

Supporting Information

Impact of Alkyl Chain Length and Water on the Structure and Properties of 1-alkyl-3-methylimidazolium Chloride Ionic Liquids

Paridhi Sanchora[†], Deepak K. Pandey[†], Hardik L. Kagdada[†], Arnulf Materny[§] and Dheeraj K. Singh^{†*}

[†]*Department of Physics, Institute of Infrastructure Technology Research and Management, Ahmedabad, 380026, India*

[§]*Department of Physics and Earth Sciences, Jacobs University Bremen, Bremen, 28759, Germany*

*Corresponding author: dheerajsingh84@gmail.com

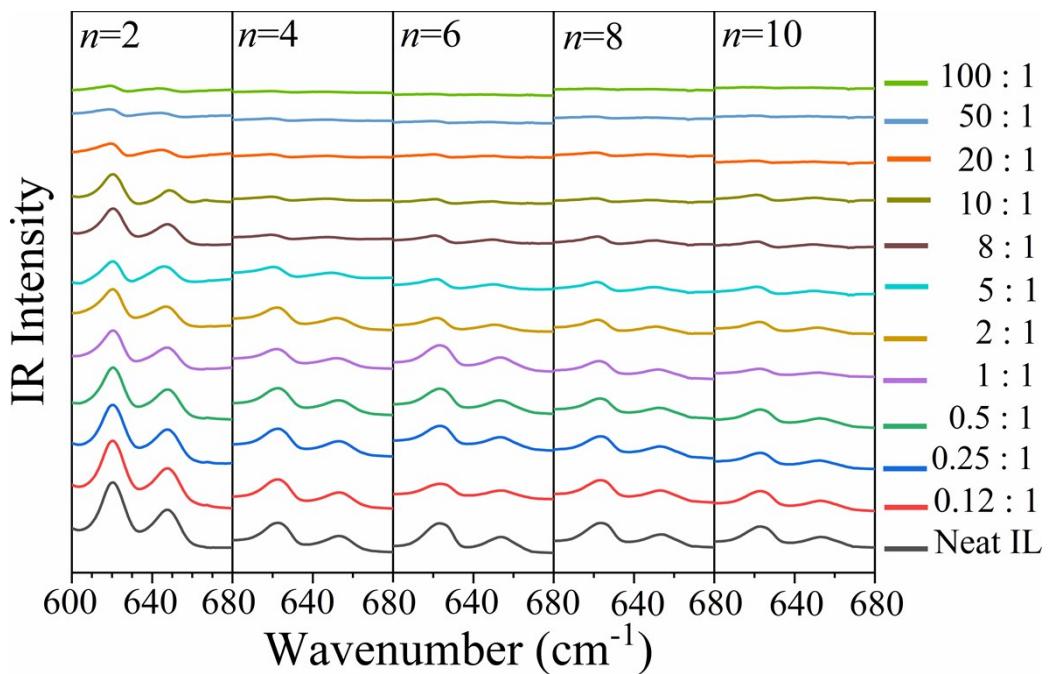


Fig. S1 ATR-IR absorption spectra of neat $\text{C}_n\text{mim Cl}$ ($n = 2, 4, 6, 8, 10$) ion pairs and their binary mixtures with H_2O in the region of $600\text{-}680 \text{ cm}^{-1}$ at various $\text{H}_2\text{O} : \text{IL}$ molar ratios. The lines at 623 and 654 cm^{-1} are assigned to ring stretching vibration and alkyl chain $\text{N}-\text{CH}_2-\text{CH}_3$ asymmetric bending, respectively.

