# Benzoic Acid Resin (BAR): A Heterogeneous Redox Organocatalyst for Continuous Flow Synthesis of Benzoquinones from $\beta$-O-4 Lignin Models 

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## Supporting Information

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## General

Chemicals: Starting materials were purchased from Sigma-Aldrich ${ }^{\circledR}$ and used without further purification. Solvents were purified by standard procedures. ${ }^{1}$

Analytical Methods: The reaction outcomes were analyzed by Merck ${ }^{\circledR}$ Thin Layer Chromatography Silica Gel $60 \mathrm{~F}_{254}$, visualized with UV light and revealed with vanillin solution in $\mathrm{H}_{2} \mathrm{SO}_{4(\mathrm{aq})}$ followed by heating. Conversions were analyzed by GC-FID, GCMS and ${ }^{1} \mathrm{H}$ NMR analysis. The GC-17A (Shimadzu ${ }^{\circledR}$ ) chromatograph with flame ionization detector (FID), equipped with Agilent ${ }^{\circledR}$ HP-5 capillary column was used. The GC-FID conditions were: injector $=260^{\circ} \mathrm{C}$; detector $=280^{\circ} \mathrm{C}$; pressure $=100 \mathrm{kPa}$. The column temperature range was $40{ }^{\circ} \mathrm{C}$ to $115{ }^{\circ} \mathrm{C}$ at $10{ }^{\circ} \mathrm{C} / \mathrm{min}$. GC-MSQP2010SE (Shimadzu ${ }^{\circledR}$ ) with low-resolution electron impact (EI, 70 eV ) chromatograph, equipped with a Restek ${ }^{\circledR}$ Rtx-5MS capillary column, was also employed. The GC-MS conditions were: injector $=260^{\circ} \mathrm{C}$; detector $=280^{\circ} \mathrm{C}$; pressure $=100 \mathrm{kPa}$. The column temperature range was $40^{\circ} \mathrm{C}$ to $115^{\circ} \mathrm{C}$ at $10^{\circ} \mathrm{C} / \mathrm{min}$ and hold time $\left.=15 \mathrm{~min}\right)$. Varian ${ }^{\circledR}$ INOVA (300 MHz ) and Bruker ${ }^{\circledR}$ Avance III ( 300 MHz ) spectrometers were used in the NMR characterizations. The chemical shifts of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were referenced either to tetramethylsilane as internal standard ( ${ }^{1} \mathrm{H}$ NMR: $\mathrm{CDCl}_{3}$ at $\delta 7.26 \mathrm{ppm}$ and acetone- $\mathrm{d}_{6}$ at $\delta 2.05 \mathrm{ppm})$ or deuterated solvent $\left({ }^{13} \mathrm{C}\right.$ NMR $=\mathrm{CDCl}_{3}$ at $\delta 77.16 \mathrm{ppm}$ and acetone- $\mathrm{d}_{6}$ at $\delta 29.92 \mathrm{ppm}$ and $\delta 206.25 \mathrm{ppm}$ ). High-resolution mass spectra were obtained on a Bruker ${ }^{\circledR}$ Daltonics MicroToF spectrometer using electrospray ionization-time of flight (ESI-TOF) techniques. The melting points were determined with a BÜCHI ${ }^{\circledR}$ B-545 apparatus. Scanning Electron Microscopy (SEM) images were obtained using a JEOL ${ }^{\circledR}$ JSM 7401F field emission gun electron microscope. Samples were prepared by the conductive double-sided carbon tape technique. All experiments described in this study were carried out at least in sextuplicate.

## Synthesis and Characterization of the Benzoic Acid Resin (BAR)

Scheme SI


Considering solvent recycle, the $E$ Factor value was calculated as follows: Obtained mass of $\mathbf{B A R}=2.49 \mathrm{~g}$; Mass of reagents $=9.25 \mathrm{~g}(1.97 \mathrm{~g}$ of TMEDA +1.54 g of $n-\mathrm{BuLi}+$ 3.00 g of $\mathrm{CO}_{2(\mathrm{~s})}+2.74 \mathrm{~g}$ of HCl$)$; Amount of poly(styrene-co-divinylbenzene) $\mathbf{1}=2.00$ g.
$E$ Factor $=(2.00+9.25-2.49) / 2.49=3.5$
Resin 1 and BAR were also characterized by SEM (Figure S1).


