

Electronic Supplementary Material for Nanoscale Advances.

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Supplemental Information

Three Dimensional Ti₃C₂ MXene Nanoribbons Frameworks with Uniform Potassiophilic Sites for Dendrite-Free Potassium Metal Anodes

Haodong Shi,^{abc} Yanfeng Dong,^d Shuanghao Zheng,^{ac} Cong Dong,^{abc} Zhong-Shuai Wu^{*ac}

Affiliations:

^aState Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian 116023, China

^bUniversity of Chinese Academy of Sciences, 19 A Yuquan Rd, Shijingshan District, Beijing 100049, China

^cDalian National Laboratory for Clean Energy, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian 116023, China

^dDepartment of Chemistry, College of Sciences, Northeastern University, 3-11 Wenhua Road, Shenyang 110819, China

*Corresponding author: wuzs@dicp.ac.cn

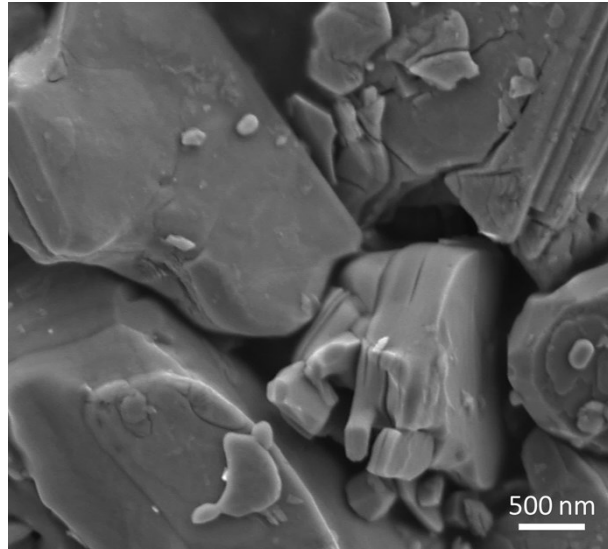


Fig. S1 SEM image of Ti₃AlC₂ MAX.

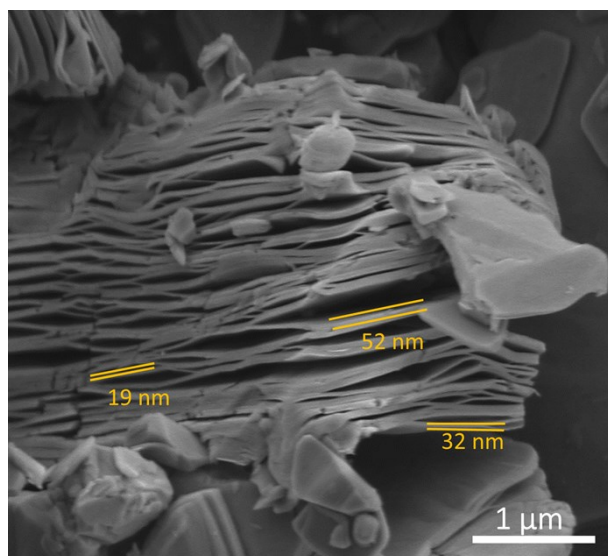


Fig. S2 SEM image of m-Ti₃C₂ MXene nanosheets.

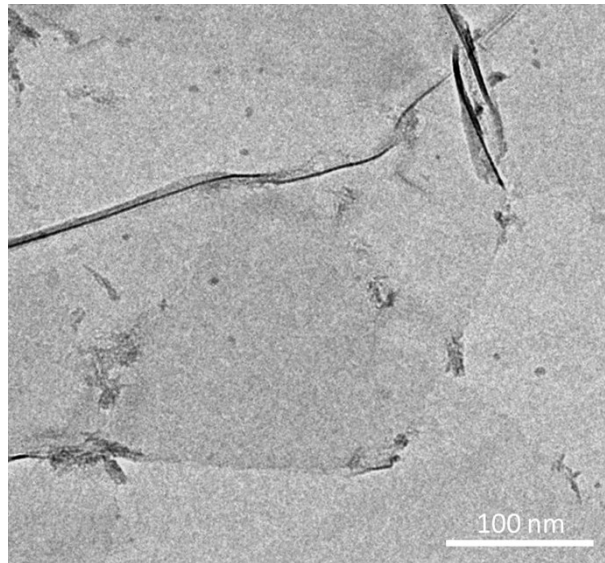


Fig. S3 TEM image of d-Ti₃C₂ MXene nanosheets.

