

Electronic Supplementary Information

For

Synthesis, Characterization, and Alkoxide Transfer Reactivity of Dimeric $Tl_2(OR)_2$ Complexes

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1. NMR Spectra

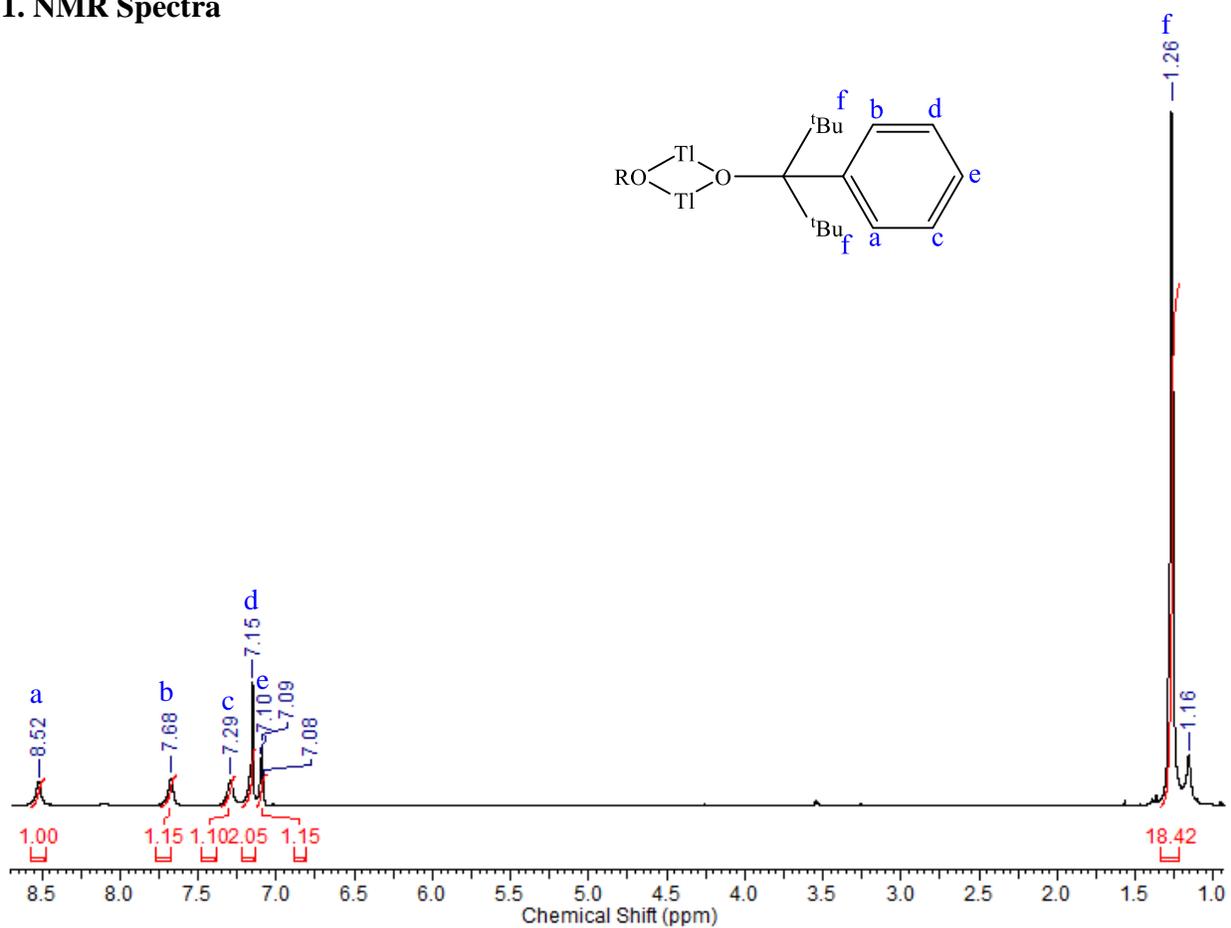


Figure S1. ^1H NMR of $\text{Tl}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ (C_6D_6 , 600 MHz, room temperature).

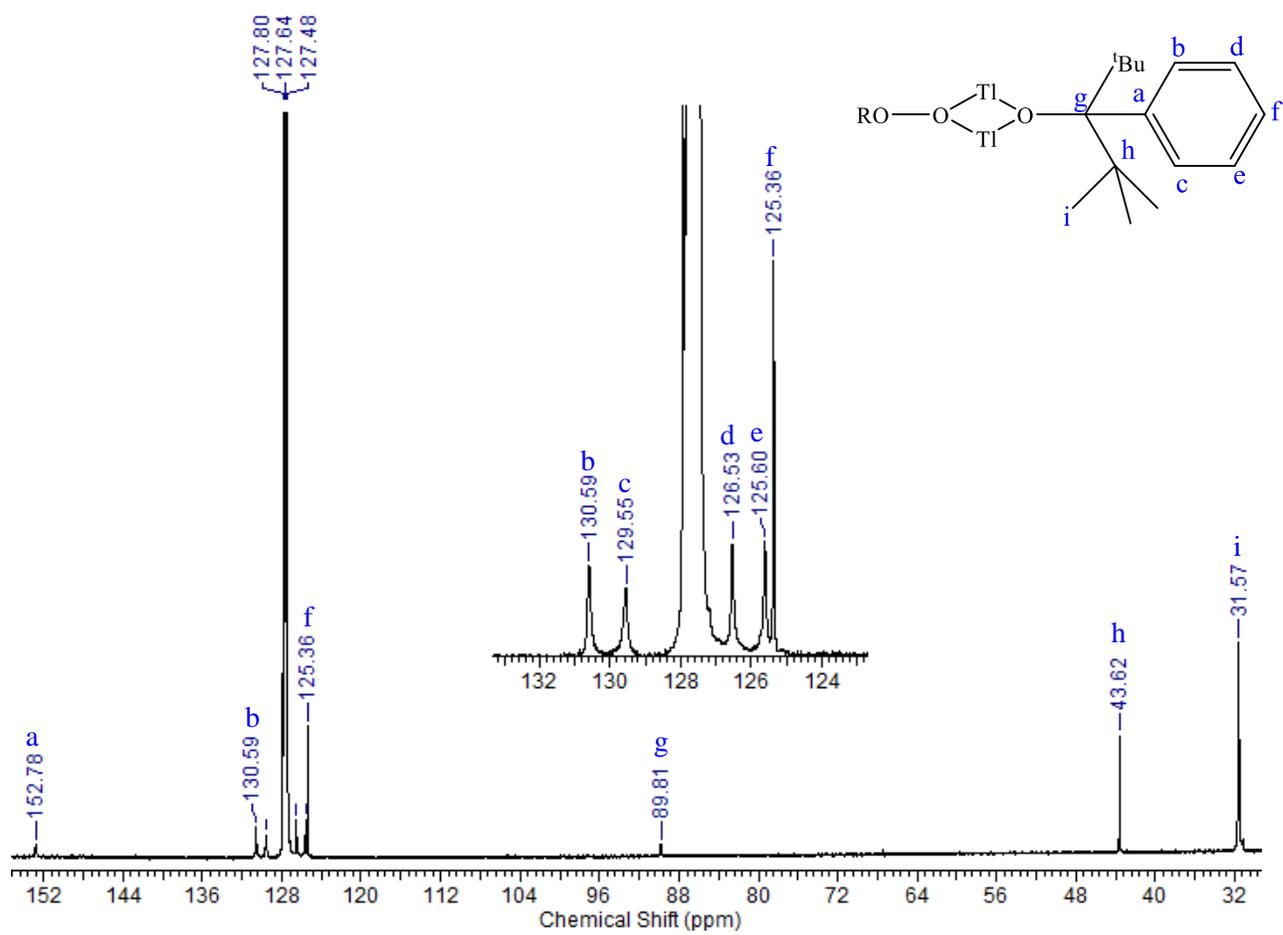


Figure S2. ^{13}C NMR of $\text{Ti}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ (C_6D_6 , 150 MHz).

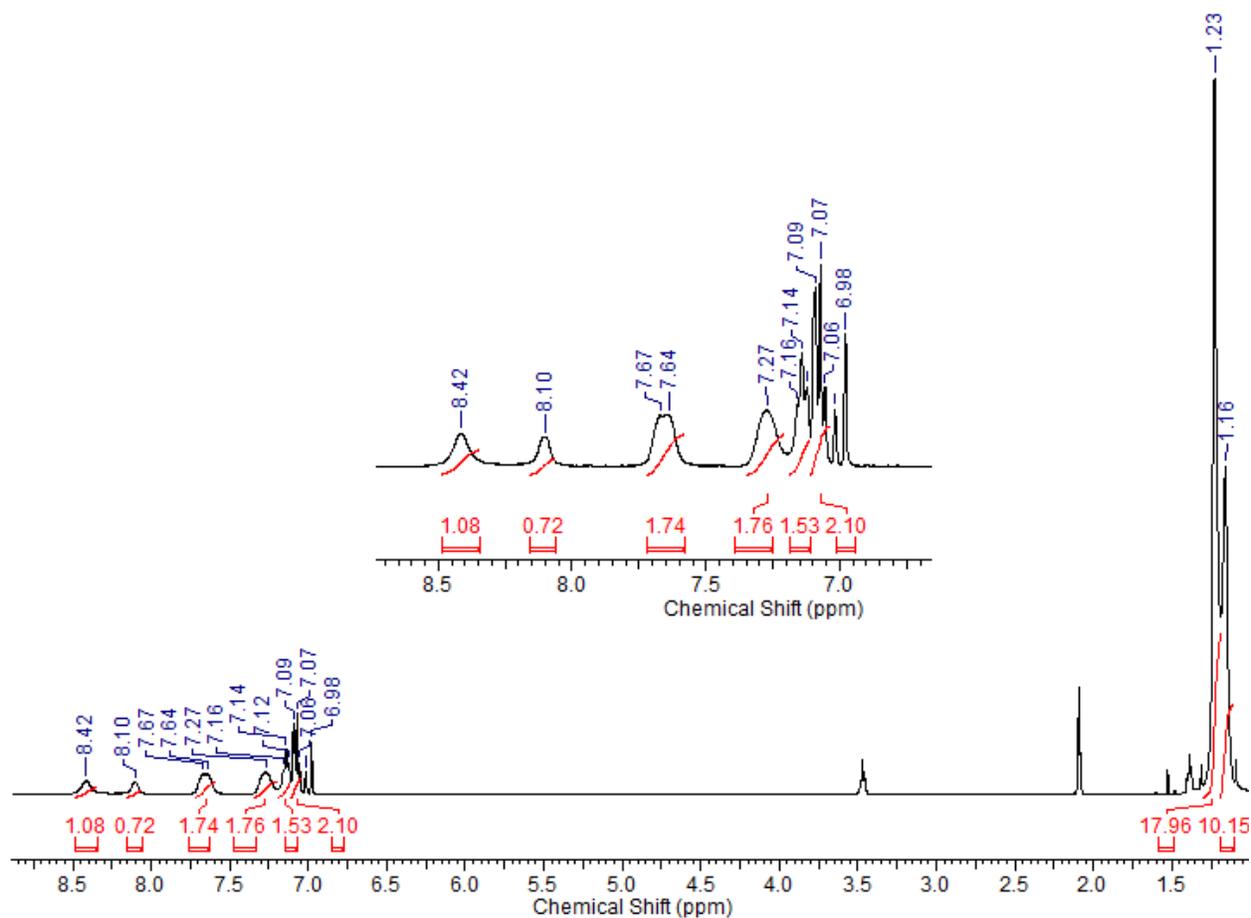


Figure S3. ¹H NMR of $\text{Ti}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ (C_7D_8 , 400 MHz, room temperature).

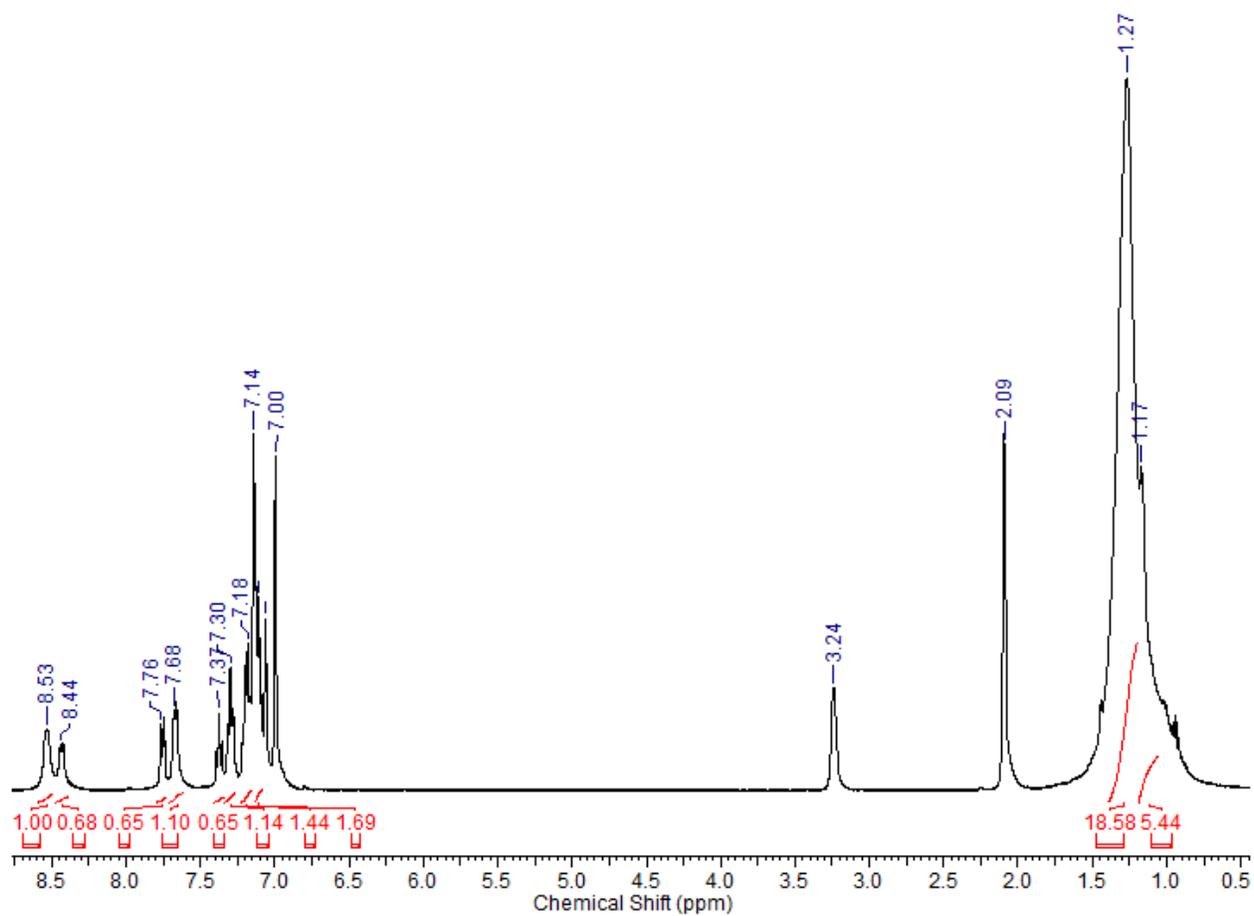


Figure S4. ^1H NMR of $\text{Tl}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ (C_7D_8 , 400 MHz, $-40\text{ }^\circ\text{C}$).

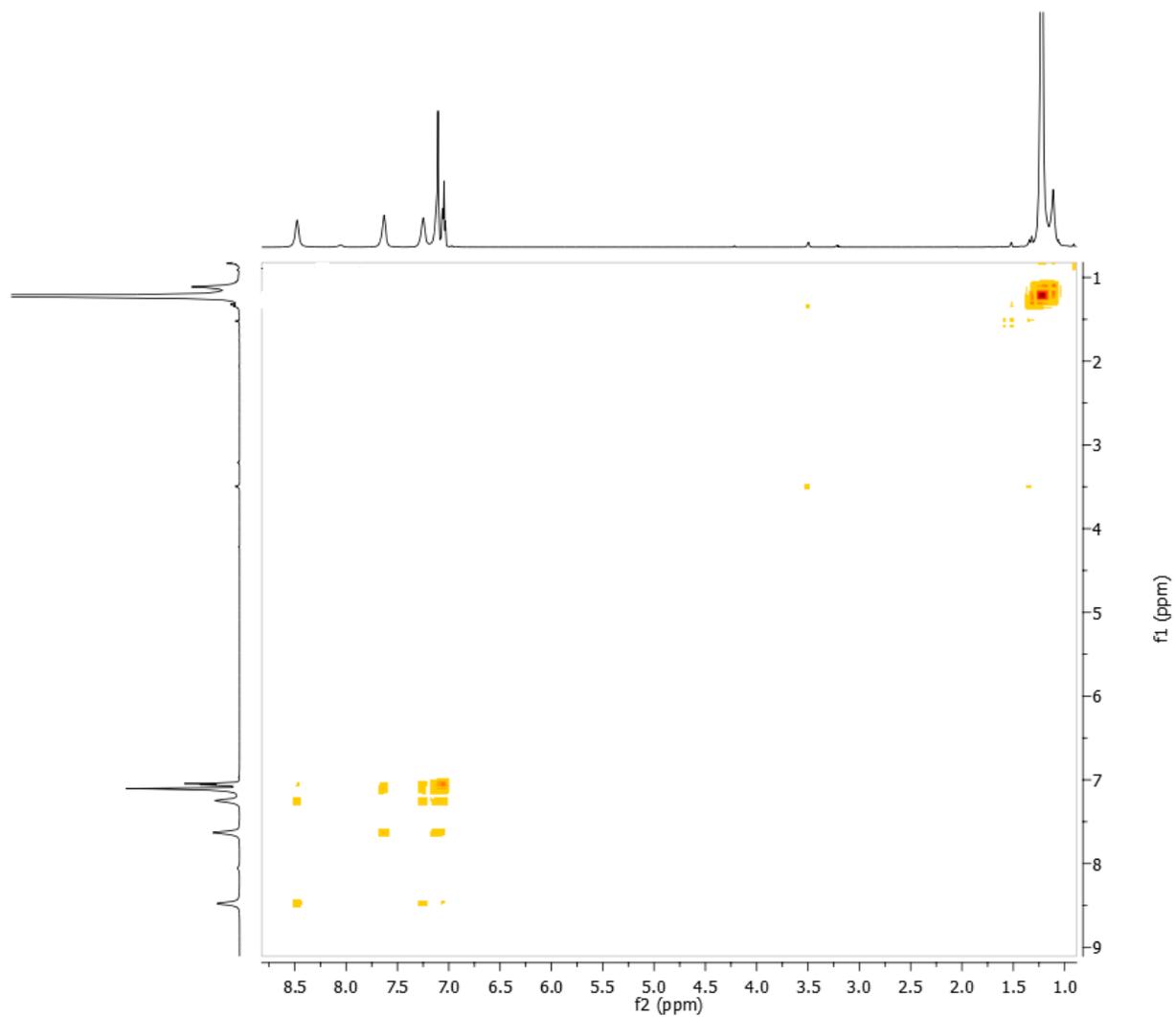


Figure S5. ¹H-¹H COSY NMR of $\text{Ti}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ (C_6D_6 , 600 MHz).