Figure $\mathrm{S1} 10 \mathrm{SEM}$ images of $\mathbf{1}(\mathrm{A})$ and $\mathbf{B A R}^{\times 350}(\mathrm{~B})$ at $\times 100$ and $\times 350$ magnifications.

Assembly for Continuous Flow Oxidations of $\boldsymbol{\beta}-\mathrm{O}-4$ Lignin Models Mediated by BAR


Figure S2: Assembly for the continuous flow system mediated by BAR.

## Synthesis of the $\boldsymbol{\beta}$-O-4 Lignin Models

The synthesis and characterization of the ether-substituted $\alpha$-phenoxyacetophenones (iiiia e) from ether-substituted acetophenones ( $\mathbf{i}_{\text {a-c }}$ ) was published by our group (Scheme S2). ${ }^{2}$

Scheme S2 ${ }^{2}$


1-(4-(benzyloxy)phenyl)ethanone ( $\mathbf{i}_{\mathrm{c}}, \mathbf{1 0}$ ): white powder.

${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ): $\delta=7.94(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, 7.44-7.34 (m, 5H), 7.00 (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.14 (s, 2H), 2.55 ( $\mathrm{s}, 3 \mathrm{H}$ ) ppm.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{7 5} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=196.9,162.8,136.2,130.7$
(2C), 128.8 (2C), 128.4, 127.6 (2C), 114.7 (2C), 70.3, 26.5
ppm.
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right): ~ 90-91$ (lit. ${ }^{4}$ 89-90).
$\boldsymbol{R}_{f}: 0.72$ (3:7 EtOAc/hexanes).
\#CAS: 54696-05-8.
1-(4-(benzyloxy)phenyl)-2-phenoxyethanone (iiia, 11 ): beige solid.

${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right): \delta=7.99(\mathrm{~d}, J=8.9 \mathrm{~Hz}$, 2H), 7.42-7.25 (m, 7H), 7.04-6.92 (m, 5H), 5.19 (s, 2 H ), 5.13 ( $\mathrm{s}, 2 \mathrm{H}$ ) ppm.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{7 5} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=193.2,163.3,158.2$,
136.1, 130.7 (2C), 129.7 (2C), 128.8 (2C), 128.4, $127.9,127.6$ (2C), 121.7, 115.0 (2C), 114.9 (2C), $70.8,70.3 \mathrm{ppm}$.
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right): 111-112$.
$\boldsymbol{R}_{\boldsymbol{f}}: 0.66$ (3:7 $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ hexanes).
\#CAS: 2173138-44-6.

For the synthesis of the ether-substituted $\alpha, \alpha$-(hydroxymethyl)phenoxyacetophenones 12-16, a slightly modified procedure for the aldol addition between formaldehyde (HCHO) and iiii ${ }_{\text {a-e }}$ (Scheme S3), previously described for Magallanes et al. ${ }^{3}$, was employed.

Scheme S3 ${ }^{3}$

$\mathrm{iii}_{\mathrm{a}-\mathrm{e}}$
$\mathrm{iii}_{\mathrm{a}}$ (11): $\mathrm{R}=p-\mathrm{OBn} ; \mathrm{R}^{1}=\mathrm{H}$
$\mathrm{iii}_{\mathrm{b}}: \mathrm{R}=p-\mathrm{OMe} ; \mathrm{R}^{1}=o-\mathrm{OMe}$
$\mathrm{iii}_{\mathrm{c}}: \mathrm{R}=p$-OMe; $\mathrm{R}^{1}=p-\mathrm{OMe}$ $\mathrm{iii}_{\mathrm{d}}: \mathrm{R}=m-\mathrm{OMe} ; \mathrm{R}^{1}=o-\mathrm{OMe}$ $\mathrm{iii}_{\mathrm{e}}: \mathrm{R}=m-\mathrm{OMe} ; \mathrm{R}^{1}=p-\mathrm{OMe}$


