

Electronic supplementary information for:

**Exploring The Anion Chemical Space of $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$
(Ln = Y, La, Gd): An Electroelastic Material with
High Mechanical Sensitivity and Energy Harvesting**

Evgenii Strugovshchikov* and Aleksandr Pishtshev

Institute of Physics, University of Tartu, W.Ostwaldi 1, 50411 Tartu, Estonia

E-mail: evgenii.strugovshchikov@ut.ee, aleksandr.pishtshev@ut.ee

CONTENTS

1.	Equilibrium atomic positions predicted for $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ (Ln = Y, La, Gd)	S2
2.	Bond distances and bond angles in $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ (Ln = Y, La, Gd).....	S6
3.	Elastic properties of $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ (Ln = Y, La, Gd)	S9
4.	Zone-centered optical vibrational modes calculated in the harmonic approximation.....	S12
5.	Simulation of X-ray diffraction patterns.....	S18
6.	Radial distribution function evaluated for the shortest interatomic distances	S22
7.	Piezoelectric properties of $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ (Ln = Y, La, Gd).....	S23
	References:.....	S25

1. Equilibrium atomic positions predicted for $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ ($\text{Ln} = \text{Y, La, Gd}$)

Supplementary Table S1: $\text{Y}_2\text{OF}_2\text{H}_2$, P3m1

P3m1 (156); $a = 3.712 \text{ \AA}$, $c = 6.941 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
Y1	1a	3m.	0.0000	0.0000	0.6421
Y2	1c	3m.	0.6667	0.3333	0.2102
O1	1c	3m.	0.6667	0.3333	0.5306
F1	1a	3m.	0.0000	0.0000	0.0000
F2	1b	3m.	0.3333	0.6667	0.7721
H1	1a	3m.	0.0000	0.0000	0.3294
H2	1b	3m.	0.3333	0.6667	0.2326

Supplementary Table S2: $\text{Y}_2\text{OF}_2\text{H}_2$, R3m

R3m (160); $a = 3.657 \text{ \AA}$, $c = 22.014 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
Y1	3a	3m	0.0000	0.0000	0.0000
Y2	3a	3m	0.0000	0.0000	0.4710
O1	3a	3m	0.0000	0.0000	0.3659
F1	3a	3m	0.0000	0.0000	0.1964
F2	3a	3m	0.0000	0.0000	0.6168
H1	3a	3m	0.0000	0.0000	0.7927
H2	3a	3m	0.0000	0.0000	0.0949

Supplementary Table S3: $\text{La}_2\text{OF}_2\text{H}_2$, P3m1

P3m1 (156); $a = 3.997 \text{ \AA}$, $c = 7.288 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	1a	3m.	0.0000	0.0000	0.6430
La2	1b	3m.	0.3333	0.6667	0.2028
O1	1b	3m.	0.3333	0.6667	0.5279
F1	1a	3m.	0.0000	0.0000	0.0000
F2	1c	3m.	0.6667	0.3333	0.7740
H1	1a	3m.	0.0000	0.0000	0.3250
H2	1c	3m.	0.6667	0.3333	0.2464

Supplementary Table S4: $\text{La}_2\text{OF}_2\text{H}_2$, R3m

R3m (160); $a = 3.978 \text{ \AA}$, $c = 21.444 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	3a	3m	0.0000	0.0000	0.0000
La2	3a	3m	0.0000	0.0000	0.4847
O1	3a	3m	0.0000	0.0000	0.3705
F1	3a	3m	0.0000	0.0000	0.2130
F2	3a	3m	0.0000	0.0000	0.6163
H1	3a	3m	0.0000	0.0000	0.8035
H2	3a	3m	0.0000	0.0000	0.1046

Supplementary Table S5: Gd₂OF₂H₂, P3m1

P3m1 (156); a = 3.783 Å, c = 7.026 Å					
Atom	Wyckoff position	Symm.	x	y	z
Gd1	1a	3m.	0.0000	0.0000	0.6414
Gd2	1c	3m.	0.6667	0.3333	0.2079
O1	1c	3m.	0.6667	0.3333	0.5287
F1	1a	3m.	0.0000	0.0000	0.0000
F2	1b	3m.	0.3333	0.6667	0.7722
H1	1a	3m.	0.0000	0.0000	0.3289
H2	1b	3m.	0.3333	0.6667	0.2338

Supplementary Table S6: Gd₂OF₂H₂, R3m

R3m (160); a = 3.742 Å, c = 21.562 Å					
Atom	Wyckoff position	Symm.	x	y	z
Gd1	3a	3m	0.0000	0.0000	0.0000
Gd2	3a	3m	0.0000	0.0000	0.4760
O1	3a	3m	0.0000	0.0000	0.3676
F1	3a	3m	0.0000	0.0000	0.2024
F2	3a	3m	0.0000	0.0000	0.6159
H1	3a	3m	0.0000	0.0000	0.7972
H2	3a	3m	0.0000	0.0000	0.0981

Supplementary Table S7: Y₂OFClH₂, P3m1

P3m1 (156); a = 3.735 Å, c = 8.181 Å					
Atom	Wyckoff position	Symm.	x	y	z
Y1	1a	3m.	0.0000	0.0000	0.0000
Y2	1b	3m.	0.3333	0.6667	0.3637
O1	1b	3m.	0.3333	0.6667	0.0891
F1	1c	3m.	0.6667	0.3333	0.8849
Cl1	1a	3m.	0.0000	0.0000	0.5930
H1	1a	3m.	0.0000	0.0000	0.2660
H2	1c	3m.	0.6667	0.3333	0.3393

Supplementary Table S8: Y₂OFClH₂, R3m

R3m (160); a = 3.712 Å, c = 25.933 Å					
Atom	Wyckoff position	Symm.	x	y	z
Y1	3a	3m	0.0000	0.0000	0.0000
Y2	3a	3m	0.0000	0.0000	0.4484
O1	3a	3m	0.0000	0.0000	0.3604
F1	3a	3m	0.0000	0.0000	0.6270
Cl1	3a	3m	0.0000	0.0000	0.1863
H1	3a	3m	0.0000	0.0000	0.7724
H2	3a	3m	0.0000	0.0000	0.0829

Supplementary Table S9: La₂OFClH₂, P3m1

P3m1 (156); $a = 4.048 \text{ \AA}$, $c = 8.374 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	1a	3m.	0.0000	0.0000	0.6153
La2	1c	3m.	0.6667	0.3333	0.2354
O1	1c	3m.	0.6667	0.3333	0.5166
F1	1b	3m.	0.3333	0.6667	0.7185
Cl1	1a	3m.	0.0000	0.0000	0.0000
H1	1a	3m.	0.0000	0.0000	0.3342
H2	1b	3m.	0.3333	0.6667	0.2672

Supplementary Table S10: La₂OFClH₂, R3m

R3m (160); $a = 3.712 \text{ \AA}$, $c = 25.933 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	3a	3m	0.0000	0.0000	0.0000
La2	3a	3m	0.0000	0.0000	0.4603
O1	3a	3m	0.0000	0.0000	0.3660
F1	3a	3m	0.0000	0.0000	0.6315
Cl1	3a	3m	0.0000	0.0000	0.2049
H1	3a	3m	0.0000	0.0000	0.7823
H2	3a	3m	0.0000	0.0000	0.0928

Supplementary Table S11: Gd₂OFClH₂, P3m1

P3m1 (156); $a = 3.812 \text{ \AA}$, $c = 8.181 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
Gd1	1a	3m.	0.0000	0.0000	0.0000
Gd2	1b	3m.	0.3333	0.6667	0.3693
O1	1b	3m.	0.3333	0.6667	0.0923
F1	1c	3m.	0.6667	0.3333	0.8874
Cl1	1a	3m.	0.0000	0.0000	0.6009
H1	1a	3m.	0.0000	0.0000	0.2706
H2	1c	3m.	0.6667	0.3333	0.3434

Supplementary Table S12: Gd₂OFClH₂, R3m

R3m (160); $a = 3.712 \text{ \AA}$, $c = 25.933 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	3a	3m	0.0000	0.0000	0.0000
La2	3a	3m	0.0000	0.0000	0.4502
O1	3a	3m	0.0000	0.0000	0.3613
F1	3a	3m	0.0000	0.0000	0.6274
Cl1	3a	3m	0.0000	0.0000	0.1887
H1	3a	3m	0.0000	0.0000	0.7735
H2	3a	3m	0.0000	0.0000	0.0843

Supplementary Table S13: LaGdOFClH₂, P3m1

P3m1 (156); $a = 3.939 \text{ \AA}$, $c = 8.292 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	1a	3m.	0.0000	0.0000	0.6028
Gd1	1c	3m.	0.6667	0.3333	0.2288
O1	1c	3m.	0.6667	0.3333	0.4995
F1	1b	3m.	0.3333	0.6667	0.7163
Cl1	1a	3m.	0.0000	0.0000	0.0000
H1	1a	3m.	0.0000	0.0000	0.3207
H2	1b	3m.	0.3333	0.6667	0.2530

Supplementary Table S14: LaGdOFClH₂, R3m

R3m (160); $a = 3.924 \text{ \AA}$, $c = 25.246 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	3a	3m	0.0000	0.0000	0.0000
Gd1	3a	3m	0.0000	0.0000	0.4563
O1	3a	3m	0.0000	0.0000	0.3667
F1	3a	3m	0.0000	0.0000	0.6277
Cl1	3a	3m	0.0000	0.0000	0.1971
H1	3a	3m	0.0000	0.0000	0.7806
H2	3a	3m	0.0000	0.0000	0.0916

Supplementary Table S15: Y₂OCl₂H₂, R3m

R3m (160); $a = 3.756 \text{ \AA}$, $c = 30.481 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
Y1	3a	3m	0.0000	0.0000	0.0000
Y2	3a	3m	0.0000	0.0000	0.4307
O1	3a	3m	0.0000	0.0000	0.3545
Cl1	3a	3m	0.0000	0.0000	0.1563
Cl2	3a	3m	0.0000	0.0000	0.6094
H1	3a	3m	0.0000	0.0000	0.7540
H2	3a	3m	0.0000	0.0000	0.0698

