

## Supplementary Information

### Design and Antibacterial Activity Assessment of “Green” Synthesized 1,4-disubstituted 1,2,3-triazoles via a Fe<sub>3</sub>O<sub>4</sub>/Silicalite-1/PVA/Cu(I) nanocomposite catalyzed three component reaction

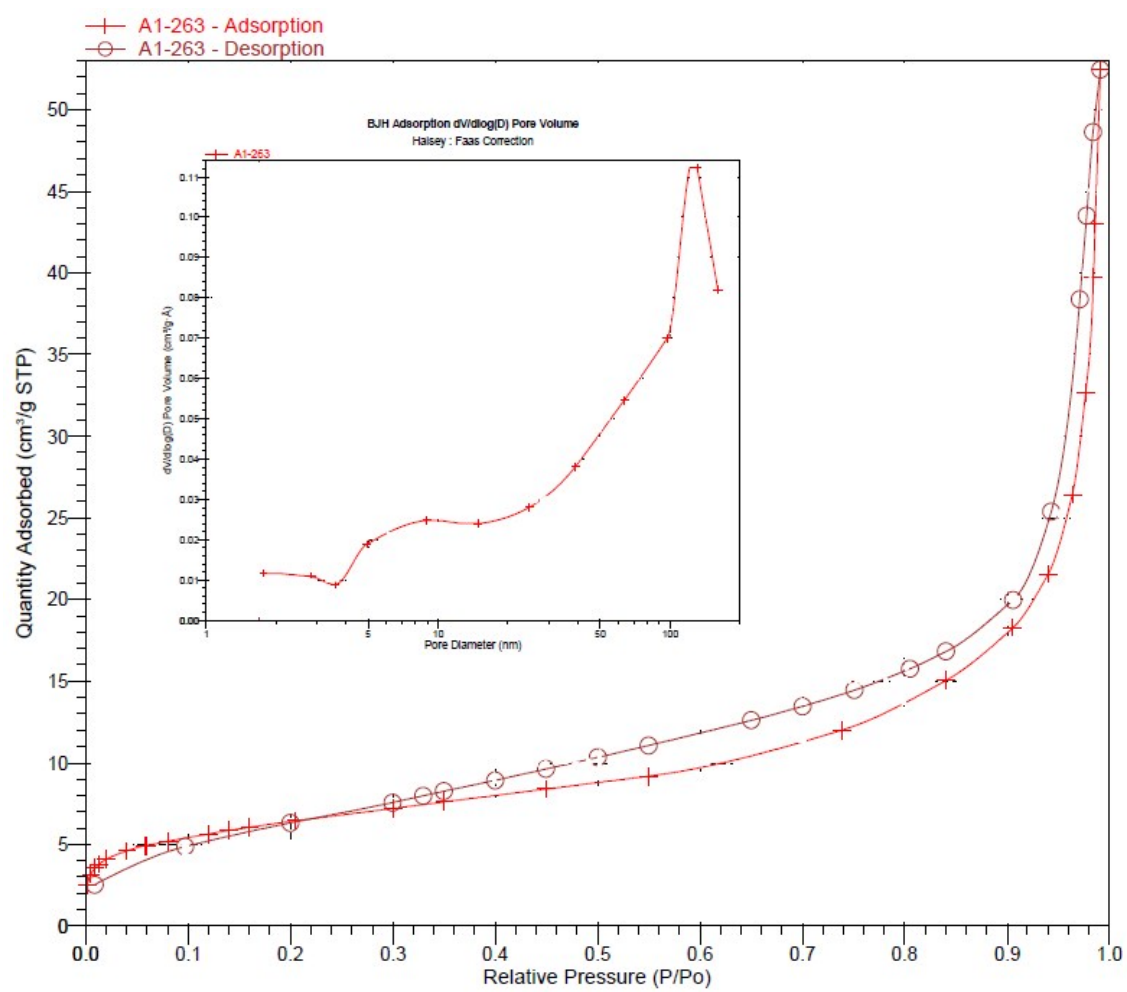
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University of Science and Technology, Tehran 16846-13114, Iran*

