## Hydrothermal Synthesis of Two 2D Uranyl Coordination

## Polymers, Structures, Luminescence, and Photocatalytic

## **Degradation of Rhodamine B**

Xiaolan Tong,<sup>a</sup> Shan Wang,<sup>a</sup> HongXia Gao,<sup>b</sup> Yingchong Ge,<sup>a</sup> Jun Zuo,<sup>a</sup> Fen Liu,<sup>a</sup>

Jianhua Ding\*a and Jianbo Xiong\*a

†a State Key Laboratory of Nuclear Resources and Environment School of Chemistry, Biology and Materials Science, East

China University of Technology, Nanchang 330013, P. R. China. Email: xiongjianbo1210@163.com

†b School of Earth Science, East China University of Technology, Nanchang 330013, P. R. China.

## **Supporting materials**

Table S1 Selected bond length (Å) and angles (°) for 1 and 2

1.747(3)	U(1)-O(6)	1.741(3)
2.361(3)	U(1)-O(1)#2	2.288(3)
2.416(3)	U(1)-O(3)#4	2.475(3)
2.595(3)		
154.30(10)	O(2)#3-U(1)-O(3)#4	75.40(9)
153.47(9)	O(5)-U(1)-O(2)#3	85.18(12)
93.43(12)	O(5)-U(1)-O(4)	90.93(12)
91.07(11)	O(5)-U(1)-O(3)#4	96.18(11)
79.36(10)	O(1)#2-U(1)-O(4)	126.29(10)
74.65(9)	O(1)#2-U(1)-O(3)#4	152.06(10)
51.74(9)	O(4)-U(1)-O(3)#4	79.79(9)
131.13(11)	O(6)-U(1)-O(2)#3	97.09(12)
177.56(13)	O(6)-U(1)-O(1)#2	86.10(12)
87.42(12)	O(6)-U(1)-O(3)	86.49(11)
85.31(11)		
1,y-1,z-1; #2 1-x,-y	,-z; #3 x+1,y,z-1; #4 -x,y+3/2,	,-z+1/2; #5 -x,y+3/2,
2.360(3)	U(1)-O(2)#1	2.321(3)
2.474(4)	U(1)-O(4)#2	2.426(4)
2.341(4) 1.756(4)	U(1)-O(6)	1.756(4)
	$\begin{array}{c} 1.747(3)\\ 2.361(3)\\ 2.416(3)\\ 2.595(3)\\ 154.30(10)\\ 153.47(9)\\ 93.43(12)\\ 91.07(11)\\ 79.36(10)\\ 74.65(9)\\ 51.74(9)\\ 131.13(11)\\ 177.56(13)\\ 87.42(12)\\ 85.31(11)\\ 1,y-1,z-1; \#2 \ 1-x,-y,\\ 2.360(3)\\ 2.474(4)\\ 2.341(4)\\ 1.756(4)\\ \end{array}$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Bond angles				
O(1)-U(1)-O(3)#2	150.17(12)	O(1)-U(1)-O(4)#2	156.85(12)	
O(2)#1-U(1)-O(1)	79.74(12)	O(7)-U(1)-O(6)	178.27(18)	
O(2)#1-U(1)-O(3)#2	130.08(12)	O(2)#1-U(1)-O(4)#2	77.17(12)	
O(2)#1-U(1)-O(5)	155.66(14)	O(5)-U(1)-O(1)	75.95(14)	
O(5)-U(1)-O(3)#2	74.22(14)	O(5)-U(1)-O(4)#2	127.16(13)	
O(6)-U(1)-O(1)	91.32(16)	O(6)-U(1)-O(2)#1	87.77(16)	
O(6)-U(1)-O(3)#2	89.42(17)	O(6)-U(1)-O(4)#2	89.53(17)	
O(6)-U(1)-O(5)	91.46(18)	O(7)-U(1)-O(1)	88.72(15)	
O(7)-U(1)-O(2)#1	90.54(16)	O(7)-U(1)-O(3)#2	91.40(17)	
O(7)-U(1)-O(4)#2	89.76(16)	O(7)-U(1)-O(5)	90.23(18)	
Symmetrical code: #1 –x,2-y,-z; #2 x-1,y,z; #3 x-1,y-1,z.				



Fig. S1 Photographs of crystal 1 and 2 under light field (forty times magnified).



Fig. S2 IR spectra of 1 and 2.



Fig. S3 Coordination modes of bpda in 1 (a) and 2 (b).



Fig. S4 Space-filling mode of the 2D honeycomb network of 1 with pore diameter of 3.0 Å.



**Fig. S5** 2D rectangular sheet structure of **2** with size of 12.4 Å×4.4 Å (the Van der Waals radius of carbons were subtracted).



Fig. S6 The PXRD patterns of 1 (a) and 2 (b).



Fig. S7 TGA profiles of 1 (a) and 2 (b).



Fig. S8 The calculated band gaps of 1 and 2 based on the (ahn)<sup>2</sup>-hv curves.



Fig. S9 UV/Vis spectra of RhB soulutions in the presence of 1 (a) or 2 (b) mearsured at different time.



Fig. S10 Comparison of the PXRD patterns of 1 (a) and 2 (b) before and after photocatalysis experiment.