Supplementary Table S16: La₂OCl₂H₂, R3m

R3m (160); $a = 4.056 \text{ \AA}$, $c = 30.999 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
La1	3a	3m	0.0000	0.0000	0.0000
La2	3a	3m	0.0000	0.0000	0.4356
O1	3a	3m	0.0000	0.0000	0.3561
Cl1	3a	3m	0.0000	0.0000	0.1619
Cl2	3a	3m	0.0000	0.0000	0.6092
H1	3a	3m	0.0000	0.0000	0.7548
H2	3a	3m	0.0000	0.0000	0.0739

Supplementary Table S17: Gd₂OCl₂H₂, R3m

R3m (160); $a = 3.824 \text{ \AA}$, $c = 30.722 \text{ \AA}$					
Atom	Wyckoff position	Symm.	x	y	z
Gd1	3a	3m	0.0000	0.0000	0.0000
Gd2	3a	3m	0.0000	0.0000	0.4316
O1	3a	3m	0.0000	0.0000	0.3549
Cl1	3a	3m	0.0000	0.0000	0.1570
Cl2	3a	3m	0.0000	0.0000	0.6096
H1	3a	3m	0.0000	0.0000	0.7545
H2	3a	3m	0.0000	0.0000	0.0704

2. Bond distances and bond angles in Ln₂OF_{2-x}Cl_xH₂ (Ln = Y, La, Gd)**Supplementary Table S18:** Bond distances (in Å) and bond angles (in degrees) in Y₂OF₂H₂

Y ₂ OF ₂ H ₂ , P3m1				Y ₂ OF ₂ H ₂ , R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
Y1 – Y2	3.68			Y1 – Y2	3.69		
Y1 – O1	2.28	Y1 – O1 – Y2	109.9	Y1 – O1	2.23	Y1 – O1 – Y2	108.8
Y2 – O1	2.22			Y2 – O1	2.31		
Y1 – H2	2.17	Y2 – H1 – H2	166.8	Y1 – H2	2.09	Y2 – H1 – H2	169.0
Y2 – H1	2.15			Y2 – H1	2.13		
H1 – H2	2.25			H1 – H2	2.22		
H2 – F1	2.29	Y1 – H2 – F1	180.0	H2 – F1	2.23	Y1 – H2 – F1	180.0
Y1 – F2	2.33	Y2 – F1 – F2	177.8	Y1 – F2	2.38	Y1 – F2 – F1	165.2
Y1 – F1	2.48			Y2 – F2	3.21		
Y2 – F1	2.59			Y2 – F1	2.48		
F2 – F1	2.66			F2 – F1	2.85		

Supplementary Table S19: Bond distances (in Å) and bond angles (in degrees) in La₂OF₂H₂

La ₂ OF ₂ H ₂ , P3m1				La ₂ OF ₂ H ₂ , R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
La1 – La2	3.95			La1 – La2	3.98		
La1 – O1	2.46	La1 – O1 – La2	110.0	La1 – O1	2.43	La1 – O1 – La2	109.1
La2 – O1	2.37			La2 – O1	2.45		
La1 – H2	2.32	La2 – H1 – H2	173.9	La1 – H2	2.24	La2 – H1 – H2	170.9
La2 – H1	2.33			La2 – H1	2.32		
H1 – H2	2.38			H1 – H2	2.40		
H2 – F1	2.37	La1 – H2 – F1	180.0	H2 – F1	2.33	La1 – H2 – F1	180.0
La1 – F2	2.50	La1 – F1 – F2	177.1	La1 – F2	2.54	La1 – F2 – F1	172.1
La1 – F1	2.60			La1 – F1	3.45		
La2 – F1	2.74			La2 – F1	2.65		
F2 – F1	2.84			F2 – F1	2.74		

Supplementary Table S20: Bond distances (in Å) and bond angles (in degrees) in $\text{Gd}_2\text{OF}_2\text{H}_2$

$\text{Gd}_2\text{OF}_2\text{H}_2$, P3m1				$\text{Gd}_2\text{OF}_2\text{H}_2$, R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
Gd1 – Gd2	3.75			Gd1 – Gd2	3.76		
Gd1 – O1	2.32	Gd1 – O1 – Gd2	109.9	Gd1 – O1	2.28	Gd1 – O1 – Gd2	108.9
Gd2 – O1	2.25			Gd2 – O1	2.34		
Gd1 – H2	2.20	Gd2 – H1 – H2	167.8	Gd1 – H2	2.11	Gd2 – H1 – H2	168.9
Gd2 – H1	2.19			Gd2 – H1	2.18		
H1 – H2	2.28			H1 – H2	2.27		
H2 – F1	2.31	Gd1 – H2 – F1	180.0	H2 – F1	2.25	Gd1 – H2 – F1	180.0
Gd1 – F2	2.37	Gd1 – F1 – F2	177.5	Gd1 – F2	2.42	Gd1 – F2 – F1	168.2
Gd1 – F1	2.52			Gd2 – F2	3.02		
Gd2 – F1	2.63			Gd2 – F1	2.51		
F2 – F1	2.71			F2 – F1	2.77		

Supplementary Table S21: Bond distances (in Å) and bond angles (in degrees) in Y_2OFClH_2

Y_2OFClH_2, P3m1				Y_2OFClH_2, R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
Y1 – Y2	3.67			Y1 – Y2	3.67		
Y1 – O1	2.28	Y1 – O1 – Y2	108.7	Y1 – O1	2.26	Y1 – O1 – Y2	108.2
Y2 – O1	2.25			Y2 – O1	2.28		
Y1 – H2	2.18	Y2 – H1 – H2	169.7	Y1 – H2	2.15	Y2 – H1 – H2	171.0
Y2 – H1	2.17			Y2 – H1	2.16		
H1 – H2	2.24			H1 – H2	2.22		
H2 – Cl1	2.68	Y1 – H2 – Cl1	180.0	H2 – Cl1	2.68	Y1 – H2 – Cl1	180.0
Y1 – F1	2.35	Y2 – Cl1 – F1	173.1	Y1 – F1	2.38	Y1 – F1 – Cl1	153.2
Y1 – Cl1	3.33			Y2 – F1	4.63		
Y2 – Cl1	2.86			Y2 – Cl1	2.83		
Cl1 – F1	3.22			Cl1 – F1	3.51		

Supplementary Table S22: Bond distances (in Å) and bond angles (in degrees) in $\text{La}_2\text{OFClH}_2$

$\text{La}_2\text{OFClH}_2$, P3m1				$\text{La}_2\text{OFClH}_2$, R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
La1 – La2	3.95			La1 – La2	3.94		
La1 – O1	2.48	La1 – O1 – La2	109.5	La1 – O1	2.47	La1 – O1 – La2	109.4
La2 – O1	2.36			La2 – O1	2.36		
La1 – H2	2.35	La2 – H1 – H2	173.0	La1 – H2	2.33	La2 – H1 – H2	173.1
La2 – H1	2.35			La2 – H1	2.35		
H1 – H2	2.40			H1 – H2	2.40		
H2 – Cl1	2.80	La1 – H2 – Cl1	180.0	H2 – Cl1	2.81	La1 – H2 – Cl1	180.0
La1 – F1	2.49	La2 – Cl1 – F1	174.9	La1 – F1	2.49	La1 – F1 – Cl1	155.7
La1 – Cl1	3.22			La2 – F1	4.29		
La2 – Cl1	3.06			La2 – Cl1	3.04		
Cl1 – F1	3.32			Cl1 – F1	3.30		

Supplementary Table S23: Bond distances (in Å) and bond angles (in degrees) in $\text{Gd}_2\text{OFClH}_2$

$\text{Gd}_2\text{OFClH}_2$, P3m1				$\text{Gd}_2\text{OFClH}_2$, R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
Gd1 – Gd2	3.74			Gd1 – Gd2	3.74		
Gd1 – O1	2.33	Gd1 – O1 – Gd2	108.9	Gd1 – O1	2.30	Gd1 – O1 – Gd2	108.4
Gd2 – O1	2.27			Gd2 – O1	2.30		
Gd1 – H2	2.21	Gd2 – H1 – H2	170.4	Gd1 – H2	2.18	Gd2 – H1 – H2	171.8
Gd2 – H1	2.21			Gd2 – H1	2.20		
H1 – H2	2.28			H1 – H2	2.26		
H2 – Cl1	2.70	Gd1 – H2 – Cl1	180.0	H2 – Cl1	2.71	Gd1 – H2 – Cl1	180.0
Gd1 – F1	2.39	Gd2 – F1 – Cl1	173.9	Gd1 – F1	2.41	Gd1 – F1 – Cl1	153.7
Gd1 – Cl1	3.26			Gd2 – F1	4.59		
Gd2 – Cl1	2.90			Gd2 – Cl1	2.87		
Cl1 – F1	3.21			Cl1 – F1	3.50		

Supplementary Table S24: Bond distances (in Å) and bond angles (in degrees) in LaGdOFClH_2

LaGdOFClH_2, P3m1				LaGdOFClH_2, R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
La1 – Gd1	3.85			La1 – Gd1	3.84		
Gd1 – O1	2.24	Gd1 – O1 – La1	110.6	Gd1 – O1	2.26	Gd1 – O1 – La1	110.4
La1 – O1	2.43			La1 – O1	2.42		
La1 – H2	2.34	Gd1 – H1 – H2	171.2	La1 – H2	2.31	Gd1 – H1 – H2	171.7
Gd1 – H1	2.28			Gd1 – H1	2.28		
H1 – H2	2.34			H1 – H2	2.33		
H2 – Cl1	2.66	La1 – H2 – Cl1	180.0	H2 – Cl1	2.66	La1 – H2 – Cl1	180.0
La1 – F1	2.46	Gd1 – Cl1 – F1	173.9	La1 – F1	2.47	La1 – F1 – Cl1	156.2
La1 – Cl1	3.29			Gd1 – F1	4.33		
Gd1 – Cl1	2.96			Gd1 – Cl1	2.93		
Cl1 – F1	3.27			Cl1 – F1	3.34		

Supplementary Table S25: Bond distances (in Å) and bond angles (in degrees) in $\text{Y}_2\text{OCl}_2\text{H}_2$ and $\text{La}_2\text{OCl}_2\text{H}_2$

