

## Investigation of High-Performance Adsorption for Benzene and Toluene Vapors by Calix[4]arene based Organosilica (CBOS)

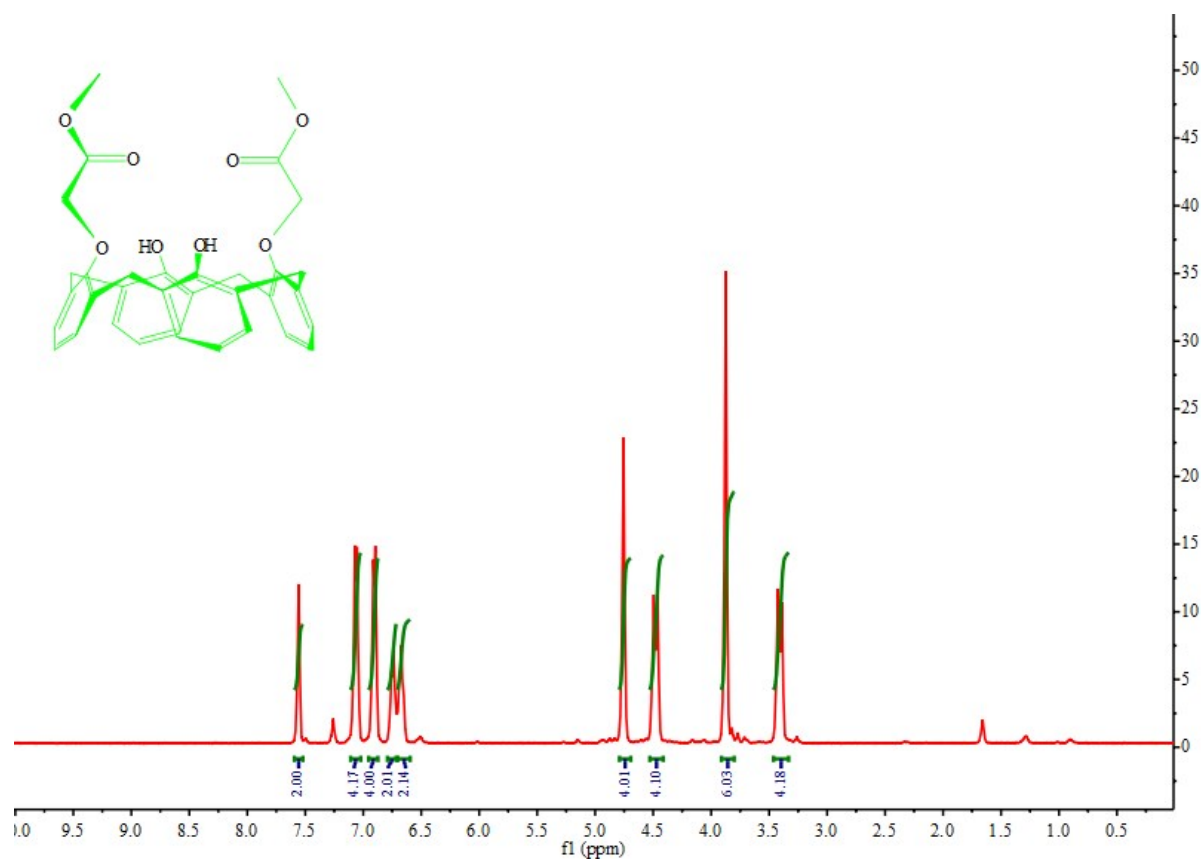
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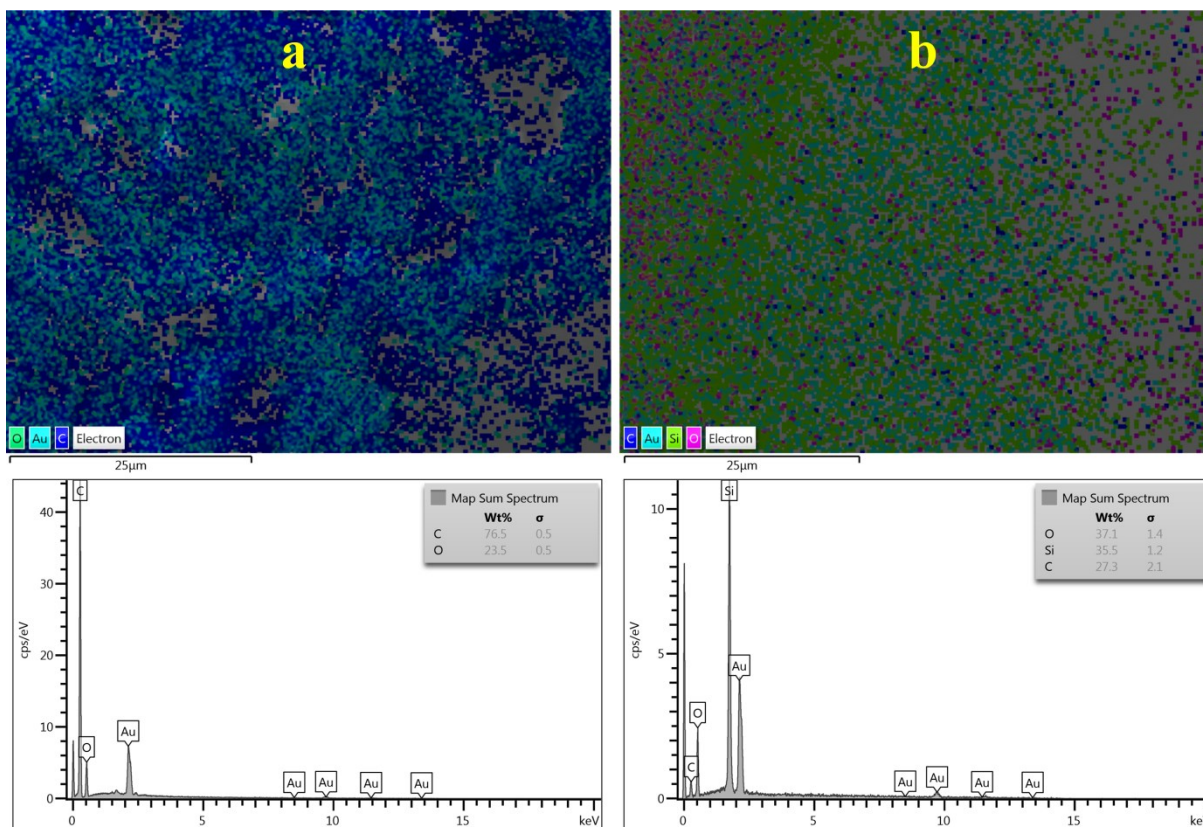
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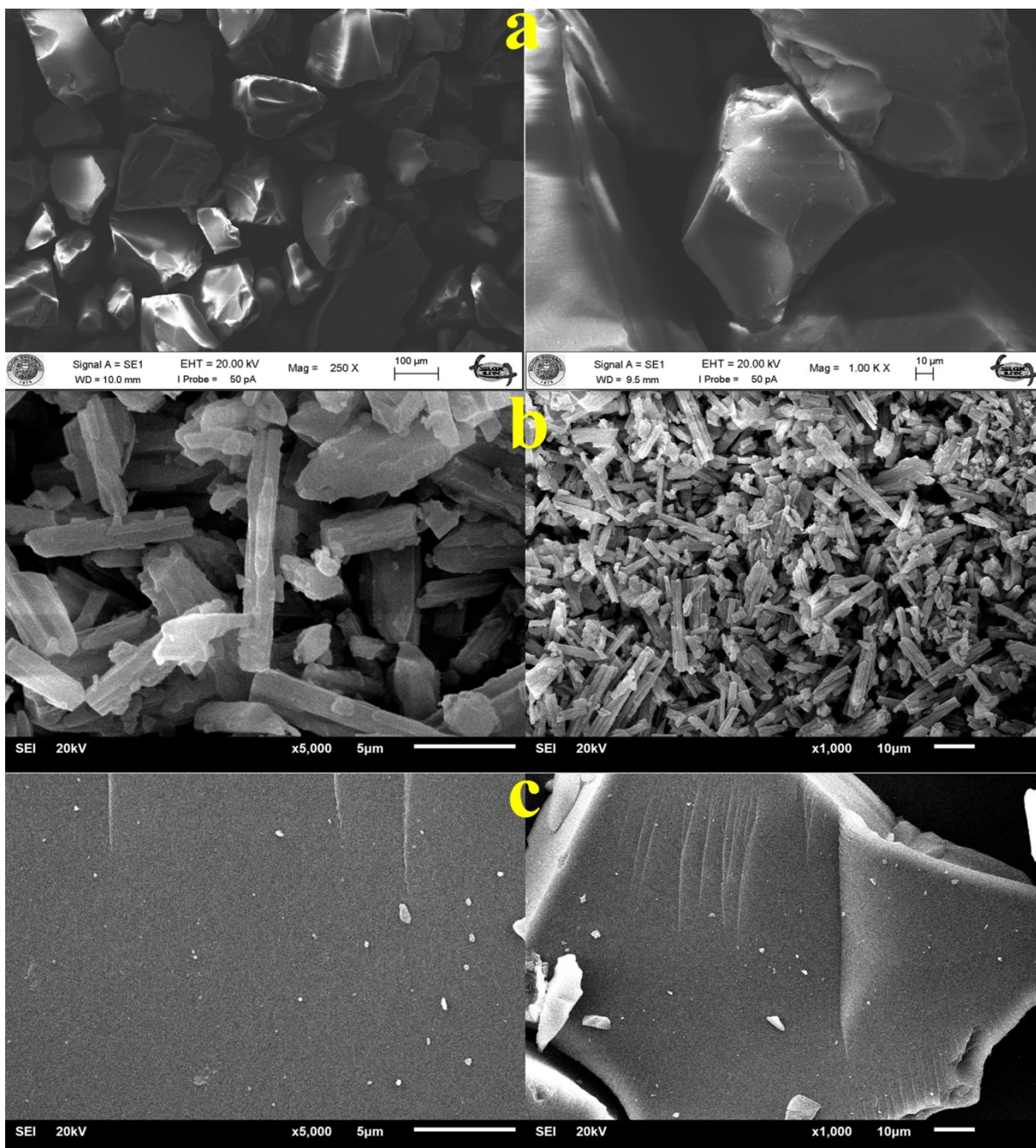


**Fig. S1.** The <sup>1</sup>H NMR spectra of calix-3.

<sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>): δ (ppm) 7.59 (s, 2H, OH), 7.11 (s, 4H, ArH), 6.95 (s, 4H, ArH), 6.79 (s, 2H, ArH), 6.70 (s, 2H, ArH), 4.79 (s, 4H, OCH<sub>2</sub>), 4.53 (d, 4H, ArCH<sub>2</sub>Ar), 3.91 (s, 6H, OCH<sub>3</sub>), 3.46 (d, 4H, ArCH<sub>2</sub>Ar). Anal. Calcd for C<sub>34</sub>H<sub>32</sub>O<sub>8</sub>: C, 71.82; H, 5.67; O, 22.51; Found: 72.91; H, 5.08; O, 22.01.