12-16
$12: \mathrm{R}=\mathrm{p}-\mathrm{OBn} ; \mathrm{R}^{1}=\mathrm{H}$
$13: \mathrm{R}=p-\mathrm{OMe} ; \mathrm{R}^{1}=o-\mathrm{OMe}$
$14: \mathrm{R}=p-\mathrm{OMe} ; \mathrm{R}^{1}=p-\mathrm{OMe}$
$15: \mathrm{R}=m-\mathrm{OMe} ; \mathrm{R}^{1}=o-\mathrm{OMe}$
$16: \mathrm{R}=m-\mathrm{OMe} ; \mathrm{R}^{1}=p-\mathrm{OMe}$

Substrates iii $_{\text {a-e }}(7.00 \mathrm{mmol})$ were solubilized in 50 mL of a mixture of acetone and ethanol ( $1: 1$ ). Potassium carbonate $(1.06 \mathrm{~g}, 7.70 \mathrm{mmol})$ was added and the reaction mixture was left stirring for 5 minutes. Then, formalin (aqueous, $37 \% \mathrm{wt}, 830 \mu \mathrm{~L}, 10.5$ mmol ) was added, stirred for 2 h at room temperature. The reaction mixture was analyzed by TLC. Due to the presence of starting material, formalin ( $37 \% \mathrm{wt}, 275 \mu \mathrm{~L}, 3.50 \mathrm{mmol}$ ) was added once again. After 30 min , the solvent was evaporated under reduced pressure, and water $(50 \mathrm{~mL})$ and $\mathrm{CHCl}_{3}(50 \mathrm{~mL})$ were added to the flask. The aqueous layer was wsahed with $\mathrm{CHCl}_{3}(3 \times 25 \mathrm{~mL})$. The combined organic layers were washed with water, brine, and then dried over anhydrous $\mathrm{MgSO}_{4}$. The organic layer was concentrated and purified by column chromatography (EtOAc/hexanes).

1-(4-(benzyloxy)phenyl)-3-hydroxy-2-phenoxypropan-1-one (12): yield $=91 \%(2.20$

g ), pearl-white solid.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ): $\delta=8.05(\mathrm{~d}, J=5.4 \mathrm{~Hz}$, 2 H ), $7.41-7.35(\mathrm{~m}, 5 \mathrm{H}), 7.24(\mathrm{dd}, J=4.5 \mathrm{~Hz}, J=5.4$ $\mathrm{Hz}, 2 \mathrm{H}), 7.01(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.96(\mathrm{t}, J=4.5 \mathrm{~Hz}$, $1 \mathrm{H}), 6.89(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.49(\mathrm{dd}, J=2.4 \mathrm{~Hz}, J=$ $3.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.13(\mathrm{~s}, 1 \mathrm{H}), 4.17-4.07(\mathrm{~m}, 2 \mathrm{H}), 2.38(\mathrm{dd}, J=3.6 \mathrm{~Hz}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm}$. ${ }^{13} \mathbf{C}$ NMR ( $75 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ): $\delta=195.1,163.5,157.4,136.1,131.4$ (2C), 129.8 (2C), 128.9 (2C), 128.5, 128.0, 127.6 (2C), 122.0, 115.4 (2C), 115.1 (2C), $81.2,70.4,63.5 \mathrm{ppm}$.
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right): 107-109$.
$\boldsymbol{R}_{\boldsymbol{f}}: 0.41$ (5:5 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{O}_{4}: 349.1440$, found: 349.1433 .
\#CAS: 2183492-22-8.
3-hydroxy-2-(2-methoxyphenoxy)-1-(4-methoxyphenyl)propan-1-one (13): yield = $83 \%$ ( 1.76 g ), pale-yellow syrup.

