## Supporting Information

# Metal-Free Visible Light-Promoted Synthesis of Isothiazoles: a Catalytic Approach for N-S Bond Formation from Iminyl Radicals under Batch and Flow 

## Conditions

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## Table of Contents

1. General Information ..... S3
2. Synthesis and Characterization of Substrates ..... S5
2.1. General Procedure and Characterization of Sulfides SI-1 ..... S5
2.2. General Procedure and Characterization of Selenides SI-10 ..... S14
2.3. General Procedure and Characterization of $\alpha$-Imino-oxy Acids $\mathbf{1}$ ..... S15
3. Optimization of the Reaction Conditions ..... S26
4. Optimization of the $\alpha$-Imino-oxy Acid Structure ..... S28
5. General Procedure and Characterization of Isothiazoles 2 ..... S29
6. Synthesis of Brassilexin Derivative 4 ..... S35
7. Flow Setup for the Synthesis of Isothiazoles 2 ..... S37
8. Mechanistic Studies ..... S39
8.1. Cyclic Voltammetry of $\alpha$-Imino-oxy Acids 1 ..... S39
8.2. Fluorescence Quenching Studies ..... S42
8.3. Radical Trapping Experiments ..... S44
9. References ..... S45
10. NMR Spectra ..... S47
10.1. NMR Spectra of Sulfides SI-1 ..... S47
10.2. NMR Spectra of Selenides $\mathbf{S I - 1 0}$ ..... S64
10.3. NMR Spectra of $\alpha$-Imino-oxy Acids 1 ..... S65
10.4. NMR Spectra of Isothiazoles 2 ..... S90
10.5. NMR Spectra of Brassilexin Derivative 4 ..... S105

## 1. General Information

Reaction temperatures are reported as the ones of the heat transfer medium surrounding the vessel unless otherwise stated.

The following solvents DCM, MeCN, PhMe and THF were purified by passing through a Pure SolvTM column drying system from Innovative Technology, Inc. Additional anhydrous solvents ( $\leq 50 \mathrm{ppm}$ water) were purchased from Acros Organics, Sigma-Aldrich or Alfa Aesar and stored over molecular sieves under a nitrogen atmosphere. Commercially available chemicals were obtained from Acros Organics, Aldrich Chemical Co., Alfa Aesar, TCI Chemicals and Fluorochem, and used as received, unless otherwise stated.

Analytical thin layer chromatography (TLC) was performed on silica gel 60 F254 aluminum plates (Merck) and they were visualized by exposure to short wave ultraviolet light ( $254 \mathrm{~nm}, 366 \mathrm{~nm}$ ) and/or by staining. For staining the TLC plates were dipped into a solution of $\mathrm{KMnO}_{4}\left(1 \mathrm{~g} \mathrm{KMnO}_{4}, 6 \mathrm{~g} \mathrm{~K}_{2} \mathrm{CO}_{3}\right.$ and 0.1 g KOH in $\left.100 \mathrm{~mL} \mathrm{H}_{2} \mathrm{O}\right)$ and developed with a heat gun if necessary. Flash chromatography was performed using Geduran® Silica Gel $60(0.040-0.063 \mathrm{~nm})$ or latrobeads 6RS -8060 silica gel with appropriate mixtures of cyclohexane and ethyl acetate and compressed air.

NMR spectra were recorded at room temperature on a Bruker AV 300 or or 500 MHz spectrometer running at 300 or 500 MHz for ${ }^{1} \mathrm{H}, 75$ or 126 MHz for ${ }^{13} \mathrm{C}, 282$ or 471 MHz for ${ }^{19} \mathrm{~F}$ in solvents as indicated. Chemical shifts ( $\delta$ ) for ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra are given in ppm relative to tetramethylsilane (TMS) using the residual solvent signals as references for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra ( $\mathrm{CDCl}_{3}: \delta_{\mathrm{H}}=7.26 \mathrm{ppm}, \delta_{\mathrm{C}}=77.16$ ppm, $\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}: \delta_{\mathrm{H}}=2.05 \mathrm{ppm}, \delta_{\mathrm{C}}=29.84,206.26 \mathrm{ppm}, \mathrm{C}_{6} \mathrm{D}_{6}: \delta_{\mathrm{H}}=7.16 \mathrm{ppm}, \delta_{\mathrm{C}}=$ 134.19, 129.26, 128.25, $\left.125.96 \mathrm{ppm}, \mathrm{CD}_{2} \mathrm{Cl}_{2}: \delta_{\mathrm{H}}=5.32 \mathrm{ppm}, \delta_{\mathrm{C}}=53.84 \mathrm{ppm}\right) .{ }^{[1]}$ ${ }^{19} \mathrm{~F}$-NMR spectra are not externally calibrated and chemical shifts is given relative to $\mathrm{CCl}_{3} \mathrm{~F}$ as received from the automatic data processing. ${ }^{13} \mathrm{C}$-NMR and ${ }^{19} \mathrm{~F}-\mathrm{NMR}$ spectra were acquired on a broadband decoupled mode. Chemical shifts are generally reported with two $\left({ }^{1} \mathrm{H}\right)$ or one (all other nuclei) digits after the decimal point. NMR-data are reported as follows: chemical shift (multiplicity [ $\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet,
$\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{p}=$ quintuplet, hept $=$ heptuplet, $\mathrm{m}=$ multiplet, $\mathrm{br}=$ broad], coupling constants $(J, \mathrm{~Hz})$ and integration). All spectra were processed using the MestReNova program.

High-Resolution Mass Spectra (HRMS) were recorded on an Agilent Technologies 6120 Quadrupole LC/MS coupled with an SFC Agilent Technologies 1260 Infinity Series instrument for the ESI-MS (Electrospray lonization) or on an Agilent Technologies 5977B MSD coupled with an Agilent Technologies 7820A GC System for the EI-MS (Electron lonization mass spectroscopy). MassWorks software version 4.0.0.0 (Cerno Bioscience) was used for the formula identification. MassWorks is an MS calibration software which calibrates isotope profiles to achieve high mass accuracy and enables elemental composition determination on conventional mass spectrometers of unit mass resolution allowing highly accurate comparisons between calibrated and theoretical spectra. ${ }^{[2]}$ Obtained data are expressed in mass/charge ( $\mathrm{m} / \mathrm{z}$ ) units.

UV-Vis measurements were acquired on an Agilent 8453 UV-Vis Spectrophotometer controlled by UV-Visible ChemStation Software. Emission spectra were recorded on a JASCO Spectrofluorometer FP-8600 equipped with a TC-815 Peltier thermostated single cell holder (water-cooled) controlled by Spectra Manager Version 2.10.01. Time resolved emission spectra were carried out using an Edinburg Instruments FS5 Spectrofluorometer, and a 450 nm EPL laser.

Cyclic Voltammetry (CV) experiments were acquired on an IVIUM Technologies CompactStat controlled by IviumSoft version 2.124 offering a compliance voltage of up to $\pm 100 \mathrm{~V}$ (available at the counter electrode), $\pm 10 \mathrm{~V}$ scan range and $\pm 1 \mathrm{~A}$ current range. HPLC grade acetone solvent was used for all measurements. Tetra- $n$-butylammonium hexafluorophosphate was used as supporting electrolyte at 0.1 M concentration. All cyclic voltammetry experiments were performed using a conventional three-electrode system, containing a coiled Pt wire acting as counter electrode, an $\mathrm{Ag} / \mathrm{AgCl}$ saturated solution as reference electrode and a glassy carbon working electrode $\left(A=0.071 \mathrm{~cm}^{2}\right)$ at $20 \mathrm{mV} / \mathrm{s}$ scan rate. All the electrodes were
purchased from Metrohm. Redox-active species were dissolved at 1.0 nM concentration and these solutions were throughout the measurements.

A custom-made temperature-controlled photoreactor setup was used for the photocatalytic reactions (Figure S1). The irradiation takes place at the desired wavelengths ( $365,385,420$, 450 or 540 nm ) using 380 mW single LEDs, located 1 cm beneath the base of the vial. Reaction temperature is kept at $20-25^{\circ} \mathrm{C}$ using a recirculating chiller.


Figure S1. Experimental setup employed during photocatalytic reactions.

## 2. Synthesis and Characterization of Substrates

### 2.1. General Procedure and Characterization of Sulfides SI-1



General procedure A for the synthesis of sulfides SI-1:

Compounds SI-1a-c were prepared following a slightly modified procedure reported in the literature. ${ }^{[3]}$ 1-(2-bromophenyl)ethan-1-one (1 equiv) was added to a stirred solution of the corresponding sodium alkanethiolate (1.1. equiv) in THF (1.3M) and heated to $75{ }^{\circ} \mathrm{C}$ for 12 h . After this time, the reaction was cooled to room temperature, diluted with dichloromethane and washed with water ( $3 x$ ) and brine. The organic phase was dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated under reduced
pressure. Then, the solid residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure products $\mathbf{S I - 1}$.

## General procedure B for the synthesis of sulfides SI-1:

Compounds Sl-1d-s were prepared following a slightly modified procedure reported in the literature. ${ }^{[3]}$ The corresponding thiol ( 1.05 equiv) was added dropwise over 0.25 h to a stirred suspension of NaH ( 1.2 equiv) in THF or DMF ( $2 \mathrm{~mL} / \mathrm{mmol}$ of thiol) at $0{ }^{\circ} \mathrm{C}$ (icebath). Afterwards, a solution of the bromide/chloride/fluoride compound (1 equiv) in THF or DMF ( $0.5 \mathrm{~mL} / \mathrm{mmol}$ ) was added and the reaction was heated to $75^{\circ} \mathrm{C}$ for 12 h . After this time, the reaction was cooled to room temperature, diluted with dichloromethane and washed with water ( $3 x$ ) and brine. The organic phase was dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Then, the solid residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure products $\mathbf{S I - 1}$.

## 1-(2-(methylthio)phenyl)ethan-1-one (SI-1a)

O Following the general procedure $A$, the reaction of 1-(2-bromophenyl)ethan-1-one ( $2.1 \mathrm{mmol}, 0.28 \mathrm{~mL}$ ) and sodium methanethiolate ( $2.3 \mathrm{mmol}, 160.1 \mathrm{mg}$ ) in THF ( 1.6 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 a}(88 \%, 305.3 \mathrm{mg})$ as a pale orange solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.71(\mathrm{dd}, J=7.9,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.37-7.29(\mathrm{~m}, 1 \mathrm{H}), 7.17(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H})$, $7.04(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.46(\mathrm{~s}, 3 \mathrm{H}), 2.27(\mathrm{~s}, 3 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[4]}$

## 1-(2-(isopropylthio)phenyl)ethan-1-one (SI-1b)



Following the general procedure $A$, the reaction of 1-(2-bromophenyl)ethan-1-one ( $2 \mathrm{mmol}, 0.27 \mathrm{~mL}$ ) and sodium 2-propanethiolate ( $2.2 \mathrm{mmol}, 215.9 \mathrm{mg}$ ) in THF ( 1.5 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I}-\mathbf{1 b}(75 \%, 292.3 \mathrm{mg})$ as a yellow oil. $\left.{ }^{1} \mathrm{H} \mathbf{N M R} \mathbf{( 3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}\right)$ $\delta 7.65$ (dd, $J=7.7,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.46-7.36$ (m, 2H), 7.21 (ddd, $J=7.7,6.7,1.8 \mathrm{~Hz}$, 1H), 3.46 (hept, $J=6.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.60 (s, 3H), 1.33 (d, $J=6.6 \mathrm{~Hz}, 6 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 200.5,138.7,137.8,131.3,129.6,129.1,124.7,36.2,29.3,22.6$.

## 1-(2-(tert-butylthio)phenyl)ethan-1-one (SI-1c)



Following the general procedure $A$, the reaction of 1-(2-bromophenyl)ethan-1-one ( $2 \mathrm{mmol}, 0.27 \mathrm{~mL}$ ) and sodium 2-methyl-2-propanethiolate ( $2.2 \mathrm{mmol}, 246.8 \mathrm{mg}$ ) in THF ( 1.5 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 c}(68 \%, 284.4 \mathrm{mg})$ as an yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, $\left.\mathrm{CDCl}_{3}\right) 7.58(\mathrm{t}, \mathrm{J}=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.38-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.32-7.25(\mathrm{~m}, 1 \mathrm{H}), 2.29(\mathrm{~s}, 3 \mathrm{H})$, $1.22(\mathrm{~s}, 9 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[5]}$

## 1-(2-(benzylthio)phenyl)ethan-1-one (SI-1d)

O Following the general procedure B , the reaction of 1-(2-bromophenyl)ethan-1-one ( $2 \mathrm{mmol}, 0.27 \mathrm{~mL}$ ), $\mathrm{NaH}(60 \%, 2.4$ $\mathrm{mmol}, 96 \mathrm{mg}$ ) and benzyl mercaptan ( $2.1 \mathrm{mmol}, 0.25 \mathrm{~mL}$ ) in THF ( 5.5 mL ) at $75{ }^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 d}(71 \%, 342.8 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.78(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{~d}, J=3.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.39(\mathrm{~d}, J=6.5$ $\mathrm{Hz}, 2 \mathrm{H}), 7.34-7.17(\mathrm{~m}, 4 \mathrm{H}), 4.13(\mathrm{~s}, 2 \mathrm{H}), 2.59(\mathrm{~s}, 3 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[6]}$

