Supporting Information:

Accurate Reproducing Kernel-Based Potential Energy Surfaces for the Triplet Ground States of N₂O and Dynamics for the N+NO \leftrightarrow O+N₂ and N₂+O \rightarrow 2N+O Reactions

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FIG. S1. Correlation between the *ab initio* and RKHS interpolated energies for the training grid (upper panel) and for the test grid (which are not part of the training grid) (lower panel) for ${}^{3}A'$ and ${}^{3}A''$ different electronic states of N₂O. Zero of energy is set to the atomization energy of N₂O i.e., energy of N(${}^{4}S$)+N(${}^{4}S$)+O(${}^{3}P$). The values of R^{2} in upper panels for the grid energies up to 6.80 eV are 0.99995 and 0.99996 for ${}^{3}A'$ and ${}^{3}A''$, respectively where as in lower panel for the test grid values of R^{2} are 0.99985 and 0.99988 for ${}^{3}A'$ and ${}^{3}A''$, respectively. The correlation for the "on grid points" is not 1 because a) a finite regularization parameter was used (see main text) and b) the global PES is generated by accumulating the weighted contributions from the three channels.

<i>T</i> (K)	$k_1^f({}^3A')$	$k_1^f({}^3A'')$	$k_1^f({}^3A') + k_1^f({}^3A'')$
100	0.000	4.879×10^{-11}	4.879×10^{-11}
200	0.000	3.928×10^{-11}	3.928×10^{-11}
300	0.000	3.333×10^{-11}	3.333×10^{-11}
500	9.743×10^{-16}	$2.837{ imes}10^{-11}$	2.837×10^{-11}
800	2.424×10^{-14}	$2.592{ imes}10^{-11}$	2.595×10^{-11}
1000	7.984×10^{-14}	2.590×10^{-11}	2.598×10^{-11}
2000	1.257×10^{-12}	2.871×10^{-11}	3.000×10^{-11}
3000	3.759×10^{-12}	3.234×10^{-11}	3.610×10^{-11}
3500	5.268×10^{-12}	3.458×10^{-11}	3.985×10^{-11}
4000	7.117×10^{-12}	3.609×10^{-11}	4.321×10^{-11}
4500	8.495×10^{-12}	3.816×10^{-11}	4.666×10^{-11}
5000	1.077×10^{-11}	4.018×10^{-11}	5.095×10^{-11}
5500	1.273×10^{-11}	4.100×10^{-11}	5.373×10^{-11}
6000	1.465×10^{-11}	4.290×10^{-11}	5.755×10^{-11}
7500	2.121×10^{-11}	4.776×10^{-11}	6.897×10^{-11}
10000	$3.157{\times}10^{-11}$	5.516×10^{-11}	8.674×10^{-11}
12500	4.014×10^{-11}	6.095×10^{-11}	1.011×10^{-10}
15000	4.884×10^{-11}	6.793×10^{-11}	1.168×10^{-10}
17500	5.508×10^{-11}	7.087×10^{-11}	1.259×10^{-10}
20000	6.025×10^{-11}	7.524×10^{-11}	1.355×10^{-10}

TABLE S1. Rate coefficients for the $N(^4S) + NO(X^2\Pi) \rightarrow O(^3P) + N_2(X^1\Sigma)$ from 100 to 20000 K calculated using QCT on the $^3A'$ and $^3A''$ PESs. Units are in cm³s⁻¹molecule⁻¹

-	T (K)	$k_2^f({}^3A')$	$k_2^f({}^3A^{\prime\prime})$	$k_2^f({}^3A') + k_2^f({}^3A'')$
-	100	0.000	0.000	0.000
	200	0.000	0.000	0.000
	300	0.000	0.000	0.000
	500	0.000	0.000	0.000
	800	0.000	0.000	0.000
	1000	0.000	0.000	0.000
	2000	1.881×10^{-16}	0.000	1.8812×10^{-16}
	3000	2.591×10^{-14}	3.029×10^{-14}	$5.6202 {\times} 10^{-14}$
	3500	$6.513 { imes} 10^{-14}$	9.275×10^{-14}	$1.5792{\times}10^{-13}$
	4000	1.469×10^{-13}	1.718×10^{-13}	3.1872×10^{-13}
	4500	2.848×10^{-13}	3.062×10^{-13}	5.9102×10^{-13}
	5000	4.408×10^{-13}	4.768×10^{-13}	9.1752×10^{-13}
	5500	7.233×10^{-13}	7.878×10^{-13}	1.5112×10^{-12}
	6000	1.051×10^{-12}	$1.086{ imes}10^{-12}$	$2.1372{\times}10^{-12}$
	7500	$2.626{\times}10^{-12}$	$2.512{ imes}10^{-12}$	5.1392×10^{-12}
	10000	6.338×10^{-12}	6.481×10^{-12}	1.2822×10^{-11}
	12500	1.187×10^{-11}	1.117×10^{-11}	2.3042×10^{-11}
	15000	$1.696{ imes}10^{-11}$	1.650×10^{-11}	3.3452×10^{-11}
	17500	$2.176{ imes}10^{-11}$	2.106×10^{-11}	4.2812×10^{-11}
	20000	2.618×10^{-11}	2.532×10^{-11}	5.1502×10^{-11}

