

Electronic Supporting Information

Complexation of triangular silver(I) or copper(I) nitropyrazolates with dibenzothiophenes having potential use in adsorptive desulfurization

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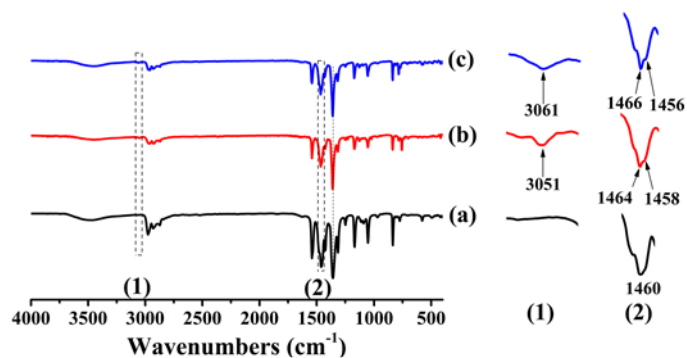


Fig. S1 IR spectra (KBr pellet) of (a) Ag_3pz_3 , (b) $\text{Ag}_3\text{pz}_3\cdot\text{DBT}$, (c) $\text{Ag}_3\text{pz}_3\cdot\text{DMDBT}$, (1) peaks at 3010-3090 cm^{-1} , (2) peaks at 1430-1500 cm^{-1} .

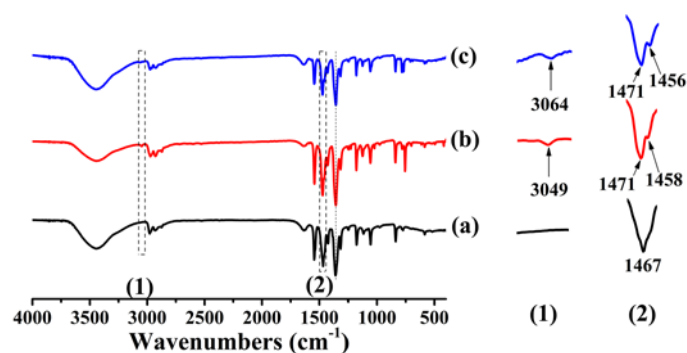


Fig. S2 IR spectra (KBr pellet) of (a) Cu_3pz_3 , (b) $\text{Cu}_3\text{pz}_3\cdot\text{DBT}$, (c) $\text{Cu}_3\text{pz}_3\cdot\text{DMDBT}$, (1) peaks at 3010-3100 cm^{-1} , (2) peaks at 1440-1500 cm^{-1} .

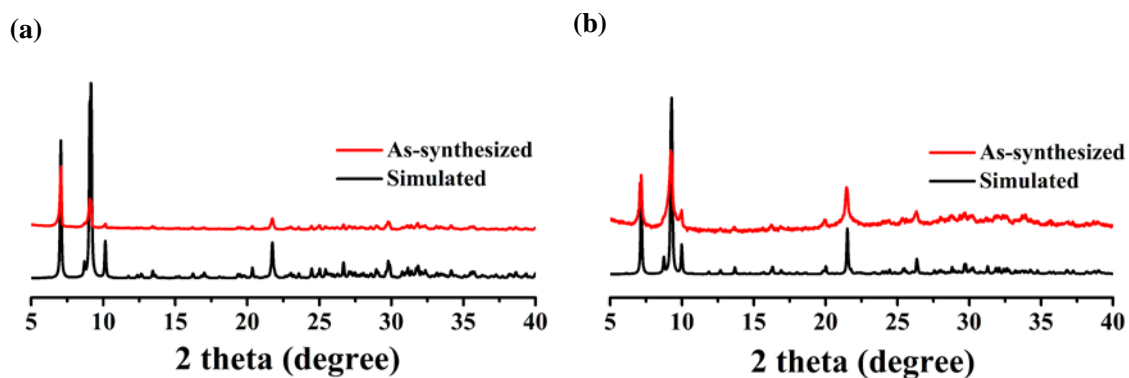


Fig. S3 Comparison of the PXR D pattern of the as-synthesized sample of Ag_3pz_3 (a) and Cu_3pz_3 (b) (red line) with the simulated pattern based on the crystal data (black line).

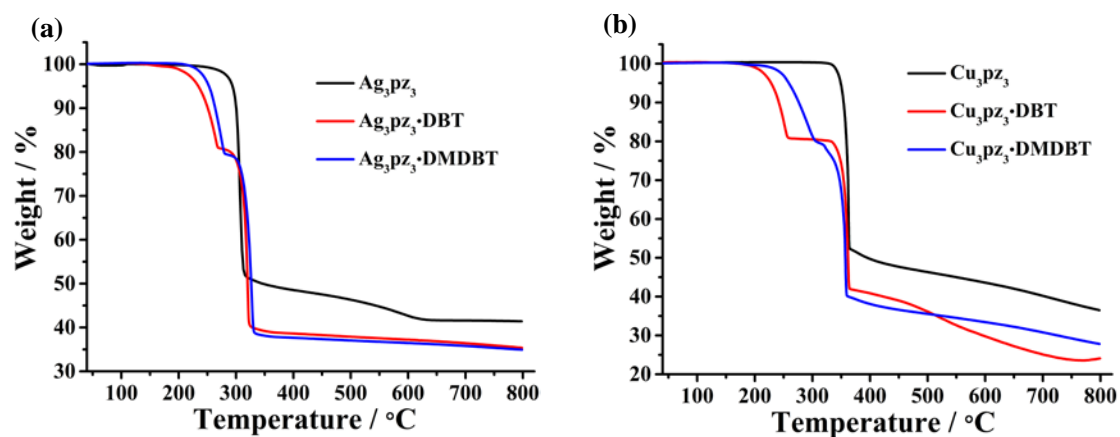


Fig. S4 Thermogravimetric curves of Ag_3pz_3 (a), Cu_3pz_3 (b) and their adducts with DBT/DMDBT.