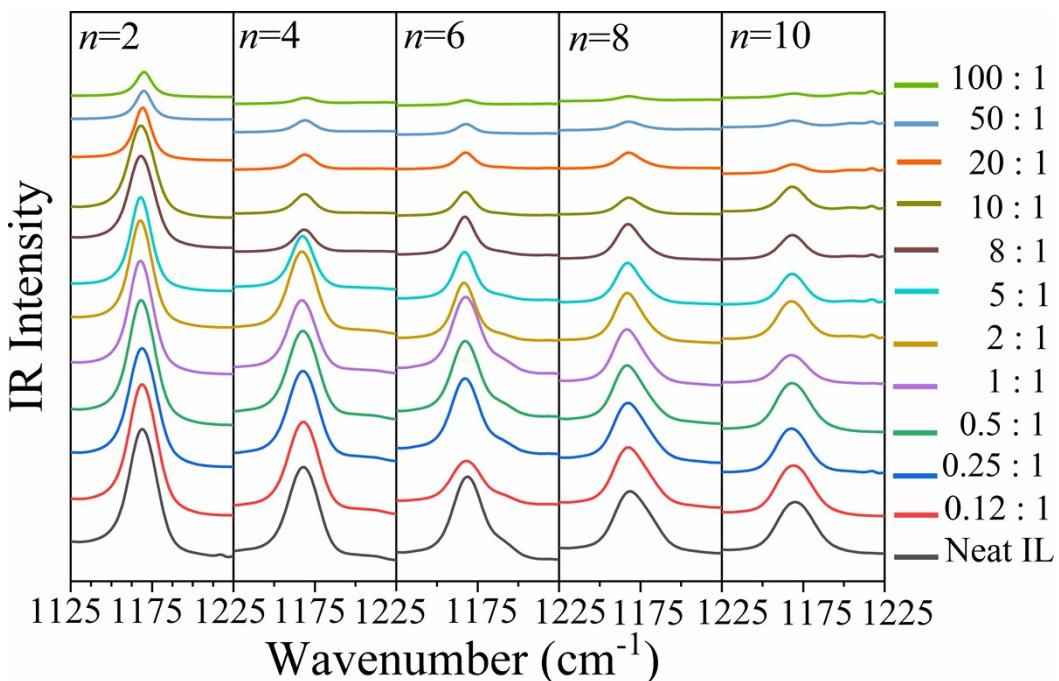


Fig. S2 ATR-IR absorption spectra of neat $\text{C}_n\text{mim Cl}$ ($n = 2, 4, 6, 8, 10$) ion pairs and their binary mixtures with H_2O in the region of $1125\text{-}1225 \text{ cm}^{-1}$ at various $\text{H}_2\text{O} : \text{IL}$ molar ratios. The line at 1167 cm^{-1} is assigned to the C-C stretching vibration of the ring.

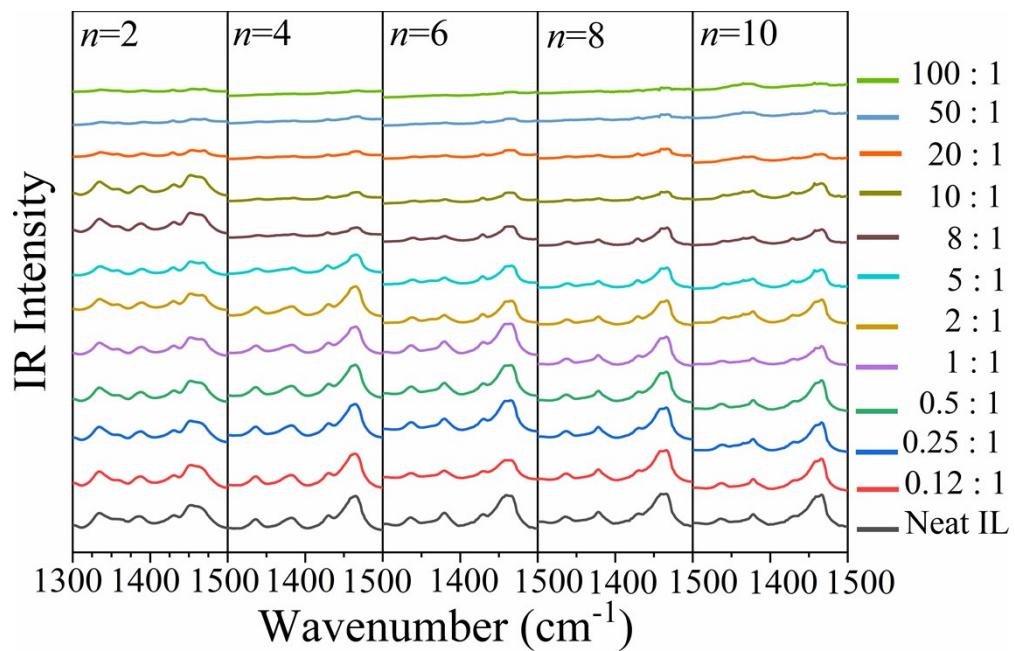


Fig. S3 ATR-IR absorption spectra of neat C_n mim Cl ($n = 2, 4, 6, 8, 10$) ion pairs and their binary mixtures with H_2O in the region of $1300\text{-}1500\text{ cm}^{-1}$ at various $H_2O : IL$ molar ratios. The line at 1325 cm^{-1} assigned to the ring C-H in plane bending. The line at 1378 cm^{-1} belongs to the alkyl chain (CH_2) stretching. The peaks at 1453 and 1470 cm^{-1} are assigned to the asymmetric ring stretching and asymmetric bending vibrations of C8-H, respectively.