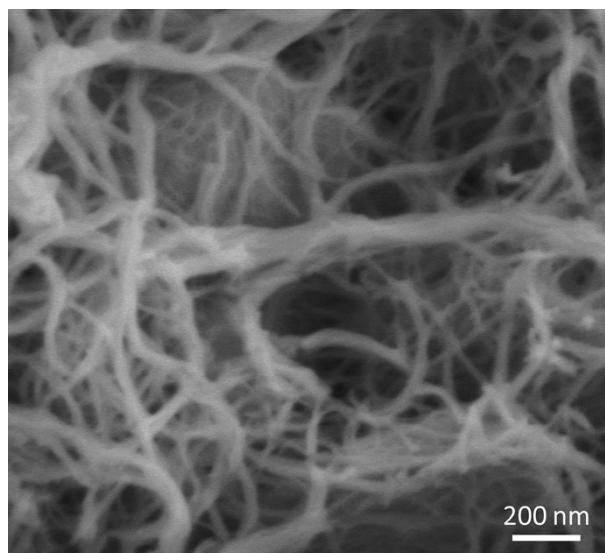


Fig. S4 SEM image of a-Ti₃C₂ MXene nanoribbon frameworks.

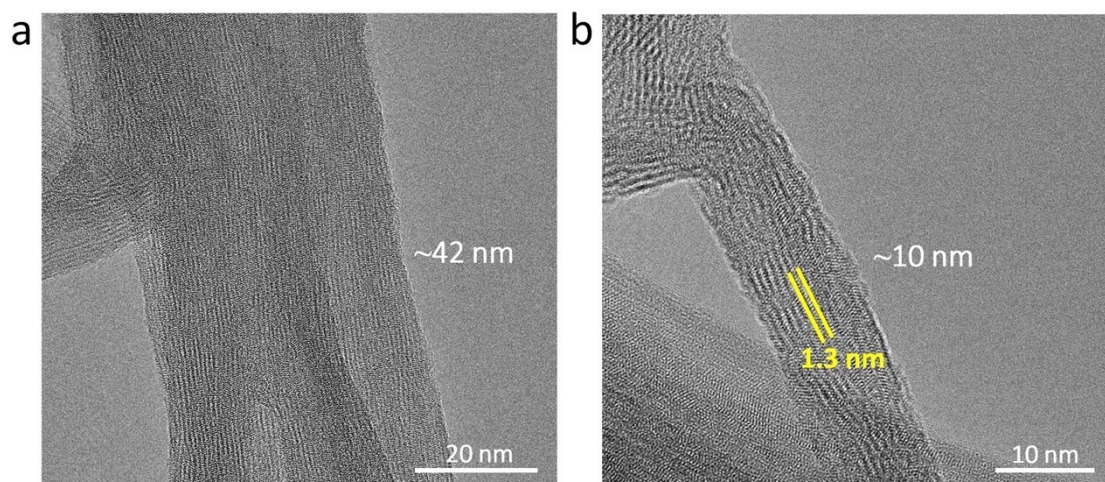


Fig. S5 (a,b) HRTEM images of a-Ti₃C₂ MXene nanoribbons.

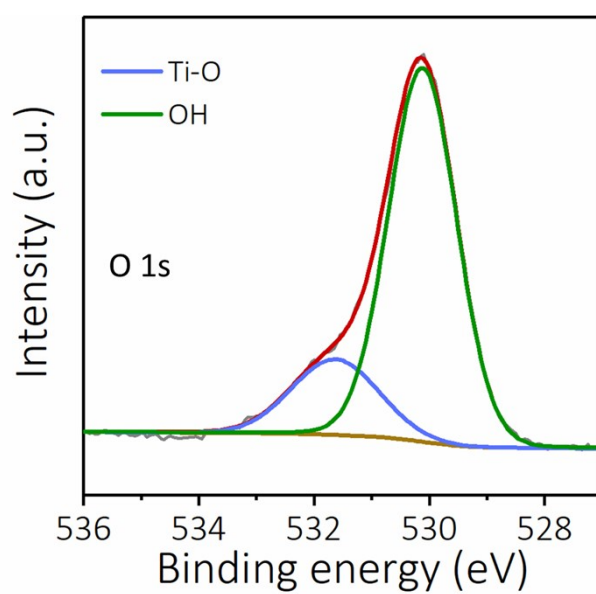


Fig. S6 High resolution O1s XPS spectrum of a-Ti₃C₂ nanoribbons.