TABLE S2. Rate coefficients for the $N_A(^4S) + N_BO(X^2\Pi) \rightarrow N_B(^4S) + N_AO(X^2\Pi)$ from 100 to 20000 K calculated using QCT on the ³A' and ³A'' PESs. Units are in cm³s⁻¹molecule⁻¹

T (K)	$k^b(^3A')$	$k^b(^3A^{\prime\prime})$	$k^b({}^3A') + k^b({}^3A'')$
2800	0.000	0.000	0.000
3000	0.000	$7.605{\times}10^{-16}$	7.605×10^{-16}
3500	2.357×10^{-16}	3.824×10^{-15}	4.059×10^{-15}
4000	3.782×10^{-15}	1.764×10^{-14}	2.142×10^{-14}
4500	1.026×10^{-14}	5.161×10^{-14}	$6.187{ imes}10^{-14}$
5000	3.082×10^{-14}	1.191×10^{-13}	1.499×10^{-13}
5500	8.772×10^{-14}	2.555×10^{-13}	3.433×10^{-13}
6000	1.575×10^{-13}	4.374×10^{-13}	5.950×10^{-13}
7000	5.105×10^{-13}	1.171×10^{-12}	1.682×10^{-12}
10000	$3.974{ imes}10^{-12}$	6.773×10^{-12}	1.075×10^{-11}
12000	8.785×10^{-12}	1.392×10^{-11}	$2.271{\times}10^{-11}$
15000	2.082×10^{-11}	2.859×10^{-11}	4.942×10^{-11}
17000	3.096×10^{-11}	4.010×10^{-11}	7.105×10^{-11}
20000	4.575×10^{-11}	5.804×10^{-11}	1.038×10^{-10}

TABLE S3. Rate coefficients for the $O(^{3}P) + N_{2}(X^{1}\Sigma) \rightarrow N(^{4}S) + NO(X^{2}\Pi)$ from 2800 to 20000 K calculated using QCT on the $^{3}A'$ and $^{3}A''$ PESs. Units are in cm³s⁻¹molecule⁻¹

TABLE S4. N₂ dissociation rates for the $O(^{3}P) + N_{2}(X^{1}\Sigma)$ collisions for 7000 to 20000 K calculated using QCT simulations on the $^{3}A'$ and $^{3}A''$ PESs. Units are in cm³s⁻¹molecule⁻¹

T (K)	$k_1^d({}^3A') + k_1^d({}^3A'')$
7000	0.000
10000	1.061×10^{-13}
12000	5.848×10^{-13}
15000	3.915×10^{-12}
17000	6.888×10^{-12}
20000	1.720×10^{-11}



FIG. S2. Vibrational relaxation rates for $O + N_2(\nu = 1) \rightarrow O + N_2(\nu' = 0)$. QR QCT-HB rates calculated from the quasi-reactive (QR) trajectories (see text). The fit to the experimental results is shown as dashed olive line.



FIG. S3. Conformational sampling (green isocontour lines) for the three different types of trajectories projected onto the relaxed $(0.9 \le r \le 1.3 a_0) \text{ O}+\text{N}_2$ PES. 4×10^5 trajectories were run at 4000 K on the ³A" PES, out of which 664, 204 and 60 trajectories were found to be PNR, QR, and reactive trajectories, respectively. The PNR trajectories sample the long range part of the PES (the maximum at $R \approx 6 a_0$) and never enter the strongly interacting region ($R < 4 a_0$) whereas the QR and reactive trajectories heavily sample the strongly interacting region. The distributions have been smoothed by using kernel density estimation (KDE)¹ as implemented in the 'R-studio' package². The contour lines represent 10 % to 90% of the maximum intensity for each case, and dark contours correspond to increased population. It is noted that all distributions are close to converged as they display near-C_{2v} symmetry as required by the underlying PES.

REFERENCES

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