Table S1. Thermogravimetric data of Ag_3pz_3 , Cu_3pz_3 and their adducts

	Weight loss (found)	Temperature (°C)	Weight loss (Calc.)	Assignment
$\text{Ag}_3\text{pz}_3 \cdot \text{DMDBT}$	20.5%	141 – 281	20.1%	–DMDBT
$\text{Ag}_3\text{pz}_3 \cdot \text{DBT}$	19.0%	130 – 270	18.2%	–DBT
$\text{Cu}_3\text{pz}_3 \cdot \text{DMDBT}$	21.0%	141 – 309	23.2%	–DMDBT
$\text{Cu}_3\text{pz}_3 \cdot \text{DBT}$	19.3%	132 – 260	20.8%	–DBT

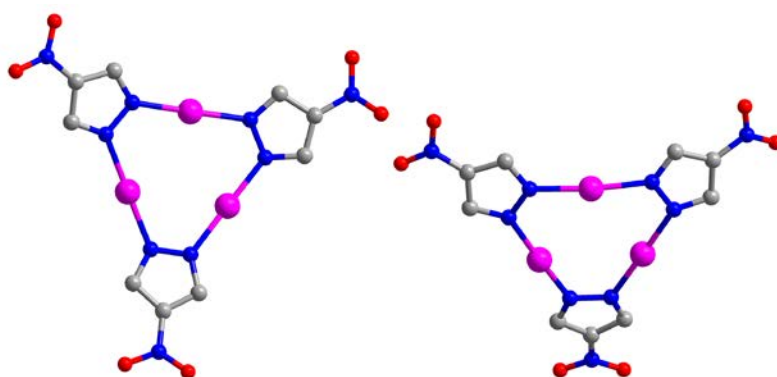


Fig. S5 The asymmetric unit of Ag_3pz_3 showing two crystallographically independent molecules. Colour code: Ag, pink; N, blue; C, gray and O, red. The diethyl groups and hydrogen atoms of pyrazole have been omitted in this and the following figures for clarity.

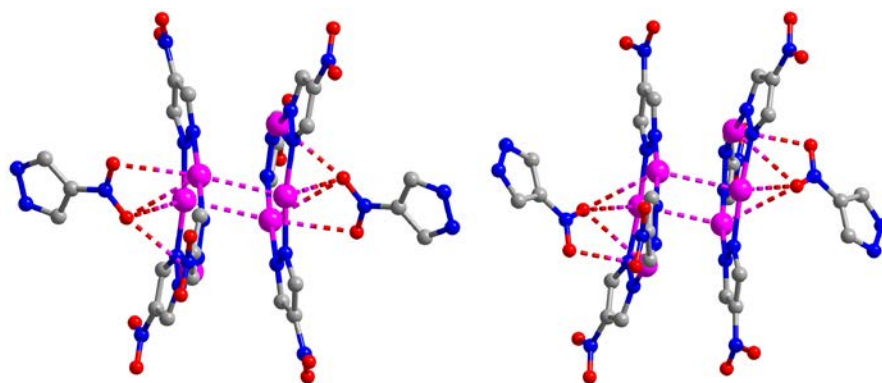


Fig. S6 The dimer of trimers of Ag_3pz_3 . Colour code: Ag, pink; N, blue; C, gray and O, red.

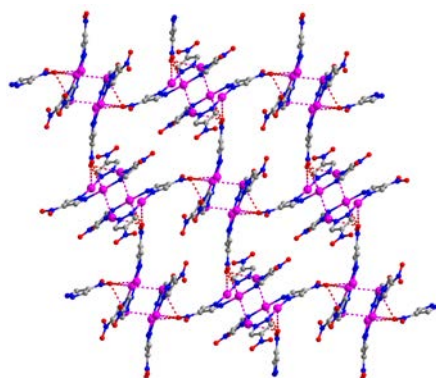


Fig. S7 A fragment of the 2D net in the crystal structure of Ag_3pz_3 . Colour code: Ag, pink; N, blue; C, gray and O, red.

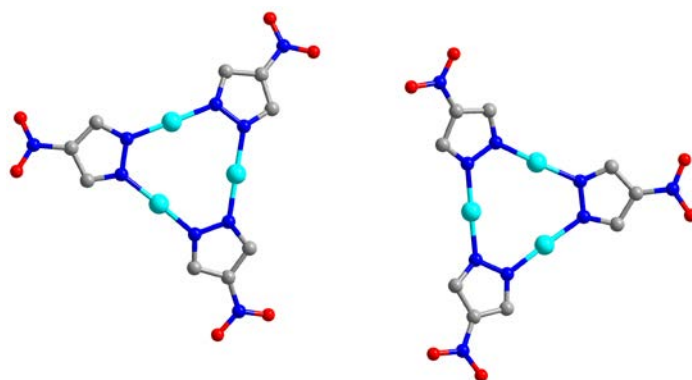


Fig. S8 The asymmetric unit of Cu_3pz_3 showing two crystallographically independent molecules. Colour code: Cu, turquoise; N, blue; C, gray and O, red.

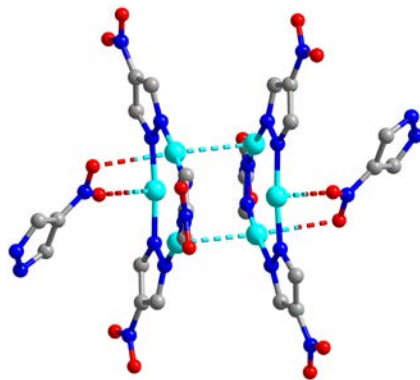


Fig. S9 The dimer of trimers of Cu_3pz_3 . Colour code: Cu, turquoise; N, blue; C, gray and O, red.

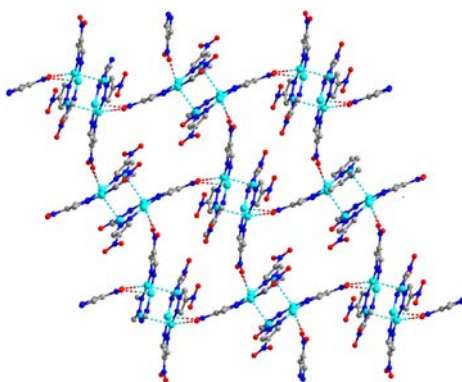


Fig. S10 A fragment of the 2D net in the crystal structure of Cu_3pz_3 . Colour code: Cu, turquoise; N, blue; C, gray and O, red.

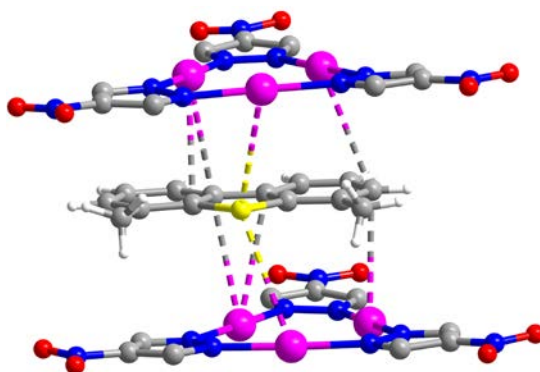


Fig. S11 A fragment of the columnar stacking of $\text{Ag}_3\text{pz}_3\cdot\text{DMDBT}$, showing $\text{Ag}\cdots\text{S}$ and $\text{Ag}\cdots\text{C}$ contacts. Ag, pink; N, blue; C, gray; H, white; O, red and S, yellow. The diethyl group of pyrazole have been omitted in this and the following figures for clarity.