## 1-(2-(phenylthio)phenyl)ethan-1-one (SI-1e)



Following the general procedure $B$, the reaction of 1-(2-bromophenyl)ethan-1-one ( $2 \mathrm{mmol}, 0.27 \mathrm{~mL}$ ), $\mathrm{NaH}(60 \%, 2.4$ mmol, 96 mg ) and thiophenol ( $2.1 \mathrm{mmol}, 0.21 \mathrm{~mL}$ ) in THF ( 5.5 mL ) at $75{ }^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 e}(72 \%, 330.0 \mathrm{mg})$ as a yellow solid. ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 7.83$ (dd, J = 7.7, 1.6 Hz, 1H), $7.56-7.52(\mathrm{~m}, 2 \mathrm{H}), 7.43-7.40(\mathrm{~m}, 3 \mathrm{H})$, $7.28-7.14(\mathrm{~m}, 2 \mathrm{H}), 6.89(\mathrm{dd}, J=8.0,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.67(\mathrm{~s}, 3 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[7]}$

## 1-(2-(benzylthio)-4-fluorophenyl)ethan-1-one (SI-1f)



Following the general procedure $B$, the reaction of 1-(2-bromo-4-fluorophenyl)ethan-1-one ( $2.8 \mathrm{mmol}, 600.0 \mathrm{mg}$ ), NaH ( $60 \%$, $3.4 \mathrm{mmol}, 134.4 \mathrm{mg}$ ) and benzyl mercaptan ( $3.0 \mathrm{mmol}, 0.35$ $\mathrm{mL})$ in THF ( 7.7 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I}-1 \mathbf{f}(64 \%, 463.7 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 7.84(\mathrm{dd}, \mathrm{J}=8.7,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.43-7.37(\mathrm{~m}, 2 \mathrm{H})$,
$7.35-7.24(\mathrm{~m}, 3 \mathrm{H}), 7.08$ (dd, $J=10.5,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.84$ (ddd, $J=8.7,7.6,2.4 \mathrm{~Hz}$, 1H), 4.08 (s, 2H), 2.56 ( $\mathrm{s}, 3 \mathrm{H}$ ). ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 197.5,164.8(\mathrm{~d}, \mathrm{~J}=255.0$ $\mathrm{Hz}), 145.8(\mathrm{~d}, J=8.9 \mathrm{~Hz}), 135.5,133.6(\mathrm{~d}, J=10.1 \mathrm{~Hz}), 130.7(\mathrm{~d}, J=2.6 \mathrm{~Hz}), 129.2$, 128.8, 127.6, $113.0(\mathrm{~d}=25.1 \mathrm{~Hz}), 110.8(\mathrm{~d}, \mathrm{~J}=22.0 \mathrm{~Hz}), 37.7,28.1$; ${ }^{19} \mathrm{~F}$ NMR (282 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$-105.6.

## 1-(2-(benzylthio)-4-methoxyphenyl)ethan-1-one (SI-1g)

 $0.3 \mathrm{~mL})$ in DMF ( 6.6 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 g}(63 \%, 412.9 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.71(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.41-7.34(\mathrm{~m}, 2 \mathrm{H})$, $7.29-7.15(\mathrm{~m}, 3 \mathrm{H}), 6.79(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.57(\mathrm{dd}, J=8.7,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.02(\mathrm{~s}$, 2H), 3.68 (s, 3H), $2.45(\mathrm{~s}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 197.1,162.5,144.8,136.3$, 133.7, 129.1, 128.6, 127.3, 127.1, 111.3, 108.8, 55.5, 37.3, 27.7.

## 1-(2-(benzylthio)-5-methoxyphenyl)ethan-1-one (SI-1h)



Following the general procedure $B$, the reaction of 1-(2-bromo-5-methoxyphenyl)ethan-1-one ( $2 \mathrm{mmol}, 458.1 \mathrm{mg}$ ), $\mathrm{NaH}(60 \%, 2.4 \mathrm{mmol}, 96 \mathrm{mg})$ and benzyl mercaptan ( 2.1 mmol , 0.25 mL ) in THF ( 5.5 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 h}(60 \%, 327.5 \mathrm{mg})$ as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 7.33-7.19(\mathrm{~m}, 6 \mathrm{H}), 7.11(\mathrm{~d}, \mathrm{~J}=2.8 \mathrm{~Hz}, 1 \mathrm{H})$, 6.91 (dd, J = 8.7, $2.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.02 (s, 2H), 3.80 (s, 3H), 2.53 (s, 3H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 200.7,157.6,140.9,136.9,131.9,128.9,128.3,127.2,127.0,117.0$, 114.5, 55.3, 39.6, 29.4.

## 1-(2-(benzylthio)-5-bromophenyl)ethan-1-one (SI-1i)

Following the general procedure $B$, the reaction of
$(60 \%, 2.4 \mathrm{mmol}, 96 \mathrm{mg})$ and benzyl mercaptan $(2.1 \mathrm{mmol}, 0.25 \mathrm{~mL})$
in THF ( 5.5 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I}-1 \mathbf{i}(57 \%, 367.3 \mathrm{mg})$ as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.84(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{dd}, J=8.5,2.2 \mathrm{~Hz}$,

1H), $7.34-7.17$ (m, 6H), 4.09 (s, 2H), 2.56 ( $\mathrm{s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) ס 198.1, 139.5, 137.1, 135.7, 134.6, 133.1, 128.9, 128.5, 128.5, 127.4, 117.6, 37.8, 28.4.

## 1-(2-(benzylthio)-5-(trifluoromethyl)phenyl)ethan-1-one (SI-1j)


i) 1) MeMgBr , $\mathrm{Et} 2 \mathrm{O}, 0^{\circ} \mathrm{C}$; 2) PCC, silica gel, DCM , rt.

Compound SI-2 was prepared following a previously reported procedure ${ }^{[8]}$ and spectral data is in accordance with the literature. ${ }^{[9]}$
 $\mathrm{mmol}, 0.22 \mathrm{~mL})$ in DMF ( 4.7 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I}-1 \mathrm{j}(43 \%, 224.6 \mathrm{mg})$ as a pale yelow solid. ${ }^{1} \mathrm{H} \mathbf{N M R}\left(\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}\right) \delta 7.98(\mathrm{~d}, \mathrm{~J}=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.57(\mathrm{dd}$, $J=8.5,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.46(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.40-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.20(\mathrm{~m}, 3 \mathrm{H})$, 4.13 (s, 2H), 2.59 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 197.9,146.4(\mathrm{q}, \mathrm{J}=1.0 \mathrm{~Hz}$ ), 135.5, 134.6, 129.0, 128.7, 128.2 (q, $J=3.5 \mathrm{~Hz}), 127.6,127.4(\mathrm{q}, J=3.8 \mathrm{~Hz}), 126.4$, $126.0(\mathrm{q}, J=33.3 \mathrm{~Hz}), 123.8(\mathrm{q}, J=271.8 \mathrm{~Hz}), 37.5,28.1 ;{ }^{19} \mathrm{~F}$ NMR (282 MHz, CDCl ${ }_{3}$ ) $\delta$-62.3.

## 1-(6-(benzylthio)benzo[d][1,3]dioxol-5-yl)ethan-1-one (SI-1k)


i) 1) $\mathrm{MeMgBr}, \mathrm{Et} 2 \mathrm{O}, 0^{\circ} \mathrm{C}$; 2) PCC, silica gel, DCM , rt.

Compound SI-3 was prepared following a previously reported procedure ${ }^{[8]}$ and spectral data is in accordance with the literature. ${ }^{[10]}$


Following the general procedure $B$, the reaction of 1-(6-bromobenzo[d][1,3]dioxol-5-yl)ethan-1-one (SI-3) ( 1.2 mmol , 291.7 mg ), $\mathrm{NaH}(60 \%, 1.4 \mathrm{mmol}, 57.6 \mathrm{mg}$ ) and benzyl mercaptan $(1.3 \mathrm{mmol}, 0.15 \mathrm{~mL})$ in DMF $(3.3 \mathrm{~mL})$ at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I}-1 \mathbf{k}(57 \%, 195.8 \mathrm{mg})$ as a pale brown solid. ${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}\right) \delta 7.40-$ $7.24(\mathrm{~m}, 5 \mathrm{H}), 7.22(\mathrm{~s}, 1 \mathrm{H}), 6.88(\mathrm{~s}, 1 \mathrm{H}), 6.01(\mathrm{~s}, 2 \mathrm{H}), 4.06(\mathrm{~s}, 2 \mathrm{H}), 2.50(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 197.6,151.1,145.1,136.3,129.2,128.7,127.4,110.2,107.4$, 102.2, 38.6, 28.6.

## 1-(3-(benzyloxy)-2-(benzylthio)phenyl)ethan-1-one (SI-1I)


i) $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{BnBr}, \mathrm{DMF}$, rt; ii) 1) MeMgBr , $\mathrm{Et}_{2} \mathrm{O}, 0^{\circ} \mathrm{C}$; 2) PCC , silica gel, DCM , rt; iii) $\mathrm{HSBn}, \mathrm{NaH}, \mathrm{DMF}, 75^{\circ} \mathrm{C}$.

Compounds SI-4 and SI-5 were prepared following previously reported procedures ${ }^{[8,11]}$ and spectral data is in accordance with the literature. ${ }^{[12]}$


Following the general procedure $B$, the reaction of 1-(3-(benzyloxy)-2-bromophenyl)ethan-1-one (SI-5) ( $0.7 \mathrm{mmol}, 213.6$ $\mathrm{mg}), \mathrm{NaH}(60 \%, 0.84 \mathrm{mmol}, 33.6 \mathrm{mg})$ and benzyl mercaptan $(0.74 \mathrm{mmol}$, $0.09 \mathrm{~mL})$ in DMF ( 1.9 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{~ S I - 1 I ~}(61 \%$, 150.0 mg ) as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~} \mathrm{CDCl}_{3}$ ) $\delta 7.45-7.39(\mathrm{~m}, 2 \mathrm{H}), 7.37-$ 7.25 (m, 3H), $7.22-7.15$ (m, 1H), 7.12 - 7.03 (m, 3H), $7.00-6.94$ (m, 2H), 6.91 (dd, $J=8.2,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{dd}, J=7.6,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.08(\mathrm{~s}, 2 \mathrm{H}), 3.95(\mathrm{~s}, 2 \mathrm{H}), 2.20(\mathrm{~s}$, 3H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) б 203.9, 159.7, 149.1, 138.1, 136.6, 130.0, 129.0, 128.8, 128.4, 128.2, 127.4, 127.1, 118.8, 118.6, 113.8, 71.1, 39.3, 31.3.

## 1-(1-(benzylthio)naphthalen-2-yl)ethan-1-one (SI-1m)



Compound SI-6 was prepared following a previously reported procedure ${ }^{[8]}$ and spectral data is in accordance with the literature. ${ }^{[13]}$


Following the general procedure $B$, the reaction of 1-(1-bromonaphthalen-2-yl)ethan-1-one (SI-6) ( $2 \mathrm{mmol}, 498.2 \mathrm{mg}$ ), $\mathrm{NaH}(60 \%, 2.4 \mathrm{mmol}, 96 \mathrm{mg})$ and benzyl mercaptan ( 2.1 mmol , $0.25 \mathrm{~mL})$ in DMF ( 5.5 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 m}(62 \%, 364.3 \mathrm{mg})$ as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 8.60(\mathrm{~d}, \mathrm{~J}=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.07-7.77(\mathrm{~m}$, $2 \mathrm{H}), 7.56(\mathrm{pd}, J=6.9,1.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.31(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.22-7.03(\mathrm{~m}, 3 \mathrm{H}), 7.09$ - $6.88(\mathrm{~m}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 2 \mathrm{H}), 2.49(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 204.6, 146.9, 137.4, 134.7, 134.1, 129.9, 128.8, 128.5, 128.3, 127.6, 127.4, 127.2, 127.0, 126.4, 122.7, 42.2, 31.6.

## 1-(3-(benzylthio)furan-2-yl)ethan-1-one (SI-1n)



Compound SI-7 was prepared following a previously reported procedure and spectral data is in accordance with the literature. ${ }^{[14]}$
Following the general procedure B , the reaction of
1-(3-bromofuran-2-yl)ethan-1-one (SI-7) $(3.45 \mathrm{mmol}, 652.1 \mathrm{mg}), \mathrm{NaH}$ DMF ( 9.5 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 n}(87 \%, 696.9 \mathrm{mg})$ as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~ C D C l}{ }_{3}$ ) $\delta 7.33(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.33-7.08(\mathrm{~m}, 5 \mathrm{H}), 6.41(\mathrm{~d}, J=$ $1.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.06 (s, 2H), 2.35 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 186.7, 146.8,
145.2, 136.1, 130.9, 128.7, 128.6, 127.4, 111.8, 37.1, 26.1.

## 1-(3-(benzylthio)thiophen-2-yl)ethan-1-one (SI-10)



Following the general procedure $B$, the reaction of 1-(3-bromothiophen-2-yl)ethan-1-one ( $2 \mathrm{mmol}, 408.1 \mathrm{mg}$ ), $\mathrm{NaH}(60 \%$, $2.4 \mathrm{mmol}, 96 \mathrm{mg}$ ) and benzyl mercaptan ( $2.1 \mathrm{mmol}, 0.25 \mathrm{~mL}$ ) in DMF ( 5.5 mL ) at $75{ }^{\circ} \mathrm{C}$ afforded the product $\mathrm{Sl}-10(83 \%, 412.9 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR (300 MHz, CDCl ${ }_{3}$ ) $\delta 7.44(\mathrm{~d}, \mathrm{~J}=5.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39-7.32(\mathrm{~m}, 2 \mathrm{H}), 7.32-7.20(\mathrm{~m}$, $3 \mathrm{H}), 7.03(\mathrm{~d}, \mathrm{~J}=5.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{~s}, 2 \mathrm{H}), 2.47(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta$ 189.8, 143.7, 136.2, 132.1, 131.4, 128.9, 128.7, 127.6, 127.3, 38.1, 28.8.