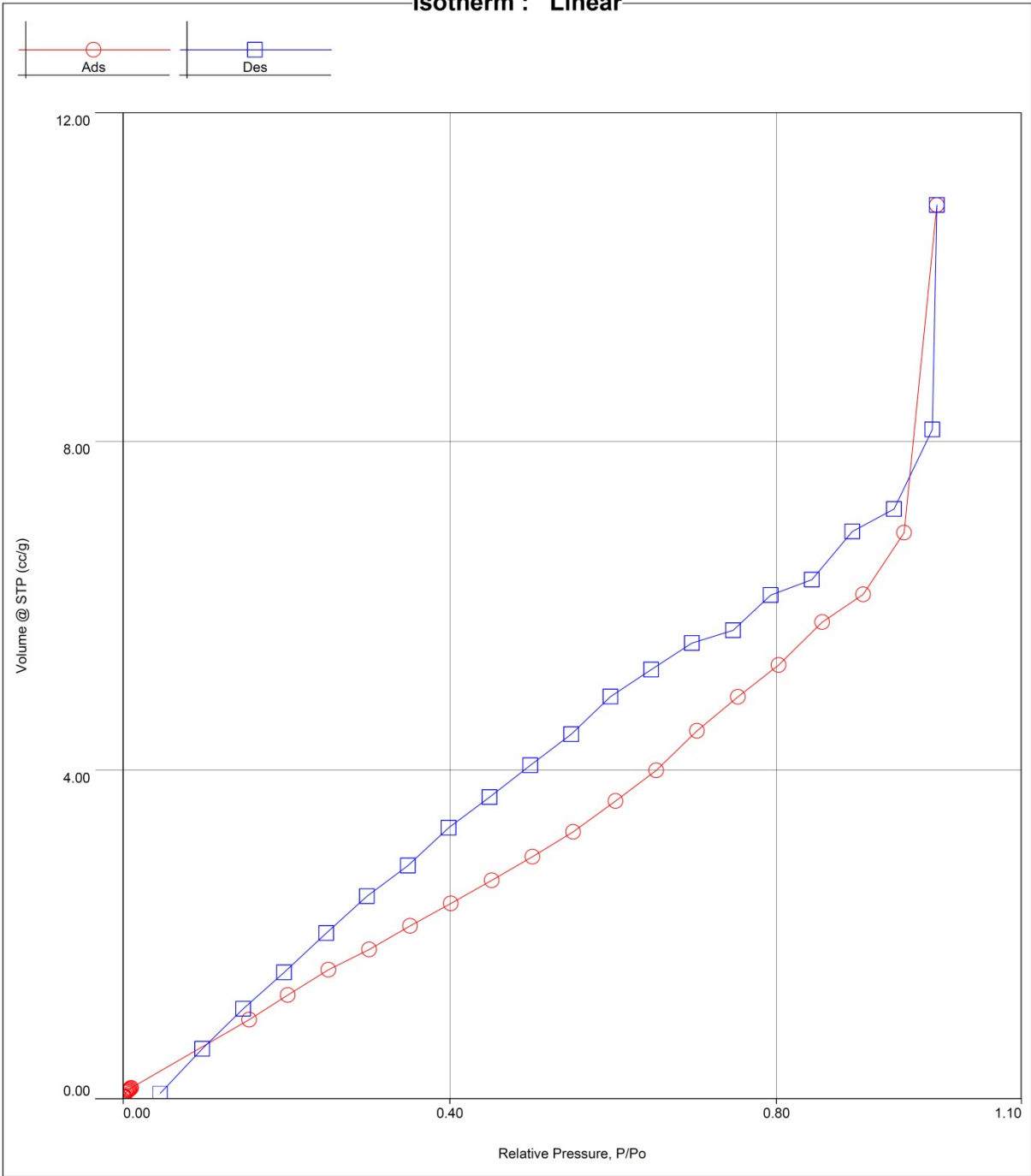


**Fig. S2.** EDS mapping of the calix-3 (a) and CBOS (b).

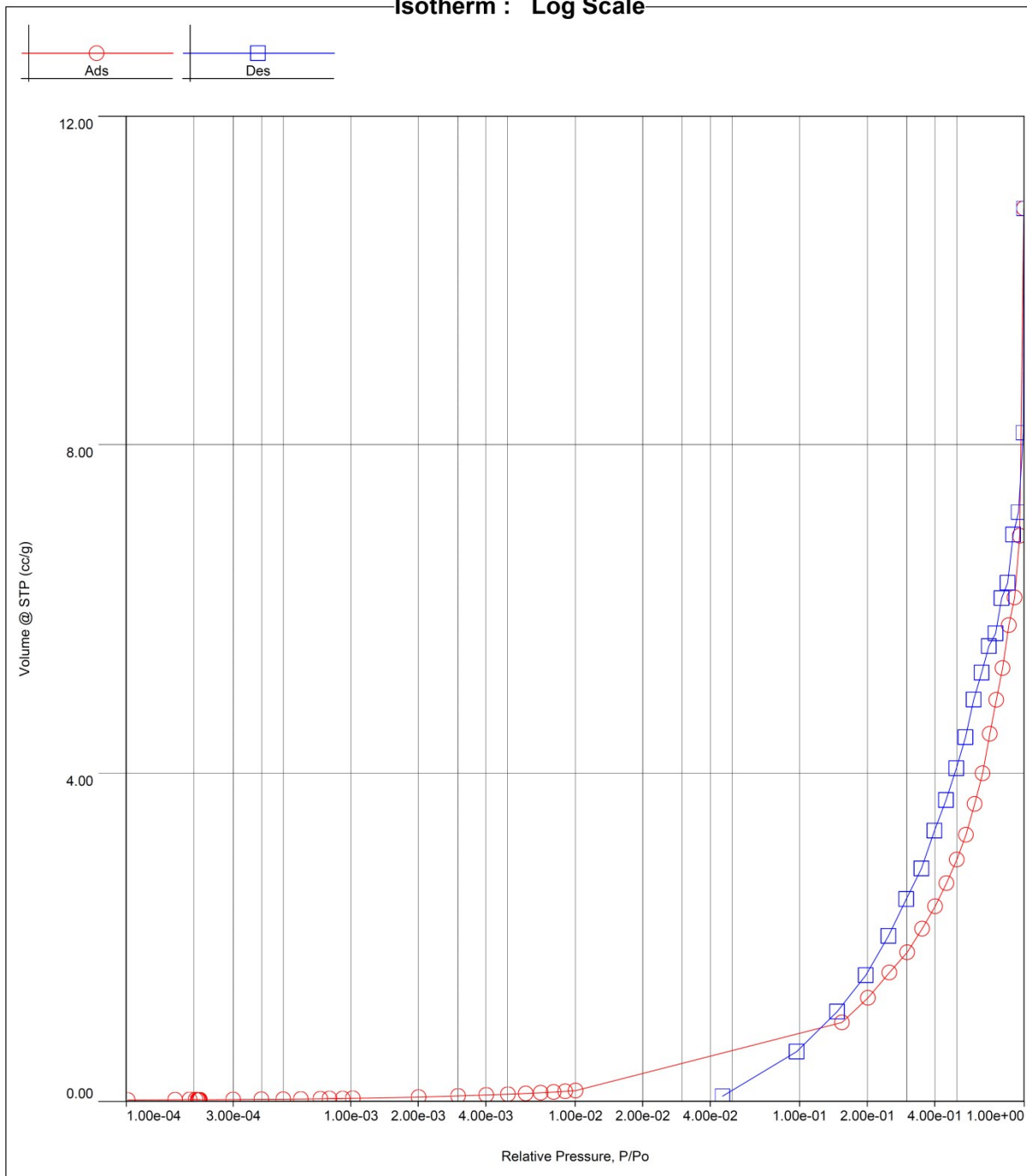


**Fig. S3.** SEM images of mesoporous silica (a), calix-3 (b), CBOS (c).

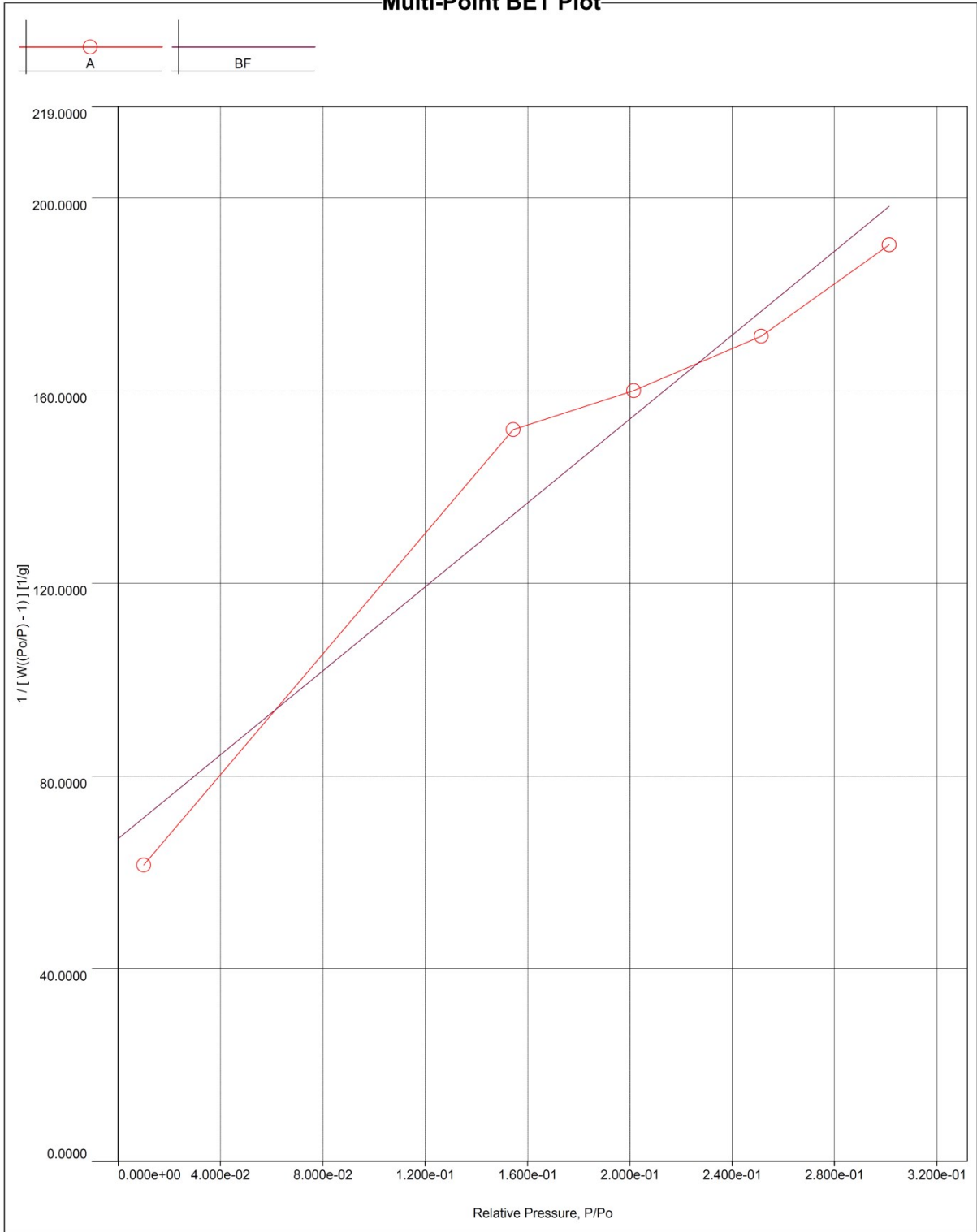
Isotherm : Linear



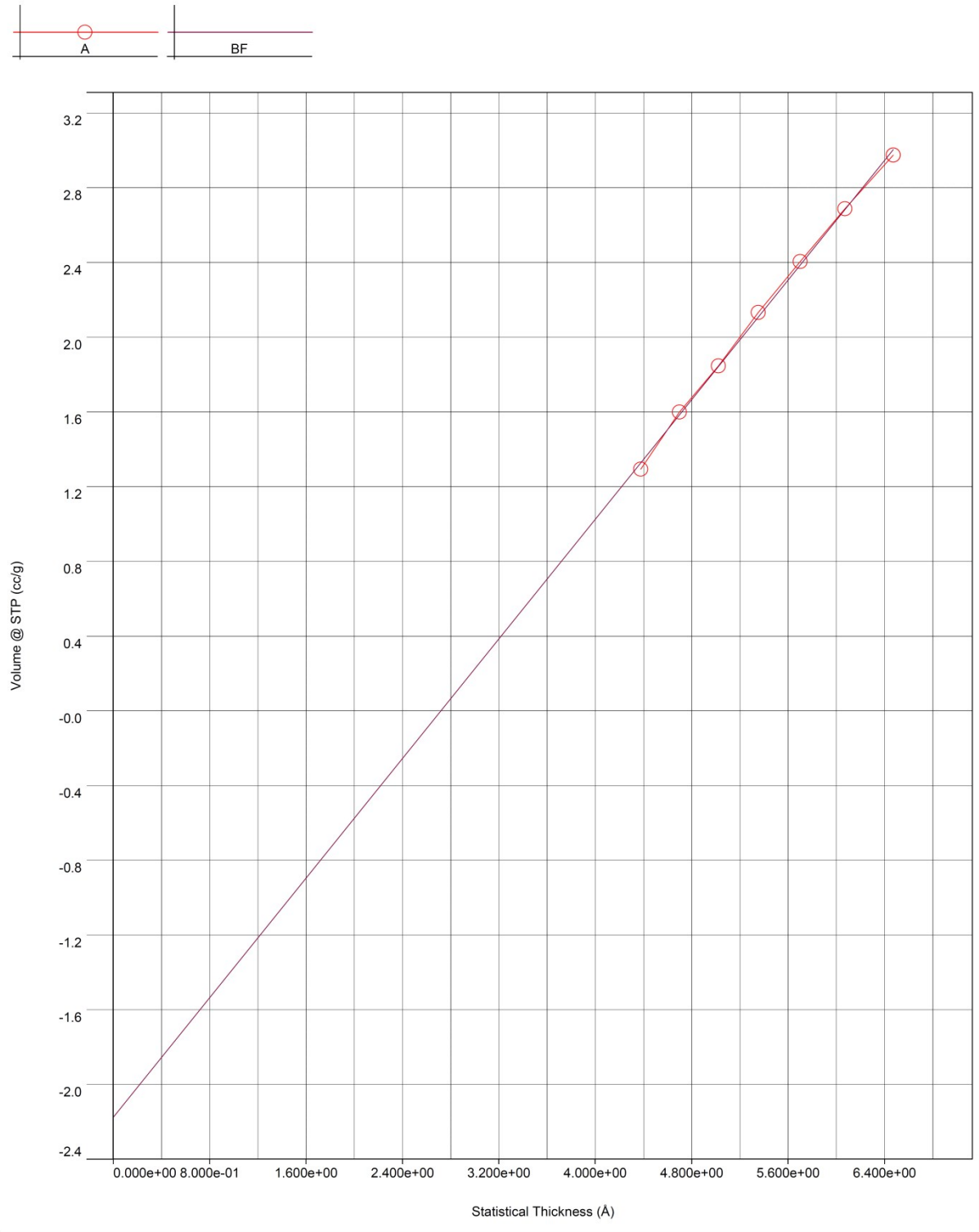