$\text{Y}_2\text{OCl}_2\text{H}_2$, R3m				$\text{La}_2\text{OCl}_2\text{H}_2$, R3m			
Bond	Value	Angle	Value	Bond	Value	Angle	Value
Y1 – Y2	3.68			La1 – La2	3.94		
Y1 – O1	2.26	Y1 – O1 – Y2	106.6	La1 – O1	2.45	La1 – O1 – La2	
Y2 – O1	2.32			La2 – O1	2.47		
Y1 – H2	2.13	Y2 – H1 – H2	174.2	La1 – H2	2.29	La2 – H1 – H2	180.0
Y2 – H1	2.19			La2 – H1	2.38		
H1 – H2	2.23			H1 – H2	2.38		
H2 – Cl1	2.64	Y1 – H2 – Cl1	180.0	H2 – Cl1	2.73	La1 – H2 – Cl1	180.0
Y1 – Cl2	2.78	Y2 – Cl1 – Cl2	159.5	La1 – Cl2	2.94	La1 – Cl2 – Cl1	160.8
Y2 – Cl2	5.45			La2 – Cl2	5.38		
Y2 – Cl1	2.78			La1 – Cl2	2.94		
Cl2 – Cl1	4.24			Cl2 – Cl1	4.24		

Supplementary Table S26: Bond distances (in Å) and bond angles (in degrees) in $\text{Gd}_2\text{OCl}_2\text{H}_2$

$\text{Gd}_2\text{OCl}_2\text{H}_2$, R3m							
Bond	Value	Angle	Value				
Gd1 – Gd2	3.74						
Gd1 – O1	2.30	Gd1 – O1 – Gd2	106.7				
Gd2 – O1	2.36						
Gd1 – H2	2.16	Gd2 – H1 – H2	174.6				
Gd2 – H1	2.23						
H1 – H2	2.27						
H2 – Cl1	2.66	Gd1 – H2 – Cl1	180.0				
Gd1 – Cl2	2.82	Gd1 – Cl2 – Cl1	159.5				
Gd2 – Cl2	5.47						
Gd1 – Cl2	2.82						
Cl2 – Cl1	4.28						

3. Elastic properties of $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ ($\text{Ln} = \text{Y}, \text{La}, \text{Gd}$)

Supplementary table S27: Components of the elasticity tensors C_{ij} (in GPa)

$\text{Y}_2\text{OF}_2\text{H}_2$ (P3m1)						$\text{Y}_2\text{OF}_2\text{H}_2$ (R3m)					
170.1	64.1	51.0	8.0	0	0	194.5	68.3	26.1	-12.1	0	0
64.1	170.1	51.0	-8.0	0	0	68.3	194.5	26.1	12.1	0	0
51.0	51.0	136.3	0	0	0	26.1	26.1	23.2	0	0	0
8.0	-8.0	0	25.6	0	0	-12.1	12.1	0	31.6	0	0
0	0	0	0	25.6	8.0	0	0	0	0	31.6	-12.1
0	0	0	0	8.0	53.0	0	0	0	0	-12.1	63.1
$\text{La}_2\text{OF}_2\text{H}_2$ (P3m1)						$\text{La}_2\text{OF}_2\text{H}_2$ (R3m)					
145.3	48.6	30.9	0.4	0	0	173.5	68.7	40.9	-17.7	0	0
48.6	145.3	30.9	-0.4	0	0	68.7	173.5	40.9	17.7	0	0
30.9	30.9	117.6	0	0	0	40.9	40.9	70.0	0	0	0
0.4	-0.4	0	20.3	0	0	-17.7	17.7	0	35.8	0	0
0	0	0	0	20.3	0.4	0	0	0	0	35.8	-17.7
0	0	0	0	0.4	48.3	0	0	0	0	-17.7	52.4
$\text{Gd}_2\text{OF}_2\text{H}_2$ (P3m1)						$\text{Gd}_2\text{OF}_2\text{H}_2$ (R3m)					
164.8	59.3	45.0	4.2	0	0	192.4	69.5	32.4	-15.0	0	0
59.3	164.8	45.0	-4.2	0	0	69.5	192.4	32.4	15.0	0	0
45.0	45.0	132.7	0	0	0	32.4	32.4	31.2	0	0	0
4.2	-4.2	0	25.5	0	0	-15.0	15.0	0	37.9	0	0
0	0	0	0	25.5	4.2	0	0	0	0	37.9	-15.0
0	0	0	0	4.2	52.7	0	0	0	0	-15.0	61.5

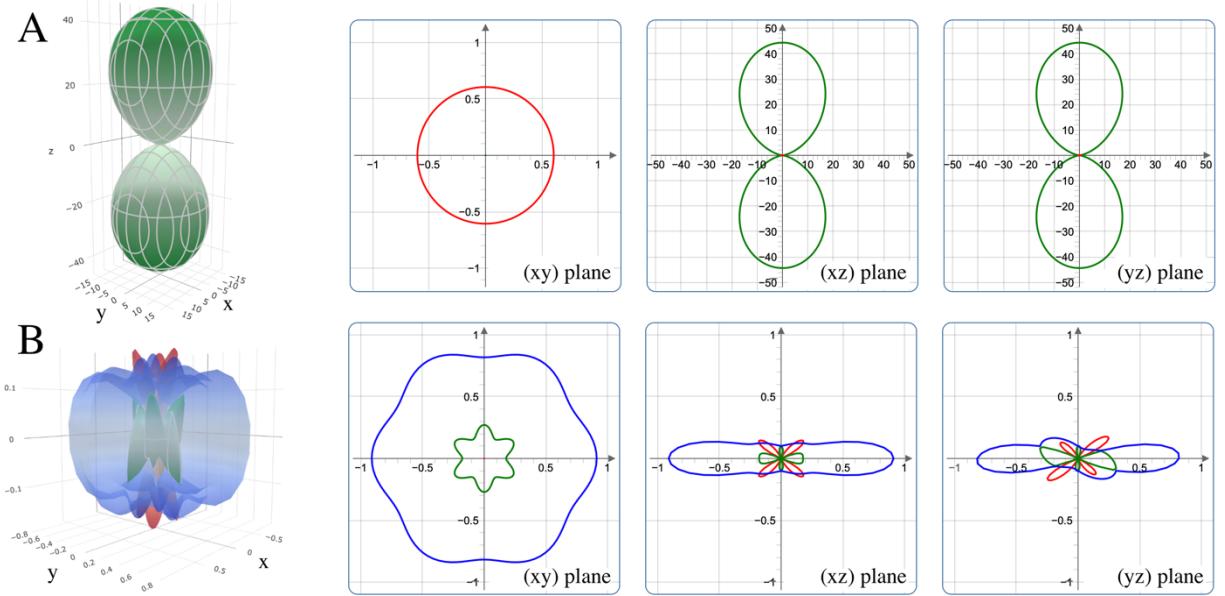
Y_2OFClH_2 (P3m1)						Y_2OFClH_2 (R3m)					
167.2	45.4	31.3	-5.2	0	0	158.9	43.2	11.2	-2.2	0	0
45.4	167.2	31.3	5.2	0	0	43.2	158.9	11.2	2.2	0	0
31.3	31.3	41.2	0	0	0	11.2	11.2	19.6	0	0	0
-5.2	5.2	0	24.4	0	0	-2.2	2.2	0	5.9	0	0
0	0	0	0	24.4	-5.2	0	0	0	0	5.9	-2.2
0	0	0	0	-5.2	60.9	0	0	0	0	-2.2	57.9
$\text{La}_2\text{OFClH}_2$ (P3m1)						$\text{La}_2\text{OFClH}_2$ (R3m)					
143.6	40.7	28.2	-4.0	0	0	139.4	44.5	21.7	-6.9	0	0
40.7	143.6	28.2	4.0	0	0	44.5	139.4	21.7	6.9	0	0

28.2	28.2	78.9	0	0	0	21.7	21.7	27.3	0	0	0
-4.0	4.0	0	18.8	0	0	-6.9	6.9	0	14.4	0	0
0	0	0	0	18.8	-4.0	0	0	0	0	14.4	-6.9
0	0	0	0	-4.0	47.0	0	0	0	0	-6.9	47.4
Gd ₂ OFClH ₂ (P3m1)						Gd ₂ OFClH ₂ (R3m)					
158.7	44.4	33.0	-4.9	0	0	151.5	41.2	12.0	-2.5	0	0
44.4	158.7	33.0	4.9	0	0	41.2	151.5	12.0	2.5	0	0
33.0	33.0	52.8	0	0	0	12.0	12.0	18.6	0	0	0
-4.9	4.9	0	24.7	0	0	-2.5	2.5	0	6.1	0	0
0	0	0	0	24.7	-4.9	0	0	0	0	6.1	-2.5
0	0	0	0	-4.9	57.1	0	0	0	0	-2.5	55.1
LaGdOFClH ₂ (P3m1)						LaGdOFClH ₂ (R3m)					
140.0	42.2	28.4	-2.5	0	0	143.6	42.7	20.4	-6.4	0	0
42.2	140.0	28.4	2.5	0	0	42.7	143.6	20.4	6.4	0	0
28.4	28.4	71.1	0	0	0	20.4	20.4	25.1	0	0	0
-2.5	2.5	0	20.0	0	0	-6.4	6.4	0	12.9	0	0
0	0	0	0	20.0	-2.5	0	0	0	0	12.9	-6.4
0	0	0	0	-2.5	48.9	0	0	0	0	-6.4	50.5
Y ₂ OCl ₂ H ₂ (R3m)						La ₂ OCl ₂ H ₂ (R3m)					
124.4	34.1	2.6	-0.2	0	0	98.8	30.0	2.9	-0.4	0	0
34.1	124.4	2.6	0.2	0	0	30.0	98.8	2.9	0.4	0	0
2.6	2.6	8.8	0	0	0	2.9	2.9	7.7	0	0	0
-0.2	0.2	0	1.5	0	0	-0.4	0.4	0	1.4	0	0
0	0	0	0	1.5	-0.2	0	0	0	0	1.4	-0.4
0	0	0	0	-0.2	45.1	0	0	0	0	-0.4	34.4
Gd ₂ OCl ₂ H ₂ (R3m)											
116.7	32.3	2.3	-0.2	0	0						
32.3	116.7	2.3	0.2	0	0						
2.3	2.3	7.7	0	0	0						
-0.2	0.2	0	1.0	0	0						
0	0	0	0	1.0	-0.2						
0	0	0	0	-0.2	42.2						