${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, acetone- $\mathbf{d}_{6}$ ): $\delta=8.15(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, 2H), 7.08-6.76 (m, 6H), 5.51 (t, $J=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.07-$ $4.045 .13(\mathrm{~m}, 2 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 2.82-2.79(\mathrm{~m}$, 1H) ppm.
${ }^{13}$ C NMR ( 75 MHz , acetone- $\mathbf{d}_{6}$ ): $\delta=196.1,164.8,151.3,148.6,132.1$ (2C), 129.7, $123.3,121.7,117.5,114.7$ (2C), 113.9, 84.3, 64.1, 56.3, 56.0 ppm .
$\boldsymbol{R}_{\boldsymbol{f}}: 0.39$ (4:6 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{5}: 303.1233$, found: 303.1231.
\#CAS: 92409-23-9.
3-hydroxy-2-(4-methoxyphenoxy)-1-(4-methoxyphenyl)propan-1-one (14): yield =
 $87 \%$ ( 1.84 g ), colorless crystals.
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, acetone-d $\mathbf{d}_{\mathbf{6}}$ ): $\delta=8.13$ (d, $J=8.8$ $\mathrm{Hz}, 2 \mathrm{H}), 7.06(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.96(\mathrm{~d}, J=5.4 \mathrm{~Hz}$, $2 \mathrm{H}), 6.89$ (d, $J=5.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.48 (d, $J=5.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), $4.23(\mathrm{dd}, J=5.5 \mathrm{~Hz}, J=5.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.80$
( $\mathrm{s}, 3 \mathrm{H}$ ), 2.85 (bs, 1H) ppm.
${ }^{13} \mathbf{C}$ NMR ( 75 MHz , acetone- $\mathbf{d}_{6}$ ): $\delta=196.3,164.9,155.3,153.0,132.0$ (2C), 129.5, 117.3 (2C), 115.4 (2C), 114.8 (2C), 83.6, 64.0, 56.1, 55.8 ppm .
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right): ~ 94-96$ (lit. ${ }^{5}$ 93-96).
$\boldsymbol{R}_{\boldsymbol{f}}: 0.34$ (4:6 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{5}$ : 303.1233, found: 303.1239.
\#CAS: 2093366-73-3.
3-hydroxy-2-(2-methoxyphenoxy)-1-(3-methoxyphenyl)propan-1-one (15): yield = $90 \%(1.90 \mathrm{~g})$, non-crystalline beige solid.

${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, acetone- $\mathbf{d}_{\mathbf{6}}$ ): $\delta=7.72$ (ddd, $J=1.2 \mathrm{~Hz}$, $J=0.9 \mathrm{~Hz}, J=1.5 \mathrm{~Hz}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.63(\mathrm{dd}, J=1.5 \mathrm{~Hz}$, $J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.43$ (t, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.19$ (ddd, $J=0.9$ $\mathrm{Hz}, J=0.9 \mathrm{~Hz}, J=2.7 \mathrm{~Hz}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.96-6.89(\mathrm{~m}$, $3 \mathrm{H}), 6.82-6.79(\mathrm{~m}, 1 \mathrm{H}), 5.59-5.56(\mathrm{~m}, 1 \mathrm{H}), 4.09-4.05(\mathrm{~m}, 2 \mathrm{H}), 3.85(\mathrm{~s}, 3 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H})$, 2.81 (bs, 1H) ppm.
${ }^{13}$ C NMR ( 75 MHz , acetone- $\mathbf{d}_{\mathbf{6}}$ ): $\delta=197.6,160.8,151.2,148.5,138.1,130.6,123.4$, 122.0, 121.6, 120.3, 117.6, 114.2, 113.8, 84.2, 64.0, 56.2, 55.8 ppm .
$\boldsymbol{R}_{f}: 0.40$ ( $4: 6 \mathrm{AcOEt} /$ hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{5}: 303.1233$, found: 303.1245 .
\#CAS: 20730-76-1.
3-hydroxy-2-(4-methoxyphenoxy)-1-(3-methoxyphenyl)propan-1-one (16): yield =

$86 \%(1.82 \mathrm{~g})$, white powder.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}$, acetone- $\mathbf{d}_{6}$ ): $\delta=7.65$ (ddd, $J=1.1$ $\mathrm{Hz}, J=0.9 \mathrm{~Hz}, J=1.5 \mathrm{~Hz}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{dd}, J$
$=1.5 \mathrm{~Hz}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.23(\mathrm{ddd}, J=0.9 \mathrm{~Hz}, J=1.1 \mathrm{~Hz}, J=$ $2.7 \mathrm{~Hz}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.94-6.83(\mathrm{~m}, 4 \mathrm{H}), 5.64(\mathrm{t}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.15-4.10(\mathrm{~m}, 2 \mathrm{H})$, $3.87(\mathrm{~s}, 3 \mathrm{H}), 3.73(\mathrm{~s}, 3 \mathrm{H}), 2.85(\mathrm{bs}, 1 \mathrm{H}) \mathrm{ppm}$.
${ }^{13} \mathbf{C}$ NMR ( 75 MHz , acetone- $\mathbf{d}_{6}$ ): $\delta=197.4,160.9,155.3,153.0,137.9,130.4,123.6$, $120.0,117.8,115.1$ (2C), 114.6 (2C), 84.5, 64.1, $55.9,55.7 \mathrm{ppm}$.
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right): ~ 90-93$.
$\boldsymbol{R}_{f}: 0.42$ ( $4: 6 \mathrm{AcOEt} /$ hexanes ).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{5}: 303.1233$, found: 303.1247.

