Electronic Supplementary Information

for

Reactions of a photoactivatable diazido Pt(IV) anticancer complex with a single-stranded oligodeoxynucleotide

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Table S1. MS data for the reaction between Pt(IV) complex 1 and ODN I at a molar ratio of 1/I = 1.0 after irradiated under blue light for 1 h (Charges for Pt moiety and the loss of protons from I for balancing the charges of the ions are omitted for clarity).

		m/z	m/z	Error
Ion	Formula	observed	calculated	(ppm)
${I + [Pt(N_3)(py)_2]}^{6-}$	C154H198N44O96P14Pt	688.962	688.961	1.45
${I + [Pt(N_3)(py)_2]_2}^{6-}$	$C_{164}H_{207}N_{49}O_{96}P_{14}Pt_2$	869.635	869.633	2.30
[I] ^{5–}	$C_{144}H_{189}N_{39}O_{96}P_{14}$	886.138	886.141	-3.39
${I + [Pt(N_3)(py)_2]_3}^{6-}$	$C_{174}H_{216}N_{54}O_{96}P_{14}Pt_3$	935.305	935.320	-16.04
${I + [Pt(py)_2]}^{5-}$	$C_{164}H_{207}N_{49}O_{96}P_{14}Pt$	956.349	956.352	-3.14
${I + [Pt(N_3)(py)_2]}^{5-}$	C154H198N44O96P14Pt	964.949	964.953	-4.15
${I + [Pt(py)_2] + [Pt(N_3)(py)_2]}^{5-}$	$C_{164}H_{206}N_{46}O_{96}P_{14}Pt_2$	1035.159	1035.156	2.90
${I + [Pt(N_3)(py)_2]_2}^{5-}$	$C_{164}H_{207}N_{49}O_{96}P_{14}Pt_2$	1043.761	1043.766	-4.79
$[I]^{4-}$	$C_{144}H_{189}N_{39}O_{96}P_{14}$	1107.931	1107.930	0.90
${I + [Pt(N_3)(py)_2]_3}^{5-}$	$C_{174}H_{216}N_{54}O_{96}P_{14}Pt_3$	1122.573	1122.570	2.67
${I + [Pt(py)_2]}^{4-}$	C154H198N41O96P14Pt	1195.686	1195.688	-1.67
${I + [Pt(N_3)(py)_2]}^{4-}$	C154H198N44O96P14Pt	1206.439	1206.445	-4.97
${I + [Pt(py)_2] + [Pt(N_3)(py)_2]}^{4-}$	$C_{164}H_{206}N_{46}O_{96}P_{14}Pt_2$	1294.204	1294.204	0.00
${\mathbf{I} + [Pt(N_3)(py)_2]_2}^{4-}$	$C_{164}H_{207}N_{49}O_{96}P_{14}Pt_2$	1304.953	1304.953	0.00
$\{I + [Pt(py)_2] + [Pt(N_3)(py)_2]_2\}^{4-}$	$C_{174}H_{216}N_{51}O_{96}P_{14}Pt_3$	1392.715	1392.711	2.87
${\mathbf{I} + [Pt(N_3)(py)_2]_3}^{4-}$	C ₁₇₄ H ₂₁₆ N ₅₄ O ₉₆ P ₁₄ Pt ₃	1403.473	1403.469	2.85
[I] ^{3–}	C ₁₄₄ H ₁₈₉ N ₃₉ O ₉₆ P ₁₄	1477.242	1477.242	0.00
$[I + Na]^{3-}$	C144H188N39O96P14Na	1484.573	1484.570	2.02
${I + [Pt(py)_2]_2 + $	C II N O D Dt	1401 226	1401 227	0.67
$[Pt(N_3)(py)_2]_2\}^{4-}$	$C_{184}H_{224}N_{56}O_{96}P_{14}Pt_{4}$	1491.220	1491.227	-0.07
$\{I + [Pt(N_3)(py)_2]_4\}^{4-}$	$C_{184}H_{225}N_{59}O_{96}P_{14}Pt_4$	1502.230	1502.234	-2.66
${I + [Pt(py)_2]}^{3-}$	$C_{154}H_{198}N_{41}O_{96}P_{14}Pt$	1594.593	1594.587	3.76
${I + [Pt(N_3)(py)_2]}^{3-}$	$C_{154}H_{198}N_{44}O_{96}P_{14}Pt$	1608.932	1608.922	6.22
${I + [Pt^{III}(N_3)(OH)(py)_2]}^{3-}$	$C_{154}H_{199}N_{44}O_{97}P_{14}Pt$	1614.594	1614.596	1.24
${[I + 3O] + [Pt^{III}(OH)(py)_2]}^{3-}$	$C_{154}H_{198}N_{41}O_{100}P_{14}Pt \\$	1616.261	1616.250	6.80
$\{[I + H_2O + (OH)_2] +$				
$[Pt(N_3)(py)_2]$ ³⁻ or {[I + O +	$C_{154}H_{202}N_{44}O_{99}P_{14}Pt$	1626.269	1626.266	1.84
$(H_2O)_2] + [Pt(N_3)(py)_2] \}^{3-}$				
$\{[I + (OH)_2] + [Pt^{III}(N_3)_2(py)_2]$				
$+ Na$ } ³⁻ or {[I + O + H ₂ O] +	$C_{154}H_{200}N_{47}O_{98}P_{14}PtNa$	1641.588	1641.594	-3.65
$[Pt^{III}(N_3)_2(py)_2] + Na\}^{3-}$				
${I + [Pt(py)_2]_2}^{3-}$	$C_{164}H_{205}N_{43}O_{96}P_{14}Pt_2$	1711.599	1711.602	-1.75
${I + [Pt(py)_2] + [Pt(N_3)(py)_2]}^{3-}$	$C_{164}H_{206}N_{46}O_{96}P_{14}Pt_2$	1725.941	1725.938	1.74