## 1-(2-(benzylthio)pyridin-3-yl)ethan-1-one (SI-1p)



Compound SI-8 was prepared following a previously reported procedure ${ }^{[8]}$ and spectral data is in accordance with the literature. ${ }^{[15]}$
 $\mathrm{mL})$ at $75{ }^{\circ} \mathrm{C}$ afforded the product $\mathrm{SI}-1 \mathrm{p}(45 \%, 108.7 \mathrm{mg})$ as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 8.58(\mathrm{dd}, J=4.7,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.04(\mathrm{dd}, J=7.8,1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $7.46-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.32-7.17(\mathrm{~m}, 3 \mathrm{H}), 7.09(\mathrm{dd}, \mathrm{J}=7.8,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.44(\mathrm{~s}, 2 \mathrm{H})$, 2.57 (s, 3H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 197.6,161.4,151.7,138.4,138.0,129.5$, $129.0,128.4,126.9,118.3,34.9,27.7$.

## 1-(2-(benzylthio)quinolin-3-yl)ethan-1-one (SI-1q)



Following the general procedure $B$, the reaction of 1-(2-chloroquinolin-3-yl)ethan-1-one ( $0.8 \mathrm{mmol}, 187.5 \mathrm{mg}$ ), NaH $(60 \%, 1 \mathrm{mmol}, 38.4 \mathrm{mg})$ and benzyl mercaptan ( $0.84 \mathrm{mmol}, 0.10$ mL ) in DMF ( 2.2 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 q}(96 \%, 225.8 \mathrm{mg})$ as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 8.40$ (s, 1H), 7.97 (d, J=8.3 Hz, 1H), 7.79 $7.70(\mathrm{~m}, 2 \mathrm{H}), 7.53(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.44(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.29(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H})$, 7.25 - 7.17 (m, 1H), 4.56 (s, 2H), 2.64 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz, CDCl $_{3}$ ) ס 197.5, 158.7, 148.4, 139.6, 138.4, 132.4, 129.5, 128.8, 128.3, 127.9, 127.7, 126.8, 125.9, 124.0, 35.0, 27.7.

## 2-(benzylthio)benzaldehyde (SI-1r)

O Following the general procedure B , the reaction of 2-bromobenzaldehyde ( $2 \mathrm{mmol}, 370.0 \mathrm{mg}$ ), $\mathrm{NaH}(60 \%, 2.4 \mathrm{mmol}, 96$ mg ) and benzyl mercaptan ( $2.1 \mathrm{mmol}, 0.25 \mathrm{~mL}$ ) in THF ( 5.5 mL ) at 75 ${ }^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I}-\mathbf{1 r}(75 \%, 341.1 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR $(\mathbf{3 0 0} \mathbf{~ M H z}$, $\left.\mathrm{CDCl}_{3}\right) \delta 10.25(\mathrm{~s}, 1 \mathrm{H}), 7.79(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{dt}, J=7.4,3.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.33-$ $7.17(\mathrm{~m}, 6 \mathrm{H}), 4.11(\mathrm{~s}, 2 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[16]}$

## (2-(benzylthio)phenyl)(phenyl)methanone (SI-1s)

 $(5.5 \mathrm{~mL})$ at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 s}(63 \%, 385.4 \mathrm{mg})$ as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 7.62(\mathrm{~d}, \mathrm{~J}=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.42(\mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.28(\mathrm{t}$, $J=7.7 \mathrm{~Hz}, 3 \mathrm{H}), 7.21(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.15-7.06(\mathrm{~m}, 6 \mathrm{H}), 3.91(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) б 196.6, 140.7, 137.3, 136.8, 135.3, 133.0, 131.1, 130.3, 130.0, 128.8, 128.7, 128.3, 127.0, 125.9, 39.6.

### 2.2. General Procedure and Characterization of Selenides

## SI-10



General procedure for the synthesis of selenides SI-10:

Compounds $\mathbf{S I - 1 0 a}$ and $\mathbf{S I - 1 0 b}$ were prepared following a reported procedure in the literature. ${ }^{[17]}$ The corresponding 2-aminoacetophenone or 2-amino-5-chlorobenzaldehyde (1.1 equiv) and an aqueous HCl solution (2M) were added to an Erlenmeyer flask. The solution was cooled to $0^{\circ} \mathrm{C}$ and a water solution of $\mathrm{NaNO}_{2}$ (1 equiv, 2M) was added dropwise. At $0^{\circ} \mathrm{C}$, sodium acetate ( 2.5 equiv) was added, followed by the addition of acetate buffer solution until pH 4.3. The KSeCN (1.5 equiv) was then added under vigorous agitation and the solution was kept 1 h at 0 ${ }^{\circ} \mathrm{C}$. Then, sodium acetate was added until pH 5.5 . The resulting solution was extracted with dichloromethane (3x) and the combined organic phases were dried over anhydrous $\mathrm{MgSO}_{4}$. The solvent was removed in vacuum and the residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure products $\mathbf{S I - 9 a}$ or $\mathbf{S I - 9 b}$.

To a two necked round-bottomed flask under nitrogen, compound SI-9a or SI-9b (1 equiv) and methanol ( $5 \mathrm{~mL} / \mathrm{mmol}$ ) were added. The solution was cooled to $0^{\circ} \mathrm{C}$ and benzylbromide (4 equiv) was added, followed by the slow addition of $\mathrm{NaBH}_{4}$ (1.1 equiv). The solution was stirred 2 h at $0^{\circ} \mathrm{C}$. The solvent was removed in vacuum and the residue was diluted in ethyl acetate followed by addition of saturated aqueous solution of $\mathrm{NH}_{4} \mathrm{Cl}$. After phase separation, the aqueous phase was extracted with ethyl acetate ( 2 x ) and the combined organic phases were dried over $\mathrm{MgSO}_{4}$. The solvent was removed in vacuum and the residue was subjected to flash
chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure products SI-10a or SI-10b.

## 1-(2-(benzylselanyl)phenyl)ethan-1-one (SI-10a)



Following the general procedure, the reaction of 1-(2-selenocyanatophenyl)ethan-1-one (SI-9a) ( $2 \mathrm{mmol}, 448.2 \mathrm{mg}$ ), $\mathrm{NaBH}_{4}(2.2 \mathrm{mmol}, 83.2 \mathrm{mg})$ and benzyl bromide ( $8.8 \mathrm{mmol}, 1.05 \mathrm{~mL}$ ) in $\mathrm{MeOH}(10 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 0 a}(71 \%, 411.7 \mathrm{mg})$ as a pale brown solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 7.93$ (dd, $J=7.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.58 (d, $J=$ $7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.42$ (dd, J = 11.9, 7.2 Hz, 3H), $7.35-7.20(\mathrm{~m}, 4 \mathrm{H}), 4.11$ (s, 2H), 2.62 (s, 3H). Spectral data is in accordance with the literature. ${ }^{[17]}$

## 2-(benzyIselanyl)-5-chlorobenzaldehyde (SI-10b)



Following the general procedure, the reaction of 5-chloro-2-selenocyanatobenzaldehyde (SI-9b) ( $2 \mathrm{mmol}, 489.0$ $\mathrm{mg}), \mathrm{NaBH}_{4}(2.2 \mathrm{mmol}, 83.2 \mathrm{mg})$ and benzyl bromide $(8.8 \mathrm{mmol}$, 1.05 mL ) in $\mathrm{MeOH}(10 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 0 b}(76 \%, 469.3 \mathrm{mg})$ as an yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 10.05$ (s, 1H), 7.79 (d, J = $2.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.52 (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{dd}, J=8.4,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.26-7.20(\mathrm{~m}, 5 \mathrm{H}), 4.10(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 191.5,136.6,134.5,133.8,133.7,133.1,132.3,129.7,129.0$, 128.7, 127.3, 31.5.

### 2.3. General Procedure and Characterization of $\alpha$-Imino-oxy

## Acids 1


$\mathrm{Sl}-1$ or ${ }^{\mathrm{SI}-10}$

reflux
$Y=S, S e$


1

Compound SI-11 was prepared following a reported procedure in the literature and spectral data is in accordance with the literature. ${ }^{[18]}$

General procedure for the synthesis of oximes 1:
Compounds 1a-u were prepared following a slightly modified procedure reported in the literature ${ }^{[18]} \mathrm{A}$ solution of ketone ( 1.0 equiv) in MeOH or $\mathrm{EtOH}(0.2 \mathrm{M})$ was treated with 2-(aminooxy)-2-methylpropanoic acid hydrochloride (SI-11) (1.2 equiv) and sodium acetate ( 2.4 equiv) and heated to reflux until completion by TLC analysis. The mixture was then allowed to cool to room temperature and water was added. The resulting aqueous solution was extracted with EtOAc (3x) and the combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated under vacuum to provide the product. Then, the residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure products 1.

2-methyl-2-(((1-(2-(methylthio)phenyl)ethylidene)amino)oxy)propanoic acid (1a)


Following the general procedure, the reaction of SI-1a (0.9 mmol, 149.6 mg ), SI-11 ( $1.08 \mathrm{mmol}, 168.0 \mathrm{mg}$ ) and sodium acetate ( $2.2 \mathrm{mmol}, 177.2 \mathrm{mg}$ ) in $\mathrm{EtOH}(4.5 \mathrm{~mL})$ at reflux afforded the product $\mathbf{1 a}(57 \%, 137.3 \mathrm{mg})$ as a white solid. ${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~ C D C l ~}{ }_{3}$ ) $\delta 10.02$ (bs, 1H), 7.35 - 7.22 (m, 3H), $7.18-7.11(\mathrm{~m}, 1 \mathrm{H}), 2.38(\mathrm{~s}, 3 \mathrm{H}), 2.28(\mathrm{~s}, 3 \mathrm{H}), 1.62(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 179.5,157.5,137.7,136.5,129.1,128.8,126.8,124.9,81.3,24.4,16.8,16.0 ;$ HRMS m/z (ESI): calcd. for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{NO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-2} 266.0856$, found 266.0862 .

2-(((1-(2-(isopropylthio)phenyl)ethylidene)amino)oxy)-2-methylpropanoic acid (1b)


Following the general procedure, the reaction of $\mathbf{S I - 1 b}$ ( 1 mmol , 194.0 mg ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( $2.4 \mathrm{mmol}, 196.9 \mathrm{mg}$ ) in EtOH ( 5 mL ) at reflux afforded the product 1b ( $75 \%$, 22.7 mg ) as a yellow oil. ${ }^{\mathbf{1}} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~}$ $\mathrm{CDCl}_{3}$ ) $\delta 8.20(\mathrm{bs}, 1 \mathrm{H}), 7.44(\mathrm{~d}, \mathrm{~J}=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.35-7.25(\mathrm{~m}$, 1H), 7.22 (d, J = $4.1 \mathrm{~Hz}, 2 \mathrm{H}$ ), $3.40-3.21$ (m, 1H), 2.28 (s, 3H), 1.59 (s, 6H), 1.22 (d, J $=6.6 \mathrm{~Hz}, 6 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz},\left(\mathrm{CDCl}_{3}\right) \delta 177.0,160.0,139.8,134.5,132.7,129.2$, 129.1, 126.8, 81.7, 38.7, 24.5, 23.0, 17.3.; HRMS m/z (ESI): calcd. for $\mathrm{C}_{15} \mathrm{H}_{20} \mathrm{NO}_{3} \mathrm{~S}$
(M-H)- 294.1169, found 249.1165.
2-(((1-(2-(tert-butylthio)phenyl)ethylidene)amino)oxy)-2-methylpropanoic acid (1c)


Following the general procedure, the reaction of $\mathbf{S I - 1 c}$ (0.3 $\mathrm{mmol}, 66.0 \mathrm{mg}$ ), SI-11 ( $0.36 \mathrm{mmol}, 56.0 \mathrm{mg}$ ) and sodium acetate ( $0.72 \mathrm{mmol}, 59.1 \mathrm{mg}$ ) in $\mathrm{EtOH}(1.5 \mathrm{~mL})$ at reflux afforded the product $1 \mathrm{c}(60 \%, 55.9 \mathrm{mg})$ as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 7.63$ - 7.54 (m, 1H), 7.38 - 7.32 (m, 2H), $7.31-7.25(\mathrm{~m}, 1 \mathrm{H}), 2.29(\mathrm{~s}, 3 \mathrm{H}), 1.58(\mathrm{~s}, 6 \mathrm{H}), 1.22(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 178.0,160.8,144.3,139.4,131.4,129.7,129.2,128.9,81.5,47.9$, 31.3, 24.6, 18.8; HRMS m/z (ESI): calcd. for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{NO}_{3} \mathrm{~S}$ (M-H)- 308.1326, found 308.1336