### Isotherm : Log Scale



# Multi-Point BET Plot

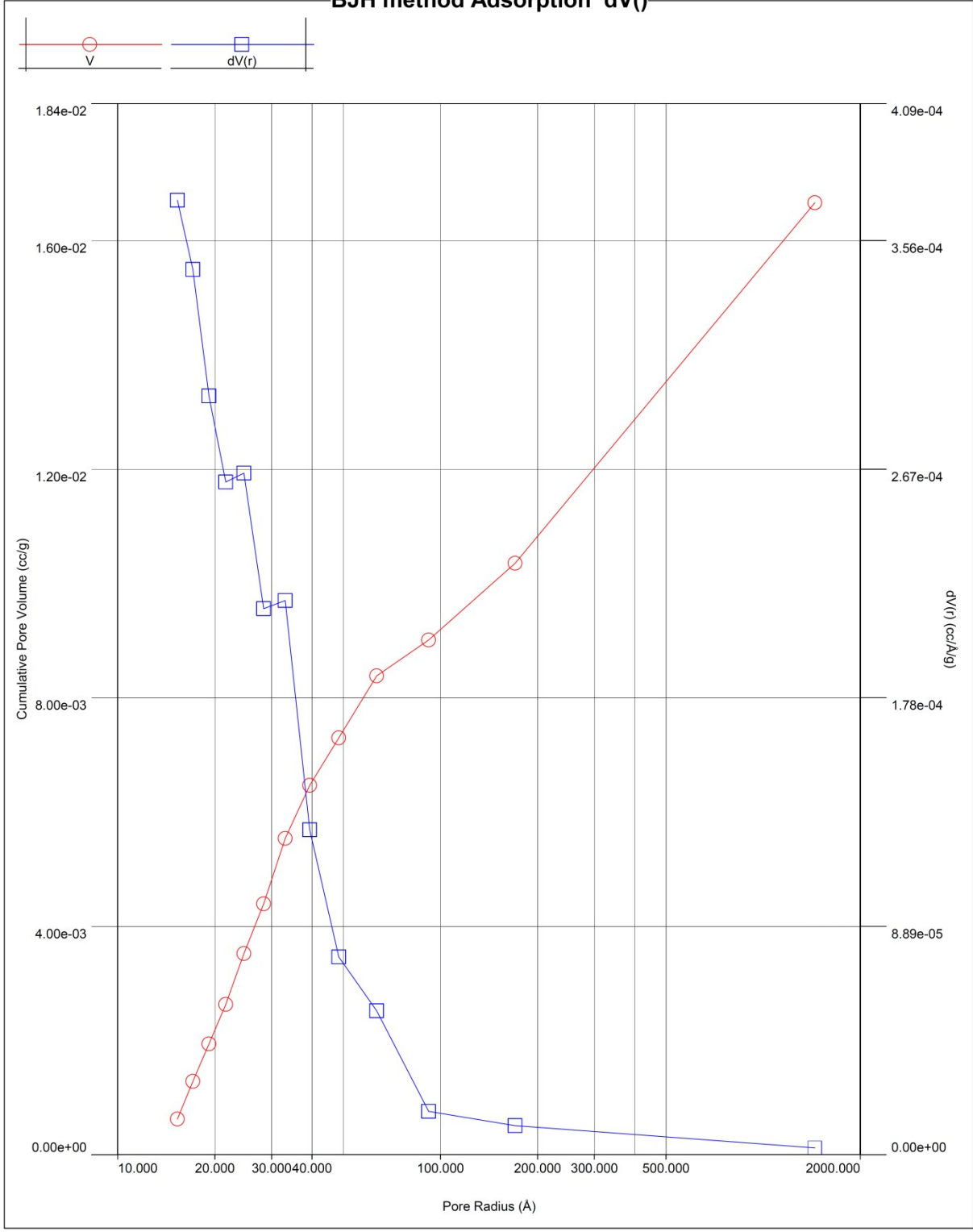


# t plot



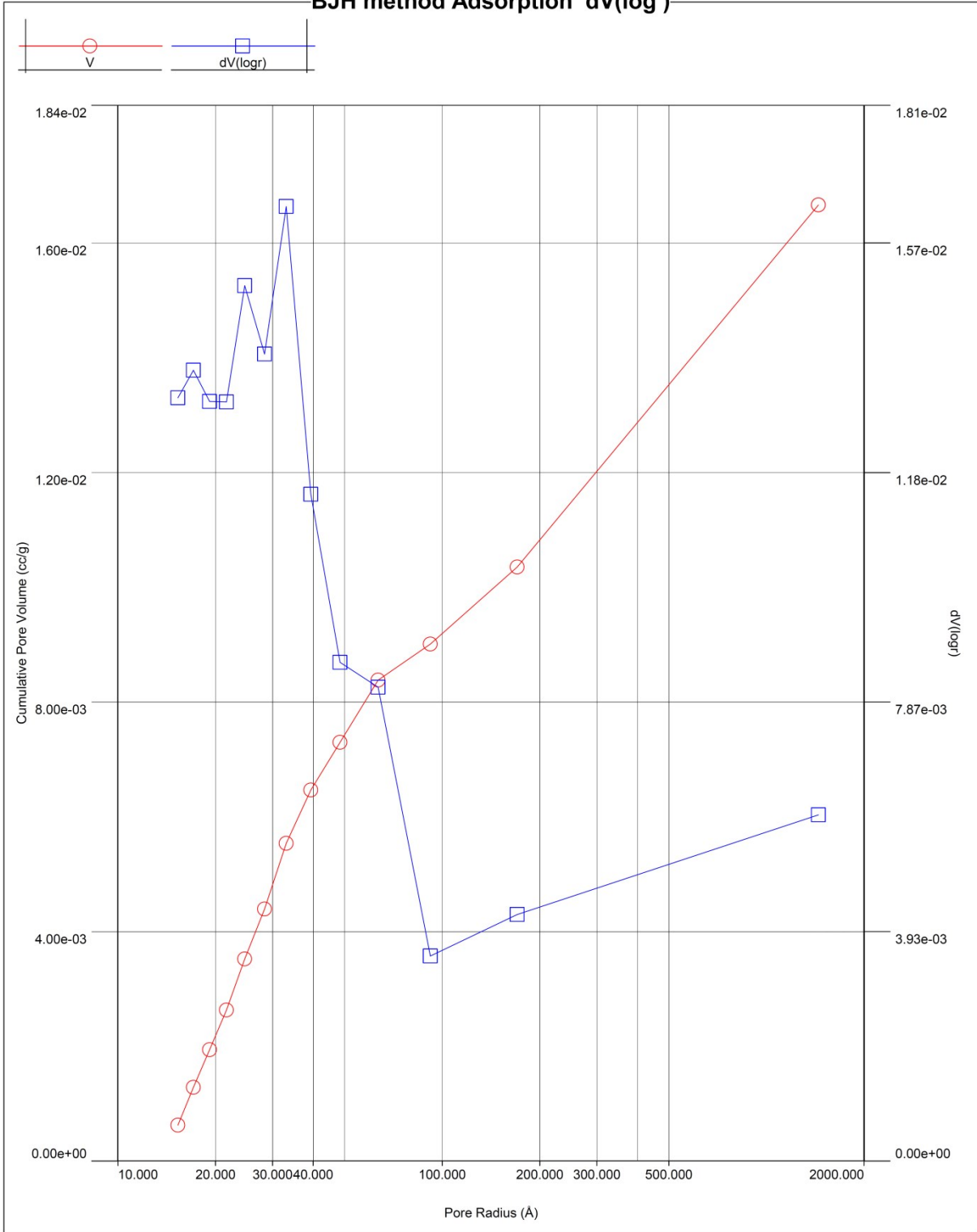


### BJH method Adsorption dV()

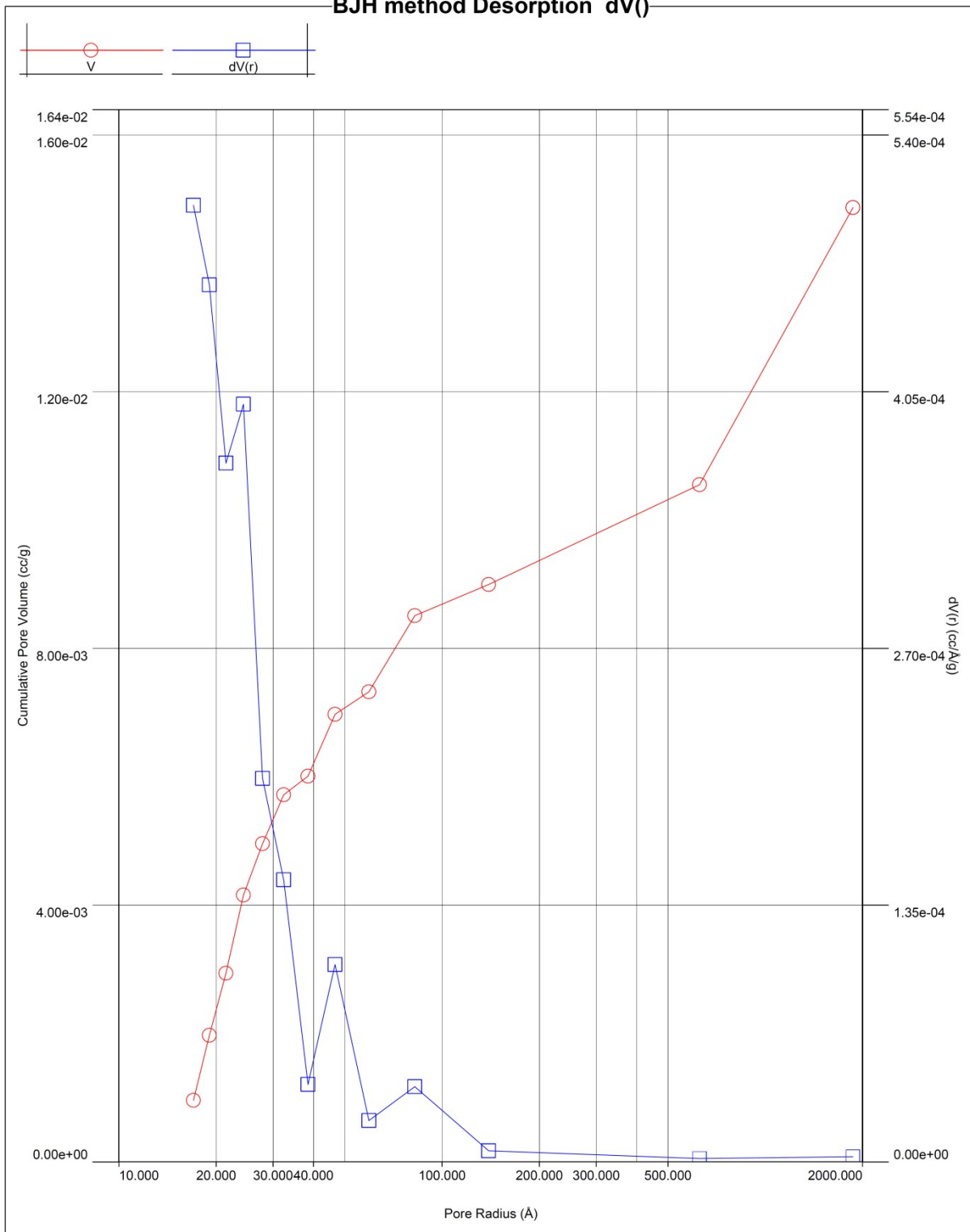




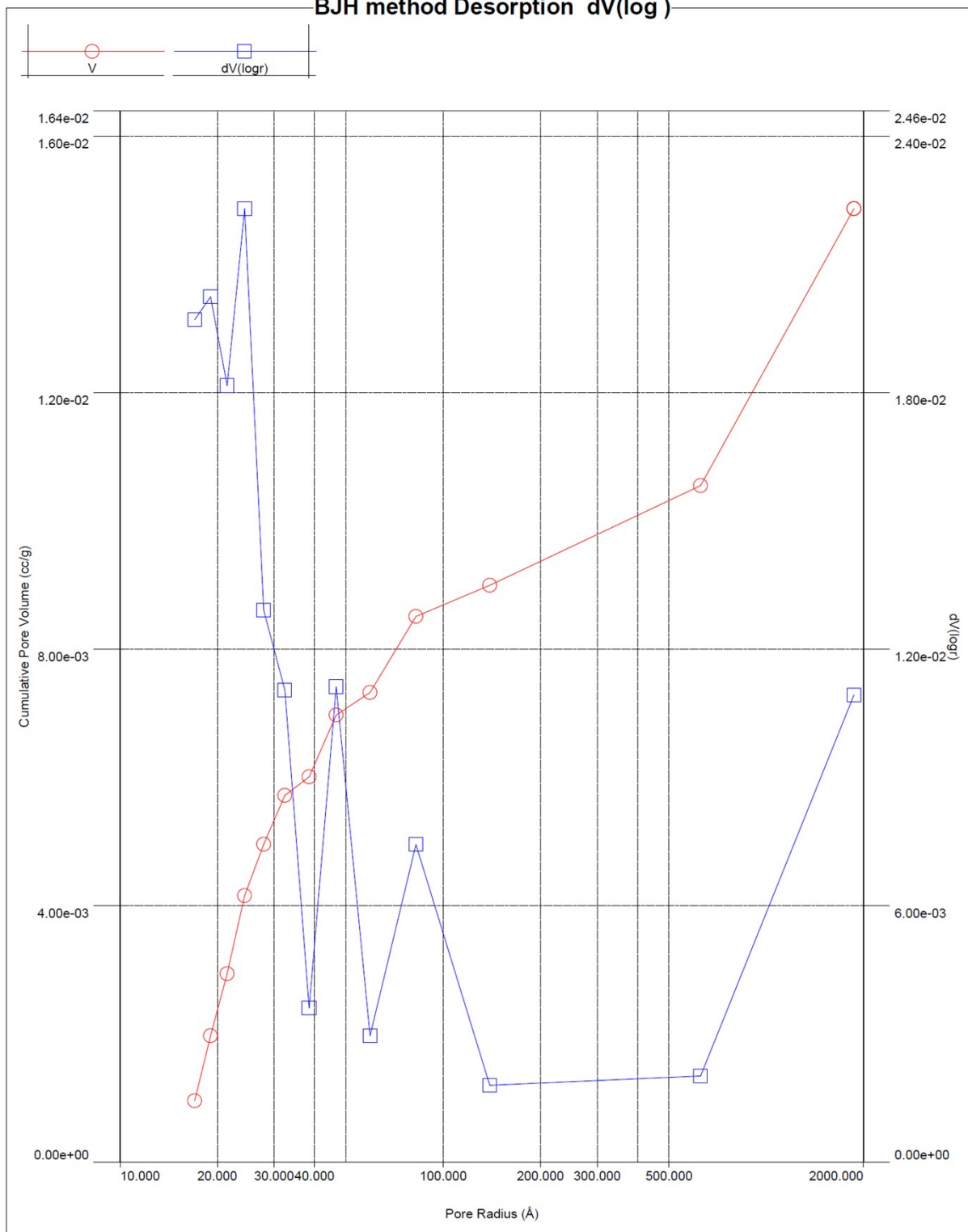