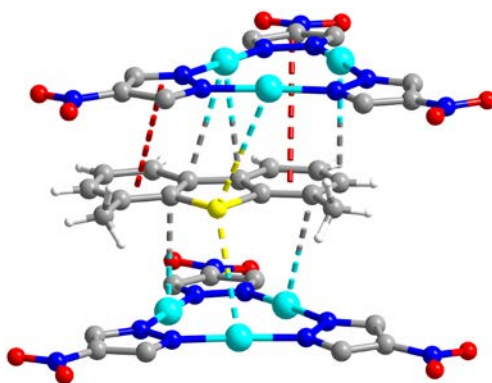


Fig. S12 A fragment of the columnar stacking of $\text{Cu}_3\text{pz}_3\cdot\text{DMDBT}$, showing $\text{Cu}\cdots\text{S}$, $\text{Cu}\cdots\text{C}$ contacts and $\pi\cdots\pi$ interactions. Colour code: Cu, turquoise; N, blue; C, gray; H, white; O, red and S, yellow.

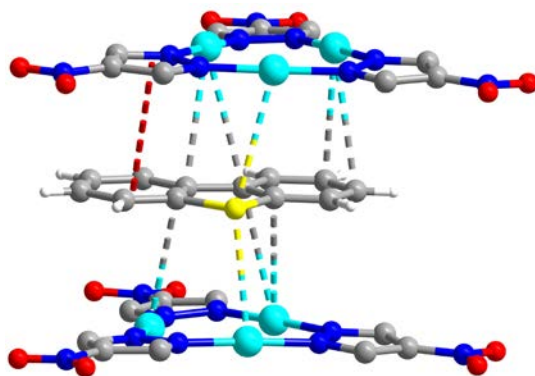


Fig. S13 A fragment of the columnar stacking of $\text{Cu}_3\text{pz}_3\cdot\text{DBT}$ showing $\text{Cu}\cdots\text{S}$, $\text{Cu}\cdots\text{C}$ contacts and $\pi\cdots\pi$ stacking interactions. Colour code: Cu, turquoise; N, blue; C, gray; H, white; O, red and S, yellow.

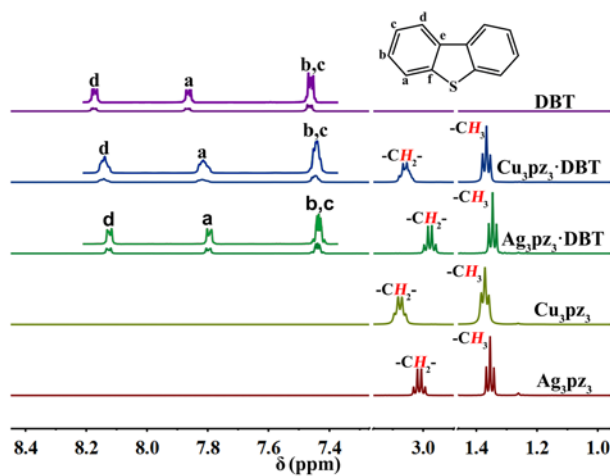
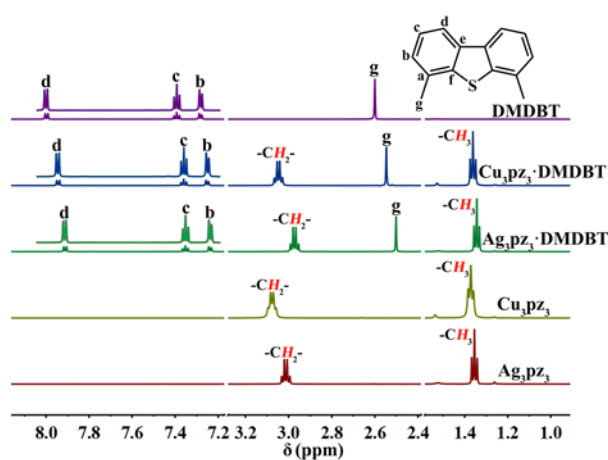


Fig. S14 Stack plot of ^1H NMR spectra for DBT, $\text{Cu}_3\text{pz}_3\cdot\text{DBT}$, $\text{Ag}_3\text{pz}_3\cdot\text{DBT}$, Cu_3pz_3 and Ag_3pz_3 in CD_2Cl_2 .

Table S2. Comparison of ^1H NMR of DBT, $\text{Cu}_3\text{pz}_3\cdot\text{DBT}$, $\text{Ag}_3\text{pz}_3\cdot\text{DBT}$, Cu_3pz_3 and Ag_3pz_3

^1H NMR/ppm	d	a	b,c	CH_2	CH_3
DBT	8.19 (dd)	7.89 (dd)	7.48 (m)		
Cu_3pz_3				3.08 (q)	1.37 (t)
$\text{Cu}_3\text{pz}_3\cdot\text{DBT}$	8.14 (t)	7.82 (dd)	7.45 (m)	3.07 (q)	1.37 (t)
Δ/ppm	0.05	0.07	0.03	0.01	0
Ag_3pz_3				3.01 (q)	1.36 (t)
$\text{Ag}_3\text{pz}_3\cdot\text{DBT}$	8.13 (dd)	7.80 (dd)	7.44 (m)	2.98 (q)	1.35 (t)
Δ/ppm	0.06	0.09	0.04	0.03	0.01

**Fig. S15** Stack plot of ^1H NMR spectra for DMDBT, $\text{Cu}_3\text{pz}_3\cdot\text{DMDBT}$, $\text{Ag}_3\text{pz}_3\cdot\text{DMDBT}$, Cu_3pz_3 and Ag_3pz_3 in CD_2Cl_2 .**Table S3.** Comparison of ^1H NMR of DMDBT, $\text{Cu}_3\text{pz}_3\cdot\text{DMDBT}$, $\text{Ag}_3\text{pz}_3\cdot\text{DMDBT}$, Cu_3pz_3 and Ag_3pz_3

	d	c	b	g	CH_2	CH_3
DMDBT	8.02 (d)	7.41 (t)	7.30 (d)	2.62 (s)		
Cu_3pz_3					3.08 (q)	1.37 (t)
$\text{Cu}_3\text{pz}_3\cdot\text{DMDBT}$	7.95 (d)	7.37 (t)	7.26 (d)	2.55 (s)	3.05 (q)	1.37 (t)
Δ/ppm	0.07	0.04	0.04	0.07	0.03	0
Ag_3pz_3					3.01 (q)	1.36 (t)
$\text{Ag}_3\text{pz}_3\cdot\text{DMDBT}$	7.92 (d)	7.36 (t)	7.24 (d)	2.51 (s)	2.98 (q)	1.35 (t)
Δ/ppm	0.1	0.05	0.06	0.11	0.03	0.01

