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- 1 Understanding the nature of NH<sub>3</sub>-coordinated active sites and complete reaction schemes for NH<sub>3</sub>-SCR using
- 2 Cu-SAPO-34 catalysts
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- 12 Fig. S1 Optimized periodic structures of Cu-SAPO-34 with (a) ZCu, (b) ZCuOH and (c) Z<sub>2</sub>Cu species, and
- 13 (d) H-SAPO-34 configuration with Si-OH-Al in  $2 \times 1 \times 1$  supercells. Green, red, purple, yellow, orange, and
- 14 white balls represent P, O, Al, Si, Cu, and H atoms, respectively. This assignment will be applied throughout
- 15 the paper.



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- 17 Fig. S2 Energy profiles for (a) NO adsorption on Z<sub>2</sub>Cu-3NH<sub>3</sub> complex, and (b) NH<sub>4</sub>NO<sub>2</sub> formation on
- 18 ZCuOH-3NH<sub>3</sub> complex. Blue balls represent N atoms, and this assignment will be applied throughout the
- 19 paper.



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- 21 Fig. S3 Energy profiles for NH<sub>2</sub>NO decomposition processes on (a) ZCuOH-2NH<sub>3</sub> complex and (b) Z<sub>2</sub>Cu-
- 22 2NH<sub>3</sub> complex.



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- 24 Fig. S4 Energy profiles for NH<sub>2</sub>NO formation and decomposition processes on ZCuOH-3NH<sub>3</sub>.



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- 26 Fig. S5 Structures of two ZCu-2NH<sub>3</sub> complexes in separate cages and one cage, and energy difference of
- 27 them.



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 $29 \quad \mbox{Fig. S6 Energy profiles for the dissociation of $O_2$ on (a) a single $ZCu-2NH_3$ site and (b) a pair of $ZCu-2NH_3$ site and (b) a pair of $ZCu-2NH_3$ site and (b) a pair of $ZCu-2NH_3$ site and (c) a pair of $ZC$ 





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- 32 Fig. S7 Energy profiles for the re-oxidation process to produce Z<sub>2</sub>Cu-2NH<sub>3</sub> complex. Atomic distances are
- 33 in Å.