### BJH method Adsorption $dV(\log)$



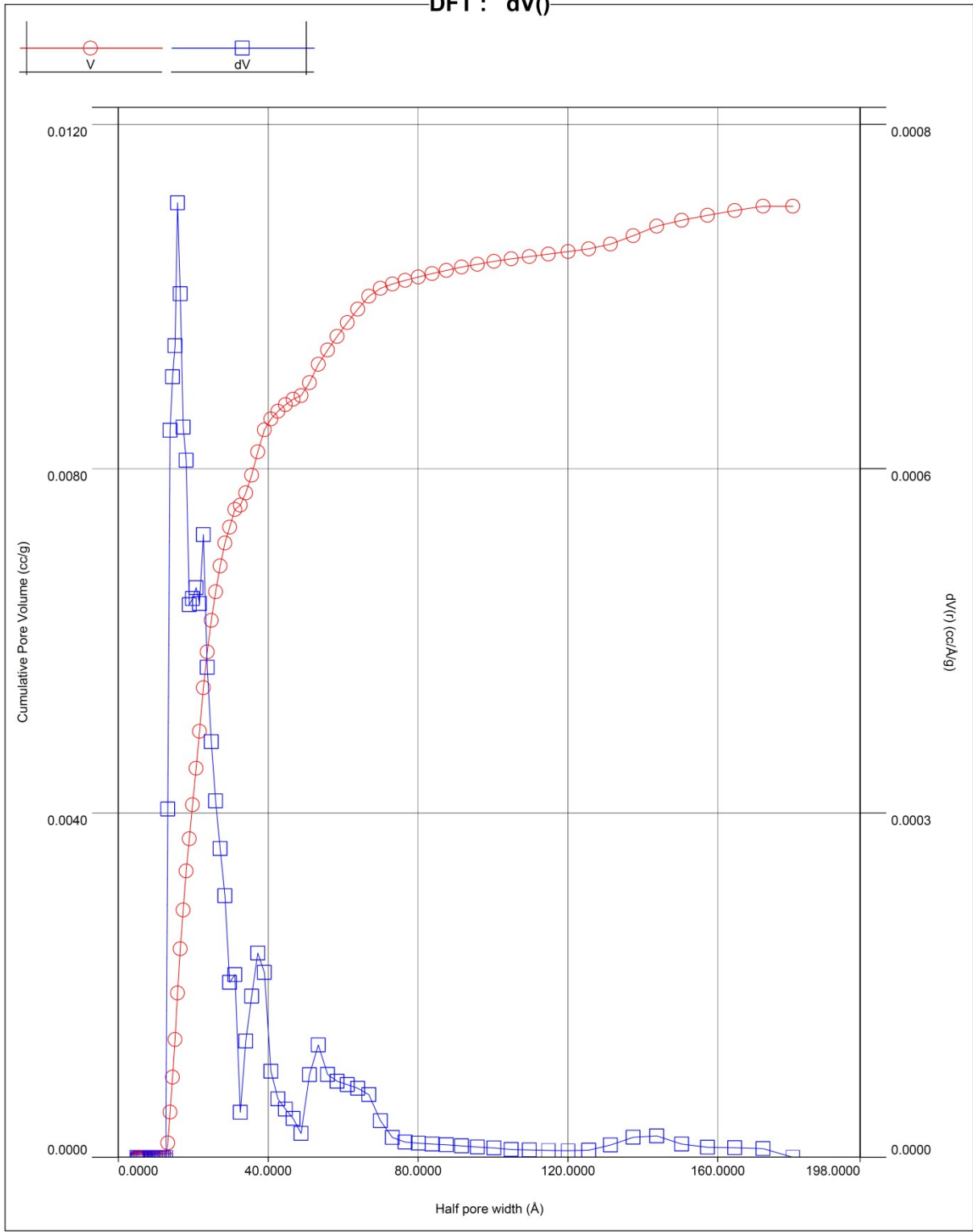
### BJH method Desorption dV()



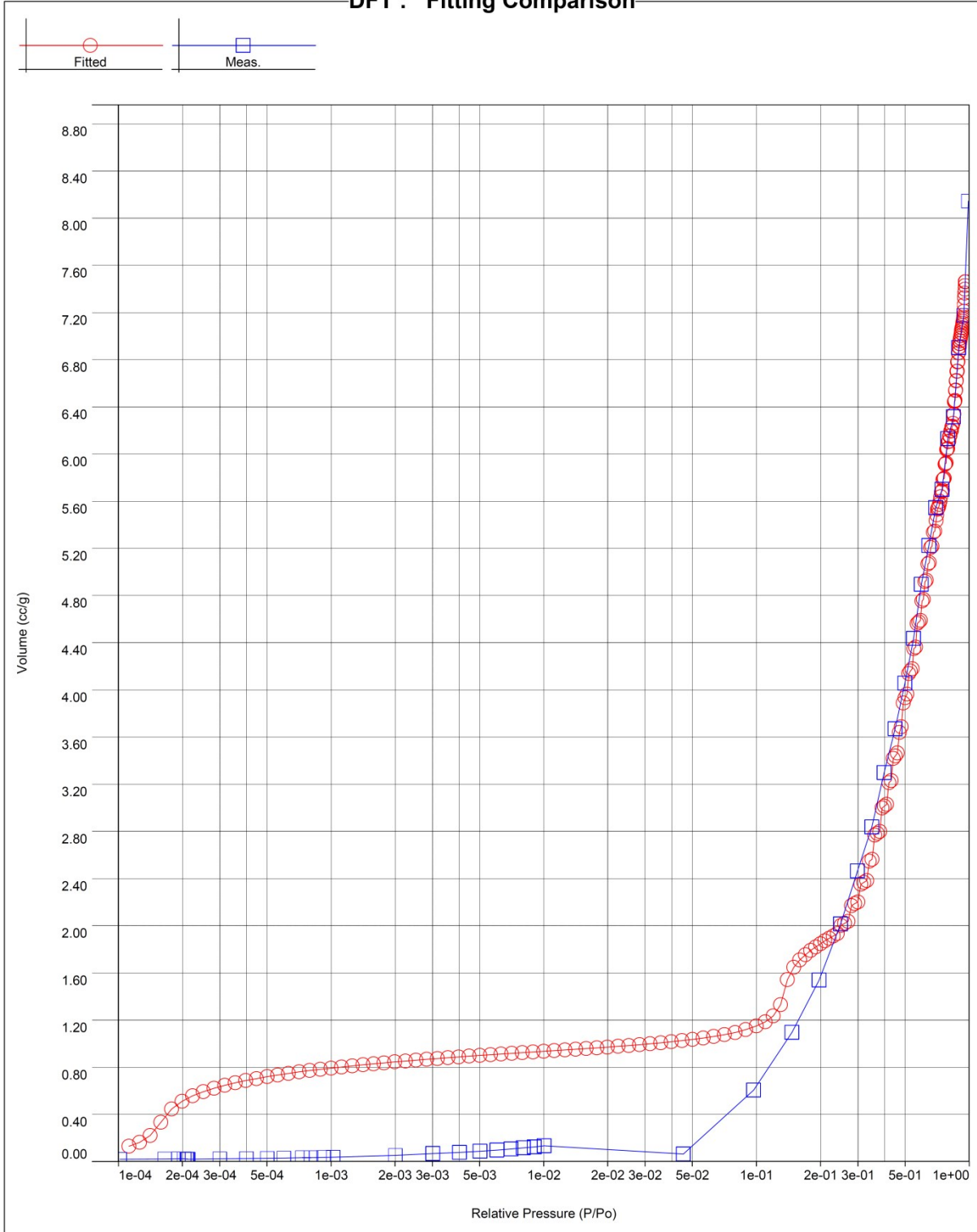
### BJH method Desorption $dV(\log)$



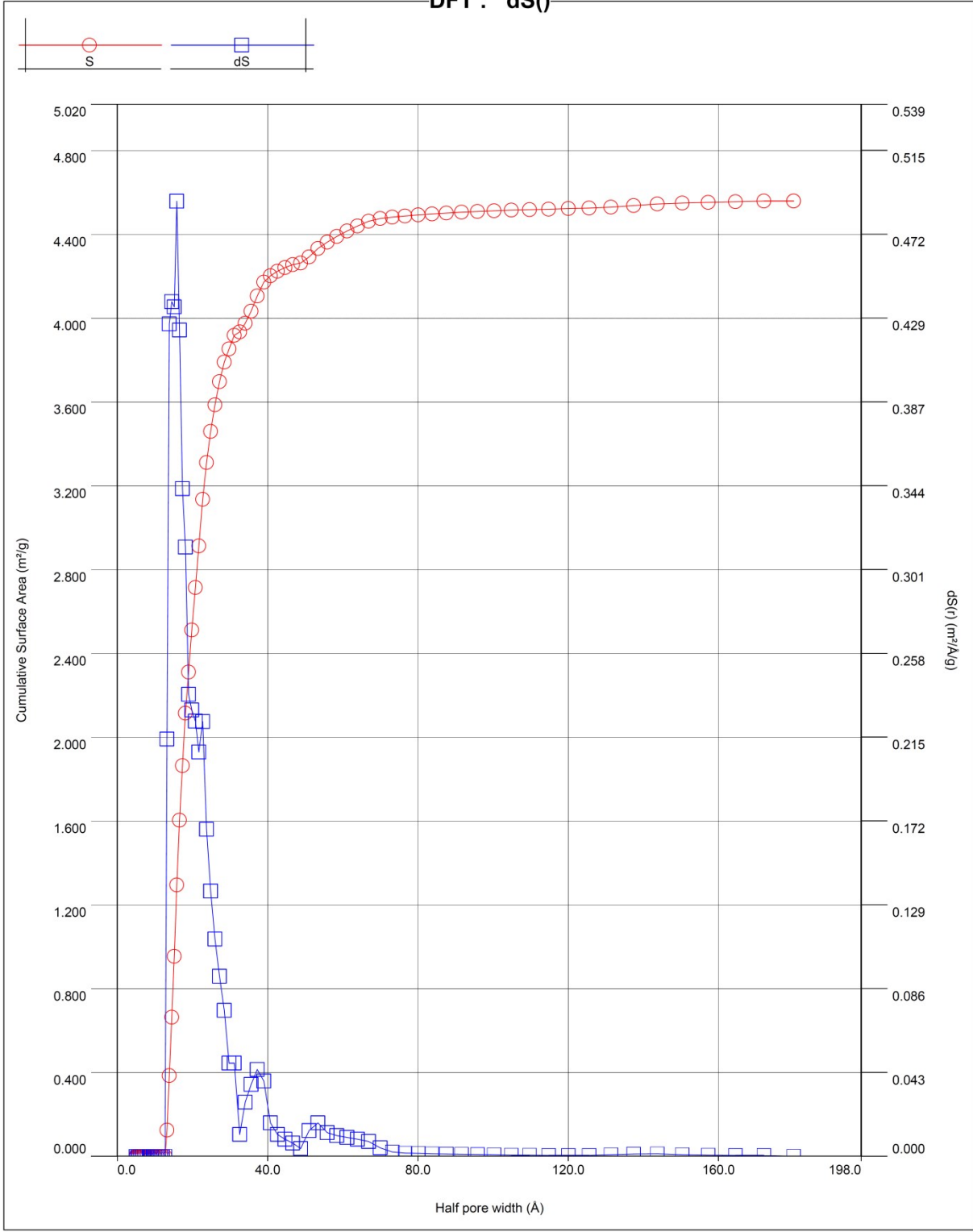
DFT : dV()



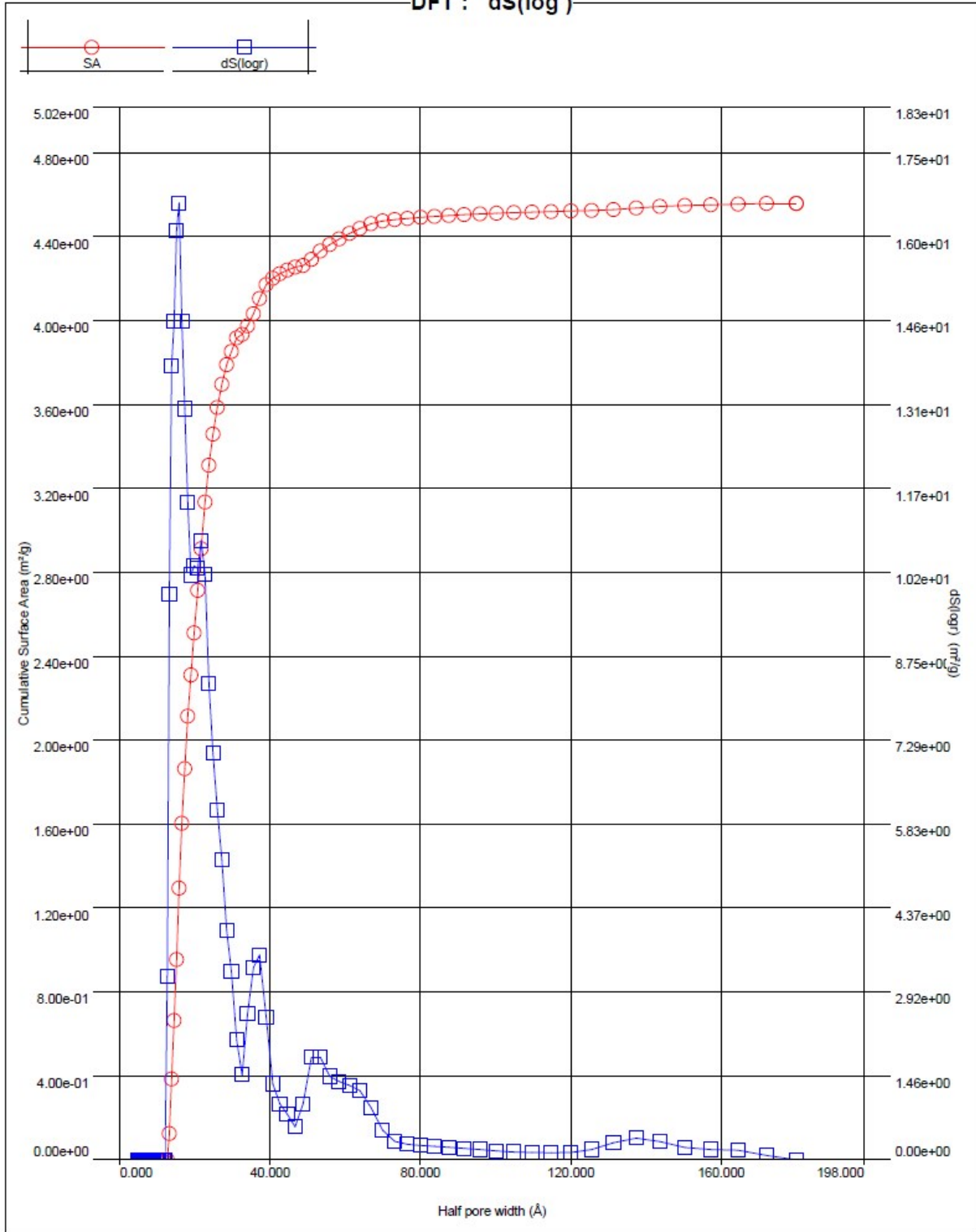
### DFT : Fitting Comparison



DFT : dS()



