## Structural Diversity of Lead Halide Chain Compounds, APbX<sub>3</sub>, Templated by Isomeric Molecular Cations

## Yuan-Yuan Guo and Philip Lightfoot\*

School of Chemistry and EaStChem, University of St Andrews, St Andrews, KY16 9ST, UK; E-Mail: <u>pl@st-andrews.ac.uk</u> (PL)

## **Supplementary Information**

Figure S1. The full-range and expanded P-XRD data for IQPbBr<sub>3</sub> at room temperature.

Figure S2. The full-range and expanded P-XRD data for QPbBr<sub>3</sub> at room temperature.

Figure S3. The full-range and expanded P-XRD data for QPbI<sub>3</sub> at room temperature.

Figure S4. The nature of the distortions within the inorganic PbX<sub>6</sub> octahedra of the three compounds.

Figure S5. Thermal Gravimetric Analysis (TGA) of IQPbBr<sub>3</sub>, QPbBr<sub>3</sub> and QPbI<sub>3</sub>.

Table S1. Crystal and Structure Refinement Data for the three compounds at 173 K.

Table S2. Selected bond lengths (Å) and bond angles (°) for IQPbBr<sub>3</sub> at 173 K and 298 K.

Table S3. Selected bond lengths (Å) and bond angles (°) for QPbBr<sub>3</sub> at 173 K and 298 K.

Table S4. Selected bond lengths (Å) and bond angles (°) for QPbI<sub>3</sub> at 173 K and 298 K.

Table S5. Hydrogen bond lengths (Å) and angles (°) for IQPbBr<sub>3</sub>, QPbBr<sub>3</sub> and QPbI<sub>3</sub> at 173 K.



**Figure S1.** The full-range (a) and expanded (b) P-XRD data for IQPbBr<sub>3</sub> at room temperature, the corresponding colours represent the experimental (black) and the calculated (red) pattens.



**Figure S2.** The full-range (a) and expanded (b) P-XRD data for QPbBr<sub>3</sub> at room temperature, the corresponding colours represent the experimental (black) and the calculated (red) pattens.



**Figure S3.** The full-range (a) and expanded (b) P-XRD data for  $QPbI_3$  at room temperature, the corresponding colours represent the experimental (black) and the calculated (red) pattens.



**Figure S4.** The nature of the distortions within the inorganic PbX<sub>6</sub> octahedra of the three compounds at 298 K. (a) Pb(1)Br<sub>6</sub> octahedra for IQPbBr<sub>3</sub>, (b) Pb(2)Br<sub>6</sub> octahedra for IQPbBr<sub>3</sub>, (c) PbBr<sub>6</sub> octahedra for QPbBr<sub>3</sub>, (d) PbI<sub>6</sub> octahedra for QPbI<sub>3</sub>.



Figure S5. Thermal Gravimetric Analysis (TGA) of IQPbBr<sub>3</sub>, QPbBr<sub>3</sub> and QPbI<sub>3</sub>.

	IQPbBr <sub>3</sub>	QPbBr <sub>3</sub>	QPbI <sub>3</sub>	
Formula	C <sub>9</sub> H <sub>8</sub> NPbBr <sub>3</sub>		C <sub>9</sub> H <sub>8</sub> NPbI <sub>3</sub>	
Formula weight	577.08	577.08	718.05	
Colour/Habit	Colorless/Chip	Colorless/Prism	Yellow/Prism	
Crystal size (mm <sup>3</sup> )	$0.32 \times 0.24 \times 0.09$	$0.25 \times 0.06 \times 0.04$	$0.32 \times 0.09 \times 0.05$	
Crystal system	Orthorhombic	Monoclinic	Monoclinic	
Space group	Pbca	$P2_{1}/n$	C2/c	
<i>a</i> (Å)	21.1904(15)	4.3397(2)	25.767(2)	
b (Å)	7.7771(5)	22.5309(17)	4.6411(4)	
c (Å)	30.828(2)	13.0048(10)	25.931(2)	
α (°)	90	90	90	
β (°)	90	98.941(6)	115.396(7)	
γ (°)	90	90	90	
V(Å <sup>3</sup> )	5080.5(6)	1256.12(15)	2801.3(4)	
Ζ	16	4	8	
$ ho_{ m calc}$ (g/cm <sup>3</sup> )	3.018	3.052	3.405	
μ (mm <sup>-1</sup> )	22.684	22.937	18.625	
F(000)	4096	1024	2480	
Refins collected	47986	9972	12590	
La dana an dana tan <b>G</b> ua	5808	2192	3132	
Independent refins	0.099	0.1118	0.086	
Goodness of Fit	1.072	0.944	0.909	
	$R_1 = 0.0397$	$R_1 = 0.0349$	$R_1 = 0.0400$	
Final K indices $(I \ge 2\sigma(I))$	$wR_2 = 0.0966$	$wR_2 = 0.0768$	$wR_2 = 0.0954$	
Largest diff. peak/hole (e Å <sup>-3</sup> )	2.408/-2.586	1.955/-1.922	2.360/-1.679	

Table S1. Single crystal and structure refinement data for IQPbBr<sub>3</sub>, QPbBr<sub>3</sub> and QPbI<sub>3</sub> at 173 K.

