

Electronic Supplementary Information for

**Multinary PtPdNiP truncated octahedral mesoporous nanocages for
enhanced methanol oxidation electrocatalysis**

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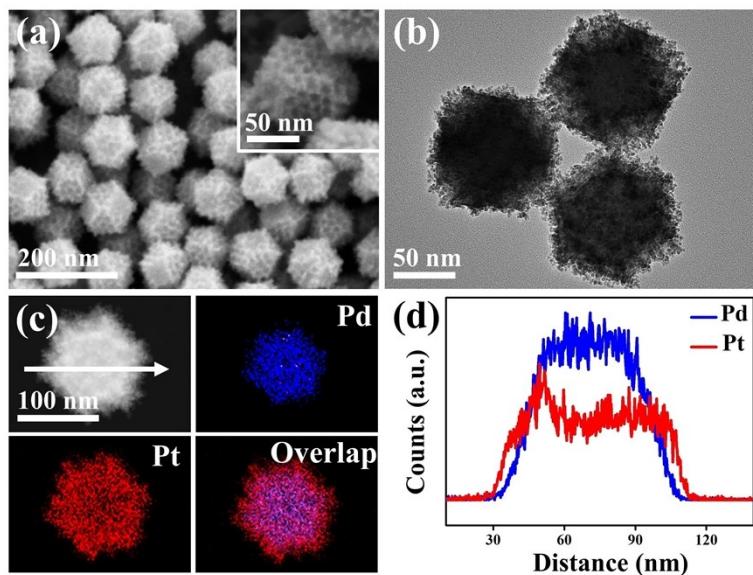


Fig. S1 (a) SEM and (b) TEM images of Pd@PtPd MTOs. (c) HAADF-STEM image and corresponding EDX elemental mappings images and (d) line-scanning profile of Pd@PtPd MTOs.

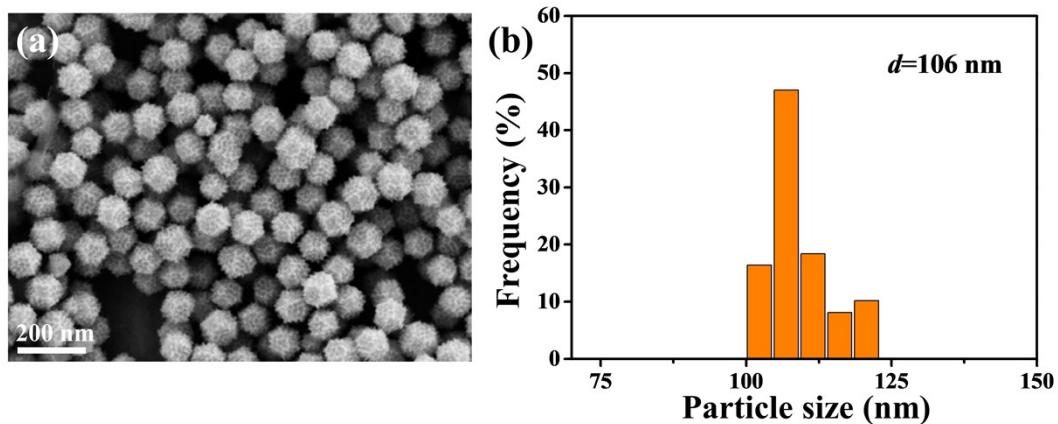


Fig. S2 Histogram of particle size distribution for the Pd@PtPd MTOs.