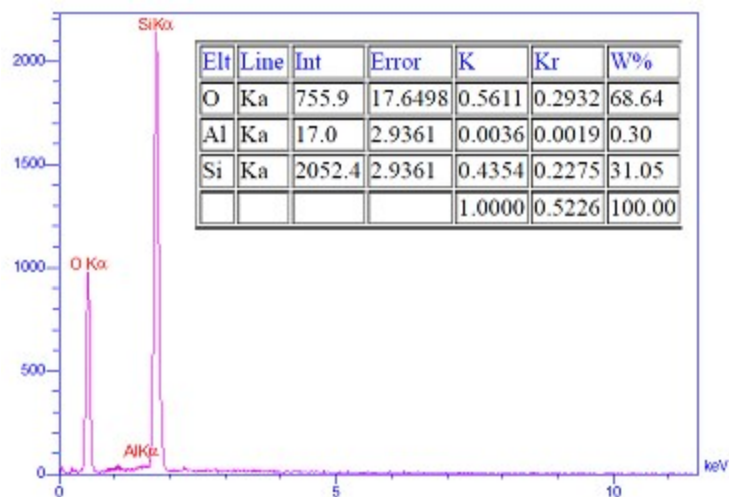
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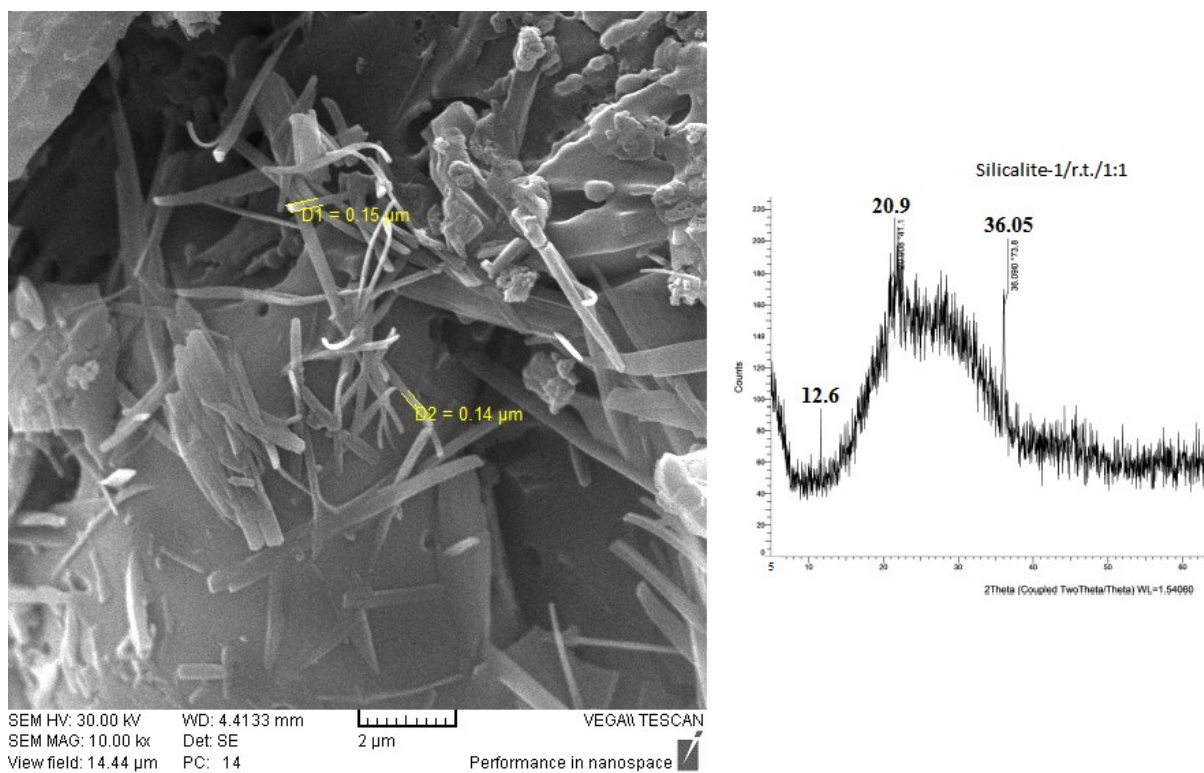
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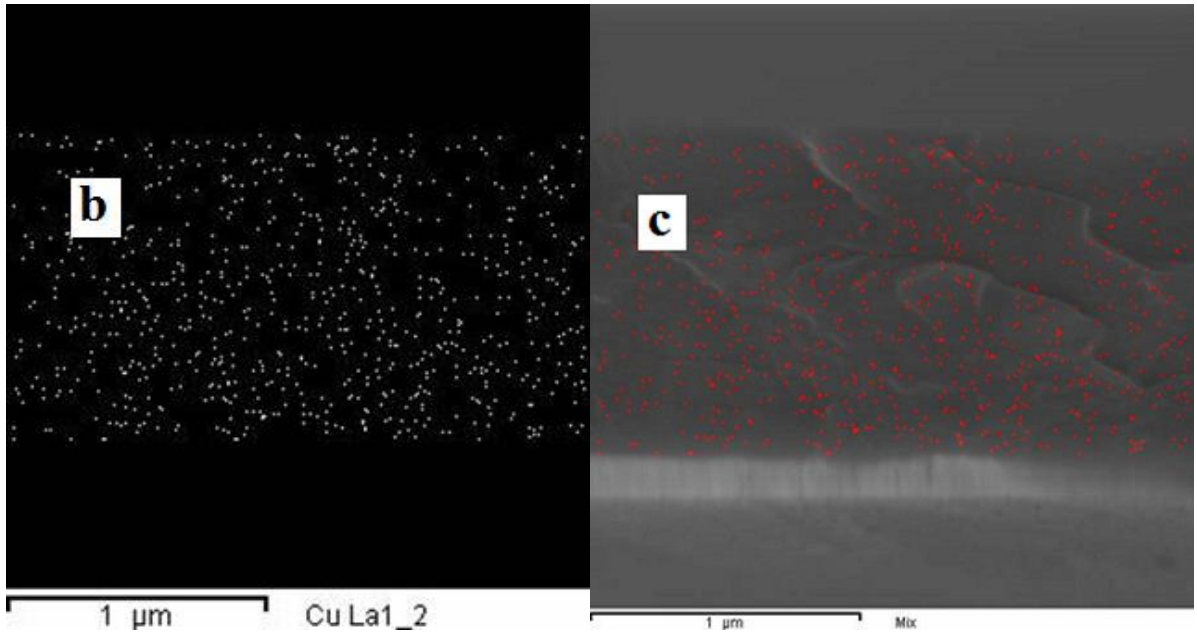
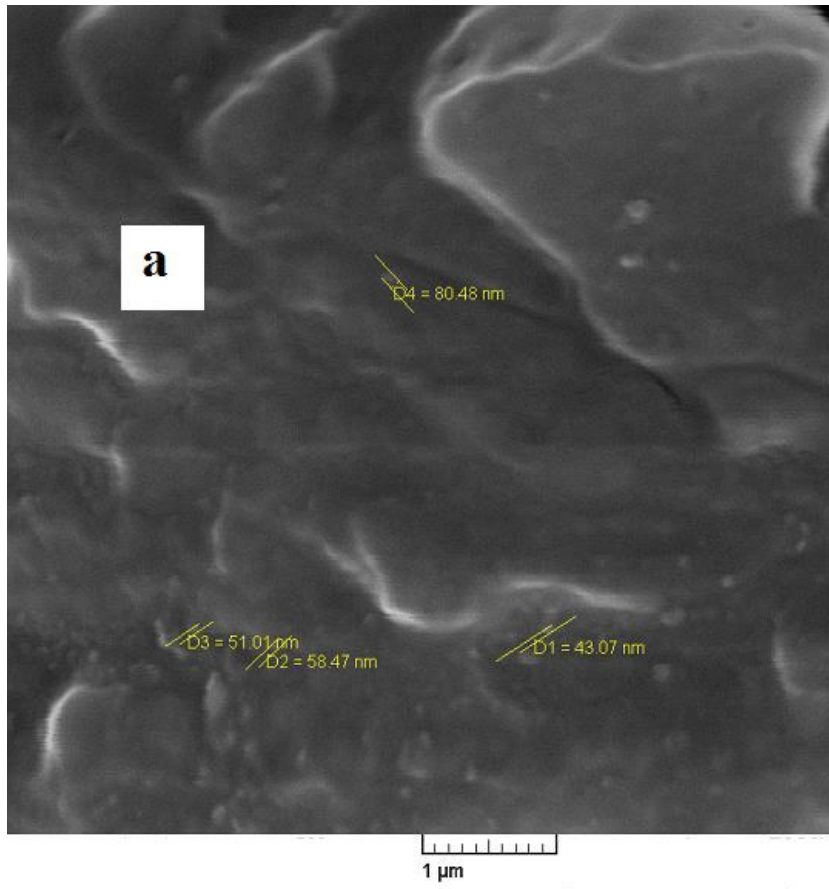
**Figure S1.** N<sub>2</sub> Adsorption-desorption isotherms and pore size distribution of Si-2.