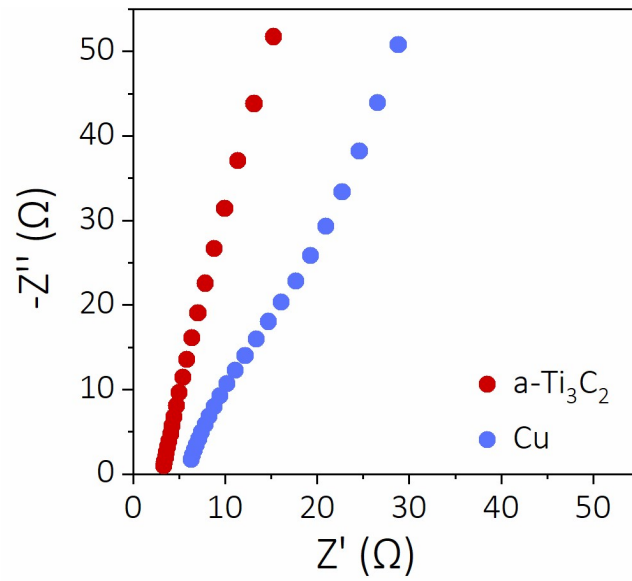


Fig. S7 Nyquist plots of the a-Ti₃C₂ MXene/K and bare Cu/K asymmetric batteries.

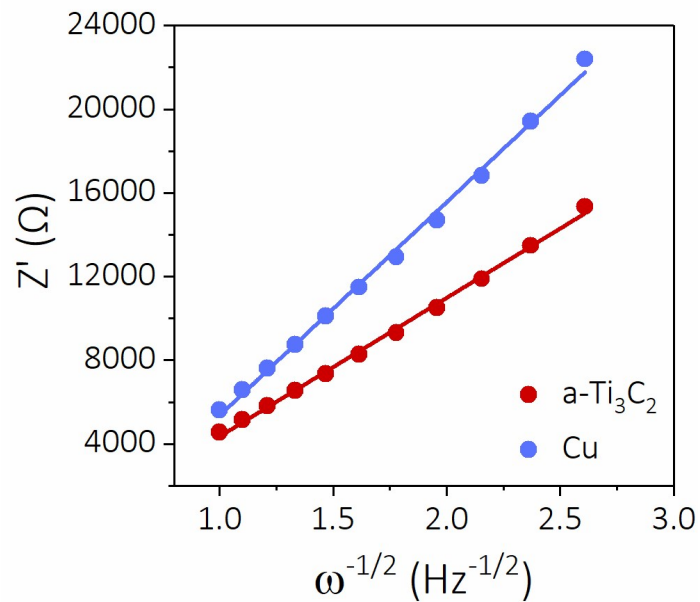


Fig. S8 The relationship of between Z' and square root of the frequency ($\omega^{-1/2}$) in the low-frequency region for the a-Ti₃C₂ MXene/K and bare Cu/K asymmetric batteries.

