

Electronic Supplementary Information

Franck–Condon simulations of transition state spectra for the OH + H₂O and OD + D₂O reactions

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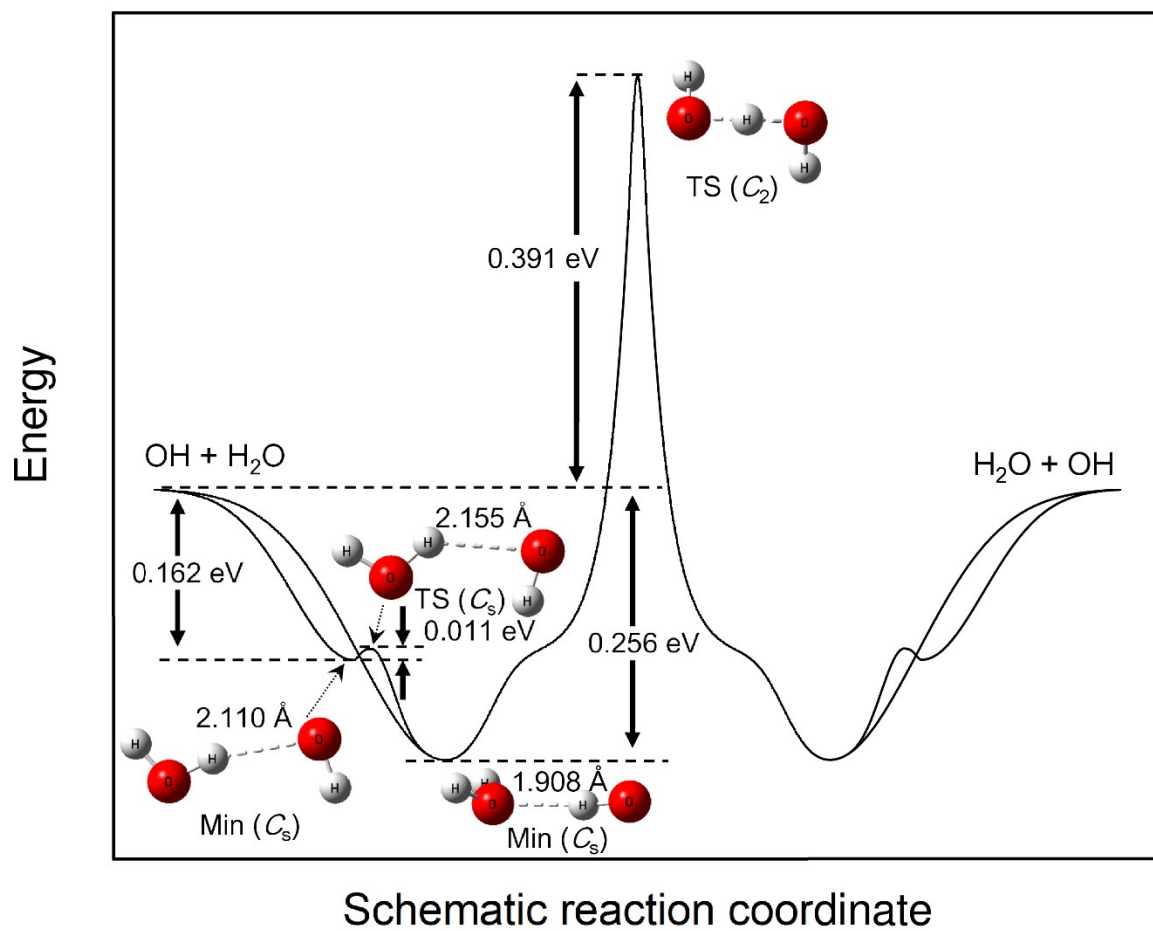


Figure S1

Schematic energy diagram of the neutral H₃O₂ reaction system. Energies at stationary points are obtained from the *ab initio*-level PIP-NN potential energy surface employed in the present study.