## NMR data for the Products (Table 3, Scheme S4)

## Scheme S4



4-(benzyloxy)phenyl acetate (17): yield for Entry 1, Table $3=16 \%(34.2 \mathrm{mg}, 0.14$ mmol ), colorless crystals.

${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.44-7.34(\mathrm{~m}, 4 \mathrm{H}), 7.31$
(ddd, $J=1.5 \mathrm{~Hz}, J=1.5 \mathrm{~Hz}, J=2.1 \mathrm{~Hz}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.02-6.94 (m, 4H), $5.04(\mathrm{~s}, 2 \mathrm{H}), 2.28(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$.
${ }^{13} \mathbf{C}$ NMR ( $75 \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=170.0,156.7,144.6,137.0$, 128.8 (2C), 128.2, 127.6 (2C), 122.5 (2C), 115.6 (2C), $70.6,21.2 \mathrm{ppm}$.
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right):$ 111-112 (lit. ${ }^{6} 100-111$ ).
$\boldsymbol{R}_{f}: 0.67$ (5:5 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{O}_{3}: 243.1021$, found: 243.1015 .
\#CAS: 6311-66-6.
1,4-benzoquinone (6): yield for Entry 6, Table $3=89 \%$ ( $56.2 \mathrm{mg}, 0.52 \mathrm{mmol}$ ), neon-
 yellow crystals.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=6.79(\mathrm{~s}, 4 \mathrm{H}) \mathrm{ppm}$.
${ }^{13} \mathbf{C}$ NMR ( $75 \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=187.1$ (2C), 136.4 (4C) ppm.
mp ( ${ }^{\circ} \mathbf{C}$ ):112-115 (lit. ${ }^{7}$ 115-116).
$\boldsymbol{R}_{\boldsymbol{f}}: 0.91$ (6:4 AcOEt/hexanes).
MS (EI+): $m / z$ (relative intensity) 108 ( $\mathbf{M}^{+}, 89$ ), 54 (100).
\#CAS: 106-51-4.
4-(benzyloxy)phenyl 2-phenoxyacetate (18): yield for Entry 3, Table $3=22 \%$ ( 64.8 mg ,
 0.19 mmol ), pale-yellow solid.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.43-7.30(\mathrm{~m}, 7 \mathrm{H})$, 7.04-6.94 (m, 7H), $5.05(\mathrm{~s}, 2 \mathrm{H}), 5.13(\mathrm{~s}, 2 \mathrm{H}), 4.86(\mathrm{~s}$, 2H) ppm.
${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ): $\delta=168.0,157.9$, $156.8,143.9,136.8,129.8$ (2C), 128.8 (2C), 128.2, 127.6 (2C), 122.2 (2C), 122.1, 115.7 (2C), 114.9 (2C), $70.6,65.6 \mathrm{ppm}$.
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right): 124-127$.
$\boldsymbol{R}_{f}: 0.59$ (5:5 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calc. for $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{O}_{4} \mathrm{Na}: 357.1103$, found: 357.1099. \#CAS: 686311-24-0.

4-(benzyloxy)phenyl 3-hydroxy-2-phenoxypropanoate (19): yield for Entry 5, Table 3 $=8 \%(16.7 \mathrm{mg}, 0.04 \mathrm{mmol})$, beige solid.

