

# Theoretical Study on the Absorption of Carbon dioxide by DBU-based Ionic Liquids

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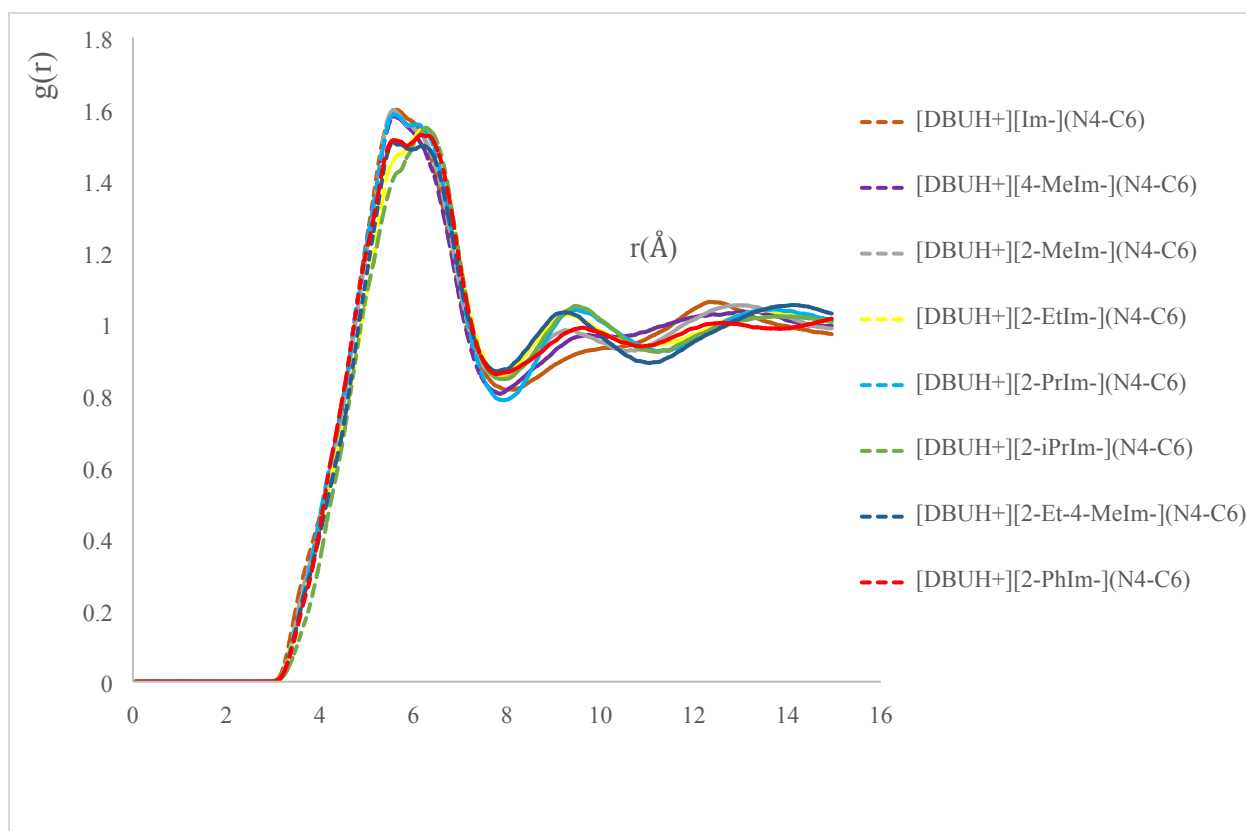


Figure S1. The RDF for interaction between cation-CO<sub>2</sub> in different DBU-based ILs during the 20 ns MD simulations

Table S1. The value of  $r_{\max}$  and  $g(r)_{\max}$  and  $N(r)_{\max}$  in different DBU-based ILs for N1-C6 and N3-C6 interactions in the presence of water molecule.

Compound	N1-C6 ( $r_{\max}$ )	$g(r)_{\max}$	$N(r)$	N3-C6 ( $r_{\max}$ )	$g(r)_{\max}$	$N(r)$
[DBUH <sup>+</sup> ][Im <sup>-</sup> ]	3.45	1.29	1.99	3.45	1.29	2.04
[DBUH <sup>+</sup> ][4-MeIm <sup>-</sup> ]	3.35	1.24	1.81	3.35	1.22	1.83
[DBUH <sup>+</sup> ][2-MeIm <sup>-</sup> ]	3.35	1.13	1.68	3.35	1.17	1.68
[DBUH <sup>+</sup> ][2-EtIm <sup>-</sup> ]	3.45	0.84	1.05	3.35	1.22	1.58
[DBUH <sup>+</sup> ][2-PrIm <sup>-</sup> ]	3.35	1.18	1.58	3.45	0.83	1.11
[DBUH <sup>+</sup> ][2-iPrIm <sup>-</sup> ]	3.35	1.03	1.26	3.35	1.00	1.28
[DBUH <sup>+</sup> ][2-Et-4-MeIm <sup>-</sup> ]	3.45	0.81	1.10	3.35	1.33	1.45
[DBUH <sup>+</sup> ][2-PhIm <sup>-</sup> ]	3.35	1.13	1.35	3.35	1.26	1.42