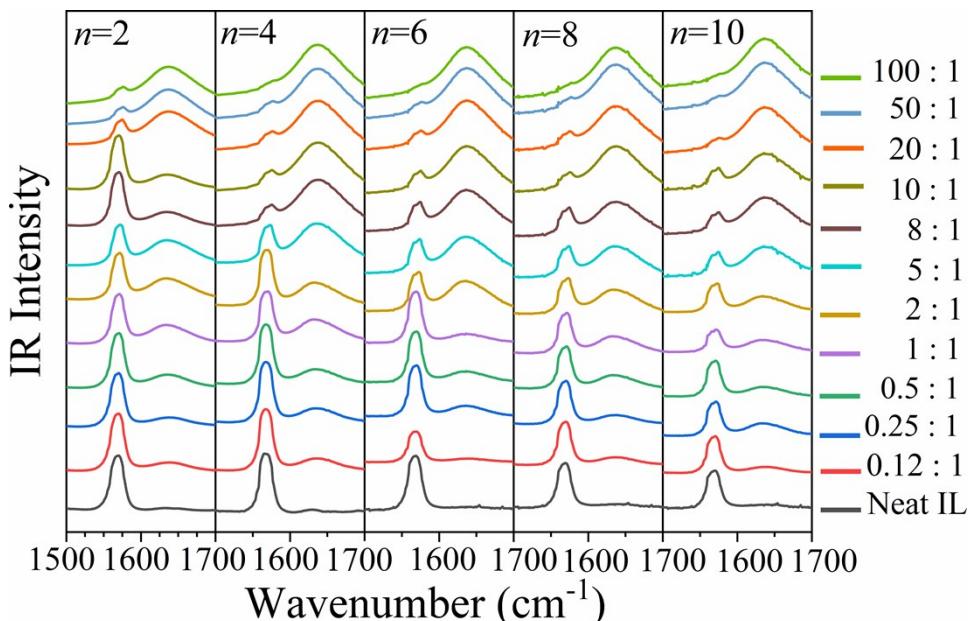


Fig. S4 ATR-IR absorption spectra of neat C_n mim Cl ($n = 2, 4, 6, 8, 10$) ion pairs and their binary mixtures with H_2O in the region of $1500\text{-}1700\text{ cm}^{-1}$ at various $H_2O : IL$ molar ratios. The line at 1569 cm^{-1} is assigned to the $C_8H_2\text{-}C_9H_2\text{-}C_{10}H_2$ scissoring vibration. A new peak arises at 1576 cm^{-1} .

