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Supporting Information for

Bottleable NiCl₂(dppe) as a catalyst for the Markovnikov-selective

hydroboration of styrenes with bis(pinacolato)diboron

Toru Hashimoto,* Toshiya Ishimaru, Keisuke Shiota, and Yoshitaka Yamaguchi*

Department of Advanced Materials Chemistry, Graduate School of Engineering,

Yokohama National University, 79-5 Tokiwadai, Hodogaya-ku, Yokohama 240-8501, Japan

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General procedures.

All manipulations involving air- and moisture-sensitive organometallic compounds were performed under an atmosphere of nitrogen dried with SICAPENT (Merck Co., Inc.) using standard Schlenk-line or high-vacuum techniques. All solvents were distilled over appropriate drying agents prior to use. NiCl₂(PPh₃)₂,¹ NiCl₂(PCy₃)₂,¹ NiCl₂(dppe),¹ NiCl₂(dppp),¹ and nickel complexes 1² were prepared according to literature procedures. Vinylarenes 2a, 2b, 2c, 2d, 2h, 13, and 15 were purchased from commercial suppliers and distilled before use. Vinylarenes 2e, 2f, 2g, 2i, 2j, 2k, 2l, and 10 were prepared by Wittig reactions involving the corresponding aldehydes. All other reagents employed in this study were obtained from commercial suppliers and used without further purification.

Proton nuclear magnetic resonance (¹H NMR) and carbon nuclear magnetic resonance (¹³C{¹H} NMR) spectra were recorded using BRUKER DRX-500 and JEOL ECX 500 spectrometers at ambient temperature. ¹H NMR chemical shifts are reported in ppm relative to tetramethylsilane (TMS) as the internal standard (TMS: 0.00 ppm). ¹³C{¹H} NMR chemical shifts are reported in ppm relative to the solvent resonance peak as the internal standard (CDCl₃: 77.0 ppm). Data are presented in the following form: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, sext = sextet, sept = septet, m = multiplet and/or multiplet resonances, br = broad), coupling constants (Hz), and signal area integration in natural numbers.

Analytical thin-layer chromatography (TLC) was performed on alumina plates (Merck, 1.05554.0001), which were visualized by exposure to ultraviolet light (254 nm) and/or by immersion in an acidic staining solution of *p*-anisaldehyde followed by heating on a hot plate. Organic solutions were concentrated by rotary evaporation at *ca*. 30–400 mmHg. Flash column chromatography was performed on Kanto silica gel 60N (spherical, neutral, 63–210 mesh). Gas chromatograms were recorded using a Shimadzu GC-17A gas chromatograph using an ULBON HR-1 capillary column (0.25 ID × 25 m, Shinwa Chemical Industries Ltd.). IR spectra were recorded using a JASCO FT/IR-460 Plus. Characteristic IR absorptions are reported in cm⁻¹. High-resolution mass spectra (HRMS) were recorded using the electron impact (EI) method with a JEOL JMS-700 Mass Spectrometer.

Experimental Procedures

Optimization of Reaction Conditions

General Procedure for the Hydroboration of 4-Methoxystyrene (2a) Catalysed by Nickel Catalysts (Table 1).

A Schlenk tube was charged with the nickel catalyst (5 mol%), KOAc (1.00 mmol), toluene (3.0 mL), and MeOH (1.0 mL). The mixture was stirred at room temperature for 5 min. Subsequently, **2a** (0.50 mmol) and **3** (1.00 mmol) were added sequentially, and stirring was continued for 24 h at 25 °C. The resulting mixture was filtered through a short pad of silica gel using EtOAc as the eluent. The yield of the product **7a** and the recovery of **2a** were determined by GC analysis of the crude product using undecane as the internal standard. The filtrate was concentrated under reduced pressure, and 1,1,2,2-tetrachloroethane was added as an internal standard. The yields of the products **4a**, **5a**, and **6a** were determined by ¹H NMR analysis.

General Procedure for the NiCl₂(dppe)-catalysed Hydroboration of 2, 10, 13, and 15 (Table 2 and Scheme 1).