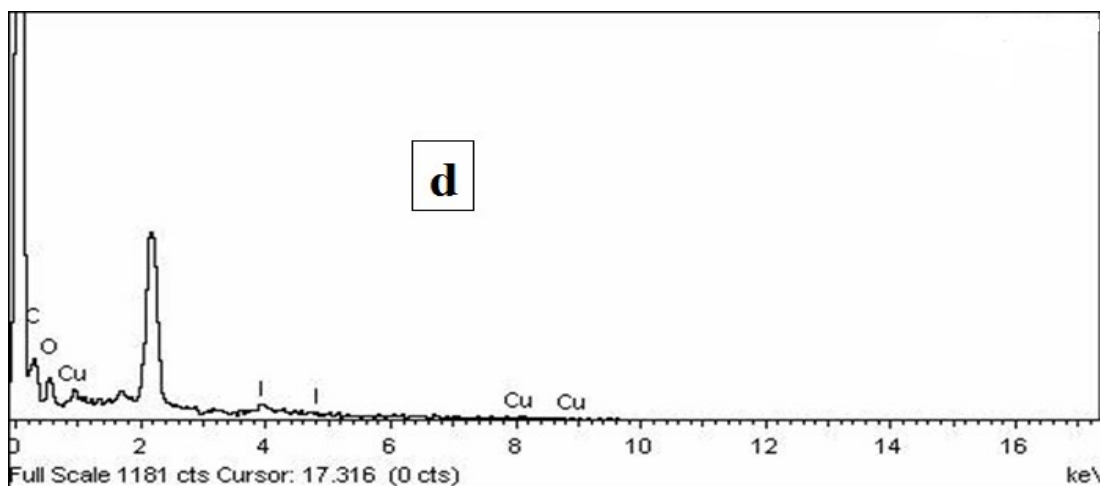


**Figure S2.** EDX data of Silicalite-1(Si-1) synthesized *via* sol-gel process.

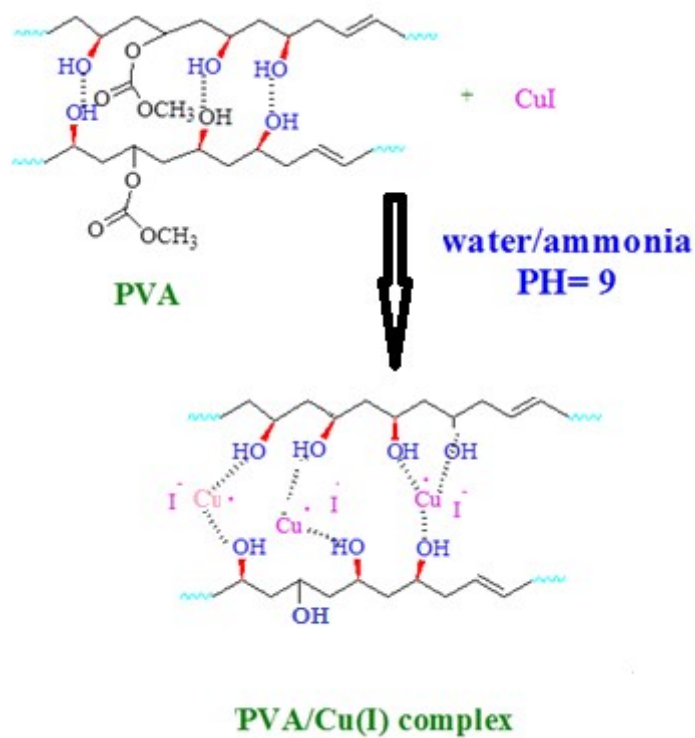


**Figure S3.** SEM image, EDX data and XRD of Silicalite-1(Si-2) synthesized *via* sol-gel process.

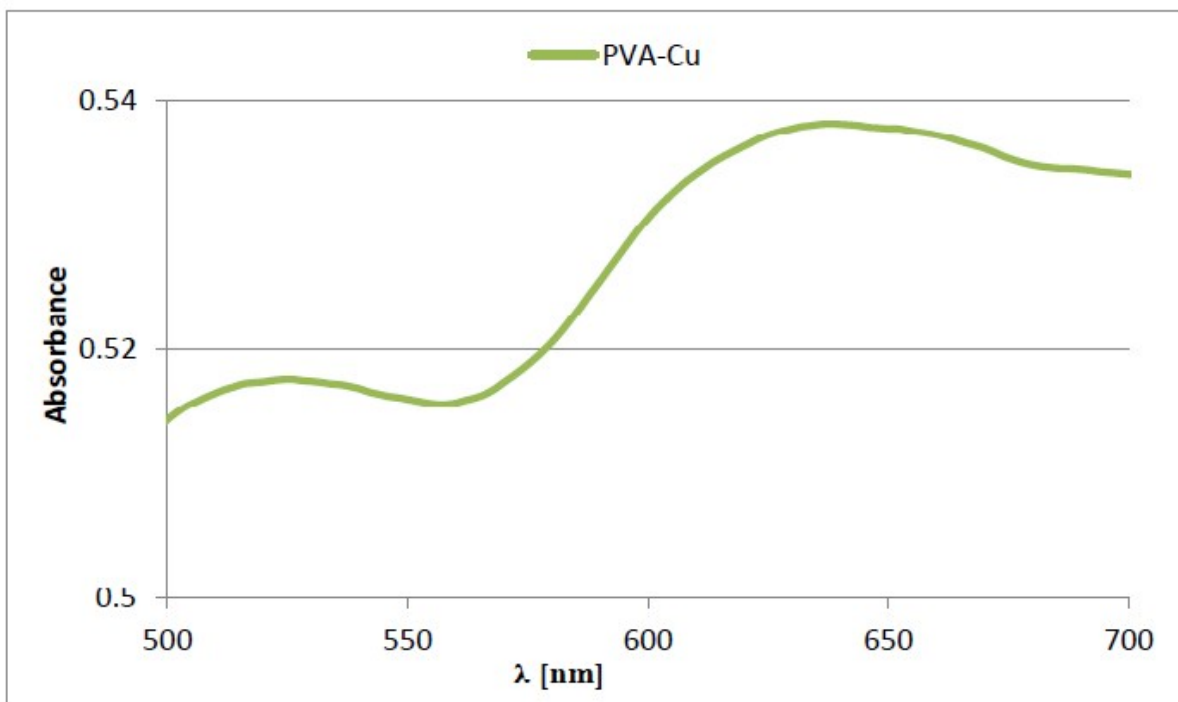