2-(((1-(2-(benzylthio)phenyl)ethylidene)amino)oxy)-2-methylpropanoic acid (1d) Following the general procedure, the reaction of SI-1d (0.4
 $\mathrm{mmol}, 100.2 \mathrm{mg}$ ), SI-11 ( $0.48 \mathrm{mmol}, 74.7 \mathrm{mg}$ ) and sodium acetate ( $0.96 \mathrm{mmol}, 78.7 \mathrm{mg}$ ) in EtOH ( 2 mL ) at reflux afforded the product $\mathbf{1 d}(96 \%, 132.8 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0}$ MHz, CDCl $_{3}$ ) $\delta 10.91$ (bs, 1H), $7.38-7.34$ (m, 1H), 7.31 - 7.24 (m, 8H), 4.09 (s, 2H), 2.29 (s, 3H), 1.65 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz, CDCl $_{3}$ ) ס 179.4, 158.3, 139.1, 137.4, 134.9, 131.1, 129.1, 129.0, 128.9, 128.4, 127.1, 126.5, 81.3, 39.5, 24.4, 16.7; HRMS m/z (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{NO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-342.1169, ~ f o u n d ~ 342.1162 . ~}$ 2-methyl-2-(((1-(2-(phenylthio)phenyl)ethylidene)amino)oxy)propanoic acid (1e)
 Following the general procedure, the reaction of $\mathbf{S I - 1 e}(1 \mathrm{mmol}$, 228.3 mg ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( 2.4 $\mathrm{mmol}, 196.9 \mathrm{mg}$ ) in EtOH ( 5 mL ) at reflux afforded the product $\mathbf{1 e}(34 \%, 111.0 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) б 10.32 (bs, 1H), 7.39 - 7.24 (m, 9H), 2.36 (s, 3H), 1.61 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\mathrm{CDCl}_{3}$ ) б 177.3, 157.1, 137.3, 134.7, 134.1, 131.2, 130.5, 128.3, 128.2, 128.2, 126.2, 125.9, 80.5, 23.3, 15.4; HRMS m/z (ESI): calcd. for $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{NO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-}$328.1013,

## 2-(((1-(2-(benzylthio)-4-fluorophenyl)ethylidene)amino)oxy)-2-methylpropanoic acid (1f)



Following the general procedure, the reaction of SI-1f (0.4 mmol, 99.2 mg ), SI-11 ( $0.48 \mathrm{mmol}, 74.7 \mathrm{mg}$ ) and sodium acetate ( $0.96 \mathrm{mmol}, 78.7 \mathrm{mg}$ ) in EtOH ( 2 mL ) at reflux afforded the product $1 \mathrm{f}(88 \%, 126.9 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~ C D C l ~}{ }_{3}$ ) $\delta 9.99$ (bs, 1H), 7.27 - 7.16 (m, 6H), 6.99 (dd, $J=9.5,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.83(\mathrm{td}, J=8.3,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.03(\mathrm{~s}, 2 \mathrm{H}), 2.20(\mathrm{~s}, 3 \mathrm{H})$, 1.57 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 179.0,162.7$ ( $\mathrm{d}, \mathrm{J}=250.1 \mathrm{~Hz}$ ), 157.3, 138.3 (d, $J=8.1 \mathrm{~Hz}$ ), 136.6, 134.1 (d, $J=3.3 \mathrm{~Hz}$ ), 130.7 (d, $J=8.8 \mathrm{~Hz}$ ), 128.9, 128.7, 127.5, 116.5 (d, $J=23.7 \mathrm{~Hz}), 113.0(\mathrm{~d}, J=21.6 \mathrm{~Hz}), 81.5,39.1,24.4,16.5 ;{ }^{19}$ F NMR (282 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$-111.98 (syn isomer), -112.59 (anti isomer); HRMS m/z (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{FNO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-} 360.1064$, found 360.1050 .

2-(((1-(2-(benzylthio)-4-methoxyhenyl)ethylidene)amino)oxy)-2-methylpropanoi c acid (1g)


Following the general procedure, the reaction of $\mathbf{S I - 1 g}$ ( $0.5 \mathrm{mmol}, 149.6 \mathrm{mg}$ ), SI-11 ( $0.6 \mathrm{mmol}, 93.3 \mathrm{mg}$ ) and sodium acetate ( $1.2 \mathrm{mmol}, 98.4 \mathrm{mg}$ ) in $\mathrm{EtOH}(2.5 \mathrm{~mL})$ at reflux afforded the product $\mathbf{1 g}(70 \%, 143.0 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~} \mathrm{CDCl}_{3}$ ) $\delta 9.89$ (bs, 1H), 7.30 $7.20(\mathrm{~m}, 5 \mathrm{H}), 7.17(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.71(\mathrm{dd}, J=8.5,2.5$ Hz, 1H), 4.04 (s, 2H), 3.71 (s, 3H), 2.23 (s, 3H), 1.59 (s, 6H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) б 178.6, 159.9, 158.3, 137.2, 136.8, 130.8, 130.2, 129.0, 128.6, 127.3, 115.8, 111.9, 81.4, 55.4, 39.3, 24.5, 16.8; HRMS m/z (ESI): calcd. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{NO}_{4} \mathrm{~S}$ (M-H)372.1264 , found 372.1264 .

2-(((1-(2-(benzylthio)-5-methoxyphenyl)ethylidene)amino)oxy)-2-methylpropano ic acid (1h)


Following the general procedure, the reaction of $\mathbf{S I - 1 h}$ (1 mmol, 272.2 mg ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( $2.4 \mathrm{mmol}, 196.9 \mathrm{mg}$ ) in EtOH ( 5 mL ) at reflux afforded the product $1 \mathrm{~h}(71 \%, 266.5 \mathrm{mg})$ as a white solid. ${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, CDCl $_{3}$ ) ס 7.33 - 7.05 (m, 6H), 6.88 6.59 (m, 2H), 3.91 (s, 2H), 3.78 (s, 3H), 2.20 (s, 3H), 1.58 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\mathrm{CDCl}_{3}$ ) б 178.1, 159.2, 159.1, 142.2, 137.8, 135.8, 128.9, 128.9, 128.3, 126.9, 124.2, 114.7, 81.4, 55.4, 41.2, 24.3, 17.1; HRMS m/z (ESI): calcd. for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{NO}_{4} \mathrm{~S}(\mathrm{M}-\mathrm{H})-$ 372.1264 , found 372.1259 .

## 2-(((1-(2-(benzylthio)-5-bromophenyl)ethylidene)amino)oxy)-2-methylpropanoic acid (1i)



Following the general procedure, the reaction of SI-1i (1 $\mathrm{mmol}, 320.0 \mathrm{mg}$ ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( $2.4 \mathrm{mmol}, 196.9 \mathrm{mg}$ ) in EtOH ( 5 mL ) at reflux afforded the product $\mathbf{1 i}(74 \%, 313.8 \mathrm{mg})$ as a brown oil. ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 7.35$ - 7.30 (m, 2H), 7.26 - 7.17 (m, 5H), 7.12 (d, J = $9.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.99$ (s, 2H), 2.19 (s, 3H), 1.57 (s, 6H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 178.3,157.1,140.6,136.9,134.1,132.5,131.9,131.8,128.8,128.5$, 127.2, 120.2, 81.5, 39.4, 24.3, 16.4; HRMS m/z (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{BrNO}_{3} \mathrm{~S}$ (M-H)420.0264, found 420.0277 .

## 2-(((1-(2-(benzylthio)-5-(trifluoromethyl)phenyl)ethylidene)amino)oxy)-2-methyl propanoic acid (1j)



Following the general procedure, the reaction of $\mathbf{S I - 1 j}$ ( 0.6 $\mathrm{mmol}, 179.8 \mathrm{mg}$ ), SI-11 ( $0.7 \mathrm{mmol}, 112.0 \mathrm{mg}$ ) and sodium acetate ( $1.4 \mathrm{mmol}, 118.1 \mathrm{mg}$ ) in EtOH ( 3 mL ) at reflux afforded the product $\mathbf{1 j}(89 \%, 211.8 \mathrm{mg})$ as a white solid. ${ }^{1} \mathbf{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) ס 10.12 (bs, 1H), 7.49 - 7.43
$(\mathrm{m}, 2 \mathrm{H}), 7.40-7.34(\mathrm{~m}, 1 \mathrm{H}), 7.31-7.24(\mathrm{~m}, 5 \mathrm{H}), 4.10(\mathrm{~s}, 2 \mathrm{H}), 2.26(\mathrm{~s}, 3 \mathrm{H}), 1.61(\mathrm{~s}$, 6 H ); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 179.4,156.3,141.1(\mathrm{q}, J=1.2 \mathrm{~Hz}), 137.8,136.3$, $128.8,128.7,128.6,127.7(q, J=32.8 H z), 127.4,125.6(q, J=3.5 H z), 125.4(q, J=$ $3.6 \mathrm{~Hz}), 123.9(\mathrm{q}, \mathrm{J}=271.9 \mathrm{~Hz}), 81.6,38.5,24.3,16.0 ;{ }^{9}$ F NMR (282 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta$ -62.43 (syn isomer), -62.48 (anti isomer); HRMS m/z (ESI): calcd. for $\mathrm{C}_{20} \mathrm{H}_{19} \mathrm{~F}_{3} \mathrm{NO}_{3} \mathrm{~S}$ (M-H)-410.1032, found 410.1026.

## 2-(((1-(6-(benzylthio)benzo[d][1,3]dioxol-5-yl)ethylidene)amino)oxy)-2-methylpr opanoic acid (1k)



Following the general procedure, the reaction of $\mathbf{S I - 1 k}$ (0.1 mmol, 37.5 mg ), SI-11 ( $0.12 \mathrm{mmol}, 18.7 \mathrm{mg}$ ) and sodium acetate ( $0.24 \mathrm{mmol}, 19.7 \mathrm{mg}$ ) in $\mathrm{EtOH}(0.5 \mathrm{~mL})$ at reflux afforded the product $1 \mathrm{k}(81 \%, 41.0 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 7.26-7.13(\mathrm{~m}, 5 \mathrm{H}), 6.79(\mathrm{~s}, 1 \mathrm{H}), 6.68(\mathrm{~s}, 1 \mathrm{H}), 5.94(\mathrm{~s}, 2 \mathrm{H})$, 3.93 (s, 2H), 2.15 (s, 3H), 1.56 (s, 6H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 178.0,158.9$, $148.1,147.2,137.5,134.6,128.8,128.4,127.1,126.6,113.4,109.2,101.7,81.4,41.1$, 24.4, 17.2; HRMS m/z (ESI): calcd. for $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{NO}_{5} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-} 386.1057$, found 386.1096.

2-(((1-(3-(benzyloxy)-2-(benzylthio)phenyl)ethylidene)amino)oxy)-2-methylpropa noic acid (1I)


Following the general procedure, the reaction of SI-1I (0.3 mmol, 99.9 mg ), SI-11 ( $0.36 \mathrm{mmol}, 56.0 \mathrm{mg}$ ) and sodium acetate ( $0.72 \mathrm{mmol}, 59.1 \mathrm{mg}$ ) in $\mathrm{EtOH}(1.5 \mathrm{~mL})$ at reflux afforded the product $11(59 \%, 75.6 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 8.63$ (bs, 1H), 7.58 - 7.51 (m, 2H), $7.48-7.38(\mathrm{~m}, 3 \mathrm{H}), 7.32-7.24(\mathrm{~m}, 1 \mathrm{H}), 7.22-7.15(\mathrm{~m}, 3 \mathrm{H}), 7.11-7.06(\mathrm{~m}, 2 \mathrm{H})$, $7.00(\mathrm{dd}, J=8.3,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.84(\mathrm{dd}, J=7.6,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{~s}, 2 \mathrm{H}), 4.06(\mathrm{~s}, 2 \mathrm{H})$, 2.07 ( $\mathrm{s}, 3 \mathrm{H}$ ), $1.60(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$ 177.7, 160.2, 159.8, 143.9, $138.3,136.7,129.6,129.0,128.8,128.4,128.3,128.2,127.4,127.0,121.6,113.2$, 81.5, 71.1, 39.1, 24.5, 17.6; HRMS m/z (ESI): calcd. for $\mathrm{C}_{26} \mathrm{H}_{26} \mathrm{NO}_{4} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-} 448.1577$, found 448.1596 .

## 2-(((1-(1-(benzylthio)naphthalen-2-yl)ethylidene)amino)oxy)-2-methylpropanoic acid (1m)



Following the general procedure, the reaction of $\mathbf{S I - 1 m}$ (1 $\mathrm{mmol}, 292.0 \mathrm{mg}$ ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( $2.4 \mathrm{mmol}, 196.9 \mathrm{mg}$ ) in $\mathrm{EtOH}(5 \mathrm{~mL})$ at reflux afforded the product $\mathbf{1 m}(78 \%, 305.3 \mathrm{mg})$ as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~ C D C l}{ }_{3}$ ) $\delta 8.60(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.88-7.81$ (m, 2H), 7.56 (pd, J $=6.8,1.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.30 (d, J=8.4 Hz, 1H), $7.21-7.09$ (m, 3H), 6.97 (dd, $J=6.8,2.9$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 3.93 (s, 2H), 2.19 (s, 3H), 1.61 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 177.3$, 160.0, 141.5, 137.2, 134.5, 133.4, 129.8, 129.2, 128.4, 128.0, 127.8, 126.9, 126.6, 126.2, 126.0, 125.5, 81.0, 41.2, 24.0, 17.4; HRMS m/z (ESI): calcd. for $\mathrm{C}_{23} \mathrm{H}_{22} \mathrm{NO}_{3} \mathrm{~S}$ (M-H)- 392.1326, found 392.1328 .

2-(((1-(3-(benzylthio)furan-2-yl)ethylidene)amino)oxy)-2-methylpropanoic acid (1n)


Following the general procedure, the reaction of $\mathbf{S I - 1 n}$ (1.3 mmol, 302.0 mg ), SI-11 ( $1.56 \mathrm{mmol}, 240.5 \mathrm{mg}$ ) and sodium acetate ( $3.1 \mathrm{mmol}, 256.0 \mathrm{mg}$ ) in $\mathrm{EtOH}(6.5 \mathrm{~mL})$ at reflux afforded the product $\mathbf{1 n}(81 \%, 349.6 \mathrm{mg})$ as a yellow solid. ${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~ C D C I ~}{ }_{3}$ ) $\delta 7.38$ (d, J = $1.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.35 - 7.20 (m, 5H), 6.40 (d, J = $1.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.05 (s, 2H), 2.23 (s, 3H), 1.60 (s, 6H); ${ }^{13}$ C NMR (75 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 178.0,150.0,145.2,142.7,137.2,128.6,128.4,127.1,118.8,113.2$, 81.8, 38.1, 24.1, 11.7. HRMS m/z (ESI): calcd. for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{NO}_{4} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-} 332.0962$, found 332.0960 .