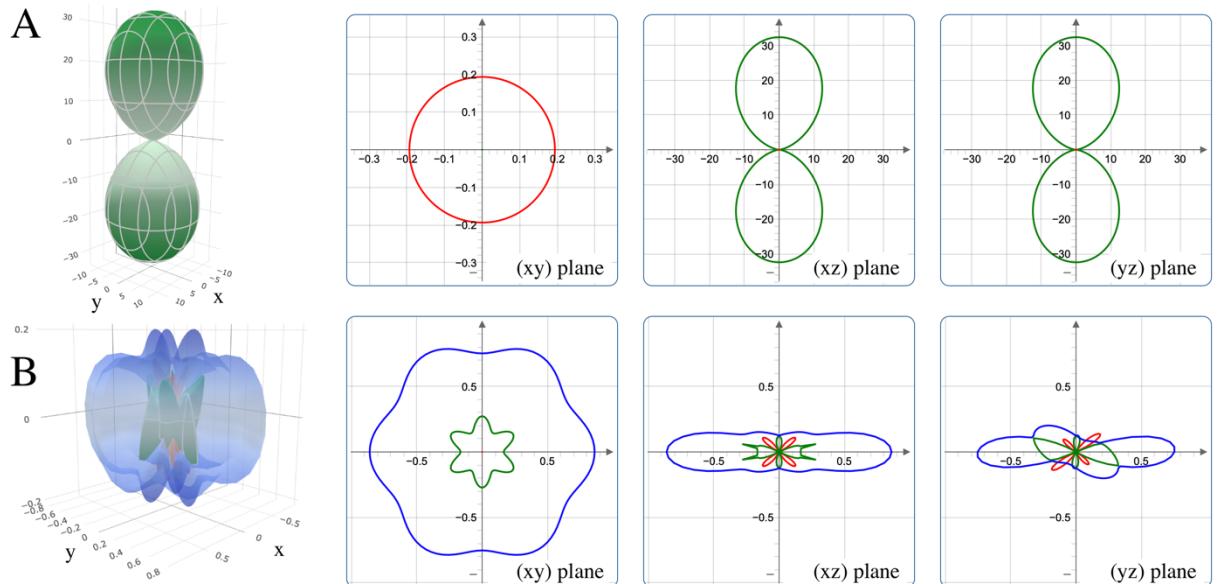
Supplementary table S28: Eigenvalues (λ) of the elasticity tensor (in Gpa)

	λ_1	λ_2	λ_3	λ_4	λ_5	λ_6
Y ₂ OF ₂ H ₂ (P3m1)	23.4	24.0	55.1	98.1	107.5	272.4
Y ₂ OF ₂ H ₂ (R3m)	17.7	27.4	28.6	67.3	129.3	268.3
La ₂ OF ₂ H ₂ (P3m1)	20.3	20.3	48.3	96.7	97.7	213.8
La ₂ OF ₂ H ₂ (R3m)	24.5	27.6	52.4	63.6	112.9	259.8
Gd ₂ OF ₂ H ₂ (P3m1)	24.9	25.1	53.4	100.1	105.9	256.7
Gd ₂ OF ₂ H ₂ (R3m)	22.5	30.6	32.9	68.7	127.9	270.7
Y ₂ OFClH ₂ (P3m1)	23.7	23.8	30.5	61.6	122.4	223.4
Y ₂ OFClH ₂ (R3m)	5.8	5.8	18.2	58.0	115.9	203.4
La ₂ OFClH ₂ (P3m1)	18.3	18.4	47.5	64.6	94.3	189.7
La ₂ OFClH ₂ (R3m)	13.0	13.2	21.5	48.8	96.1	189.7
Gd ₂ OFClH ₂ (P3m1)	24.0	24.2	39.5	57.9	114.8	216.4
Gd ₂ OFClH ₂ (R3m)	6.0	6.0	17.0	55.3	110.4	194.3
LaGdOFClH ₂ (P3m1)	19.8	19.9	49.1	58.1	97.9	195.1
LaGdOFClH ₂ (R3m)	11.8	12.0	20.0	51.1	101.9	191.3
Y ₂ OCl ₂ H ₂ (R3m)	1.5	1.5	8.8	45.1	90.3	158.5
La ₂ OCl ₂ H ₂ (R3m)	1.4	1.4	7.6	34.4	68.8	128.9
Gd ₂ OCl ₂ H ₂ (R3m)	1.0	1.0	7.6	42.2	84.3	149.0

Spatial distributions of the shear modulus and Poisson's ratio were analyzed and visualized by using program tools implemented in ELATE open-source online application [1,2].



Supplementary Figure S1: Spatial distribution of the A) linear compressibility (in TPa^{-1}) and B) Poisson's ratio in $\text{Y}_2\text{OF}_2\text{H}_2$ (R3m). Green and blue colors represent the minimal and maximal values, respectively. Red color denotes the negative values



Supplementary Figure S2: Spatial distribution of the A) linear compressibility (in TPa^{-1}) and B) Poisson's ratio in $\text{Gd}_2\text{OF}_2\text{H}_2$ (R3m). Green and blue colors represent the minimal and maximal values, respectively. Red color denotes the negative values

4. Zone-centered optical vibrational modes calculated in the harmonic approximation

The frequencies were classified by using the program tools described in Ref. [3]

Supplementary Table S29: $\text{Y}_2\text{OF}_2\text{H}_2$ (P3m1). $\Gamma_{\text{optic}} = 6\text{A}_1 + 6\text{E}$

Frequency	Mode symmetry	Vibrational displacements
1317 cm^{-1}	A_1	$\text{H}2 - \text{H}1$
1311 cm^{-1}	E	$\text{H}2 - \text{H}2$
843 cm^{-1}	E	$\text{H}1 - \text{H}2 - \text{O}1$
673 cm^{-1}	A_1	$\text{H}1 - \text{Y}2$
495 cm^{-1}	A_1	$\text{O}1 - \text{F}1 - \text{Y}1$
408 cm^{-1}	E	$\text{F}1 - \text{O}1 - \text{Y}1$
347 cm^{-1}	A_1	$\text{F}1 - \text{F}2 - \text{Y}1 - \text{Y}2$
286 cm^{-1}	E	$\text{F}2 - \text{O}1 - \text{Y}1$
207 cm^{-1}	A_1	$\text{F}1 - \text{F}2 - \text{Y}2$
194 cm^{-1}	A_1	$\text{Y}1 - \text{Y}2 - \text{F}2$
168 cm^{-1}	E	$\text{F}1 - \text{Y}1 - \text{Y}2$
106 cm^{-1}	E	$\text{Y}1 - \text{Y}2$

Supplementary Table S30: $\text{Y}_2\text{OF}_2\text{H}_2$ (R3m). $\Gamma_{\text{optic}} = 6\text{A}_1 + 6\text{E}$

Frequency	Mode symmetry	Vibrational displacements
1434 cm^{-1}	A_1	$\text{H}2 - \text{H}1$
1352 cm^{-1}	E	$\text{H}1 - \text{H}2$
850 cm^{-1}	E	$\text{H}2 - \text{H}1 - \text{O}1$
693 cm^{-1}	A_1	$\text{H}1 - \text{Y}2$
465 cm^{-1}	A_1	$\text{O}1 - \text{Y}1 - \text{F}1$
425 cm^{-1}	E	$\text{O}1 - \text{F}1 - \text{Y}2$
359 cm^{-1}	A_1	$\text{F}1 - \text{Y}1 - \text{Y}2$
251 cm^{-1}	E	$\text{F}1 - \text{F}2 - \text{Y}1$
238 cm^{-1}	A_1	$\text{F}2 - \text{F}1 - \text{Y}2$
164 cm^{-1}	E	$\text{F}1 - \text{Y}2$
161 cm^{-1}	A_1	$\text{F}2 - \text{F}1 - \text{Y}1 - \text{Y}2$
102 cm^{-1}	E	$\text{Y}1 - \text{Y}2 - \text{F}1$

Supplementary Table S31: $\text{La}_2\text{OF}_2\text{H}_2$ (P3m1). $\Gamma_{\text{optic}} = 6\text{A}_1 + 6\text{E}$

Frequency	Mode symmetry	Vibrational displacements
1243 cm^{-1}	A_1	$\text{H}2 - \text{H}1$
1155 cm^{-1}	E	$\text{H}1 - \text{H}2$
723 cm^{-1}	E	$\text{H}2 - \text{H}1$
583 cm^{-1}	A_1	$\text{H}1 - \text{La}2$
429 cm^{-1}	A_1	$\text{O}1 - \text{La}2$
338 cm^{-1}	E	$\text{O}1 - \text{F}1$
283 cm^{-1}	A_1	$\text{O}1 - \text{F}1 - \text{La}1$
242 cm^{-1}	E	$\text{F}2 - \text{O}1 - \text{La}1$
203 cm^{-1}	A_1	$\text{F}2 - \text{F}1 - \text{La}1 - \text{La}2$
151 cm^{-1}	A_1	$\text{La}1 - \text{La}2 - \text{F}1$
145 cm^{-1}	E	$\text{F}1 - \text{La}2$
82 cm^{-1}	E	$\text{La}1 - \text{La}2 - \text{F}2$

Supplementary Table S32: La₂OF₂H₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1319 cm ⁻¹	A ₁	H2 – H1
1183 cm ⁻¹	E	H1 – H2
721 cm ⁻¹	E	H2 – H1 – O1
558 cm ⁻¹	A ₁	H1 – La2
406 cm ⁻¹	A ₁	O1 – La1 – F1
333 cm ⁻¹	E	O1 – F2
291 cm ⁻¹	A ₁	F1 – F2 – La1 – La2
248 cm ⁻¹	E	F2 – O1 – La1
178 cm ⁻¹	A ₁	F1 – F2 – La1
153 cm ⁻¹	E	F1 – La2
133 cm ⁻¹	A ₁	La1 – La2 – F1
82 cm ⁻¹	E	La1 – La2 – F1

Supplementary Table S33: Gd₂OF₂H₂ (P3m1). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1303 cm ⁻¹	A ₁	H2 – H1
1269 cm ⁻¹	E	H1 – H2
788 cm ⁻¹	E	H2 – H1
652 cm ⁻¹	A ₁	H1 – Gd2
477 cm ⁻¹	A ₁	O1 – F1 – Gd1
389 cm ⁻¹	E	O1 – F2
320 cm ⁻¹	A ₁	F1 – F2 – Gd1 – Gd2
269 cm ⁻¹	E	F2 – O1 – Gd1
200 cm ⁻¹	A ₁	F2 – F1 – Gd2
159 cm ⁻¹	E	F1 – Gd1
153 cm ⁻¹	A ₁	F2 – Gd1 – Gd2
81 cm ⁻¹	E	Gd1 – Gd2 – F2