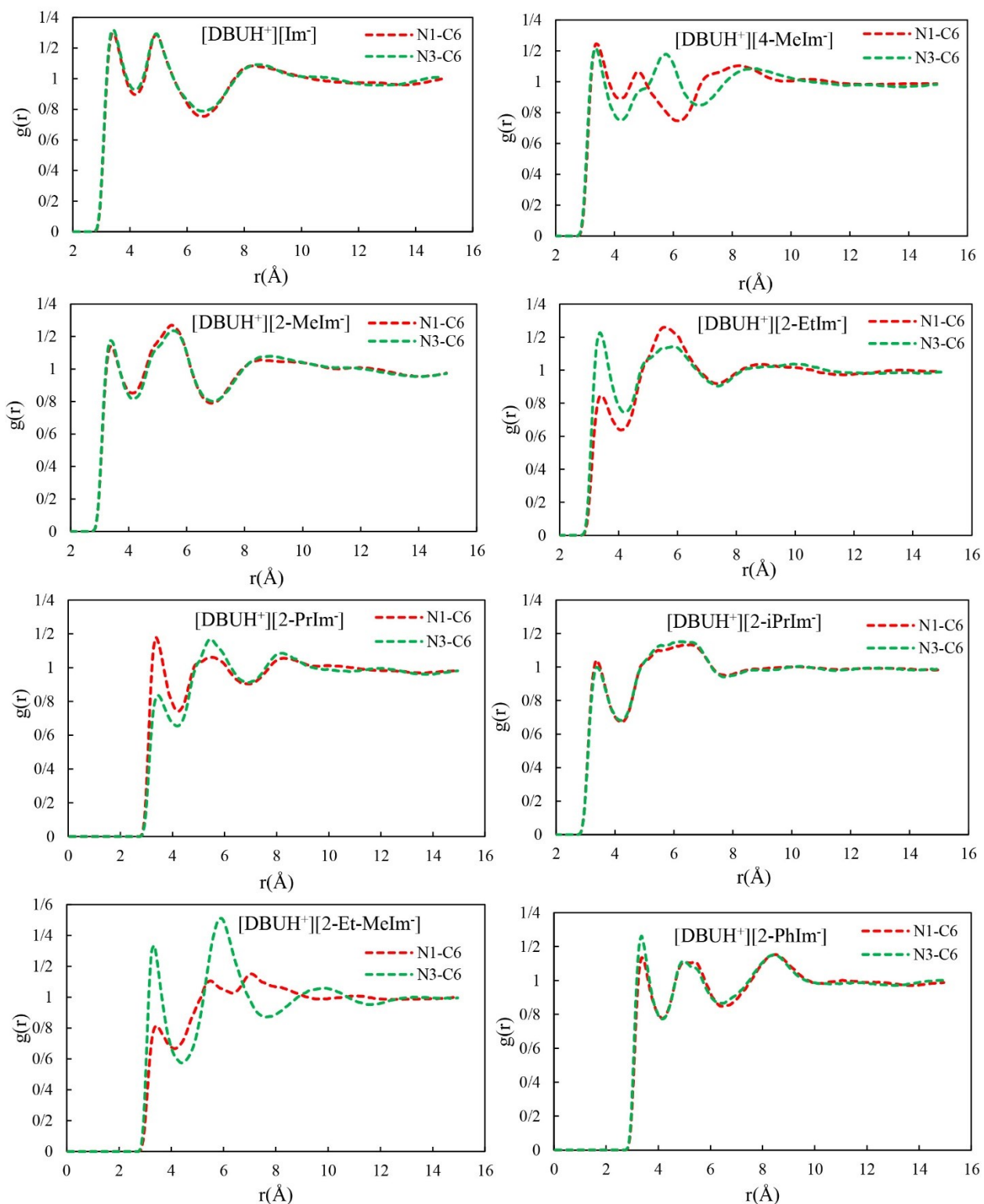


Figure S2. Calculated RDF for interaction between the functionalized imidazole anion and  $\text{CO}_2$  after 20 ns MD simulation in N1... C6 and N3... C6 atoms in the presence of water molecules.