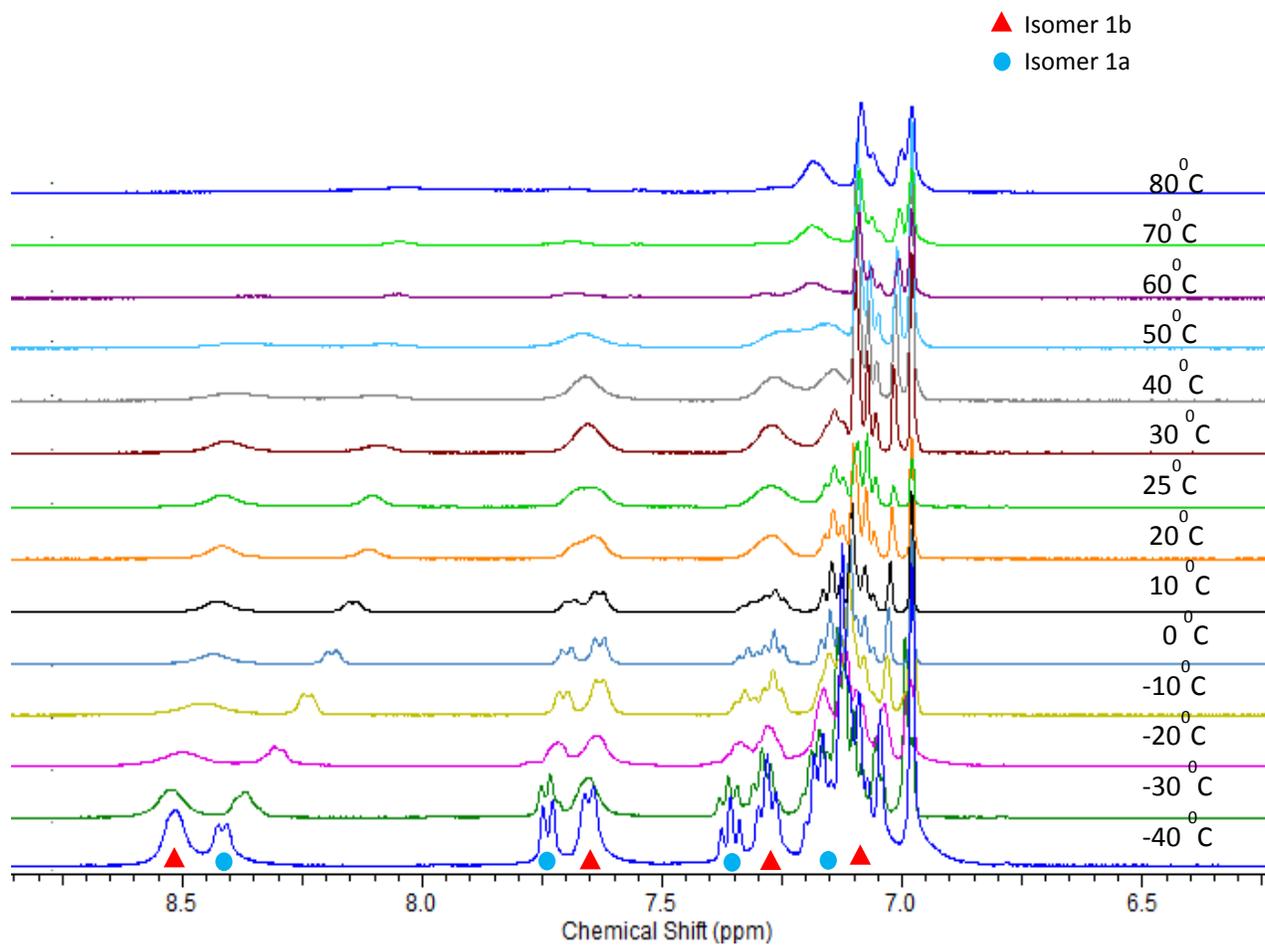


Figure S6. VT ¹H NMR of $\text{Ti}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ (C_7D_8 , 400 MHz).

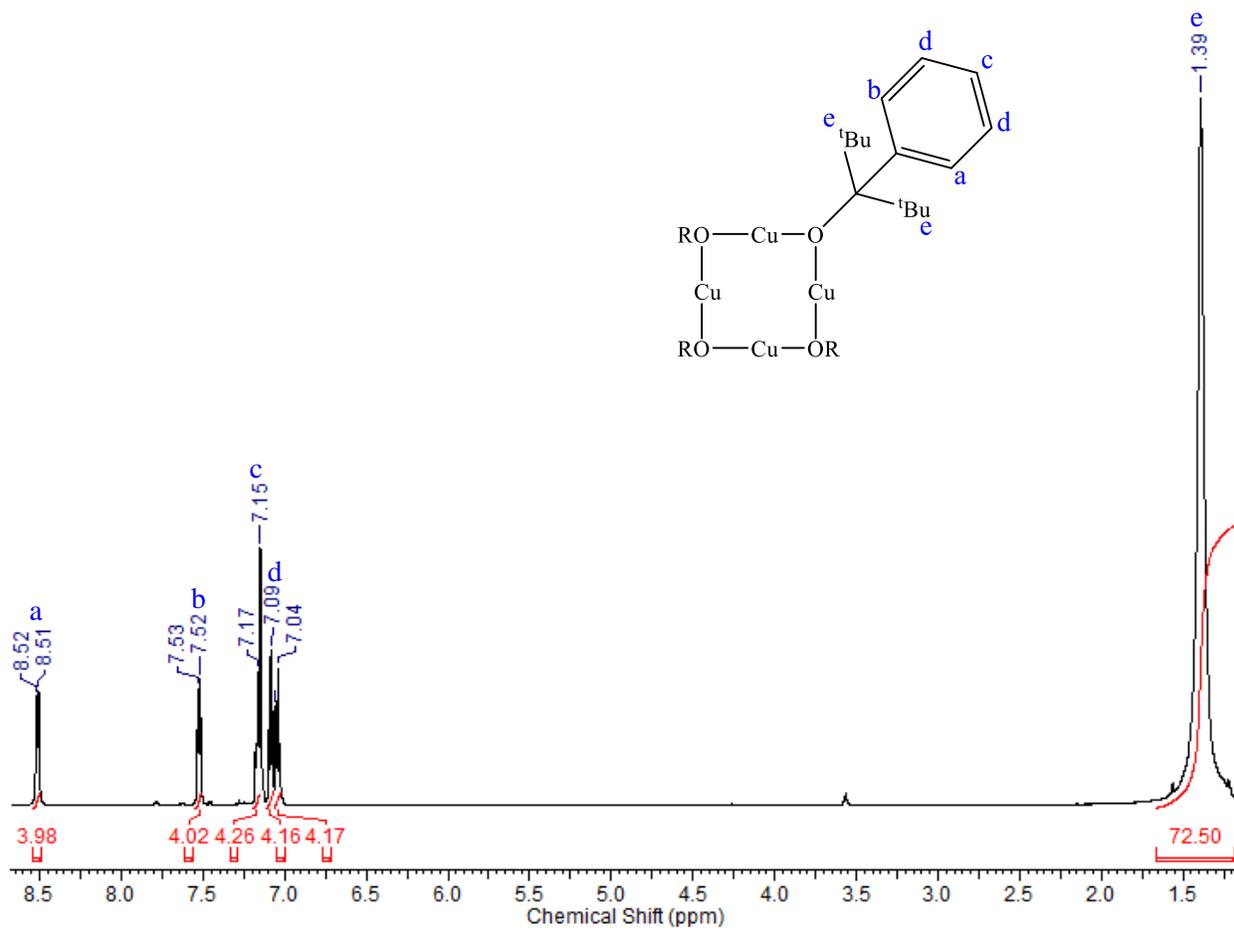


Figure S7. ^1H NMR of $\text{Cu}_4(\text{OC}^t\text{Bu}_2\text{Ph})_4$ (C_6D_6 , 600 MHz).

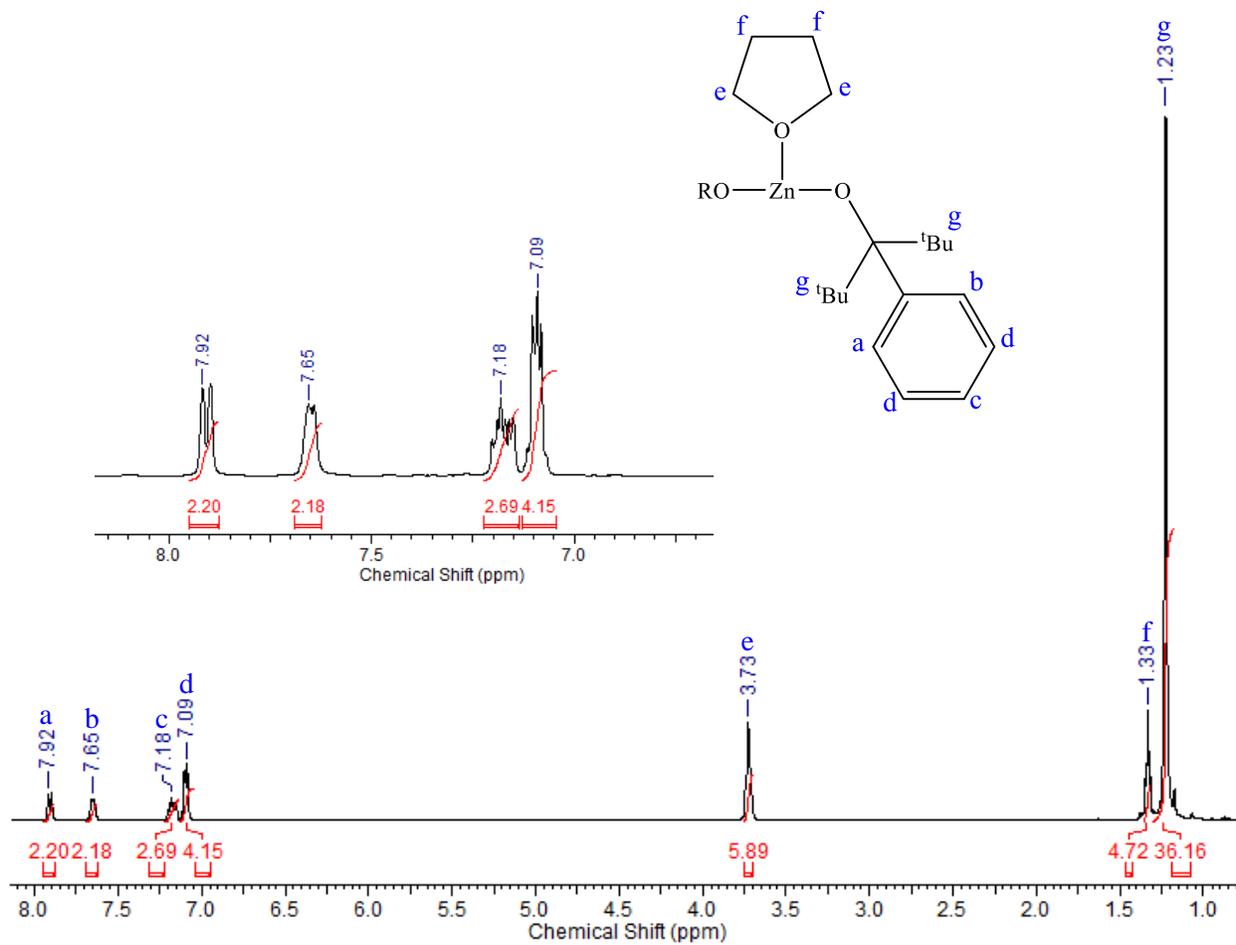


Figure S8. ^1H NMR of $\text{Zn}(\text{OC}^t\text{Bu}_2\text{Ph})_2(\text{THF})$ (C_6D_6 , 400 MHz).

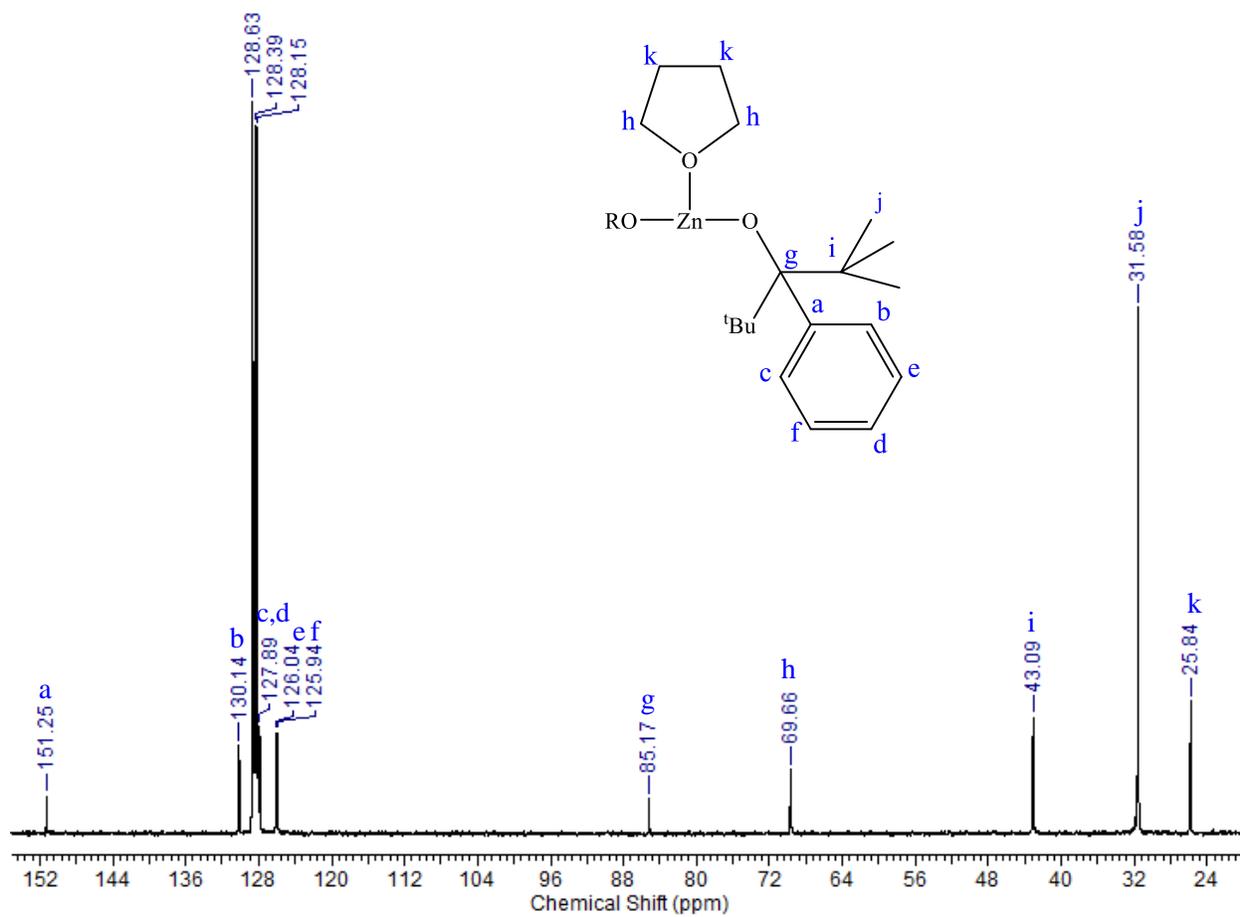


Figure S9. ^{13}C NMR of $\text{Zn}(\text{OR})_2(\text{THF})$ (C_6D_6 , 100 MHz).

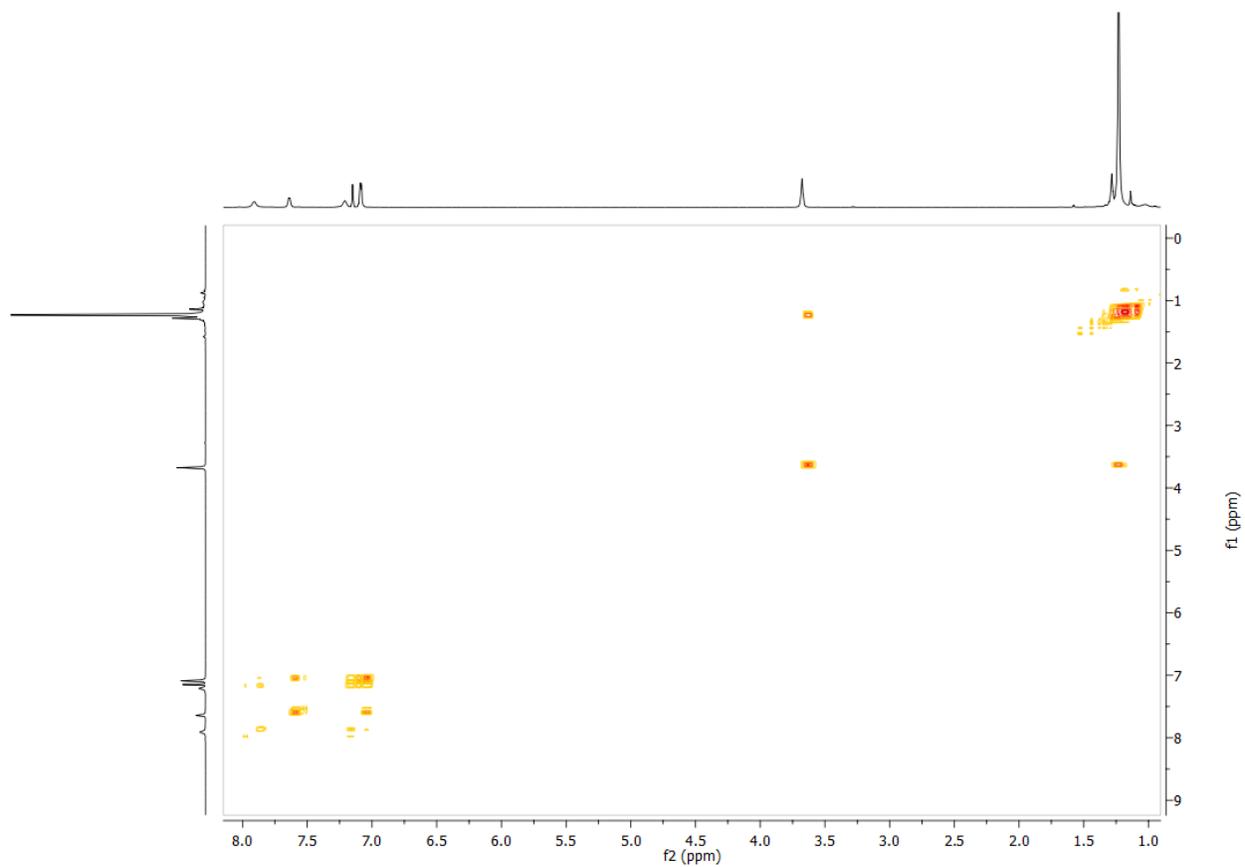


Figure S10. ^1H - ^1H COSY NMR of $\text{Zn}(\text{OR})_2(\text{THF})$ (C_6D_6 , 600 MHz).

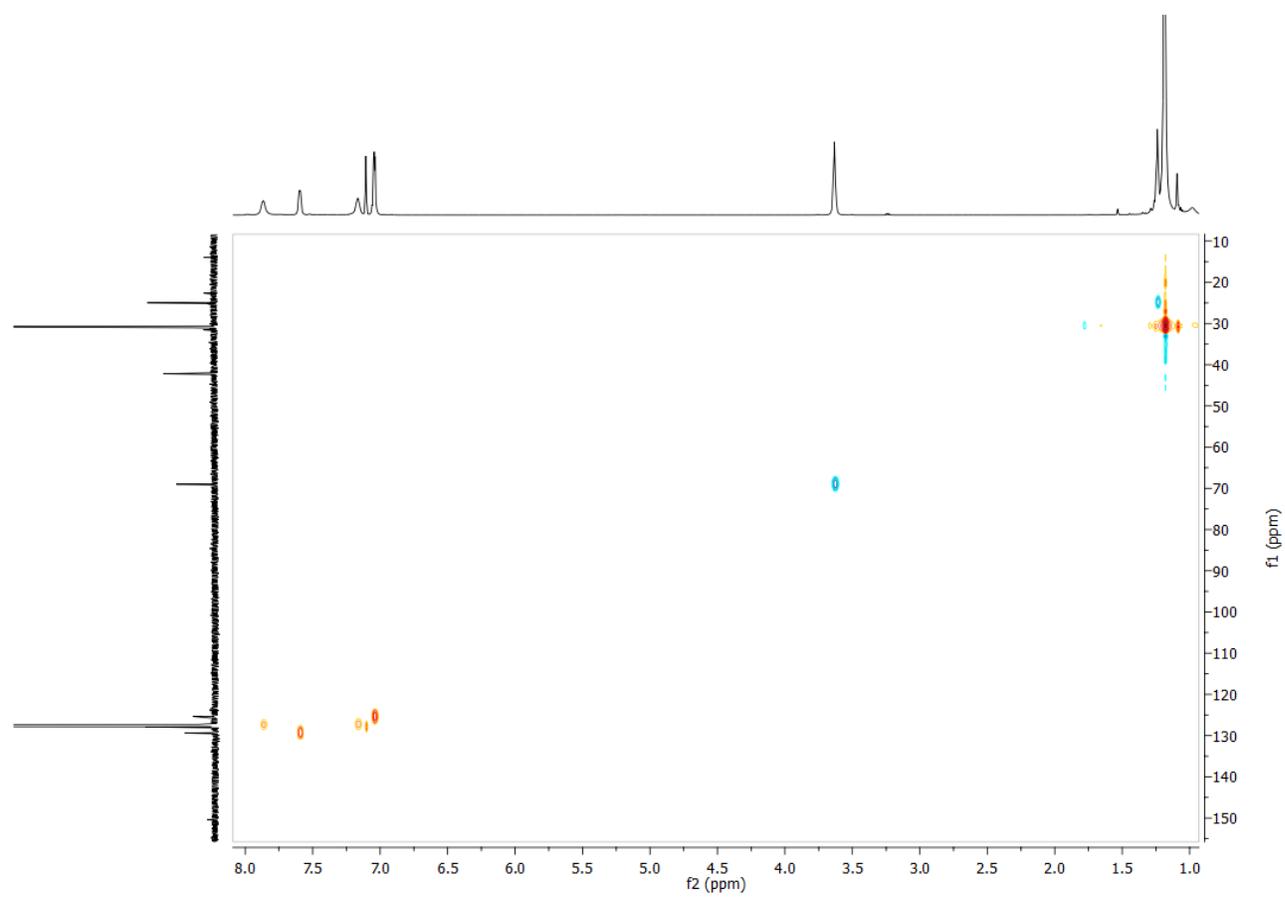


Figure S11. ^1H - ^{13}C HSQCAD NMR of $\text{Zn}(\text{OR})_2(\text{THF})$ (C_6D_6 , 600 MHz).