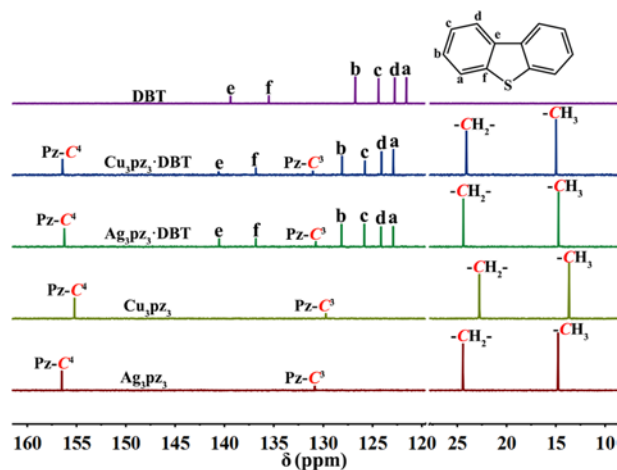


Fig. S16 Stack plot of ^{13}C NMR spectra for DBT, $\text{Cu}_3\text{pz}_3\cdot\text{DBT}$, $\text{Ag}_3\text{pz}_3\cdot\text{DBT}$, Cu_3pz_3 and Ag_3pz_3 in CD_2Cl_2 .

Table S4. Comparison of ^{13}C NMR of DBT, $\text{Cu}_3\text{pz}_3\cdot\text{DBT}$, $\text{Ag}_3\text{pz}_3\cdot\text{DBT}$, Cu_3pz_3 and Ag_3pz_3 .

^{13}C NMR/ppm	e	f	b	c	d	a	pz- C^3	pz- C^4	CH_2	CH_3
DBT	139.81	135.92	127.18	124.83	123.18	122.00				
Cu_3pz_3							130.17	155.64	23.19	14.09
$\text{Cu}_3\text{pz}_3\cdot\text{DBT}$	139.71	135.91	127.21	124.87	123.19	121.99	130.14	155.52	23.17	14.08
Δ/ppm	0.1	0.01	-0.03	-0.04	-0.01	0.01	0.03	0.12	0.02	0.01
Ag_3pz_3							129.94	155.60	23.51	13.87
$\text{Ag}_3\text{pz}_3\cdot\text{DBT}$	139.63	135.91	127.25	124.93	123.24	122.02	129.86	155.34	23.48	13.83
Δ/ppm	0.18	0.01	-0.07	-0.1	-0.06	-0.02	0.08	0.26	0.03	0.04

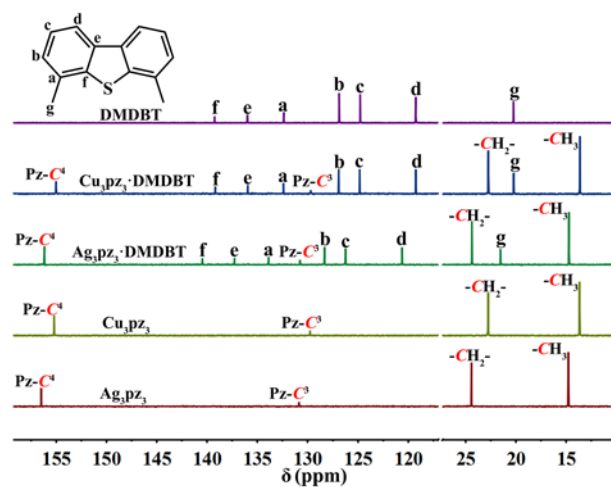


Fig. S17 Stack plot of ^{13}C NMR spectra for DMDBT, $\text{Cu}_3\text{pz}_3\cdot\text{DMDBT}$, $\text{Ag}_3\text{pz}_3\cdot\text{DMDBT}$, Cu_3pz_3 and Ag_3pz_3 in CD_2Cl_2 .

Table S5. Comparison of ^{13}C NMR of DMDBT, Cu_3pz_3 -DMDBT, Ag_3pz_3 -DMDBT, Cu_3pz_3 and Ag_3pz_3 .

^{13}C NMR/ppm	f	e	a	b	c	d	g	pz- C^3	pz- C^4	CH_2	CH_3
DMDBT	139.67	136.40	132.78	127.32	125.22	119.69	20.67				
Cu_3pz_3								130.17	155.64	23.19	14.09
Cu_3pz_3 -DMDBT	139.60	136.37	132.82	127.36	125.27	119.69	20.65	130.11	155.45	23.17	14.06
Δ /ppm	0.07	0.03	-0.04	-0.04	-0.05	0	0.02	0.06	0.19	0.02	0.03
Ag_3pz_3								129.94	155.60	23.51	13.87
Ag_3pz_3 -DMDBT	139.56	136.36	132.97	127.44	125.35	119.72	20.63	129.84	155.27	23.47	13.82
Δ /ppm	0.11	0.04	-0.19	-0.12	-0.13	-0.03	0.04	0.1	0.33	0.04	0.05

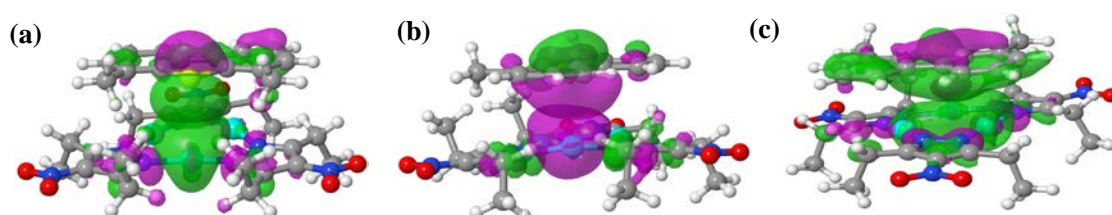


Fig. S18 The three most important MBFO pairs for the interactions between Ag_3pz_3 and DMBT. The bond order for the MBFO pair is 0.050, 0.019 and 0.015 in (a), (b) and (c), respectively. The orbitals are shown with an isovalue of 0.03.

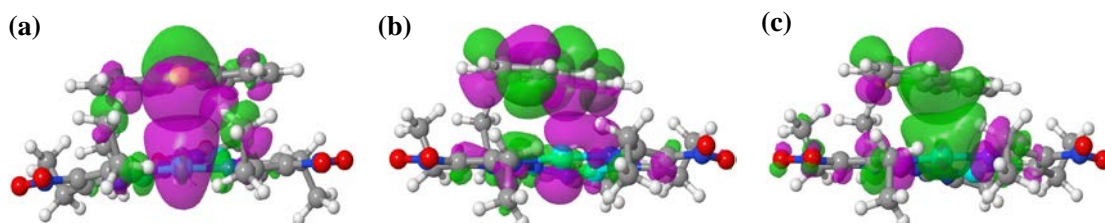


Fig. S19 The three most important MBFO pairs for the interactions between Cu_3pz_3 and DBT. The bond order for the MBFO pair is 0.029, 0.024 and 0.023 in (a), (b) and (c), respectively. The orbitals are shown with an isovalue of 0.03.