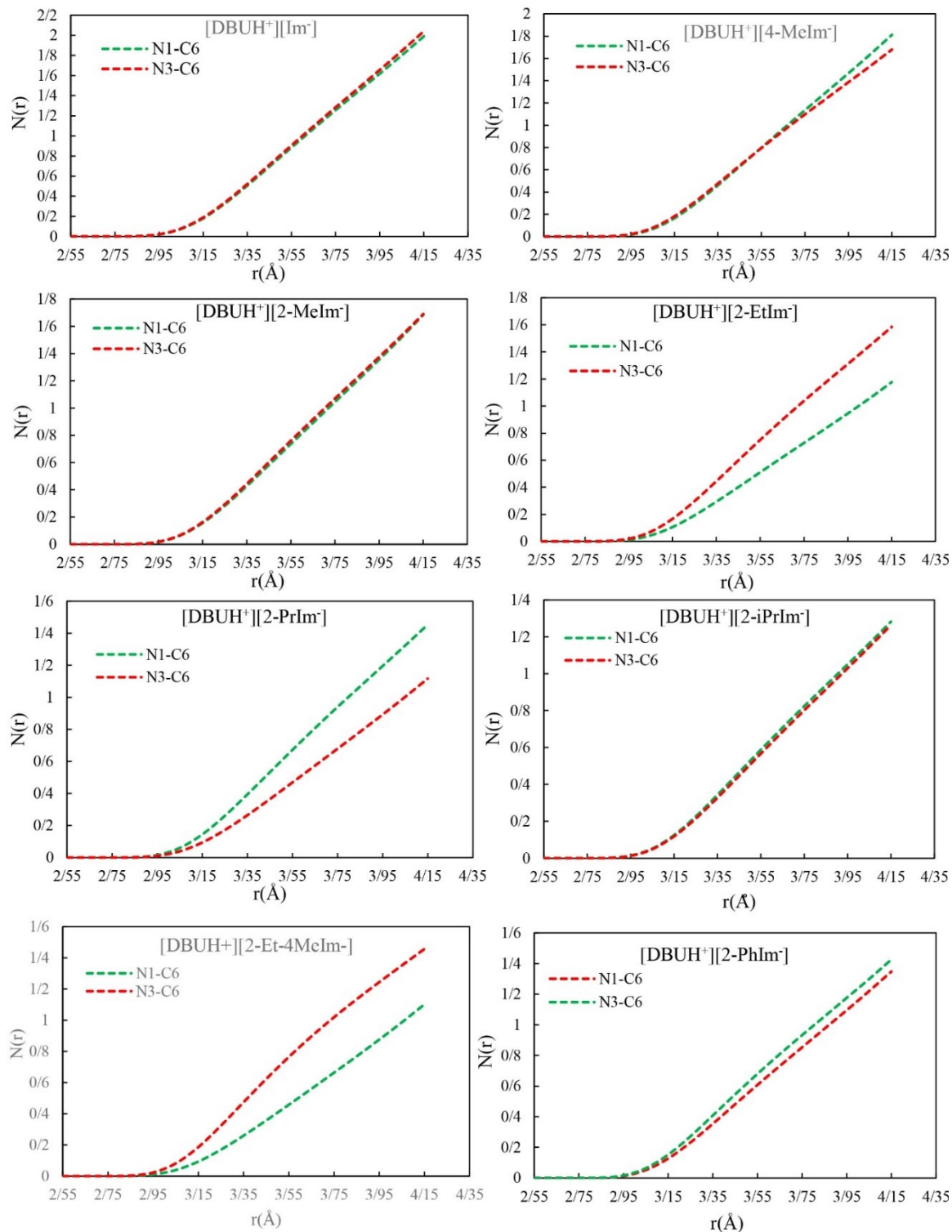


Figure S3. The coordination number of  $\text{CO}_2$  around the Imidazole anion for  $\text{N1}\cdots\text{C6}$  and  $\text{N3}\cdots\text{C6}$  interactions in different DBU-based ILs after 20 ns MD simulations in the presence of water molecules.

Table S2. The  $r_{\max}$  and  $g(r)_{\max}$  for interaction between N1,N3, O9 and H10 atoms of imidazole and water molecules.