DFT : dS(log)





### Multi-Point BET

Relative Pressure [P/Po]	Volume @ STP [cc/g]	1 / [ W((Po/P) - 1) ] [1/g]	Relative Pressure [P/Po]	Volume @ STP [cc/g]	1 / [ W((Po/P) - 1) ] [1/g]
1.00529e-02	0.1321	6.1507e+01	2.51416e-01	1.5687	1.7131e+02
1.54432e-01	0.9619	1.5191e+02	3.01487e-01	1.8150	1.9027e+02
2.01540e-01	1.2624	1.5998e+02			

### MBET summary

Slope =	435.370 1/g
Intercept =	6.698e+01 1/g
Correlation coefficient, r =	0.973157
C constant =	7.500
Surface Area =	6.932 m <sup>2</sup> /g

### Total Pore Volume data

<b>Total Pore Volume</b>
Total pore volume = 1.682e-02 cc/g for pores smaller than 2660.8 Å (Radius) at P/Po = 0.99639

### Average PoreSize data

Average pore Radius =	4.85356e+01 Å
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### Single Point Surface Area

Relative Pressure [P/Po]	Volume @ STP [cc/g]	1 / [ W((P/Po) - 1) ]	Slope	Surf. Area [m <sup>2</sup> /g]
3.01499e-03	0.0665	3.6394e+01	12071.1264	0.2885
4.02271e-03	0.0774	4.1728e+01	10373.0756	0.3357
5.02977e-03	0.0873	4.6344e+01	9214.0394	0.3780
6.02829e-03	0.0958	5.0664e+01	8404.3486	0.4144
7.02700e-03	0.1053	5.3760e+01	7650.5087	0.4552
8.02484e-03	0.1151	5.6219e+01	7005.6147	0.4971
9.03171e-03	0.1232	5.9168e+01	6551.1703	0.5316
1.00529e-02	0.1321	6.1507e+01	6118.3167	0.5692
1.54432e-01	0.9619	1.5191e+02	983.7030	3.5402
2.01540e-01	1.2624	1.5998e+02	793.7939	4.3872
2.51416e-01	1.5687	1.7131e+02	681.3636	5.1111
3.01487e-01	1.8150	1.9027e+02	631.1063	5.5181
3.51471e-01	2.1020	2.0629e+02	586.9313	5.9334
4.01494e-01	2.3740	2.2609e+02	563.1194	6.1843
4.51354e-01	2.6569	2.4774e+02	548.8922	6.3446
5.01473e-01	2.9445	2.7334e+02	545.0750	6.3891

### Thickness summary

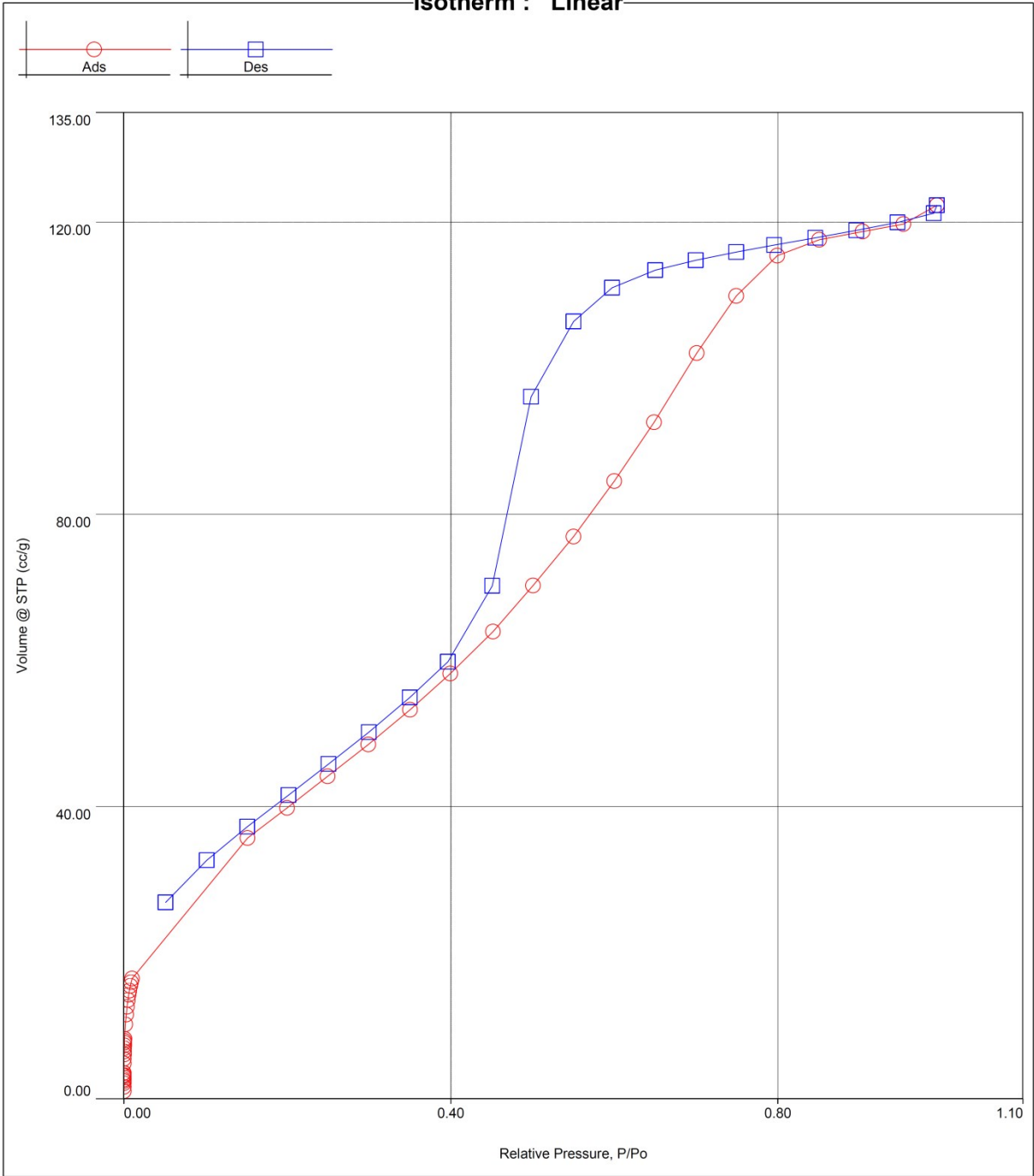
Thickness method: DeBoer	
Slope =	0.800
Intercept =	-2.206
Correlation coefficient, r =	0.999237

### V-t method summary

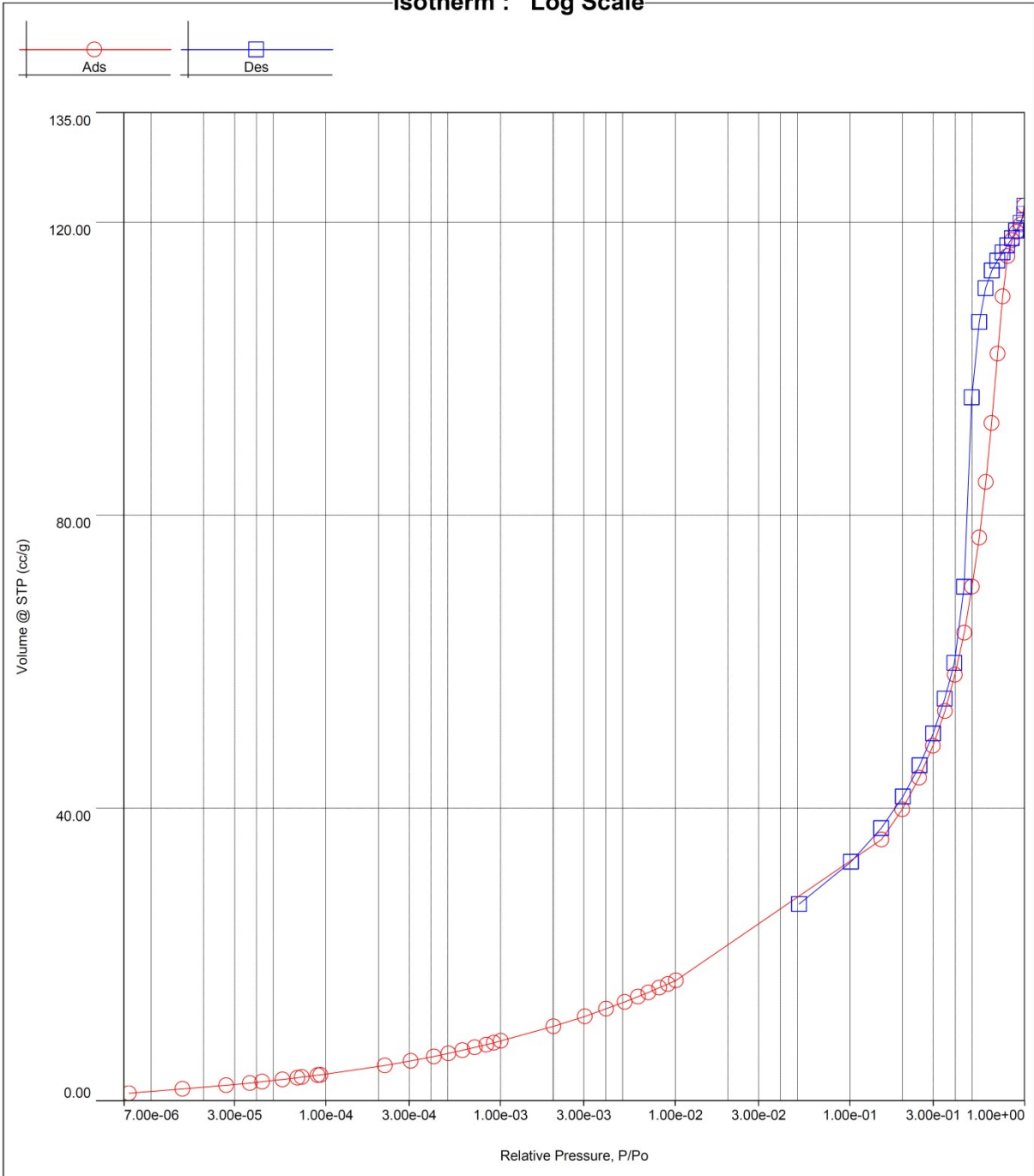
Thickness method: DeBoer	
Slope =	0.800
Intercept =	-2.206
Correlation coefficient, r =	0.999237
Micropore volume =	0.000 cc/g
Micropore area =	0.000 m <sup>2</sup> /g
External surface area =	6.932 m <sup>2</sup> /g

Fig. S4. The BET plot and analysis data for calix-3 precursor.