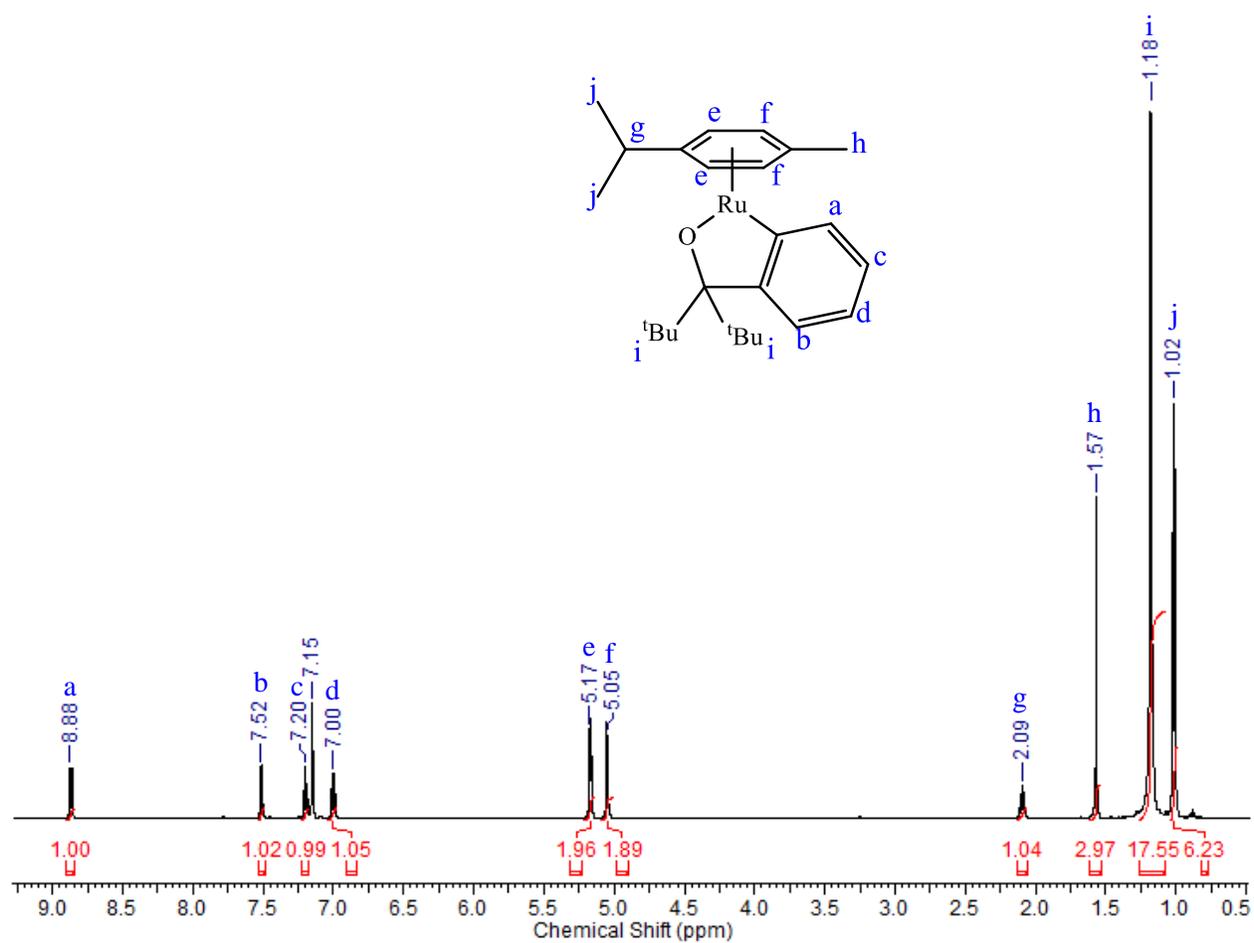


Figure S12. ^1H NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2\text{C}_6\text{H}_2)$ (C_6D_6 , 600 MHz).

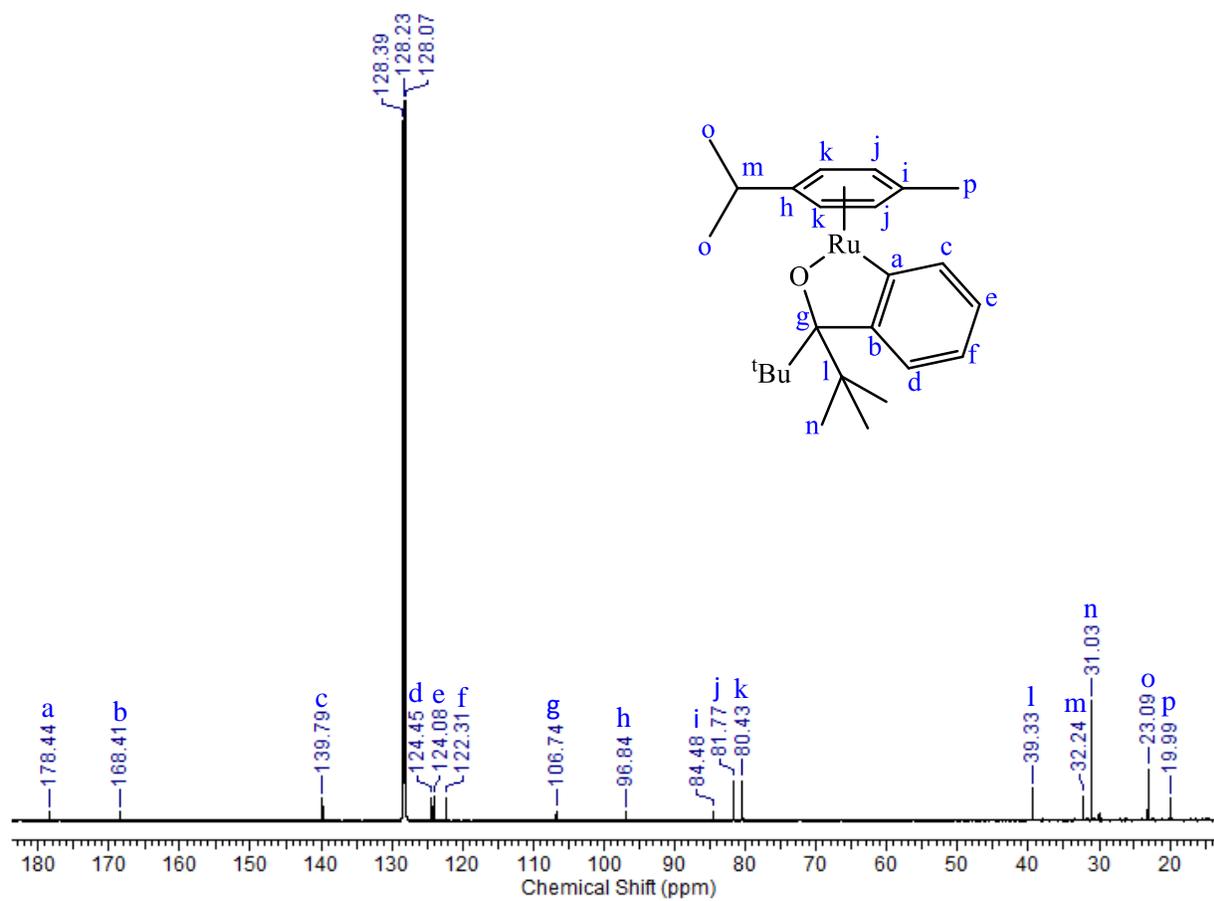


Figure S13. ^{13}C NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2\text{C}_6\text{H}_2)$ (C_6D_6 , 150 MHz).

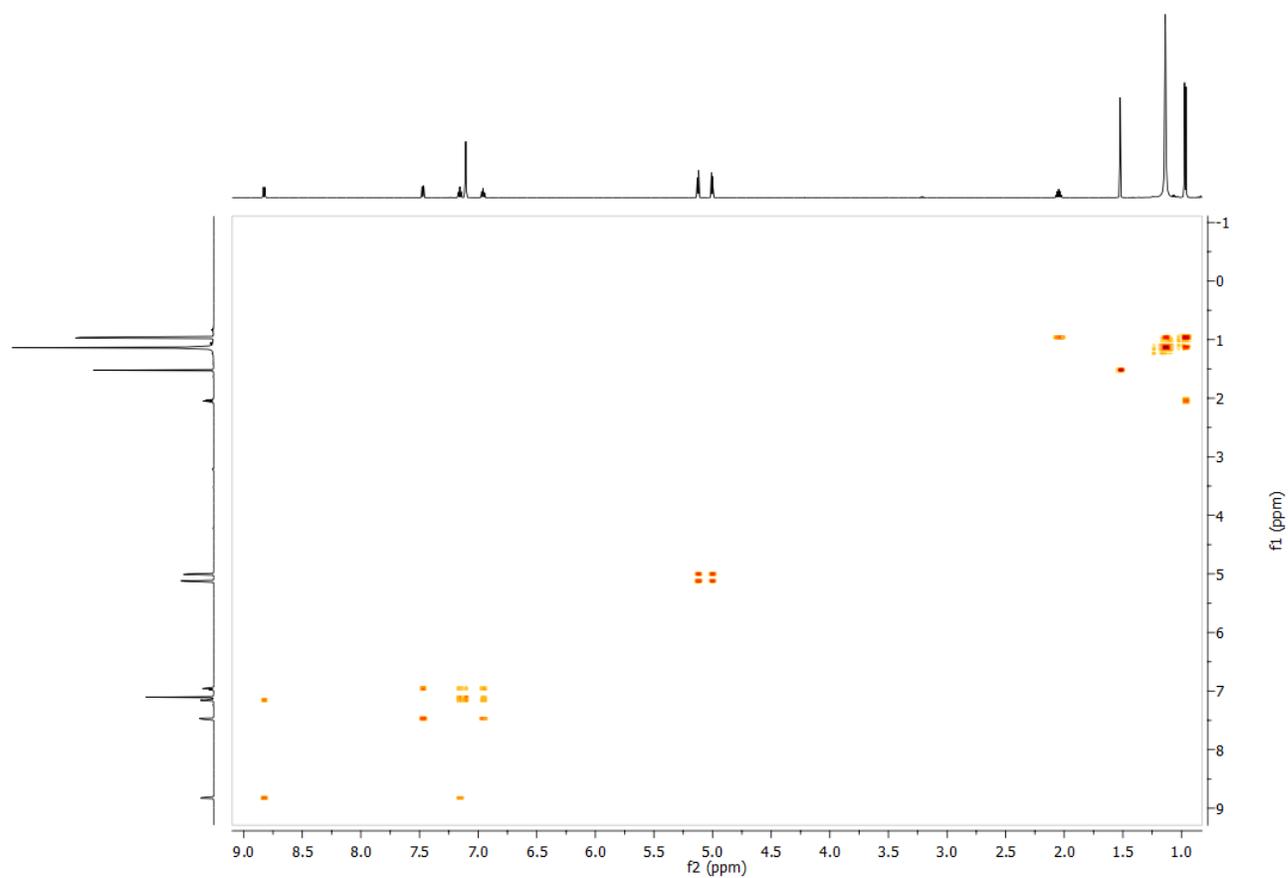


Figure S14. ¹H-¹H COSY NMR of Ru(cymene)(κ^2 -OC^tBu₂C₆H₂) (C₆D₆, 600 MHz).

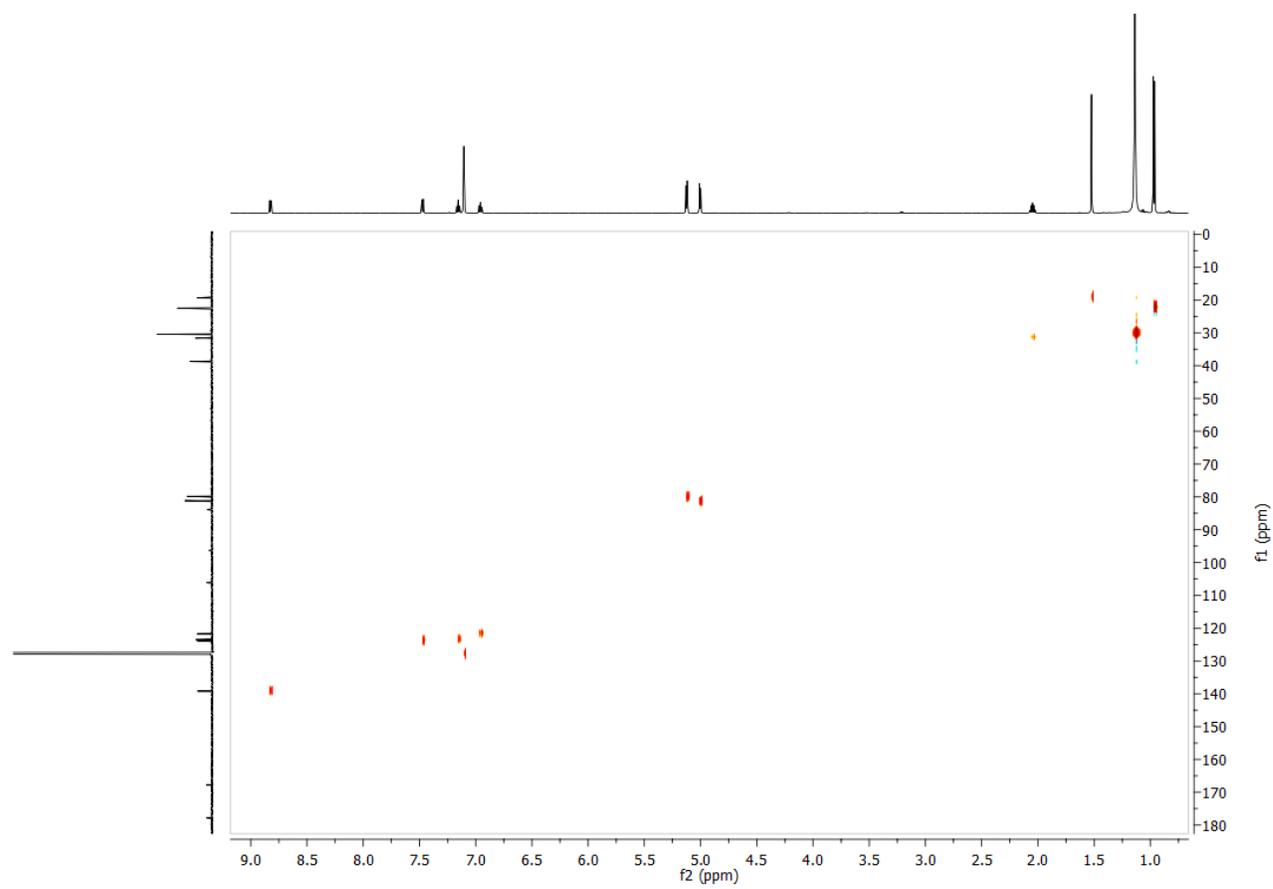


Figure S15. ^1H - ^{13}C HSQCAD NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2\text{C}_6\text{H}_2)$ (C_6D_6 , 600 MHz).

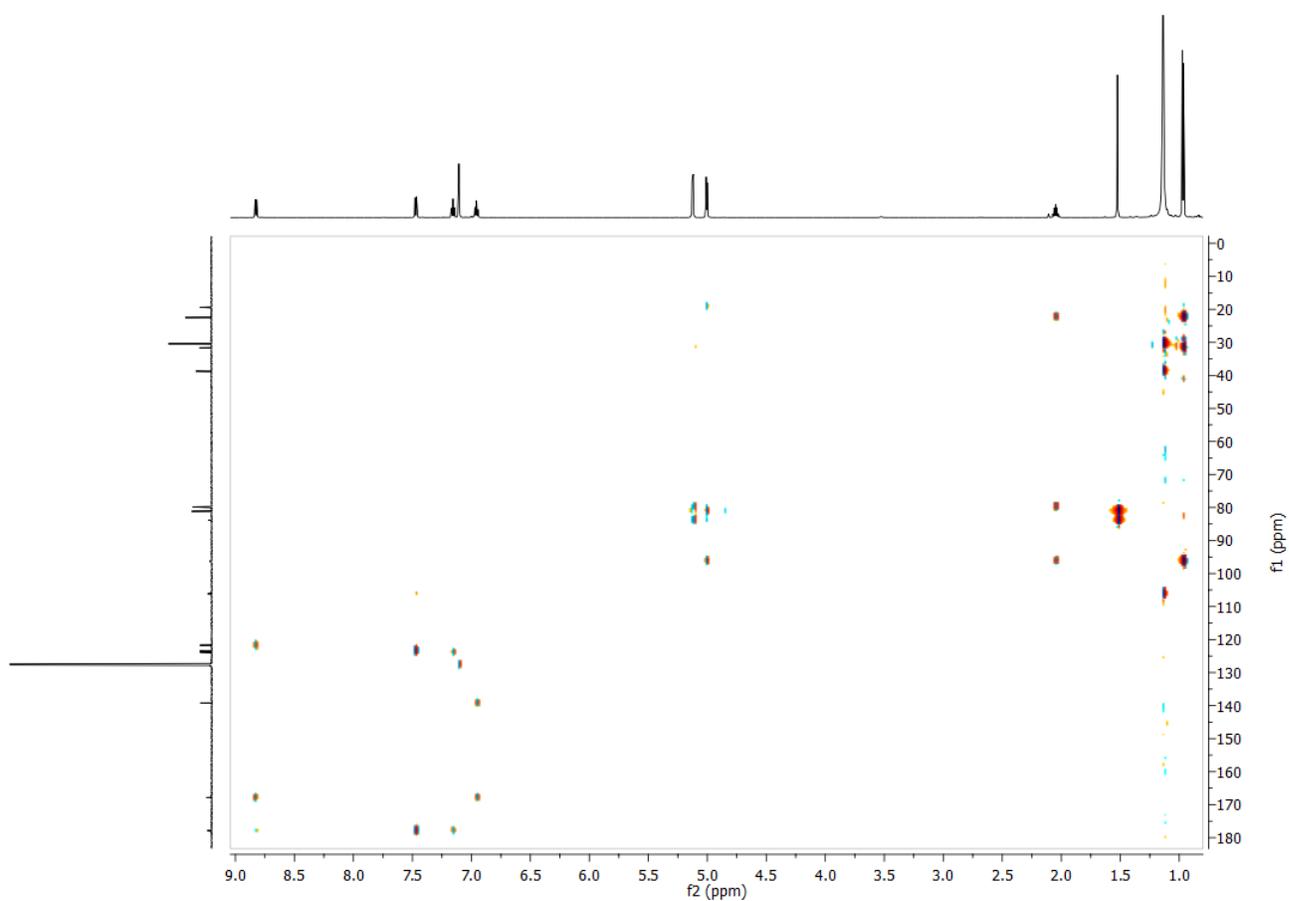


Figure S16. ^1H - ^{13}C HMBC NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2\text{C}_6\text{H}_2)$ (C_6D_6 , 600 MHz).