A Schlenk tube was charged with NiCl₂(dppe) (5–10 mol%), KOAc (2.0–3.0 equiv), toluene (3.0 mL), and MeOH (1.0 mL). The mixture was stirred at room temperature for 5 min. Then, **2**, **10**, **13**, or **15** (0.50–1.0 mmol) and **3** (2.0–3.0 equiv) and were added sequentially, and stirring was continued for 24–48 h at 25 °C. The resulting solution was filtered through a pad of silica gel using EtOAc as the eluent, and the resulting filtrates were concentrated. The residue was re-solvated in THF (6.0 mL), and a mixture of aqueous NaOH (1.8 mL, 3.0 M) and hydrogen peroxide (0.20 mL, 30% aqueous solution) was added at 0 °C. The mixture was stirred at room temperature for 0.5–1.5 h and subsequently extracted with EtOAc (3 × 5 mL). The combined organic layers were washed with brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. The crude reaction mixture was purified by column chromatography on silica gel to afford the corresponding alcohol **8** or **14**.

Effect of the Alcohol on the NiCl₂(dppe)-catalysed Hydroboration of 2a.

A Schlenk tube was charged with NiCl₂(dppe) (5 mol%), KOAc (1.00 mmol), toluene (3.0 mL), and alcohol (1.0 mL). The mixture was stirred at room temperature for 5 min. Subsequently, **2a** (0.50 mmol) and **3** (1.00 mmol) were added sequentially, and stirring was continued for 24 h at 25 °C. The resulting mixture was filtered through a short pad of silica gel using EtOAc as the eluent. The yield of the product **7a** and the recovery of **2a** were determined by GC analysis of the crude product using undecane as the internal standard. The filtrate was concentrated under reduced pressure, and 1,1,2,2-tetrachloroethane was added as an internal standard. The yields of the products **4a**, **5a**, and **6a** were determined by ¹H NMR analysis.



^a Determined by NMR analysis using an internal standard.

^b Determined by GC analysis using an internal standard.

Procedure for the NiCl₂(dppe)-Catalyzed Hydroboration of 2a using Different Amounts of 3.

A Schlenk tube was charged with NiCl₂(dppe) (5 mol%), KOAc (1.00 mmol), toluene (3.0 mL), and MeOH (1.0 mL). The mixture was stirred at room temperature for 5 min. Subsequently, **2a** (0.50 mmol) and **3** (0.1–2.0 equiv, 0.05–1.00 mmol) were added sequentially, and stirring was continued for 24 h at 25 °C. The resulting mixture was filtered through a short pad of silica gel using EtOAc as the eluent. The recovery of **2a** was determined by GC analysis of the crude product using undecane as the internal standard. The filtrate was concentrated under reduced pressure, and 1,1,2,2-tetrachloroethane was added as an internal standard. The yield of the product **4a** was determined by ¹H NMR analysis.

MeO	tol 2a	NiCl ₂ (dppe). (5.0 mo 3 (X equiv) KOAc (2.0 equiv) uene/MeOH (3:1, v/v °C, 24 h	DI%) BPin BPin MeO 4a
Entry	3 (X equiv)	Yield of 4a (%) ^a	Recovery of 2a (%) ^b
1	0.1	n.d	>98
2	0.2	n.d	>98
3	0.3	10	90
4	0.4	15	79
5	0.5	29	69
6	0.8	35	59
7	1.0	54	42
8	1.5	76	19
9	2.0	97	0

[°]Determined by NMR analysis using an internal standard.

^b Determined by GC analysis using an internal standard.

Isotope-labelling Experiment (Scheme 2)