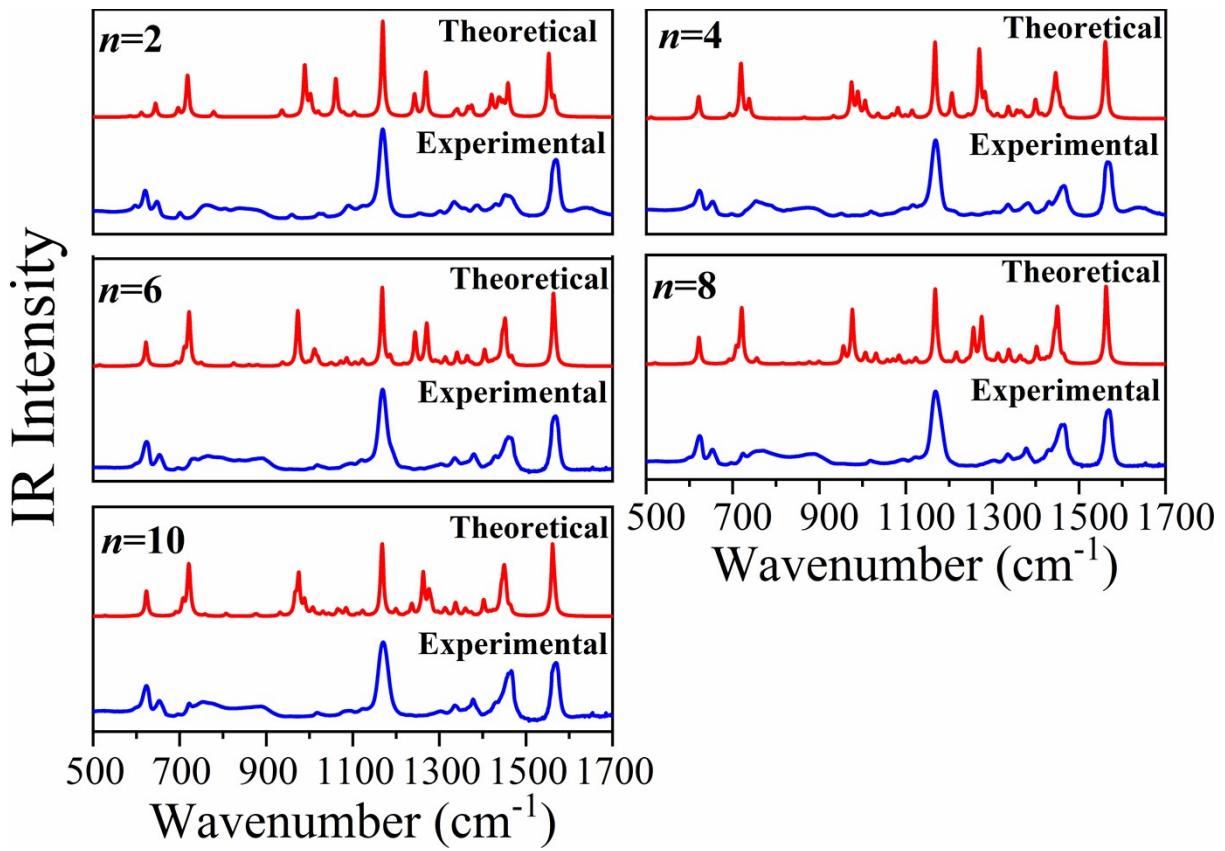


Fig. S5 Comparison of experimental and theoretical IR absorption spectra of the most stable geometries for the $\text{C}_n\text{mim Cl}$ ($n = 2, 4, 6, 8, 10$) ion-pairs in the range of $500\text{-}1700 \text{ cm}^{-1}$.

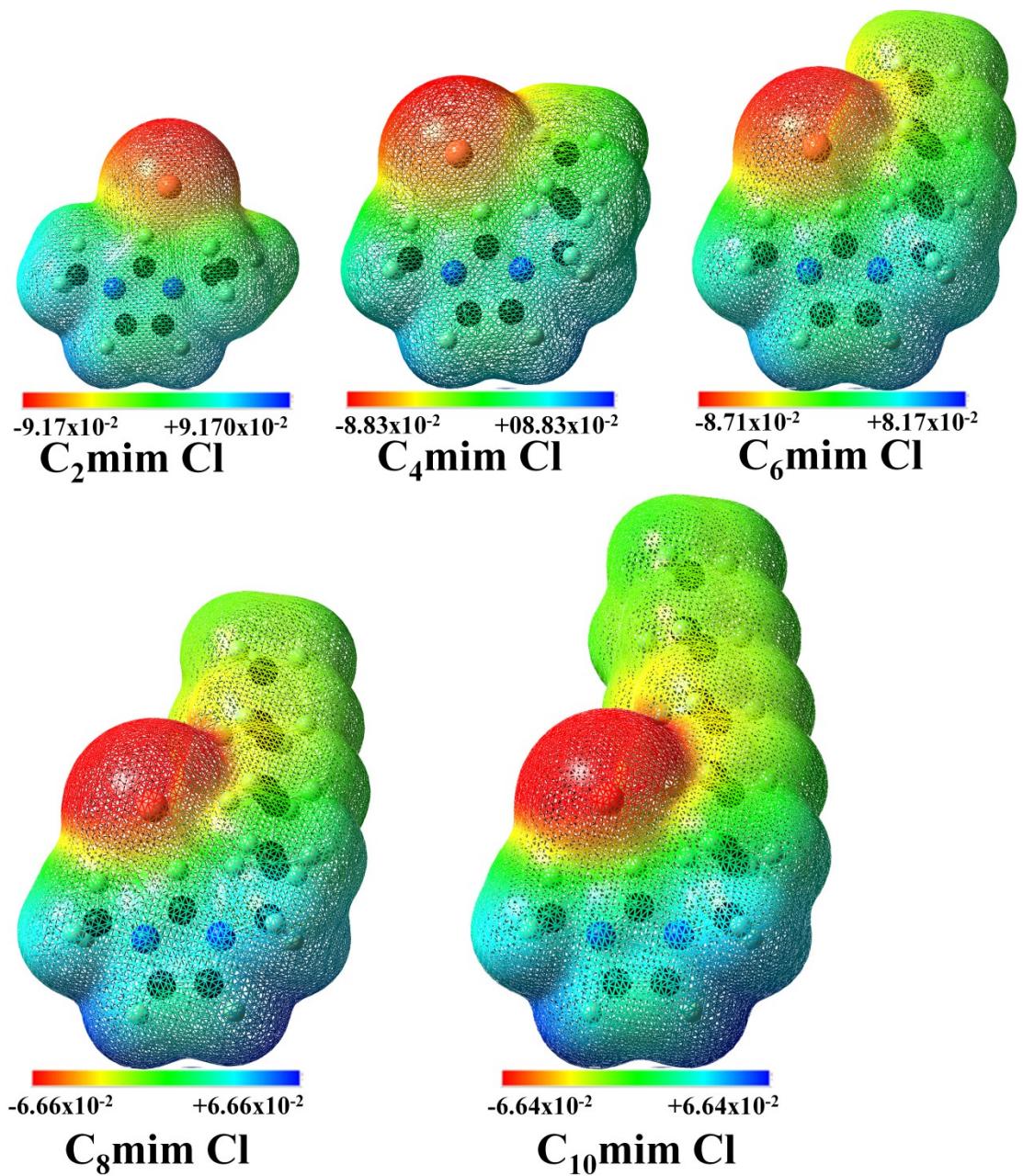


Fig. S6 Electrostatic potential mapping (ESP) from total self-consistent field (SCF) for $C_n\text{mim Cl}$ ($n = 2, 4, 6, 8, 10$) ion-pairs.