$\{[I + H_2O] + [Pt(py)_2] +$	CiceHanoNicOozPiePta	1731 938	1731 944	-3 46
$[Pt(N_3)(py)_2]$ ³⁻	C164112091 46 C9/1 141 t2	1/51.750	1751.744	-3.40
$\{[I + 2O] + [Pt(py)_2] +$	C164H206N46O08P14Pt2	1736 608	1736 602	3 46
$[Pt(N_3)(py)_2]$ ³⁻	- 104 200- 140 - 98- 14- 12	- /		
${\mathbf{I} + [Pt(N_3)(py)_2]_2}^{3-}$	$C_{164}H_{207}N_{49}O_{96}P_{14}Pt_2$	1740.284	1740.273	6.32
$\{I + [Pt^{III}(N_3)(OH)(py)_2] + [Pt(N_1)(py)_1]\}^{3-1}$	$C_{164}H_{208}N_{49}O_{97}P_{14}Pt_2$	1745.952	1745.945	4.01
$\{\mathbf{I} + [Pt(N_3)(py)_2]_2 + Na\}^{3-}$	C ₁₆₄ H ₂₀₆ N ₄₉ O ₉₆ P ₁₄ Pt ₂ Na	1747.606	1747.602	2.29
$\{[I + H_2O] +$		1959 (19		0.00
$[Pt^{III}(N_3)(OH)(py)_2]_2\}^{3-}$	$C_{164}H_{211}N_{49}O_{99}P_{14}Pt_2$	1/5/.61/	1/5/.61/	0.00
${[I + 4O] +}$	C II N O D Dt	1772 027	1772 029	0.56
$[Pt^{III}(N_3)(OH)(py)_2]_2\}^{3-}$	$C_{164}H_{209}N_{49}O_{102}P_{14}Pt_2$	1//2.93/	1//2.938	-0.36
${I + [Pt(py)_2]_2 + $	C U N O D Dt	1012 200	18/2 201	2 80
$[Pt(N_3)(py)_2]$ ³⁻	C ₁₇₄ I ₂₁₅ IN48O96F 14F t3	1043.200	1043.201	5.80
${I + [Pt(py)_2] + $	CHau Neu Oac Pu Pta	1857 297	1857 289	431
$[Pt(N_3)(py)_2]_2\}^{3-}$	C ₁₇₄ I ₂₁₆ I V51O96I 14I V3	1037.277	1057.207	4.51
${[I + O] + [Pt(py)_2] + }$	C174HayNerOozP14Pta	1862 957	1862 953	2 1 5
$[Pt(N_3)(py)_2]_2\}^{3-}$	C1/4112161 (5109/1 141 (3	1002.757	1002.755	2.10
${I + [Pt(N_3)(py)_2]_3}^{3-}$	$C_{174}H_{216}N_{54}O_{96}P_{14}Pt_3$	1871.633	1871.625	4.27
${I + [Pt^{III}(N_3)(OH)(py)_2] +$	C174H217N54O07P14Pt3	1877.307	1877.297	5.33
$[Pt(N_3)(py)_2]_2\}^{3-}$	-1/4 21/ 54-9/ 14 5			