Compound	N1-H10 ( $r_{\max}$ )	$g(r)_{\max}$	N3-H10 ( $r_{\max}$ )	$g(r)_{\max}$	N3-H10 ( $r_{\max}$ )	$g(r)_{\max}$
[DBUH <sup>+</sup> ][Im <sup>-</sup> ]	1.85	9.40	1.85	9.46	1.85	1.33
[DBUH <sup>+</sup> ][4-MeIm <sup>-</sup> ]	1.85	12.06	1.85	7.58	1.85	1.87
[DBUH <sup>+</sup> ][2-MeIm <sup>-</sup> ]	1.85	8.95	1.85	8.43	1.85	2.41
[DBUH <sup>+</sup> ][2-EtIm <sup>-</sup> ]	1.85	10.85	1.85	5.67	1.85	1.88
[DBUH <sup>+</sup> ][2-PrIm <sup>-</sup> ]	1.85	7.23	1.85	9.50	1.85	2.41
[DBUH <sup>+</sup> ][2-iPrIm <sup>-</sup> ]	1.85	7.23	1.85	8.35	1.85	3.46
[DBUH <sup>+</sup> ][2-Et-4-MeIm <sup>-</sup> ]	1.85	12.71	1.85	5.83	1.85	3.77
[DBUH <sup>+</sup> ][2-PhIm <sup>-</sup> ]	1.85	7.89	1.85	7.08	1.85	4.36

Table S3. The Average H-bond angle (°) for interaction between anion and water molecules.

Compound	Average H-bond angle (°)
[DBUH <sup>+</sup> ][Im <sup>-</sup> ]	162.307
[DBUH <sup>+</sup> ][4-MeIm <sup>-</sup> ]	162.70
[DBUH <sup>+</sup> ][2-MeIm <sup>-</sup> ]	161.85
[DBUH <sup>+</sup> ][2-EtIm <sup>-</sup> ]	161.95
[DBUH <sup>+</sup> ][2-PrIm <sup>-</sup> ]	160.74
[DBUH <sup>+</sup> ][2-iPrIm <sup>-</sup> ]	160.73
[DBUH <sup>+</sup> ][2-Et-4-MeIm <sup>-</sup> ]	161.59
[DBUH <sup>+</sup> ][2-PhIm <sup>-</sup> ]	161.00

Table S4. C-N bond length changes in the reactants, TSs and product in gas phase.

ILs	Bond Length C6-N(1,3) (Å)				
	Paths	Reactant	Product	TS	$\Delta r^\ddagger = r_{TS} - r_R$
[DBUH <sup>+</sup> ][Im <sup>-</sup> ]	A	2.68	1.51	2.20	-0.48
	B	2.63	1.51	2.25	-0.38
[DBUH <sup>+</sup> ][4-MeIm <sup>-</sup> ]	A	2.61	1.49	2.22	-0.39
	B	2.60	1.50	2.25	-0.35
[DBUH <sup>+</sup> ][2-MeIm <sup>-</sup> ]	A	2.70	1.50	2.21	-0.49
	B	2.69	1.50	2.28	-0.41
[DBUH <sup>+</sup> ][2-EtIm <sup>-</sup> ]	A	2.64	1.52	2.22	-0.42
	B	2.67	1.51	2.13	-0.54
[DBUH <sup>+</sup> ][2-PrIm <sup>-</sup> ]	A	2.65	1.47	2.21	-0.44
	B	2.64	1.53	2.26	-0.38
[DBUH <sup>+</sup> ][2-iPrIm <sup>-</sup> ]	A	2.66	1.49	2.20	-0.46
	B	2.66	1.50	2.15	-0.51
[DBUH <sup>+</sup> ][2-Et-4-MeIm <sup>-</sup> ]	A	2.63	1.52	2.23	-0.40
	B	5.68	1.55	2.14	-0.54
[DBUH <sup>+</sup> ][2-PhIm <sup>-</sup> ]	A	----	----	----	----
	B	2.70	1.52	1.84	-0.86