${ }^{1} \mathbf{H}$ NMR $\left(\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right): \delta=7.42-7.22(\mathrm{~m}$, $7 \mathrm{H}), 7.0-6.88(\mathrm{~m}, 7 \mathrm{H}), 5.10(\mathrm{~s}, 2 \mathrm{H}), 4.95(\mathrm{dd}, J=5.7$ $\mathrm{Hz}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.22-4.16(\mathrm{~m}, 2 \mathrm{H}), 2.33(\mathrm{bs}, 1 \mathrm{H})$ ppm.
${ }^{13}$ C NMR ( $75 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ): $\delta=168.9$, 158.1, $156.9,144.5,135.1$ (2C), 130.0 (2C), 128.3, 127.7 (2C), 122.5, 122.4 (2C), 115.8 (2C), 115.0 (2C), 89.5, 70.5, 63.4 ppm .
mp ( ${ }^{\circ} \mathbf{C}$ ): 137-139.
$\boldsymbol{R}_{\boldsymbol{f}}: 0.62$ (6:4 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{Na}]^{+}$calc. for $\mathrm{C}_{22} \mathrm{H}_{20} \mathrm{O}_{5} \mathrm{Na}: 387.1209$, found: 387.1218. 4-benzyloxyphenol (24): Table 3, conjoined traces ( 8.5 mg ),
 white powder.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.42-7.29$ (m, 5H), 6.85 (d, , $J=5.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.75(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 2 \mathrm{H}), 5.82(\mathrm{bs}, 1 \mathrm{H}), 4.99(\mathrm{~s}$, 2H) ppm.
m.p. $\left({ }^{\circ} \mathbf{C}\right):$ 120-122 (lit. ${ }^{6}$ 121-122).
$\boldsymbol{R}_{\boldsymbol{f}}: 0.12$ (6:4 AcOEt/hexanes).
\#CAS: 103-16-2.
4-methoxyphenyl 3-hydroxy-2-(2-methoxyphenoxy)propanoate (20): yield for Entry


6 , Table $3=9 \%(16.5 \mathrm{mg}, 0.05 \mathrm{mmol})$, opaque white crystals.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.26-6.76(\mathrm{~m}, 8 \mathrm{H}), 4.88$ (dd, $J=3.9 \mathrm{~Hz}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.19$ (ddd, $J=6.3 \mathrm{~Hz}, J$ $=3.9 \mathrm{~Hz}, J=11.7 \mathrm{~Hz}, J=26.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.86(\mathrm{~s}, 3 \mathrm{H}), 3.75$
( $\mathrm{s}, 3 \mathrm{H}$ ), 3.30 (bs, 1H) ppm.
${ }^{13}$ C NMR ( $75 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ): $\delta=168.6,150.1,147.4,143.8,143.3,123.1,122.0$ (2C), $120.9,115.6,114.9$ (2C), 112.6, 81.0, $67.3,56.0,55.6 \mathrm{ppm}$.
mp ( ${ }^{\circ} \mathbf{C}$ ): 146-149.
$\boldsymbol{R}_{\boldsymbol{f}}: 0.70$ ( $6: 4 \mathrm{AcOEt} /$ hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{6}$ : 319.1182, found: 319.1190 .
4-methoxyphenol (5): Table 3, conjoined traces ( 6.5 mg ), off white
 powder.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=6.81-6.75(\mathrm{~m}, 4 \mathrm{H}), 4.60(\mathrm{bs}, 1 \mathrm{H})$, $3.76(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$.
mp ( ${ }^{\circ} \mathbf{C}$ ): 56-57 (lit. ${ }^{8}$ 54-56).
$\boldsymbol{R}_{\boldsymbol{f}}: 0.16$ (6:4 AcOEt/hexanes).
\#CAS: 150-76-5.
4-methoxyphenyl 3-hydroxy-2-(4-methoxyphenoxy)propanoate (21): yield for Entry


