

Supporting information for

**Structural analysis and catalytic activity of tetranuclear metal carboxylate clusters with
[KZn₃(μ₃-OH)(OOCCPh₃)₆] or [Zn₄(μ₄-O)(OOCCPh₃)₆] central motif**

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Crystallographic Data for Compounds 1-3.

Table S1. Crystal and data collection parameters for compounds **1-3**.

| Crystal | 1·4CH₃CN·0.2C₆H₁₄ | 1·2·3.58CH₃CN·0.21C₆H₁₄ | 3·5CH₃CN |
|---|--|---|---|
| Chemical formula | C _{131.2} H _{108.8} N ₅ O ₁₃ Zn ₃ K | C _{130.42} H _{107.6} N _{4.58} O ₁₃ Zn _{3.08} K _{0.92} | C ₁₂₄ H ₉₆ N ₂ O ₁₃ Zn ₄ |
| Formula Mass | 2198.71 | 2184.33 | 2288.86 |
| Crystal system | Triclinic | Triclinic | Triclinic |
| Space group | <i>P</i> ī | <i>P</i> ī | <i>P</i> ī |
| <i>a</i> /Å | 14.302(4) | 14.301(4) | 13.610 (3) |
| <i>b</i> /Å | 15.362(5) | 15.348 (4) | 15.338 (4) |
| <i>c</i> /Å | 27.872 (9) | 27.869 (8) | 29.649 (8) |
| <i>α</i> /° | 74.68 (3) | 74.64 (3) | 92.10 (3) |
| <i>β</i> /° | 84.79 (3) | 84.79 (3) | 97.87 (3) |
| <i>γ</i> /° | 66.28 (3) | 66.31 (3) | 112.25 (3) |
| Unit cell volume/Å ³ | 5406 (3) | 5401 (3) | 5648(3) |
| Temperature/K | 100(2) | 100(2) | 100(2) |
| <i>Z</i> | 2 | 2 | 2 |
| Radiation type | MoKα | MoKα | MoKα |
| Absorption coefficient, μ/mm^{-1} | 0.767 | 0.781 | 0.907 |
| No. of reflections measured | 60695 | 71595 | 49124 |
| No. of independent reflections | 27833 | 23571 | 24365 |
| No. of observed reflections ($I > 2\sigma(I)$) | 22907 | 19892 | 14784 |
| R_{int} | 0.0282 | 0.0297 | 0.0525 |
| Final R_I values ($I > 2\sigma(I)$) | 0.0393 | 0.0404 | 0.0539 |
| Final $wR(F^2)$ values ($I > 2\sigma(I)$) | 0.1134 | 0.1046 | 0.1167 |
| Final R_I values (all data) | 0.0476 | 0.0515 | 0.0998 |
| Final $wR(F^2)$ values (all data) | 0.1166 | 0.1112 | 0.1279 |
| Goodness of fit on F^2 | 1.104 | 1.037 | 0.927 |
| $\Delta\rho_{\text{max}}/\text{e}\text{\AA}^{-3}$ | 1.03 | 1.55 | 0.72 |
| $\Delta\rho_{\text{min}}/\text{e}\text{\AA}^{-3}$ | -0.75 | -1.03 | -0.82 |

