

Supplementary Information

NMR Probe Effects on *trans*-Philicity and *trans*-Influence Ladders in Square Planar Pt(II) Complexes

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Table S1. ^1H NMR Hydride Shifts (in ppm vs TMS) for selected $\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{L})]^{0/+}$ complexes calculated by the GIAO (SO-ZORA)/PBE0/TZ2P/COSMO computational protocol in benzene solution and compared to ^1H NMR Hydride Shifts given by Kaupp's group³⁰ and to experimental data available.⁴⁰⁻⁴⁴

Complex	$\sigma^{\text{iso}}(^1\text{H})^{\text{a}}$	$\delta_{\text{calcd}}(^1\text{H})^{\text{a}}$	$\delta_{\text{calcd}}(^1\text{H})^{\text{b}}$	$\delta_{\text{exptl}}(^1\text{H})$	Ref.
$\text{trans-}[\text{Pt}(\text{PMe}_3)(\text{H})(\text{H})]$	35.3 (35.1) ^c	-3.3 (-3.1) ^c		-2.7	43
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{Cl})]$	54.0 (54.1)	-22.0 (-22.1)	-16.2	[-16.2] -16.9	[44], 40 ^d
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{Br})]$	51.8 (51.9)	-19.8 (-19.9)	-15.0	-15.6	40
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{CN})]$	43.4 (43.3)	-11.4 (-11.3)	-6.7	-7.8	40
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{NCMe})]^+$	56.9 (57.0)	-24.9 (-15.0)		-18.7	42 ^e
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{SCN})]$	50.5 (50.6)	-18.5 (-18.6)	-12.2	-13.3	40
$\text{trans-}[\text{Pt}(\text{PMe}_2)(\text{H})(\text{NO}_2)]$	56.8 (56.9)	-24.8 (-24.9)	-19.0	-19.7	40
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{NO}_3)]$	61.6 (61.7)	-29.6 (-29.7)	-23.6	-23.8	40
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{OH})]$	54.7 (54.7)	-22.7 (-22.7)		-10.0	42
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{OH}_2)]^+$	67.0 (67.3)	-35.0 (-35.3)		-25.4	42
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{PMe}_3)]^+$	44.0 (44.0)	-12.0 (-12.0)		-12.8	43
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{Me})]$	37.4 (37.3)	-5.4 (-5.3)	-4.2	-3.8 [-2.1]	[41] ^f
$\text{trans-}[\text{Pt}(\text{PMe}_2)(\text{H})(\text{Ph})]$	39.8 (39.7)	-7.8 (-7.7)	-6.3	-5.7	

^a ^1H NMR Hydride Shifts calculated at the GIAO (SO-ZORA)/PBE0/TZ2P/COSMO level. ^b ^1H NMR Hydride Shifts calculated by Kaupp's group (taken from Ref. 30). ^c Figures in parentheses are the ^1H NMR Hydride Shifts calculated at the GIAO (SO-ZORA)/PBE0/TZ2P-J/COSMO level of theory. ^d The $\delta_{\text{exptl}}(^1\text{H})$ data given in Ref. 40 refer to the $\text{trans-}[\text{Pt}(\text{PEt}_3)_2(\text{H})(\text{Cl})]$ complexes. ^e The $\delta_{\text{exptl}}(^1\text{H})$ data given in Ref. 42 refer to the $\text{trans-}[\text{Pt}(\text{PPh}_3)_2(\text{H})(\text{OH}_2)]^+$ and $\text{trans-}[\text{Pt}(\text{PPh}_3)_2(\text{H})(\text{NCMe})]^+$ complexes. ^f The $\delta_{\text{exptl}}(^1\text{H})$ data given in Ref. 41 refer to the $\text{trans-}[\text{Pt}(\text{PPh}_3)_2(\text{H})(\text{Me})]$ complexes.