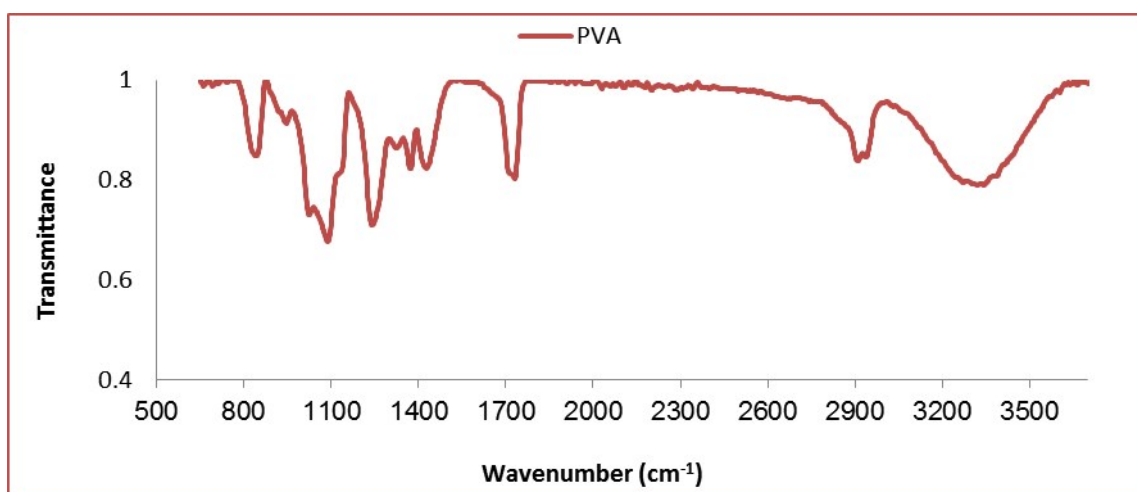
**Figure S4.** SEM image of PVA/Cu(I) catalyst (a), dispersion of PVA/Cu(I) catalyst (b and c) and EDX data of PVA/Cu(I) (d).



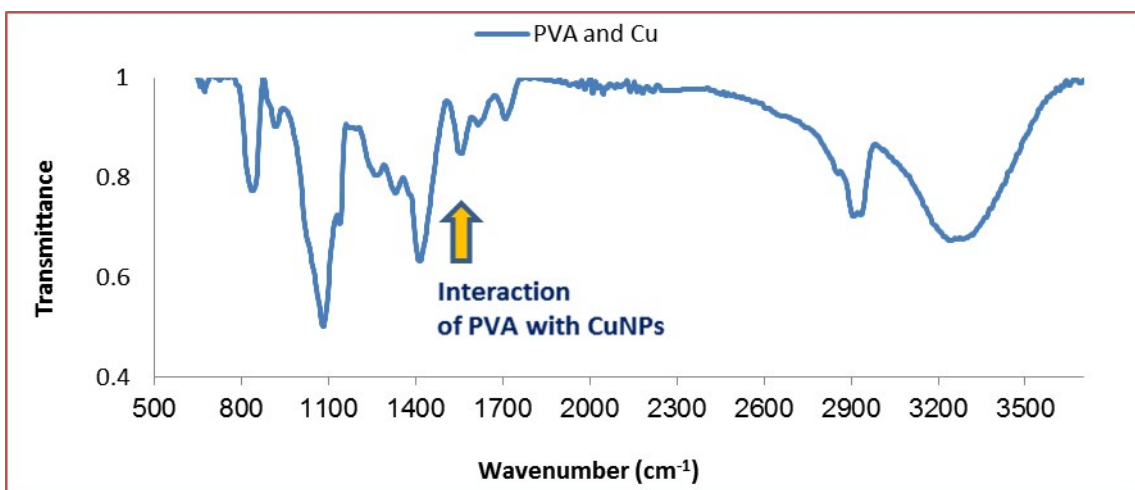
**Scheme S1.** Mechanism of reaction between PVA and CuI in formation of PVA/Cu(I) nanocomposite.