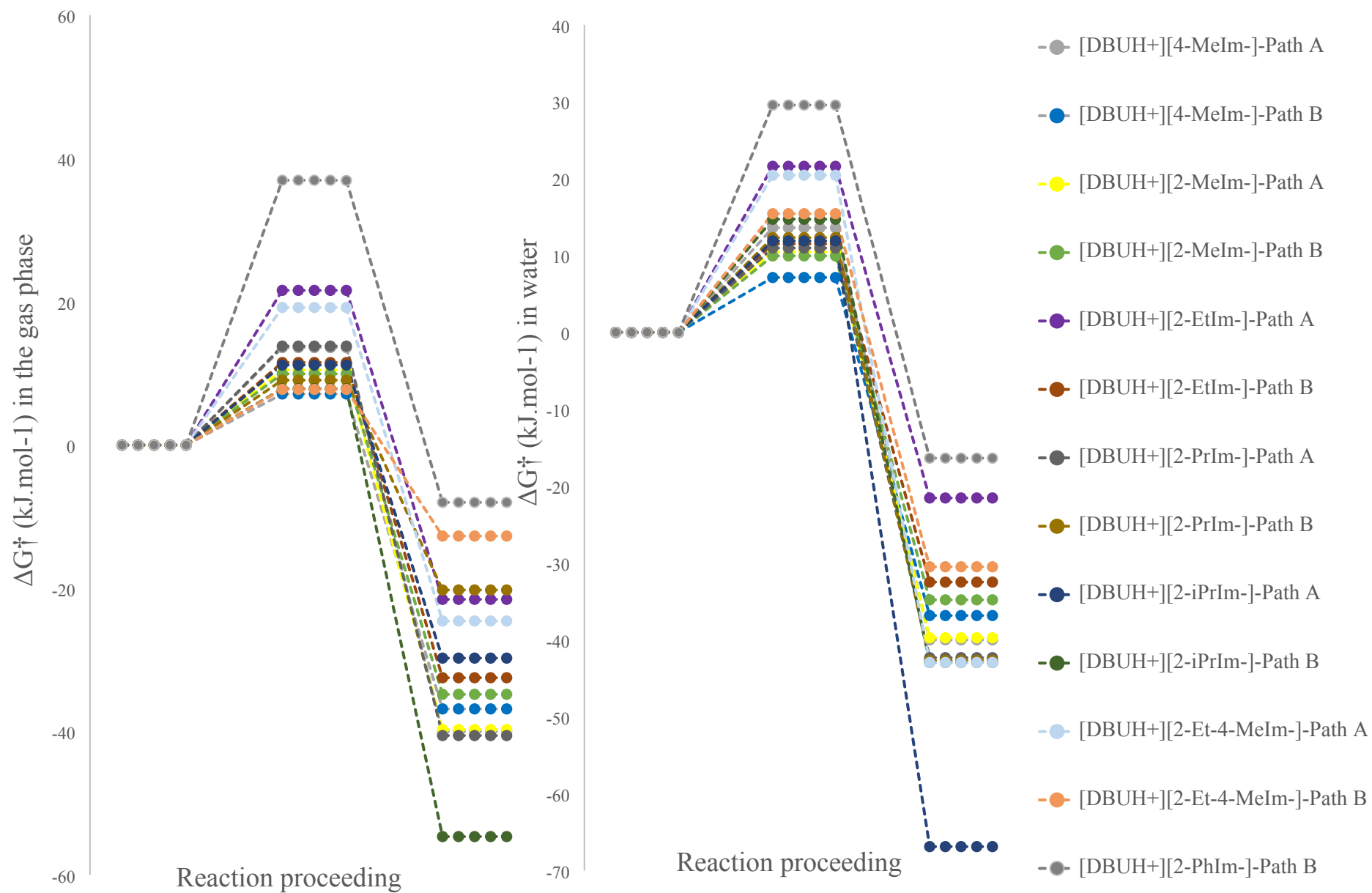


Figure S4. The potential energy diagram of CO<sub>2</sub> absorption by [DBUH<sup>+</sup>][RIm<sup>-</sup>] in the gas phase and water.

Table S5. Main donor-acceptor interaction energies in the reactants, TSs and product at M06-2X level of theory.