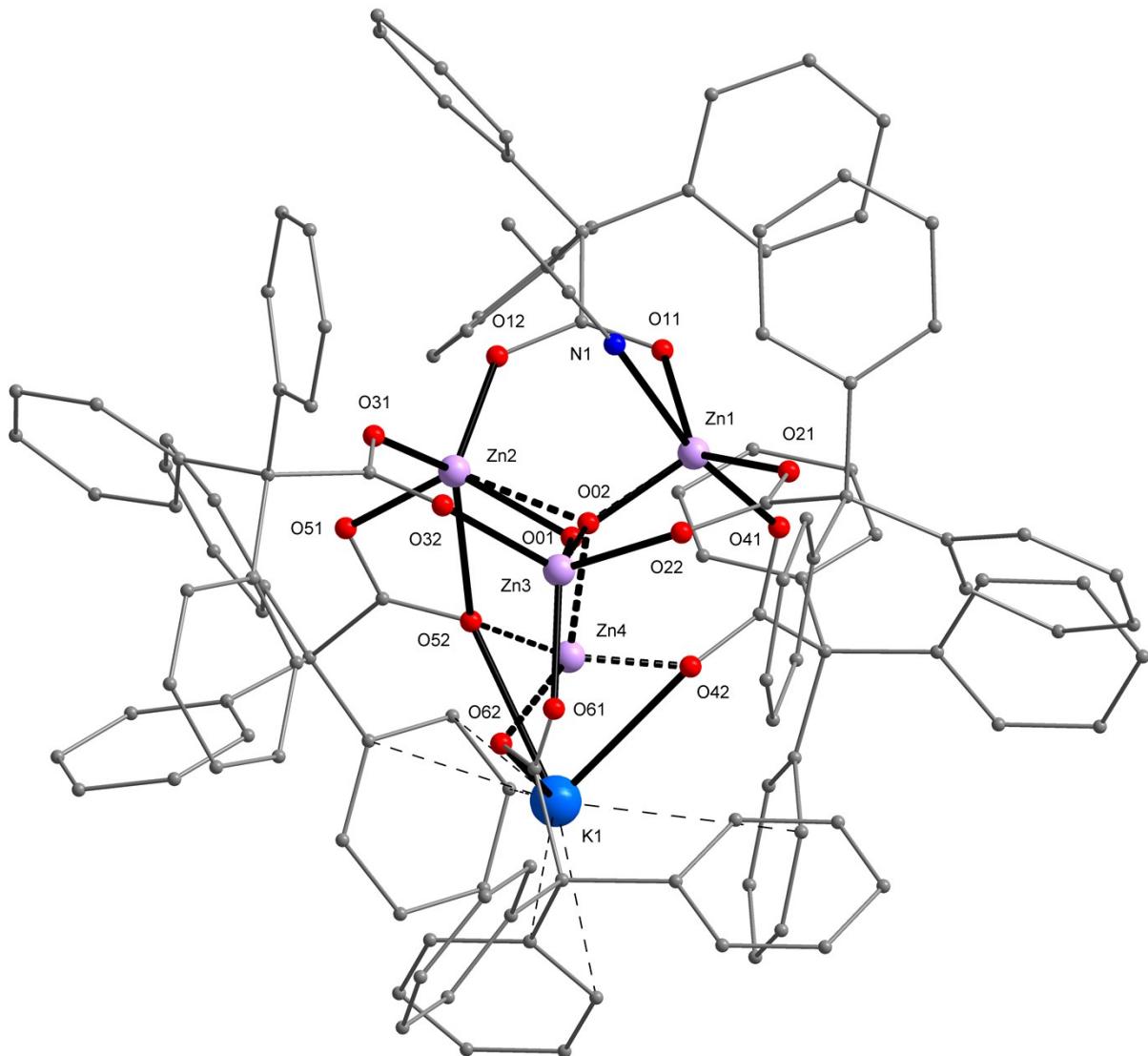


Figure S1. Molecular structures of $[KZn_3(\mu_3\text{-OH})(Ph_3CCOO)_6(MeCN)]_{0.92} \cdot [Zn_4(\mu_4\text{-O})(Ph_3CCOO)_6(MeCN)_2]_{0.08} \cdot 3.58\text{MeCN} \cdot 0.21C_6H_{14}$, (**1**·**2**·3.58MeCN·0.21C₆H₁₄). Hydrogen atoms and solvent molecules are omitted for clarity.