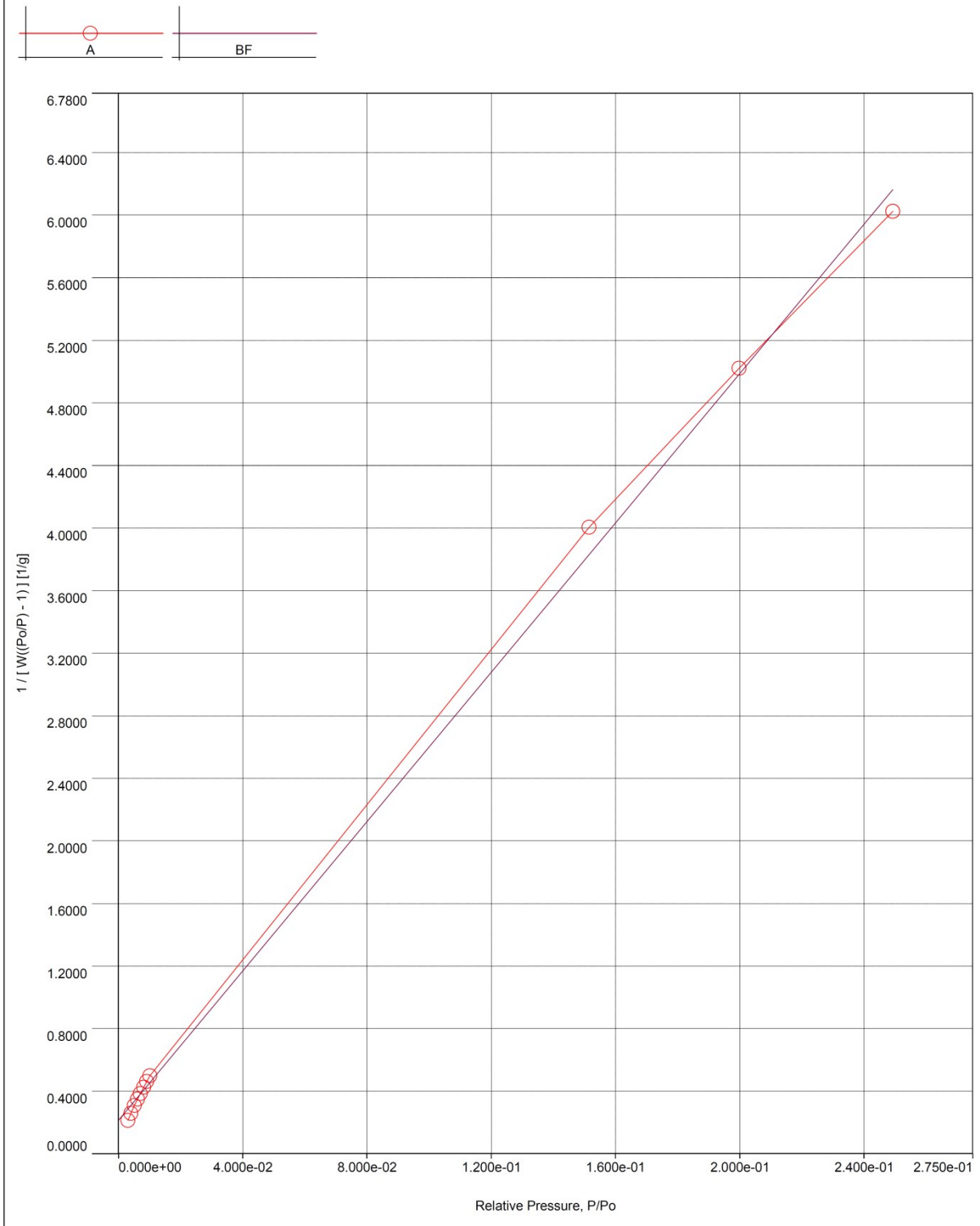
Isotherm : Linear



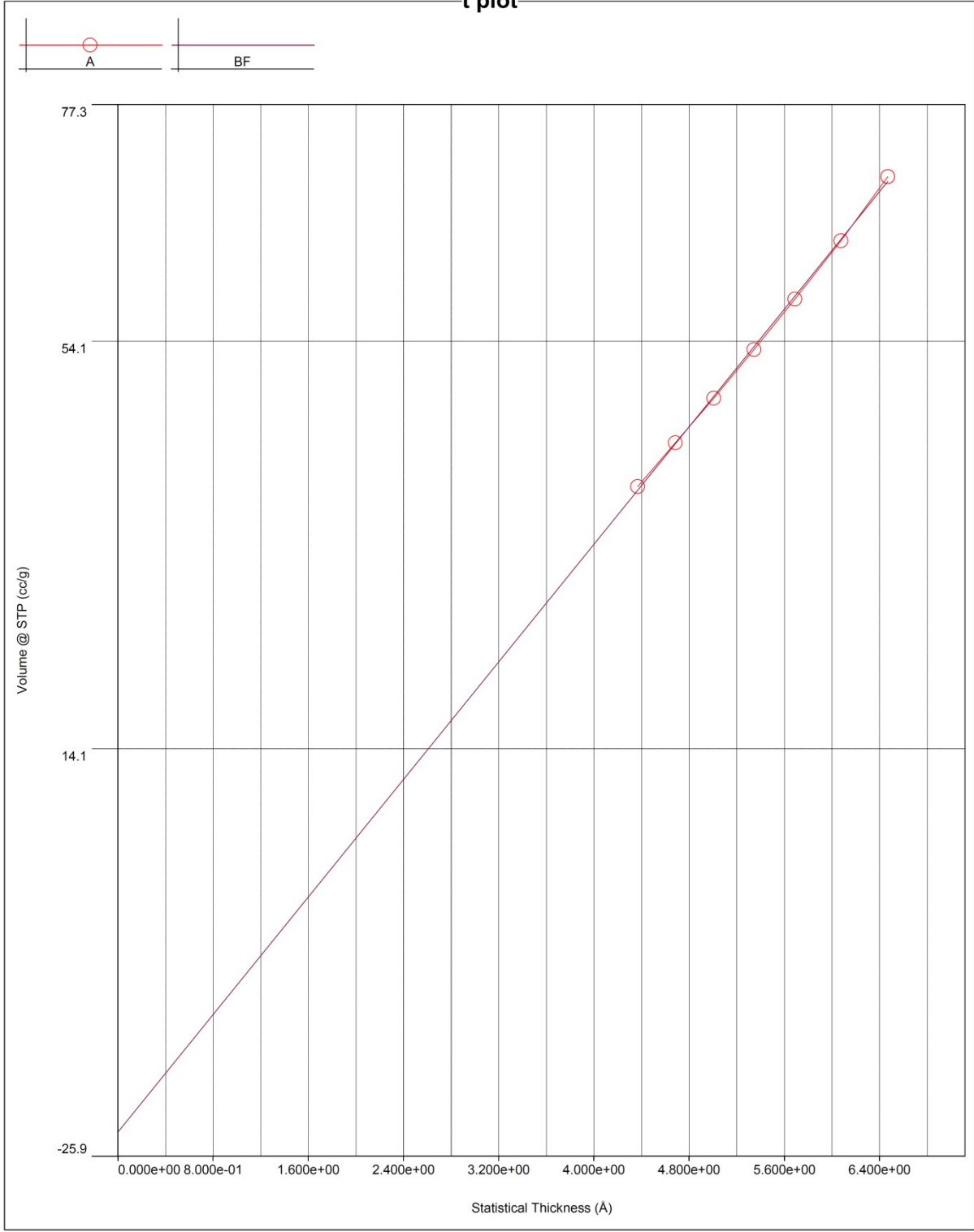
### Isotherm : Log Scale



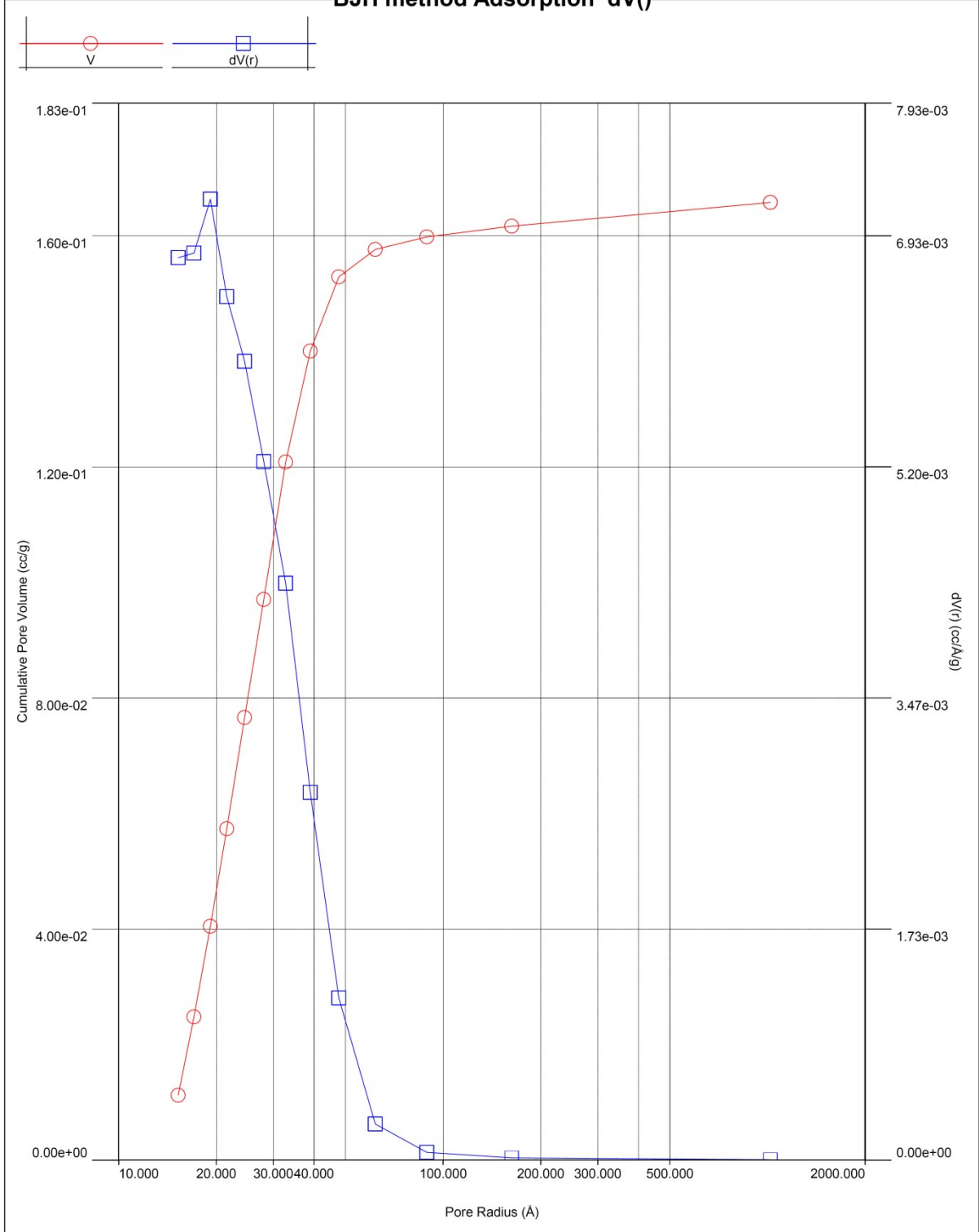
### Multi-Point BET Plot



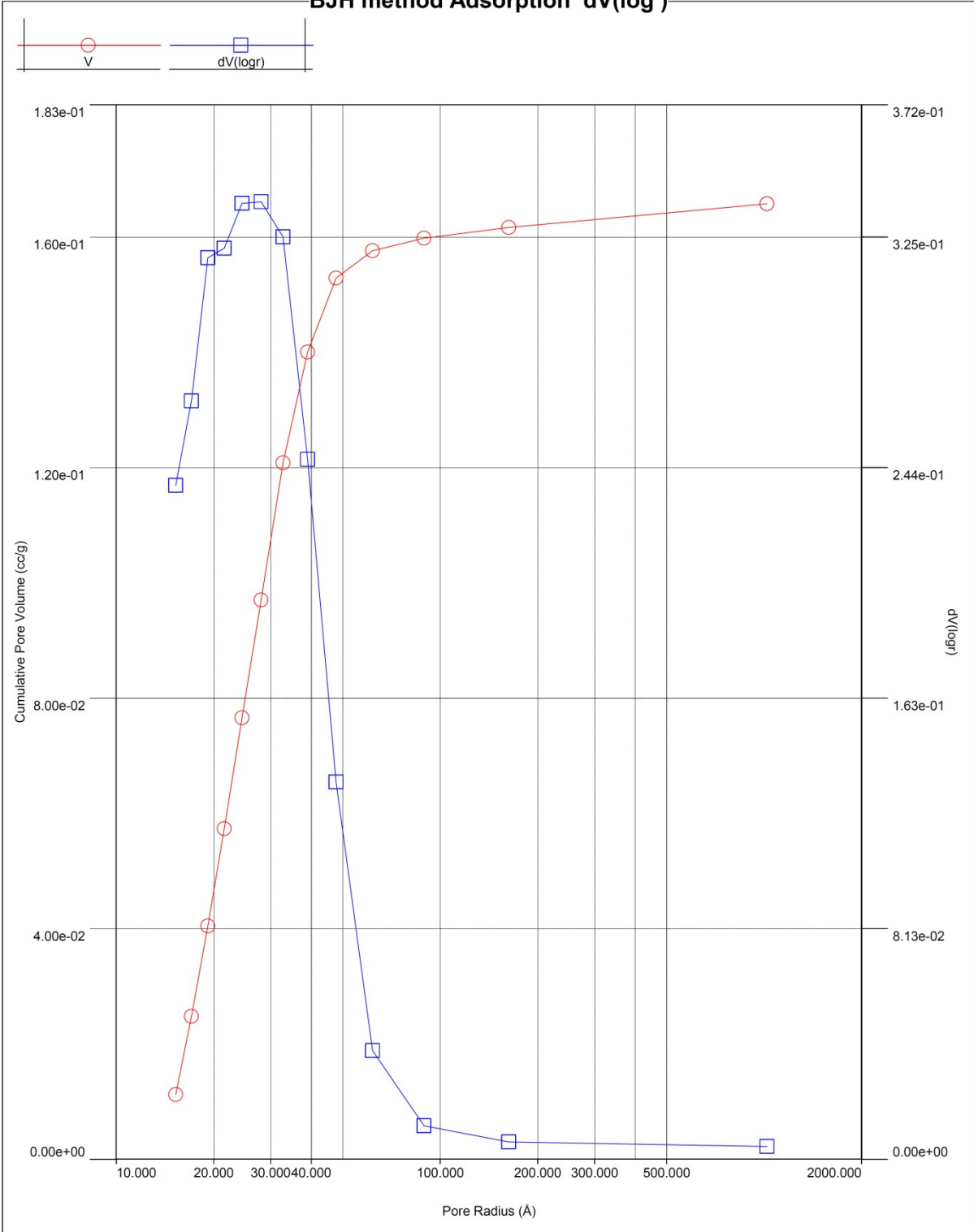