	Donor $\rightarrow$ Acceptor (Reactant)	$E(2)_R^*$ (kcal.mol <sup>-1</sup> )	Donor $\rightarrow$ Acceptor (TS)	$E(2)_{TS}^*$ (kcal.mol <sup>-1</sup> )	Donor $\rightarrow$ Acceptor (Product)	$E(2)_P^*$ (kcal.mol <sup>-1</sup> )
[DBUH <sup>+</sup> ][Im <sup>-</sup> ] (Path A)	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	5.78	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	28.69	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	34.60
	lp <sub>C6-08</sub> $\rightarrow\sigma^*$ <sub>N3</sub>	8.43	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	4.20	lp <sub>O7</sub> $\rightarrow\sigma^*$ <sub>N1-C6</sub>	36.98
	$\sum E(2)_R$	14.21	$\sum E(2)_{TS}$	32.89	$\sum E(2)_P$	71.58
[DBUH <sup>+</sup> ][Im <sup>-</sup> ] (Path B)	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	5.78	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	2.99	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	27.71
	lp <sub>C6-08</sub> $\rightarrow\sigma^*$ <sub>N3</sub>	8.43	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	24.41	lp <sub>O7</sub> $\rightarrow\sigma^*$ <sub>N3-C6</sub>	41.89
	$\sum E(2)_R$	14.21	$\sum E(2)_{TS}$	27.40	$\sum E(2)_P$	69.60
[DBUH <sup>+</sup> ][4-MeIm <sup>-</sup> ] (Path A)	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	60.54	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	26.82	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	35.56
	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	19.94	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	3.96	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>N1-C6</sub>	36.44
	$\sum E(2)_R$	80.48	$\sum E(2)_{TS}$	30.78	$\sum E(2)_P$	72.00
[DBUH <sup>+</sup> ][4-MeIm <sup>-</sup> ] (Path B)	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	60.54	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	26.05	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	26.79
	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	19.94	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	2.22	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>N1-C6</sub>	19.71
	$\sum E(2)_R$	80.48	$\sum E(2)_{TS}$	28.27	$\sum E(2)_P$	46.50
[DBUH <sup>+</sup> ][2-MeIm <sup>-</sup> ] (Path A)	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	4.36	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	26.56	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	36.50
	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	14.48	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	4.81	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>N1-C6</sub>	34.65
	$\sum E(2)_R$	18.84	$\sum E(2)_{TS}$	31.37	$\sum E(2)_P$	71.15
[DBUH <sup>+</sup> ][2-MeIm <sup>-</sup> ] (Path B)	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	4.36	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	23.34	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	24.94
	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	14.48	lp <sub>N3</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	2.61	lp <sub>O7</sub> $\rightarrow\pi^*$ <sub>N3-C6</sub>	41.08
	$\sum E(2)_R$	18.84	$\sum E(2)_{TS}$	25.95	$\sum E(2)_P$	39.02
[DBUH <sup>+</sup> ][2-EtIm <sup>-</sup> ] (Path A)	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	1.40	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	3.73	lp <sub>N1-C2</sub> $\rightarrow\pi^*$ <sub>C6-08</sub>	22.71
	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	11.77	lp <sub>N1</sub> $\rightarrow\pi^*$ <sub>C6-07</sub>	30.85	lp <sub>O8</sub> $\rightarrow\pi^*$ <sub>N1-C6</sub>	40.16
	$\sum E(2)_R$	13.17	$\sum E(2)_{TS}$	34.58	$\sum E(2)_P$	62.87



[DBUH <sup>+</sup> ][2-EtIm <sup>-</sup> ] (Path B)	lp <sub>N1</sub> →π* <sub>C6-07</sub>	1.40	lp <sub>N3</sub> →π* <sub>C6-08</sub>	43.78	lp <sub>N3</sub> →π* <sub>C6-08</sub>	33.34
	lp <sub>O8</sub> →π* <sub>C6-07</sub>	11.77	lp <sub>N3</sub> →π* <sub>C6-07</sub>	1.75	lp <sub>O8</sub> →π* <sub>N3-C6</sub>	36.64
	ΣE(2) <sub>R</sub>	13.17	ΣE(2) <sub>TS</sub>	45.53	ΣE(2) <sub>P</sub>	69.98
[DBUH <sup>+</sup> ][2-PrIm <sup>-</sup> ] (Path A)	lp <sub>N1</sub> →π* <sub>C6-07</sub>	5.45	lp <sub>N1</sub> →π* <sub>C6-08</sub>	6.34	lp <sub>N1</sub> →π* <sub>C6-07</sub>	44.38
	lp <sub>O8</sub> →π* <sub>C6-07</sub>	37.61	lp <sub>N1</sub> →π* <sub>C6-07</sub>	9.56	lp <sub>O8</sub> →π* <sub>N1-C6</sub>	31.35
	ΣE(2) <sub>R</sub>	43.06	ΣE(2) <sub>TS</sub>	15.9	ΣE(2) <sub>P</sub>	75.73
[DBUH <sup>+</sup> ][2-PrIm <sup>-</sup> ] (Path B)	lp <sub>N3</sub> →π* <sub>C6-08</sub>	5.45	lp <sub>N3</sub> →π* <sub>C6-08</sub>	25.02	lp <sub>N3</sub> →π* <sub>C6-08</sub>	29.71
	lp <sub>O8</sub> →π* <sub>C6-07</sub>	37.61	lp <sub>N3</sub> →π* <sub>C6-07</sub>	2.50	lp <sub>O8</sub> →π* <sub>N3-C6</sub>	38.87
	ΣE(2) <sub>R</sub>	43.06	ΣE(2) <sub>TS</sub>	27.52	ΣE(2) <sub>P</sub>	68.58
[DBUH <sup>+</sup> ][2-iPrIm <sup>-</sup> ] (Path A)	lp <sub>N1</sub> →π* <sub>C6-07</sub>	5.20	lp <sub>N1</sub> →π* <sub>C6-08</sub>	6.34	lp <sub>N1</sub> →π* <sub>C6-08</sub>	37.81
	lp <sub>O8</sub> →π* <sub>C6-07</sub>	14.46	lp <sub>N1</sub> →π* <sub>C6-07</sub>	9.56	lp <sub>O8</sub> →π* <sub>N1-C6</sub>	34.28
	ΣE(2) <sub>R</sub>	19.66	ΣE(2) <sub>TS</sub>	15.90	ΣE(2) <sub>P</sub>	72.09
[DBUH <sup>+</sup> ][2-iPrIm <sup>-</sup> ] (Path B)	lp <sub>N3</sub> →π* <sub>C6-07</sub>	5.20	lp <sub>N3</sub> →π* <sub>C6-07</sub>	38.42	lp <sub>N3</sub> →π* <sub>C6-07</sub>	24.42
	lp <sub>O7</sub> →π* <sub>C6-08</sub>	14.46	lp <sub>N3</sub> →π* <sub>C6-08</sub>	1.79	lp <sub>O7</sub> →π* <sub>N3-C6</sub>	41.10
	ΣE(2) <sub>R</sub>	19.66	ΣE(2) <sub>TS</sub>	40.21	ΣE(2) <sub>P</sub>	65.52
[DBUH <sup>+</sup> ][2-Et-4-MeIm <sup>-</sup> ] (Path A)	lp <sub>N1</sub> →π* <sub>C6-07</sub>	1.49	lp <sub>N1</sub> →π* <sub>C6-08</sub>	3.57	lp <sub>O7</sub> →π* <sub>N1-C6</sub>	38.43
	lp <sub>O8</sub> →π* <sub>C6-07</sub>	11.72	lp <sub>N1</sub> →π* <sub>C6-07</sub>	29.35	lp <sub>O8</sub> →π* <sub>N1-C6</sub>	39.44
	ΣE(2) <sub>R</sub>	13.21	ΣE(2) <sub>TS</sub>	32.92	ΣE(2) <sub>P</sub>	77.87
[DBUH <sup>+</sup> ][2-PhIm <sup>-</sup> ] (Path B)	lp <sub>N3</sub> →π* <sub>C6</sub>	6.25	lp <sub>N3</sub> →π* <sub>C6-08</sub>	1.12	lp <sub>N3</sub> →π* <sub>C6-07</sub>	25.72
	lp <sub>O8</sub> →π* <sub>C6-07</sub>	15.61	lp <sub>N3</sub> →π* <sub>C6-07</sub>	2.14	lp <sub>O7</sub> →π* <sub>N3-C6</sub>	43.00
	ΣE(2) <sub>R</sub>	21.86	ΣE(2) <sub>TS</sub>	3.26	ΣE(2) <sub>P</sub>	68.72