2-(((1-(3-(benzylthio)thiophen-2-yl)ethylidene)amino)oxy)-2-methylpropanoic acid (10)


Following the general procedure, the reaction of SI-10 (0.6 $\mathrm{mmol}, 149.2 \mathrm{mg}$ ), SI-11 ( $0.72 \mathrm{mmol}, 112.1 \mathrm{mg}$ ) and sodium acetate ( $1.44 \mathrm{mmol}, 118.1 \mathrm{mg}$ ) in EtOH ( 3 mL ) at reflux afforded the product $\mathbf{1 0}(84 \%, 176.3 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR (300

MHz, CDCl $_{3}$ ) $\delta 10.03$ (bs, 1H), 7.32 - 7.19 (m, 6H), $6.95(\mathrm{~d}, \mathrm{~J}=5.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.09$ (s, 2 H ), 2.30 ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.62 ( $\mathrm{s}, 6 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 178.5,152.4,137.3$, 133.1, 132.7, 129.5, 128.8, 128.6, 127.3, 125.7, 81.9, 39.5, 24.3, 15.4; HRMS m/z (ESI): calcd. for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{NO}_{3} \mathrm{~S}_{2}(\mathrm{M}-\mathrm{H})^{-} 348.0723$, found 348.0746 .

2-(((1-(2-(benzylthio)pyridin-3-yl)ethylidene)amino)oxy)-2-methylpropanoic acid (1p)


Following the general procedure, the reaction of $\mathbf{S I - 1 p}$ (0.2 mmol, 56.3 mg ), SI-11 ( $0.24 \mathrm{mmol}, 37.4 \mathrm{mg}$ ) and sodium acetate ( $0.48 \mathrm{mmol}, 39.4 \mathrm{mg}$ ) in $\mathrm{EtOH}(1 \mathrm{~mL})$ at reflux afforded the product 1p $(77 \%, 61.7 \mathrm{mg})$ as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0}$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.92(\mathrm{bs}, 1 \mathrm{H}), 8.45(\mathrm{dd}, J=4.8,1.7 \mathrm{~Hz}, 1 \mathrm{H})$, 7.46 (dd, $J=7.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.41-7.35$ (m, 2H), $7.31-7.18$ (m, 3H), 7.00 (dd, $J=$ $7.6,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.41(\mathrm{~s}, 2 \mathrm{H}), 2.23(\mathrm{~s}, 3 \mathrm{H}), 1.62(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 179.0, 157.3, 155.2, 148.7, 138.2, 135.9, 131.2, 129.2, 128.3, 126.9, 119.0, 81.7, 35.2, 24.4, 15.2; HRMS m/z (ESI): calcd. for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ (M-H)- 343.1111, found 343.1099 .

2-(((1-(2-(benzylthio)quinolin-3-yl)ethylidene)amino)oxy)-2-methylpropanoic acid (1q)


Following the general procedure, the reaction of $\mathbf{S I - 1 q}$ ( $0.7 \mathrm{mmol}, 200.0 \mathrm{mg}$ ), SI-11 ( $0.8 \mathrm{mmol}, 130.7 \mathrm{mg}$ ) and sodium acetate ( $1.68 \mathrm{mmol}, 137.8 \mathrm{mg}$ ) in EtOH ( 3.5 mL ) at reflux afforded the product $\mathbf{1 q}(80 \%, 215.1 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 10.87$ (bs, 1H), 7.98 (d, J = $8.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.87 (s, 1H), $7.77-7.63(\mathrm{~m}, 2 \mathrm{H}), 7.55-7.46(\mathrm{~m}, 2 \mathrm{H}), 7.46-$ 7.38 (m, 1H), $7.35-7.18$ (m, 3H), 4.57 (s, 2H), 2.34 (s, 3H), 1.68 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 179.3,157.4,155.4,147.4,138.5,135.2,130.3,129.8,129.4$, 128.4, 127.9, 127.8, 126.9, 125.7, 125.3, 81.8, 35.1, 24.5, 15.7; HRMS m/z (ESI): calcd. for $\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-}$393.1278, found 393.1285.

## 2-(((2-(benzylthio)benzylidene)amino)oxy)-2-methylpropanoic acid (1r)



Following the general procedure, the reaction of $\mathbf{S I - 1 r}(1 \mathrm{mmol}$, 228.1 mg ), $\mathbf{~ S I - 1 1 ~ ( 1 . 2 ~ m m o l , ~} 186.7 \mathrm{mg}$ ) and sodium acetate ( $2.4 \mathrm{mmol}, 196.9 \mathrm{mg}$ ) in $\mathrm{EtOH}(5 \mathrm{~mL})$ at reflux afforded the product $1 \mathbf{r}(88 \%, 290.9 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz, CDCl $_{3}$ ) $\delta 11.29$ (bs, 1H), $8.63(\mathrm{~s}, 1 \mathrm{H}), 7.73(\mathrm{dd}, \mathrm{J}=7.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.32(\mathrm{dd}, \mathrm{J}$ $=7.7,1.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.28-7.05(\mathrm{~m}, 7 \mathrm{H}), 3.98(\mathrm{~s}, 2 \mathrm{H}), 1.60(\mathrm{~s}, 6 \mathrm{H}){ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) б 179.6, 148.3, 136.9, 135.6, 133.1, 132.7, 130.1, 128.8, 128.4, 127.3, 127.2, 127.2, 81.6, 40.3, 24.0; HRMS m/z (ESI): calcd. for $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{NO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-}$328.1013, found 328.1015.

## 2-((((2-(benzylthio)phenyl)(phenyl)methylene)amino)oxy)-2-methylpropanoic acid (1s)



Following the general procedure, the reaction of $\mathbf{S I - 1 s}(1 \mathrm{mmol}$, 304.1 mg ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( 2.4 $\mathrm{mmol}, 196.9 \mathrm{mg}$ ) in $\mathrm{MeOH}(5 \mathrm{~mL})$ at reflux afforded the product 1s $(67 \%, 272.8 \mathrm{mg})$ as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~} \mathrm{CDCl}_{3}$ ) б $7.45-7.25$ (m, 9H), $7.24-7.10(\mathrm{~m}, 5 \mathrm{H}), 4.00(\mathrm{~d}, \mathrm{~J}=9.8 \mathrm{~Hz}$, 2H), 1.55 (d, J = 4.4 Hz, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 176.2, 157.6, 137.1, 135.6, 134.8, 133.7, 131.6, 129.9, 129.1, 129.0, 128.9, 128.4, 128.4, 127.3, 127.2, 127.0, 82.7, 39.1, 24.3; HRMS m/z (ESI): calcd. for $\mathrm{C}_{24} \mathrm{H}_{22} \mathrm{NO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-} 404.1326$, found 404.1325

2-(((1-(2-(benzylselanyl)phenyl)ethylidene)amino)oxy)-2-methylpropanoic acid (1t)


Following the general procedure, the reaction of $\mathbf{S I - 1 0 a}$ ( 1 mmol , 289.2 mg ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( 2.4 $\mathrm{mmol}, 196.9 \mathrm{mg}$ ) in $\mathrm{MeOH}(5 \mathrm{~mL})$ at reflux afforded the product 1 t ( $76 \%, 295.5 \mathrm{mg}$ ) as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.57-7.47(\mathrm{~m}, 1 \mathrm{H}), 7.37-7.20(\mathrm{~m}, 8 \mathrm{H}), 4.08(\mathrm{~s}, 2 \mathrm{H})$,
132.6, 131.1, 129.2, 128.9, 128.7, 128.4, 126.8, 126.5, 81.5, 32.3, 24.5, 16.1; HRMS $\mathrm{m} / \mathrm{z}$ (ESI): calcd. for $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{NO}_{3} \mathrm{Se}(\mathrm{M}-\mathrm{H})^{-3} 390.0614$, found 390.0623.

2-(((2-(benzylselanyl)-5-chlorobenzylidene)amino)oxy)-2-methylpropanoic acid (1u)


Following the general procedure, the reaction of $\mathbf{S I - 1 0 b}$ (1 $\mathrm{mmol}, 309.7 \mathrm{mg}$ ), SI-11 ( $1.2 \mathrm{mmol}, 186.7 \mathrm{mg}$ ) and sodium acetate ( $2.4 \mathrm{mmol}, 196.9 \mathrm{mg}$ ) in $\mathrm{MeOH}(5 \mathrm{~mL})$ at reflux afforded the product $1 \mathbf{u}(65 \%, 268.2 \mathrm{mg})$ as a yellow solid.
${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, CDCl $_{3}$ ) $\delta 8.53(\mathrm{~s}, 1 \mathrm{H}), 7.75(\mathrm{~d}, \mathrm{~J}=2.4 \mathrm{~Hz}$, 1H), 7.45 (d, J = $8.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.33-7.16$ (m, 6H), 4.04 (s, 2H), 1.68 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 178.6,149.3,137.6,136.8,135.6,134.2,130.0,129.4,128.8$, 128.5, 127.3, 127.1, 82.0, 33.1, 24.1; HRMS m/z (ESI): calcd. for $\mathrm{C}_{18} \mathrm{H}_{17} \mathrm{CINO}_{3} \mathrm{Se}$ (M-H)- 410.0068, found 410.0072 .

Procedure for the synthesis of oxime 1a':


Compound $\mathbf{S I}-12$ was prepared following a reported procedure in the literature. ${ }^{[18]}$
Compound 1a' was prepared following a slightly modified procedure reported in the literature. ${ }^{[18]} \mathrm{A}$ solution of ketone (1.0 equiv) in $\mathrm{EtOH}(0.2 \mathrm{M})$ was treated with 2-(aminooxy)propanoic acid hydrochloride (SI-12) (1.2 equiv), sodium acetate (2.4 equiv) and heated to reflux until completion by TLC analysis. The mixture was then allowed to cool to room temperature and water was added. The resulting aqueous solution was extracted with EtOAc (3x) and the combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and concentrated under vacuum to provide the product. Then, the residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure product 1a'.

## 2-(((1-(2-(methylthio)phenyl)ethylidene)amino)oxy)propanoic acid (1a')



Following the procedure, the reaction of SI-1a (1 mmol, 166.2 mg ), SI-12 ( $1.2 \mathrm{mmol}, 168.6 \mathrm{mg}$ ) and sodium acetate ( 2.4 mmol , 196.9 mg ) in EtOH ( 5 mL ) at reflux afforded the product 1a' ( $78 \%, 198.0 \mathrm{mg}$ ) as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 10.19 (s, 1H), $7.27-7.00(\mathrm{~m}, 4 \mathrm{H}), 4.77$ (q, J = 7.0 Hz, 1H), 2.31 (s, 3H), 2.21 (s, 3H), 1.51 (d, J = $7.0 \mathrm{~Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz},\left(\mathrm{CDCl}_{3}\right) \delta 178.4$, 158.4, 137.5, 136.6, 129.2, 128.8, 127.1, 125.1, 21.4, 16.9, 16.8, 16.3; HRMS m/z (ESI): calcd. for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{NO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-2} 252.0700$, found 252.0704 .

Procedure for the synthesis of oxime 1a":


Compound 1a" was prepared following a slightly modified procedure reported in the literature ${ }^{[19]}$ A solution of $\mathbf{S I - 1 a}(0.8 \mathrm{mmol}, 139.2 \mathrm{mg}, 1.0$ equiv) in EtOH ( 0.5 M ) was treated with hydroxylamine hydrochloride ( $1.3 \mathrm{mmol}, 93.1 \mathrm{mg}, 1.6$ equiv), sodium acetate ( $1.7 \mathrm{mmol}, 137.4 \mathrm{mg}, 2.0$ equiv) and heated to reflux for 12 h . The mixture was then allowed to cool to room temperature and concentrated under vacuum to provide the product $\mathbf{S I - 1 3}$ as a white solid, which was used for the next step without purification. Then, SI-13 was added dropwise to a stirred suspension of $\mathrm{NaH} 60 \%$ in mineral oil ( $2.5 \mathrm{mmol}, 60.3 \mathrm{mg}, 3.0$ equiv) in dry THF ( 1.5 M ) at $0^{\circ} \mathrm{C}$ (ice bath) under inert atmosphere and the mixture was stirred for 15 min . Afterwards, a solution of 2-bromo-2-phenylacetic acid ( $0.9 \mathrm{mmol}, 198.1 \mathrm{mg}, 1.1$ equiv) in dry THF ( 0.5 M ) was added and the reaction was heated to $75^{\circ} \mathrm{C}$ for 12 h . After this time, the reaction was cooled to room temperature and concentrate hydrochloric acid was added until pH 2. The resulting solution was extracted with EtOAc (3x) and the combined organic layers were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated under vacuum to provide the
product. Then, the residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure product 1a".

## 2-(((1-(2-(methylthio)phenyl)ethylidene)amino)oxy)-2-phenylacetic acid (1a")



Following the procedure, the reaction of $\mathbf{S I - 1 a}(0.8 \mathrm{mmol}, 139.2$ mg ) afforded the product 1a" (40\% for two reaction steps, 106.5 mg ) as white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 11.00$ (bs, 1H), $7.66-7.58(\mathrm{~m}, 2 \mathrm{H}), 7.47-7.40(\mathrm{~m}, 3 \mathrm{H}), 7.37-7.26$ (m, 3H), 7.23 - 7.16 (m, 1H), 5.81 (s, 1H), 2.41 (s, 3H), 2.37 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz},\left(\mathrm{CDCl}_{3}\right)$ б 176.7, 159.1, 137.6, 136.5, 134.7, 129.3, 129.1, 128.9, 128.7, 127.7, 127.2, 125.1, 83.2, 16.9, 16.6; HRMS m/z (ESI): calcd. for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{NO}_{3} \mathrm{~S}(\mathrm{M}-\mathrm{H})^{-3} 314.0856$, found 314.0856 .