Table S2. $R(\text{Pt-H})$ bond lengths (in Å) for selected $\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{L})]^{0/+}$ complexes calculated by the PBE0/SARC-ZORA(Pt) \cup 6-31+G(d)(E)/PCM computational protocol and compared to $R(\text{Pt-H})$ bond lengths calculated by the PBE0/Def2-TZVP (ECP) computational protocol in the gas phase.^a

Complex	$R(\text{Pt-H})$		$\Delta R(\text{Pt-H})$
	PBE0/SARC-ZORA(Pt) \cup 6-31+G(d)(E)/PCM(benzene)	PBE0/Def2-TZVP (ECP) (gas phase)	
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{Cl})]$	1.616	1.565	0.051
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{Br})]$	1.620	1.566	0.054
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{CN})]$	1.654	1.609	0.045
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{SCN})]$	1.625	1.576	0.049
$\text{trans-}[\text{Pt}(\text{PMe}_2)(\text{H})(\text{NO}_2)]$	1.627	1.582	0.045
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{NO}_3)]$	1.599	1.548	0.051
$\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{H})(\text{Me})]$	1.699	1.636	0.063
$\text{trans-}[\text{Pt}(\text{PMe}_2)(\text{H})(\text{Ph})]$	1.693	1.628	0.065

MO. = **0.053**

^a Taken from Ref. 30.

Table S3. $\sigma^{\text{iso}}(\text{SO})$ ^1H NMR *trans*-philicity descriptors (in ppm) for the *trans*-[Pt(PMe₃)₂(H)L]^{0/+} complexes calculated at the 2c-SO-ZORA level of theory employing the GIAO(SO)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

Ligand	$\sigma^{\text{iso}}(\text{SO})$ ^1H NMR
F ⁻	60.80
H ₂ O	67.02
OCN ⁻	63.01
OBr	59.29
NO ₃ ⁻	61.57
OCl ⁻	59.29
CCl ₃ COO ⁻	60.60
CH ₃ COO ⁻	58.76
HCOO ⁻	58.84
C ₆ H ₅ COO ⁻	58.45
OF ⁻	57.54
OH ⁻	54.68
NO ₂ ⁻	56.79
NCS ⁻	54.80
NCMe	56.89
Py	57.60
NCH	57.05
NH ₃	56.49
NCPH	56.03
N ₃ ⁻	54.92
N ₂	57.97
Cl ⁻	53.97
NH ₂ ⁻	46.95
Br ⁻	51.80
SCN ⁻	50.54
H ₂ S	52.13
SH ⁻	46.17
CN ⁻	43.40
CNMe	44.56
NHC	43.91
CNH	43.90
CNPh	43.55
<i>t</i> -Bu ⁻	41.37
Ph ⁻	39.83
PMe ₃	44.00
CO	42.34
PPh ₃	43.82
Me ⁻	37.40
PH ₃	43.01
PF ₃	41.53
SnCl ₃ ⁻	44.07
H ⁻	35.29
B(OH) ₂ ⁻	29.27
BH ₂ ⁻	28.30

Table S4. $\sigma^{\text{iso}}(\text{SO})$ ^{13}C NMR *trans*-philicity descriptors (in ppm) for the *trans*-[Pt(PMe₃)₂(CH₃)L]^{0/+} complexes calculated at the 2c-SO-ZORA level of theory employing the GIAO(SO)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

Ligand	$\sigma^{\text{iso}}(\text{SO})$ ^{13}C NMR
BH ₂ ⁻	193.86
Ph ⁻	215.24
B(OH) ₂ ⁻	192.91
Me ⁻	210.27
CN ⁻	214.48
F ⁻	239.83
H ⁻	201.13
OF ⁻	233.19
OH ⁻	232.07
NH ₂ ⁻	220.16
OCl ⁻	233.19
OB ^r	234.79
Cl ⁻	224.55
NO ₂ ⁻	229.94
<i>t</i> -Bu ⁻	223.47
NCS ⁻	228.17
Br ⁻	219.59
N ₃ ⁻	225.45
SH ⁻	214.84
CH ₃ COO ⁻	232.36
C ₆ H ₅ COO ⁻	232.16
HCOO ⁻	231.47
NHC	211.13
NO ₃ ⁻	233.12
CCl ₃ COO ⁻	231.15
OCN ⁻	232.53
SCN ⁻	215.77
Py	223.82
CNPh	204.51
NCP ^h	219.14
NCMe	219.98
CNMe	204.50
NCH	217.93
CNH	202.44
NH ₃	219.14
SnCl ₃ ⁻	200.82
H ₂ O	225.03
CO	196.78
PMe ₃	199.17
N ₂	211.72
H ₂ S	206.55
PP ^h ₃	197.60
PH ₃	176.78
PF ₃	189.50