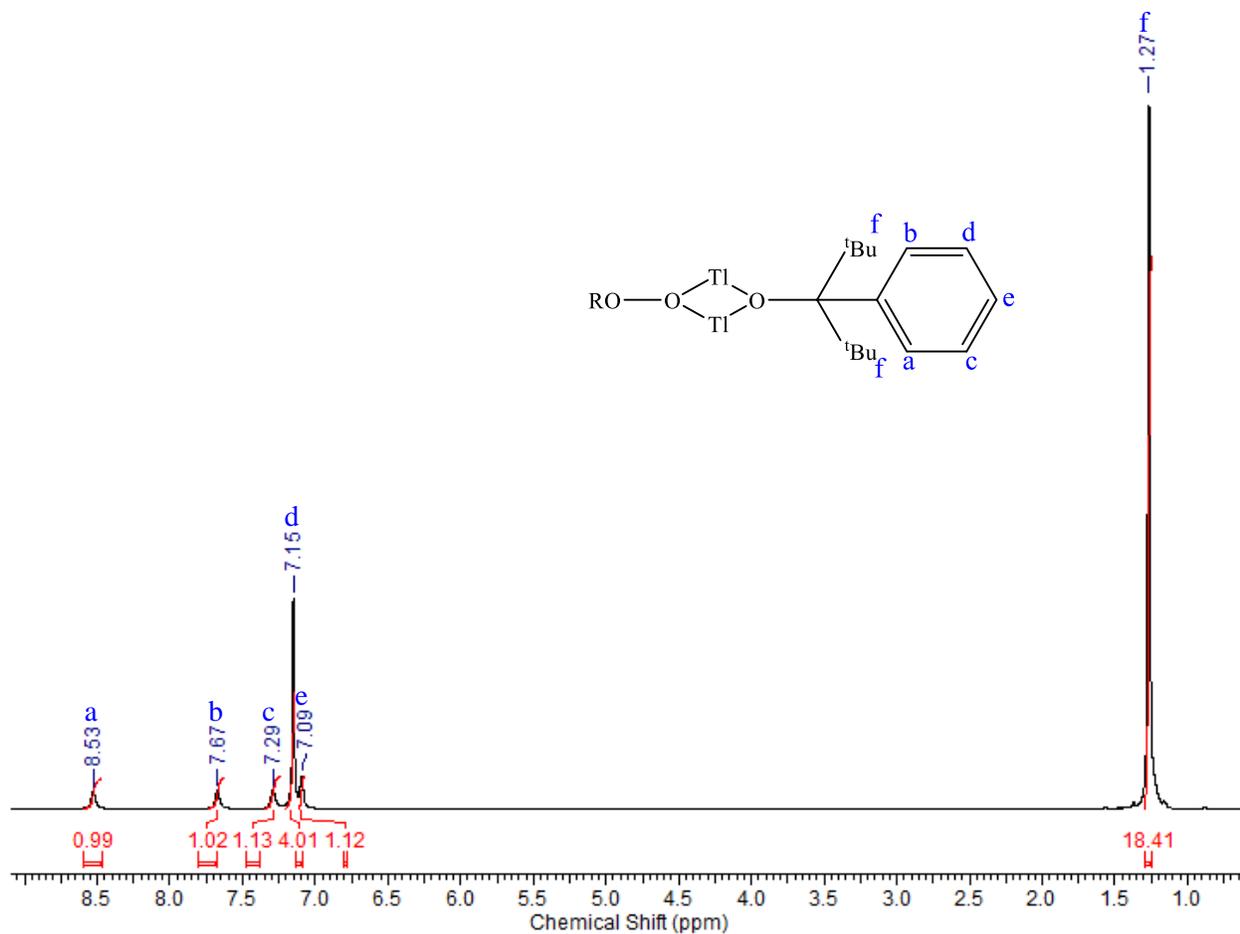


Figure S17. ^1H NMR of $\text{Tl}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ as obtained from the attempted preparation of “ $\text{Ni}(\text{OR})_2$ ” (C_6D_6 , 600 MHz).

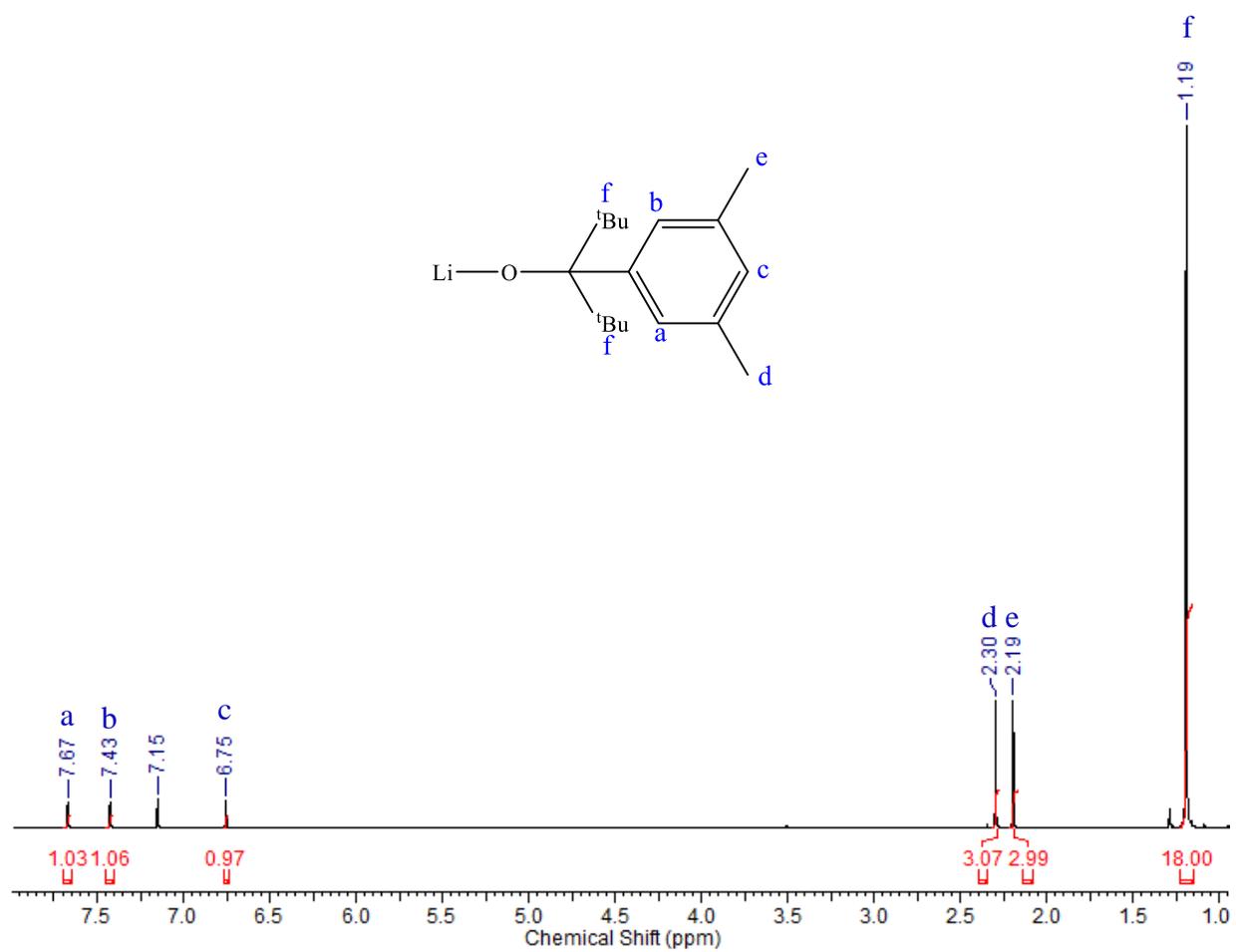


Figure S18. ^1H NMR of $\text{LiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 600 MHz).

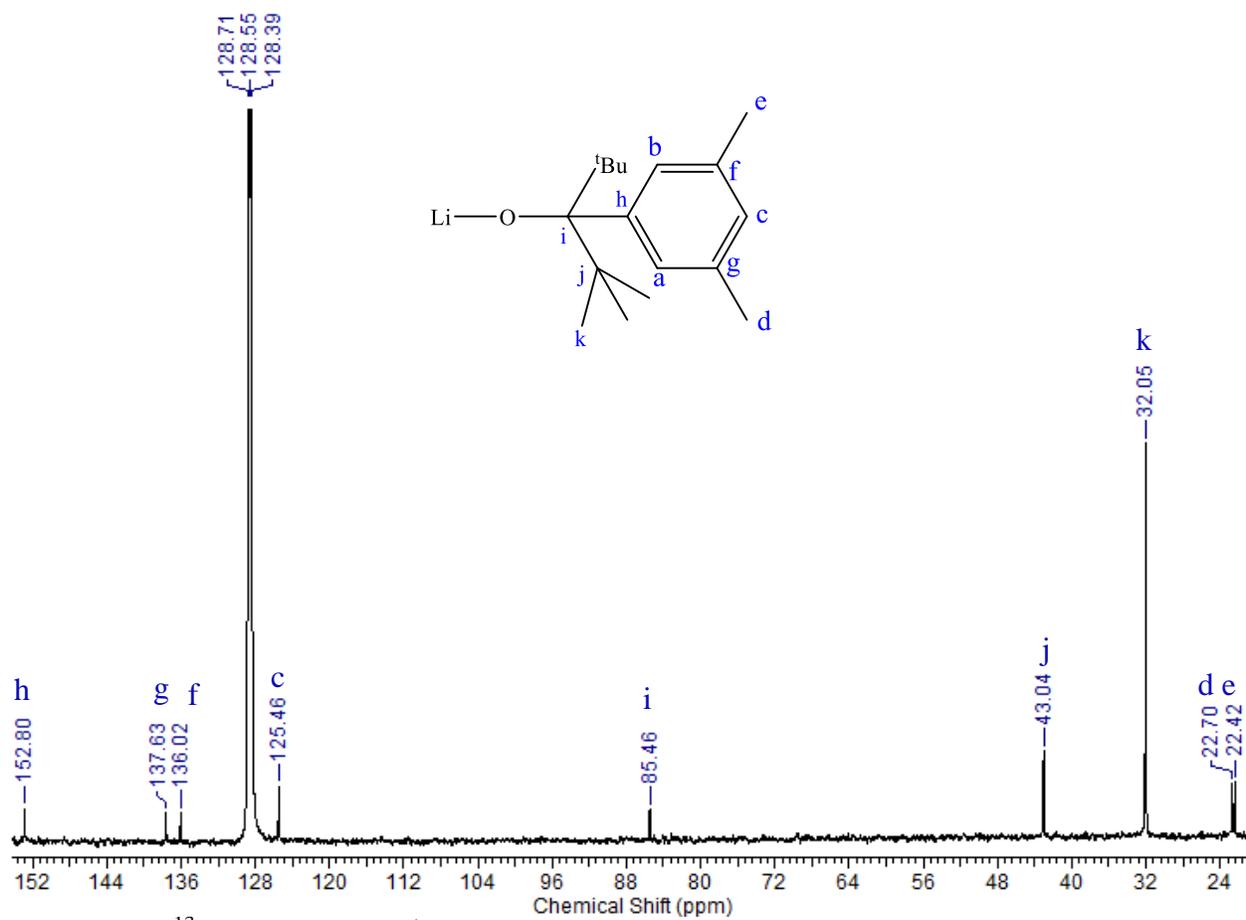


Figure S19. ^{13}C NMR of $\text{LiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 150 MHz). *Carbons **a** and **b** appear under C_6D_6 peaks, and were detected by HSQCAD, HMBC and HMQC.

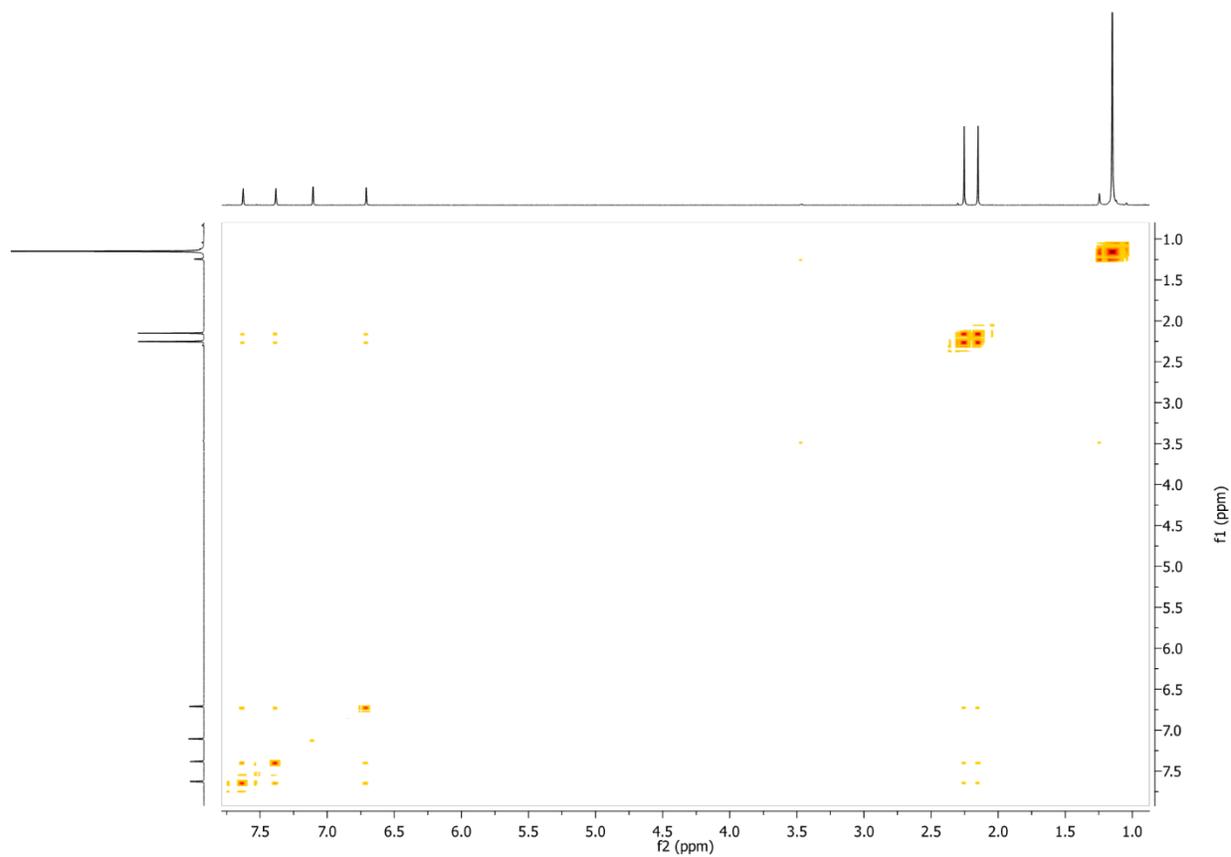


Figure S20. ^1H - ^1H COSY NMR of $\text{LiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 600 MHz).

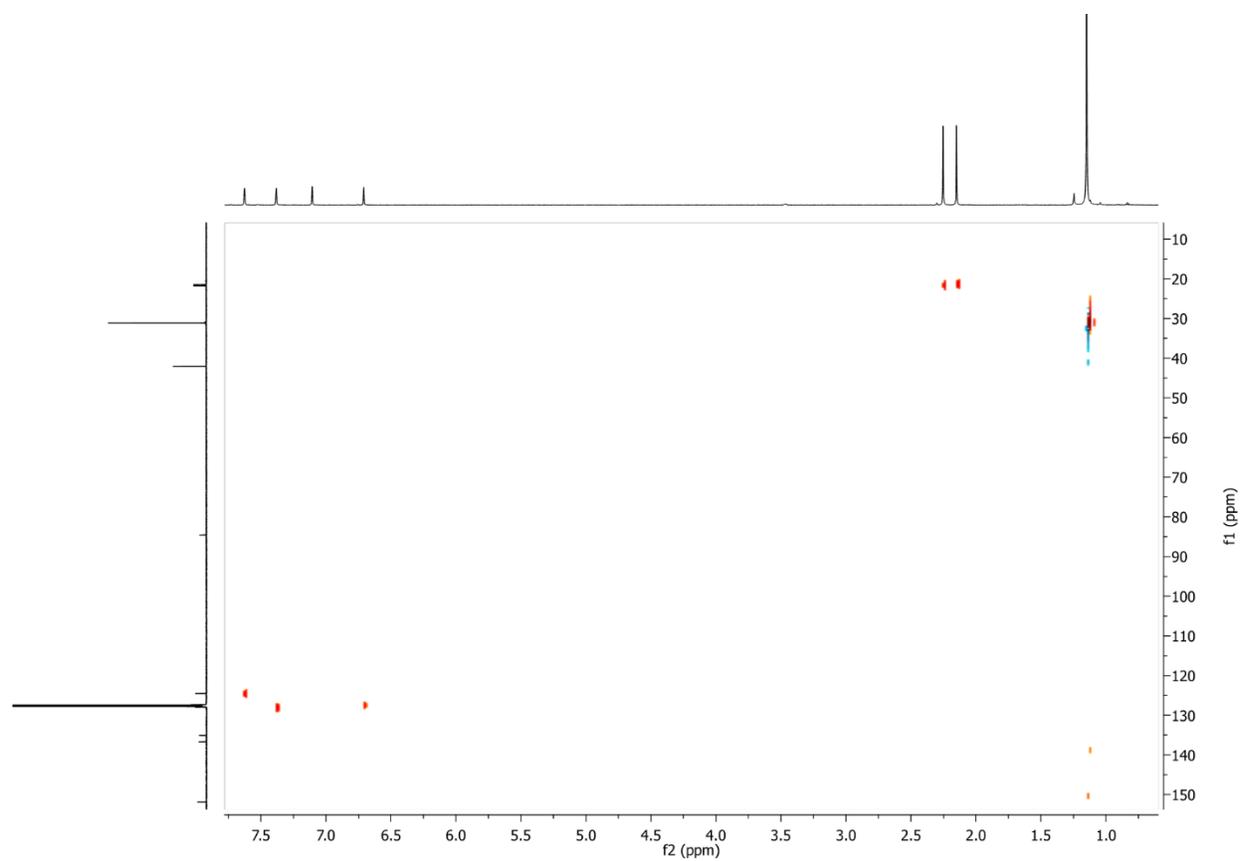


Figure S21. ^1H - ^{13}C HSQC NMR of $\text{LiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 600 MHz).

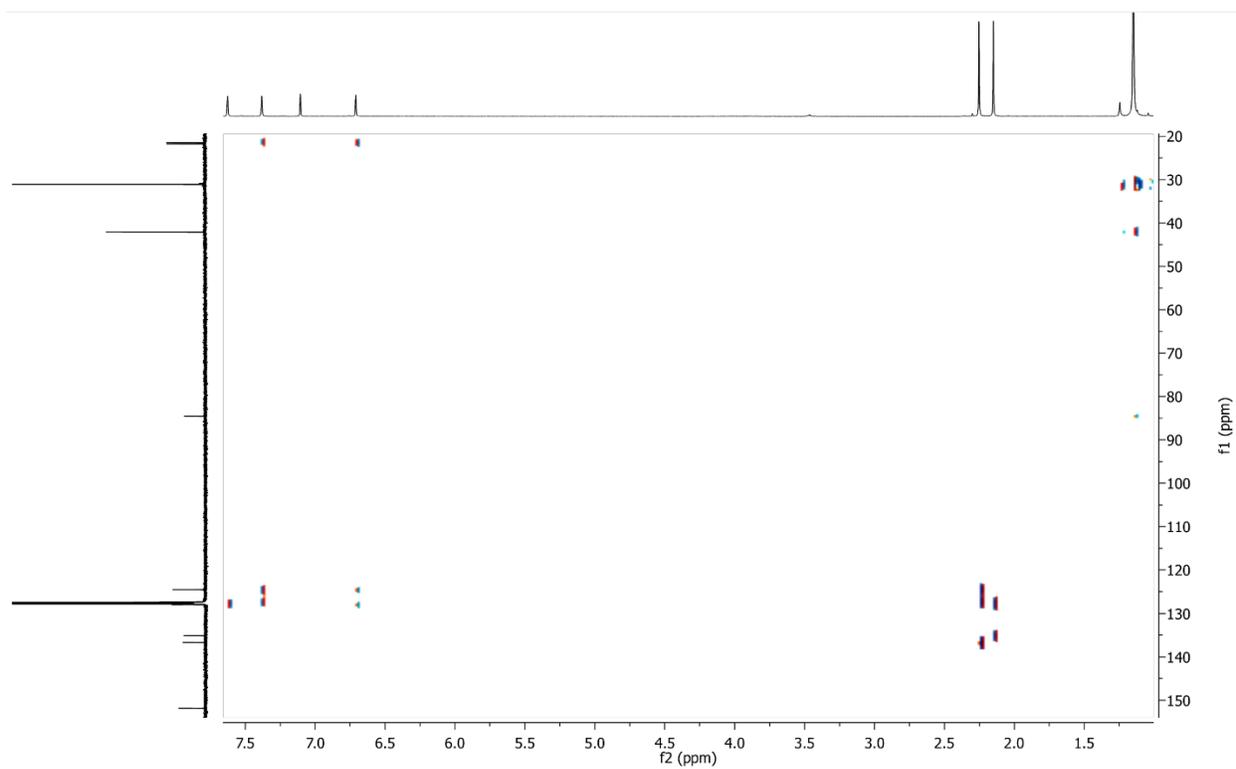


Figure S22. ^1H - ^{13}C HMBCAD NMR of $\text{LiOC}'\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 600 MHz).