Supplementary Table S34: Gd₂OF₂H₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1418 cm ⁻¹	A ₁	H2 – H1
1298 cm ⁻¹	E	H1 – H2
799 cm ⁻¹	E	H2 – H1 – O1
661 cm ⁻¹	A ₁	H1 – Gd2
448 cm ⁻¹	A ₁	O1 – Gd1 – F2
395 cm ⁻¹	E	O1 – F1
333 cm ⁻¹	A ₁	F1 – F2 – Gd1 – Gd2
249 cm ⁻¹	E	F2 – O1 – Gd1
206 cm ⁻¹	A ₁	F1 – F2 – Gd2
163 cm ⁻¹	E	F1 – Gd2
128 cm ⁻¹	A ₁	Gd1 – Gd2 – F1
80 cm ⁻¹	E	Gd1 – Gd2

Supplementary Table S35: Y_2OFClH_2 (P3m1). $\Gamma_{\text{optic}} = 6\text{A}_1 + 6\text{E}$

Frequency	Mode symmetry	Vibrational displacements
1272 cm^{-1}	E	H1 – H2
1233 cm^{-1}	A_1	H2 – H1
847 cm^{-1}	E	H1 – H2 – O1
681 cm^{-1}	A_1	H1 – Y2
488 cm^{-1}	A_1	F1 – O1 – Y1
403 cm^{-1}	E	O1 – F1 – Y1
315 cm^{-1}	A_1	F1 – Cl1 – Y1 – Y2
246 cm^{-1}	E	F1 – O1 – Y1
221 cm^{-1}	A_1	F1 – Cl1 – Y2
156 cm^{-1}	A_1	Cl1 – Y1 – Y2
152 cm^{-1}	E	Cl1 – Y2
96 cm^{-1}	E	Y1 – Y2 – Cl1

Supplementary Table S36: Y_2OFClH_2 (R3m). $\Gamma_{\text{optic}} = 6\text{A}_1 + 6\text{E}$

Frequency	Mode symmetry	Vibrational displacements
1285 cm^{-1}	E	H1 – H2
1274 cm^{-1}	A_1	H2 – H1
856 cm^{-1}	E	H2 – H1 – O1
686 cm^{-1}	A_1	H1 – Y2
479 cm^{-1}	A_1	O1 – Y1 – F1
411 cm^{-1}	E	O1 – F1 – Y1
314 cm^{-1}	A_1	F1 – Cl1 – Y1 – Y2
240 cm^{-1}	A_1	Cl1 – F1 – Y2
231 cm^{-1}	E	F1 – Y2 – O1
154 cm^{-1}	A_1	Cl1 – F1 – Y1 – Y2
154 cm^{-1}	E	Cl1 – Y2
90 cm^{-1}	E	Y1 – Y2 – Cl1

Supplementary Table S37: $\text{La}_2\text{OFClH}_2$ (P3m1). $\Gamma_{\text{optic}} = 6\text{A}_1 + 6\text{E}$

Frequency	Mode symmetry	Vibrational displacements
1121 cm^{-1}	E	H1 – H2
1119 cm^{-1}	A_1	H2 – H1
762 cm^{-1}	E	H2 – H1
562 cm^{-1}	A_1	H1 – La2
438 cm^{-1}	A_1	O1 – F1 – La1
332 cm^{-1}	E	O1 – F1
256 cm^{-1}	A_1	F1 – Cl1 – La1 – La2
222 cm^{-1}	E	F1 – O1 – La1
180 cm^{-1}	A_1	Cl1 – F1 – La2
140 cm^{-1}	A_1	La1 – La2 – Cl1
128 cm^{-1}	E	Cl1 – La2
75 cm^{-1}	E	La1 – La2 – Cl1

Supplementary Table S38: La₂OFClH₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1144 cm ⁻¹	A ₁	H2 – H1
1127 cm ⁻¹	E	H1 – H2
768 cm ⁻¹	E	H2 – H1 – O1
565 cm ⁻¹	A ₁	H1 – La2
435 cm ⁻¹	A ₁	O1 – La1 – F1
333 cm ⁻¹	E	O1 – F1
258 cm ⁻¹	A ₁	F1 – Cl1 – La1 – La2
225 cm ⁻¹	E	F1 – O1 – La1
176 cm ⁻¹	A ₁	F1 – Cl1 – La2
130 cm ⁻¹	A ₁	La1 – La2 – Cl1
128 cm ⁻¹	E	Cl1 – La2
73 cm ⁻¹	E	La1 – La2 – Cl1 – F1

Supplementary Table S39: Gd₂OFClH₂ (P3m1). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1228 cm ⁻¹	E	H1 – H2
1208 cm ⁻¹	A ₁	H2 – H1
789 cm ⁻¹	E	H2 – H1
656 cm ⁻¹	A ₁	H1 – Gd2
474 cm ⁻¹	A ₁	O1 – F1 – Gd1
382 cm ⁻¹	E	O1 – F1
289 cm ⁻¹	A ₁	F1 – Cl1 – Gd1 – Gd2
235 cm ⁻¹	E	F1 – O1 – Gd1
195 cm ⁻¹	A ₁	Cl1 – F1 – Gd2
139 cm ⁻¹	E	Cl1 – Gd2
131 cm ⁻¹	A ₁	Cl1 – Gd1 – Gd2
77 cm ⁻¹	E	Gd1 – Gd2 – Cl1 – F1

Supplementary Table S40: Gd₂OFClH₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1254 cm ⁻¹	A ₁	H2 – H1
1242 cm ⁻¹	E	H1 – H2
808 cm ⁻¹	E	H2 – H1 – O1
661 cm ⁻¹	A ₁	H1 – Gd2
464 cm ⁻¹	A ₁	O1 – Gd1 – F1
389 cm ⁻¹	E	O1 – F1
286 cm ⁻¹	A ₁	F1 – Cl1 – Gd1 – Gd2
221 cm ⁻¹	E	F1 – O1 – Gd1
215 cm ⁻¹	A ₁	Cl1 – F1 – Gd2
138 cm ⁻¹	E	Cl1 – Gd2
126 cm ⁻¹	A ₁	Gd1 – Gd2 – Cl1 – F1
72 cm ⁻¹	E	Gd1 – Gd2 – Cl1 – F1

Supplementary Table S41: LaGdOFClH₂ (P3m1). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1171 cm ⁻¹	A ₁	H2 – H1
1111 cm ⁻¹	E	H1 – H2
656 cm ⁻¹	A ₁	H1 – La1
644 cm ⁻¹	E	H2 – H1
468 cm ⁻¹	A ₁	O1 – F1 – Gd1
350 cm ⁻¹	E	O1 – F1
275 cm ⁻¹	A ₁	F1 – Cl1 – La1 – Gd1
253 cm ⁻¹	E	F1 – O1 – La1
184 cm ⁻¹	A ₁	Cl1 – F1 – Gd1
139 cm ⁻¹	A ₁	Cl1 – La1 – Gd1
121 cm ⁻¹	E	Cl1 – Gd1
75 cm ⁻¹	E	Gd1 – La1 – Cl1 – F1

Supplementary Table S42: LaGdOFClH₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1199 cm ⁻¹	A ₁	H2 – H1
1117 cm ⁻¹	E	H1 – H2
661 cm ⁻¹	E	H2 – H1 – O1
660 cm ⁻¹	A ₁	H1 – Gd1
462 cm ⁻¹	A ₁	O1 – La1 – F1
352 cm ⁻¹	E	O1 – F1
275 cm ⁻¹	A ₁	F1 – Cl1 – La1 – Gd1
249 cm ⁻¹	E	F1 – O1 – La1
188 cm ⁻¹	A ₁	Cl1 – F1 – Gd1
134 cm ⁻¹	A ₁	Gd1 – La1 – Cl1 – F1
119 cm ⁻¹	E	Cl1 – Gd1
72 cm ⁻¹	E	La1 – Gd1 – Cl1 – F1

Supplementary Table S43: Y₂OCl₂H₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

Frequency	Mode symmetry	Vibrational displacements
1325 cm ⁻¹	A ₁	H2 – H1
1226 cm ⁻¹	E	H1 – H2
787 cm ⁻¹	E	H2 – H1 – O1
688 cm ⁻¹	A ₁	H1 – Y2
456 cm ⁻¹	A ₁	O1 – Y1
383 cm ⁻¹	E	O1 – Y1
280 cm ⁻¹	A ₁	Cl1 – Cl2 – Y1 – Y2
250 cm ⁻¹	A ₁	Cl1 – Cl2 – Y2 – Y1
166 cm ⁻¹	E	Cl2 – Y2 – O1
154 cm ⁻¹	E	Cl1 – Y2
140 cm ⁻¹	A ₁	Cl2 – Cl1 – Y2 – Y1
84 cm ⁻¹	E	Y1 – Y2 – Cl1

Supplementary Table S44: La₂OCl₂H₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

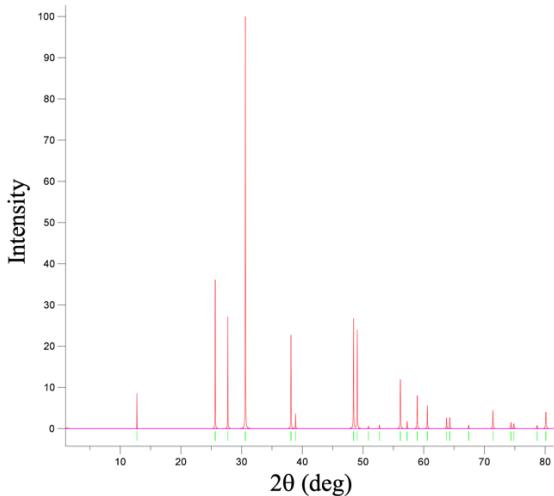
Frequency	Mode symmetry	Vibrational displacements
1228 cm ⁻¹	A ₁	H2 – H1
1073 cm ⁻¹	E	H1 – H2
684 cm ⁻¹	E	H2 – H1 – O1
584 cm ⁻¹	A ₁	H1 – La2
395 cm ⁻¹	A ₁	O1 – La1
305 cm ⁻¹	E	O1 – Cl2
232 cm ⁻¹	A ₁	Cl1 – Cl2 – La1 – La2
211 cm ⁻¹	A ₁	Cl2 – Cl1 – La2
149 cm ⁻¹	E	Cl2 – O1 – La1
134 cm ⁻¹	E	Cl1 – La2
113 cm ⁻¹	A ₁	La1 – La2 – Cl1
66 cm ⁻¹	E	La1 – La2 – Cl1 – Cl2