Table S5. $\sigma^{\text{iso}}(\text{SO})$ ^{13}C O NMR *trans*-philicity descriptors (in ppm) for the *trans*-[Pt(PMe₃)₂(CO)L]⁺²⁺ complexes calculated at the 2c-SO-ZORA level of theory employing the GIAO(SO)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

Ligand	$\sigma^{\text{iso}}(\text{SO})$ ^{13}C O NMR
H ₂ O	48.79
N ₂	42.27
NCH	36.23
OCN ⁻	34.49
NcMe	34.18
NcPh	32.79
NO ₃ ⁻	29.83
NH ₃	29.62
F ⁻	29.34
Py	28.98
CCl ₃ COO ⁻	28.16
H ₂ S	27.10
HCOO ⁻	25.53
CH ₃ COO ⁻	25.08
NCS ⁻	24.61
C ₆ H ₅ COO ⁻	24.17
OCl ⁻	23.72
OB ⁻	23.15
Cl ⁻	22.65
OF ⁻	22.52
Br ⁻	22.39
CO	22.10
NO ₂ ⁻	20.63
OH ⁻	19.88
N ₃ ⁻	19.66
PF ₃	18.00
CNMe	17.87
CNPh	17.08
SCN ⁻	14.77
PH ₃	13.93
PMe ₃	11.31
NHC	10.69
CNH	10.69
PPh ₃	9.73
NH ₂ ⁻	9.40
SH ⁻	8.71
CN ⁻	8.33
SnCl ₃ ⁻	2.95
Ph ⁻	-2.13
Me ⁻	-2.57
H ⁻	-7.85
t-Bu ⁻	-9.55
B(OH) ₂ ⁻	-13.76
BH ₂ ⁻	-17.16

Table S6. $\sigma^{\text{iso}}(\text{SO})$ $^{15}\text{NH}_2$ NMR *trans*-philicity descriptors (in ppm) for the *trans*-[Pt(PMe₃)₂(NH₂)L]^{0/+} complexes calculated at the 2c-SO-ZORA (SO) level of theory employing the GIAO(SO)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

Ligand	$\sigma^{\text{iso}}(\text{SO})$
OH ⁻	322.54
HCOO ⁻	315.45
CH ₃ COO ⁻	317.53
NH ₂ ⁻	313.56
C ₆ H ₅ COO ⁻	316.72
NO ₃ ⁻	305.93
F ⁻	306.50
Py	298.64
NH ₃	297.78
H ₂ O	285.89
OBr	305.20
B(OH) ₂ ⁻	311.65
OCl ⁻	304.63
NCH	293.78
OF ⁻	300.35
CCl ₃ COO ⁻	311.66
<i>t</i> -Bu ⁻	280.36
OCN ⁻	288.35
NO ₂ ⁻	288.65
N ₂	277.00
N ₃ ⁻	284.99
NCS ⁻	290.13
Ph ⁻	277.06
CNH	279.14
CNMe	279.46
Cl ⁻	277.26
PMe ₃	271.20
Me ⁻	266.73
PH ₃	266.41
Br ⁻	268.73
BH ₂ ⁻	226.80
SCN ⁻	264.58
SH ⁻	263.87
H ⁻	255.34
NCMe	267.37
NCPH	266.50
CN ⁻	268.83
NHC	257.33
PF ₃	250.14
H ₂ S	240.63
CNPh	240.05
SnCl ₃ ⁻	230.32
PPh ₃	223.89
CO	224.48

Table S7. $\sigma^{\text{iso}}(\text{SO})$ $^{17}\text{OH}_2$ NMR *trans*-philicity descriptors (in ppm) for the *trans*- $[\text{Pt}(\text{PMe}_3)_2(\text{OH}_2)\text{L}]^{+2+}$ complexes calculated at the 2c-SO-ZORA (SO) level of theory employing the GIAO(SO)/PBE0/TZ2P/COSMO computational protocols in benzene solution.