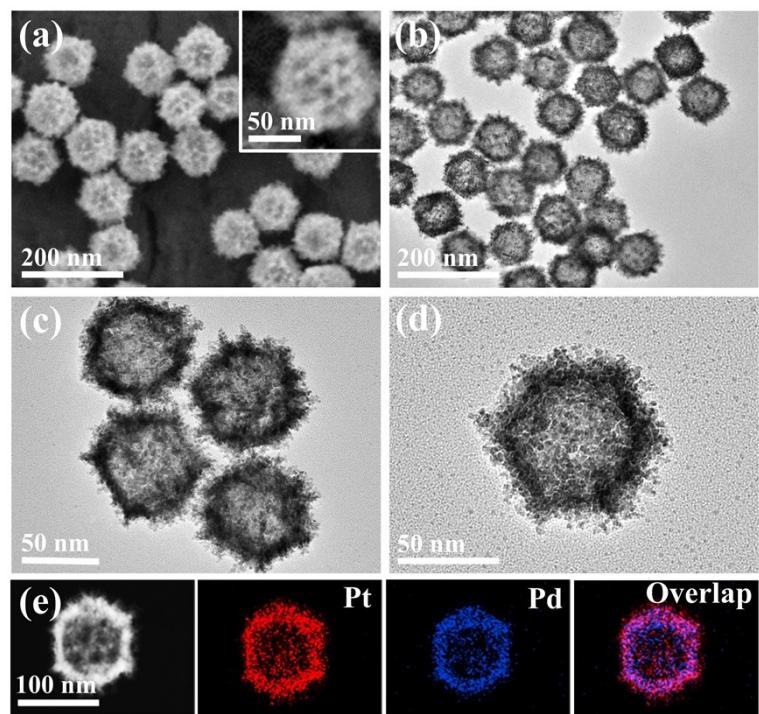


Fig. S3 (a) SEM and (b-d) TEM images of PtPd TOMNs. (e) HAADF-STEM image of one PtPd TOMN nanoparticle and corresponding EDX elemental mappings images.

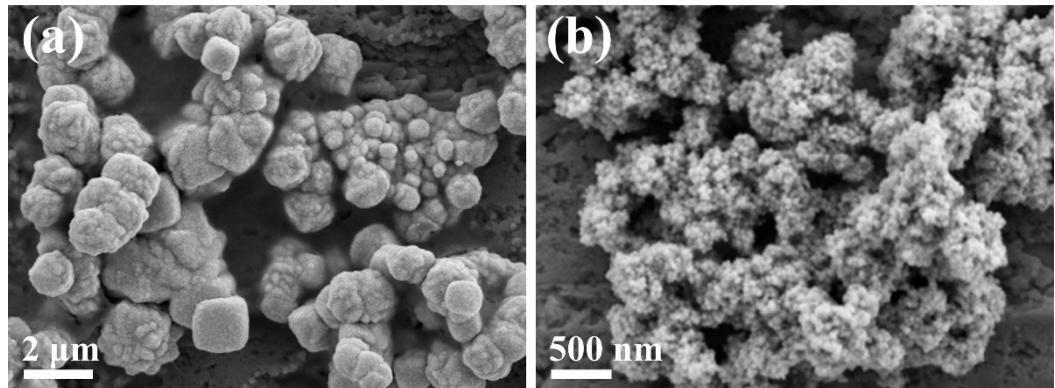


Fig. S4 SEM images of the samples prepared under the typical conditions without (a) F127 and by replacing F127 with (b) PVP.

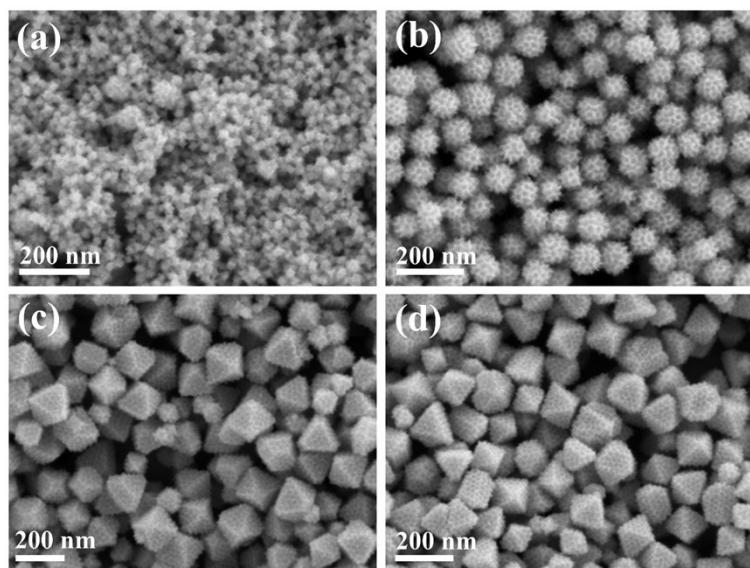


Fig. S5 SEM images of the samples prepared with different amounts of HCl under the typical synthesis: (a) 0 mL, (b) 0.1 mL, (c) 0.3 mL (d) 0.4 mL.