Temperature	Temperature173 K29		
Pb(1)-Br(1)	2.9955(7)	3.0034(11)	
Pb(1)-Br(1)#1	3.0752(8)	3.0763(11)	
Pb(1)-Br(2)	3.2403(8)	3.2299(12)	
Pb(1)-Br(2)#1	2.8753(8)	2.8846(12)	
Pb(1)-Br(3)	2.9449(8)	2.9535(11)	
Pb(1)-Br(3)#1	3.0364(8)	3.0398(11)	
Pb(2)-Br(4)	3.0923(8)	3.0837(11)	
Pb(2)-Br(4)#2	2.9921(8)	3.0088(11)	
Pb(2)-Br(5)	2.8328(8)	2.8425(11)	
Pb(2)-Br(5)#2	3.3111(8)	3.2973(12)	
Pb(2)-Br(6)	3.0377(8)	3.0295(11)	
Pb(2)-Br(6)#2	2.9515(8)	2.9704(11)	
Br(2)#1-Pb(1)-Br(3)	95.50(2)	95.80(3)	
Br(2)#1-Pb(1)-Br(1)	93.95(2)	94.23(3)	
Br(3)-Pb(1)-Br(1)	88.38(2)	88.14(3)	
Br(2)#1-Pb(1)-Br(3)#1	84.87(2)	84.86(3)	
Br(3)-Pb(1)-Br(3)#1	94.52(3)	94.55(4)	
Br(2)#1-Pb(1)-Br(1)#1	82.36(2)	82.49(3)	
Br(1)-Pb(1)-Br(1)#1	91.78(2)	92.00(3)	
Br(3)#1-Pb(1)-Br(1)#1	85.30(2)	85.29(3)	
Br(3)-Pb(1)-Br(2)	80.21(2)	80.45(3)	
Br(1)-Pb(1)-Br(2)	77.81(2)	78.14(3)	
Br(3)#1-Pb(1)-Br(2)	103.56(2)	102.93(3)	
Br(1)#1-Pb(1)-Br(2)	101.91(2)	101.24(3)	
Br(5)-Pb(2)-Br(6)#2	92.21(2)	93.52(3)	
Br(5)-Pb(2)-Br(4)#2	94.04(2)	95.05(3)	
Br(6)#2-Pb(2)-Br(4)#2	88.90(2)	88.23(3)	
Br(5)-Pb(2)-Br(6)	85.20(2)	85.37(3)	
Br(6)#2-Pb(2)-Br(6)	94.89(3)	95.01(4)	
Br(5)-Pb(2)-Br(4)	84.15(2)	84.34(3)	
Br(4)#2-Pb(2)-Br(4)	90.64(2)	90.97(3)	
Br(6)-Pb(2)-Br(4)	85.53(2)	85.81(3)	
Br(6)#2-Pb(2)-Br(5)#2	78.60(2)	78.73(3)	
Br(4)#2-Pb(2)-Br(5)#2	78.02(2)	78.18(3)	
Br(6)-Pb(2)-Br(5)#2	103.33(2)	101.83(3)	
Br(4)-Pb(2)-Br(5)#2	103.90(2)	103.28(3)	
Pb(1)-Br(1)-Pb(1)#2	79.66(2)	79.53(3)	
Pb(2)-Br(4)-Pb(2)#1	79.43(2)	79.33(3)	
Pb(1)#2-Br(2)-Pb(1)	78.72(2)	78.77(3)	
Pb(2)-Br(5)-Pb(2)#1	78.11(2)	78.22(3)	
Pb(2)#1-Br(6)-Pb(2)	80.96(2)	80.81(3)	
Pb(1)-Br(3)-Pb(1)#2	81.09(2)	80.91(3)	