Table S1. Bond lengths and dihedral angles of the optimized geometries of the C_nmim Cl ($n = 2, 4, 6, 8, 10$) ion-pairs obtained from the DFT calculations.

	C ₂ mim Cl	C ₄ mim Cl	C ₆ mim Cl	C ₈ mim Cl	C ₁₀ mim Cl
Bond Length (Å)					
N1–C2	1.33	1.33	1.33	1.33	1.33
C2–N3	1.33	1.33	1.33	1.33	1.33
N3–C4	1.37	1.37	1.37	1.37	1.37
C4–C5	1.35	1.35	1.35	1.35	1.35
C5–N1	1.38	1.37	1.37	1.37	1.37
C2–H9	1.12	1.12	1.11	1.11	1.11
C5–H11	1.07	1.07	1.07	1.07	1.07
C4–H10	1.07	1.07	1.07	1.07	1.07
N1–C6	1.45	1.46	1.46	1.46	1.46
N3–C7	1.47	1.47	1.47	1.47	1.47
C6–H12	1.09	1.09	1.09	1.09	1.09
C6–H13	1.09	1.09	1.09	1.09	1.09
C6–H14	1.08	1.08	1.08	1.09	1.09
C7–H15	1.09	1.09	1.09	1.09	1.09
C7–H16	1.09	1.09	1.09	1.09	1.09
C8–H17	1.09	1.09	1.09	1.09	1.09
C8–H18	1.09	1.09	1.09	1.09	1.09
C9–H19	1.09	1.09	1.09	1.09	1.09
C9–H20		1.09	1.09	1.09	1.09
C10–H21		1.09	1.09	1.09	1.09
C10–H22		1.09	1.09	1.09	1.09
C10–H23		1.09	1.09	1.09	1.09
C11–H24			1.09	1.09	1.09
C12–H25			1.09	1.09	1.09
C12–H26			1.09	1.09	1.09
C13–H27				1.09	1.09
C13–H28				1.09	1.09
C14–H29				1.09	1.09
C14–H30				1.09	1.09
C14–H31				1.09	1.09
C15–H32					1.09
C15–H33					1.09
C16–H34					1.09
C16–H35					1.09
C16–H36					1.09
Cl20–H9	1.99	2.00	2.02	2.02	2.02
Dihedral Angle (deg)					
∠ N1–C7–C8–C9		-64.52	-64.54	-64.62	-64.73

Table S2. Wavenumbers of vibrational modes for C₂mim⁺ and the energetically optimized configuration of C_nmim Cl ($n = 2, 4, 6, 8, 10$), calculated using DFT and observed in the IR spectra.

Vibration	Computational		Experimental
	C ₂ mim ⁺ (cm ⁻¹)	IR C ₂ mim Cl (cm ⁻¹)	
v _s (C4-H10, C5-H11)	3308 (10)	3279 (6)	3144 (W)
v _{as} (C4-H10, C5-H11)	3290 (16)		
v (C2-H9)	3292 (29)	2631 (1420)	3054 (S)
v _{as} (H12,13-C6- H14)	3189 (0.3)	3152 (1)	
v _{as} (H12-C6-H13)	3178 (0.1)	3132 (6)	
v _{as} (H15-C7-H16, H17,18-C8-H19)	3160 (8)	3124 (4)	2976 (M)
v _{as} (H15-C7-H16, H17,18-C8-H19)	3154 (5)	3115 (8)	
v _{as} (H15-C7-,H16); v _s (H17,18-C8-H19)	3138 (2)	3100 (14)	
v _s (H15-C7-H16)	3098 (6)	3055 (33)	2931 (W)
v _s (H12,13-C6-H14)	3085 (4)	3046 (22)	
v _s (H17-C8-H18,19)	3063 (4)	3020 (24)	2868 (W)

The C–H region for C₂mim Cl is scaled by 0.991. The IR intensities are shown in km/mol in parentheses. Key: W, weak; M, medium; S, strong; v, stretch; v_s, symmetric stretch; v_{as}, asymmetric stretch.