$\{[\mathbf{I} + \mathbf{H}_2\mathbf{O}] +$				
$[Pt^{III}(N_3)(OH)(py)_2]_2 +$	$C_{174}H_{220}N_{54}O_{99}P_{14}Pt_3$	1888.973	1888.961	6.35
$[Pt(N_3)(py)_2]$ ³⁻				
$\{\mathbf{I} + [Pt(py)_2] + [Pt(py)_2] \}$	$C_{184}H_{224}N_{56}O_{96}P_{14}Pt_4$	1988.986	1988.969	8.55
$[Pt(N_3)(py)_2]_3$		2002 225	2002 212	5.00
$\{I + [Pt(N_3)(py)_2]_4\}^3$	$C_{184}H_{225}N_{59}O_{96}P_{14}Pt_{4}$	2003.325	2003.313	5.99
$\{[I + (OH)_2 + H_2O] + [Pt(N_1)(m_1)_1]\}^3 - \text{or } \{[I + O + O] + [Pt(N_2)(m_1)_1]\}^3 - \text{or } \{[I + O] + O] + [Pt(N_2)(m_1)_1]\}^3 - (Pt(N_2)(m_1)_1]^3 - (Pt(N_2)$	CHNOP. Pt.	2020 665	2020 648	8 / 1
$(H_{2}O)_{2}$ + $[Pt(N_{2})(py)_{2}]_{4}^{3-}$	C184112291159C991 141 t4	2020.003	2020.040	0.41
$\{[I]_{2} + [Pt(nv)_{2}] + [I(1v_{3})(py)_{2}]_{4}\}$				
$[Pt(N_2)(nv)_2]_2$ ⁵⁻	$C_{328}H_{413}N_{95}O_{192}P_{28}Pt_4$	2080.143	2080.133	4.81
$\{\mathbf{I} + [Pt(N_3)(pv)_2]_2\}_2^{5-}$	C328H414N08O102P28Pt4	2088.748	2088.734	6.70
$\{\mathbf{I} + [Pt(pv)_2] + \}$	0 5201 4141 (98 0 1921 281 4	200001710	2000.701	0170
$[Pt(N_3)(py)_2]_4\}^{3-}$	$C_{194}H_{233}N_{61}O_{96}P_{14}Pt_5$	2120.340	2120.320	9.43
$\{\mathbf{I} + [Pt(N_3)(py)_2]_5\}^{3-1}$	$C_{194}H_{234}N_{64}O_{96}P_{14}Pt_5$	2134.681	2134.656	11.71
$\{[\mathbf{I} + (\mathbf{OH})_2 + \mathbf{H}_2\mathbf{O}] +$				
$[Pt(N_3)(py)_2]_5\}^{3-}$ or $\{[I + O +$	$C_{194}H_{238}N_{64}O_{99}P_{14}Pt_5$	2152.017	2152.000	7.90
$(H_2O)_2] + [Pt(N_3)(py)_2]_5\}^{3-}$				
${I + [Pt(N_3)(py)_2]_3}_2^{5-}$	$C_{348}H_{432}N_{108}O_{192}P_{28}Pt_6$	2246.381	2246.352	12.91