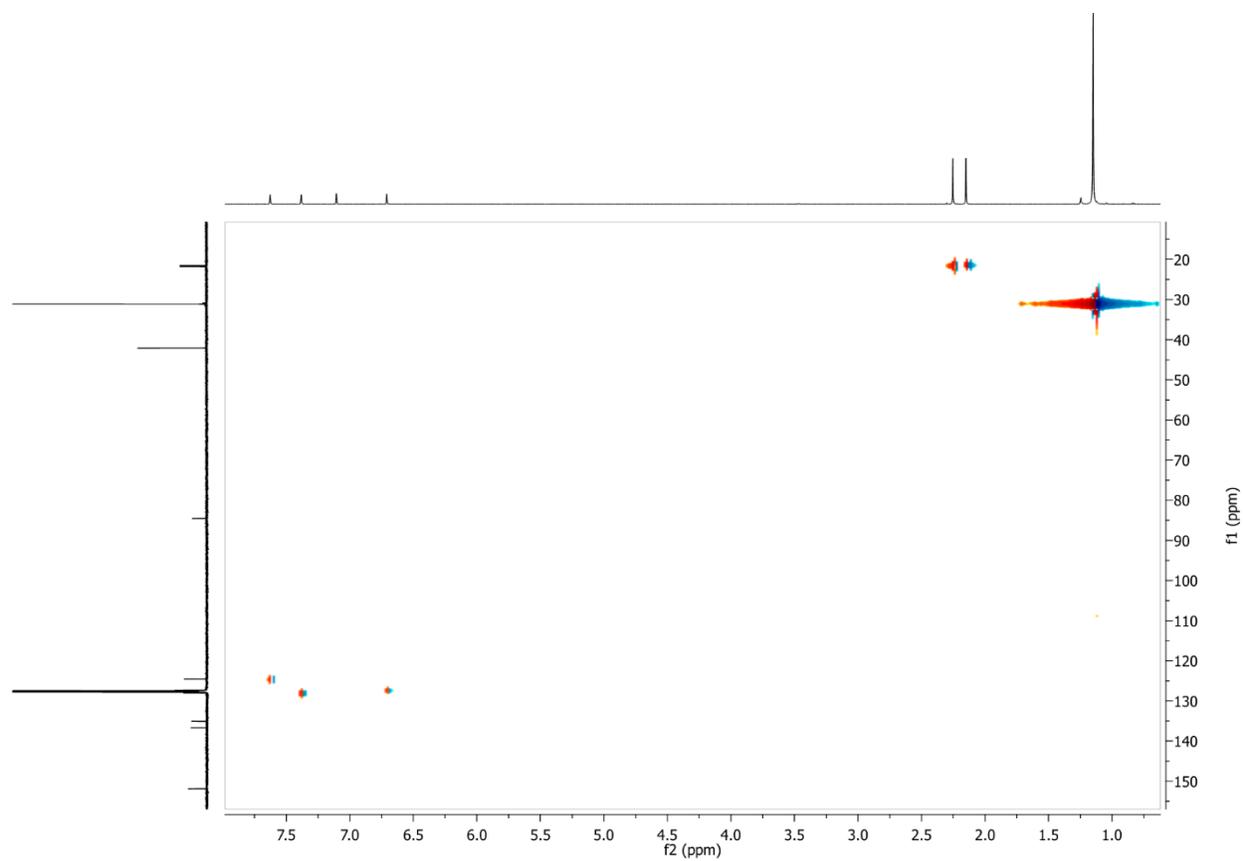


Figure S23. ^1H - ^{13}C HMQC NMR of $\text{LiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 600 MHz).

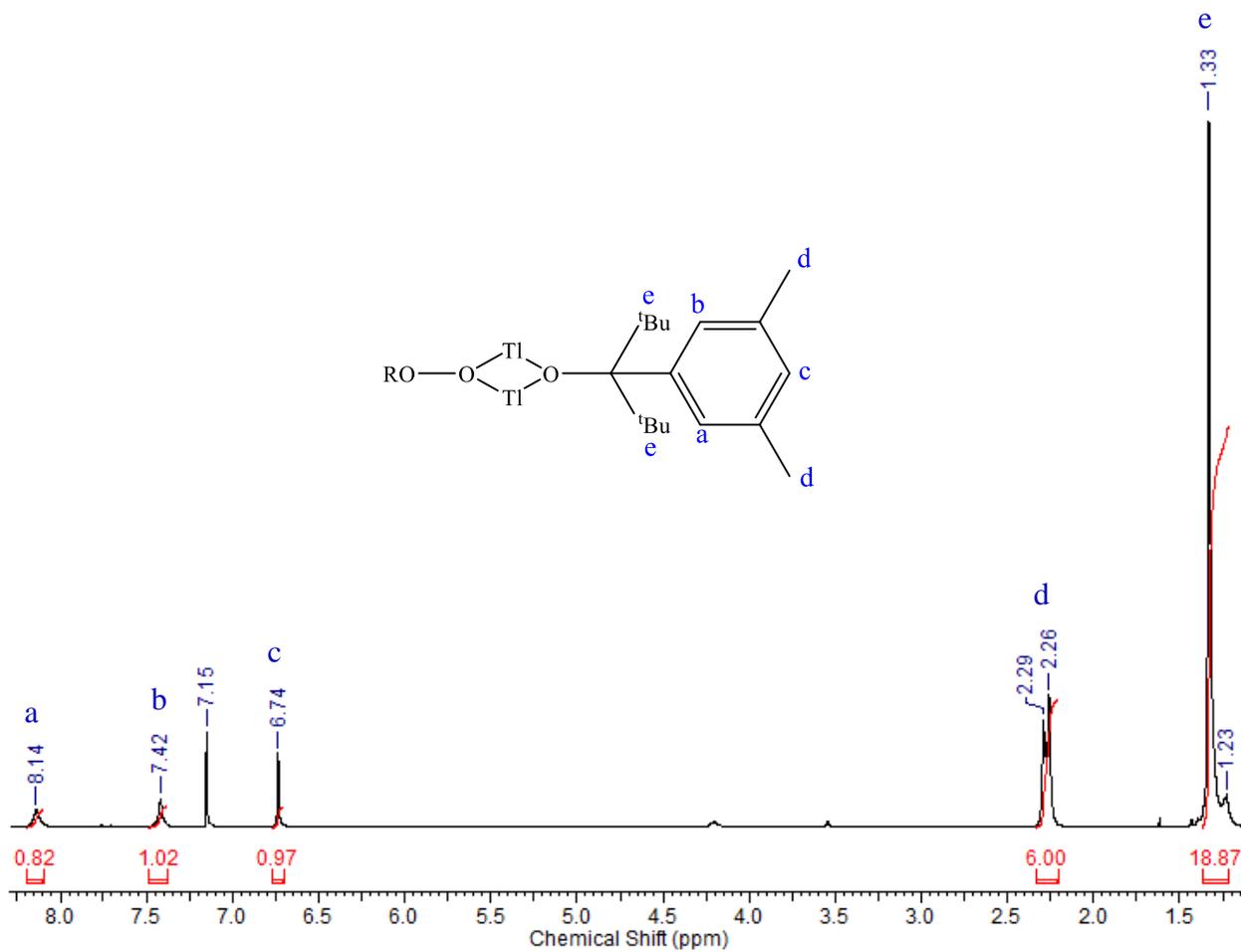


Figure S24. ^1H NMR of $\text{TiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 600 MHz) (single isomer, obtained after prolonged storage at $-35\text{ }^\circ\text{C}$).

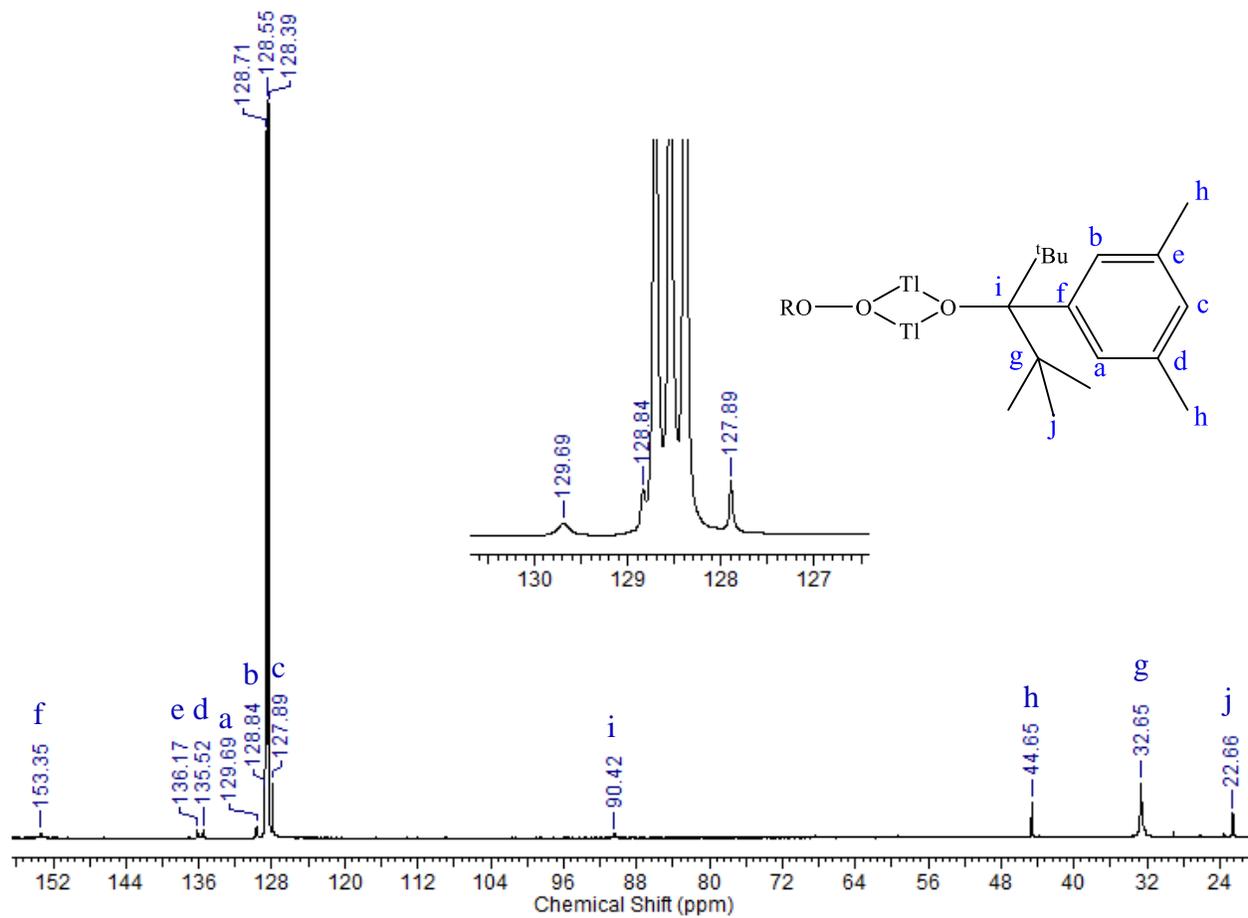


Figure S25. ^{13}C NMR of $\text{TiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (C_6D_6 , 150 MHz).

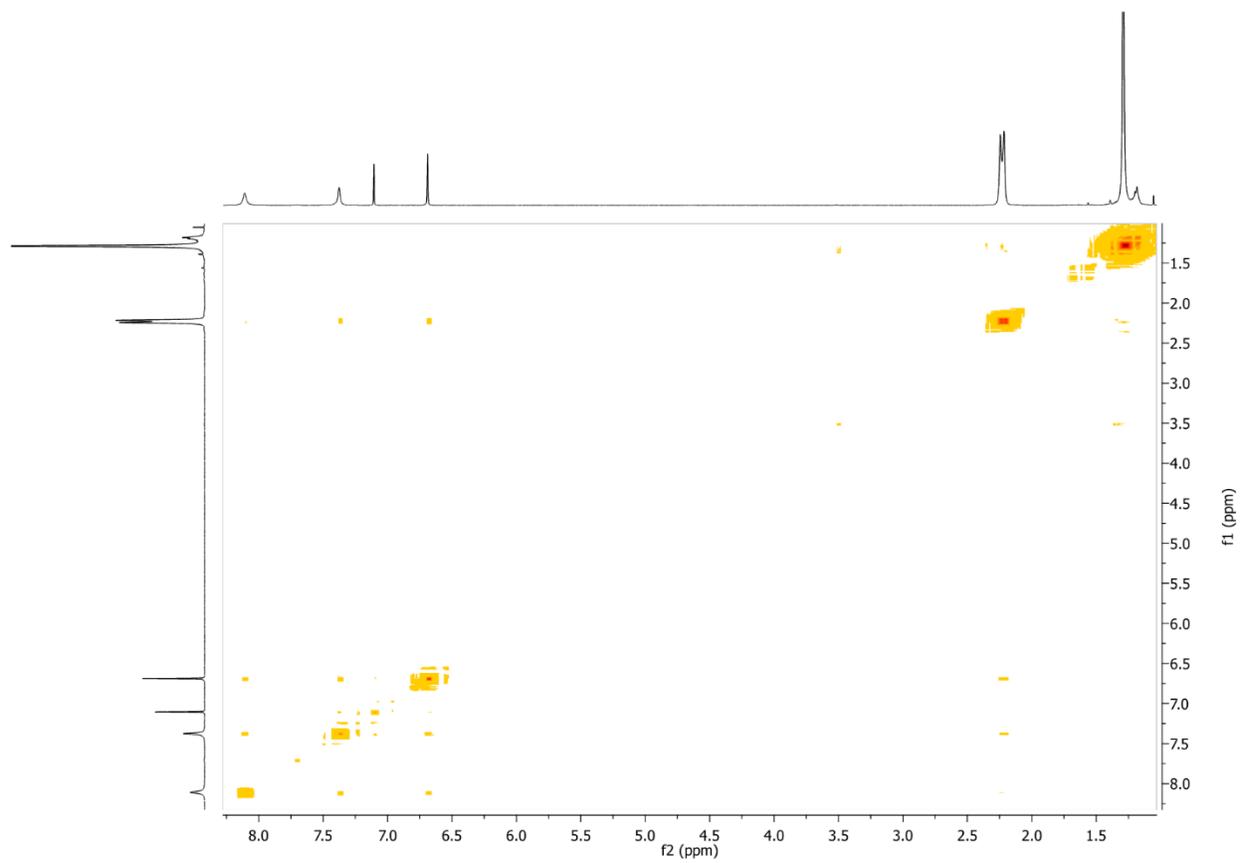


Figure S26. ¹H-¹H COSY NMR of TiOC^tBu₂(3,5-Me₂C₆H₃) (C₆D₆, 600 MHz).

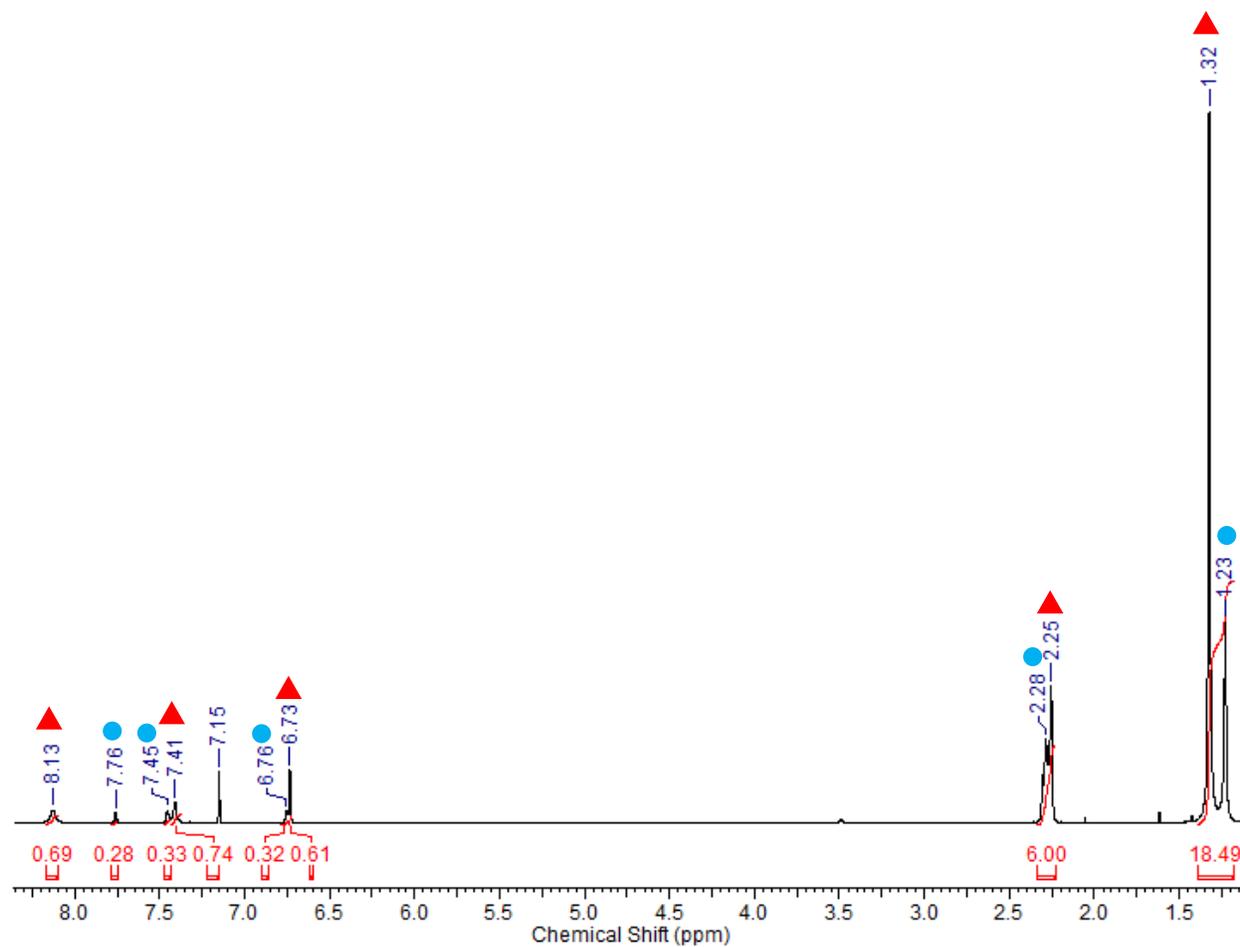


Figure S27. ^1H NMR of freshly prepared $\text{Tl}_2(\text{OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))_2$ (C_6D_6 , 600 MHz, room temperature) demonstrating two isomers.