Vibration	Computational		Experimental
	C ₄ mim ⁺ (cm ⁻¹)	IR C ₄ mim Cl (cm ⁻¹)	
v _s (C4-H10, C5-H11)	3310 (10)	3134 (6)	3143 (W)
v _{as} (C4-H10, C5-H11)	3291 (19)		
v (C2-H9)	3292 (23)	2555 (1239)	3057 (M)
v _{as} (H12,13-C6- H14)	3177 (0.1)	2988 (7)	
v _{as} (H12-C6-H13)	3178 (0.1)	2971 (19)	
v _{as} (H21-22-C10-H12); v _s (H19-C9-H20)	3140 (16)	2958 (14)	2957 (S)
v _{as} (H15-C7-H16, H17-C8-H18)	3160 (8)	2950 (11)	
v _s (H19-C9-H20)	3125 (33)	2944 (43)	2930 (M)
v _{as} (H17-C8-H18), v _{as} (H21,22-C10-,H23); v _{as} (H19-C9-H20)		2929 (18)	
v _s (H15-C7-H16); v _{as} (H17-C8-H18); v _{as} (H19-C9-H20);	3102 (19)	2908 (9)	
v _{as} (H21,22-C10-,H23)			
v _s (H15-C7-H16); v _{as} (H17-C8-H18);	3093 (3.3)	2900 (28)	
v _s (H17-C8-H18);		2884 (11)	
v _s (H12,13-C6-H14)	3085 (4)	2879 (65)	
v _s (H21-C10-H22,23)	3051 (22)	2870 (42)	2867 (M)
v _s (H17-C8-H18)	3046 (14)	2858 (45)	
v _s (H19-C9-H20)	3030 (20)		

The C–H region for C₄mim Cl is scaled by 0.954. The IR intensities are shown in km/mol in parentheses. Key: W, weak; M, medium; S, strong; v, stretch; v_s, symmetric stretch; v_{as}, asymmetric stretch.

Vibration	Computational		Experimental
	C ₆ mim ⁺ (cm ⁻¹)	IR C ₆ mim Cl (cm ⁻¹)	
v _s (C4-H10, C5-H11)	3310 (10)	3054 (2)	3140 (W)
v _{as} (C4-H10, C5-H11)	3292 (17)	3036 (6)	
v (C2-H9)	3291 (25)	2526 (1128)	3052 (M)
v _{as} (H12,13-C6- H14)	3189 (0.4)	2919 (7)	
v _{as} (H25-C12-H26,27); v _s (H23-C11-H24)	3178 (0.2)	2900 (21)	2955 (M)
v _{as} (H15-C7-H16, H17-C8-H18)	3151 (4.2)	2880 (16)	
v _{as} (H25-C12-H26)	3130 (25)	2879 (37)	
v _s (H23-C11-H24), v _s (H21-C10-H22)	3118 (48)	2869 (61)	2929 (S)
v _s (H15-C7-H16); v _{as} (H17-C8-H18); v _{as} (H19-C9-H20), v _{as} (H21-C10-H22)	3103 (30)		
v _{as} (H17-C8-H18); v _{as} (H19-C9-H20),		2860 (34)	

Vibration	Computational	Experimental
	IR $C_8mim^+ (cm^{-1})$	IR $C_8mim Cl (cm^{-1})$
$v_{as}(H21-C10-H22)$		
$v_{as}(H17-C8-H18); v_{as}(H19-C9-H20),$		2849 (6)
$v_{as}(H21-C10-H22); v_{as}(H23-C11-H24);$		
$v_{as}(H25-C12-H26)$		
$v_s(H15-C7-H16); v_{as}(H17-C8-H18); v_{as}(H19-C9-$		
$H20), v_{as}(H21-C10-H22); v_{as}(H23-C11-H24);$		2837 (4)
$v_{as}(H25-C12-H26)$		
$v_s(H15-C7-H16)$	3093 (7)	2831 (28)
$v_{as}(H23-C11-H24), v_{as}(H21-C10-H22)$	3083 (36)	2830 (10)
$v_s(H18-C8-H19)$	3046 (17)	2815 (11)
$v_s(H12,13-C6-H14)$	3085 (4)	2810 (75)
$v_s(H25-C12-H26)$	3045 (24)	2802 (41)
$v_s(H23-C11-H24); v_s(H19-C9-H20)$	3034 (60)	2788 (77) 2857 (M)

The C–H region for C_6mim Cl is scaled by 0.923. The IR intensities are shown in km/mol in parentheses. Key: W, weak; M, medium; S, strong; v, stretch; v_s , symmetric stretch; v_{as} , asymmetric stretch.