Table S2. Fragment ions observed by MS/MS analysis in negative-ion mode of mono-platinated I ($[I + 1']^{3-}$, *m/z* 1608.932) produced by the reaction of complex 1 with ODN I at 310 K after irradiation under blue light for 1 h. (Charges for Pt moiety and the loss of protons from I for balancing the charges of the ions are omitted for clarity). $\mathbf{1'} = [Pt(N_3)(py)_2]^+$.

	/ - 1		/ - 1 •	
Fragments	m/z^{a} observed	Fragments	m/z^{a} observed	
	(calculated)		(calculated)	
w_1^-	306.045 (306.047)	$[a_2 - T_2^b]^-$	386.068 (386.078)	
W_2^-	610.087 (610.094)	a_2^-	512.106 (512.117)	
W3 ⁻	899.136 (899.141)	$[a_3 - C_3^b]^-$	690.117 (690.125)	
w_4^-	1203.182 (1203.188)	$\{[a_3 - C_3] + 1'\}^-$	1084.163 (1084.172)	
W5 ⁻	1507.229 (1507.234)	$[a_4 - T_4]^-$	979.152 (979.168)	
W_5^{2-}	753.100 (753.117)	a_4^-	1105.212 (1105.211)	
$[w_5 + 1']^-$	1902.280 (1902.289)	$[a_5 - C_5]^-$	1283.208 (1283.211)	
w_{6}^{2-}	897.630 (897.633)	$\{[a_5 - C_5] + 1'\}^-$	1678.269 (1678.266)	
W7 ⁻	2100.332 (2100.328)	$[a_8 - G_8^b]^-$	2180.369 (2180.352)	
W7 ²⁻	1049.656 (1049.656)	$[a_8 - G_8]^{2-}$	1089.669 (1089.672)	
$[w_7 + 1']^{2-}$	1247.175 (1247.188)	$\{[a_8-G_8]+1'\}^{2-}$	1287.186 (1287.195)	
w_{10}^{-}	1518.731 (1518.734)	$[a_{10} - C_{10}]^{2-}$	1406.720 (1406.727)	
$[w_{10} + 1']^{2-}$	1715.761 (1715.758)	$\{[a_{10} - C_{10}] + \mathbf{1'}\}^{2-}$	1603.748 (1603.750)	
W_{12}^{2-}	1815.267 (1815.281)	$[a_{13} - C_{13}]^{2-}$	1855.293 (1855.290)	
$[w_{12} + 1']^{2-}$	2012.315 (2012.305)	$\{[a_{13} - C_{13}] + \mathbf{1'}\}^{2-}$	2052.337 (2052.333)	
$[w_{14} + 1']^{3-}$	1539.235 (1539.234)	$\{[a_{15} - C_{15}] + \mathbf{1'}\}^{2-}$	1565.909 (1565.906)	
$\{w_{14} + [\mathbf{1'} - N_3]\}^{3-}$	1524.899 (1524.891)			
$\{w_{14} + [\mathbf{1'} - py]\}^{3-}$	1512.887 (1512.883)			
$[w_{14} + 1' - C^c]^{3-}$	1502.217 (1502.219)			
$[w_{14} + 1' - G^c]^{3-}$	1488.886 (1488.883)			
$[I + 1']^{3-}$	1608.926 (1608.922)	$\{[I + 1'] - N_3\}^{3-}$	1594.588 (1594.586)	
$[I - G]^{3-}$	1426.890 (1426.891)	$\{[I + 1'] - py\}^{3-}$	1582.581 (1582.578)	
$\{[I + 1'] - C\}^{3-}$	1571.912 (1571.904)	$\{[I + 1'] - N_2 - py\}^{3-}$	1573.245 (1571.242)	
$\{[I + 1'] - G\}^{3-}$	1558.241 (1558.242)	$\{[I + 1'] - py - C\}^{3-}$	1545.566 (1545.563)	
$\{[I + 1'] - 2C\}^{3-}$	1534.894 (1534.898)	$\{[I + 1'] - py - G\}^{3-}$	1532.228 (1532.227)	
$\{[I + 1'] - G - C\}^{3-}$	1521.562 (1521.563)			
Internal fragments ^d				
$[T_6:T_9]^-$	1418.181 (1418.188)	[T _x :T _x] ⁻ (x=2,4,9,12,14)	481.035 (481.039)	
$\{[T_6:T_9] - G\}^-$	1267.127 (1267.144)	[T ₄ :T ₇] ⁻ /[T ₉ :T ₁₂] ⁻ / [T ₁₁ :T ₁₄] ⁻	1378.177 (1378.180)	