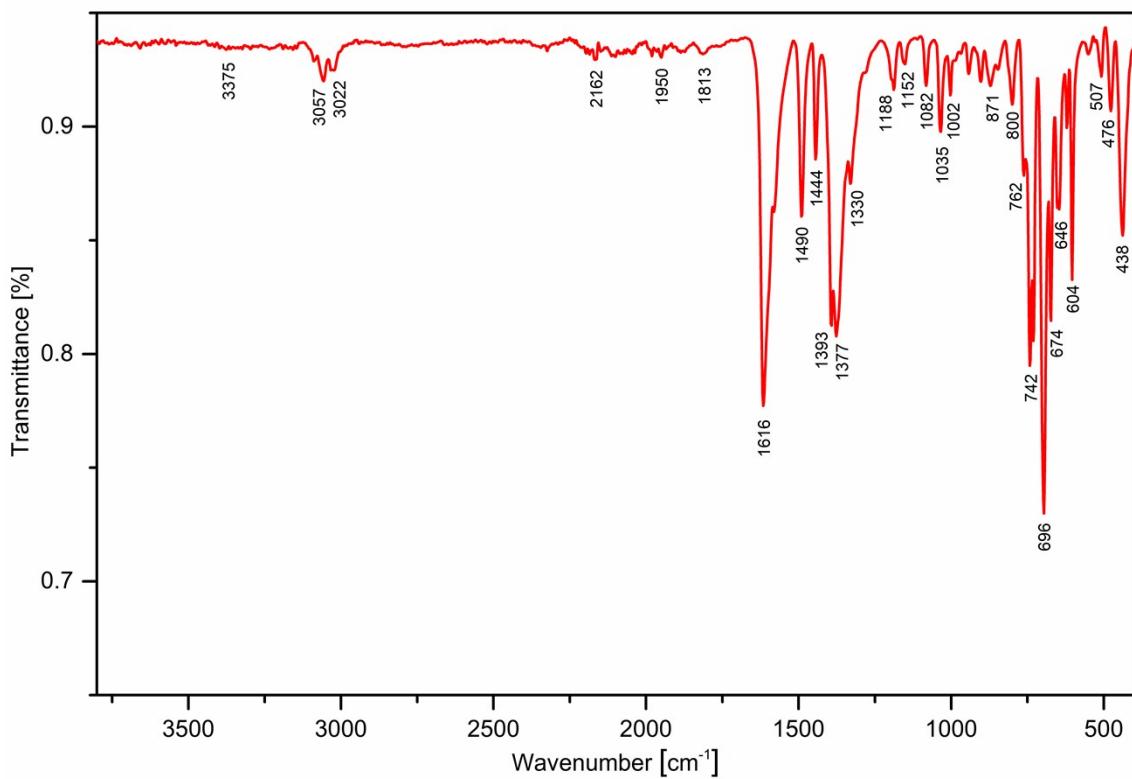


Figure S2. FTIR-ATR spectra of **1**.

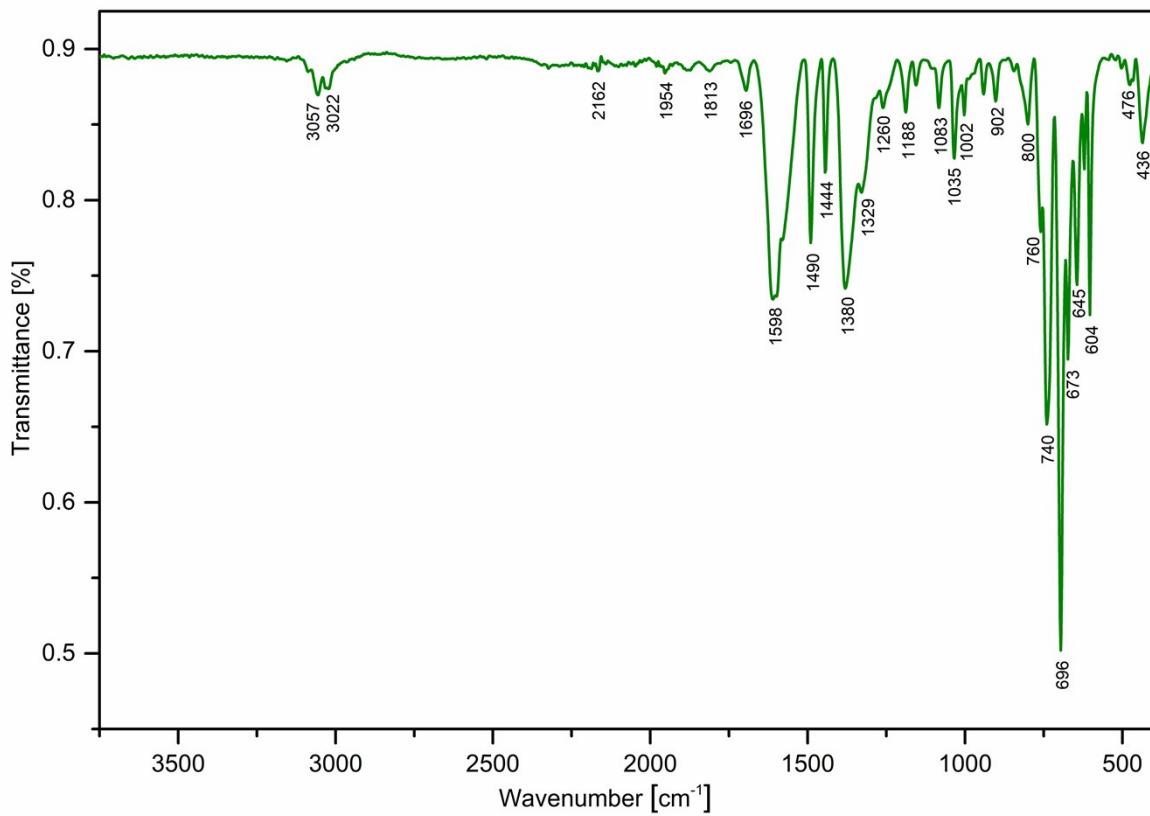


Figure S3. FTIR-ATR spectra of **1·2**.