Vibration	Computational	Experimental
	IR $C_8mim^+ (cm^{-1})$	IR $C_8mim Cl (cm^{-1})$
$v_s(C4-H10, C5-H11)$	3310 (10)	3180 (2)
$v_{as}(C4-H10, C5-H11)$	3291 (20)	3160 (6)
$v(C2-H9)$	3292 (21)	2637 (106) 3055 (W)
$v_{as}(H12,13-C6-H14)$	3189 (0.4)	3038 (7)
$v_{as}(H12-C6-H13)$	3177 (0.2)	3018 (22)
$v_{as}(H15-C7-H16, H17,18-C8-H19)$	3152 (4)	2997 (16)
$v_{as}(H29,30-C14-H31)$	3126 (30)	2990 (46)
$v_{as}(H29,30-C14-H31); v_{as}(H27-C13-H28)$	3114 (56)	2986 (69) 2956 (M)
$v_s(H15-C7-H16); v_{as}(H17-C8-H18);$		
$v_{as}(H19-C9-H20), v_{as}(H21-C10-H22)$	3104 (34)	
$v_{as}(H15-C7-H16); v_{as}(H17-C8-H18);$		2979 (49)
$v_{as}(H19-C9-H20), v_{as}(H21-C10-H22);$		
$v_{as}(H23-C11-H24); v_{as}(H25-C12-H26);$	3086 (80)	2970 (16)
$v_{as}(H27-C13-H28); v_{as}(H29,30-C14-H31)$		
$v_s(H15-C7-H16); v_{as}(H17-C8-H18);$		
$v_{as}(H19-C9-H20), v_{as}(H21-C10-H22);$	2958 (20)	
$v_{as}(H23-C11-H24); v_{as}(H25-C12-H26);$		
$v_{as}(H27-C13-H28); v_{as}(H29,30-C14-H31)$		
$v_s(H15-C7-H16, H17-C8-H18)$	2948 (28)	
$v_{as}(H23-C11-H24); v_{as}(H27-C13-H28)$	2941 (2)	
$v_s(H17-C8-H18); v_{as}(H23-C11-H24);$	2932 (19)	
$v_{as}(H25-C12-H26); v_{as}(H27-C13-H28)$		
$v_s(H17-C7-H18)$	2931 (5)	
$v_s(H17-C8-H18)$	3084 (4)	2924 (80) 2924 (S)
$v_s(H29,30-C14-H31)$	3042 (33)	2915 (39) 2853 (M)
$v_s(H17-C8-H18); v_{as}(H23-C11-H24);$		
$v_s(H25-C12-H26); v_s(H27-C13-H28)$	3034 (101)	
$v_s(H27-C13-H28)$		2909 (60)
$v_{as}(H17-C8-H18)$		2900 (44)
$v_s(H21-C10-H22); v_{as}(H23-C11-H24);$		2899 (57)
$v_s(H25-C12-H26); v_s(H27-C13-H28)$		
$v_s(H21-C10-H22); v_s(H25-C12-H26)$	2896 (0.4)	
$v_s(H23-C11-H24)$	2892 (15)	

The C–H region for C_8mim Cl is scaled by 0.960. The IR intensities are shown in km/mol in parentheses. Key: W, weak; M, medium; S, strong; v, stretch; v_s , symmetric stretch; v_{as} , asymmetric stretch.

Vibration	Computational	Experimental
	IR $C_8mim^+ (cm^{-1})$	IR $C_8mim Cl (cm^{-1})$

	$C_{10}im^+$ (cm^{-1})	IR $C_{10}mim Cl$ (cm^{-1})	
v_s (C4-H10, C5-H11)	3311 (10)	3180 (2)	3141 (W)
v_{as} (C4-H10, C5-H11)	32901(19)	3161 (6)	
v (C2-H9)	3292 (22)	2640 (1104)	3057 (W)
v_{as} (H12,13-C6- H14)	3189 (0.4)	3039 (7)	
v_{as} (H12-C6-H13)	3178 (0.2)	3018 (22)	
v_{as} (H15-C7-H16, H17-C8-H18)	3152 (4)	2998 (15)	
v_{as} (H33,34-C16- H35)	3123 (34)	2992 (45)	
v_{as} (H33,34-C16- H35); v_{as} (H31-C15- H32)	3113 (60)	2987 (70)	2955 (W)
v_{as} (H15-C7-H16); v_{as} (H17-C8-H18); v_{as} (H19-C9-H20); v_{as} (H21-C10-H22); v_{as} (H23-C11-H24)		2980 (57)	
v_s (H15-C7-H16); v_{as} (H17-C8-H18); v_{as} (H19-C9-H20); v_{as} (H21-C10-H22); v_{as} (H23-C11-H24); v_{as} (H25-C12-H26); v_{as} (H27-C13-H28); v_{as} (H29-C14-H30)	3103 (37)	2971 (26)	
v_s (H15-C7-H16); v_{as} (H17-C8-H18); v_{as} (H19-C9-H20); v_{as} (H21-C10-H22); v_{as} (H23-C11-H24); v_{as} (H25-C12-H26); v_{as} (H27-C13-H28); v_{as} (H29-C14-H30); v_{as} (H33,34-C16- H35); v_{as} (H31-C15- H32)	3087 (126)	2962 (72)	
v_s (H15-C7-,H16); v_{as} (H17,18-C8-H19)		2948 (28)	
v_{as} (H19-C9-H20), v_{as} (H21-C10-H22); v_{as} (H23-C11-H24); v_{as} (H25-C12-H26); v_s (H27-C13-H28); v_s (H29-C14-H30)	3034 (150)	2924 (80)	2922 (S)
v_s (H33,34-C16- H35)		2917 (39)	
v_s (H31-C15- H32)		2912 (78)	2852 (M)
v_s (H19-C9-H20), v_s (H21-C10-H22); v_s (H23-C11-H24); v_s (H25-C12-H26); v_s (H27-C13-H28); v_s (H29-C14-H30)	3034 (150)	2904 (83)	
v_s (H19-C9-H20)	3016 (4)	2902 (40)	

The C–H region for $C_{10}mim Cl$ is scaled by 0.961. The IR intensities are shown in km/mol in parentheses. Key: W, weak; M, medium; S, strong; v, stretch; v_s , symmetric stretch; v_{as} , asymmetric stretch.