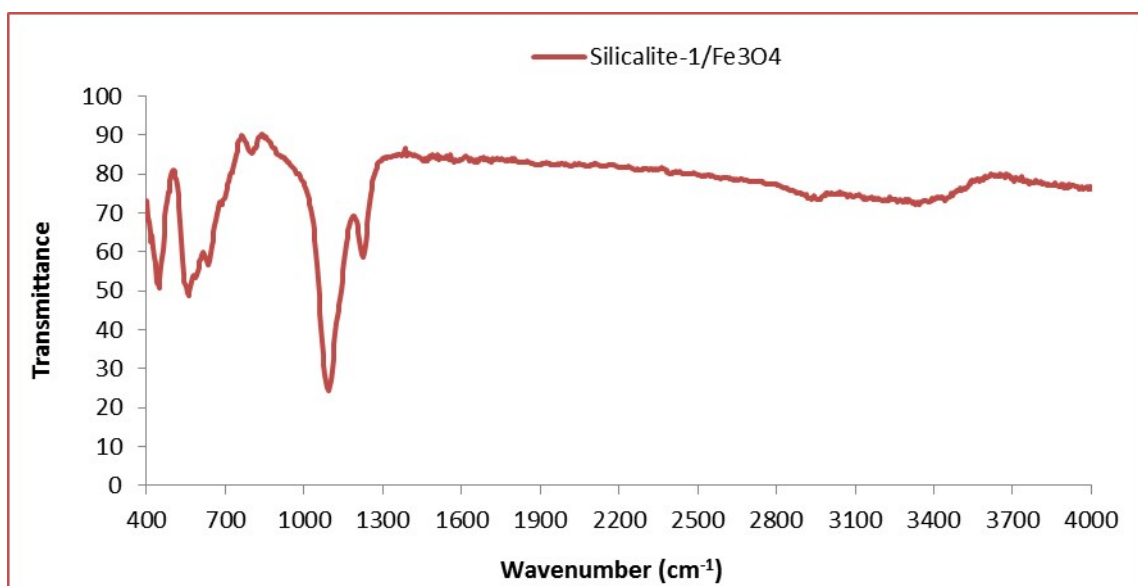
**Figure S5.** UV-Vis spectrum of PVA/Cu(I) nanocatalyst.



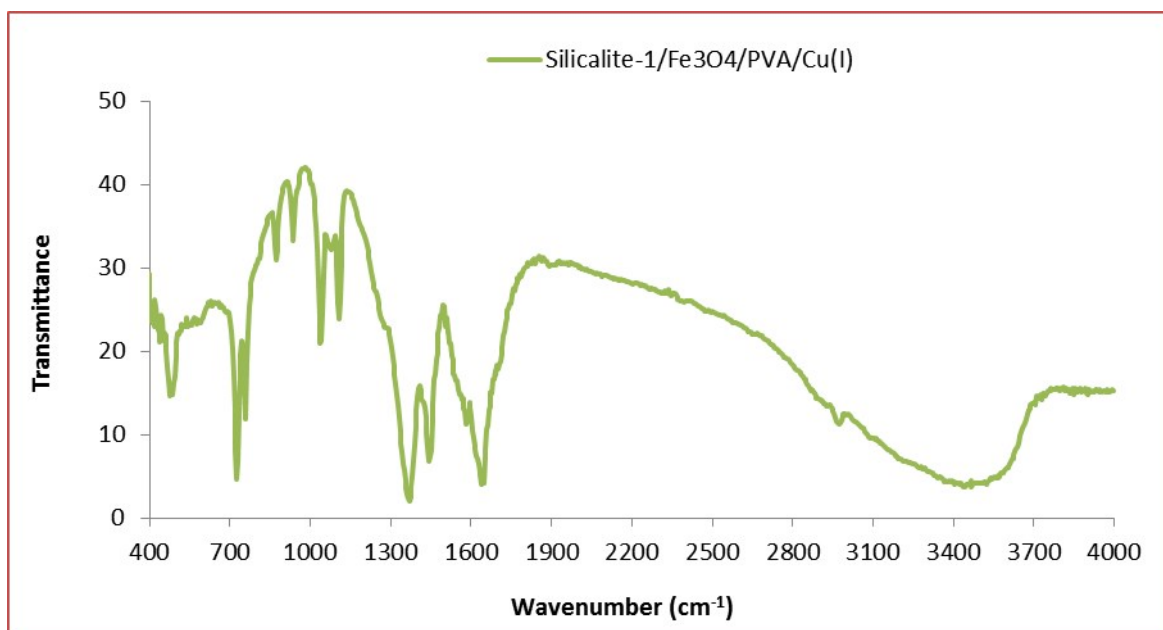
**Figure S6.** FT-IR spectrum of PVA.



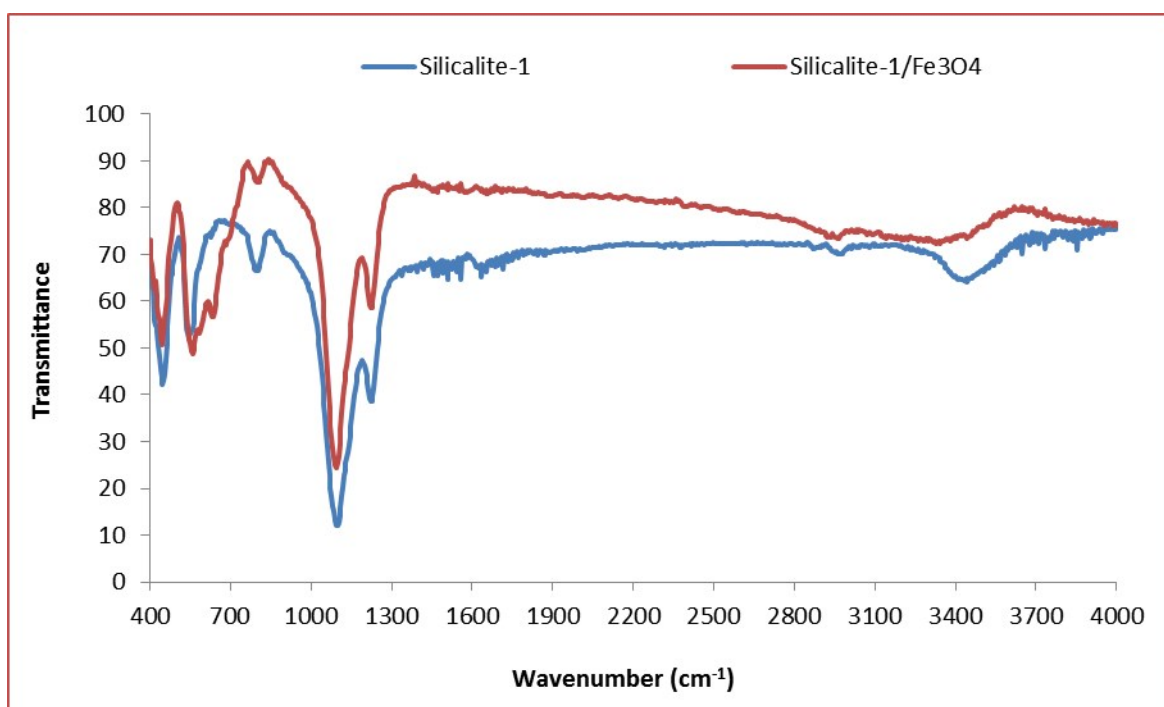
**Figure S7.** FT-IR spectrum of PVA/Cu(I).