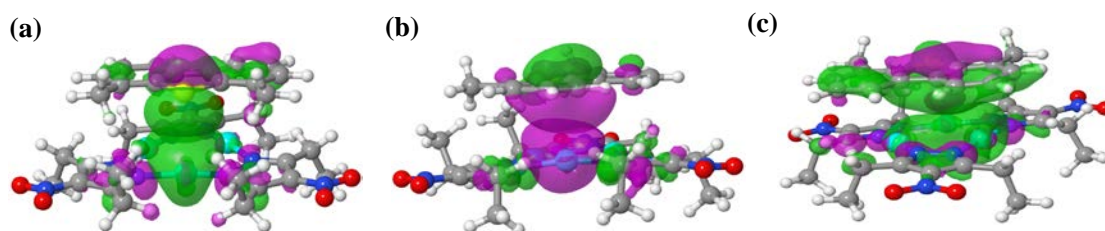


Fig. S20 The three most important MBFO pairs for the interactions between Cu_3pz_3 and DMBT. The bond order for the MBFO pair is 0.031, 0.017 and 0.014 in (a), (b) and (c), respectively. The orbitals are shown with an isovalue of 0.03.

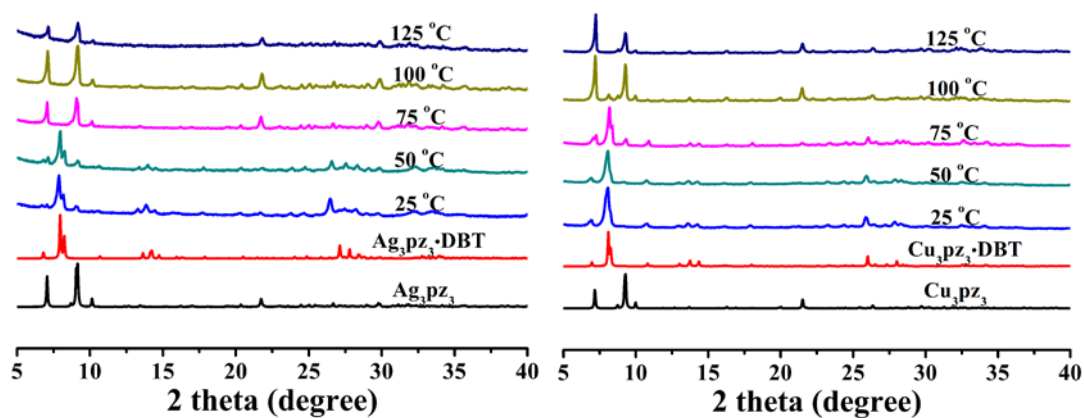


Fig. S21 Comparison of the PXRD patterns of DBT-adsorbed samples of $\text{Ag}_3\text{pz}_3/\text{Cu}_3\text{pz}_3$ at different temperatures with those simulated from their crystal data (red and black line). (Experimental condition: 0.02 mmol M_3pz_3 in 10 mL of *n*-octane containing 1000 mg S/L of DBT with the molar ratio of S: $\text{M}_3\text{pz}_3 = 15.6$)

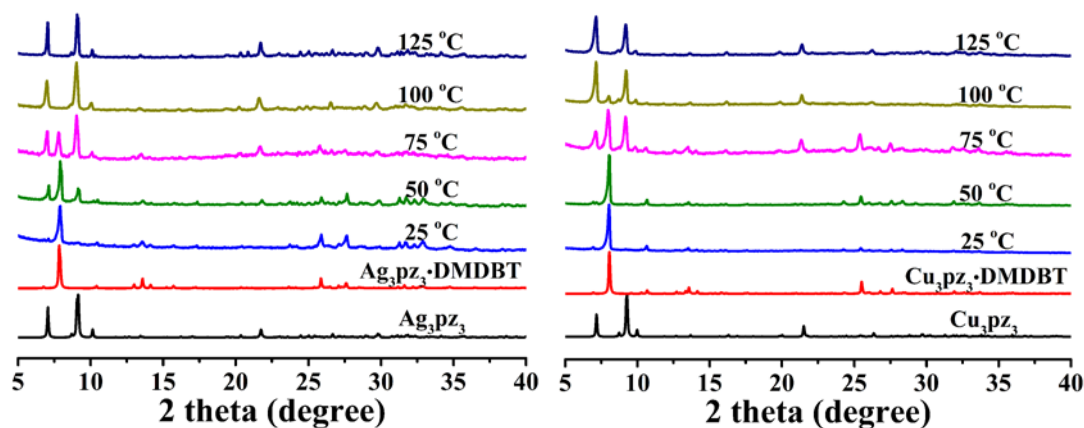


Fig. S22 Comparison of the PXRD patterns of DMBDT-adsorbed samples of $\text{Ag}_3\text{pz}_3/\text{Cu}_3\text{pz}_3$ at different temperatures with those simulated from their crystal data (red and black line). (Experimental condition: 0.02 mmol M_3pz_3 in 10 mL of *n*-octane containing 800 mg S/L of DMBDT with the molar ratio of S: $\text{M}_3\text{pz}_3 = 12.5$)

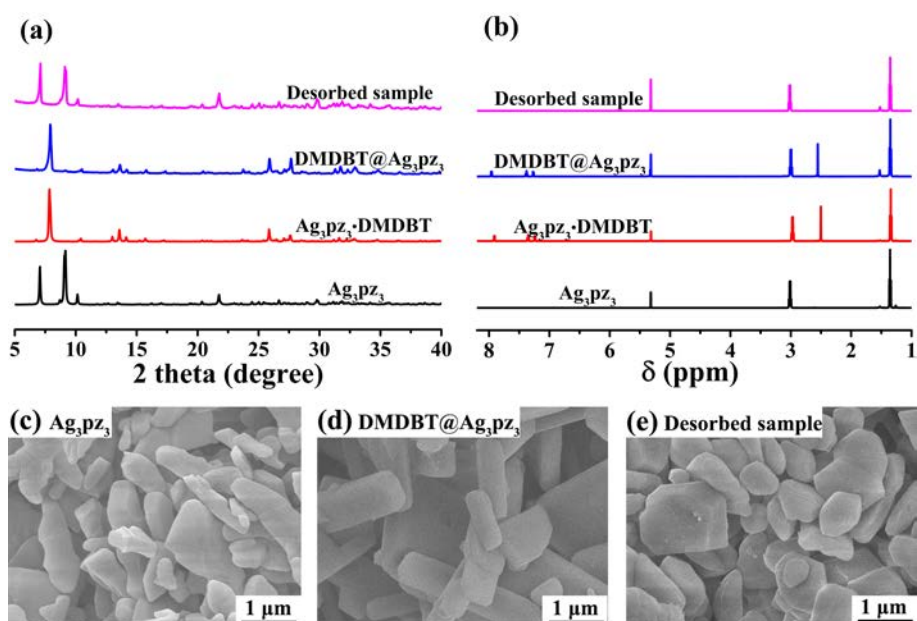


Fig. S23 PXR patterns (a), ¹H NMR spectra (b) and SEM images (c,d,e) of the DMDBT-adsorbed and desorbed samples of Ag₃pz₃ in comparison with those of Ag₃pz₃ or Ag₃pz₃·DMDBT.