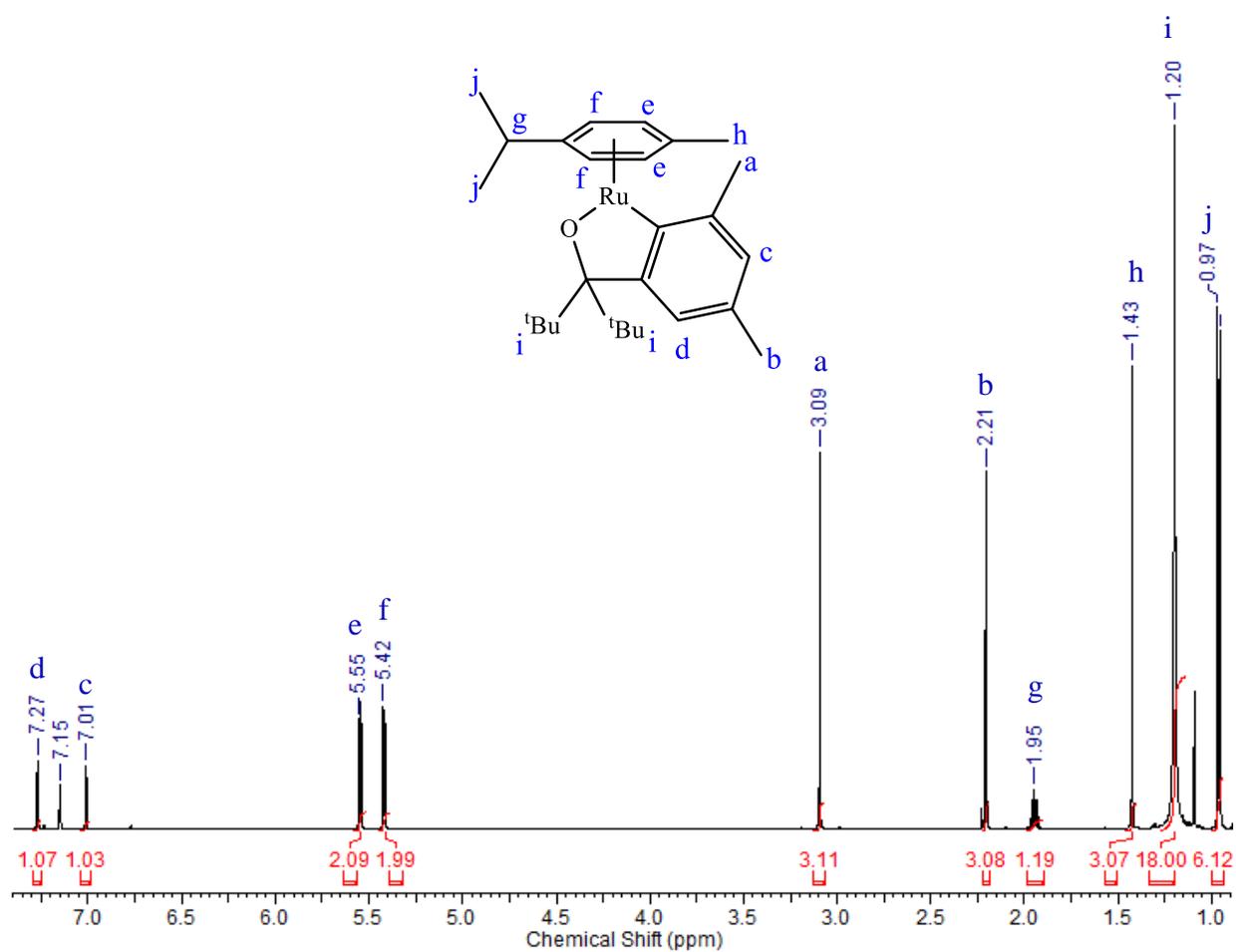


Figure S28. ¹H NMR of Ru(cymene)(κ²-OC^tBu₂(3,5-Me₂C₆H₃)) (C₆D₆, 600 MHz).

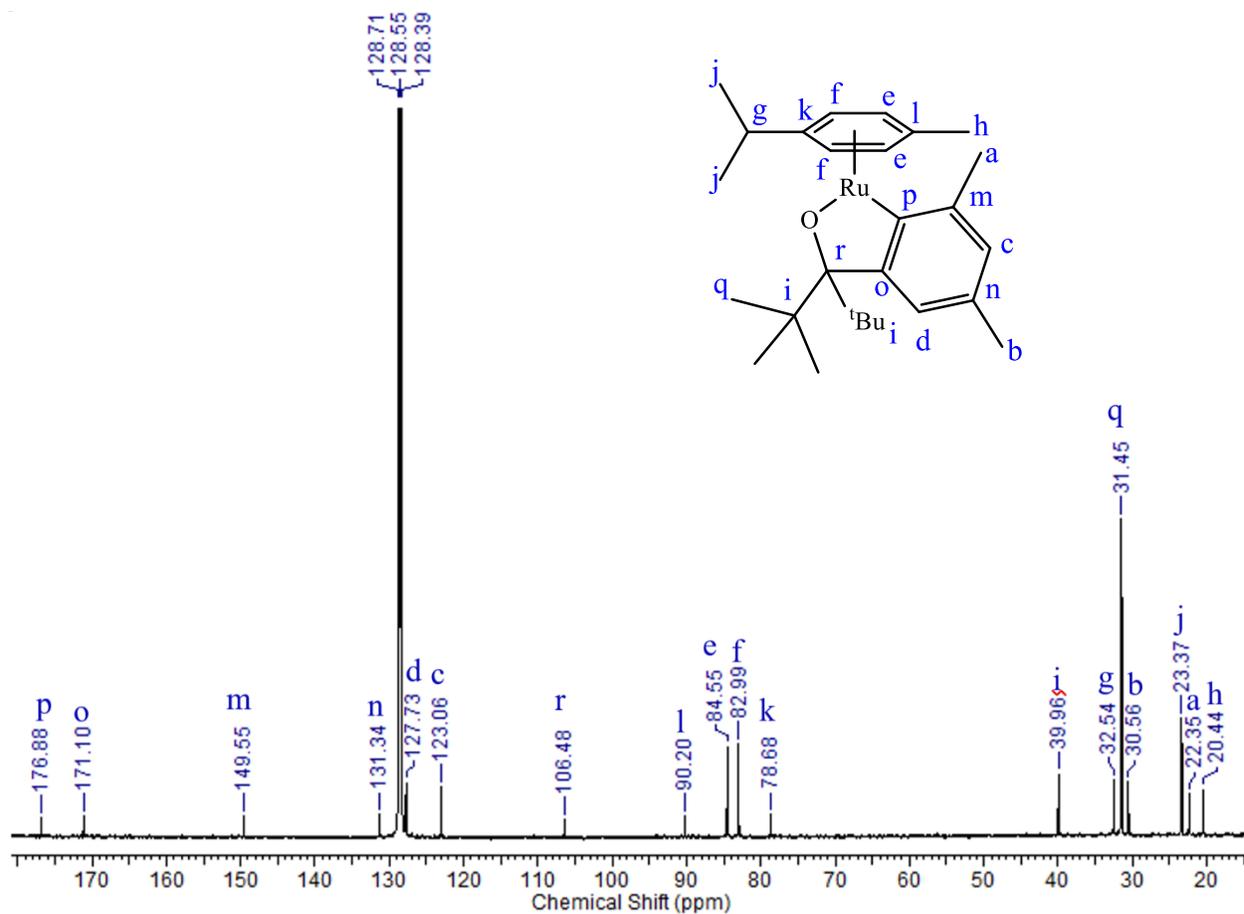


Figure S29. ^{13}C NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))$ (C_6D_6 , 150 MHz).

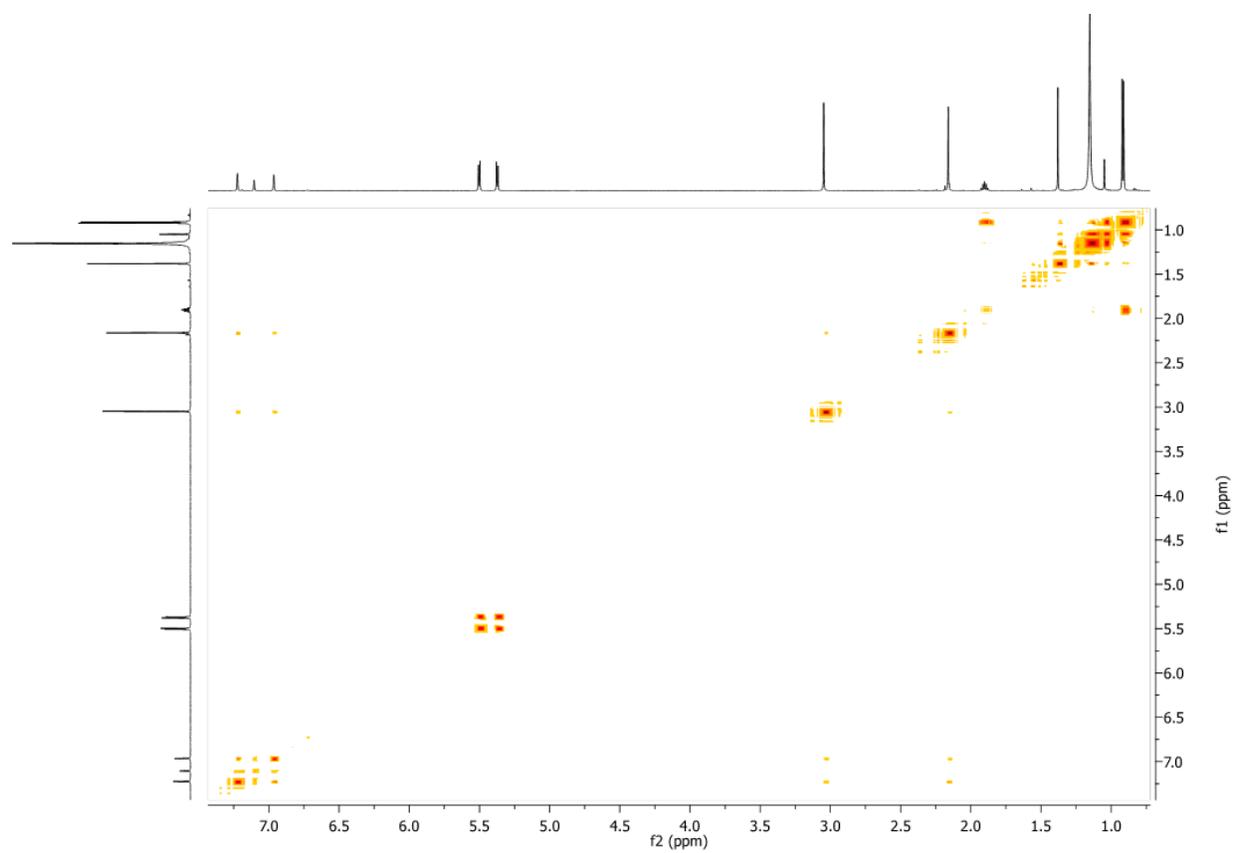


Figure S30. ^1H - ^1H COSY NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))$ (C_6D_6 , 600 MHz).

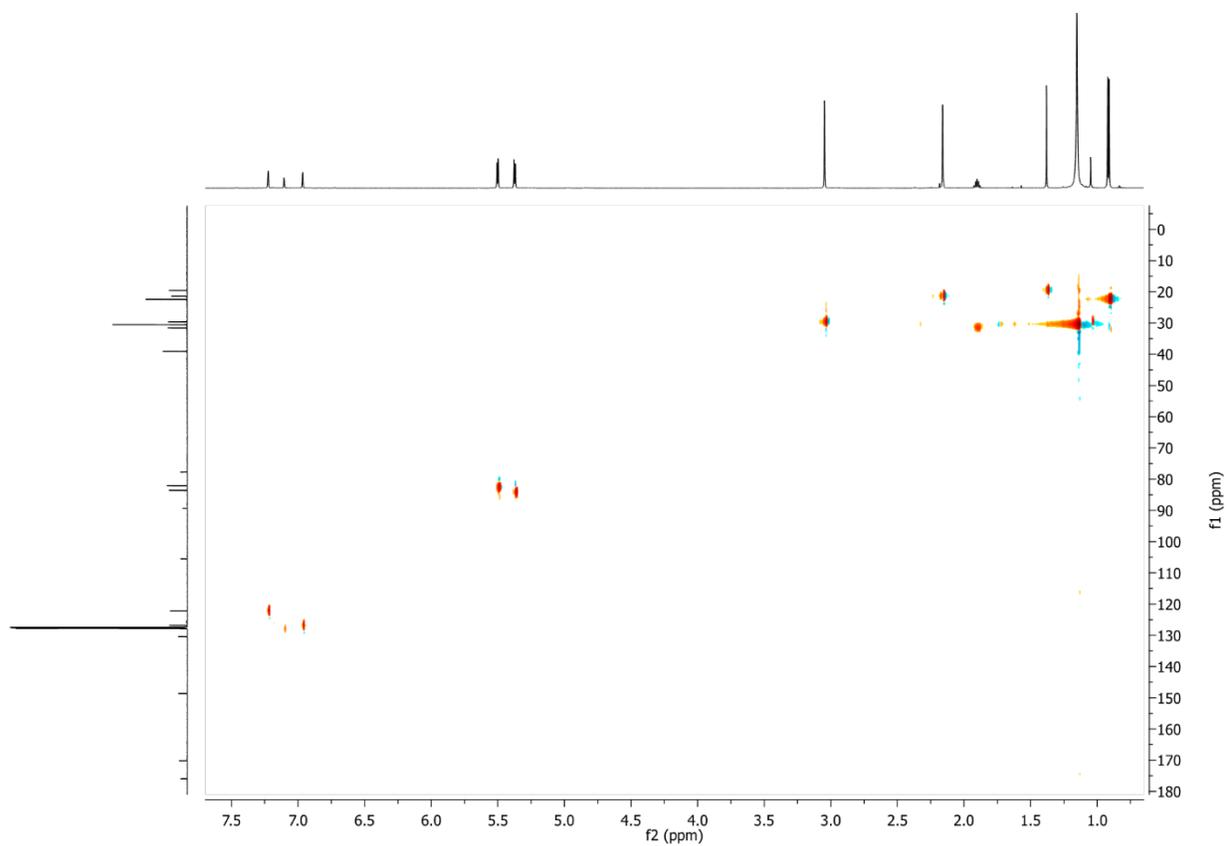


Figure S31. ^1H - ^{13}C HSQCAD NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))$ (C_6D_6 , 600 MHz).

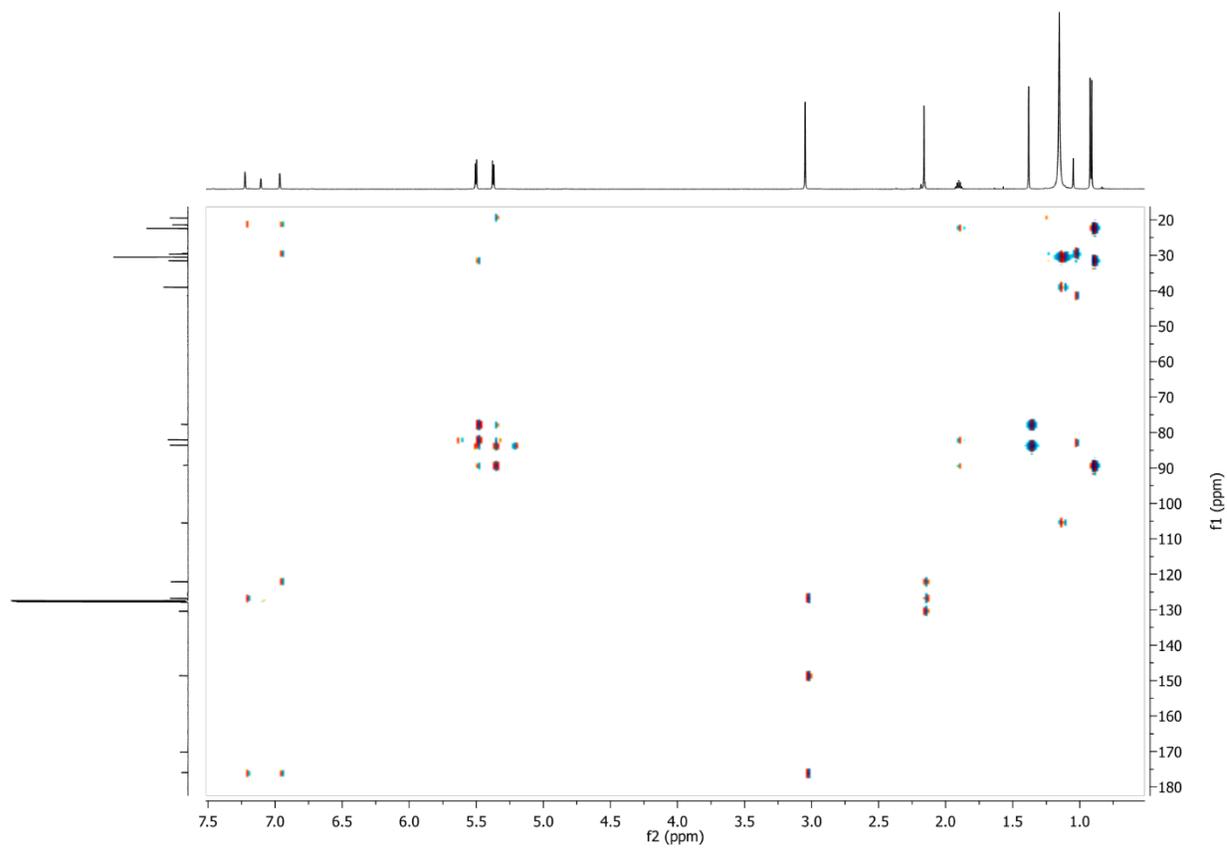


Figure S32. ^1H - ^{13}C HMBCAD NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))$ (C_6D_6 , 600 MHz).

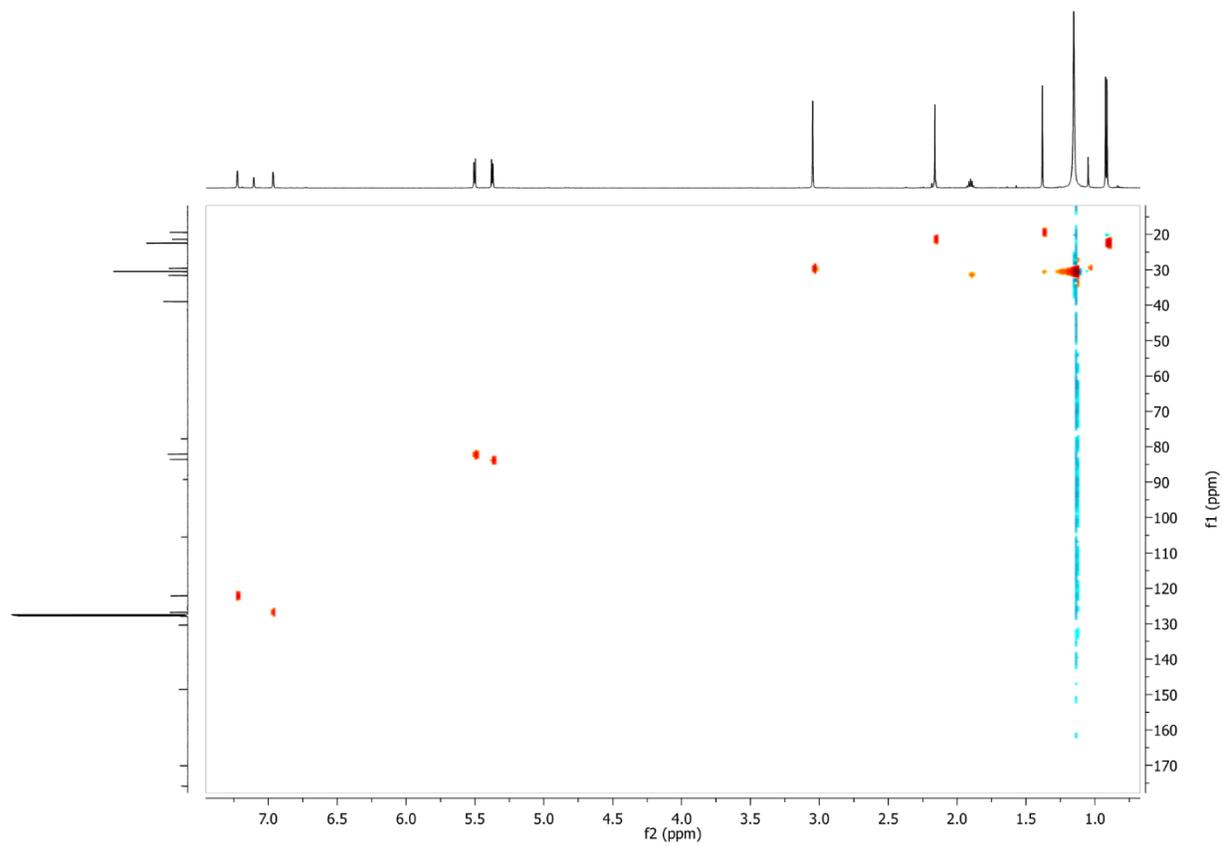


Figure S33. ^1H - ^{13}C HMQC NMR of $\text{Ru}(\text{cymene})(\kappa^2\text{-OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))$ (C_6D_6 , 600 MHz).