## 3. Optimization of the Reaction Conditions

General procedure for the optimization of the reaction conditions of the synthesis of isothiazoles 2:

In an oven-dried glass vial equipped with a stirring bar, 2-methyl-2-(((1-(2-(methylthio)phenyl)ethylidene)amino)oxy)propanoic acid (1a, 0.05 $\mathrm{mmol}, 13.4 \mathrm{mg}$, 1 equiv), the base ( $0.05 \mathrm{mmol}, 1$ equiv), the photocatalyst ( $2.5 \cdot 10^{-3}$ $\mathrm{mmol}, 0.05$ equiv) and the respective solvent $(0.67 \mathrm{~mL})$ were added under air. Then, the vial was closed with a PTFE/rubber septum perforated several times to enable the air enter into the reaction. Afterwards, the mixture was stirred and irradiated at 450 nm using 380 mW single LEDs in a custom-made temperature-controlled system (see Figure S1) for 16 h at $20^{\circ} \mathrm{C}$. Then, the volatiles were removed and yield was determined by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ analysis using 1,3,5-trimethoxybenzene ( $8.4 \mathrm{mg}, 0.05 \mathrm{mmol}$ ) as internal standard.

Table S1. Optimization of reaction conditions with 1a.



| entry | PC | base | solvent | yield (\%) ${ }^{[a]}$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | PC1 | CsF | DCE | 17 |
| 2 | PC1 | CsF | DCM | 20 |
| 3 | PC1 | CsF | THF | 9 |
| 4 | PC1 | CsF | DMF | 21 |
| 5 | PC1 | CsF | MeCN | 10 |
| 6 | PC1 | CsF | Acetone | 20 |
| 7 | PC1 | $\mathrm{Cs}_{2} \mathrm{CO}_{3}$ | Acetone | 26 |
| 8 | PC1 | $\mathrm{K}_{2} \mathrm{CO}_{3}$ | Acetone | 26 |
| 9 | PC1 | $\mathrm{Na}_{2} \mathrm{CO}_{3}$ | Acetone | 53 |
| 10 | PC1 | $\mathrm{Li}_{2} \mathrm{CO}_{3}$ | Acetone | 9 |
| 11 | PC1 | NaF | Acetone | 17 |
| 12 | PC1 | $\mathrm{Na}_{2} \mathrm{HPO}_{4}$ | Acetone | 41 |


| 13 | PC1 | $\mathrm{NaHCO}_{3}$ | Acetone | 23 |
| :--- | :---: | :--- | :--- | :--- |
| 14 | PC1 | NaOAc | Acetone | 56 |
| 15 | PC2 | NaOAc | Acetone | 35 |
| 16 | PC3 | NaOAc | Acetone | 31 |
| 17 | PC4 | NaOAc | Acetone | 32 |
| 18 | PC5 | NaOAc | Acetone | 44 |
| $19^{[b]}$ | PC1 | NaOAc | Acetone | 54 |
| $22^{[c]}$ | PC1 | NaOAc | Acetone | 4 |
| 21 | - | NaOAc | Acetone | 0 |
| 22 | PC1 | - | Acetone | 0. |
| $23^{[d]}$ | PC1 | NaOAc | Acetone | 0. |

${ }^{[a]}$ The yield was determined from the crude ${ }^{1} \mathrm{H}-\mathrm{NMR}$. ${ }^{[b]}$ Oxygen was used instead air. ${ }^{[c]}$ Argon was used instead air. ${ }^{[d]}$ The reaction was performed in the dark. DCE $=$ 1,2-Dichloroethane. DCM = Dichloromethane, THF = Tetrahydrofuran, DMF = $\mathrm{N}, \mathrm{N}$-Dimethylformamide.

## 4. Optimization of the $\alpha$-Imino-oxy Acid Structure

General procedure for the optimization of the a-imino-oxy acid structure in the synthesis of isothiazoles 2 :

In an oven-dried glass vial equipped with a stirring bar, the corresponding $\alpha$-imino-oxy acid 1 ( $0.05 \mathrm{mmol}, 1$ equiv), NaOAc ( $0.05 \mathrm{mmol}, 4.1 \mathrm{mg}, 1$ equiv), photocatalyst PC1, 9-mesityl-10-methyl acridinium perchlorate, ( $2.5 \cdot 10^{-3} \mathrm{mmol}, 1.05$ $\mathrm{mg}, 0.05$ equiv) and acetone ( 0.67 mL ) were added under air. Then, the vial was closed with a PTFE/rubber septum perforated several times to enable the air enter into the reaction. Afterwards, the mixture was stirred and irradiated at 450 nm using 380 mW single LEDs in a custom-made temperature-controlled system (see Figure $\mathrm{S} 1)$ for 16 h at $20^{\circ} \mathrm{C}$. Then, the volatiles were removed and yield was determined by ${ }^{1} \mathrm{H}$-NMR analysis using 1,3,5-trimethoxybenzene ( $8.4 \mathrm{mg}, 0.05 \mathrm{mmol}$ ) as internal standard.

Table S2. Optimization of $\alpha$-imino-oxy acid structure.


| entry | $\mathbf{R}_{\mathbf{1}}$ | $\mathbf{R}_{\mathbf{2}}$ | $\mathbf{R}_{\mathbf{3}}$ | yield (\%) ${ }^{[\mathrm{ar]}}$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | Me | H | Me | 33 |
| 2 | Me | Me | Me | 56 |
| 3 | Ph | H | Me | 58 |
| 4 | Me | Me | $i-\mathrm{Pr}$ | 44 |
| 5 | Me | Me | $t-\mathrm{Bu}$ | 68 |
| 6 | Me | Me | Ph | 4 |
| 7 | Me | Me | Bn | $\mathbf{8 5 ( 8 1 ) ^ { [ b ] }}$ |
| $8^{[\mathrm{cc]}}$ | Me | Me | Bn | 85 |

[a] The yield was determined from the crude ${ }^{1} \mathrm{H}-\mathrm{NMR}$. ${ }^{[b]}$ Isolated yield in 0.1 mmol scale in parentheses. [c] The reaction was performed with a blue LED spot of 40 W instead of 380 mW single LEDs.

## 5. General Procedure and Characterization of Isothiazoles 2



General procedure for the synthesis of isothiazoles 2:

In an oven-dried glass vial equipped with a stirring bar, 1 ( $0.1 \mathrm{mmol}, 1$ equiv), NaOAc ( $0.1 \mathrm{mmol}, 8.2 \mathrm{mg}, 1$ equiv), photocatalyst PC1, 9-mesityl-10-methyl acridinium perchlorate, ( $5 \cdot 10^{-3} \mathrm{mmol}, 2.1 \mathrm{mg}, 0.05$ equiv) and acetone ( 1.33 mL ) were
added under air. Then, the vial was closed with a PTFE/rubber septum perforated several times to enable the air enter into the reaction. Afterwards, the mixture was stirred and irradiated at 450 nm using 380 mW single LEDs in a custom-made temperature-controlled system (see Figure S1) for 16 h at $20^{\circ} \mathrm{C}$. Then, the volatiles were removed and the analytically pure product 2 was obtained by flash chromatography (latrobeads silica gel; cyclohexane/EtOAc).

## 3-methylbenzo[d]isothiazole (2a)



Following the general procedure, the reaction of $1 \mathbf{d}(0.1 \mathrm{mmol}, 34.0$ mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1$ $\mathrm{mg})$ in acetone $(1.33 \mathrm{~mL})$ irradiated at 450 nm afforded the product $\mathbf{2 a}$ $(81 \%, 11.9 \mathrm{mg})$ as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z , ~} \mathrm{CDCl}_{3}$ ) $\delta 7.97-7.90(\mathrm{~m}, 2 \mathrm{H}), 7.56$ - $7.49(\mathrm{~m}, 1 \mathrm{H}), 7.48-7.40(\mathrm{~m}, 1 \mathrm{H}), 2.76(\mathrm{~s}, 3 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[20]}$

## 6-fluoro-3-methylbenzo[d]isothiazole (2b)



Following the general procedure, the reaction of $1 \mathbf{f}(0.1 \mathrm{mmol}, 35.4$ mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( $5 \mathrm{~mol} \%$, 2.1 mg ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product 2b ( $82 \%, 13.4 \mathrm{mg}$ ) as a white solid. ${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C}_{6} \mathrm{D}_{6}\right) \delta 7.05(\mathrm{dd}, \mathrm{J}=$ 8.8, 4.9 Hz, 1H), 6.86 (dd, $J=8.5,2.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), $6.70(\mathrm{td}, J=8.7,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.26$ (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 162.6(\mathrm{~d}, \mathrm{~J}=249.6 \mathrm{~Hz}$ ), 162.1, $154.2(\mathrm{~d}, \mathrm{~J}=10.4 \mathrm{~Hz})$, 132.4, $124.9(\mathrm{~d}, J=10.2 \mathrm{~Hz}), 113.6(\mathrm{~d}, J=25.4 \mathrm{~Hz}), 105.9(\mathrm{~d}, J=25.6 \mathrm{~Hz}), 17.0 ;{ }^{19} \mathrm{~F}$ NMR (282 MHz, $\mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta$-113.77; HRMS m/z (ESI): calcd. for $\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{FNS}(\mathrm{M})^{+}$167.0205, found 167.0200.

## 6-methoxy-3-methylbenzo[d]isothiazole (2c)



Following the general procedure, the reaction of $1 \mathrm{~g}(0.1 \mathrm{mmol}$, 37.1 mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg})$ and photocatalyst PC1 (5 $\mathrm{mol} \%, 2.1 \mathrm{mg}$ ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 c}(89 \%, 15.8 \mathrm{mg})$ as a white solid. ${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C}_{6} \mathrm{D}_{6}\right) \delta 7.26(\mathrm{~d}, \mathrm{~J}$ $=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{dd}, J=8.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~d}, \mathrm{~J}=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.23(\mathrm{~s}, 3 \mathrm{H})$,
2.36 (s, 3H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta$ 162.1, 160.0, 155.1, 130.3, 124.2, 115.9, 101.1, 55.1, 17.1; HRMS m/z (ESI): calcd. for $\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{NOS}(\mathrm{M}+\mathrm{H})^{+}$180.0478, found 180.0480.

## 5-methoxy-3-methylbenzo[d]isothiazole (2d)

Following the general procedure, the reaction of $\mathbf{1 h}(0.1 \mathrm{mmol}$,
 37.3 mg ), NaOAc ( $0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 (5 $\mathrm{mol} \%, 2.1 \mathrm{mg}$ ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 d}(75 \%, 13.4 \mathrm{mg})$ as a white solid. ${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{3 0 0} \mathbf{~ M H z},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 7.96$ (d, J = $8.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.52(\mathrm{~d}, \mathrm{~J}=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{dd}, \mathrm{J}=8.8,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.93(\mathrm{~s}, 3 \mathrm{H})$, 2.67 (s, 3H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) ~ \delta 163.1,159.1,146.0,137.5,121.7,120.0$, 105.4, 56.2, 17.6; HRMS m/z (ESI): calcd. for $\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{NOS}(\mathrm{M}+\mathrm{H})^{+}$180.0478, found 180.0482.

## 5-bromo-3-methylbenzo[d]isothiazole (2e)



Following the general procedure, the reaction of $\mathbf{1 i}(0.1 \mathrm{mmol}, 42.0$ mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( $5 \mathrm{~mol} \%$, 2.1 mg ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 e}(68 \%, 15.6 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 8.28(\mathrm{~s}$, 1 H ), 8.09 (d, J = $8.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.72 (d, J = $8.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.73 ( $\mathrm{s}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 163.2,152.1,137.9,131.2,127.4,122.9,119.2,17.5 ;$ HRMS m/z (ESI): calcd. for $\mathrm{C}_{8} \mathrm{H}_{6} \mathrm{BrNS}(\mathrm{M})^{+}$227.9477, found 226.9404.

## 3-methyl-5-(trifluoromethyl)benzo[d]isothiazole (2f)

Following the general procedure, the reaction of $1 \mathbf{j}(0.1 \mathrm{mmol}$,
 40.6 mg ), NaOAc ( $0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 (5 $\mathrm{mol} \%, 2.1 \mathrm{mg}$ ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $2 \mathrm{f}(71 \%, 15.2 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H} \mathbf{N M R}\left(\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D}_{2} \mathrm{Cl}_{2}\right) \delta 8.28$ $8.22(\mathrm{~m}, 1 \mathrm{H}), 8.07(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.74(\mathrm{dd}, J=8.5,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.78(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, CD $\mathbf{C l}_{2}$ ) $\delta 163.9,155.8,135.4,127.6(q, J=32.6 \mathrm{~Hz}), 125.0(\mathrm{q}, \mathrm{J}=$ $272.0 \mathrm{~Hz}), 124.2(\mathrm{q}, J=3.2 \mathrm{~Hz}), 121.5(\mathrm{q}, J=4.3 \mathrm{~Hz}), 121.4,17.8 ;{ }^{19} \mathrm{~F}$ NMR (282 MHz, $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ ) $\delta$-61.17; HRMS m/z (ESI): calcd. for $\mathrm{C}_{9} \mathrm{H}_{6} \mathrm{~F}_{3} \mathrm{NS}(\mathrm{M})^{+}$217.0173, found

## 3-methyl-[1,3]dioxolo[4',5':4,5]benzo[1,2-d]isothiazole (2g)

Following the general procedure, the reaction of $\mathbf{1 k}(0.1 \mathrm{mmol}$,
 36.0 mg ), NaOAc ( $0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( 5 $\mathrm{mol} \%, 2.1 \mathrm{mg}$ ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 g}(66 \%, 11.8 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 6.77(\mathrm{~s}$, 1H), 6.65 (s, 1H), 5.25 (s, 2H), $2.27(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 161.4,149.8$, 148.7, 147.4, 130.8, 102.0, 101.2, 98.6, 17.1; HRMS m/z (ESI): calcd. for $\mathrm{C}_{9} \mathrm{H}_{8} \mathrm{NO}_{2} \mathrm{~S}$ $(\mathrm{M}+\mathrm{H})^{+}$194.0270, found 194.0265.