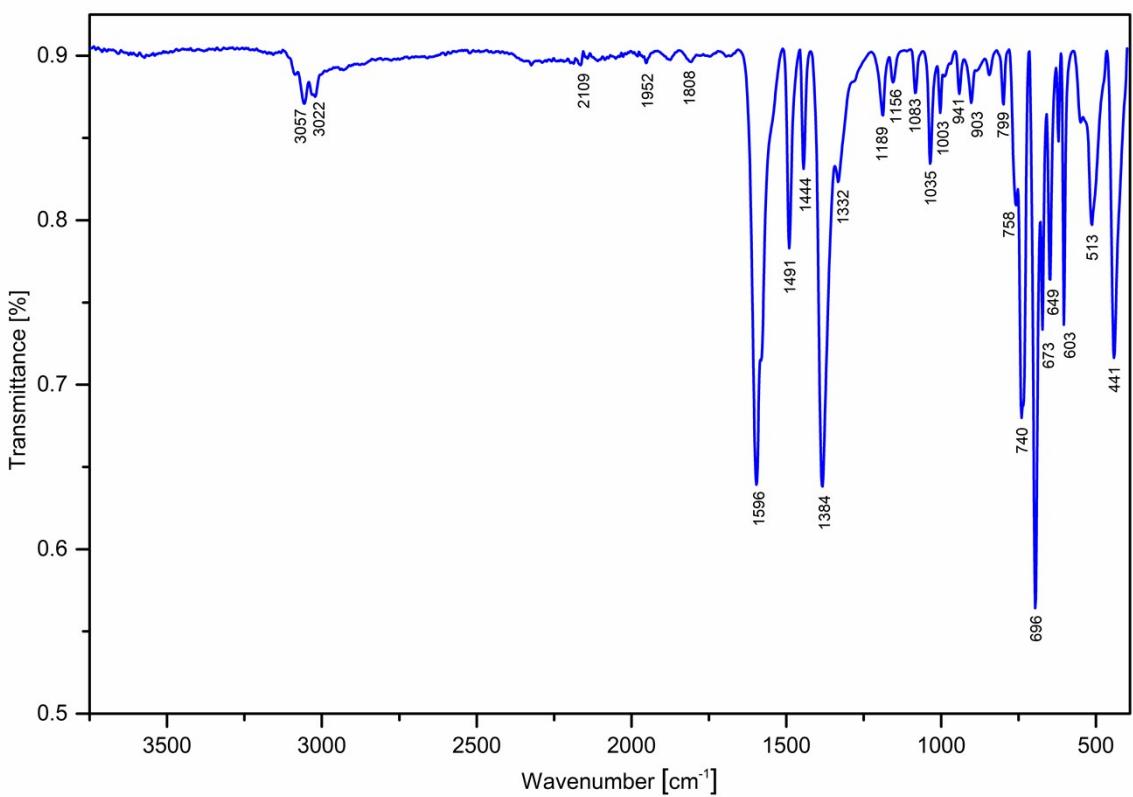


Figure S4. FTIR-ATR spectra of **3**.