Ligand	$\sigma^{\text{iso}}(\text{SO})$ $^{17}\text{OH}_2$ NMR
H ₂ O	513.76
F ⁻	463.62
OCN ⁻	452.51
NO ₃ ⁻	456.87
CCl ₃ COO ⁻	447.87
CH ₃ COO ⁻	441.37
C ₆ H ₅ COO ⁻	441.72
N ₂	450.34
HCOO ⁻	441.52
OBr	425.93
NCH	446.24
NH ₃	439.67
OH ⁻	426.28
OCI ⁻	423.54
NCS ⁻	431.83
Py	434.83
OF ⁻	419.11
NO ₂ ⁻	417.74
Cl ⁻	409.95
NCMe	427.00
N ₃ ⁻	402.78
H ₂ S	407.34
NCPH	434.02
CNMe	406.90
CNH	411.75
CN ⁻	408.87
NH ₂ ⁻	389.45
SCN ⁻	386.89
NHC	372.09
CO	397.80
PF ₃	383.92
Br	383.52
PH ₃	379.36
CNPh	386.44
SH ⁻	361.08
Me ⁻	362.59
B(OH) ₂ ⁻	350.33
PPh ₃	358.63
PMe ₃	359.46
Ph ⁻	359.72
H ⁻	356.79
SnCl ₃ ⁻	344.63
<i>t</i> -Bu ⁻	323.37
BH ₂ ⁻	327.04

Table S8. $\sigma^{\text{iso}}(\text{SO})$ ^{35}Cl NMR *trans*-philicity descriptors (in ppm) for the *trans*-[Pt(PMe₃)₂(Cl)L]^{0/+} complexes calculated at the 2c-SO-ZORA (SO) level of theory employing the GIAO(SO)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

Ligand	$\sigma^{\text{iso}}(\text{SO})$ ^{35}Cl NMR
F ⁻	1225.20
OBr ⁻	1203.85
OCN ⁻	1176.67
OCl ⁻	1202.23
NO ₃ ⁻	1186.38
OH ⁻	1204.76
CH ₃ COO ⁻	1189.16
OF ⁻	1193.92
C ₆ H ₅ COO ⁻	1188.74
CCl ₃ COO ⁻	1177.68
HCOO ⁻	1182.18
NO ₂ ⁻	1187.39
NCS ⁻	1173.53
H ₂ O	1096.56
N ₃ ⁻	1160.22
NH ₂ ⁻	1166.19
Py	1129.75
Cl ⁻	1129.33
Ph ⁻	1154.24
Br ⁻	1112.91
NH ₃	1101.96
Me ⁻	1137.19
NCMe	1097.13
CN ⁻	1129.32
NCPH	1095.76
H ⁻	1129.00
SH ⁻	1105.04
BH ₂ ⁻	1123.03
NCH	1079.90
SCN ⁻	1086.11
B(OH) ₂ ⁻	1117.91
NHC	1096.59
<i>t</i> -Bu ⁻	1094.34
N ₂	1021.80
CNMe	1052.94
CNPh	1041.67
CNH	1037.34
H ₂ S	1002.84
SnCl ₃ ⁻	1005.93
PPh ₃	998.39
PMe ₃	995.56
PH ₃	983.81
CO	989.68
PF ₃	940.66