$[T_2:T_4]^-/[C_5:T_7]^-/$ $[C_{10}:T_{12}]^-$	1074.127 (1074.133)	${[T_4:T_7]/[T_9:T_{12}]/ [T_{11}:T_{14}] + 1'}^-$	1773.223 (1773.234)
${[T_2:T_4]/[C_5:T_7]/[C_{10}:T_{12}] + 1'}^-$	1468.187 (1468.188)	$[T_6:T_7]^-/[T_{11}:T_{12}]^-$	785.079 (785.086)
$[T_2:T_7]^{2-}/[T_9:T_{14}]^{2-}$	985.129 (985.133)	$\{[T_2:T_{14}] + \mathbf{1'}\}^{3-1}$	1496.219 (1496.211)
$[T_2:T_7]^-/[T_9:T_{14}]^-$	1971.275 (1971.273)	$[T_2:T_9]^{2-}/[C_5:T_{12}]^{2-}$	1302.184 (1302.180)
${[T_2:T_7]/[T_9:T_{14}] + 1'}^{2-}$	1182.656 (1182.656)	$\{[T_2:T_9]/[C_5:T_{12}] + 1'\}^{2-}$	1499.206 (1499.211)
$[T_2:T_{12}]^{2-}/[T_4:T_{14}]^{2-}$	1750.765 (1750.750)	$\{[T_4:T_9] - G\}^-$	1860.234 (1860.227)
${[T_2:T_{12}]/[T_4:T_{14}] + 1'}^{2-}$	1947.790 (1947.781)	$[T_4:T_{12}]^{2-}/[T_6:T_{14}]^{2-}$	1454.210 (1454.203)
		${[T_4:T_{12}]/[T_6:T_{14}] + 1'}^{2-}$	1651.226 (1651.234)

^aThe most abundant isotopic mass-to-charge ratio.

 ${}^{b}T_{n}$, C_{n} and G_{n} represent the loss of a cytosine and a guanine base, respectively, followed by elimination of a H₂O molecule to form a furan ring, n indicates the position of the base.

°C and G represent the neutral loss of a cytosine and a guanine base, respectively.

^dThe internal fragment $B_m:B_n$ results from fragmentation at both the a- and w-sites, having a phosphate group at the 5'-terminus and a furan ring at the 3'-terminus.

Table S3. Fragment ions observed by MS/MS analysis in negative-ion mode of diplatinated I ($[I + 1'_2]^{3-}$, *m/z* 1740.284) produced by the reaction of complex 1 with ODN I at 310 K after irradiation under blue light for 1 h. (Charges for Pt moiety and the loss of protons from I for balancing the charges of the ions are omitted for clarity). 1' = $[Pt(N_3)(py)_2]^+$.