A Schlenk tube was charged with NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%), KOAc (98.1 mg, 1.00 mmol), toluene (3.0 mL), and CH₃OD (1.0 mL). The mixture was stirred at room temperature for 5 min. Subsequently, 2a (58.6 mg, 0.50 mmol) and 3 (254.0 mg, 1.00 mmol) were added sequentially and stirring was continued for 24 h at 25 °C. The resulting solution was filtered through a pad of silica gel using EtOAc as the eluent and concentrated. To a solution of the residue in THF (4.0 mL), hydrogen peroxide (0.20 mL, 30% aqueous solution) and an aqueous solution of NaOH (1.8 mL, 3.0 M) were added at 0 °C. The mixture was stirred at room temperature for 1 h and subsequently extracted with EtOAc (3×5 mL). The combined organic layers were washed with brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. The crude reaction mixture was purified by column chromatography on silica gel to afford the corresponding alcohol 8a-D (70.3 mg, 0.46 mmol, 90%). The amount of deuterium incorporated into the alcohol **8a-D** was measured using ¹H NMR analysis. ¹H NMR (500 MHz, CDCl₃) δ 1.46–1.49 (m, 2H), 1.72 (br, 1H), 4.86 (t, J = 6.0 Hz, 1H), 6.87–6.90 (m, 2H), 7.29–7.32 (m, 2H) ppm; ${}^{13}C{}^{1}H$ NMR (125 MHz, CDCl₃) δ 24.7 (t, J = 19.1 Hz), 55.3, 69.9, 113.8, 126.6, 138.0, 159.0 ppm. IR (Diamond-ATR, neat) 3370, 2928, 2850, 1612, 1510, 1461, 1299, 1241, 1174, 1064, 1032, 829, 545 cm⁻¹, HRMS (EI⁺): calcd for C₉H₁₁DO₂ [M]^{+•}: 153.0900. found: 153.0901. ¹H and ¹³C $\{$ ¹H $\}$ NMR spectra have been attached.

Characterization Data for the Isolated Products Synthesis of 1-(4-Methoxyphenyl)ethanol (8a)

MeO Compound **8a** was prepared according to the general procedure using **2a** (68.4 mg, 0.51 mmol), **3** (254.0 mg, 1.00 mmol), KOAc (98.1 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

OH

The product 8a was obtained in 85% yield (65.9 mg, 0.433 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 1.48 (d, J = 6.5 Hz, 3H), 1.78 (br, 1H), 3.81 (s, 3H), 4.86 (q, J = 6.5 Hz, 1H), 6.87–6.90 (m, 2H), 7.29–7.32 (m, 2H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 25.0, 55.3, 70.0, 113.8, 126.6, 138.0, 159.0 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.³

Synthesis of 1-Phenylethanol (8b)

OH

Compound **8b** was prepared according to the general procedure using **2b** (53.5 mg, 0.51 mmol), **3** (253.9 mg, 1.00 mmol), KOAc (98.3 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 8b was obtained in 84% yield (52.8 mg, 0.43 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 1.51 (d, *J* = 6.5 Hz, 3H), 1.81 (br, 1H), 4.91 (q, *J* = 6.5 Hz, 1H), 7.26–7.29 (m, 2H), 7.34–7.39 8m, 2H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 25.1, 70.4, 125.3, 127.4, 128.5, 145.8 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.³

Synthesis of 1-(*p*-Tolyl)ethanol (8c)

ОН

Compound **8c** was prepared according to the general procedure using **2c** (59.6 mg, 0.50 mmol), **3** (253.9 mg, 1.00 mmol), KOAc (98.1 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 8c was obtained in 89% yield (61.0 mg, 0.45 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 1.48 (d, *J* = 6.5 Hz, 3H), 1.77 (br, 1H), 4.87 (q, *J* = 6.5 Hz, 1H), 7.15–7.17 (m, 2H), 7.26–7.28 (m, 2H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 21.0, 25.0, 70.2, 125.3,

129.1, 137.1, 142.8 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.³

Synthesis of 1-(4-t-Butylphenyl)ethanol (8d)

^{t-Bu} Compound **8d** was prepared according to the general procedure using **2d** (80.1 mg, 0.50 mmol), **3** (253.9 mg, 1.00 mmol), KOAc (98.1 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 8d was obtained in 86% yield (77.0 mg, 0.43 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 1.32 (s, 9H), 1.49 (d, *J* = 6.0 Hz, 3H), 1.87 (br, 1H), 4.86 (q, *J* = 6.0 Hz, 1H), 7.30–7.31 (m, 2H), 7.37–7.38 (m, 2H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 24.9, 31.3, 34.5, 70.2, 125.1, 125.4, 142.8, 150.4 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.⁴

Synthesis of 1-(4-Biphenyl)ethanol (8e)

^{Ph} Compound **8e** was prepared according to the general procedure using **2e** (90.1 mg, 0.50 mmol), **3** (253.9 mg, 1.00 mmol), KOAc (98.2 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 8e was obtained in 73% yield (72.1 mg, 0.36 mmol) as a white solid.