Supplementary Table S45: Gd₂OCl₂H₂ (R3m). $\Gamma_{\text{optic}} = 6A_1 + 6E$

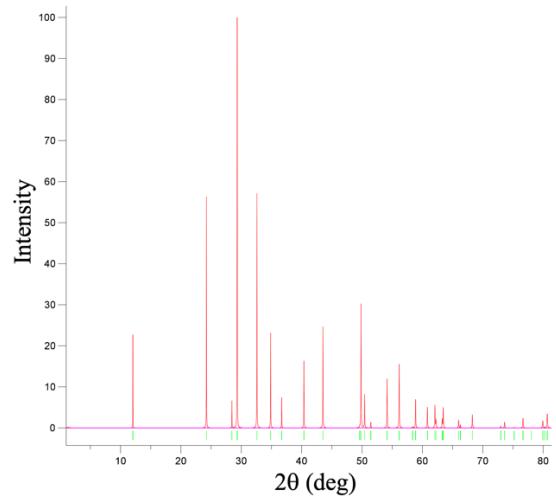
Frequency	Mode symmetry	Vibrational displacements
1304 cm ⁻¹	A ₁	H2 – H1
1189 cm ⁻¹	E	H1 – H2
738 cm ⁻¹	E	H2 – H1 – O1
661 cm ⁻¹	A ₁	H1 – Gd2
436 cm ⁻¹	A ₁	O1 – Gd1
361 cm ⁻¹	E	O1 – Gd2
249 cm ⁻¹	A ₁	Cl1 – Cl2 – Gd1 – Gd2
230 cm ⁻¹	A ₁	Cl1 – Cl2 – Gd2
152 cm ⁻¹	E	Cl2 – Gd1
140 cm ⁻¹	E	Cl1 – Gd2
116 cm ⁻¹	A ₁	Gd1 – Gd2 – Cl1 – Cl2
68 cm ⁻¹	E	Gd1 – Gd2 – Cl1 – Cl2

5. Simulation of X-ray diffraction patterns

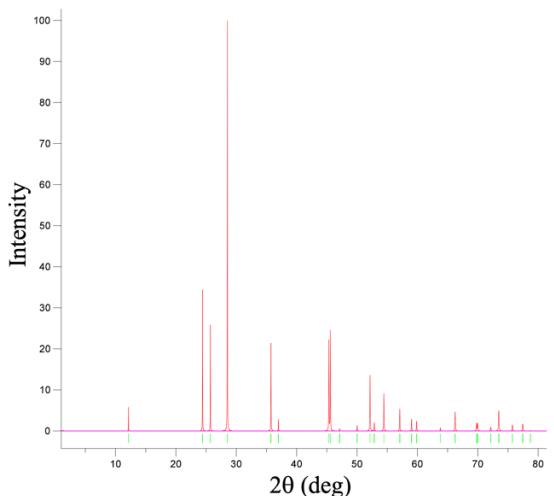
The Cu K α monochromatic beam with wavelength $\lambda=1.5406 \text{ \AA}$ was set as a parameter of calculation to determine the theoretical X-ray diffraction patterns. The illustrations have been made by using the VESTA program [4].



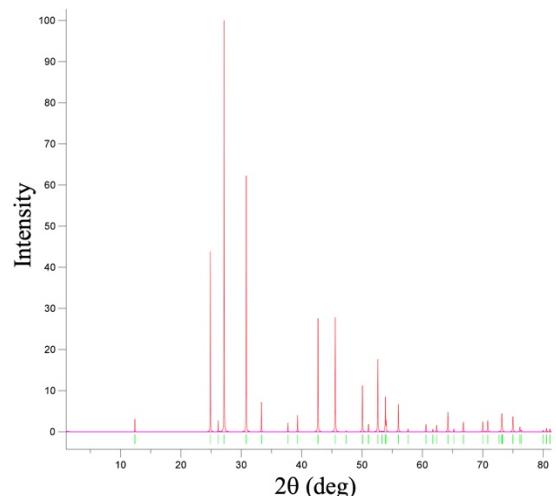
Supplementary Figure S3: $\text{Y}_2\text{OF}_2\text{H}_2$ ($\text{P}3\text{m}1$)



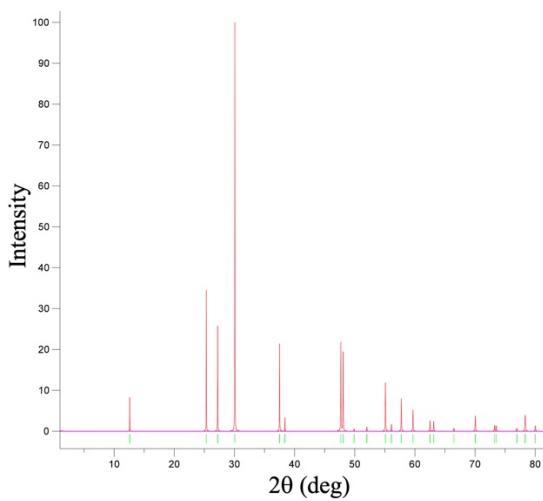
Supplementary Figure S4: $\text{Y}_2\text{OF}_2\text{H}_2$ ($\text{R}3\text{m}$)



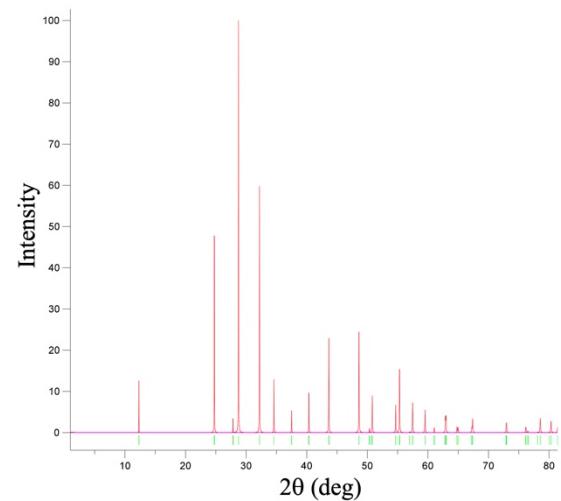
Supplementary Figure S5: $\text{La}_2\text{OF}_2\text{H}_2$ ($\text{P}3\text{m}1$)



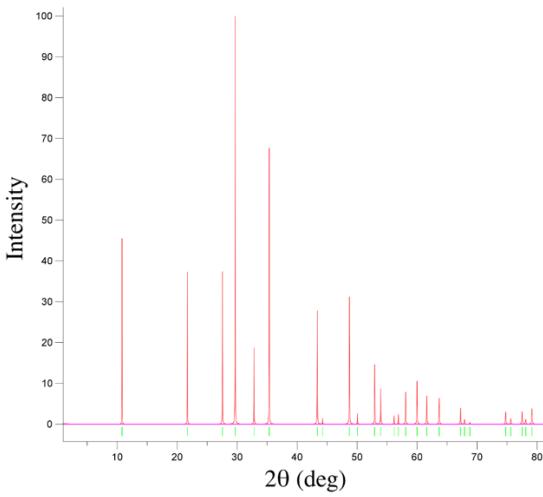
Supplementary Figure S6: $\text{La}_2\text{OF}_2\text{H}_2$ ($\text{R}3\text{m}$)



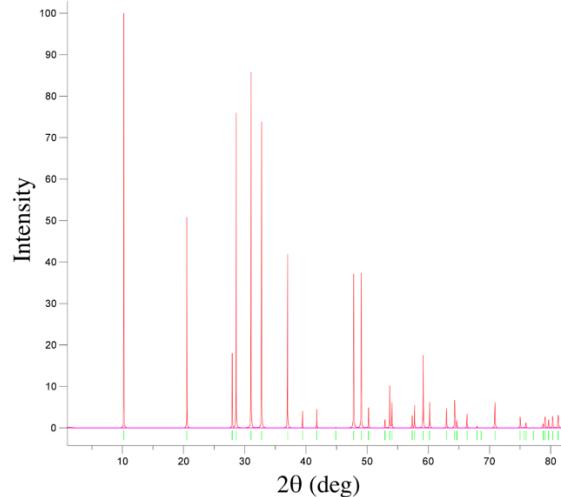
Supplementary Figure S7: $\text{Gd}_2\text{OF}_2\text{H}_2$ (P3m1)



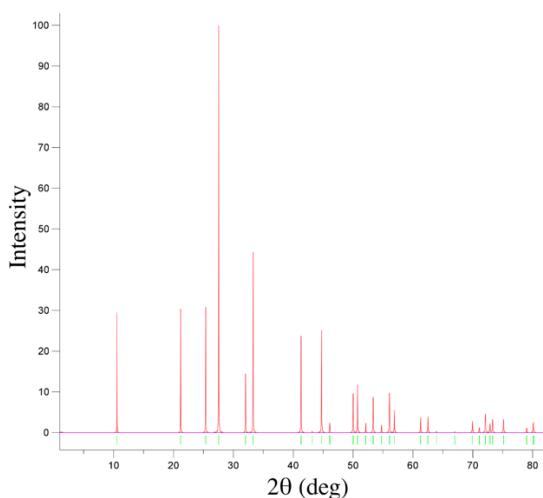
Supplementary Figure S8: $\text{Gd}_2\text{OF}_2\text{H}_2$ (R3m)



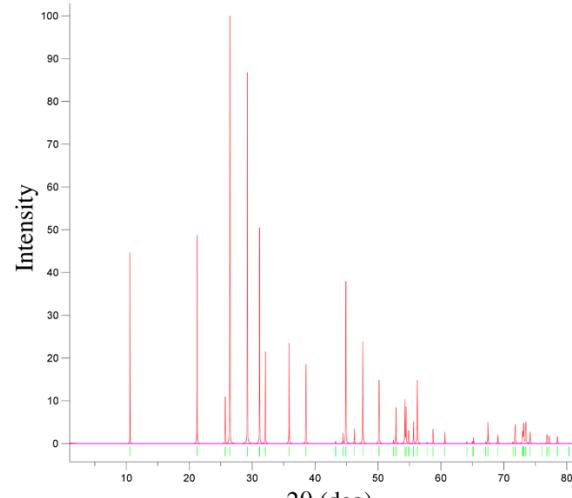
Supplementary Figure S9: Y_2OFClH_2 (P3m1)