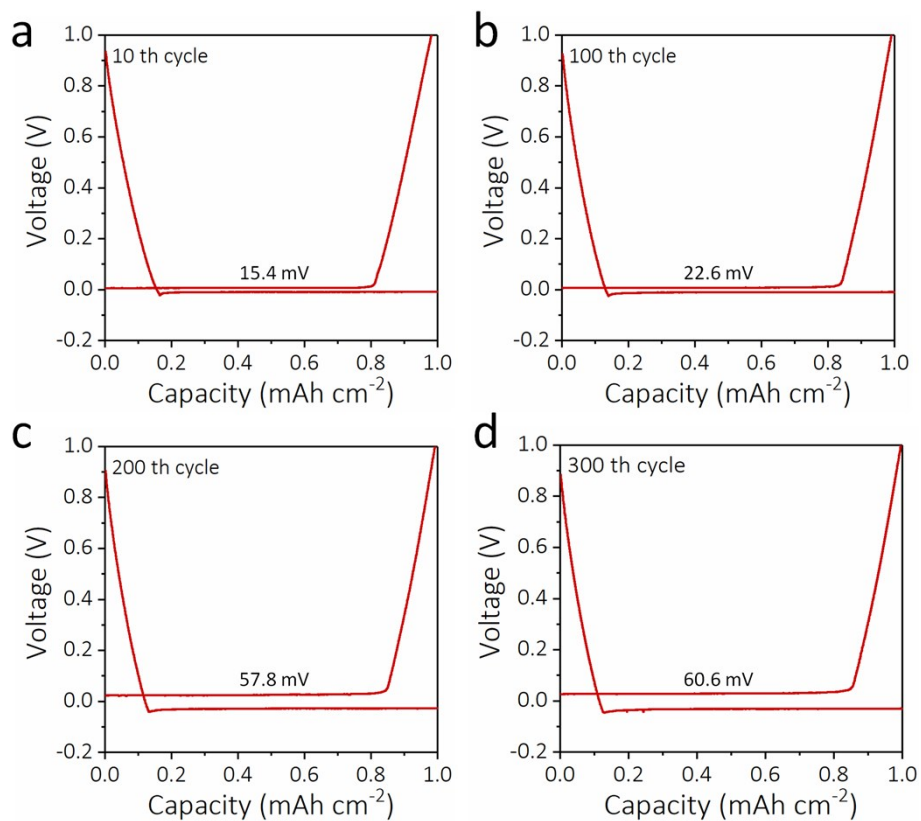


Fig. S9 Voltage profiles of a-Ti₃C₂ MXene electrode tested at different cycles. Plant 1 mAh cm⁻² of K at a current density 1 mA cm⁻² at (a) 10th cycle, (b) 100th cycle, (c) 200th cycle, and (d) 300th cycle.

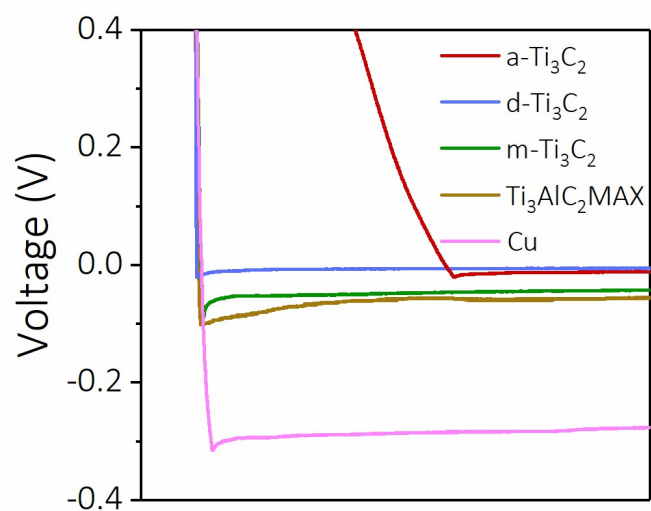


Fig. S10 Nucleation section on the voltage profiles of a-Ti₃C₂ MXene, d-Ti₃C₂ MXene, m-Ti₃C₂ MXene, Ti₃AlC₂ MAX and bare Cu electrodes with K deposition capacity of 1 mAh cm⁻² at 1 mA cm⁻².