**Figure S8.** FT-IR spectrum of Fe<sub>3</sub>O<sub>4</sub>/Silicalite-1 nanocomposite.



**Figure S9.** FT-IR spectrum of Fe<sub>3</sub>O<sub>4</sub>/Silicalite-1/PVA/Cu(I).



**Figure S10.** FT-IR spectra of Fe<sub>3</sub>O<sub>4</sub>/Silicalite-1 nanocomposite and Fe<sub>3</sub>O<sub>4</sub>/Silicalite-1/PVA/Cu(I).



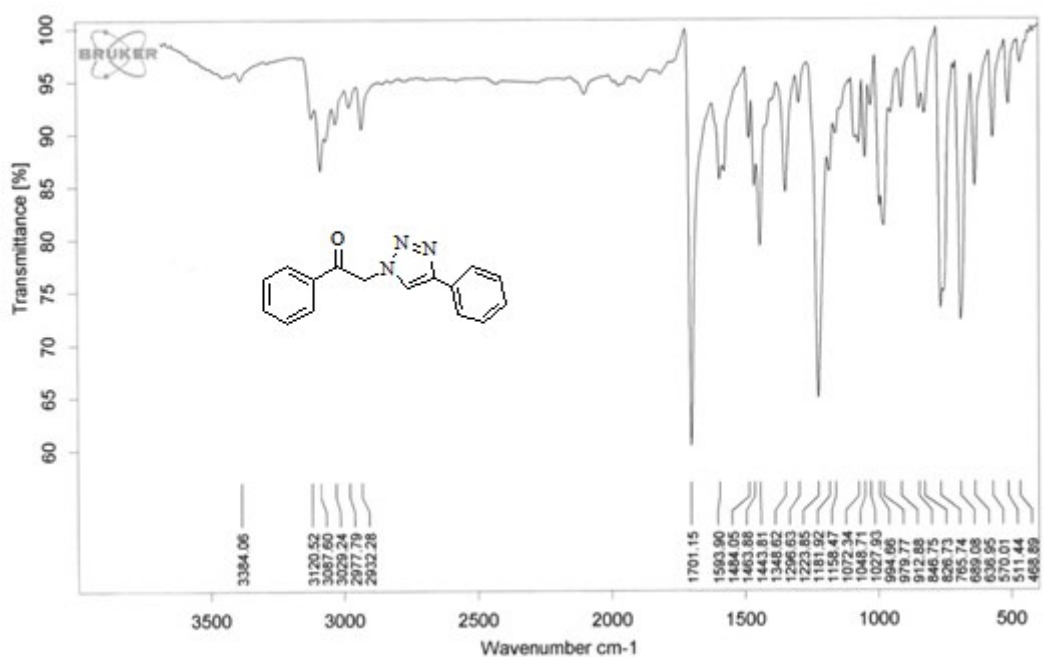


Figure S11. IR Spectrum of 1-Phenyl-2-(4-phenyl-[1,2,3]triazol-1-yl)ethanone **5a**.