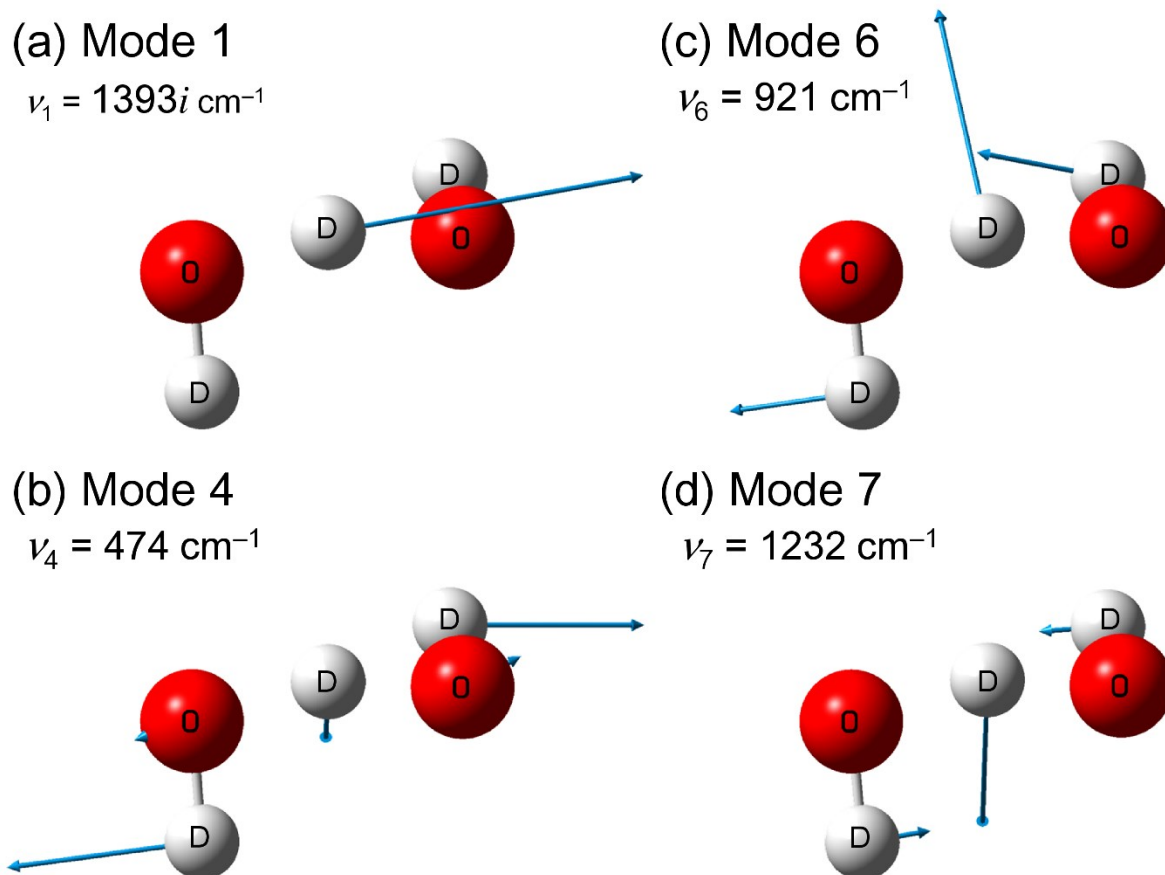


Figure S2

Normal-mode coordinates used in the quantum dynamics calculations as active nuclear degrees of freedom, obtained from vibrational frequency analysis of the C_2 transition-state structure for the $\text{OD} + \text{D}_2\text{O} \rightarrow \text{D}_2\text{O} + \text{OD}$ reaction. Mode 1 corresponds to the vibrational mode with an imaginary frequency.