¹H NMR (500 MHz, CDCl₃) δ 1.53 (d, J = 6.5 Hz, 3H), 1.81 (br, 1H), 4.94–4.97 (m, 1H), 7.33–7.36 (m, 1H), 7.42–7.46 (m, 4H), 7.58–7.60 (m, 4H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 25.2, 70.2, 125.8, 127.1, 127.3, 128.9, 140.5, 140.9, 144.8 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.⁵

Synthesis of 1-[(4-tert-Butyldimethylsilyloxy)phenyl]ethanol (8f)

Compound **8f** was prepared according to the general procedure using **2f** (118.3 mg, 0.50 mmol), **3** (253.9 mg, 1.00 mmol), KOAc (98.2 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

ОН

The product 8f was obtained in 70% yield (89.9 mg, 0.36 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 0.19 (s, 6H), 0.98 (s, 9H), 1.48 (d, J = 6.5 Hz, 3H), 1.71 (br, 1H), 4.85 (q, J = 6.5 Hz, 1H), 6.80–6.82 (m, 2H), 7.22–7.25 (m, 2H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ -4.4, 18.2, 25.0, 25.7, 70.1, 120.0, 126.6, 138.5, 155.0 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.⁶

Synthesis of 1-[4-(Benzyloxy)phenyl]ethanol (8g)

BnO Compound **8g** was prepared according to the general procedure using **2g** (105.2 mg, 0.50 mmol), **3** (253.9 mg, 1.00 mmol), KOAc (98.1 mg, 1.00 mmol), and NiCl₂(dppe) (13.3 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 8g was obtained in 53% yield (60.2 mg, 0.26 mmol) as a white solid.

¹H NMR (500 MHz, CDCl₃) δ 1.48 (d, *J* = 6.5 Hz, 3H), 1.69 (br, 1H), 4.86 (dq, *J* = 3.5 and 6.5 Hz, 1H), 5.07 (s, 2H), 6.95–6.98 (m, 2H), 7.29–7.34 (m, 3H) 7.37–7.44 (m, 4H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 25.0, 69.9, 70.0, 114.8, 126.7, 127.4, 127.9, 128.6, 137.0, 138.2, 158.2 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.⁶

Synthesis of 1-(4-Fluorophenyl)ethanol (8h)

F Compound **8h** was prepared according to the general procedure using **2h** (123.1 mg, 1.01 mmol), **3** (507.8 mg, 2.00 mmol), KOAc (196.3 mg, 2.00 mmol), and NiCl₂(dppe) 52.8 mg, 0.10 mmol, 10 mol%).

Reaction conditions: 24 h, 25 °C.

OH

The product 8h was obtained in 41% yield (57.8 mg, 0.41 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 1.49 (d, J = 6.5 Hz, 3H), 1.78 (br, 1H), 4.90 (qd, J = 3.5 and 6.5 Hz, 1H), 7.01–7.06 (m, 2H), 7.33–7.37 (m, 2H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 25.2, 69.7, 115.2 (J = 21.4 Hz), 127.0 (J = 8.4 Hz), 141.5 (J = 2.4 Hz), 162.1 (J = 243.3 Hz) ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.³

Compound **8i** was prepared according to the general procedure using **2i** (58.8 mg, 0.50 mmol), **3** (253.8 mg, 1.00 mmol), KOAc (98.3 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 8i was obtained in 70% yield (47.5 mg, 0.35 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 1.47 (d, J = 6.5 Hz, 3H), 1.70 (br, 1H), 2.35 (s, 3H), 5.14 (dq, J = 3.0 and 6.5 Hz, 1H), 7.13–7.19 8m, 2H), 7.22–7.26 (m, 1H), 7.51–7.52 (m, 1H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 18.9, 23.9, 66.8, 124.4, 126.4, 127.2, 130.4, 134.2, 143.8 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.³

Synthesis of 1-Mesitylethan-1-ol (8j)

OH

Compound **8j** was prepared according to the general procedure using **2j** (74.0 mg, 0.51 mmol), **3** (253.8 mg, 1.00 mmol), KOAc (98.3 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 8j was obtained in 63% yield (52.2 mg, 0.32 mmol) as a colourless oil.