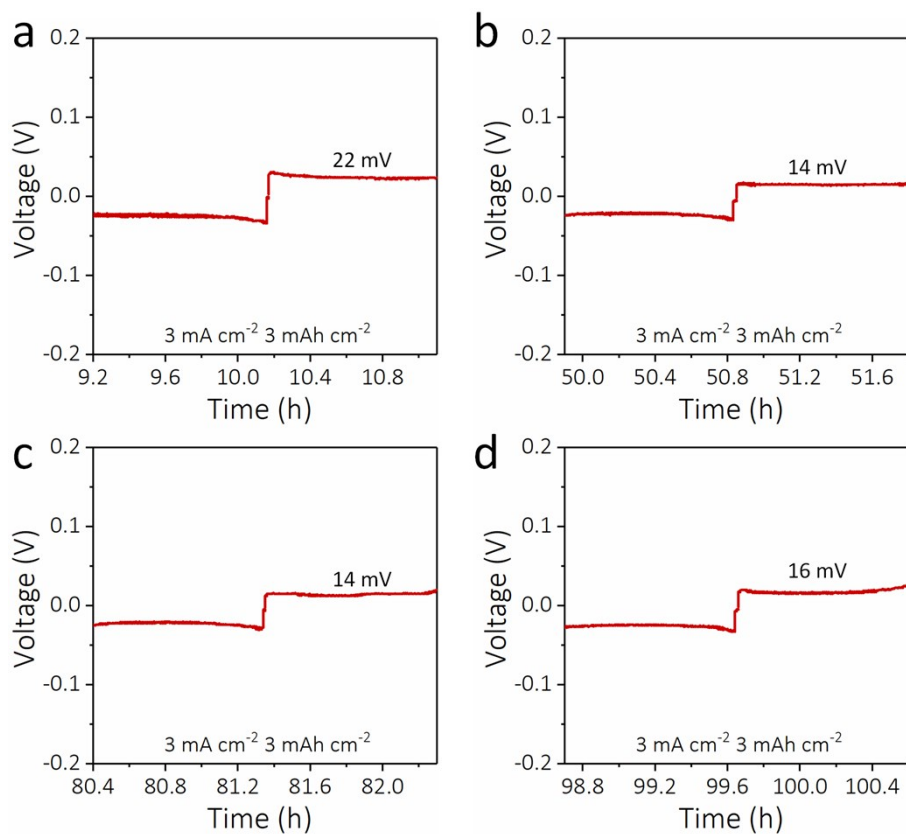


Fig. S11 Detailed voltage profiles of a-Ti₃C₂-K symmetric battery at different time. Planting 3 mAh cm⁻² of K at current density of 3 mA cm⁻² for (a) 10 h, (b) 50 h, (c) 80 h, and (d) 100 h, respectively.

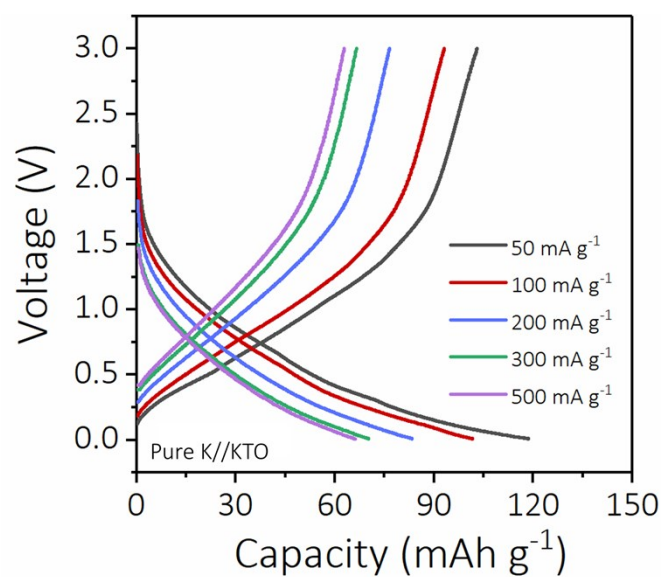


Fig. S12 Charge and discharge profiles of pure K//KTO battery tested at different rates.

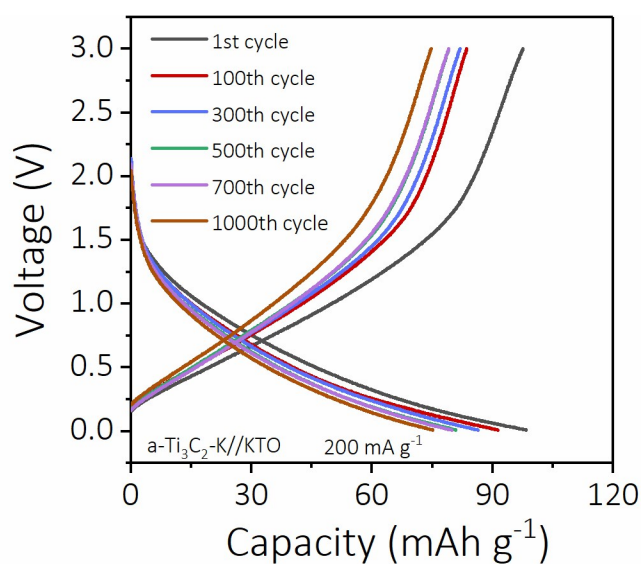


Fig. S13 Charge and discharge profiles of a-Ti₃C₂-K//KTO battery obtained at different cycles at 200 mA g⁻¹.