## 7-(benzyloxy)-3-methylbenzo[d]isothiazole (2h)



Following the general procedure, the reaction of $\mathbf{1 I}(0.1 \mathrm{mmol}, 44.8$
mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1$ mg ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 h}$ $(88 \%, 22.4 \mathrm{mg})$ as a white solid. ${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CD}_{2} \mathrm{Cl}_{2}$ ) $\delta 7.55$ (dd, $J=8.0,0.8$ $\mathrm{Hz}, 1 \mathrm{H}), 7.52-7.46(\mathrm{~m}, 2 \mathrm{H}), 7.46-7.33(\mathrm{~m}, 4 \mathrm{H}), 6.98(\mathrm{~d}, \mathrm{~J}=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.29(\mathrm{~s}$, 2H), 2.71 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $\mathbf{7 5} \mathrm{MHz}, \mathrm{CD}_{2} \mathrm{Cl}_{2}$ ) $\delta$ 163.7, 152.6, 143.0, 137.9, 137.0, 129.2, 128.7, 128.0, 127.0, 116.4, 108.8, 71.1, 17.9; HRMS m/z (ESI): calcd. for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{NOS}(\mathrm{M}+\mathrm{H})^{+}$256.0791, found 256.0800

## 3-methyInaphtho[2,1-ণ]isothiazole (2i)



Following the general procedure, the reaction of $\mathbf{1 m}(0.1 \mathrm{mmol}, 39.3$
mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1$ mg ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 i}(70 \%, 14.3 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta$ $8.24-8.17(\mathrm{~m}, 1 \mathrm{H}), 8.16-8.10(\mathrm{~m}, 1 \mathrm{H}), 8.00(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.92(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, 1H), 7.76 - 7.68 (m, 2H), 2.79 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) ~ \delta ~ 164.7, ~ 153.9$, 133.9, 133.1, 130.0, 129.1, 128.4, 127.4, 127.3, 126.1, 121.6, 17.7; HRMS m/z (ESI): calcd. for $\mathrm{C}_{12} \mathrm{H}_{9} \mathrm{NS}(\mathrm{M})^{+}$199.0456, found 199.0465 .

## 3-methylfuro[2,3-વ]isothiazole (2j)

Following the general procedure, the reaction of $\mathbf{1 n}(0.1 \mathrm{mmol}, 33.3 \mathrm{mg})$, $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg})$ and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1 \mathrm{mg}$ ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product 2 j ( $32 \%$, $4.5 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 7.91$ (dd, $J=3.7,2.1 \mathrm{~Hz}$, 1 H ), 7.01 (dd, $J=3.7,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.50(\mathrm{~d}, \mathrm{~J}=3.7 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 126 MHz , $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 152.0,148.3,146.4,135.3,107.4,15.8$; HRMS m/z (ESI): calcd. for $\mathrm{C}_{6} \mathrm{H}_{6} \mathrm{NOS}(\mathrm{M}+\mathrm{H})^{+} 140.0165$, found 140.0169 .

## 3-methylthieno[2,3-d]isothiazole (2k)



Following the general procedure, the reaction of $10(0.1 \mathrm{mmol}, 35.0 \mathrm{mg})$, $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg})$ and photocatalyst PC1 $(5 \mathrm{~mol} \%, 2.1 \mathrm{mg})$ in acetone $(1.33 \mathrm{~mL})$ irradiated at 450 nm afforded the product $\mathbf{2 k}(36 \%$, 5.6 mg ) as a brown oil. ${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z , ~ ( C D} \mathbf{H}_{\mathbf{2}} \mathbf{C O}\right) \delta 7.89(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.48$ $(\mathrm{d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.57(\mathrm{~s}, 3 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[21]}$

## 3-methylisothiazolo[5,4-b]pyridine (2I)



Following the general procedure, the reaction of $\mathbf{1 p}(0.1 \mathrm{mmol}, 34.2$
mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1$ mg ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 l}$ $(86 \%, 12.8 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 8.40(\mathrm{dd}, J=4.5,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 7.19$ (dd, $J=8.2,1.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.52 (dd, $J=8.2,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.19(\mathrm{~s}, 3 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[22]}$

## 3-methylisothiazolo[5,4-b]quinoline (2m)



Following the general procedure, the reaction of $1 \mathbf{q}(0.1 \mathrm{mmol}$, 39.2 mg ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 (5 $\mathrm{mol} \%, 2.1 \mathrm{mg}$ ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 m}(69 \%, 13.8 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR $\left(\mathbf{3 0 0} \mathbf{~ M H z},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 12$ (s, 1H), 8.22 (d, J = $8.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 8.13 (d, J = $8.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.99-7.90(\mathrm{~m}, 1 \mathrm{H}), 7.72-$ $7.64(\mathrm{~m}, 1 \mathrm{H}), 2.83(\mathrm{~s}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $\left.75 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 171.3,163.9,149.2$, 133.9, 132.8, 130.6, 129.3, 127.5, 126.9, 125.7, 18.3; HRMS m/z (ESI): calcd. for
$\mathrm{C}_{11} \mathrm{H}_{9} \mathrm{~N}_{2} \mathrm{~S}(\mathrm{M}+\mathrm{H})^{+}$201.0481, found 201.0481.

## benzo[0]isothiazole (2n)



Following the general procedure, the reaction of $1 \mathbf{r}(0.1 \mathrm{mmol}, 32.9 \mathrm{mg})$, $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg})$ and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1 \mathrm{mg}$ ) in acetone $(1.33 \mathrm{~mL})$ irradiated at 450 nm afforded the product $\mathbf{2 n}(27 \%, 3.6 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 8.93(\mathrm{~s}, 1 \mathrm{H}), 8.08(\mathrm{~d}, \mathrm{~J}=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, 7.98 (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.58-7.50(\mathrm{~m}, 1 \mathrm{H}), 7.45$ (ddd, $J=8.0,7.0,1.0 \mathrm{~Hz}, 1 \mathrm{H})$. Spectral data is in accordance with the literature. ${ }^{[23]}$

## 3-phenylbenzo[d]isothiazole (20)



Following the general procedure, the reaction of $1 \mathrm{~s}(0.1 \mathrm{mmol}, 28.3$ mg ), NaOAc ( $0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1$ mg ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 o}$ $(85 \%, 17.9 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 8.27(\mathrm{~d}, \mathrm{~J}=8.1 \mathrm{~Hz}$, $1 \mathrm{H}), 8.24$ (d, J = $8.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.93 (dd, $J=8.0,1.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.70-7.64$ (m, 2H), 7.63
 129.9, 129.7, 128.8, 126.4, 125.7, 121.4; HRMS m/z (ESI): calcd. for $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{NS}$ $(\mathrm{M}+\mathrm{H})^{+} 212.0528$, found 212.0534 .

## 3-methylbenzo[d][1,2]selenazole (2p)



Following the general procedure, the reaction of $\mathbf{1 t}(0.1 \mathrm{mmol}, 39.0 \mathrm{mg})$, $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg})$ and photocatalyst PC1 ( $5 \mathrm{~mol} \%, 2.1 \mathrm{mg}$ ) in acetone $(1.33 \mathrm{~mL})$ irradiated at 450 nm afforded the product $\mathbf{2 p}(82 \%$, $16.1 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z , ~}\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 8.19(\mathrm{~d}, \mathrm{~J}=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, 8.03 (dd, $J=8.0,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.60-7.54(\mathrm{~m}, 1 \mathrm{H}), 7.51$ (ddd, $J=8.0,7.1,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, 2.65 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}$ ) $\delta 167.6,153.0,139.4,128.8,126.9,125.9$, 125.2, 20.1; HRMS m/z (ESI): calcd. for $\mathrm{C}_{8} \mathrm{H}_{8} \mathrm{NS}(\mathrm{M}+\mathrm{H})^{+}$197.9816, found 197.9819 .

## 5-chlorobenzo[d][1,2]selenazole (2q)



Following the general procedure, the reaction of $\mathbf{1 u}(0.1 \mathrm{mmol}$, 41.1 mg ), NaOAc ( $0.1 \mathrm{mmol}, 8.2 \mathrm{mg}$ ) and photocatalyst PC1 (5 $\mathrm{mol} \%, 2.1 \mathrm{mg}$ ) in acetone ( 1.33 mL ) irradiated at 450 nm afforded the product $\mathbf{2 q}$
( $60 \%, 13.0 \mathrm{mg}$ ) as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathbf{3 0 0} \mathbf{~ M H z , ~}\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 9.47(\mathrm{~s}, 1 \mathrm{H}), 8.27$ (d, $J=8.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), $8.24(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.57(\mathrm{dd}, J=8.6,2.0 \mathrm{~Hz}, 1 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( 75 MHz, (CD $\left.3_{2}\right)_{2} C O$ ) $\overline{160.6}, 151.7,142.2,132.0,129.1,126.7,126.6 ;$ HRMS m/z (ESI): calcd. for $\mathrm{C}_{7} \mathrm{H}_{4} \mathrm{ClNSe}(\mathrm{M})^{+}$216.9197, found 216.9202.

## 6. Synthesis of Brassilexin Derivative 4



Following the general procedure for the synthesis of sulfides, benzyl mercaptan ( $0.53 \mathrm{mmol}, 0.06 \mathrm{~mL}, 1.05$ equiv) was added dropwise for 0.25 h to a stirred suspension of $\mathrm{NaH}\left(60 \%, 0.6 \mathrm{mmol}, 24 \mathrm{mg}, 1.2\right.$ equiv) in DMF $(1.1 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ (ice bath). Afterwards, a solution of the 3-acetyl-2-chloro-1-methyl-1H-indole ( 0.5 mmol , 103.8 mg , 1 equiv) in DMF ( 0.25 mL ) was added and the reaction was heated to $75^{\circ} \mathrm{C}$ for 12 h . After this time, the reaction was cooled to room temperature, diluted with dichloromethane and washed with water ( 3 x ) and brine. The organic phase was dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Then, the solid residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure product 1-(2-(benzylthio)-1-methyl-1H-indol-3-yl)ethan-1-one.

## 1-(2-(benzylthio)-1-methyl-1H-indol-3-yl)ethan-1-one (SI-1t)



Following procedure, the reaction of 3-acetyl-2-chloro-1-methyl-1H-indole ( $0.5 \mathrm{mmol}, 103.8 \mathrm{mg}$ ), $\mathrm{NaH}(60 \%, 0.6 \mathrm{mmol}, 24$ mg ) and benzyl mercaptan ( $0.53 \mathrm{mmol}, 0.06 \mathrm{~mL}$ ) in DMF ( 1.35 mL ) at $75^{\circ} \mathrm{C}$ afforded the product $\mathbf{S I - 1 t}(89 \%, 131.2 \mathrm{mg})$ as a pale yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.39-8.26(\mathrm{~m}, 1 \mathrm{H}), 7.35-7.23(\mathrm{~m}, 4 \mathrm{H}), 7.19$ (d, J $=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.99(\mathrm{~d}, \mathrm{~J}=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.04(\mathrm{~s}, 2 \mathrm{H}), 3.52(\mathrm{~s}, 3 \mathrm{H}), 2.74(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$

NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 194.8,137.4,137.1,135.8,128.7,128.6,127.6,126.5,123.9$, 122.6, 122.5, 121.4, 109.9, 42.4, 30.6, 28.4 .

Following the general procedure for the synthesis of oximes, a solution of 1-(2-(benzylthio)-1-methyl-1H-indol-3-yl)ethan-1-one (SI-1t, $0.44 \mathrm{mmol}, 131.2 \mathrm{mg}, 1$ equiv) in EtOH ( 2.2 mL ) was treated with 2-(aminooxy)-2-methylpropanoic acid hydrochloride (SI-11, $0.53 \mathrm{mmol}, 82.2 \mathrm{mg}, 1.2$ equiv), sodium acetate $(1.06 \mathrm{mmol}$, $86.6 \mathrm{mg}, 2.4$ equiv) and heated to reflux until completion by TLC analysis. The mixture was then allowed to cool to room temperature and water was added. The resulting aqueous solution was extracted with EtOAc (3x) and the combined organic layers were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and concentrated under vacuum to provide the product. Then, the residue was subjected to flash chromatography (silica gel, cyclohexane/EtOAc) to give the analytically pure product 3.

## 2-(((1-(2-(benzylthio)-1-methyl-1H-indol-3-yl)ethylidene)amino)oxy)-2-methylpro panoic acid (3)



Following the procedure, the reaction of $\mathbf{S I - 1 t}(0.44 \mathrm{mmol}$, 131.2 mg ), SI-11 ( $0.53 \mathrm{mmol}, 82.2 \mathrm{mg}$ ) and sodium acetate ( $1.06 \mathrm{mmol}, 86.6 \mathrm{mg}$ ) in $\mathrm{EtOH}(2.2 \mathrm{~mL}$ ) at reflux afforded the product $3(84 \%, 145.7 \mathrm{mg})(75 \%$ for two steps) as a yellow solid. ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\delta 7.75(\mathrm{~d}, \mathrm{~J}=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 7.32-7.11$ (m, 6H), 6.95 (dd, J=7.4, $1.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 3.88 (s, 2H), 3.49 (s, 3H), 2.39 (s, 3H), 1.65 (s, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 176.6,155.8,137.6,137.3$, 129.7, 128.7, 128.5, 127.4, 125.5, 123.5, 121.0, 120.5, 118.3, 110.0, 81.5, 42.6, 29.8, 24.6, 16.4; HRMS m/z (ESI): calcd. for $\mathrm{C}_{22} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ (M-H)- 395.1435, found 395.1241 .