**Table S2.** Selected bond lengths (Å) and bond angles (°) for IQPbBr<sub>3</sub> at 173 K and 298 K.

#1 -x+1/2,y+1/2,z #2 -x+1/2,y-1/2,z

Temperature	173 K	298 K
Pb(1)-Br(1)	2.8241(9)	2.8247(6)
Pb(1)-Br(2)#1	2.9985(9)	3.0046(6)
Pb(1)-Br(2)	3.0201(9)	3.0267(6)
Pb(1)-Br(3)	3.0317(9)	3.0352(6)
Pb(1)-Br(3)#1	3.2822(9)	3.2890(6)
Pb(1)-Br(3)#2	3.0713(9)	3.0766(6)
Br(1)-Pb(1)-Br(2)#1	90.42(3)	90.37(2)
Br(1)-Pb(1)-Br(2)	88.91(3)	89.17(2)
Br(2)-Pb(1)-Br(3)	87.04(2)	86.43(2)
Br(2)#1-Pb(1)-Br(2)	92.28(2)	92.68(2)
Br(1)-Pb(1)-Br(3)	91.97(3)	92.27(2)
Br(2)#1-Pb(1)-Br(3)	89.10(2)	88.67(2)
Br(1)-Pb(1)-Br(3)#2	90.47(3)	91.05(2)
Br(2)#2-Pb(1)-Br(3)#1	92.06(3)	92.20(2)
Br(2)-Pb(1)-Br(3)#2	87.97(2)	87.51(2)
Br(3)-Pb(1)-Br(3)#1	92.07(2)	91.38(2)
Br(3)#1-Pb(1)-Br(3)#2	87.12(2)	87.19(2)
Br(3)-Pb(1)-Br(3)#2	90.64(2)	91.11(2)
Pb(1)#2-Br(2)-Pb(1)	92.28(2)	92.68(2)
Pb(1)#2-Br(3)-Pb(1)#1	92.88(2)	93.57(2)
Pb(1)#2-Br(3)-Pb(1)	87.93(2)	88.62(2)
Pb(1)-Br(3)-Pb(1)#1	90.64(2)	91.11(2)

Table S3. Selected bond lengths (Å) and bond angles (°) for QPbBr<sub>3</sub> at 173 K and 298 K.

#1 x-1,y,z #2 x+1,y,z

Temperature	173 K	298 K
Pb(1)-I(1)	3.0001(7)	2.9956(6)
Pb(1)-I(2)	3.2051(6)	3.2092(5)
Pb(1)-I(2)#1	3.2945(6)	3.2996(5)
Pb(1)-I(3)	3.2174(6)	3.2207(5)
Pb(1)-I(3)#1	3.2732(6)	3.2749(5)
Pb(1)-I(3)#2	3.5061(6)	3.5212(5)
I(1)-Pb(1)-I(2)	90.87(2)	91.19(1)
I(1)-Pb(1)-I(3)	91.76(2)	92.40(1)
I(2)-Pb(1)-I(3)	89.92(2)	89.67(1)
I(1)-Pb(1)-I(3)#1	92.89(2)	93.08(1)
I(3)-Pb(1)-I(3)#1	91.29(2)	91.53(1)
I(1)-Pb(1)-I(2)#1	92.00(2)	91.88(1)
I(2)-Pb(1)-I(2)#1	91.12(2)	91.28(1)
I(3)#1-Pb(1)-I(2)#1	87.42(2)	87.20(1)
I(2)-Pb(1)-I(3)#2	89.16(2)	89.13(1)
I(3)-Pb(1)-I(3)#2	92.04(2)	91.71(1)
I(3)#1-Pb(1)-I(3)#2	87.19(2)	86.69(1)
I(2)#1-Pb(1)-I(3)#2	84.21(2)	84.04(1)
Pb(1)-I(2)-Pb(1)#2	91.12(2)	91.28(1)
Pb(1)-I(3)-Pb(1)#2	91.29(2)	91.53(1)
Pb(1)-I(3)-Pb(1)#3	92.81(2)	93.31(1)
Pb(1)#2-I(3)-Pb(1)#3	87.96(2)	88.29(1)

Table S4. Selected bond lengths (Å) and bond angles (°) for QPbI<sub>3</sub> at 173 K and 298 K.

#1 x,y+1,z #2 x,y-1,z #3 x-1/2,-y+1/2,-z+1

APbX <sub>3</sub>	D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
	N(1)-H(2)Br(1)	0.86	3.13	3.537(8)	111.8
	N(1)-H(2)Br(1)#1	0.86	3.08	3.694(9)	129.9
IQPbBr <sub>3</sub>	N(1)-H(2)Br(6)#3	0.86	2.98	3.613(8)	132.3
	N(2)-H(10)Br(3)	0.86	3.10	3.413(8)	104.4
	N(2)-H(10)Br(4)	0.86	2.72	3.462(9)	145.5
QPbBr <sub>3</sub>	N(1)-H(1)Br(1)	0.86	2.50	3.324(8)	161.9
QPbI <sub>3</sub>	N(1)-H(1)I(1)	0.86	2.93	3.604(8)	136.5
	N(1)-H(1)I(1)#3	0.86	3.24	3.829(8)	128.0

Table S5. Hydrogen bond lengths (Å) and angles (°) for IQPbBr<sub>3</sub>, QPbBr<sub>3</sub> and QPbI<sub>3</sub> at 173 K.

#1 -x+1/2,y+1/2,z #3 x,-y+1/2,z+1/2 (**IQPbBr**<sub>3</sub>); #3 -x

#3 -x+1,y,-z+3/2 (**QPbI**<sub>3</sub>)