# t plot



### BJH method Adsorption dV()

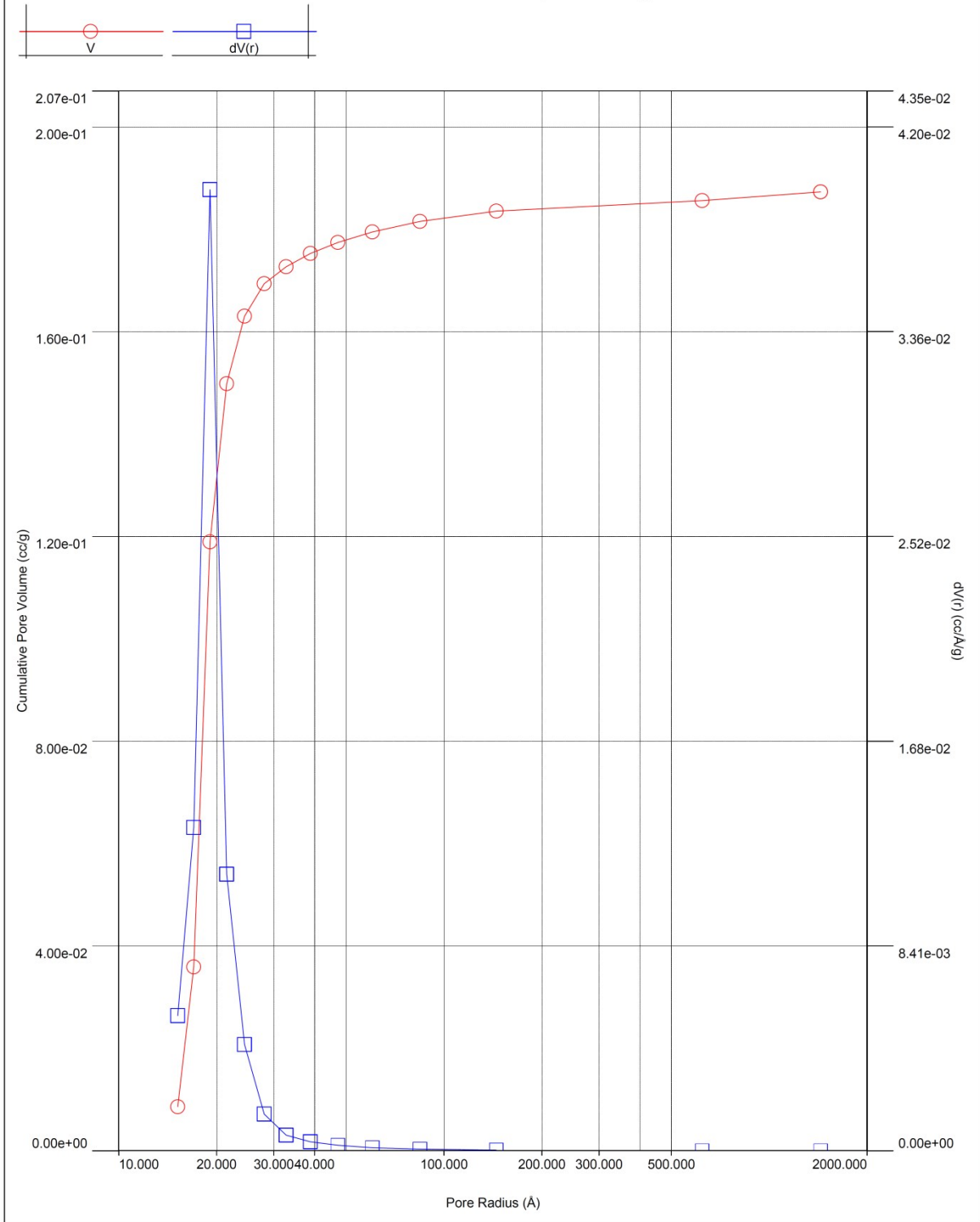


### BJH method Adsorption dV(log)

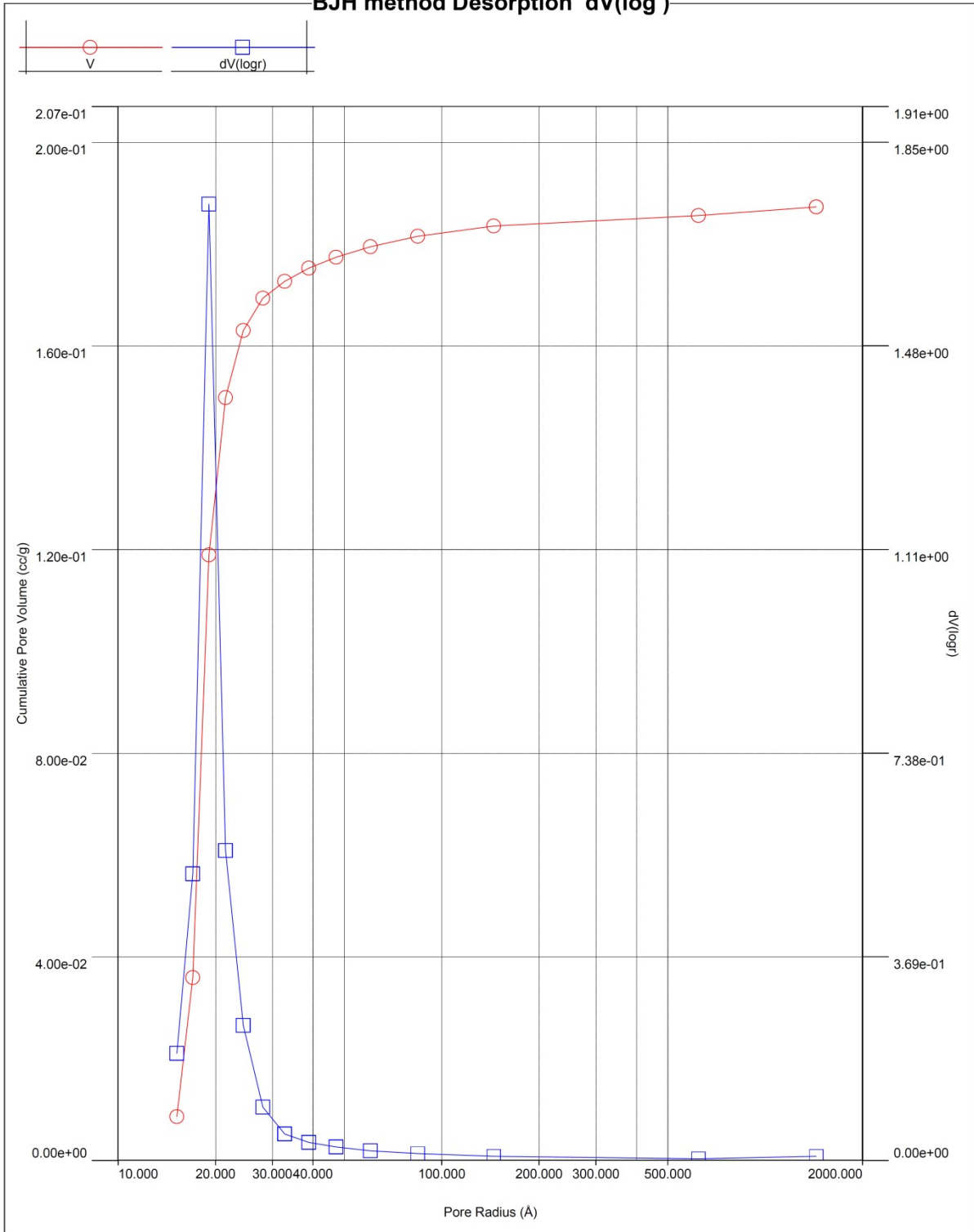




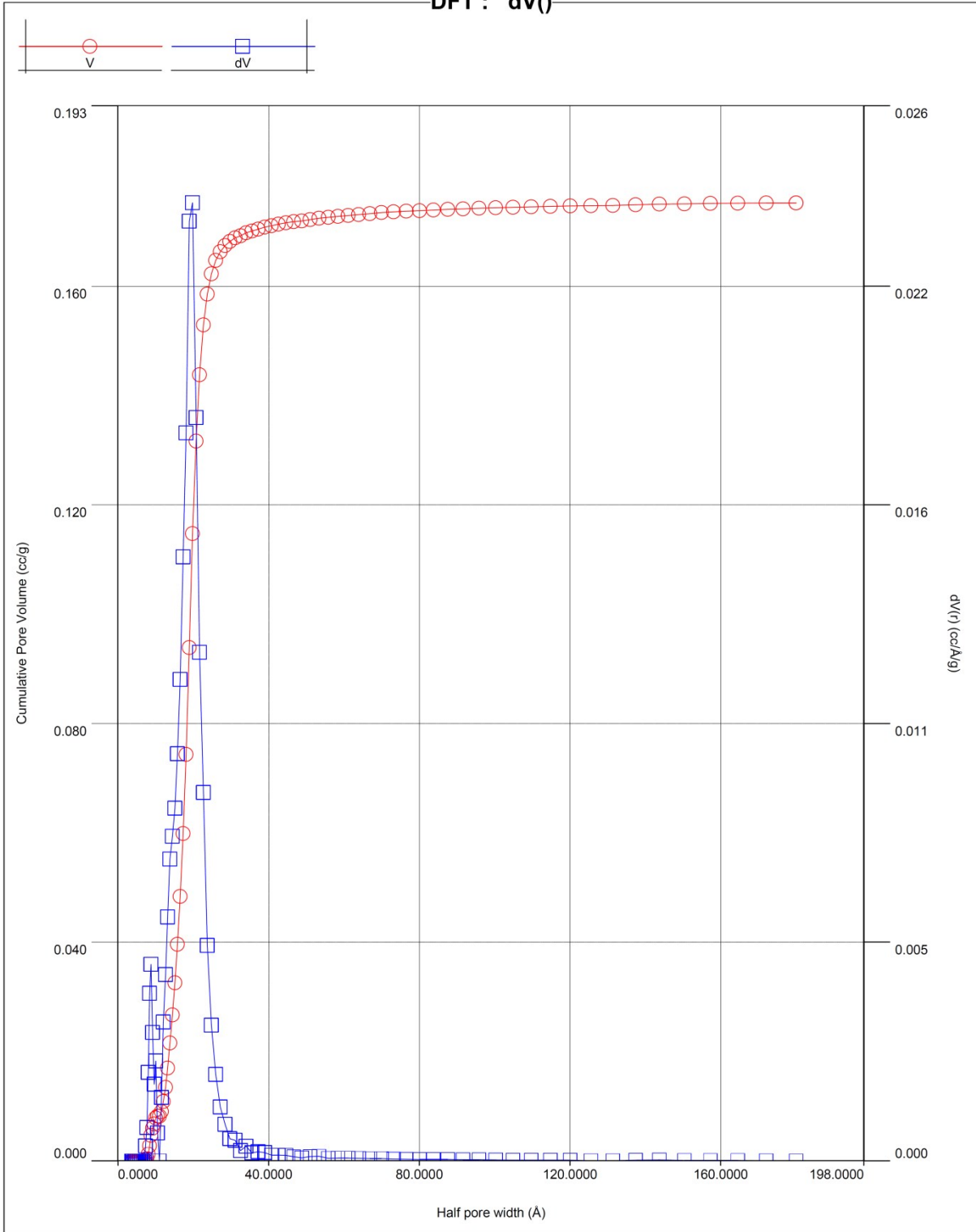
### BJH method Desorption dV()



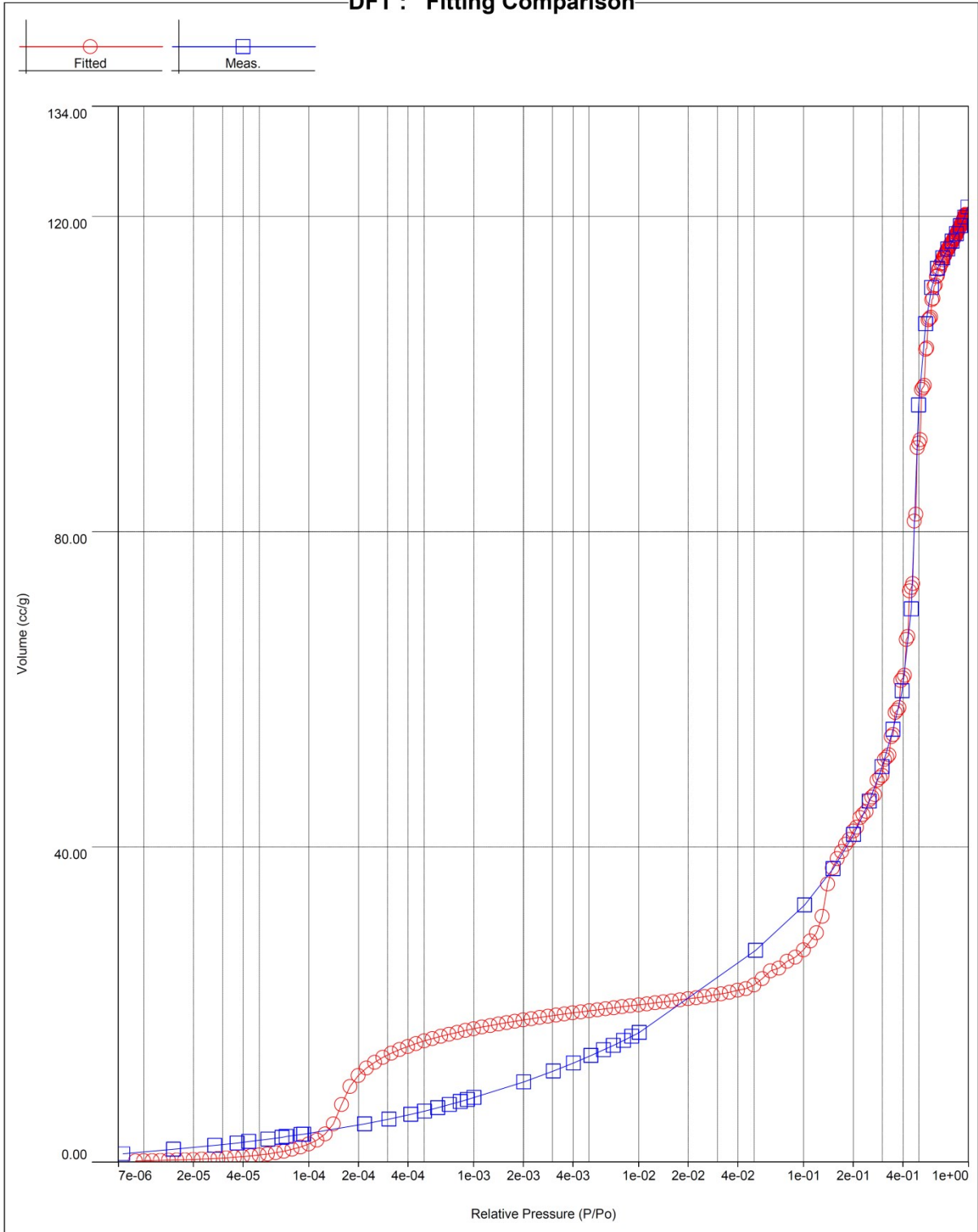
### BJH method Desorption dV(log)