2. UV-Vis Spectra

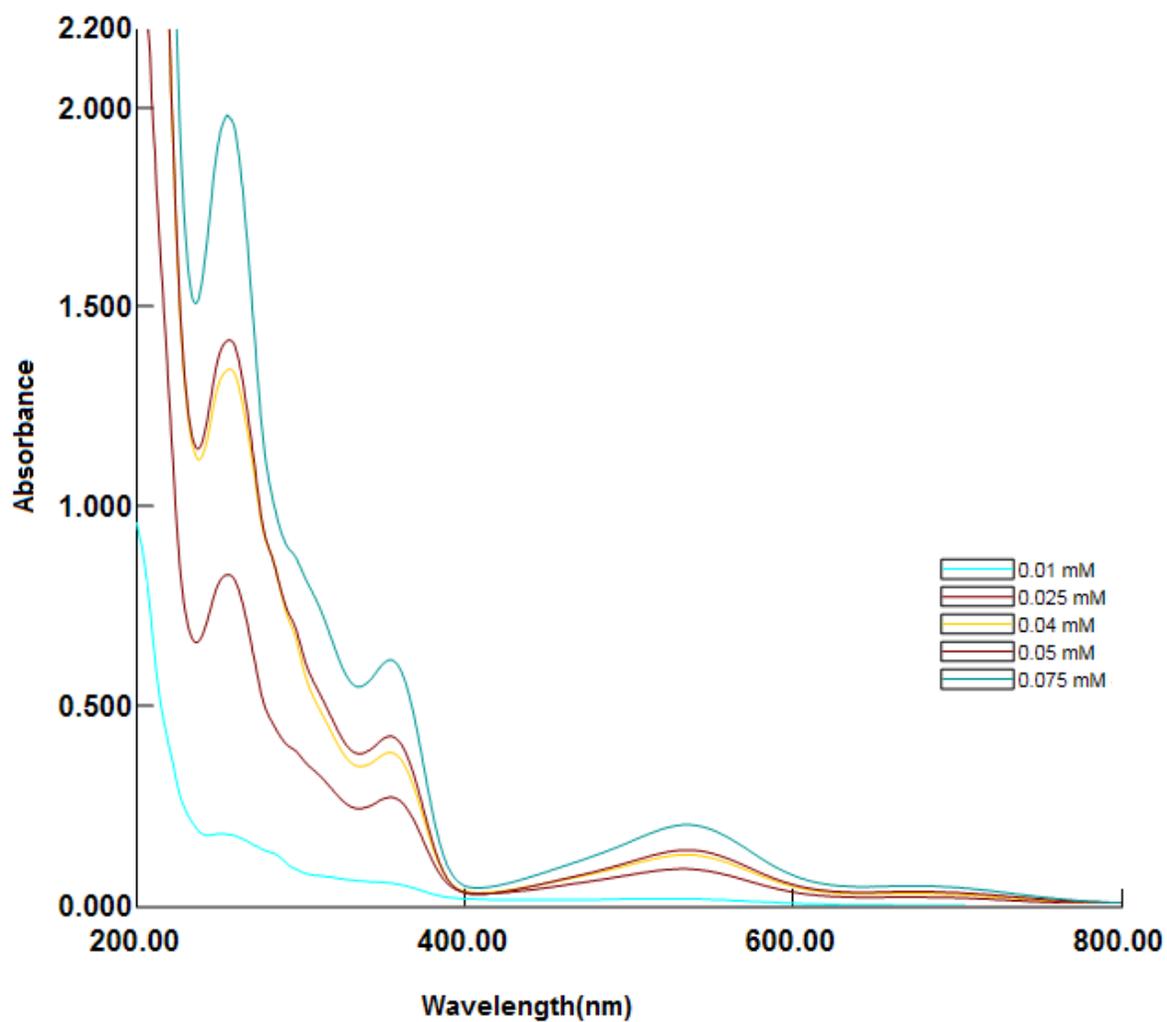


Figure S34. UV-Vis spectra for Ru(cymene)(κ^2 -OC^tBu₂C₆H₂) at five different concentrations. λ_{max} , nm (ϵ_M , L mol⁻¹ cm⁻¹): 539 (2800), 355 (8600), 296 (sh, 13400), 256 (28300).

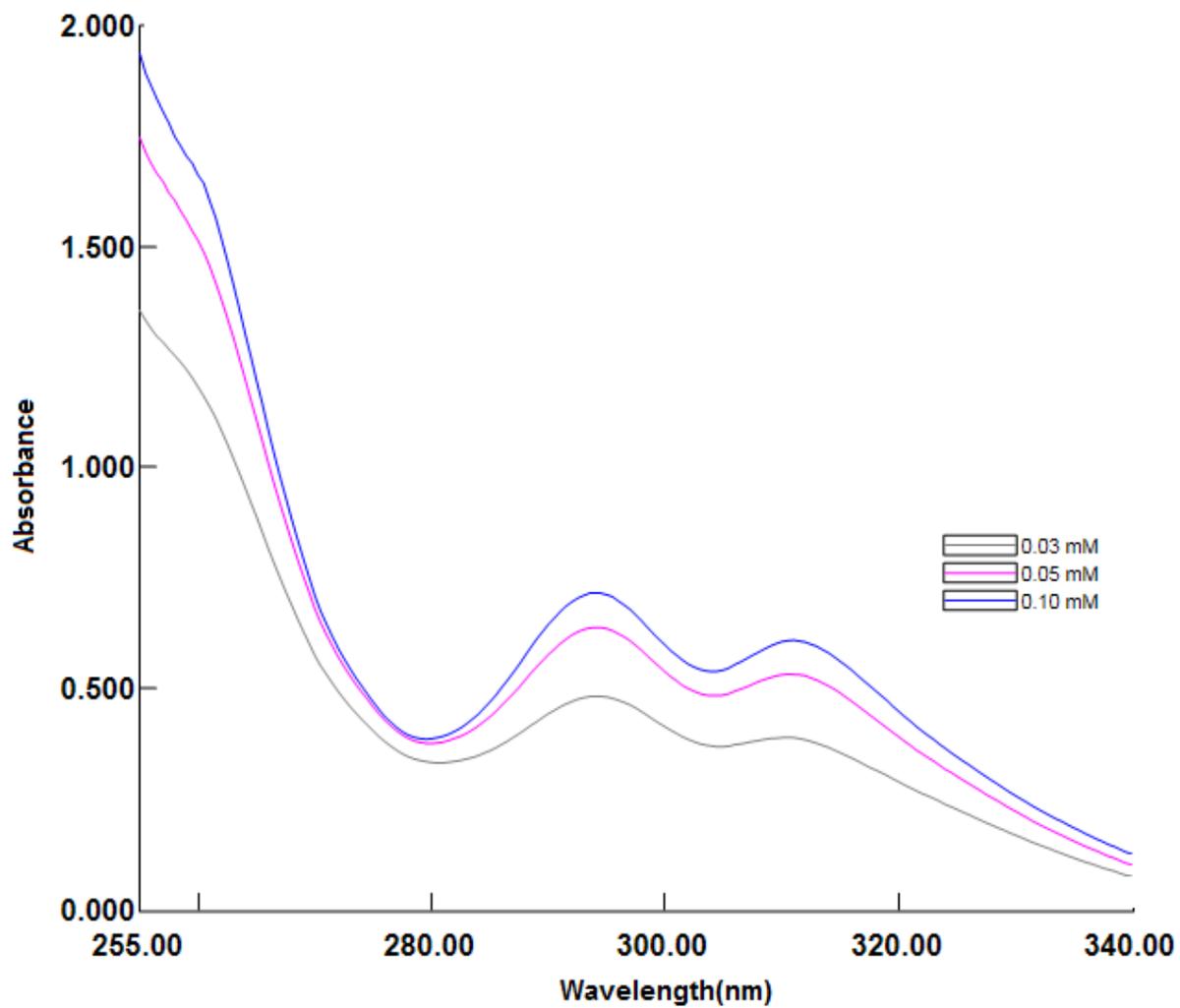


Figure S35. UV-Vis spectrum for $\text{Ti}_2(\text{OC}^t\text{Bu}_2\text{Ph})_2$ at three different concentrations. λ_{max} , nm (ϵ_M , $\text{L mol}^{-1} \text{cm}^{-1}$): 311 (3100), 294 (3300), 261(sh, 6500).

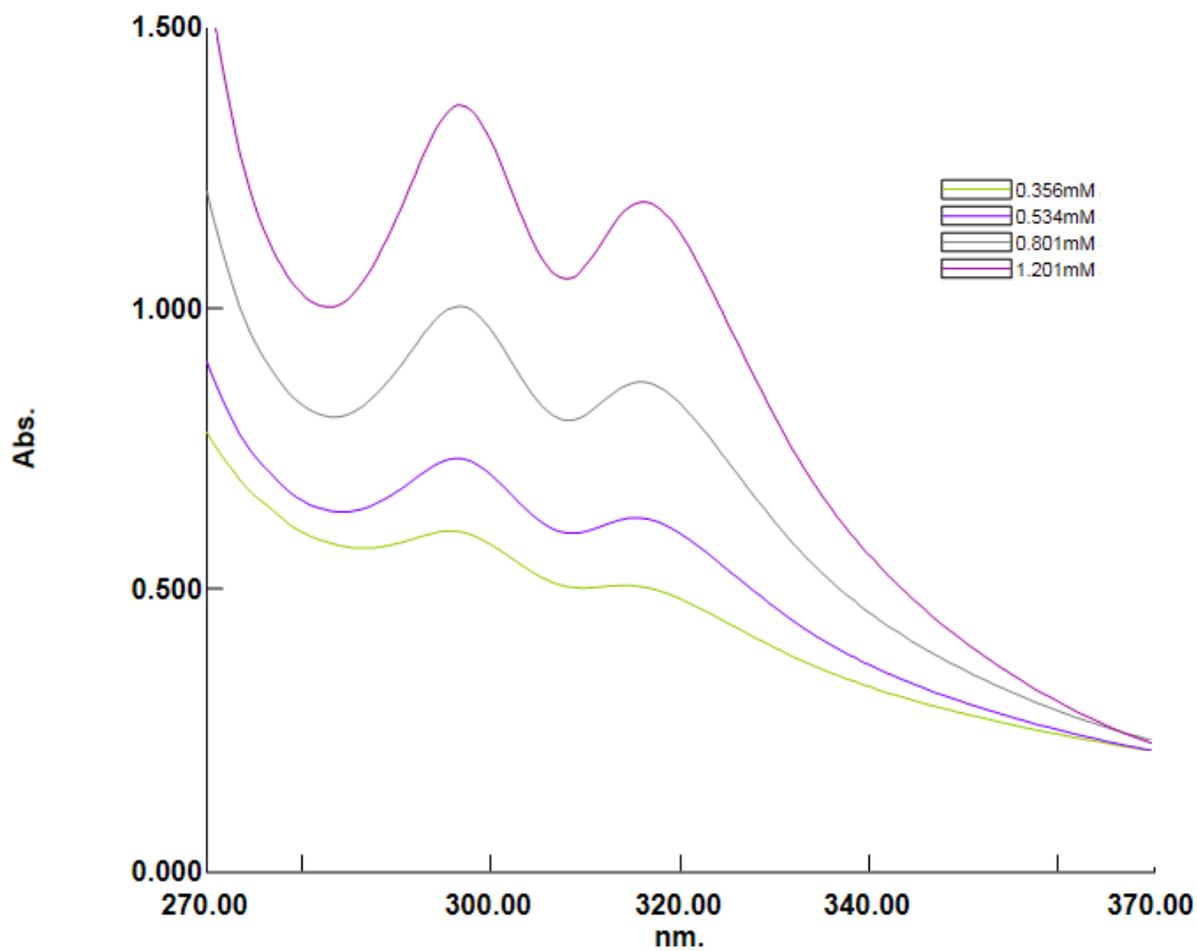


Figure S36. UV-Vis spectrum for $\text{Ti}_2(\text{OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))_2$ at four different concentrations. λ_{max} , nm (ϵ_{M} , $\text{L mol}^{-1} \text{cm}^{-1}$): 316 (900), 297 (1200).

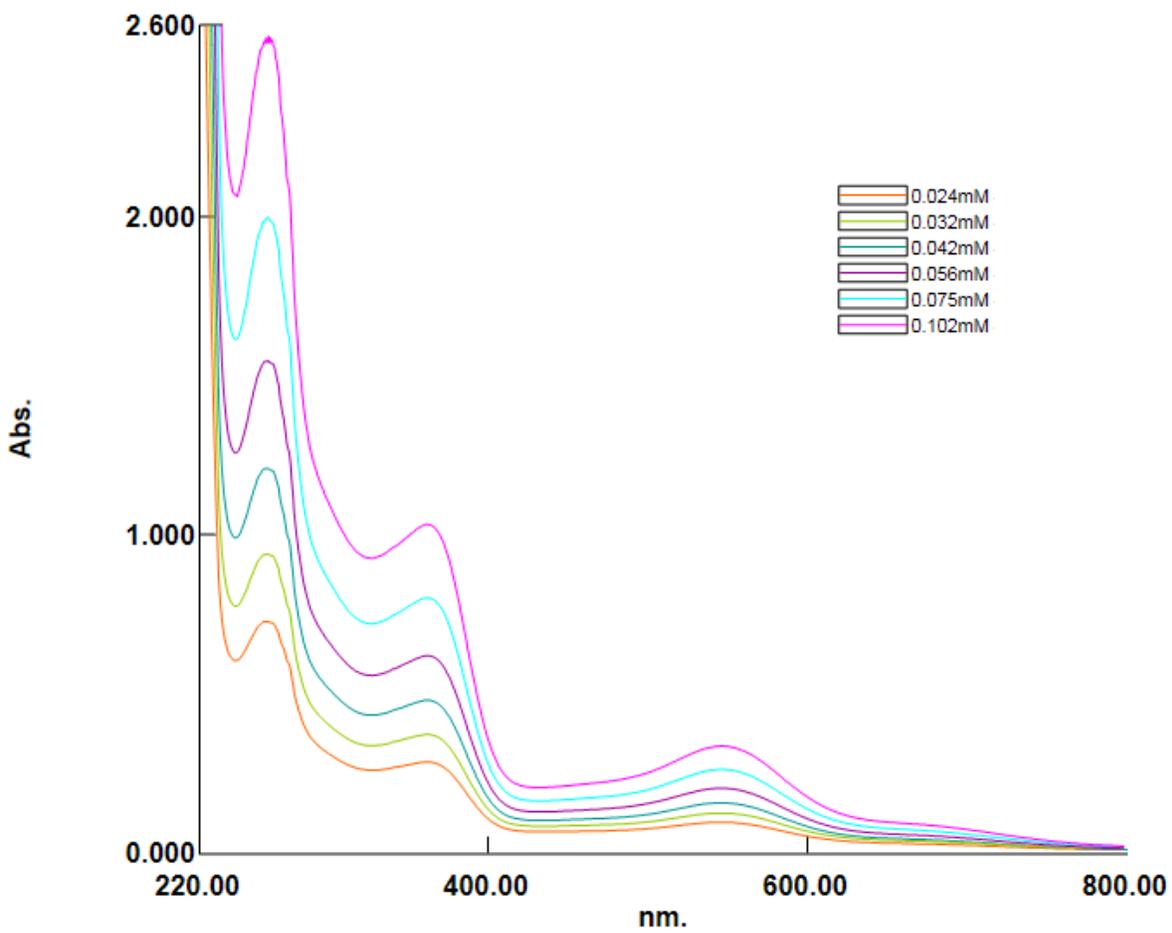


Figure S37. UV-Vis spectra for Ru(cymene)(κ^2 -OC^tBu₂(3,5-Me₂C₆H₃)) at six different concentrations. λ_{max} , nm (ϵ_M , L mol⁻¹ cm⁻¹): 545 (5200), 361 (15400), 263 (30000).

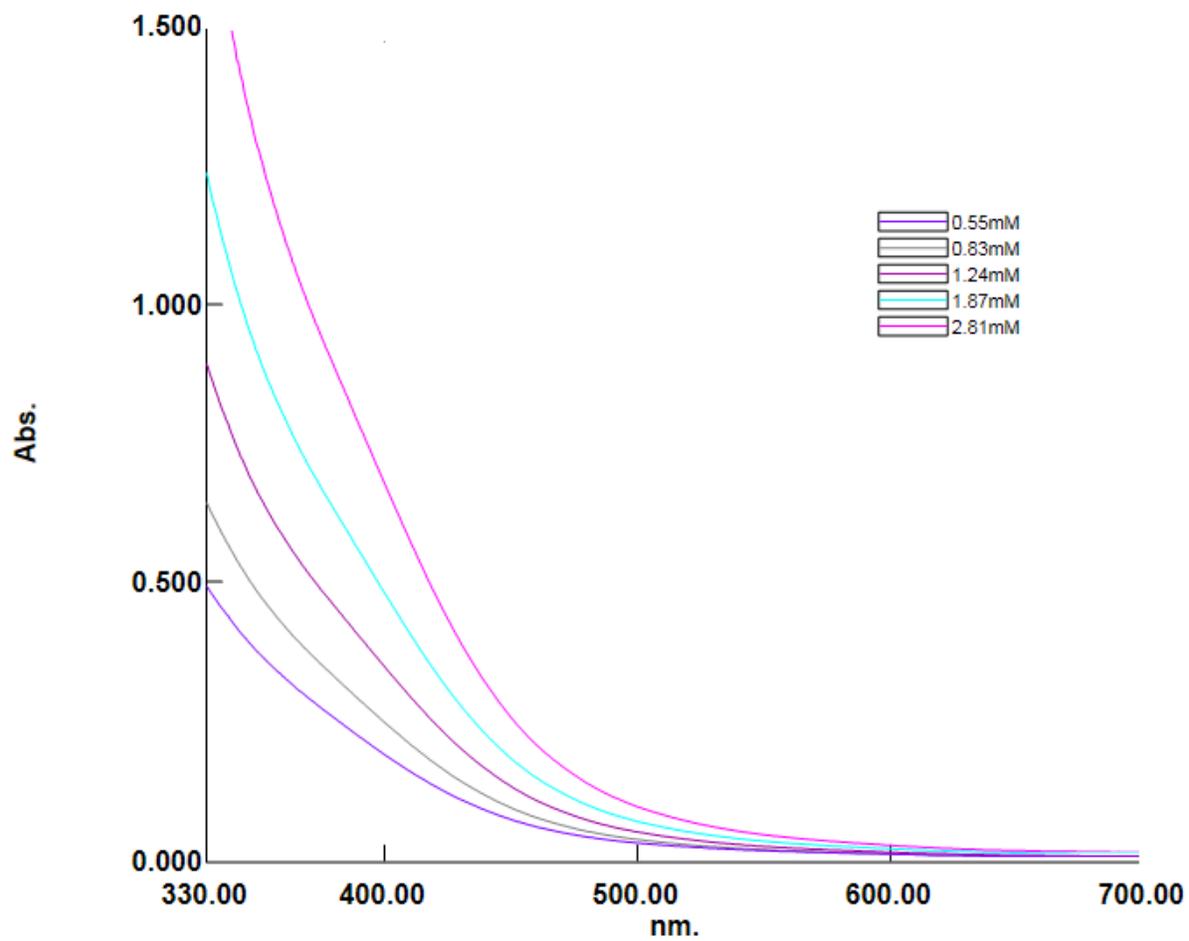


Figure S38. UV-Vis spectrum of Fe(OCtBu₂(3,5-Me₂C₆H₃))₂(THF)₂ at five different concentrations. λ_{max} , nm (ϵ_M , L mol⁻¹ cm⁻¹): 407 (sh, 250).