¹H NMR (500 MHz, CDCl₃) δ 1.53 (d, J = 6.5 Hz, 3H), 2.10 (br, 1H), 2.25 (s, 3H), 2.41 (s, 6H), 5.37 (q, J = 6.5 Hz, 1H), 6.82 (s, 2H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 20.5, 20.7, 21.6, 67.5, 130.1, 135.6, 136.4, 137.6 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.⁷

Synthesis of 1-(1-Naphthyl)ethanol (8k)

Compound **8k** was prepared according to the general procedure using **2k** (77.6 mg, 0.50 mmol), **3** (254.1 mg, 1.00 mmol), KOAc (98.7 mg, 1.01 mmol), and NiCl₂(dppe) (13.4 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

OH

The product 8k was obtained in 83% yield (71.4 mg, 0.41 mmol) as a white solid.

¹H NMR (500 MHz, CDCl₃) δ 1.68 (d, *J* = 6.5 Hz, 3H), 1.89 (br, 1H), 5.70 (dq, *J* = 3.5 and 6.5 Hz, 1H), 7.47–7.55 (m, 3H), 7.68–7.70 (m, 1H), 7.78–7.80 (m, 1H), 7.87–7.89 (m, 1H), 8.13–8.14 (m,

1H) ppm; ${}^{13}C{}^{1}H$ NMR (125 MHz, CDCl₃) δ 24.3, 67.1, 122.0, 123.1, 125.52, 125.54, 126.0, 127.9, 128.9, 130.3, 133.8, 141.3 ppm. ${}^{1}H$ and ${}^{13}C{}^{1}H$ NMR spectra have been attached. The product was characterized by comparison with previously reported ${}^{1}H$ and ${}^{13}C{}^{1}H$ NMR data.⁸

Synthesis of 1-(2-Naphthyl)ethanol (8l)

Compound **81** was prepared according to the general procedure using **21** (77.0 mg, 0.50 mmol), **3** (253.9 mg, 1.00 mmol), KOAc (98.1 mg, 1.00 mmol), and NiCl₂(dppe) (13.2 mg, 0.025 mmol, 5 mol%).

Reaction conditions: 24 h, 25 °C.

The product 81 was obtained in 88% yield (75.3 mg, 0.44 mmol) as a white solid.

¹H NMR (500 MHz, CDCl₃) δ 1.55 (d, *J* = 6.5 Hz, 3H), 2.11 (br, 1H), 5.02 (q, *J* = 6.5 Hz, 1H), 7.43– 7.48 (m, 3H), 7.79–7.82 (m, 4H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 25.1, 70.5, 123.75, 123.78, 125.8, 126.1, 127.6, 127.9, 128.3, 132.9, 133.3, 143.1 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.³

Synthesis of 1,2-Diphenylethan-1-ol (14)

OH

^{Ph} Compound 14 was prepared according to the general procedure using 13 (90.5 mg, 0.50 mmol), 3 (381.1 mg, 1.50 mmol), KOAc (147.3 mg, 1.50 mmol), and NiCl₂(dppe) (26.4 mg, 0.05 mmol, 10 mol%).

Reaction conditions: 48 h, 25 °C.

The product 14 was obtained in 85% yield (84.2 mg, 0.42 mmol) as a white solid.

¹H NMR (500 MHz, CDCl₃) δ 1.93 (d, J = 3.0 Hz, 1H), 2.99 (dd, J = 8.5 and 13.5 Hz, 1H), 3.05 (dd, J = 4.5 and 13.5 Hz, 1H), 4.89–4.93 (m, 1H), 7.20–7.38 (m, 10H) ppm; ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 46.1, 125.9, 126.6, 127.6, 128.4, 128.5, 129.5, 138.0, 143.8 ppm. ¹H and ¹³C{¹H} NMR spectra have been attached. The product was characterized by comparison with previously reported ¹H and ¹³C{¹H} NMR data.³

Compound 14 was also prepared according to the general procedure using 15 (90.8 mg, 0.50 mmol), 3 (381.3 mg, 1.50 mmol), KOAc (147.2 mg, 1.50 mmol), and $NiCl_2(dppe)$ (26.5 mg, 0.050 mmol, 10 mol%).

Reaction conditions: 48 h, 25 °C.

The product 14 was obtained in 81% yield (81.3 mg, 0.41 mmol) as a white solid.

¹H NMR and ¹³C NMR Spectra











































8h



















8k













14









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