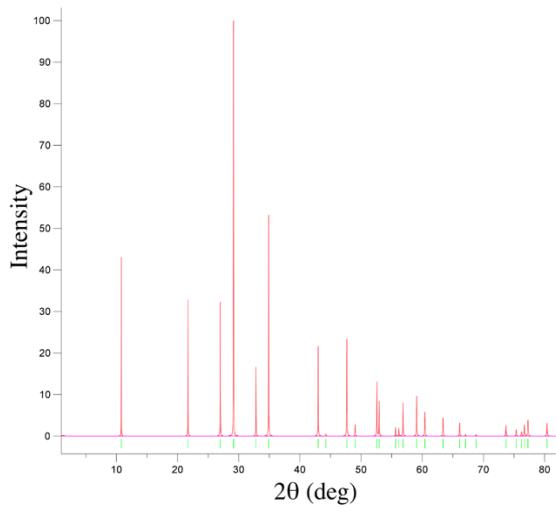
Supplementary Figure S10: Y_2OFClH_2 (R3m)



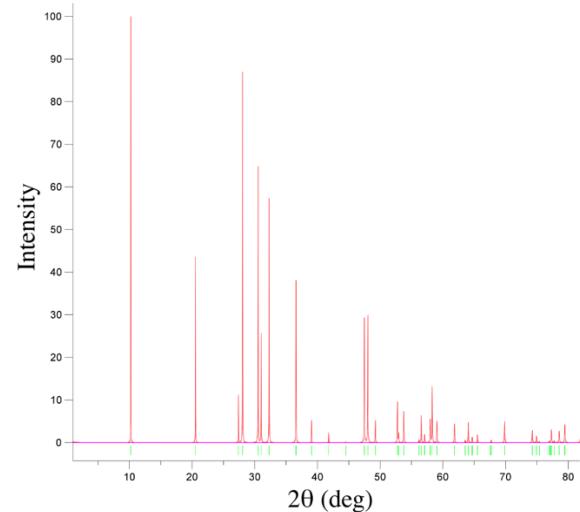
Supplementary Figure S11: $\text{La}_2\text{OFClH}_2$ (P3m1)



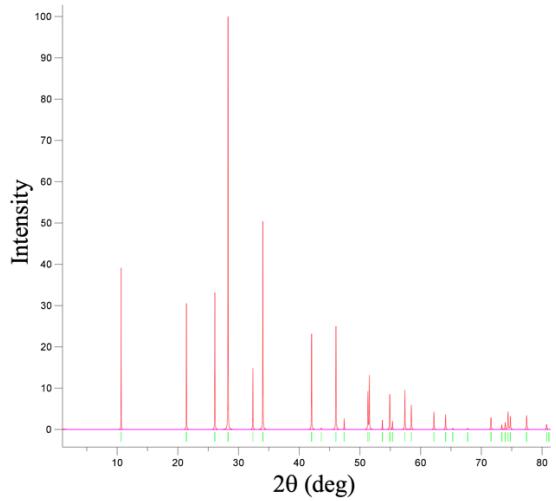
Supplementary Figure S12: $\text{La}_2\text{OFClH}_2$ (R3m)



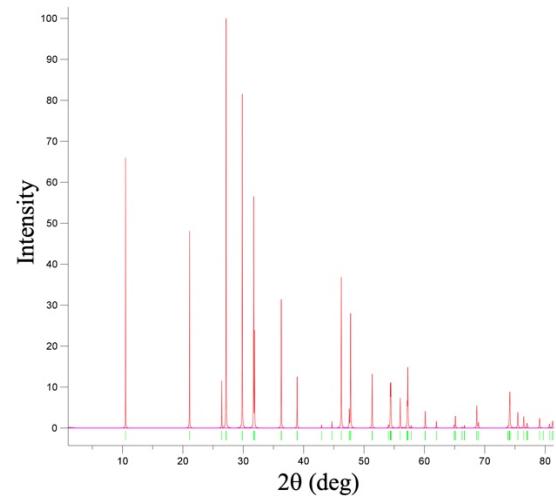
Supplementary Figure S13: Gd_2OClH_2 (P3m1)



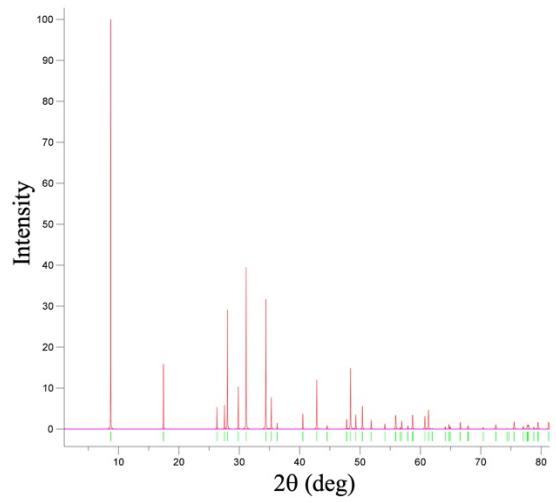
Supplementary Figure S14: Gd_2OClH_2 (R3m)



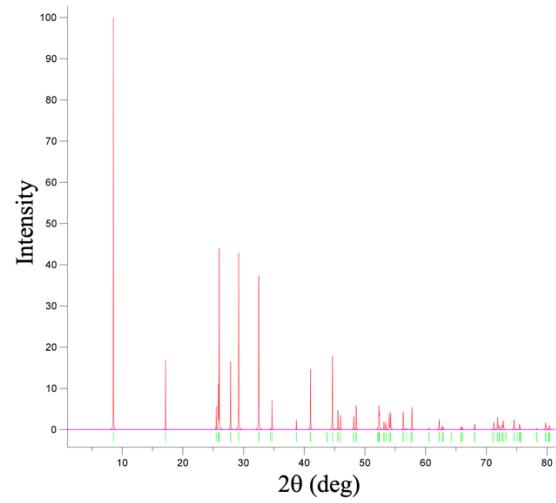
Supplementary Figure S15: LaGdOClH_2 (P3m1)



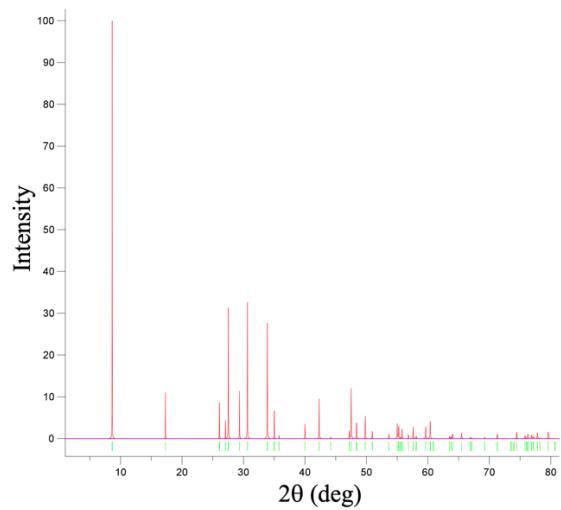
Supplementary Figure S16: LaGdOClH_2 (R3m)



Supplementary Figure S17: $\text{Y}_2\text{OCl}_2\text{H}_2$ (R3m)



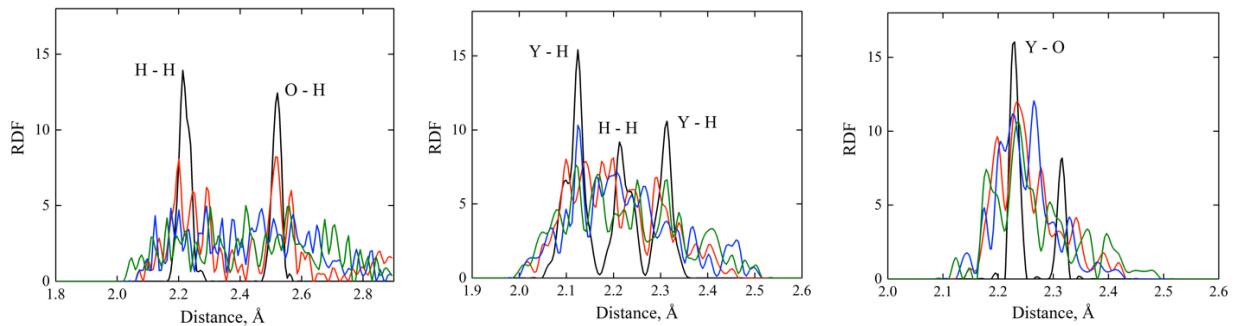
Supplementary Figure S18: $\text{La}_2\text{OCl}_2\text{H}_2$ (R3m)



Supplementary Figure S19: $\text{Gd}_2\text{OCl}_2\text{H}_2$ (R3m)

6. Radial distribution function evaluated for the shortest interatomic distances

Calculations have been performed for four temperatures: T= 12 K (results are shown in black), 100 K (shown in red), 200 K (shown in blue) and 300 K (shown in green).