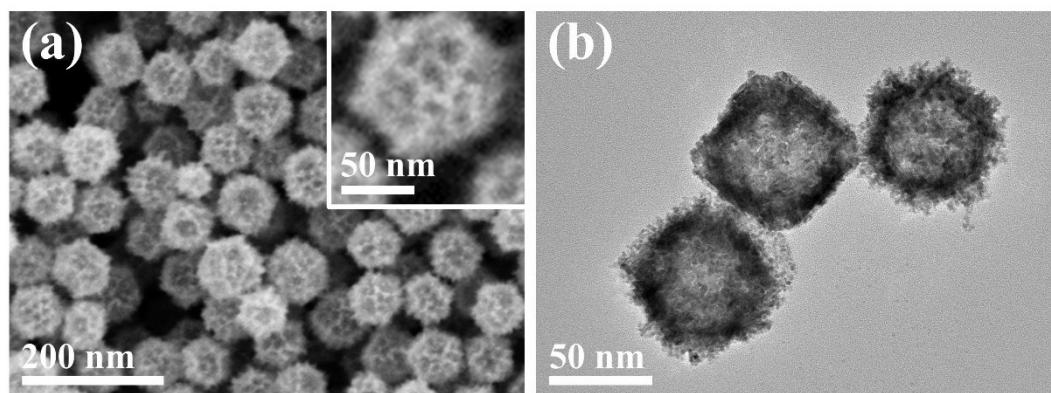


Fig. S6 (a) SEM and (b) TEM images of PtPdNi TOMNs.

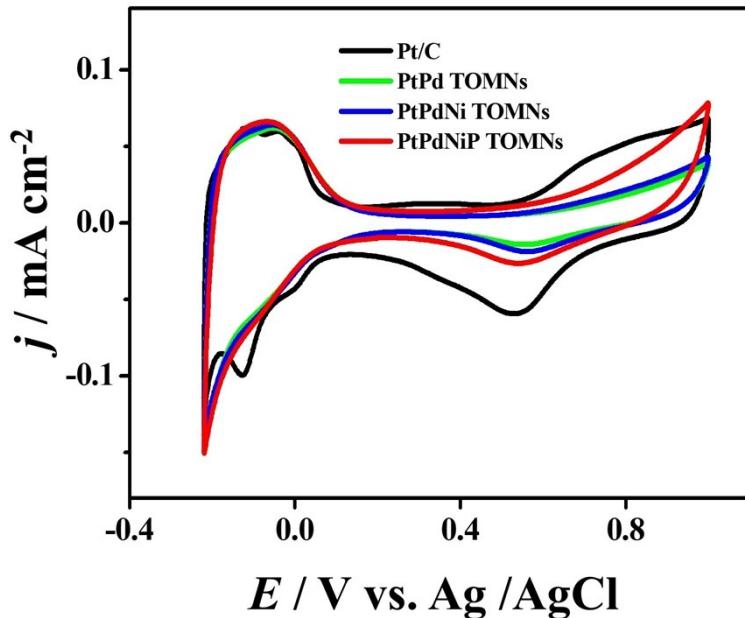


Fig. S7 CVs of the catalysts recorded in N_2 -saturated 0.5 M H_2SO_4 solution with a scan rate of 50 mV s^{-1} .

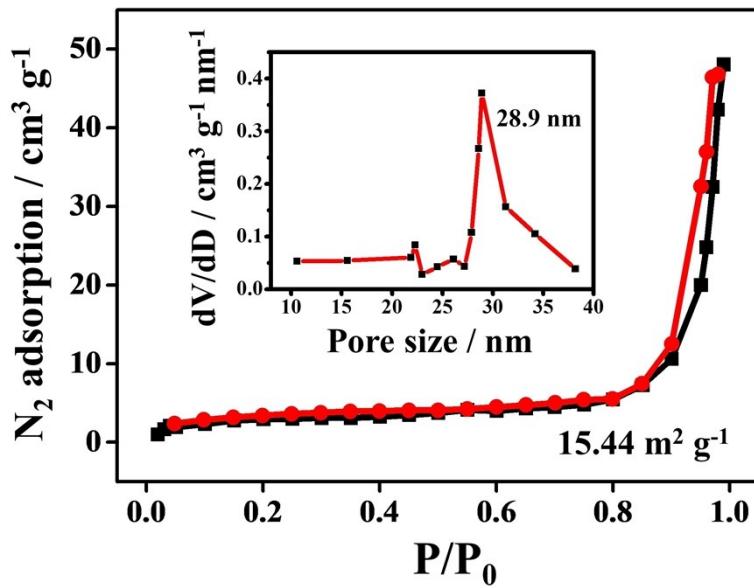


Fig. S8 N_2 adsorption-desorption isotherms for PtPdNiP TOMNs. The inset is the BJH pore-size distribution curves.