Table S6. Topological parameters at the BCPs of N1-C6 and N3-C6 in the gas phase and water.

ILs	Path	BCP	R				TS			
			$\rho(r)$	$\nabla^2\rho$	V(r)	-G/V	$\rho(r)$	$\nabla^2\rho$	V(r)	-G/V
[DBUH <sup>+</sup> ][Im <sup>-</sup> ]	A (g)	N1-C6	0.0190	0.0631	-0.0134	1.09	0.0489	0.1034	-0.0367	0.852
	B (g)	N3-C6	0.0190	0.0631	-0.0134	1.09	0.0417	0.0969	-0.0303	0.900
[DBUH <sup>+</sup> ][4-MeIm <sup>-</sup> ]	A (g)	N1-C6	0.0203	0.0660	-0.0145	1.07	0.0475	0.1029	-0.0355	0.862
	B (g)	N3-C6	0.0203	0.0660	-0.0145	1.07	0.0444	0.0995	-0.0326	0.881
[DBUH <sup>+</sup> ][2-MeIm <sup>-</sup> ]	A (g)	N1-C6	0.0170	0.0567	-0.117	1.05	0.0486	0.1019	-0.0362	0.852
	B (g)	N3-C6	0.0170	0.0567	-0.117	1.05	0.0417	0.0969	-0.0303	0.900
[DBUH <sup>+</sup> ][2-EtIm <sup>-</sup> ]	A (g)	N1-C6	0.0188	0.0642	-0.0136	1.09	0.0477	0.1021	-0.0357	0.857
	B (g)	N3-C6	0.0188	0.0642	-0.0136	1.09	0.0575	0.0983	-0.0427	0.787
[DBUH <sup>+</sup> ][2-PrIm <sup>-</sup> ]	A (g)	N1-C6	0.0190	0.0633	-0.0135	1.09	0.0480	0.1022	-0.0359	0.856
	B (g)	N3-C6	0.0190	0.0633	-0.0135	1.09	0.0432	0.0976	-0.0314	0.888
[DBUH <sup>+</sup> ][2-iPrIm <sup>-</sup> ]	A (g)	N1-C6	0.0180	0.0607	-0.0127	1.10	0.0497	0.1008	-0.0369	0.842
	B (g)	N3-C6	0.0180	0.0607	-0.0127	1.10	0.0554	0.0993	-0.0411	0.802
[DBUH <sup>+</sup> ][2-Et-4-MeIm <sup>-</sup> ]	A (g)	N1-C6	0.0191	0.0647	-0.0138	1.09	0.0465	0.101	-0.0347	0.866
	B (g)	N3-C6	0.0191	0.0647	-0.0138	1.09	0.0564	0.0975	-0.0416	0.793
[DBUH <sup>+</sup> ][2-PhIm <sup>-</sup> ]	B (g)	N3-C6	0.0163	0.0570	-0.0115	1.12	0.0110	0.0400	-0.1030	0.548