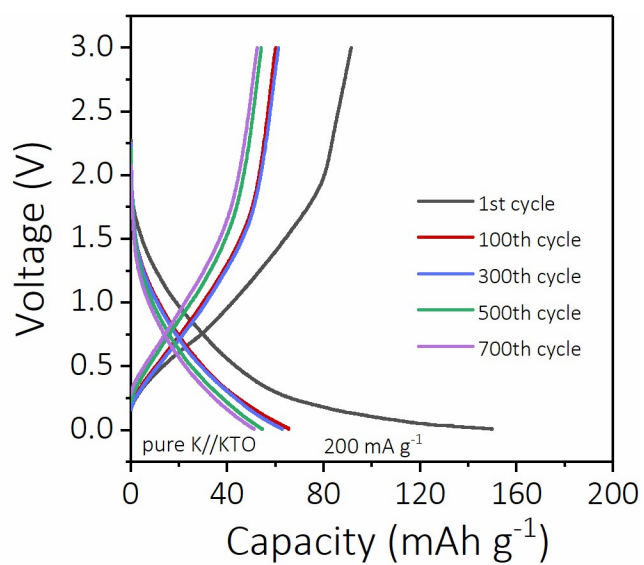


Fig. S14 Charge and discharge profiles of pure K//KTO battery obtained at different cycles at 200 mA g⁻¹.

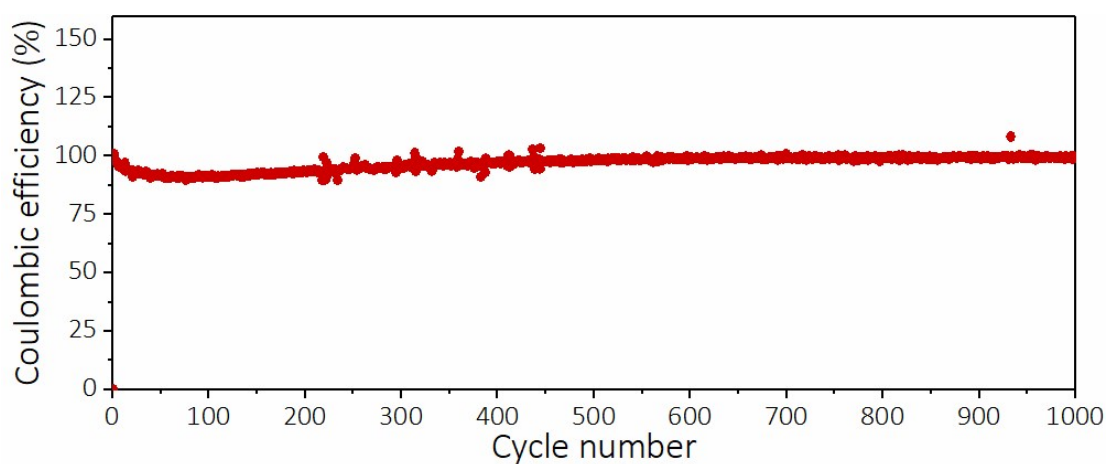


Fig. S15 Coulombic efficiency of a-Ti₃C₂-K//KTO battery tested at 200 mA g⁻¹.

Table S1. Performance comparison of a-Ti₃C₂-K electrode with the state-of-the-art K

Anode materials	Current density (mA cm⁻²)	Areal capacity (mAh cm⁻²)	Cycle time (h)	Refs.
rGO@3D-Cu	0.5	0.5	200	Ref. [1]
ACM	5	1	58	Ref. [2]
PCNF@SnO ₂	1	10	350	Ref. [3]
DN-MXene/CNT	0.5	0.5	300	Ref. [4]
HNCP/G	1	1	100	Ref. [5]
PM/NiO	0.4	0.1	100	Ref. [6]
K-Hg	0.5	0.5	250	Ref. [7]
Polished K metal	0.1	0.02	80	Ref. [8]
a-Ti₃C₂	5	10	700	This work
	5	5	800	

anodes recently reported

ACM: ligned carbon nanotube membrane; PCNF@SnO₂: SnO₂-coated conductive porous carbon nanofiber framework; DN-MXene: defect-rich and nitrogen-containing MXene; HNCP/G: hollow N-doped C polyhedrons/graphene; PM: puffed millet.

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