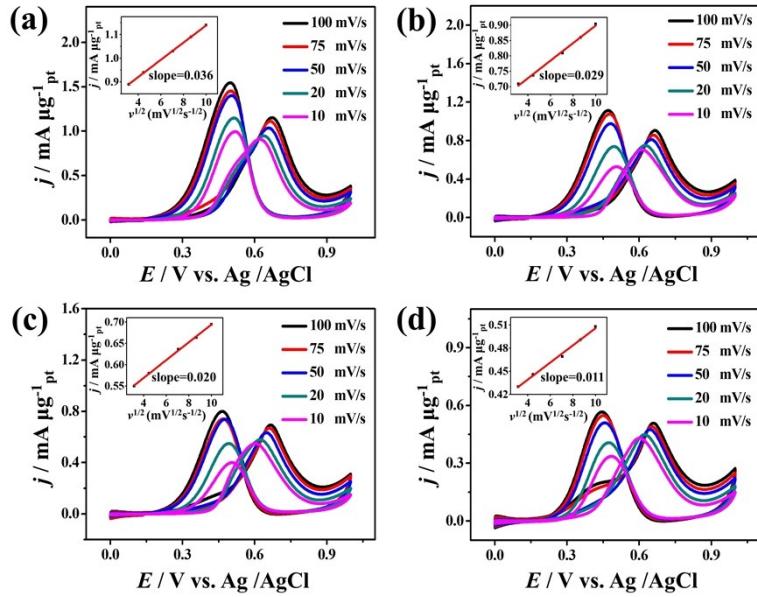


Fig. S9 CVs at different scan rates and their corresponding fitted plot of the j_m vs. the $v^{1/2}$ for the (a) PtPdNiP TOMNs, (b) PtPdNi TOMNs, (c) PtPd TOMNs and (d) Pt/C.

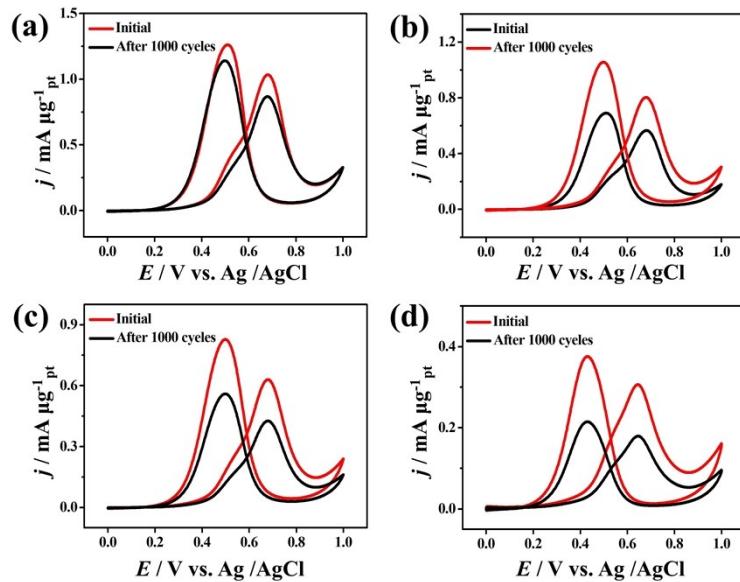


Fig. S10 CVs of MOR recorded in 0.5 M H_2SO_4 solution with 1.0 M CH_3OH at scan rate of 50 mV s^{-1} before and after durability test for (a) PtPdNiP TOMNs, (b) PtPdNi TOMNs, (c) PtPd TOMNs and (d) Pt/C.

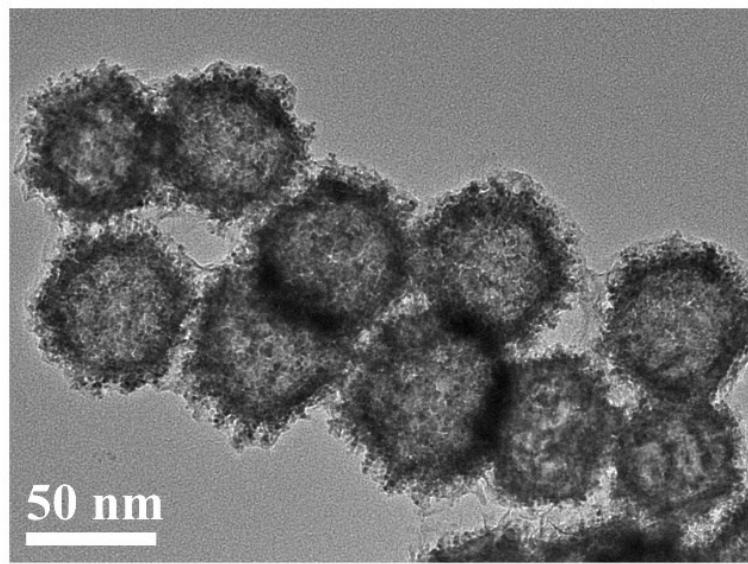


Fig. S11 TEM image of the PtPdNiP TOMNs after the durability test.