(bs, 1 H ) ppm.
${ }^{13} \mathbf{C} \mathbf{N M R}\left(75 \mathbf{M H z}, \mathbf{C D C l}_{3}\right): \delta=168.9,157.7,155.2,151.6,143.8,122.2$ (2C), 117.2 (2C), 114.9 (2C), 114.6 (2C), 79.1, 63.6, 55.7, 55.6 ppm .
$\mathbf{m p}\left({ }^{\circ} \mathbf{C}\right): 150-152$.
$\boldsymbol{R}_{f}: 0.72$ (6:4 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{6}$ : 319.1182, found: 319.1196.
3-methoxyphenyl 3-hydroxy-2-(2-methoxyphenoxy)propanoate (22): yield for Entry
 7, Table $3=8 \%(14.6 \mathrm{mg}, 0.05 \mathrm{mmol})$, white powder. ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.02(\mathrm{~d}, J=9.0 \mathrm{~Hz}$, 2H), $6.95-6.82(\mathrm{~m}, 6 \mathrm{H}), 4.89$ (dd, $J=4.5 \mathrm{~Hz}, J=5.1$ $\mathrm{Hz}, 1 \mathrm{H}), 4.22(\mathrm{ddd}, J=5.1 \mathrm{~Hz}, J=4.5 \mathrm{~Hz}, J=12.0$ $\mathrm{Hz}, J=14.1 \mathrm{~Hz}, 2 \mathrm{H}), 5.13(\mathrm{~s}, 2 \mathrm{H}), 3.76(\mathrm{~s}, 6 \mathrm{H}), 2.70$


8 , Table $3=11 \%$ ( $20.0 \mathrm{mg}, 0.06 \mathrm{mmol}$ ), yellow oil.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.21(\mathrm{t}, J=5.6 \mathrm{~Hz}, 1 \mathrm{H})$, $7.01-6.77$ (m, 7H), 4.89 (dd, $J=3.9 \mathrm{~Hz}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.08 (ddd, $J=6.3 \mathrm{~Hz}, J=3.9 \mathrm{~Hz}, J=11.7 \mathrm{~Hz}, J=27.3$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 3.87 (s, 3H), 3.84 (s, 3H), 3.16 (bs, 1H) ppm.
${ }^{13} \mathbf{C}$ NMR ( $75 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ): $\delta=169.1,160.4,151.9,147.2,143.6,130.0,123.2,120.6$, $116.0,114.2,112.7,111.3,107.5,83.0,68.1,56.0,55.6 \mathrm{ppm}$.
$\boldsymbol{R}_{\boldsymbol{f}}: 0.75$ (6:4 AcOEt/hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{6}: 319.1182$, found: 319.1177.
3-methoxyphenol (7): Table 3, conjoined traces ( 6.2 mg ), deep-brown liquid.

${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.14(\mathrm{t}, J=5.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.54-6.44$ $(\mathrm{m}, 3 \mathrm{H}), 5.56(\mathrm{bs}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. $\boldsymbol{R}_{f}: 0.15$ (6:4 AcOEt/hexanes).
\#CAS: 150-19-6.
2-methoxycyclohexa-2,5-diene-1,4-dione (8): yield for Entry 8, Table 3 = 84\% (56.2
 $\mathrm{mg}, 0.52 \mathrm{mmol}$ ), beige powder.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=6.71(\mathrm{~m}, 2 \mathrm{H}), 5.95(\mathrm{~m}, 1 \mathrm{H}), 3.74(\mathrm{~s}$, 3H) ppm.
${ }^{13} \mathbf{C}$ NMR ( $75 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ): $\delta=187.4,181.5,158.6,137.0,134.3$,
107.5, 56.2 ppm .
mp ( ${ }^{\circ} \mathbf{C}$ ): 144-145 (lit. ${ }^{9}$ 144-146).
$\boldsymbol{R}_{\boldsymbol{f}}: 0.88$ (6:4 AcOEt/hexanes).
MS (EI ${ }^{+}$): $m / z$ (relative intensity) $138\left(\mathrm{M}^{+}, 53\right), 108$ (68), 69 (100).
\#CAS: 2880-58-2.
3-methoxyphenyl 3-hydroxy-2-(4-methoxyphenoxy)propanoate (23): yield for Entry 9 , Table $3=10 \%(18.4 \mathrm{mg}, 0.06 \mathrm{mmol})$, pale-yellow non-crystalline solid.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{3 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ): $\delta=7.20(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.00-6.83(\mathrm{~m}, 7 \mathrm{H}), 4.90(\mathrm{dd}$,
 $J=4.5 \mathrm{~Hz}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{ddd}, J=5.1 \mathrm{~Hz}, J$ $=4.5 \mathrm{~Hz}, J=11.7 \mathrm{~Hz}, J=15.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H})$, 3.75 (s, 3H), 2.93 (bs, 1H) ppm.
${ }^{13} \mathbf{C}$ NMR $\left(75 \mathrm{MHz}, \mathbf{C D C l}_{3}\right): \delta=169.0,160.2,152.1$, 151.2, 145.9, 129.9, 115.2 (2C), 114.7 (2C), 113.8,
111.7, 107.2, 80.5, 68.0, 55.7, 55.6 ppm .
$\boldsymbol{R}_{f}: 0.73$ ( $6: 4 \mathrm{AcOEt} /$ hexanes).
HRMS (ESI-TOF): $m / z[\mathrm{M}+\mathrm{H}]^{+}$calc. for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{O}_{6}: 319.1182$, found: 319.1185 .