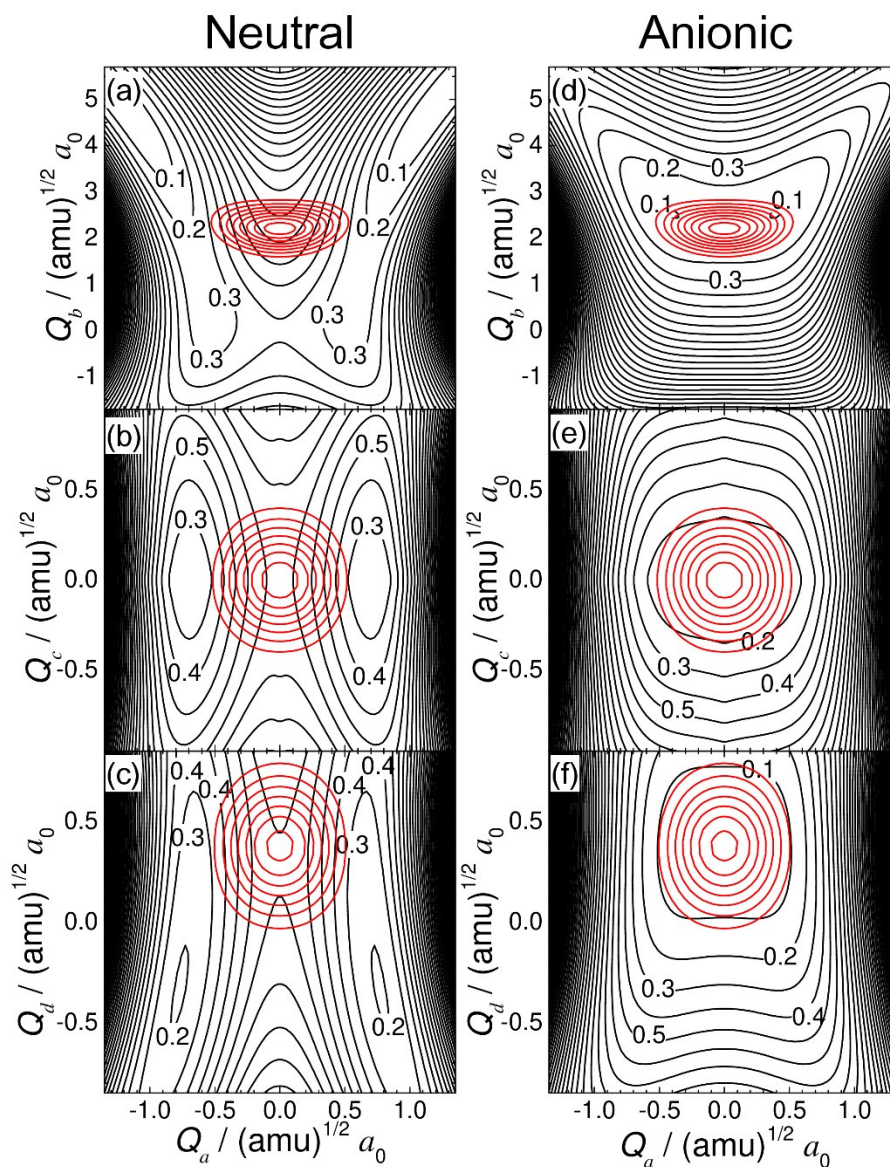


Figure S3

Two-dimensional contour plots of the potential energy surfaces of D_3O_2 and D_3O_2^- . The contour increment was set to 0.1 eV. Left and right panels show the neutral and anionic potential energy surfaces, respectively. Panels (a) and (d) depict the surfaces plotted as a function of Q_a and Q_b . Panels (b) and (e) depict the surfaces plotted as a function of Q_a and Q_c . Panels (c) and (f) depict the surfaces plotted as a function of Q_a and Q_d . The superimposed red contour lines indicate the projection of the initial anion wavefunction density in the ground vibrational state.