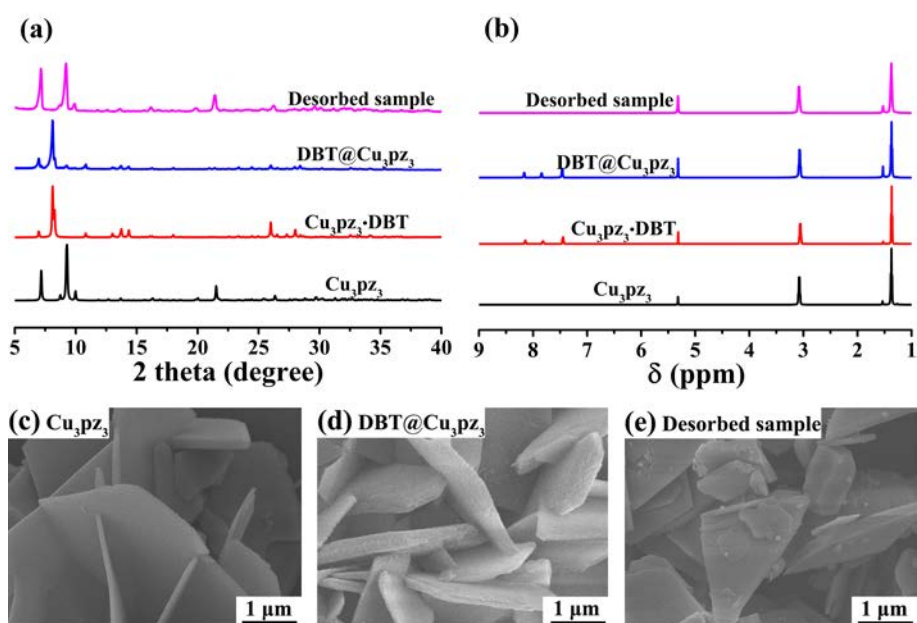


Fig. S24 PXR patterns (a), ¹H NMR spectra (b) and SEM images (c,d,e) of the DBT-adsorbed and desorbed samples of Cu₃pz₃ in comparison with those of Cu₃pz₃ or Cu₃pz₃·DBT.

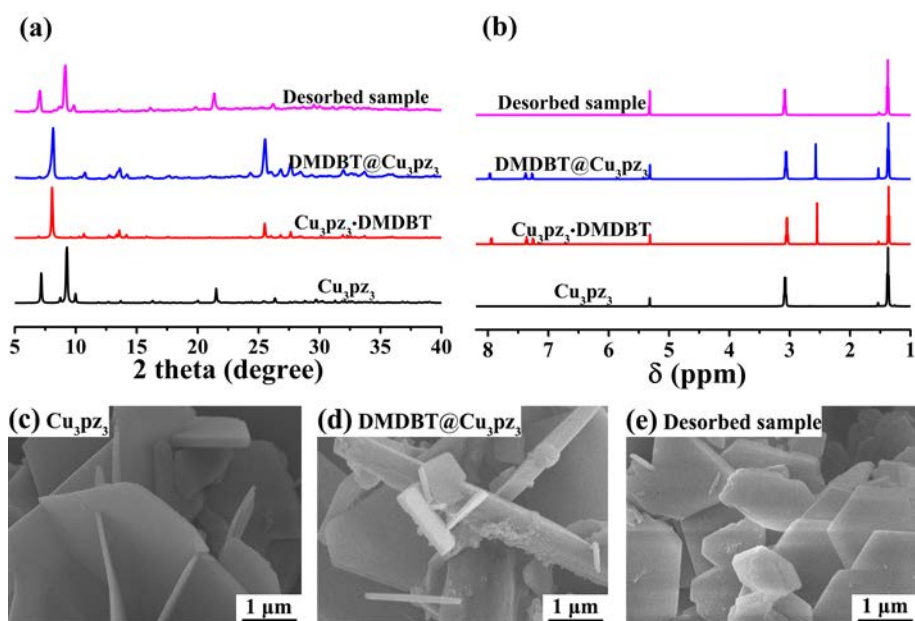


Fig. S25 PXR patterns (a), ¹H NMR spectra (b) and SEM images (c,d,e) of the DMDBT-adsorbed and desorbed samples of Cu₃pz₃ in comparison with those of Cu₃pz₃ or Cu₃pz₃-DMDBT.

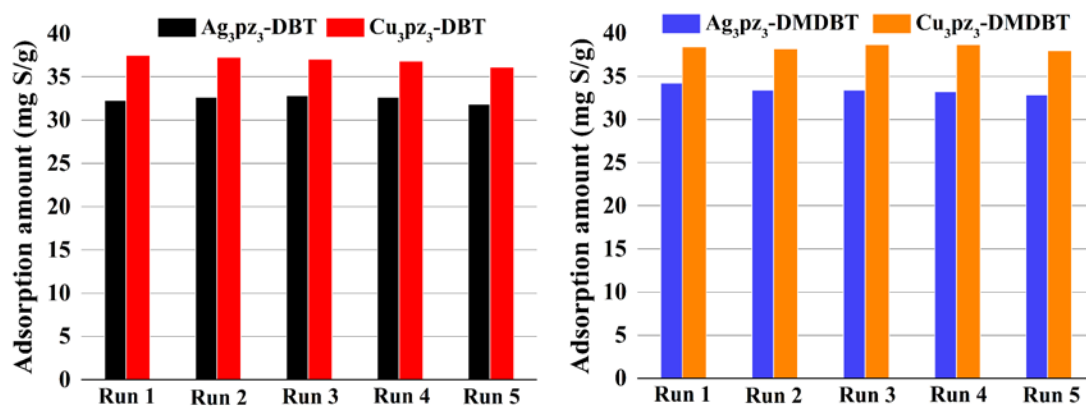


Fig. S26 Recycling test of Ag₃pz₃ (0.02 mmol) and Cu₃pz₃ (0.02 mmol) for the adsorption of DBT and DMDBT from *n*-octane containing 1000 mg S/L of DBT or 800 mg S/L of DMDBT (10 mL).