At the end of this study, after approximately 700 reactions showing the same catalytic activity, BAR was unloaded from the packed-bed reactor and analyzed by SEM (Figure S3).


Figure S3: SEM images of recycled BAR at x100 and x350 magnifications.

Considering solvent recycle, the $E$ Factor was calculated as follows: Obtained mass of products $=72.7 \mathrm{mg}(56.2 \mathrm{mg}$ of $\mathbf{6}+16.5 \mathrm{mg}$ of $\mathbf{2 0})$; Mass of reagents $=1008.3 \mathrm{mg}(60.7$ mg of $\mathrm{H}_{2} \mathrm{O}_{2}+342.8 \mathrm{mg}$ of $\mathrm{MeSO}_{3} \mathrm{H}+604.8 \mathrm{mg}$ of $\mathrm{NaHCO}_{3}$ ); Amount of the starting material $\mathbf{1 3}=179.7 \mathrm{mg}$.
$E$ Factor $=(179.7+1008.3-72.7) / 72.7=15.3$.

## Experimental

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#### Abstract

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Figure S4. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 0}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S5. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 0}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.



Figure S6. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 1}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S7. ${ }^{13} \mathrm{C}$ NMR Spectrum of $11\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S8. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 2}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S9. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 2}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.






Figure S10．${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 3}\left(300 \mathrm{MHz}\right.$ ，acetone－ $\left.\mathrm{d}_{6}\right)$ ．


Figure S11．${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 3}\left(75 \mathrm{MHz}\right.$ ，acetone－ $\left.\mathrm{d}_{6}\right)$ ．
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Figure S12. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 4}\left(300 \mathrm{MHz}\right.$, acetone- $\left.\mathrm{d}_{6}\right)$.


Figure S13. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 4}\left(75 \mathrm{MHz}\right.$, acetone- $\left.\mathrm{d}_{6}\right)$.




Figure S14. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 5}\left(300 \mathrm{MHz}\right.$, acetone- $\left.\mathrm{d}_{6}\right)$.


Figure S15. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 5}\left(75 \mathrm{MHz}\right.$, acetone- $\left.\mathrm{d}_{6}\right)$.


Figure S16. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 6}\left(300 \mathrm{MHz}\right.$, acetone- $\left.\mathrm{d}_{6}\right)$.


Figure S17. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 6}\left(75 \mathrm{MHz}\right.$, acetone- $\left.\mathrm{d}_{6}\right)$.


Figure S18. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 7}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S19. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 7}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S20. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{6}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S21. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{6}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S22. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 8}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S23. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 8}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S24. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{1 9}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S25. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{1 9}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S26. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{2 0}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S27. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{2 0}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.



Figure S28. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{2 1}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S29. ${ }^{13} \mathrm{C}$ NMR Spectrum of $21\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S30. ${ }^{1} \mathrm{H}$ NMR Spectrum of $22\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S31. ${ }^{13} \mathrm{C}$ NMR Spectrum of $22\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S32. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{8}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.



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Figure S33. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{8}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.

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Figure S34. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{2 3}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S35. ${ }^{13} \mathrm{C}$ NMR Spectrum of $\mathbf{2 3}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S36. ${ }^{1} \mathrm{H}$ NMR Spectrum of $5\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S37. ${ }^{1} \mathrm{H}$ NMR Spectrum of $7\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.


Figure S39. ${ }^{1} \mathrm{H}$ NMR Spectrum of $\mathbf{2 4}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$.