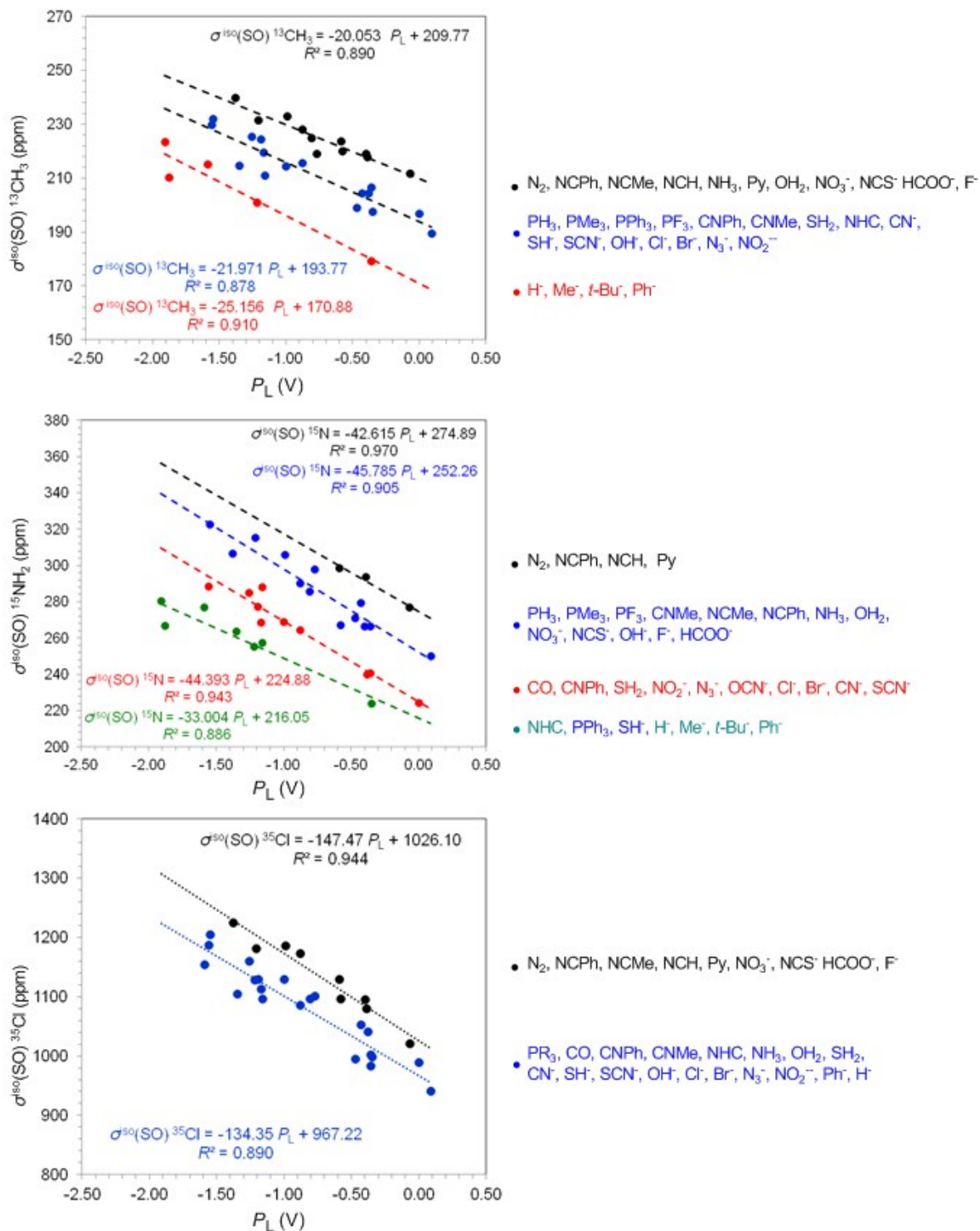


Figure S1. Linear plots of the $\sigma^{\text{iso}}(\text{SO})$ X (X = $^{13}\text{CH}_3$, $^{15}\text{NH}_2$, ^{35}Cl) vs P_L correlations for square planar *trans*- $[\text{Pt}(\text{PMe}_3)_2(\text{X})\text{L}]^{0/+}$ (X = X = CH₃, NH₂, Cl) complexes calculated by the GIAO(2c-SO-ZORA)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