Table S6 Crystallographic data and structure refinements

	Ag₃pz₃	Ag₃pz₃·DBT	Ag₃pz₃·DMDBT	(Ag₃pz₃)₂·BT
Formula	C ₄₂ H ₆₀ Ag ₆ N ₁₈ O ₁₂	C ₃₃ H ₃₈ Ag ₃ N ₉ O ₆ S	C ₃₅ H ₄₂ Ag ₃ N ₉ O ₆ S	C ₅₀ H ₆₆ Ag ₆ N ₁₈ O ₁₂ S
Mol. wt.	1656.30	1012.39	1040.44	1790.48
Crystal system	Monoclinic	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P2₁/c</i>	<i>P2₁/c</i>	<i>P2₁/c</i>	<i>C2/c</i>
<i>a</i> (Å)	25.0348(10)	13.1758(2)	13.2738(6)	19.2517(4)
<i>b</i> (Å)	17.4283(7)	21.4602(4)	22.3282(7)	24.1571(5)
<i>c</i> (Å)	13.2012(5)	13.1785(3)	13.8197(6)	14.3647(3)
α (deg)				
β (deg)	91.0170(10)	100.0406(18)	100.078(4)	105.5782(8)
γ (deg)				
<i>V</i> (Å ³)	5759.0(4)	3669.21(13)	4032.7(3)	6435.1(2)
<i>Z</i>	4	4	4	4
<i>T</i> (K)	298.15(10)	120(10)	298.15(10)	298.15(10)
$\rho_{\text{calcd.}}$ (Mg/m ³)	1.910	1.833	1.714	1.848
μ (mm ⁻¹)	2.071	1.699	1.549	1.893
Reflns collected	73119	27145	28415	41721
Reflns unique	10151	6466	7123	5677
	(<i>R</i> _{int} = 0.0723)	(<i>R</i> _{int} = 0.0964)	(<i>R</i> _{int} = 0.0532)	(<i>R</i> _{int} = 0.0757)
Final <i>R</i> indices	<i>R</i> ₁ = 0.0404	<i>R</i> ₁ = 0.0366	<i>R</i> ₁ = 0.0626	<i>R</i> ₁ = 0.0480
	<i>wR</i> ₂ = 0.0825	<i>wR</i> ₂ = 0.0797	<i>wR</i> ₂ = 0.1762	<i>wR</i> ₂ = 0.1056
[<i>I</i> > 2 σ (<i>I</i>)]				
<i>S</i>	1.026	1.030	1.067	1.062
$\Delta\rho_{\text{max}}$ (eÅ ⁻³)	1.20	1.10	0.79	0.69
$\Delta\rho_{\text{min}}$ (eÅ ⁻³)	-0.88	-1.35	-0.93	-0.66

Table S7 Crystallographic data and structure refinements

	Cu₃pz₃	Cu₃pz₃·DBT	Cu₃pz₃·DMDBT
Formula	C ₂₁ H ₃₀ Cu ₃ N ₉ O ₆	C ₃₃ H ₃₈ Cu ₃ N ₉ O ₆ S	C ₃₅ H ₄₂ Cu ₃ N ₉ O ₆ S
Mol. wt.	695.16	879.40	907.45
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P2₁/c</i>	<i>P2₁/c</i>	<i>P2₁/c</i>
<i>a</i> (Å)	24.6486(15)	12.8446(18)	12.8398(6)
<i>b</i> (Å)	17.7157(10)	21.390(3)	21.8537(10)
<i>c</i> (Å)	12.7565(7)	13.7791(19)	14.0702(6)
α (deg)			
β (deg)	90.671(2)	99.202(4)	98.4970(10)
γ (deg)			
<i>V</i> (Å ³)	5570.0(6)	3737.0(9)	3904.7(3)
<i>Z</i>	8	4	4
<i>T</i> (K)	298.15(10)	298.15(10)	298.15(10)
$\rho_{\text{calcd.}}$ (Mg/m ³)	1.658	1.563	1.544
μ (mm ⁻¹)	2.323	1.803	1.728
Reflns collected	69497	45165	61638
Reflns unique	9813	6597	6899
	(<i>R</i> _{int} = 0.0426)	(<i>R</i> _{int} = 0.0466)	(<i>R</i> _{int} = 0.0352)
Final <i>R</i> indices	<i>R</i> ₁ = 0.0519	<i>R</i> ₁ = 0.0500	<i>R</i> ₁ = 0.0357
[<i>I</i> > 2σ(<i>I</i>)]	<i>wR</i> ₂ = 0.1267	<i>wR</i> ₂ = 0.1300	<i>wR</i> ₂ = 0.0844
<i>S</i>	1.059	1.022	1.019
$\Delta\rho_{\text{max}}$ (eÅ ⁻³)	0.97	0.62	0.67
$\Delta\rho_{\text{min}}$ (eÅ ⁻³)	-0.65	-0.47	-0.52

Table S8 Selected bond distances (Å) and bond angles (°).

Ag₃pz₃			
Ag(1)-N(1)	2.095(5)	Ag(1)-N(6)	2.091(5)
Ag(2)-N(2)	2.097(5)	Ag(2)-N(3)	2.084(5)
Ag(3)-N(4)	2.073(5)	Ag(3)-N(5)	2.082(5)
Ag(4)-N(10)	2.058(5)	Ag(4)-N(15)	2.067(5)
Ag(5)-N(11)	2.085(5)	Ag(5)-N(12)	2.087(5)
Ag(6)-N(13)	2.093(5)	Ag(6)-N(14)	2.077(5)
Ag(1)⋯Ag(3) ^{#1}	3.0950(7)	Ag(4)⋯Ag(6) ^{#2}	3.1172(7)
Ag(1)⋯O(1) ^{#1}	3.0346(53)	Ag(1)⋯O(2) ^{#1}	3.0583(62)
Ag(2)⋯O(2) ^{#1}	2.9467(58)	Ag(3)⋯O(2) ^{#1}	3.1832(67)
Ag(4)⋯O(10)	3.1947(69)	Ag(5)⋯O(10)	3.0791(71)
Ag(5)⋯O(9)	2.9765(53)	Ag(6)⋯O(10)	2.9633(71)
N(6)-Ag(1)-N(1)	177.41(18)	N(3)-Ag(2)-N(2)	173.7(2)
N(4)-Ag(3)-N(5)	172.7(2)	N(10)-Ag(4)-N(15)	174.4(2)
N(11)-Ag(5)- N(12)	176.1(2)	N(14)-Ag(6)-N(13)	175.4(2)
^{#1} 1-x, 1-y, -z; ^{#2} -x, 2-y, -z			

