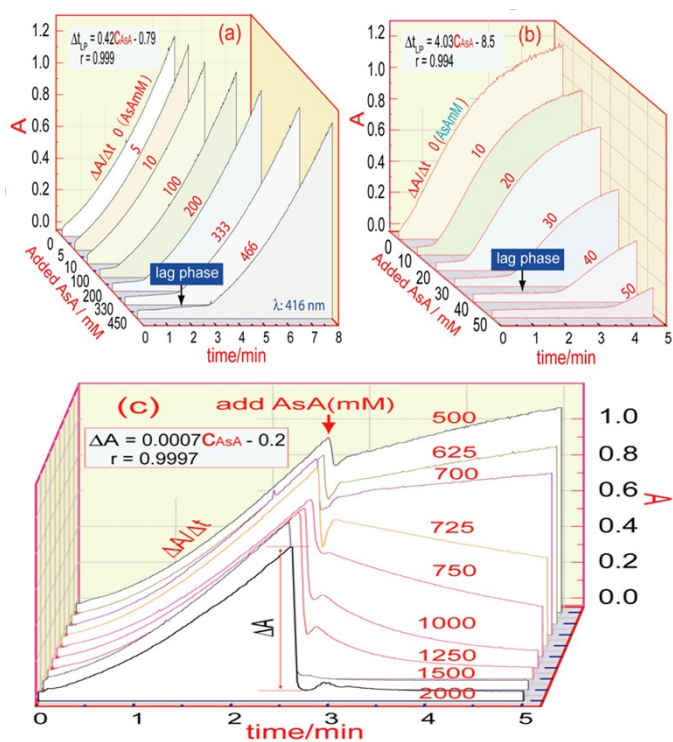
**Fig. S1.** The effect of pH on AsA reduction ability was investigated with the AsA/H<sub>2</sub>O<sub>2</sub> reaction (265 nm). Specific experimental process: Use of 1 M of HCl or NaOH prepared a buffer solution having appropriate pH (0.3~13), and then in 2.975 mL of the buffer solution, 0.025 mL of 5 mmol/L or 10 mmol/L AsA was added to form a reaction solution which was continuously detected for 5 min at the characteristic wavelength (265 nm) of AsA to obtain its absorbance ( $\Delta A/\Delta t$ ).



**Fig. S2.** Effects of AsA on the ABTS/POD/H<sub>2</sub>O<sub>2</sub> reaction (425 nm). (a). Spectrum scanning curves; (b). Lag phases of kinetic curves containing different-concentration AsA; (c). The different-concentration AsA was added into the ABTS/POD/H<sub>2</sub>O<sub>2</sub> solution after reacting for 3.5 min.



**Fig. S3.** Impacts of AsA on two chromogenic reactions. (a). Scanning curves of 4-AP/phenol/POD/H<sub>2</sub>O<sub>2</sub> mixture; (b). Lag phases of kinetic curves with different-concentration AsA; (c). Different concentrations of AsA were added into 4-AP/phenol/POD/H<sub>2</sub>O<sub>2</sub> reaction after reacting for 3 min; (d). Correlations of extinction degree of chromogenic product with the AsA concentration in 4-AP/phenol/POD/H<sub>2</sub>O<sub>2</sub> reaction; (e). Spectrum scanning of TMB/POD/H<sub>2</sub>O<sub>2</sub> reaction; (f). Lag phases of kinetic curves containing AsA; (g). Adding different concentrations of AsA in TMB/POD/H<sub>2</sub>O<sub>2</sub> mixture after reacting for 2 min; (h). Correlation of the extinction degree of the chromogenic product formed in TMB/POD/H<sub>2</sub>O<sub>2</sub> reaction with the AsA concentration.



**Fig. S4.** Effects of AsA on other chromogenic reactions for determining POD activity. (a). Lag phases of kinetic curves in the *o*-phenylenediamine/POD/H<sub>2</sub>O<sub>2</sub> reaction which has different concentrations of AsA; (b). Lag phases of kinetic curves in the pyrogallol/POD/H<sub>2</sub>O<sub>2</sub> reaction containing AsA; (c). Lag phases of kinetic curves in the quinol/POD/H<sub>2</sub>O<sub>2</sub> reaction with different-concentration AsA.