3. IR Spectra

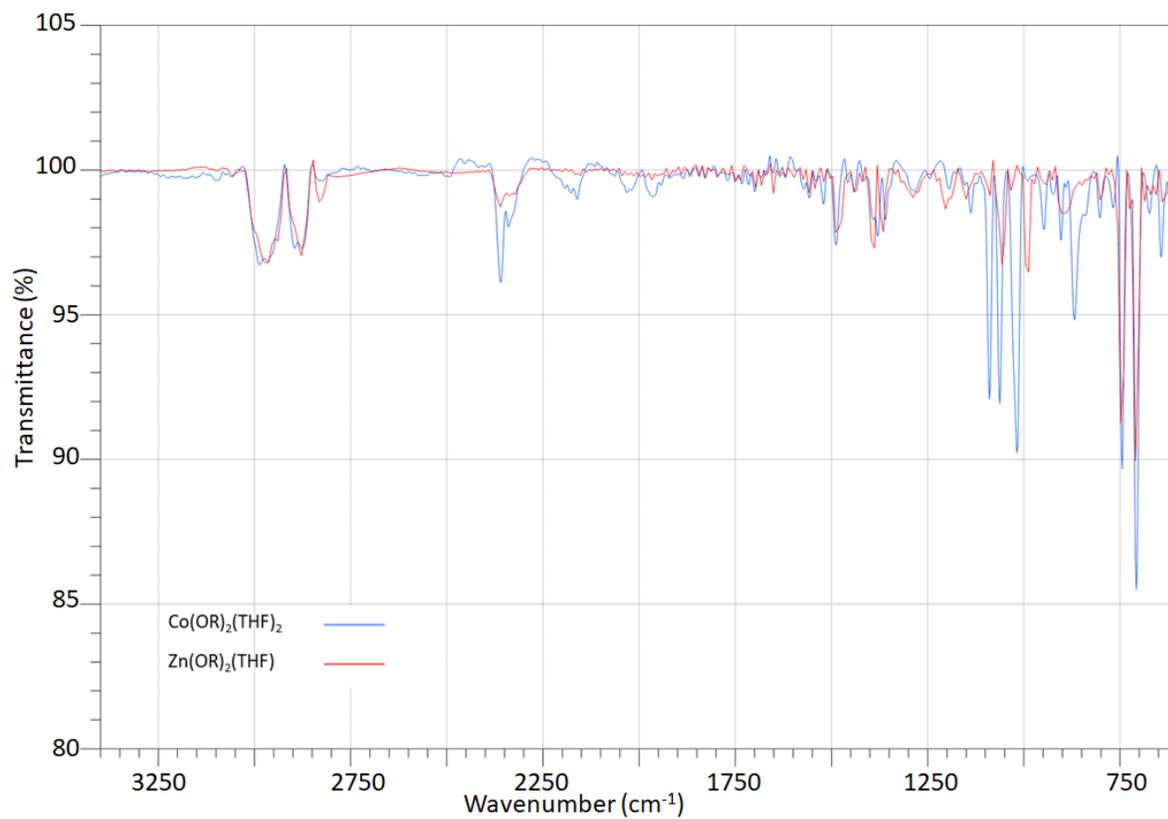


Figure S39. Overlaid IR spectra of $\text{Co}(\text{OC}^t\text{Bu}_2\text{Ph})_2(\text{THF})_2$ and $\text{Zn}(\text{OC}^t\text{Bu}_2\text{Ph})_2(\text{THF})$ in the 3400-600 cm⁻¹ range.

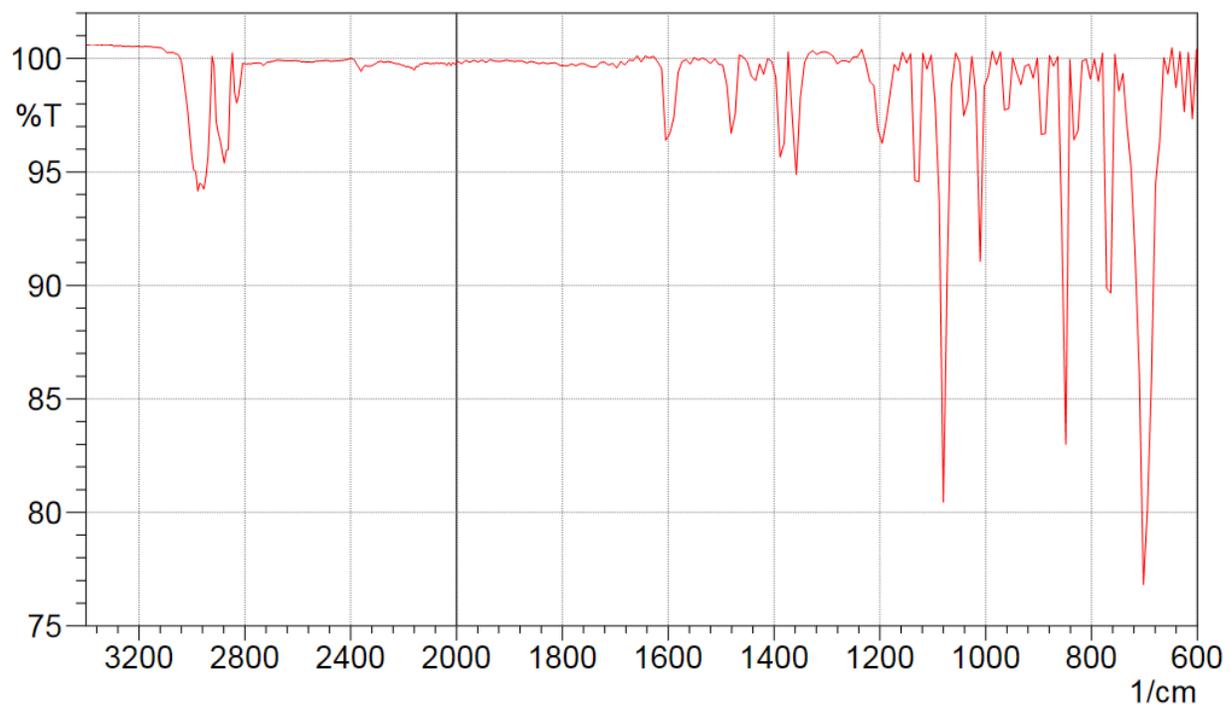


Figure S40. IR spectra of $\text{LiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ in the $3400\text{-}600\text{ cm}^{-1}$ range.

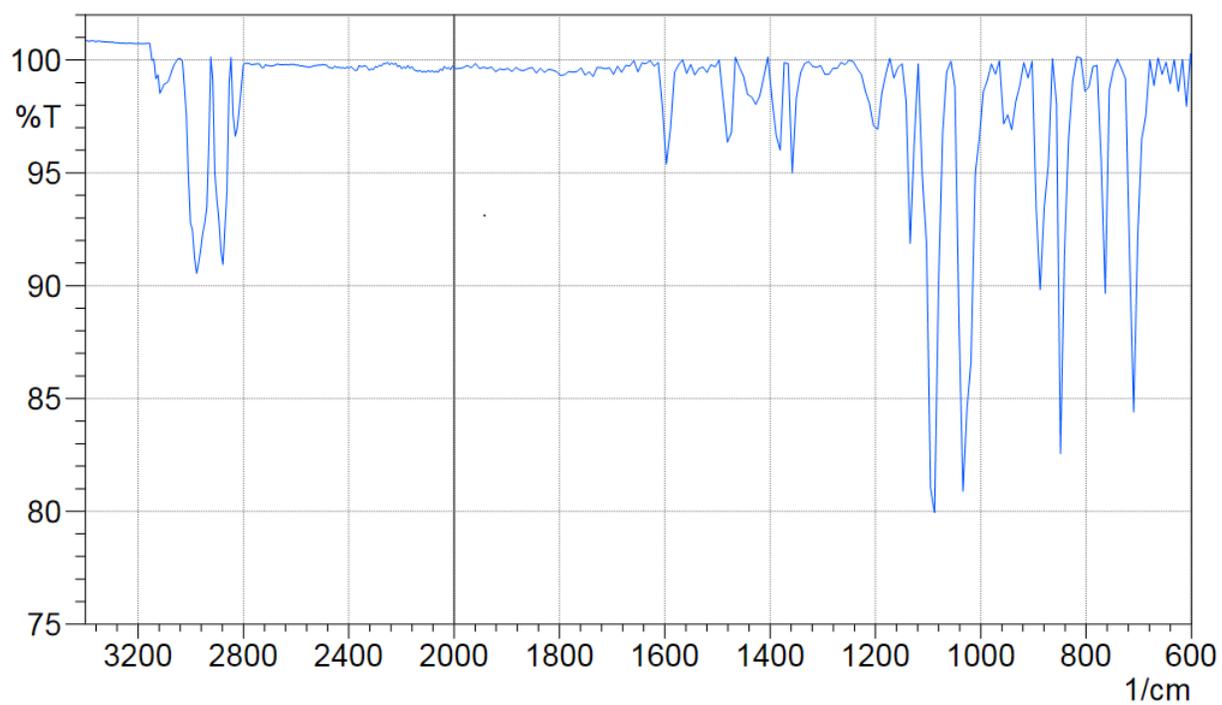


Figure S41. IR spectra of $\text{Fe}(\text{OC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3))_2(\text{THF})_2$ in the $3400\text{-}600\text{ cm}^{-1}$ range.

4. ORTEP plots of 1a and 1b demonstrating agostic interactions

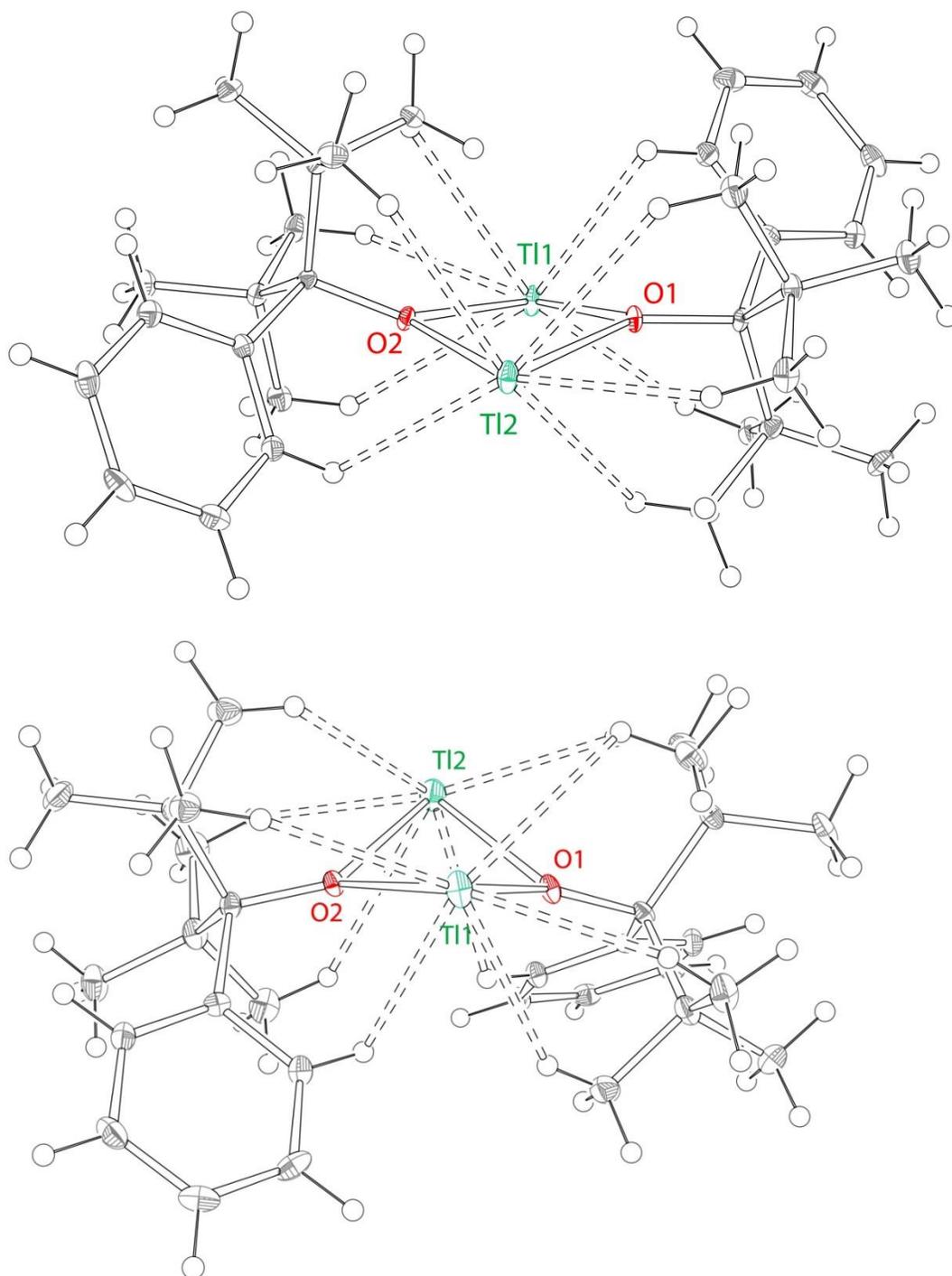


Figure S42. ORTEP diagram (50% probability ellipsoids) demonstrating agostic interactions for **1a** (top) and **1b** (bottom). Ti...H-C bond lengths are in 2.7-3.0 Å range.

5. X-ray structure of $\text{LiOC}^t\text{Bu}_2(3,5\text{-Me}_2\text{C}_6\text{H}_3)$ (**7**)

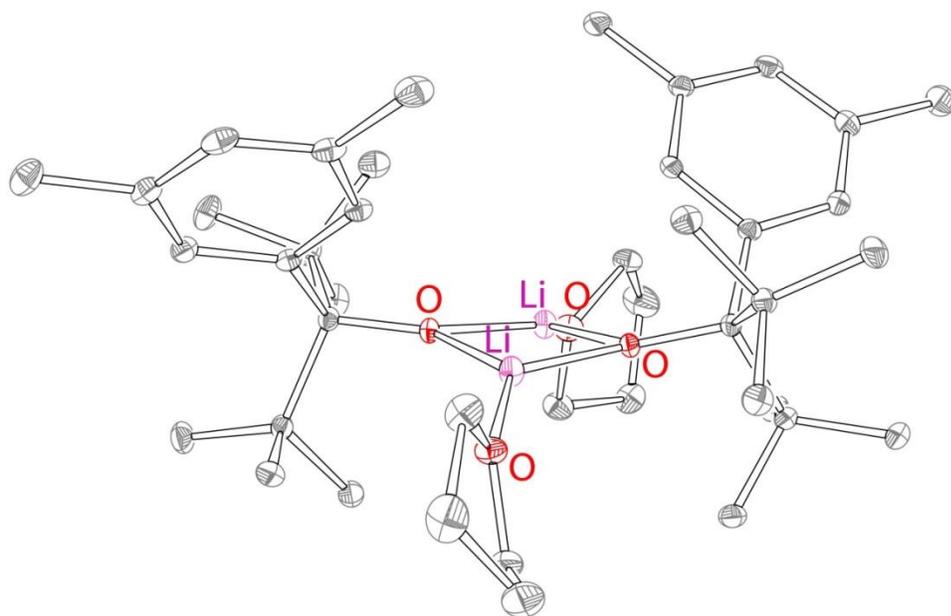


Figure S43. ORTEP diagram (50% probability ellipsoids) of the X-ray structure of **7**. H atoms were omitted for clarity.

6. Table S1. Crystal and structure refinement data.

	1a	1b	5	6
Formula	C ₃₀ H ₄₆ O ₂ Tl ₂	C ₃₀ H ₄₆ O ₂ Tl ₂	C ₃₄ H ₅₄ O ₃ Zn	C ₂₅ H ₃₆ ORu
Fw, g/mol	847.41	847.41	576.14	453.61
temperature	100(2) K	100(2) K	100(2) K	100(2) K
cryst syst	Triclinic	triclinic	monoclinic	triclinic
space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> -1
Colour	colorless	bright yellow	colorless	purple
Z	2	2	2	2
a, Å	7.9029(9)	8.0763(5)	10.0341(11)	9.3633(8)
b, Å	12.1061(14)	12.3094(7)	23.577(3)	11.1104(9)
c, Å	15.5684(18)	15.4883(9)	13.7200(17)	11.4477(9)
α, deg	88.456(4)	101.592(2)	90.00	91.329(4)
β, deg	80.664(4)	94.016(3)	106.953(4)	102.082(4)
γ, deg	74.856(4)	105.845(2)	90.00	112.349(4)
V, Å ³	1418.5(3)	1438.24(15)	3104.8(6)	1070.07(15)
d _{calcd} , g/cm ³	1.984	1.957	1.233	1.408
μ, mm ⁻¹	11.369	11.214	0.822	0.744
2θ, deg	55.98	50.00	60.06	56.74
R ₁ ^a (all data)	0.0224	0.0315	0.1088	0.0354
wR ₂ ^b (all data)	0.0392	0.0675	0.1009	0.0697
R ₁ ^a [(I>2σ)]	0.0176	0.0262	0.0477	0.0319
wR ₂ ^b [(I>2σ)]	0.0376	0.0651	0.0835	0.0682
GOF (F ²)	1.024	1.151	1.000	1.108

^a $R_1 = \sum ||F_o - |F_c|| / \sum |F_o|$. ^b $wR_2 = (\sum (w(F_o^2 - F_c^2)^2) / \sum (w(F_o^2)^2))^{1/2}$. ^c $GOF = (\sum w(F_o^2 - F_c^2)^2 / (n - p))^{1/2}$ where n is the number of data and p is the number of parameters refined.

Table S1. Crystal and structure refinement data (continued).

	7	8	9	10
Formula	C ₄₂ H ₇₀ O ₄ Li ₂	C ₃₄ H ₅₄ O ₂ Tl ₂	C ₄₂ H ₇₀ O ₄ Fe	C ₂₇ H ₄₀ ORu
Fw, g/mol	652.86	903.51	694.83	481.66
temperature	100(2) K	100(2) K	100(2) K	100(2) K
cryst syst	monoclinic	monoclinic	monoclinic	triclinic
space group	<i>P2₁/c</i>	<i>P2₁</i>	<i>P2₁/n</i>	<i>P-1</i>
Colour	colorless	colorless	colorless	violet
Z	4	2	4	2
a, Å	11.9523(13)	8.0476(3)	10.4095(5)	8.1664(4)
b, Å	21.215(3)	15.1435(6)	14.9890(7)	10.4284(5)
c, Å	15.9314(18)	13.6841(5)	25.9262(11)	14.7777(7)
α, deg	90.00	90.00	90.00	106.093(2)
β, deg	93.335(5)	98.9240(10)	96.890(2)	98.321(2)
γ, deg	90.00	90.00	90.00	93.777(2)
V, Å ³	4032.9(9)	1647.48(11)	4016.0(3)	1189.00(10)
d _{calcd} , g/cm ³	1.075	1.821	1.149	1.452
μ, mm ⁻¹	0.066	9.796	0.413	0.674
2θ, deg	55.84	55.14	49.70	61.18
R ₁ ^a (all data)	0.0589	0.0312	0.1070	0.0367
wR ₂ ^b (all data)	0.1126	0.0519	0.1063	0.0762
R ₁ ^a [(I>2σ)]	0.0422	0.0264	0.0516	0.0303
wR ₂ ^b [(I>2σ)]	0.1027	0.0503	0.0910	0.0734
GOF (F ²)	1.034	1.033	1.005	1.090

^a $R_1 = \sum ||F_o - |F_c|| / \sum |F_o|$. ^b $wR_2 = (\sum (w(F_o^2 - F_c^2)^2) / \sum (w(F_o^2)^2))^{1/2}$. ^c $GOF = (\sum w(F_o^2 - F_c^2)^2 / (n - p))^{1/2}$ where n is the number of data and p is the number of parameters refined.