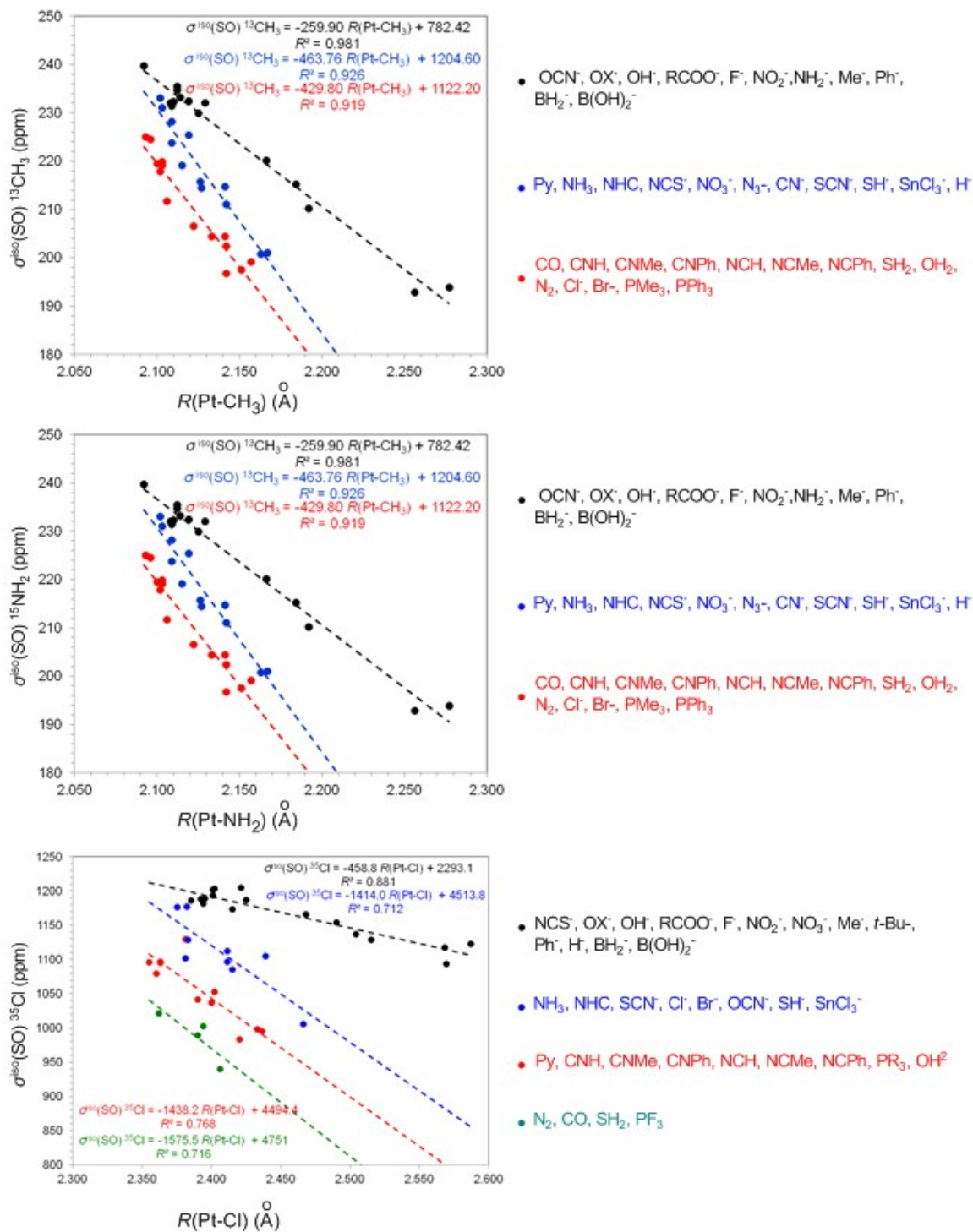


Figure S2. Linear plots of the $\sigma^{\text{iso}}(\text{SO})$ X (X = CH₃, NH₂, Cl) shieldings vs $R(\text{Pt-X})$ correlations for square planar *trans*-[Pt(PMe₃)₂(X)L]^{0/+} (X = CH₃, NH₂, Cl) complexes calculated by the GIAO(2c-SO-ZORA)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

