Electronic Supplementary Material (ESI) for Journal of Analytical Atomic Spectrometry. This journal is © The Royal Society of Chemistry 2020

## Supplementary data

## Deep eutectic solvent-based extraction of uranium(VI) from a wide range acidity and subsequent determination by direct loading in thermal ionization mass spectrometry

Raju V. Shah<sup>a,b,\*</sup>, Ashok K. Pandey<sup>b,c</sup>, K. Sasi Bhushan<sup>a</sup>, S. Jagadish Kumar<sup>a</sup>, Radhika M. Rao<sup>a</sup>, P.G. Jaison<sup>a</sup>

<sup>a</sup>Fuel Chemistry Division, Bhabha Atomic Research Centre, Trombay, Mumbai 400 085, India

<sup>b</sup>Homi Bhabha National Institute, Anushaktinagar, Mumbai 400 094, India

<sup>c</sup>Radiochemistry Division, Bhabha Atomic Research Centre, Trombay, Mumbai 400 085, India

Table S1 :Composition of DES:

TOPO:IA DES	Mole Fraction of TOPO	Mole Fraction of Itaconic Acid	Appearance of Mixture
Α	0.1	0.9	Thick Gel
В	0.3	0.7	Highly viscous liquid
С	0.4	0.6	Highly viscous liquid
D	0.5	0.5	Free flowing liquid
E	0.6	0.4	Free flowing liquid
F	0.8	0.2	Highly viscous liquid

**Fig. S1**.DSC thermogram of the TOPO:IA DES having 0.6:0.4 mole proportions.



Fig. S2.Effect of time on extraction of Uranium.



**Table S2.** Variation of extraction efficiency and distribution coefficient ( $K_d$ , mL.g<sup>-1</sup>) of U(VI) in the TOPO:IA-PP as a function pH in the equilibrating solution.

рН	Extraction Efficiency (%)	Weight of Membrane (g)	K <sub>d</sub> (mL.g <sup>-1</sup> )
1	86	0.0214	2871
3	87	0.0186	3521
5	90	0.0214	4109
7	89	0.0219	3608
9	88	0.0213	3527

Chemical Structure and properties of TOPO and Itaconic Acid:



**Trioctylphosphineoxide:** Molecular Formula :  $C_{24}H_{51}OP$ ; Molar Mass: 386.635 g/mol; Density :0.861g/cm<sup>3</sup>; Melting Point : 50-55°C; Solubility in water: Insoluble.



**Itaconic Acid (methylidenesuccinic acid):** Molecular Formula:  $C_5H_6O_4$ ; Molar Mass: 128.084 g/mol; Melting Point: 166-167 °C; Solubility in water: 1 g/12 mL.

<b>Chemical Constituent</b>	Concentration(mg/L)
pH (units)	8.1
Ca <sup>2+</sup>	35
Mg <sup>2+</sup>	80
Na +	740
$K^+$	7
CO <sub>3</sub> <sup>2-</sup>	25
HCO <sub>3</sub> -	435
Cl-	1015
SO4 <sup>2-</sup>	185
NO <sub>3</sub> -	29
F-	1

Synthetic ground water was prepared as per composition mentioned in reference.<sup>28</sup>

Synthetic urine was prepared as per composition mentioned in reference.<sup>29</sup>

## Composition of synthetic urine sample

Chemical Constituent	Concentration (g L <sup>-1</sup> )
pH (units)	6
Urea	20
NaCl	6.3
K <sub>2</sub> HPO <sub>4</sub>	4.9
MgCl <sub>2</sub> anhydrous	0.35
CaCl <sub>2</sub> dihydrate	0.43
Citric acid monohydrate	1.0

## **Composition of Seawater**

U concentration =  $1.4x10^{-8}$  mol/L (3.3 mg/m<sup>3</sup>). *pH* = 7.5-8.5,

*Salt conc*.: [NaCl] = 0.55 mol/L, [HCO<sub>3</sub>-] = 2.5 mmol/L, [SO<sub>4</sub><sup>2-</sup>] = 27 mmol/L *Na*=10770 *ppm, Mg* = 1290 *ppm,* 

Ca = 412 ppm,

K =3 80 ppm

Si, Li, Rb, I, Ba, Mo = 2-0.01 ppm,

As, U, V, Ti, Fe = 0.004 - 0.001 ppm

Zn, Ni, Al, Cs, Cr, Sb, Se, Mn,Cu,W = 0.0005-0.0001 ppm