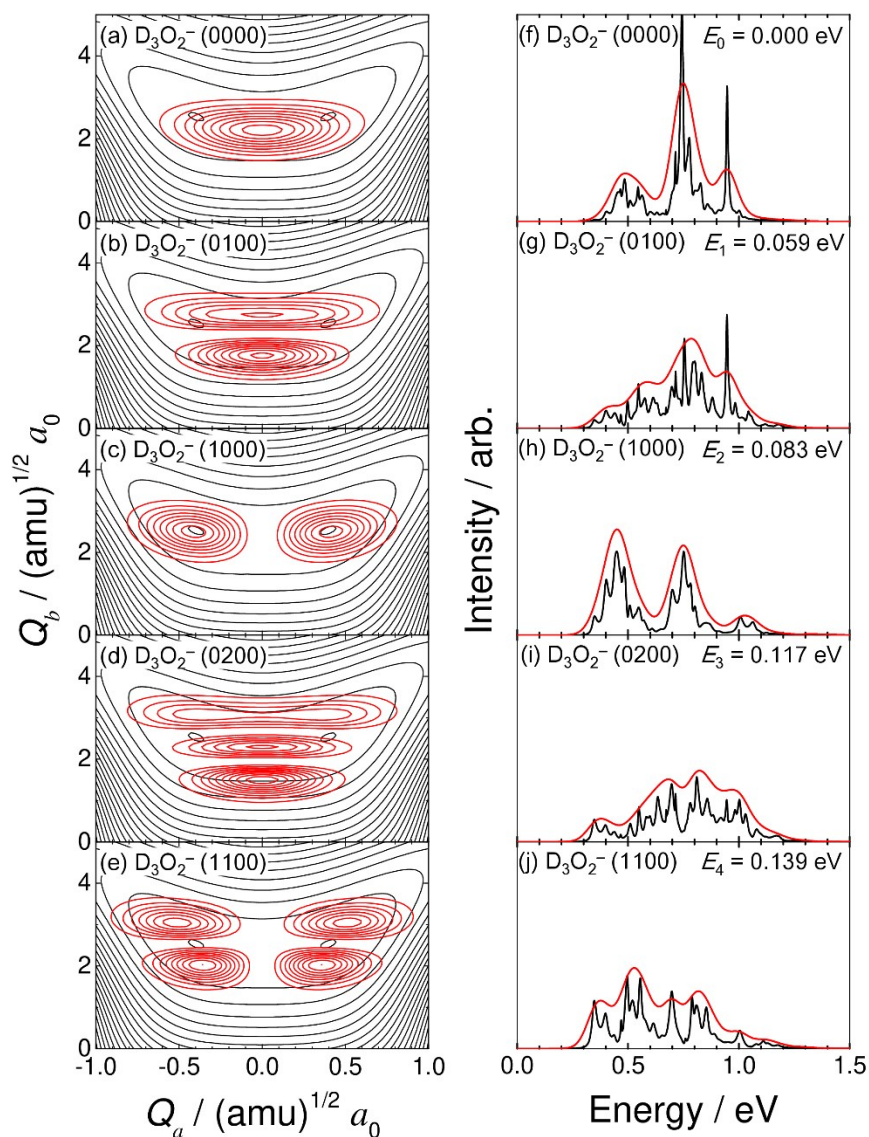


Figure S4

Left panels: vibrational wavefunction densities of the $D_3O_2^-$ anion plotted as a function of Q_a and Q_b coordinates. The state can be described by a set of quantum numbers (v_a, v_b, v_c, v_d) , where v_a , v_b , v_c , and v_d are the vibrational quantum numbers for the Q_a , Q_b , Q_c , and Q_d coordinates, respectively. Right panels: photodetachment spectrum calculated using each vibrational wavefunction as an initial wave packet. The calculated spectra are plotted as a function of the energy defined by the neutral potential energy surface.