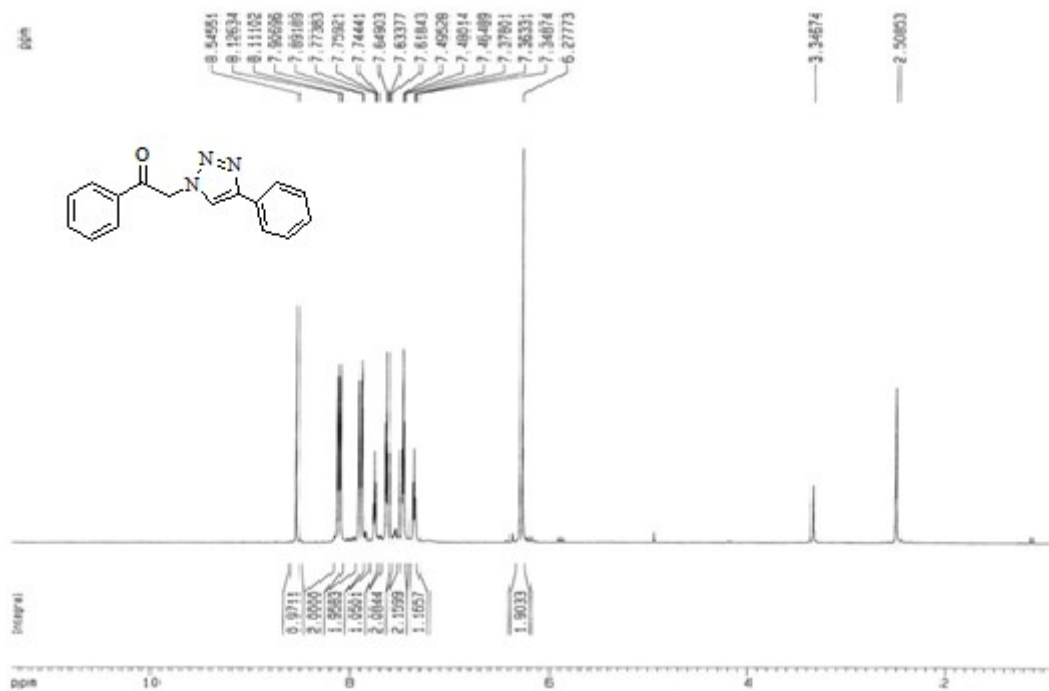
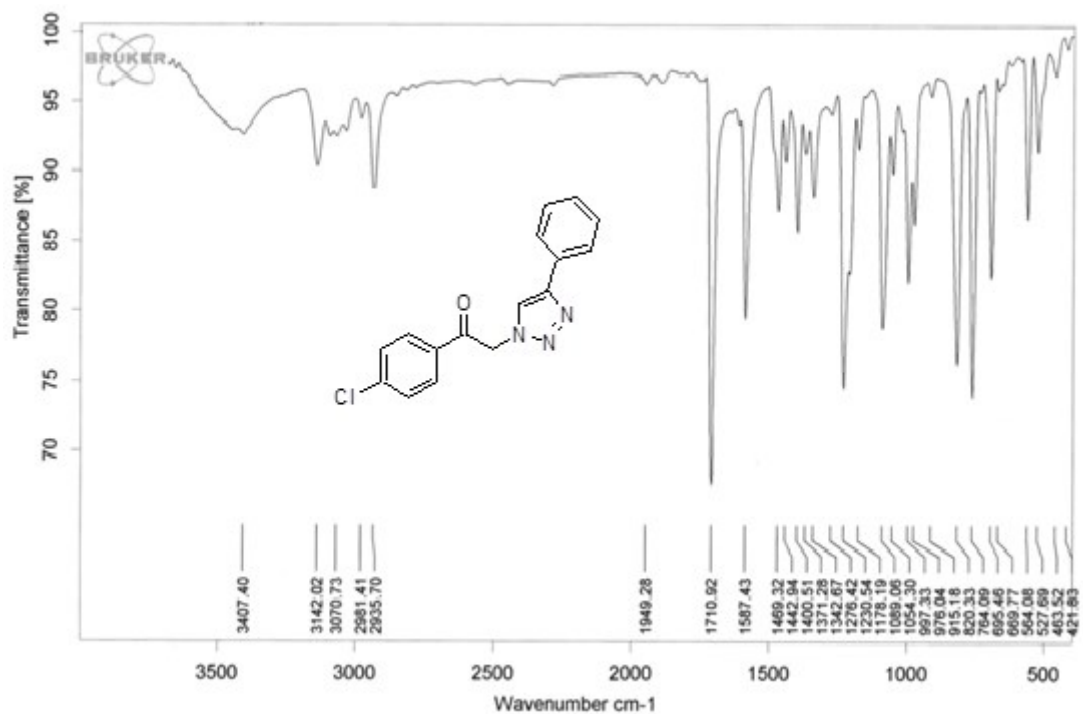
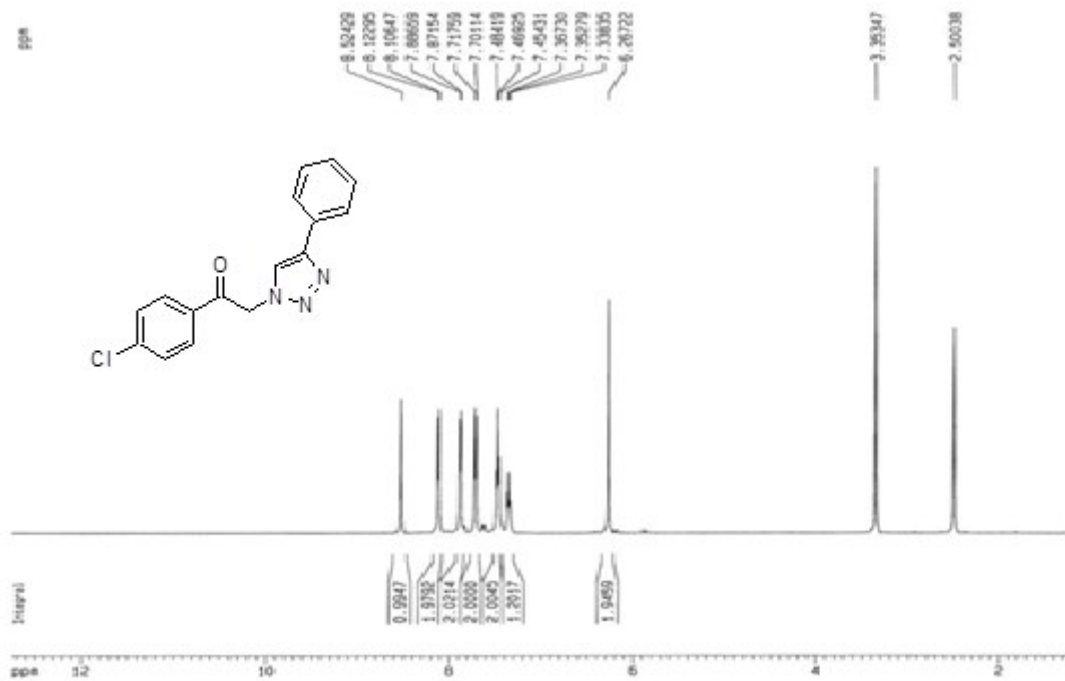