Following the general procedure for the synthesis of isothiazoles, in an oven-dried glass vial equipped with a stirring bar, 3 ( $0.1 \mathrm{mmol}, 39.6 \mathrm{mg}, 1$ equiv), NaOAc ( 0.1 mmol, $8.2 \mathrm{mg}, 1$ equiv), photocatalyst PC1, 9-mesityl-10-methyl acridinium perchlorate, ( $5 \cdot 10^{-3} \mathrm{mmol}, 2.1 \mathrm{mg}, 0.05$ equiv) and acetone ( 1.33 mL ) were added under air. Then, the vial was closed with a PTFE/rubber septum perforated several
times to enable the air enter into the reaction. Afterwards, the mixture was stirred and irradiated at 450 nm using 380 mW single LEDs in a custom-made temperature-controlled system (see Figure S1) for 16 h at $20^{\circ} \mathrm{C}$. Then, the volatiles were removed and the analytically pure product 4 was obtained by flash chromatography (latrobeads silica gel; cyclohexane/EtOAc).

## 3,8-dimethyl-8H-isothiazolo[5,4-b]indole (4)



Following the procedure, the reaction of 3 ( $0.1 \mathrm{mmol}, 39.6 \mathrm{mg}$ ), $\mathrm{NaOAc}(0.1 \mathrm{mmol}, 8.2 \mathrm{mg})$ and photocatalyst PC1 (5 mol\%, 2.1 mg ) in acetone $(1.33 \mathrm{~mL})$ irradiated at 450 nm afforded the product 4 $(71 \%, 14.4 \mathrm{mg})$ as a white solid. ${ }^{1} \mathrm{H}$ NMR ( $\left.300 \mathrm{MHz},\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 7.90(\mathrm{~d}, \mathrm{~J}=7.8 \mathrm{~Hz}$, $1 \mathrm{H}), 7.53(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.36(\mathrm{ddd}, J=8.2,7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.30-7.18(\mathrm{~m}, 1 \mathrm{H})$, $3.94(\mathrm{~s}, 3 \mathrm{H}), 2.72(\mathrm{~d}, \mathrm{~J}=1.3 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (126 MHz, $\left.\left(\mathrm{CD}_{3}\right)_{2} \mathrm{CO}\right) \delta 161.4,157.3$, 144.7, 123.9, 123.0, 120.7, 120.2, 118.9, 109.8, 32.2, 17.9; HRMS m/z (ESI): calcd. for $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{~S}(\mathrm{M}+\mathrm{H})^{+}$202.0565, found 202.0559 .

## 7. Flow Setup for the Synthesis of Isothiazoles 2

General procedure for the optimization of the reaction conditions for the flow setup:

In a coil $(V=18 \mathrm{~mL})$ made of perfluoroalkoxy (PFA) tubing (i.d. $=1.6 \mathrm{~mm}$ ) irradiated with a blue LED spot (40 W), 2-(((1-(2-(benzylthio)phenyl)ethylidene) amino)oxy)-2-methylpropanoic acid (1d, $0.15 \mathrm{mmol}, 36.3 \mathrm{mg}, 1$ equiv), NaOAc ( 0.15 mmol, $12.3 \mathrm{mg}, 1$ equiv), photocatalyst PC1, 9-mesityl-10-methyl acridinium perchlorate and acetone ( 2 mL ) were injected at $20^{\circ} \mathrm{C}$. Then, the volatiles were removed and yield was determined by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ analysis using nitromethane ( $8 \mu \mathrm{~L}$, 0.15 mmol ) as the internal standard. $\mathrm{O}_{2}$ gas was employed in a segmented flow fashion and a BPR (1.6 bar) was added at the end of the coil to increase the pressure of the system.


Table S3. Optimization of reaction conditions for the flow setup.

| entry | conditions $^{[\mathrm{ab]}}$ | $\mathrm{PC} 1(\mathrm{~mol} \%)$ | $\boldsymbol{t}_{\boldsymbol{R}}(\mathbf{m i n})$ | conv. (\%) ${ }^{[\mathrm{bl]}}$ | yield (\%) ${ }^{[\mathrm{b}]}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | No oxygen | 10 | 80 | 50 | 13 |
| 2 | Acetone sat. <br> with $\mathrm{O}_{2}$ <br> $\mathrm{BPR}(1.6$ bar $)$ | 10 | 80 | 68 | 41 |
| 3 | $\mathrm{O}_{2}$ | 10 | 80 | 84 | 47 |
| 4 | $\mathrm{O}_{2}$ | 10 | 180 | 88 | 48 |
| 5 | $\mathrm{O}_{2}$ | 10 | 80 | 100 | 26 |
| $\mathbf{6}$ | $\mathrm{O}_{2}$ | $\mathbf{5 P R}(1.6$ bar) | $\mathbf{6 0}$ | $\mathbf{8 2}$ | $\mathbf{4 7}$ |

[a] Segmented flow ( 2 mm slugs) was made by passing oxygen gas and reaction solution through a T-union ( 1.25 mm thru-hole; $17.5 \mu \mathrm{l}$ swept volume). ${ }^{[b]}$ The yield was determined from the crude ${ }^{1} \mathrm{H}-\mathrm{NMR}$. BPR = black pressure regulator

General procedure for the synthesis of isothiazoles $\mathbf{2}$ in the flow setup:
The corresponding oxime derivative 1 ( $0.33 \mathrm{mmol}, 1$ equiv), photocatalyst PC1, 9-mesityl-10-methyl acridinium perchlorate ( $0.017 \mathrm{mmol}, 6.8 \mathrm{mg}, 0.05$ equiv) and acetone $(4.4 \mathrm{~mL})$ were mixed at $20^{\circ} \mathrm{C}$ and passed through a packed bed reactor filled with NaOAc (Figure S2a). Afterwards, reaction solution and $\mathrm{O}_{2}$ gas, used in a segmented flow fashion, were mixed through a T-union (Figure S2b), controlling the gas slugs ( 2 mm ) with a micrometric valve. Compounds $\mathbf{1 b}, 1 \mathrm{c}$ and 1 g were injected consecutively (separated by 4 mL of acetone) in the coil ( $\mathrm{V}=18 \mathrm{~mL}$ ) made of perfluoroalkoxy (PFA) tubing (i.d. $=1.6 \mathrm{~mm}$ ) and irradiated with a blue LED spot ( 40 W) (Figure S3). After 2 h , isothiazoles 2 were collected in different vials. Then, the volatiles were removed and yield was determined by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ analysis using
nitromethane ( $17.7 \mu \mathrm{~L}, 0.33 \mathrm{mmol}$ ) as the internal standard: 2b ( $55 \%$; 0.18 mmol ), 2c ( $58 \% ; 0.19 \mathrm{mmol}$ ) and $\mathbf{2 g}$ ( $44 \% ; 0.14 \mathrm{mmol}$ ). In all the transformations the throughput of the reaction is $\approx 140$ times higher than in conventional conditions and around $15 \%$ of starting material is recovered in all the cases.
a)

b)


Figure S2. a) Packed bed reactor filled with NaOAc and b) T-union with a micrometric valve employed in flow setup.


Figure S3. Coil of PFA with liquid and gas slugs employed in flow setup.

## 8. Mechanistic Studies

### 8.1. Cyclic Voltammetry of $\alpha$-Imino-oxy Acids 1

Electrochemical studies of substrates $\mathbf{1 a} \mathbf{- d}, \mathbf{1 a} \mathbf{a n d}^{\mathbf{1 a}} \mathbf{1 0}$ were performed employing

1 nM solution of the corresponding a-imino-oxy acid $\mathbf{1}$ freshly prepared in HPLC grade acetone along with 1 nM sodium acetate and 0.1 M supporting electrolyte (tetrabutylammonium hexafluorophosphate) solutions. Nitrogen was passed through each sample before measurements to avoid the influence of oxygen reduction.


Figure S4. Cyclic voltammetry of a-imino-oxy acids 1a, 1a'and 1a".

Table S4. Oxidation potentials of $\alpha$-imino-oxy acids 1a, 1a'and 1a'.

| $\alpha$-Imino-oxy acids | $\mathrm{E}_{1 / 2}{ }^{\text {ox }}(\mathrm{V})$ vs SCE |
| :---: | :---: |
|  <br> 1a | +1.58 |
|  | +1.58 |
|  | +1.58 |



Figure S5. Cyclic voltammetry of $\alpha$-imino-oxy acids 1a-d.

Table S5. Oxidation potentials of $\alpha$-imino-oxy acids 1a-d.
(V) vs SCE



Figure S6. Electrochemical scale of $\alpha$-imino-oxy acids 1a-d.

### 8.2. Fluorescence Quenching Studies

Emission intensities and time resolved emission spectra were recorded using an Edinburg Instruments FS5 Spectrofluorometer, and a 450 nm EPL laser.

For the steady-state and time resolved fluorescence quenching studies, increasing concentrations of quencher were added to a solution $3,7 \mathrm{mM}$ of PC1 in acetone ( $\lambda_{\mathrm{exc}}=$ 450 nm ).
a)




$K q=4.2 \times 10^{9} \mathrm{M}^{-1} \mathrm{~s}^{-1}$
b)



$K q=6.95 \times 10^{9} \mathrm{M}^{-1} \mathrm{~s}^{-1}$
c)


Figure S7. Fluorescence quenching studies. a) Quenching studies of PC1 with 1d: (1) Steady state luminescence quenching spectrum and (2) Stern-Volmer plot of the steady state; (3) Time resolved luminescence quenching and (4) Stern-Volmer plot to obtain $K q$; b) Quenching studies of PC1 with 1d + NaOAc: (1) Steady state luminescence quenching spectrum and (2) Stern-Volmer plot of the steady state; (3) Time resolved luminescence quenching and (4) Stern-Volmer plot to obtain $K q$; c) Quenching studies of PC1 with NaOAc: (1) Steady state luminescence quenching spectrum and (2) Stern-Volmer plot of the steady state; (3) Time resolved luminescence quenching and (4) Stern-Volmer plot to obtain Kq.

### 8.3. Radical Trapping Experiments

Table S6. Radical Trapping Experiments.


| entry | Radical scavenger | yield (\%) $^{[a]}$ |
| :---: | :---: | :---: |
| 1 | None | 85 |
| 2 | TEMPO | 33 |
| 3 | BHT | 39 |

${ }^{[a]}$ The yield was determined from the crude ${ }^{1} \mathrm{H}-\mathrm{NMR}$.

Following the general procedure for the synthesis of isothiazoles 2, in an oven-dried glass vial equipped with a stirring bar, 2-methyl-2-(((1-(2-(methylthio) phenyl)ethylidene)amino)oxy)propanoic acid (1a, $0.05 \mathrm{mmol}, 13.4 \mathrm{mg}$, 1 equiv), the $\mathrm{NaOAc}(0.05 \mathrm{mmol}, 4.1 \mathrm{mg}, 1$ equiv), photocatalyst PC1, 9-mesityl-10-methyl acridinium perchlorate, $\left(2.5 \cdot 10^{-3} \mathrm{mmol}, 1 \mathrm{mg}, 0.05\right.$ equiv), the corresponding radical scavenger ( $0.25 \mathrm{mmol}, 5$ equiv) and acetone ( 0.67 mL ) were added under air. Then, the vial was closed with a PTFE/rubber septum perforated several times to enable the air enter into the reaction. Afterwards, the mixture was stirred and irradiated at 450 nm using 380 mW single LEDs in a custom-made temperature-controlled system (see Figure S1) for 16 h at $20^{\circ} \mathrm{C}$. Then, the volatiles were removed and yield was determined by ${ }^{1} \mathrm{H}$-NMR analysis using 1,3,5-trimethoxybenzene ( $8.4 \mathrm{mg}, 0.05 \mathrm{mmol}$ ) as internal standard.

## 9. References

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## 10. NMR Spectra

### 10.1. NMR Spectra of Sulfides $\mathrm{SI}-1$


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10.2. NMR Spectra of Selenides $\mathrm{SI}-10$



10.3. NMR Spectra of $\alpha$-Imino-oxy Acids 1









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$\begin{array}{lllllllllllllllllllllllllllllll}10 & 0 & -10 & -20 & -30 & -40 & -50 & -60 & -70 & -80 & -90 & -100 & -110 & -120 & -130 & -140 & -150 & -160 & -170 & -180 & -190 & -200 & -210\end{array}$






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| 200 | 190 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 | $\begin{gathered} 100 \\ \mathrm{f} 1(\mathrm{ppm}) \end{gathered}$ | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 | 0 |




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$\begin{array}{llllllllllllllllllllll}210 & 200 & 190 & 180 & 170 & 160 & 150 & 140 & 130 & 120 & 110 & 100 & 90 & 80 & 70 & 60 & 50 & 40 & 30 & 20 & 10 & 0 \\ f 1(\mathrm{ppm})\end{array}$






10.4. NMR Spectra of Isothiazoles 2



$\underset{\sim}{\infty} \underset{\sim}{\infty}$






$$
\begin{aligned}
& \hat{\vec{i}} \\
& \text { N }
\end{aligned}
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### 10.5. NMR Spectra of Brassilexin Derivative 4











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