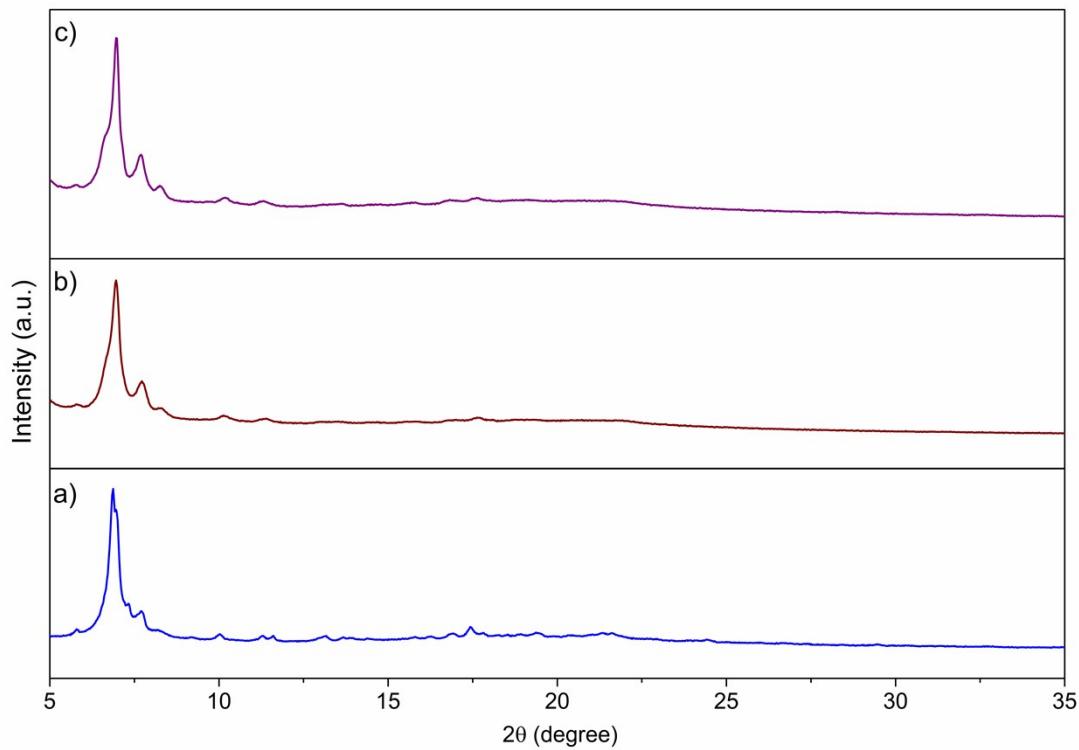


Figure S5. Comparison of PXRD patterns of **3** measured at 25 °C for the ground crystals (a), crystals without grounding and crystalline precipitate isolated from the reaction (c).

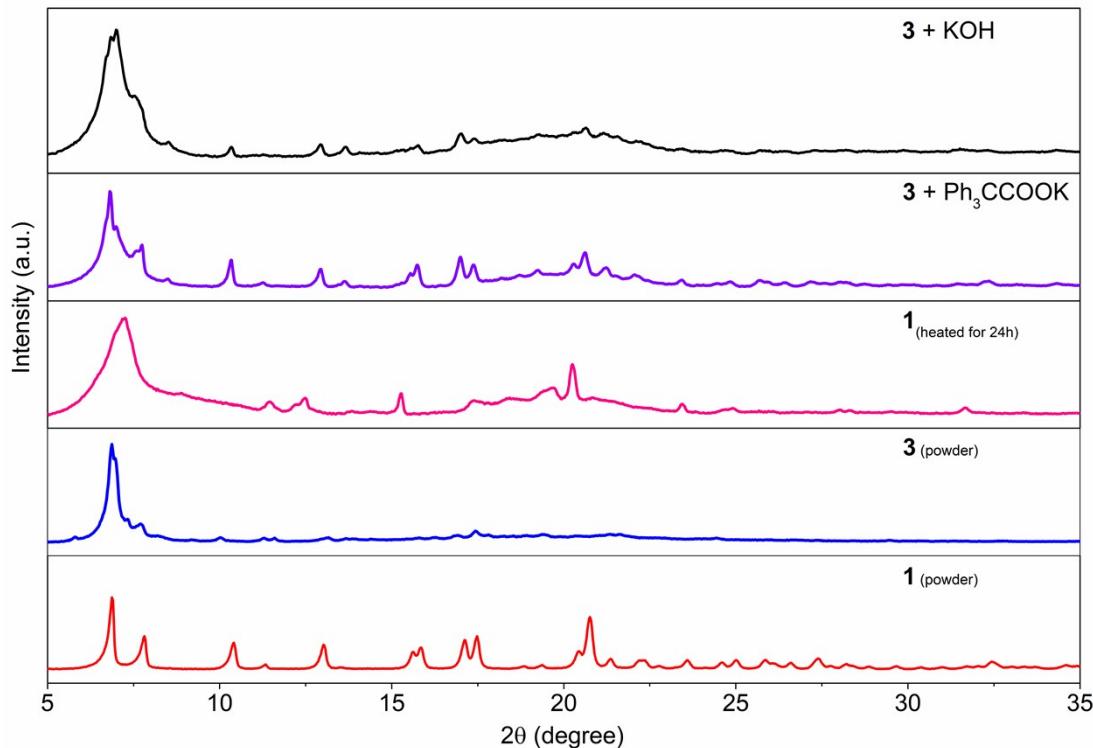


Figure S6. Comparison of PXRD patterns of **1** and **3** together with patterns of crystalline materials obtained by the heating of **1** for 24h in CH₃CN solution (pink), and reaction of **3** with Ph₃CCOOK (violet) or KOH (black).