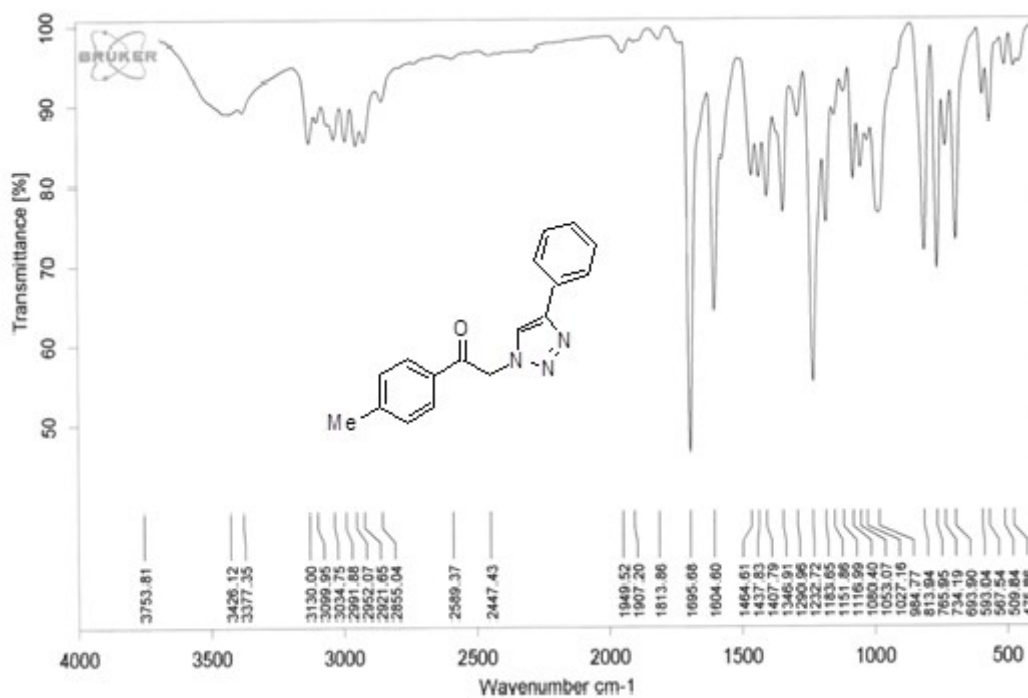
Figure S12. <sup>1</sup>H NMR Spectrum of 1-Phenyl-2-(4-phenyl-[1,2,3] triazol-1-yl) ethanone **5a**.



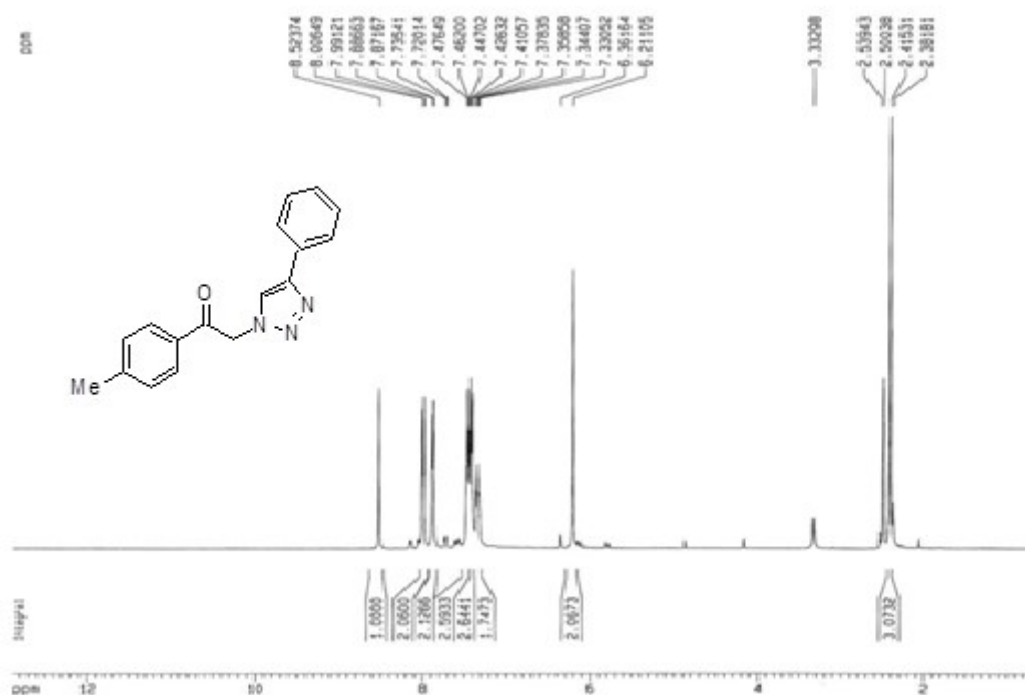
**Figure S13.** IR Spectrum of 1-(4-Chlorophenyl)-2-(4-phenyl-1H-1,2,3-triazol-1-yl)ethanone **5d**.



**Figure S14.** <sup>1</sup>H NMR Spectrum of 1-(4-Chlorophenyl)-2-(4-phenyl-1H-1,2,3-triazol-1-yl)ethanone **5d**.



**Figure S15.** IR Spectrum of 2-(4-Phenyl-1H-1,2,3-triazol-1-yl)-1-p-tolyl-ethanone **5e**.



**Figure S16.** <sup>1</sup>H NMR Spectrum of 2-(4-Phenyl-1H-1,2,3-triazol-1-yl)-1-p-tolyl-ethanone **5e**.

### Materials and equipment used in this study

All starting chemicals and reagents were commercially available and were utilized without further purification. Melting points were obtained through an Electrothermal 9100 apparatus. Fourier transform infrared spectroscopy (FT-IR) spectra were recorded by the film without KBr with a Shimadzu IR-470 spectrometer. Scanning electron micrograph (SEM) images were taken with a VEGA-TESCAN microscope with an attached camera. Energy-dispersive X-ray (EDX) analysis recorded by Numerix DXP-X10P. The XRD measurements were carried out by using a DRON-8 X-ray diffractometer, thermal analysis (TGA) was done by using of Bahr-STA 504 instrument under an argon atmosphere, and the magnetic properties of the sample were detected at room temperature using a VSM (Meghnatis Kavir Kashan Co., Kashan, Iran). Analytical thin-layer chromatography (TLC) was performed using

Merck silica gel GF254 plates. The surface areas of the catalyst sample were measured by N<sub>2</sub> adsorption at -195 °C with Autosorb-1 (Quanta Chrome), the BET equation being applied to the isotherm. The external surface area was obtained from a t-plot.

### Acknowledgements

The authors are deeply grateful to Iran National Science Foundation (INSF) for financial support (Cod number 97009059) and Iran University of Science and Technology for partial support of this research project.

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