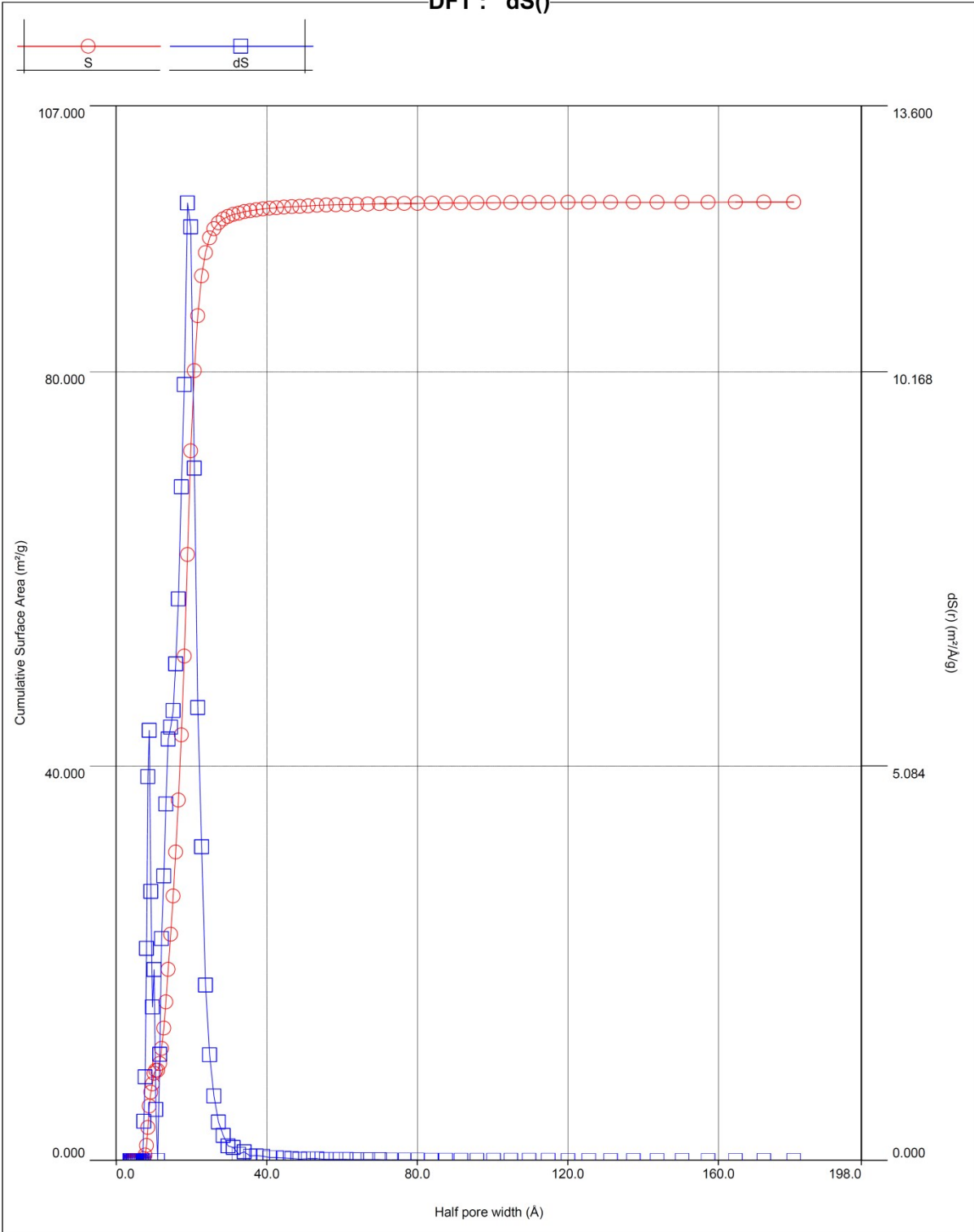
DFT : dV()



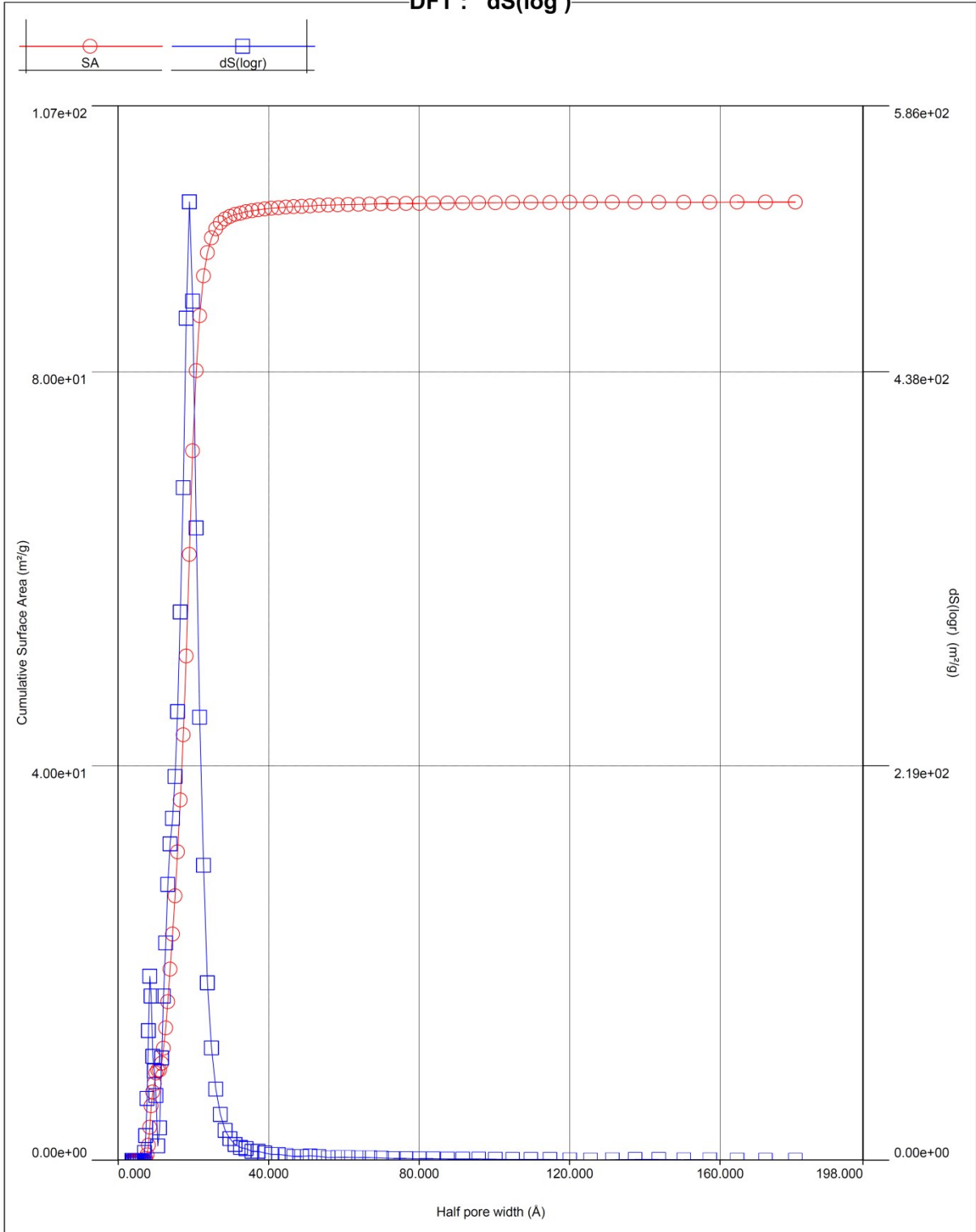
### DFT : Fitting Comparison



DFT : dS()



DFT : dS(logr)



### Multi-Point BET

Relative Pressure [P/Po]	Volume @ STP [cc/g]	1 / [ W((Po/P) - 1) ] [1/g]	Relative Pressure [P/Po]	Volume @ STP [cc/g]	1 / [ W((Po/P) - 1) ] [1/g]
3.04515e-03	11.5254	2.1204e-01	9.09683e-03	15.9404	4.6080e-01
4.03642e-03	12.5579	2.5822e-01	1.01236e-02	16.4313	4.9800e-01
5.14660e-03	13.4983	3.0664e-01	1.51489e-01	35.6733	4.0043e+00
6.13671e-03	14.2184	3.4746e-01	1.99807e-01	39.7915	5.0208e+00
7.02144e-03	14.7876	3.8259e-01	2.49312e-01	44.1070	6.0246e+00
8.12461e-03	15.4301	4.2474e-01			

### MBET summary

Slope = 23.865 1/g  
 Intercept = 2.135e-01 1/g  
 Correlation coefficient, r = 0.999354  
 C constant = 112.785  
 Surface Area = 144.633 m<sup>2</sup>/g

### Total Pore Volume data

#### Total Pore Volume

Total pore volume = 1.891e-01 cc/g for  
 pores smaller than 1825.8 Å (Radius)  
 at P/Po = 0.99473

### Average PoreSize data

Average pore Radius = 2.61543e+01 Å

### Single Point Surface Area

Relative Pressure [P/Po]	Volume @ STP [cc/g]	1 / [ W((P/Po) - 1) ]	Slope	Surf. Area [m <sup>2</sup> /g]
3.04515e-03	11.5254	2.1204e-01	69.6337	50.0120
4.03642e-03	12.5579	2.5822e-01	63.9722	54.4380
5.14660e-03	13.4983	3.0664e-01	59.5814	58.4497
6.13671e-03	14.2184	3.4746e-01	56.6205	61.5062
7.02144e-03	14.7876	3.8259e-01	54.4895	63.9117
8.12461e-03	15.4301	4.2474e-01	52.2787	66.6145
9.09683e-03	15.9404	4.6080e-01	50.6546	68.7503
1.01236e-02	16.4313	4.9800e-01	49.1923	70.7940
1.51489e-01	35.6733	4.0043e+00	26.4332	131.7479
1.99807e-01	39.7915	5.0208e+00	25.1285	138.5886
2.49312e-01	44.1070	6.0246e+00	24.1648	144.1154
2.99142e-01	48.4679	7.0460e+00	23.5541	147.8520
3.50211e-01	53.2253	8.1020e+00	23.1345	150.5337
3.99716e-01	58.1846	9.1567e+00	22.9079	152.0224
4.51855e-01	63.9190	1.0319e+01	22.8362	152.4997
5.00869e-01	70.2177	1.1434e+01	22.8291	152.5472

### Thickness summary

Thickness method: DeBoer  
 Slope = 14.414  
 Intercept = -23.515  
 Correlation coefficient, r = 0.999575

### V-t method summary

Thickness method: DeBoer  
 Slope = 14.414  
 Intercept = -23.515  
 Correlation coefficient, r = 0.999575  
 Micropore volume = 0.000 cc/g  
 Micropore area = 0.000 m<sup>2</sup>/g  
 External surface area = 144.633 m<sup>2</sup>/g

Fig. S5. The BET plot and analysis data for CBOS precursor.



**Table S1.** ANOVA results of the quadratic surface model for the benzene adsorption process.

Source	Sum of Squares	Degree of Freedom	Mean Square	F-value	p-value	
Model	6.722E+05	9	74688.59	81.89	< 0.0001	significant
X <sub>1</sub> -Time	2.238E+05	1	2.238E+05	245.42	< 0.0001	
X <sub>2</sub> -Concentration	1.627E+05	1	1.627E+05	178.36	< 0.0001	
X <sub>3</sub> -Temperature	55919.32	1	55919.32	61.31	< 0.0001	
X <sub>1</sub> X <sub>2</sub>	10989.03	1	10989.03	12.05	0.0060	
X <sub>1</sub> X <sub>3</sub>	5371.66	1	5371.66	5.89	0.0356	
X <sub>2</sub> X <sub>3</sub>	127.20	1	127.20	0.1395	0.7166	
X <sub>1</sub> <sup>2</sup>	1.062E+05	1	1.062E+05	116.46	< 0.0001	
X <sub>2</sub> <sup>2</sup>	1.098E+05	1	1.098E+05	120.37	< 0.0001	
X <sub>3</sub> <sup>2</sup>	35291.14	1	35291.14	38.70	< 0.0001	
Residual	9120.21	10	912.02	-	-	
Lack of Fit	122.79	5	122.79	0.1269	0.8215	insignificant
Pure Error	6.25	5	1.25	-	-	
Cor Total	6.813E+05	19	-	-	-	

R<sup>2</sup> = 0.9866, Adjusted R<sup>2</sup> = 0.9746, Predicted R<sup>2</sup> = 0.8994, Adequate Precision = 28.2665

**Table S2.** ANOVA results of the quadratic surface model for the toluene adsorption process.

Source	Sum of Squares	Degree of Freedom	Mean Square	F-value	p-value	
Model	8.065E+05	9	89606.56	72.92	< 0.0001	significant
X <sub>1</sub> -Time	2.661E+05	1	2.661E+05	216.53	< 0.0001	
X <sub>2</sub> -Concentration	2.082E+05	1	2.082E+05	169.43	< 0.0001	
X <sub>3</sub> -Temperature	60054.58	1	60054.58	48.87	< 0.0001	
X <sub>1</sub> X <sub>2</sub>	14377.84	1	14377.84	11.70	0.0065	
X <sub>1</sub> X <sub>3</sub>	6191.06	1	6191.06	5.04	0.0486	
X <sub>2</sub> X <sub>3</sub>	0.0003	1	0.0003	2.543E-07	0.9996	
X <sub>1</sub> <sup>2</sup>	1.339E+05	1	1.339E+05	108.98	< 0.0001	
X <sub>2</sub> <sup>2</sup>	1.264E+05	1	1.264E+05	102.83	< 0.0001	
X <sub>3</sub> <sup>2</sup>	34457.52	1	34457.52	28.04	0.0003	
Residual	12288.28	10	1228.83	-	-	
Lack of Fit	117.58	5	117.58	0.1023	0.9510	insignificant
Pure Error	1.05	5	0.2097	-	-	
Cor Total	8.187E+05	19	-	-	-	

R<sup>2</sup> = 0.9850, Adjusted R<sup>2</sup> = 0.9715, Predicted R<sup>2</sup> = 0.8854, Adequate Precision = 26.9512