Table S3. Natural bond orbital (NBO) electron density in the cation and four configurations of $C_nmim Cl$ ($n=2,4,6,8$, and 10), calculated at the wB97XD/6-311++G(d,p) level of theory.

	C_2mim	$C_2mim Cl$	C_4mim	$C_4mim Cl$	C_6mim	$C_6mim Cl$	C_8mim	$C_8mim Cl$	$C_{10}mim$	$C_{10}mim Cl$
N1	-0.34	-0.37	-0.34	-0.36	-0.34	-0.36	-0.34	-0.36	-0.34	-0.36
C2	0.30	0.28	0.29	0.27	0.29	0.27	0.29	0.27	0.29	0.27
H9	0.24	0.29	0.24	0.29	0.24	0.29	0.24	0.29	0.29	0.29
N3	-0.34	-0.36	-0.34	-0.37	-0.34	-0.36	-0.34	-0.36	-0.36	-0.36
C4	-0.01	-0.02	-0.00	-0.03	-0.00	-0.03	-0.00	-0.03	-0.00	-0.03
H10	0.25	0.23	0.25	0.23	0.25	0.23	0.25	0.23	0.25	0.23
C5	-0.01	-0.04	-0.01	-0.03	-0.01	-0.03	-0.01	-0.03	-0.01	-0.03
H11	0.25	0.23	0.25	0.23	0.25	0.23	0.25	0.23	0.25	0.23
C6	-0.36	-0.35	-0.36	-0.37	-0.36	-0.37	-0.36	-0.37	-0.36	-0.37
H12	0.23	0.21	0.23	0.20	0.23	0.20	0.23	0.20	0.23	0.20
H13	0.23	0.21	0.23	0.20	0.23	0.20	0.23	0.20	0.23	0.20
H14	0.22	0.24	0.22	0.27	0.22	0.27	0.22	0.27	0.22	0.27
C7	-0.17	-0.18	-0.17	-0.15	-0.16	-0.15	-0.16	-0.16	-0.16	-0.15
H15	0.21	0.26	0.22	0.20	0.22	0.21	0.22	0.20	0.22	0.20
H16	0.22	0.19	0.22	0.21	0.22	0.20	0.22	0.20	0.22	0.20
C8	-0.59	-0.59	-0.40	-0.42	-0.39	-0.41	-0.39	-0.41	-0.39	-0.41
H17	0.21	0.24	0.20	0.24	0.20	0.21	0.20	0.21	0.20	0.21

H18	0.21	0.19	0.22	0.21	0.22	0.23	0.22	0.23	0.22	0.23
H19	0.23	0.21	0.18	0.18	0.19	0.18	0.18	0.18	0.18	0.18
C9		-0.39	-0.39	-0.39	-0.38	-0.39	-0.39	-0.39	-0.39	-0.38
H20		0.18	0.20	0.18	0.20	0.18	0.20	0.19	0.20	
C10		-0.27	-0.58	-0.38	-0.39	-0.38	-0.39	-0.38	-0.39	-0.39
H21		0.20	0.18	0.19	0.17	0.20	0.17	0.20	0.17	
H22		0.21	0.24	0.19	0.23	0.19	0.23	0.19	0.23	
C11				-0.38	-0.38	-0.38	-0.38	-0.37	-0.37	
H23		0.21	0.20	0.18	0.17	0.19	0.17	0.19	0.17	
H24				0.19	0.20	0.20	0.20	0.19	0.20	
H25				0.19	0.19	0.19	0.18	0.18	0.18	
C12				-0.57	-0.57	-0.38	-0.38	-0.38	-0.37	
H26				0.20	0.18	0.18	0.20	0.18	0.20	
H27				0.21	0.20	0.19	0.18	0.19	0.19	
C13						-0.38	-0.38	-0.37	-0.37	
H28						0.19	0.19	0.18	0.19	
H29						0.19	0.19	0.20	0.19	
C14						-0.57	-0.57	-0.38	-0.38	
H30						0.19	0.19	0.19	0.18	
H31						0.19	0.19	0.19	0.19	
C15								-0.38	-0.38	
H32								0.18	0.18	
H33								0.19	0.18	
C16								0.18	0.19	
H34								-0.57	-0.57	
H35								0.19	0.19	
Cl38	-0.86									