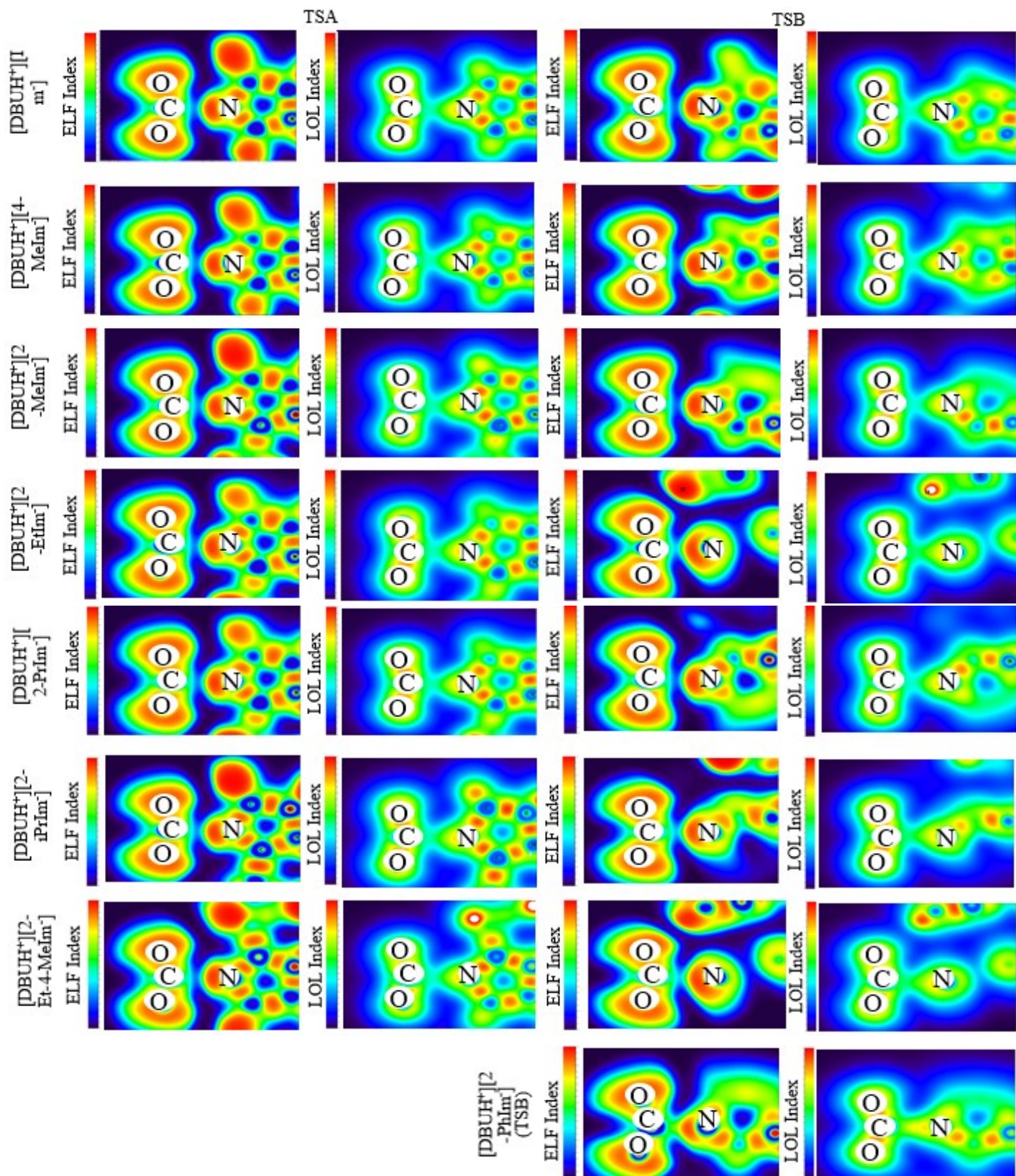


Figure S5. The ELF and LOL diagrams of the N1...C6 and N3...C6 interactions at the BCPs of the TSA and TSB.