Ag₃pz₃·DBT			
Ag(1)-N(1)	2.084(3)	Ag(1)-N(6)	2.087(3)
Ag(2)-N(2)	2.104(3)	Ag(2)-N(3)	2.082(3)
Ag(3)-N(4)	2.068(3)	Ag(3)-N(5)	2.073(3)
Ag(1)⋯S(1) [#]	3.3752(11)	Ag(3)⋯S(1)	3.4064(11)
Ag(1)⋯C(28)	3.3284(40)	Ag(2)⋯C(26)	3.3155(39)
Ag(2)⋯C(26) [#]	3.6627(41)	Ag(2)⋯C(25) [#]	3.5113(39)
Ag(2)⋯C(25)	3.2411(42)	Ag(3)⋯C(28)	3.2855(40)
N(1)-Ag(1)-N(6)	175.42(11)	N(3)-Ag(2)-N(2)	177.15(11)
N(4)-Ag(3)-N(5)	179.48(11)		
[#] x, 1.5-y, -0.5+z			

Ag₃pz₃·DMDBT			
Ag(1)-N(1)	2.081(6)	Ag(1)-N(6)	2.075(6)
Ag(2)-N(2)	2.076(6)	Ag(2)-N(3)	2.076(6)
Ag(3)-N(4)	2.076(6)	Ag(3)-N(5)	2.076(6)
Ag(1)⋯S(1)	3.8077(21)	Ag(2)⋯S(1)	3.6251(21)
Ag(1)⋯C(27)	3.5101(70)	Ag(1)⋯C(22)	3.3629(74)
Ag(2)⋯C(27) [#]	3.4680(71)	Ag(2)⋯C(28) [#]	3.5070(77)
Ag(3)⋯C(31)	3.5654(95)	Ag(3)⋯C(31) [#]	3.4990(103)
N(6)-Ag(1)-N(1)	177.9(2)	N(3)-Ag(1)-N(2)	176.6(2)
N(4)-Ag(3)-N(5)	178.8(3)		
[#] x, 0.5-y, 0.5+z			

(Ag₃pz₃)₂·BT			
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Ag(1)-N(1)	2.071(5)	Ag(1)-N(6)	2.067(5)
Ag(2)-N(2)	2.068(5)	Ag(2)-N(3)	2.076(6)
Ag(3)-N(4)	2.092(5)	Ag(3)-N(5)	2.086(5)
Ag(1)···Ag(2) [#]	3.091(7)	Ag(2)···C(23A)	3.468(273)
Ag(2)···C(24A)	3.400(311)	Ag(1)···C(25A)	3.467(263)
Ag(1)···C(26A)	3.481(274)	Ag(3)···C(28B)	3.534(228)
Ag(3)···C(27B)	3.395(170)	Ag(2)···C(23B)	3.390(221)
Ag(1)···C(25B)	3.153(225)	Ag(2)···S(1A)	3.468(273)
Ag(1)···C(28A)	3.322(329)	Ag(3)···C(26A)	3.482(321)
Ag(2)···S(1B)	3.326(175)	Ag(1)···C(28B)	3.412(240)
Ag(3)···C(25B)	3.701(218)	Ag(3)···C(24B)	3.717(221)
N(6)-Ag(1)-N(1)	175.9(2)	N(2)-Ag(2)-N(3)	177.3(2)
N(5)-Ag(3)-N(4)	177.9(2)		

[#] 0.5-x, 0.5-y, 1-z

Cu₃pz₃

Cu(1)-N(1)	1.858(4)	Cu(1)-N(6)	1.855(4)
Cu(2)-N(2)	1.870(5)	Cu(2)-N(3)	1.861(5)
Cu(3)-N(4)	1.850(5)	Cu(3)-N(5)	1.850(4)
Cu(4)-N(10)	1.854(4)	Cu(4)-N(15)	1.854(4)
Cu(5)-N(11)	1.861(4)	Cu(5)-N(13)	1.867(4)
Cu(6)-N(13)	1.862(4)	Cu(6)-N(14)	1.869(4)
Cu(2)···Cu(3) ^{#1}	3.0199(11)	Cu(4)···Cu(6) ^{#2}	2.9349(10)
Cu(1)···O(1)	2.9699(53)	Cu(2)···O(2)	2.9529(79)
Cu(5)···O(9)	2.8145(78)	Cu(6)···O(10)	3.0976(53)
N(6)-Cu(1)-N(1)	176.5(2)	N(3)-Cu(2)-N(2)	174.5(2)
N(5)-Cu(3)-N(4)	173.4(2)	N(15)-Cu(4)-N(10)	171.8(2)
N(11)-Cu(5)-N(12)	175.6(2)	N(13)-Cu(6)-N(14)	174.61(19)

^{#1} 1-x, 1-y, -z; ^{#2} -x, 1-y, -1-z

Cu₃pz₃·DBT

Cu(1)-N(1)	1.861(4)	Cu(1)-N(6)	1.855(4)
Cu(2)-N(2)	1.859(4)	Cu(2)-N(3)	1.854(3)
Cu(3)-N(4)	1.854(4)	Cu(3)-N(5)	1.854(4)
Cu(2)···S(1) [#]	3.2871(19)	Cu(3)···S(1)	3.5658(18)
Cu(1)···C(24)	3.5669(43)	Cu(1)···C(25)	3.5136(42)
Cu(1)···C(22) [#]	3.5040(34)	Cu(1)···C(27) [#]	3.4207(33)
Cu(2)···C(27)	3.5889(33)	Cu(2)···C(28)	3.5128(42)
Cu(3)···C(33) [#]	3.4782(43)	N(6)-Cu(1)-N(1)	178.16(18)
N(3)-Cu(2)-N(2)	176.16(16)	N(5)-Cu(3)-N(4)	178.88(18)

[#] x, 0.5-y, -0.5+z

Cu₃pz₃·DMDBT

Cu(1)-N(6)	1.854(2)	Cu(1)-N(1)	1.852(2)
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Cu(2)-N(2)	1.854(2)	Cu(2)-N(3)	1.854(3)
Cu(3)-N(4)	1.845(3)	Cu(3)-N(5)	1.851(3)
Cu(1)···S(1)	3.5872(10)	Cu(3)···S(1) [#]	3.9414(10)
Cu(1)···C(27) [#]	3.4863(32)	Cu(1)···C(28) [#]	3.6231(29)
Cu(2)···C(24) [#]	3.5847(44)	Cu(2)···C(23)	3.4631(31)
Cu(3)···C(29)	3.3329(29)	N(1)-Cu(1)-N(6)	176.98(11)
N(2)-Cu(2)-N(3)	177.20(12)	N(4)-Cu(3)-N(5)	177.93(11)
[#] x, 0.5-y, 0.5+z			