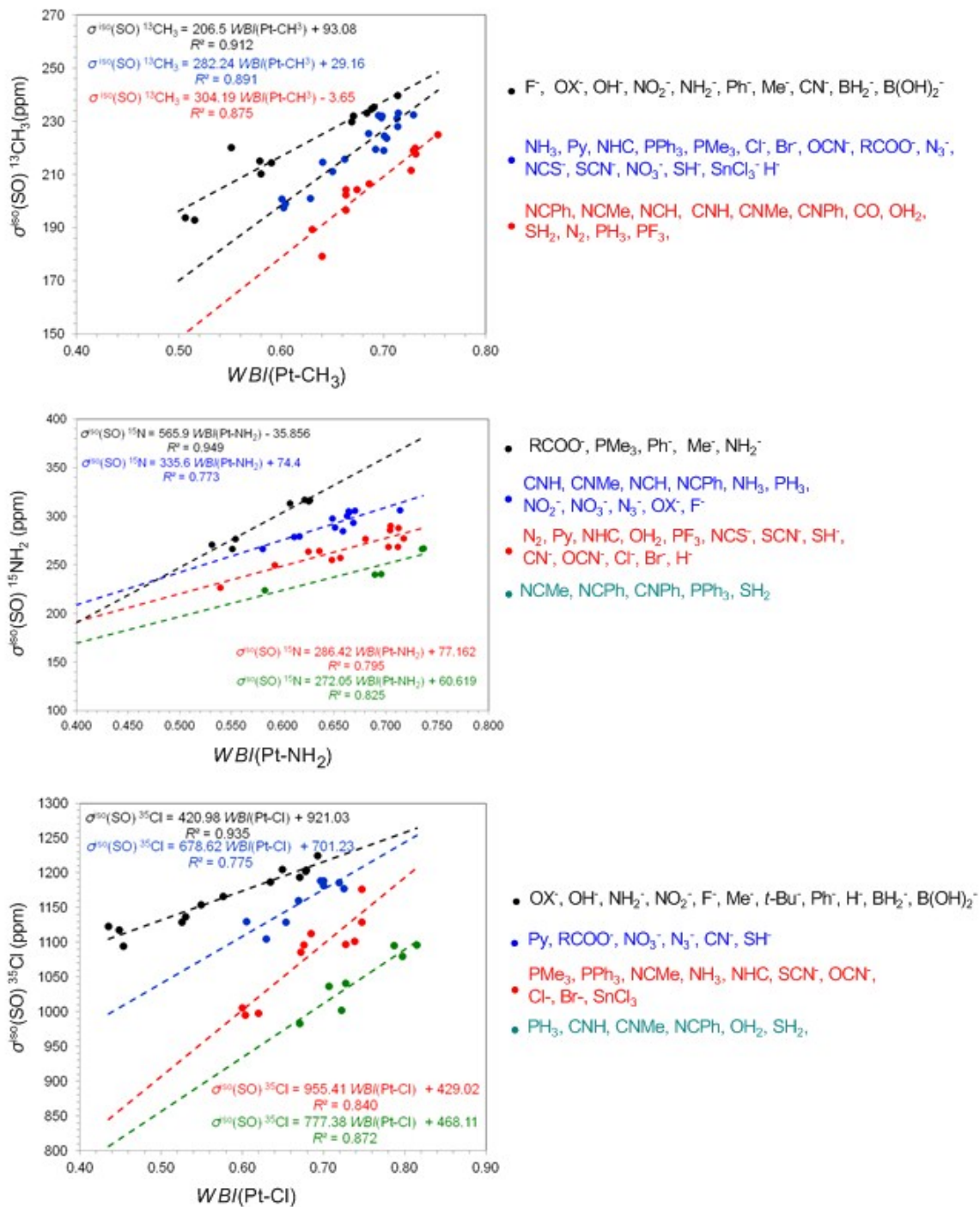


Figure S3. Linear plots of the $\sigma^{\text{iso}}(\text{SO})$ X (X = CH₃, NH₂, Cl) shieldings vs $\text{WBI}(\text{Pt-X})$ correlations for square planar $\text{trans-}[\text{Pt}(\text{PMe}_3)_2(\text{X})\text{L}]^{0/+}$ (X = CH₃, NH₂, Cl) complexes calculated by the GIAO(2c-SO-ZORA)/PBE0/TZ2P/COSMO computational protocol in benzene solution.