Supplementary Figure S20: Radial distribution functions for the shortest interatomic distances between atoms in $\text{Y}_2\text{OF}_2\text{H}_2$ (R3m)

7. Piezoelectric properties of $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ ($\text{Ln} = \text{Y}, \text{La}, \text{Gd}$)

Supplementary Table S46: Independent components of electronic and ionic contributions to direct piezoelectric tensor (e_{ij}) and dielectric tensor (ε_{ij}) in $\text{Ln}_2\text{OF}_2\text{H}_2$

		e_{15}	e_{22}	e_{31}	e_{33}	ε_{11}	ε_{33}
$\text{Y}_2\text{OF}_2\text{H}_2$ (P3m1)	Elect.	-0.03	-0.03	0.08	-0.28	4.67	3.38
	Ion.	0.92	0.89	1.15	-0.06	13.21	10.09
$\text{Y}_2\text{OF}_2\text{H}_2$ (R3m)	Elect.	-0.04	0.10	0.002	0.04	4.47	3.13
	Ion.	-0.06	-0.82	0.22	-0.08	11.86	5.04
$\text{La}_2\text{OF}_2\text{H}_2$ (P3m1)	Elect.	-0.01	0.03	0.11	-0.34	4.48	3.44
	Ion.	0.60	-0.41	1.02	0.44	12.86	8.71
$\text{La}_2\text{OF}_2\text{H}_2$ (R3m)	Elect.	-0.02	0.07	-0.02	0.08	4.56	3.61
	Ion.	-0.09	-0.08	0.38	-0.62	13.91	10.96

Supplementary Table S47: Independent components of electronic and ionic contributions to direct piezoelectric tensor (e_{ij}) and dielectric tensor (ε_{ij}) in $\text{Ln}_2\text{OFClH}_2$

		e_{15}	e_{22}	e_{31}	e_{33}	ε_{11}	ε_{33}
Y_2OFClH_2 (P3m1)	Elect.	0.01	0.002	0.07	-0.34	4.74	3.68
	Ion.	-0.03	0.25	0.23	-1.06	9.35	5.05
Y_2OFClH_2 (R3m)	Elect.	-0.01	0.03	0.02	-0.22	4.46	3.78
	Ion.	-0.14	-0.36	0.09	-0.43	9.26	2.44
$\text{La}_2\text{OFClH}_2$ (P3m1)	Elect.	0.04	0.01	0.10	-0.35	4.50	3.75
	Ion.	-0.05	-0.05	0.37	-0.54	10.13	5.74
$\text{La}_2\text{OFClH}_2$ (R3m)	Elect.	0.01	0.04	0.02	-0.21	4.55	3.65
	Ion.	-0.23	0.14	0.26	-0.89	10.69	5.39
$\text{Gd}_2\text{OFClH}_2$ (P3m1)	Elect.	0.02	-0.01	0.08	-0.37	4.70	3.80
	Ion.	-0.03	0.16	0.30	-0.99	10.10	5.49
$\text{Gd}_2\text{OFClH}_2$ (R3m)	Elect.	0.01	0.05	0.01	-0.20	4.62	3.34
	Ion.	-0.17	-0.24	0.12	-0.48	9.02	2.69
LaGdOFClH_2 (P3m1)	Elect.	0.06	0.003	0.05	-0.36	4.67	3.80
	Ion.	-0.05	0.20	0.36	-0.75	12.71	5.65
LaGdOFClH_2 (R3m)	Elect.	0.04	0.04	-0.02	-0.22	4.62	3.95
	Ion.	-0.22	-0.18	0.23	-0.82	12.51	4.31

Supplementary Table S48: Independent components of electronic and ionic contributions to direct piezoelectric tensor (e_{ij}) and dielectric tensor (ε_{ij}) in $\text{Ln}_2\text{OCl}_2\text{H}_2$

		e_{15}	e_{22}	e_{31}	e_{33}	ε_{11}	ε_{33}
$\text{Y}_2\text{OCl}_2\text{H}_2$ (R3m)	Elect.	-0.02	0.004	0.02	-0.06	4.40	3.69
	Ion.	-0.01	-0.27	0.02	-0.04	8.05	0.96
$\text{La}_2\text{OCl}_2\text{H}_2$ (R3m)	Elect.	0.01	0.01	0.03	-0.04	4.25	3.40
	Ion.	-0.06	-0.10	0.06	-0.03	8.39	1.14
$\text{Gd}_2\text{OCl}_2\text{H}_2$ (R3m)	Elect.	-0.002	0.02	0.03	-0.04	4.42	3.65
	Ion.	-0.02	-0.22	0.02	-0.04	7.89	0.95

Supplementary Table S49: Comparison of the electromechanical properties of $\text{Ln}_2\text{OF}_{2-x}\text{Cl}_x\text{H}_2$ expressed in terms of piezoelectric voltage constant g_{ij} ($10^{-3} \times [\text{Vm/N}]$), piezoelectric coupling factors k_{ij} and k_t , and piezoelectric transduction coefficients $d_{ij} \cdot g_{ij}$ ($10^{-15} \times [\text{m}^2/\text{N}]$)

Chem. Formula	Struc-ture	g_{33}	k_{33}	k_t	$d_{33} \cdot g_{33}$	g_{31}	k_{31}	$d_{31} \cdot g_{31}$	g_{15}	k_{15}	$d_{15} \cdot g_{15}$
$\text{Y}_2\text{OF}_2\text{H}_2$	P3m1	-64.8	23.9%	8.4%	500	58.3	23.2%	405	271	53.1%	11576
	R3m	-64.2	7.3%	3.1%	298	18.0	5.9%	23	9.3	1.9%	12
$\text{La}_2\text{OF}_2\text{H}_2$	P3m1	-22.7	7.7%	2.8%	55	57.8	21.2%	360	210.7	37.2%	6811
	R3m	-91.3	24.6%	17.6%	1074	27.1	10.7%	95	-22.1	4.9%	80
Y_2OFClH_2	P3m1	-600.0	94.4%	61.7%	27841	106.8	35.0%	882	-14.8	2.6%	27
	R3m	-723.3	69.3%	55.5%	26205	45.7	12.2%	105	-177	14.6%	3663
$\text{La}_2\text{OFClH}_2$	P3m1	-179.8	43.6%	33.3%	2719	60.4	18.8%	307	-0.9	0.1%	0.1
	R3m	-637.7	86.0%	58.9%	33367	93.6	28.2%	719	-143.6	19.4%	2809
$\text{Gd}_2\text{OFClH}_2$	P3m1	-427.5	79.5%	54.6%	15028	92.1	30.4%	697	-6.2	1.1%	5
	R3m	-770.9	73.7%	56.9%	31715	60.2	16.1%	193	-221.1	18.8%	5897
LaGdOFClH_2	P3m1	-238.0	54.4%	41.4%	4749	64.0	20.3%	342	-0.2	0.1%	0.1
	R3m	-724.3	88.9%	60.9%	38354	94.4	27.6%	651	-91.0	12.3%	1256
$\text{Y}_2\text{OCl}_2\text{H}_2$	R3m	-290.4	16.2%	15.7%	2999	10.4	2.1%	3.8	303.7	11.7%	9412
$\text{La}_2\text{OCl}_2\text{H}_2$	R3m	-249.6	13.8%	12.8%	2504	22.9	4.3%	21	-273.7	10.9%	8386
$\text{Gd}_2\text{OCl}_2\text{H}_2$	R3m	-260.6	14.5%	14.0%	2765	12.0	2.5%	6	-196.1	6.4%	4188

Supplementary Table S50: Comparison of the relaxed lattice geometry within the PBE+U approximation and the relevant experimental data for rare-earth oxides, fluorides, chlorides and hydrides

	La ₂ O ₃ , P-3m1		LaH ₃ , Fm-3m		LaF ₃ , P-3c1		LaCl ₃ , P63/m	
	PBE+U	exp.[5]	PBE+U	exp.[6]	PBE+U	exp.[7]	PBE+U	exp.[8]
a, Å	3.96	3.96	5.59	5.62	7.15	7.19	7.45	7.48
b, Å	3.96	3.96	5.59	5.62	7.15	7.19	7.45	7.48
c, Å	6.21	6.14	5.59	5.62	7.33	7.35	4.33	4.37
V, Å ³	84.17	83.24	174.7	177.5	324.3	328.6	208.4	211.2
	Gd ₂ O ₃ , Ia-3		GdH ₃ , P-3c1		GdF ₃ , Pnma		GdCl ₃ , P63/m	
	PBE+U	exp.[9]	PBE+U	exp.[10]	PBE+U	exp.[11]	PBE+U	exp.[8]
a, Å	10.85	10.79	6.44	6.47	6.55	6.57	7.32	7.37
b, Å	10.85	10.79	6.44	6.47	7.05	6.98	7.32	7.37
c, Å	10.85	10.79	6.70	6.72	4.48	4.39	4.05	4.11
V, Å ³	1276.7	1256.2	241.0	243.2	207.6	201.3	188.3	193.0

References:

1. R. Gaillac, P. Pullumbi, F.X. Coudert. *J. Phys.: Condens. Matter.* **2016**, *28*, 27.
2. R. Gaillac, F.X. Coudert, <http://progs.coudert.name/elite>
3. Kroumova, E.; Aroyo, M. I.; Perez-Mato, J. M.; Kirov, A.; Capillas, C.; Ivantchev, S.; Wondratschek, H. Bilbao Crystallographic Server I: Databases and crystallographic computing programs. *Phase Transitions* **2003**, *76*, 155–170.
4. Momma, K.; Izumi, F. VESTA 3 for three-dimensional visualization of crystal, volumetric and morphology data. *J. Appl. Crystallogr.* **2011**, *44*, 1272-1276
5. Cao Wang, Ming-qiu Tan, Chun-mu Feng, Zhi-feng Ma, Shuai Jiang, Zhu-an Xu, Guanghan Cao, Kazuyuki Matsubayashi, and Yoshiya Uwatoko. La₂Co₂Se₂O₃: A Quasi-Two-Dimensional Mott Insulator with Unusual Cobalt Spin State and Possible Orbital Ordering, *J. Am. Chem. Soc.*, **2010**, *132* (20), 7069-7073
6. H. Meng, M.A. Kuzovnikov, M. Tkacz, Phase stability of some rare earth trihydrides under high pressure, *Int. J. Hydrogen Energy*, **2017**, *42*, 29344-29349
7. Cheetham, A. K., Fender, B. E. F., Fuess, H. & Wright, A. F. A powder neutron diffraction study of lanthanum and cerium trifluorides, *Acta Cryst.*, **1976**, *B32*, 94-97
8. B. Morosin. Crystal Structures of Anhydrous Rare-Earth Chlorides, *J. Chem. Phys.*, **1968**, *49*, 3007-3012
9. Bartos, A., Lieb, K. P., Uhrmacher, M. & Wiarda, D. Refinement of atomic positions in bixbyite oxides using perturbed angular correlation spectroscopy, *Acta Cryst.* **1993**, *B49*, 165-169.
10. M. Ellner, H. Reule, E.J. Mittemeijer, The structure of the trihydride GdH₃, *J. Alloys Compd.*, **2000**, *309*, 1–2, 127-131
11. I.M. Ranieri, S.L. Baldochi, D. Klimm, The phase diagram GdF₃–LuF₃, *J. Solid State Chem.*, **2008**, *181*, 5, 1070-1074