Fragmonts	m/z^{a} observed	Fragmonts	m/z^{a} observed
	(calculated)	ragments	(calculated)
w_1^-	306.045 (306.047)	$[a_2 - T_2^b]^-$	386.068 (386.078)
W_2^-	610.093 (610.094)	a_2^-	512.119 (512.117)
W3 ⁻	899.136 (899.141)	$[a_3 - C_3^b]^-$	690.117 (690.125)
w_4^-	1203.182 (1203.188)	$\{[a_3 - C_3] + 1'\}^-$	1084.168 (1084.172)
W5 ⁻	1507.237 (1507.234)	$[a_4 - T_4]^-$	979.159 (979.168)
w ₅ ²⁻	753.107 (753.117)	a_4^-	1105.212 (1105.211)
$[w_5 + 1']^-$	1902.298 (1902.289)	$[a_5 - C_5]^-$	1283.208 (1283.211)
W6 ²⁻	897.630 (897.633)	$\{[a_5 - C_5] + 1'\}^-$	1678.261 (1678.266)
w_7^{2-}	1049.656 (1049.656)	$[a_8 - G_8^b]^-$	2180.379 (2180.352)
$[w_7 + 1']^{2-}$	1247.182 (1247.188)	$[a_8 - G_8]^{2-}$	1089.669 (1089.672)
w ₇ ⁻	2100.351 (2100.328)	$\{[a_8-G_8]+1'\}^{2-}$	1287.193 (1287.195)
$[w_{10} + 1']^{2-}$	1715.769 (1715.758)	$\{[a_{10} - C_{10}] + \mathbf{1'}\}^{2-}$	1603.748 (1603.750)
$[w_{10} + \mathbf{1'}_2]^{2-}$	1912.795 (1912.781)	$\{[a_{10} - C_{10}] + \mathbf{1'}_2\}^{2-}$	1800.775 (1800.773)
$[w_{12} + 1']^{2-}$	2012.333 (2012.305)	$\{[a_{13} - C_{13}] + \mathbf{1'}\}^{2-}$	2052.330 (2052.333)
$[w_{12} + \mathbf{1'}_2]^{2-}$	2209.858 (2209.828)	$\{[a_{13} - C_{13}] + \mathbf{1'}_2\}^{2-}$	2249.848 (2249.844)
$[w_{14} + 1']^{3-}$	1539.235 (1539.234)	$\{[a_{15} - C_{15}] + \mathbf{1'}_2\}^{3-}$	1697.276 (1697.258)
$\{[w_{14} + 1'] - C^c\}^{3-}$	1502.217 (1502.219)		
$\{[w_{14} + 1'] - G^c\}^{3-}$	1488.894 (1488.883)		
$[w_{14} + \mathbf{1'}_2]^{3-}$	1670.591 (1670.586)		
$\{[w_{14} + \mathbf{1'}_2] - py\}^{3-}$	1644.242 (1644.234)		
$\{[w_{14}+{\bm 1'}_2]-C\}^{3-}$	1633.582 (1633.570)		
$\{[w_{14}+{\bm 1'}_2]-G\}^{3-}$	1620.231 (1620.234)		
$[I + 1'_2]^{3-}$	1740.287 (1740.273)	$\{[\mathbf{I} + \mathbf{1'}_2] - \mathbf{G}\}^{3-}$	1689.934 (1689.930)
$\{[I + 1'_2] - N_3\}^{3-}$	1725.953 (1725.938)	$\{[I + 1'] - N_3\}^{3-}$	1594.590 (1594.586)
${[I + 1'_2] - py}^{3-}$	1713.935 (1713.930)	$\{[I + 1'] - C\}^{3-}$	1571.906 (1571.904)
$\{[I + 1'_2] - C\}^{3-}$	1703.271 (1703.258)	$\{[I + 1'] - G\}^{3-}$	1558.249 (1558.242)
$\{[\mathbf{I} + \mathbf{1'}_2] - 2py\}^{3-}$	1687.599 (1687.578)	${[I + 1'] - py - G}^{3-}$	1532.222 (1532.227)
$\{[I + 1'_2] - C - py\}^{3-}$	1676.917 (1676.914)	${[I + 1'] - G - C}^{3-}$	1521.564 (1521.563)
Internal fragments ^d			
$[T_6:T_9]^-$	1418.181 (1418.188)	$[T_x:T_x]^-$ (x=2,4,9,12,14)	481.035 (481.039)

$\{[T_6:T_9] - G\}^-$	1267.127 (1267.144)	$[T_4:T_7]^{-/}[T_9:T_{12}]^{-/}$ $[T_{11}:T_{14}]^{-}$	1378.177 (1378.180)
$[T_2:T_4]^-/[C_5:T_7]^-/$ $[C_{10}:T_{12}]^-$	1074.127 (1074.133)	${[T_4:T_7]/[T_9:T_{12}]/[T_{11}:T_{14}] + 1'}^-$	1773.250 (1773.234)
${[T_2:T_4]/[C_5:T_7]/[C_{10}:T_{12}] + 1'}^-$	1468.187 (1468.188)	$[T_6:T_7]^-/[T_{11}:T_{12}]^-$	785.084 (785.086)
$[T_2:T_7]^{2-}/[T_9:T_{14}]^{2-}$	985.129 (985.133)	$\{[T_2:T_{14}] + \mathbf{1'}_2\}^{3-}$	1627.570 (1627.563)
$[T_2:T_7]^-/[T_9:T_{14}]^-$	1971.284 (1971.273)	$[T_6:T_{14}]^{2-}$	1454.178 (1454.203)
${[T_2:T_7]/[T_9:T_{14}] + 1'}^{2-}$	1182.642 (1182.656)	$\{[T_4:T_9] - G\}^-$	1860.234 (1860.227)
${[T_2:T_9]/[C_5:T_{12}] + 1'}^{2-}$	1499.222 (1499.211)	${[T_2:T_{12}]/[T_4:T_{14}] + 1'_2}^{2-}$	2145.318 (2145.305)