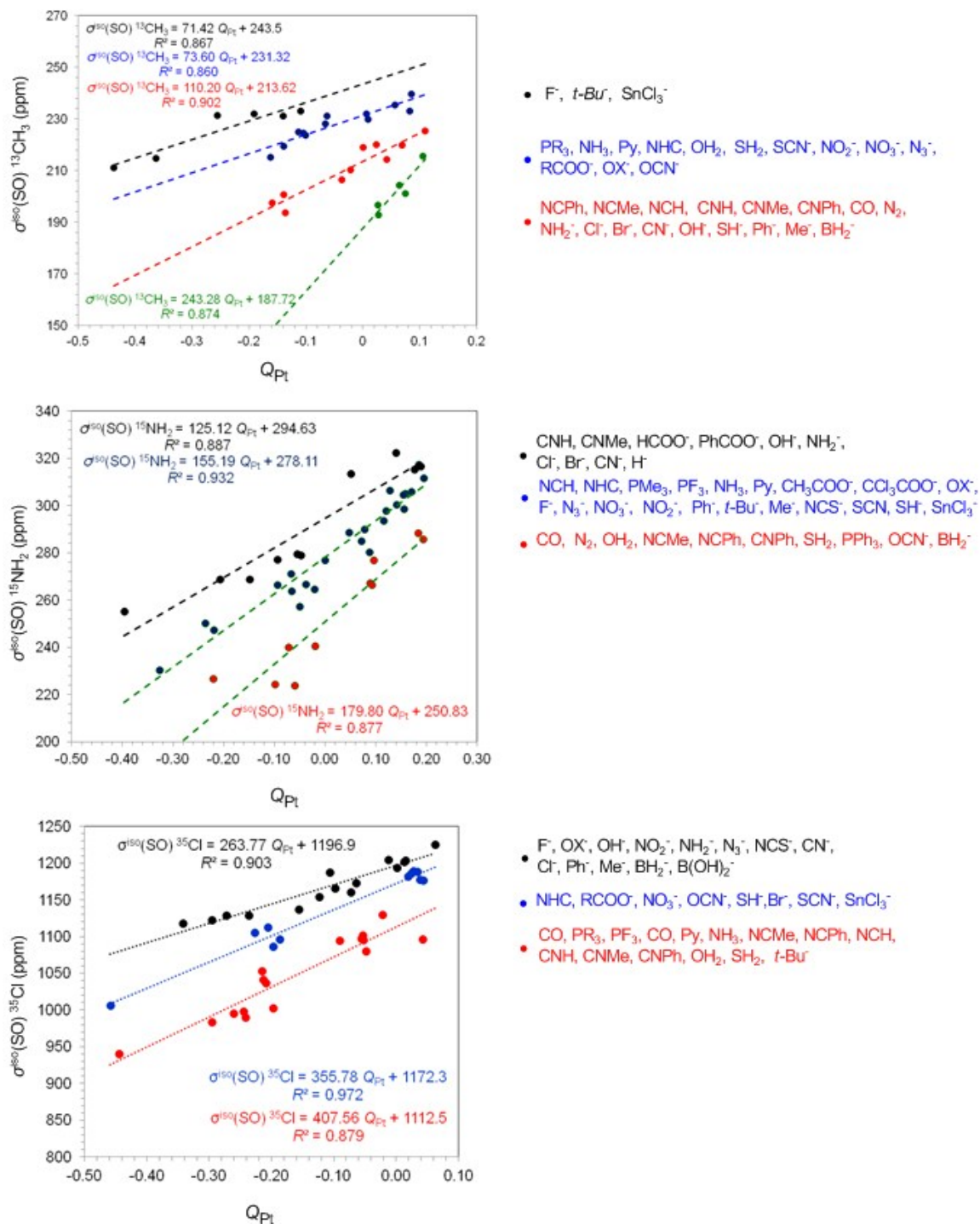


Figure S4. Linear plots of the $\sigma^{\text{iso}}(\text{SO}) \text{ X}$ ($\text{X} = \text{CH}_3, \text{NH}_2, \text{Cl}$) shieldings vs Q_{Pt} correlations for square planar *trans*- $[\text{Pt}(\text{PMe}_3)_2(\text{X})\text{L}]^{0/+}$ ($\text{X} = \text{CH}_3, \text{NH}_2, \text{Cl}$) complexes calculated by the GIAO(2c-SO-ZORA)/PBE0/TZ2P/COSMO computational protocol in benzene solution.