^aThe most abundant isotopic mass-to-charge ratio.

 ${}^{b}T_{n}$, C_{n} and G_{n} represent the loss of a cytosine and a guanine base, respectively, followed by elimination of a H₂O molecule to form a furan ring, n indicates the position of the base.

°C and G represent the neutral loss of a cytosine and a guanine base, respectively.

^dThe internal fragment $B_m:B_n$ results from fragmentation at both the a- and w-sites, having a phosphate group at the 5'-terminus and a furan ring at the 3'-terminus.

Table S4. Fragment ions observed by MS/MS analysis in negative-ion mode of triplatinated I ($[I + 1'_3]^{3-}$, *m/z* 1857.297) produced by the reaction of complex 1 with ODN I at 310 K after irradiation under blue light for 1 h. (Charges for Pt moiety and the loss of protons from I for balancing the charges of the ions are omitted for clarity). $1' = [Pt(N_3)(py)_2]^+$.

Fragments	m/z^{a} observed	Encomenta	m/z^{a} observed
	(calculated)	riagineius	(calculated)
w ₁ ⁻	306.042 (306.047)	$[a_2 - T_2^b]^-$	386.073 (386.078)
W_2^-	610.087 (610.094)	$[a_3 - C_3^b]^-$	690.111 (690.125)
W3 ⁻	899.128 (899.141)	$\{[a_3 - C_3] + 1'\}^-$	1084.160 (1084.172)
W_4^-	1203.192 (1203.188)	$[a_4 - T_4]^-$	979.162 (979.168)
W5 ⁻	1507.214 (1507.234)	$[a_5 - C_5]^-$	1283.203 (1283.211)
$[w_5 + 1']^-$	1902.280 (1902.289)	$\{[a_5 - C_5] + 1'\}^-$	1678.253 (1678.266)
W6 ²⁻	897.628 (897.633)	$[a_8 - G_8{}^b]^{2-}$	1089.658 (1089.672)
w_7^{2-}	1049.646 (1049.656)	$\{[a_8 - G_8] + \mathbf{1'}\}^{2-}$	1287.188 (1287.195)
$[w_7 + 1']^{2-}$	1247.177 (1247.188)	$\{[a_{10} - C_{10}] + \mathbf{1'}\}^{2-}$	1603.749 (1603.750)
$[w_7 + 1'_2]^{2-}$	1444.203 (1444.211)	$\{[a_{10} - C_{10}] + \mathbf{1'}_2\}^{2-}$	1800.807 (1800.773)
$[w_{10} + \mathbf{1'}_2]^{2-}$	1912.795 (1912.781)	$\{[a_{13} - C_{13}] + \mathbf{1'}_2\}^{2-}$	2249.862 (2249.844)
$[w_{10} + \mathbf{1'}_3]^{2-}$	2110.333 (2110.313)	$\{[a_{13} - C_{13}] + \mathbf{1'}_3\}^{2-}$	2446.894 (2446.867)
$[w_{12} + \mathbf{1'}_2]^{2-}$	2209.866 (2209.828)	$\{[a_{15}-C_{15}]+1'_3-py\}^{3-1}$	1802.272 (1802.258)
$[w_{12} + \mathbf{1'}_3]^{2-}$	2406.894 (2406.859)		
$[w_{14} + \mathbf{1'}_2]^{3-}$	1670.575 (1670.585)		
$\{[w_{14} + \mathbf{1'}_2] - py\}^{3-}$	1644.210 (1644.234)		
$[w_{14} + \mathbf{1'}_3]^{3-}$	1801.932 (1801.930)		
$\{[w_{14}+{\bf 1'}_3]-py\}^{3-}$	1775.583 (1775.586)		
	1051 (05 (1051 (05)		1705 000 (1705 000)
$[\mathbf{I} + \mathbf{I'}_3]^{3-1}$	18/1.635 (18/1.625)	$\{[\mathbf{I} + \mathbf{I'}_2] - \mathbf{N}_3\}^{3-1}$	1725.928 (1725.938)
$\{[\mathbf{I} + \mathbf{1'}_3] - N_3\}^{3-1}$	1857.299 (1857.289)	$\{[\mathbf{I} + \mathbf{1'}_2] - \mathbf{C}^c\}^{3-1}$	1703.255 (1703.258)
$\{[\mathbf{I} + \mathbf{1'_3}] - \mathbf{py}\}^{3-}$	1845.281 (1845.281)	${[I + 1'_2] - G^c}^{3-}$	1689.926 (1689.930)
${[I + 1'_3] - C}^{3-}$	1834.620 (1834.609)	${[I + 1'_2] - G - py}^{3-}$	1663.567 (1663.578)
$\{[\mathbf{I} + \mathbf{1'}_3] - \mathbf{G}\}^{3-}$	1821.273 (1821.273)	$\{[I + 1'_2] - G - C\}^{3-}$	1652.903 (1652.914)
$\{[I + 1'_3] - 2py\}^{3-}$	1818.930 (1818.928)		
$\{[I + 1'_3] - C -$	1808.267 (1808.266)		
py} ³⁻			
	Internal	fragments ^d	
$[T_2:T_4]^-/[C_5:T_7]^-/$ $[C_{10}:T_{12}]^-$	1074.124 (1074.133)	$[T_x:T_x]^-$ (x=2,4,9,12,14)	481.035 (481.039)
$[T_2:T_7]^{2-}/[T_9:T_{14}]^{2-}$	985.126 (985.133)	[T ₄ :T ₇] ⁻ /[T ₉ :T ₁₂] ⁻ /	1378.171 (1378.180)

		$[T_{11}:T_{14}]^{-}$	
${[[T_2:T_7]^{2-}/[T_9:T_{14}]^{2-} + 1'}$	1182.652 (1182.656)	${[T_4:T_7]^-/[T_9:T_{12}]^-/[T_{11}:T_{14}]^- + 1'}$	1773.216 (1773.234)
$[T_6:T_7]^-/[T_{11}:T_{12}]^-$	785.077 (785.086)		

^aThe most abundant isotopic mass-to-charge ratio.

 ${}^{b}T_{n}$, C_{n} and G_{n} represent the loss of a cytosine and a guanine base, respectively, followed by elimination of a H₂O molecule to form a furan ring, n indicates the position of the base.

°C and G represent the neutral loss of a cytosine and a guanine base, respectively.

^dThe internal fragment $B_m:B_n$ results from fragmentation at both the a- and w-sites, having a phosphate group at the 5'-terminus and a furan ring at the 3'-terminus.



Figure S1. The isotopic models (dots) and mass spectra (lines) for mono-, di-, tri-, tetra- and penta-platinated ODN I by complex 1 from their reaction mixture after irradiation under blue light for 1 h.



Figure S2. The isotopic models (dots) and mass spectra (lines) for the adducts at *m/z* 1614.594, 1616.261, 1731.938, 1736.608, 1757.617, 1862.957 and 1888.973.



Figure S3. The isotopic models (dots) and mass spectra (lines) for the fragments of $[w_5 + 1']^-$ and $\{[a_3 - C_3] + 1'\}^-$ observed in the CID fragmentation of $\{I + 1'\}^+$ (A, B), $\{I + 1'_2\}^+$ (C, D) and $\{I + 1'_3\}^+$ (E, F), respectively. $1' = [Pt(N_3)(py)_2]^+$.



Figure S4. Tandem mass spectra of parent ion $[I + 1'_3]^+$ with the parent ion shown in red (A) and the enlarged spectrum during the MS range of m/z 1590-1860 (B). The corresponding MS/MS data in detail are listed in Tables S4. $1' = [Pt(N_3)(py)_2]^+$.