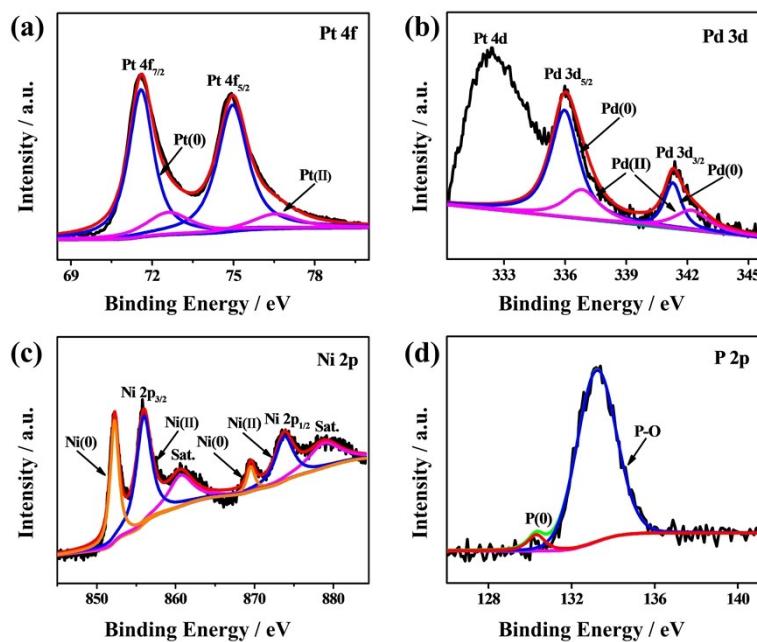


Fig. S12 XPS spectra of PtPdNiP TOMNs in (a) Pt 4f (b) Pd 3d (c) Ni 2p and (d) P 2p regions after the durability test.

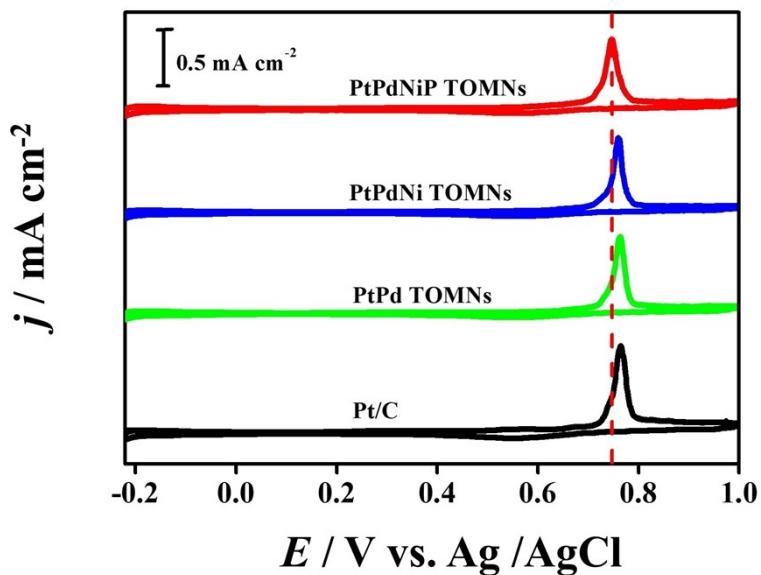


Fig. S13 CO-stripping curves of different catalysts tested in a CO-saturated 0.5 M H₂SO₄ solution at a scan rate of 50 mV s⁻¹.

Table S1 Comparisons of the MOR performance of PtPdNiP TOMNs with other reported Pt-based electrocatalysts.

Catalysts	Condition	Scan rate (mV s ⁻¹)	Mass activity (A mg ⁻¹ _{Pt})	Ref.
PtPdNiP TOMNs	0.5 M H₂SO₄ containing 1.0M CH₃OH	50	1.04	This work
PtPdCo MHNPs	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.91	1
Pt Nanoparticles	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.87	2
Pt-Pd-Cu Nanodendrites	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.69	3
PtPdTe Nanowire	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.59	4
Hollow Pt–Pd Nanocages	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.58	5
Pt Nanowires	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.46	6
Au@Pd@Pt Core-Shell Nanoparticles	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.43	7
PtAg Octahedral Nanocrystals	0.5 M H ₂ SO ₄ containing 1.0M CH ₃ OH	50	0.37	8

References

1. Y. Xu, Y. Li, X. Qian, D. Yang, X. Chai, Z. Wang, X. Li, L. Wang and H. Wang, *Nanoscale*, 2019, **11**, 4781-4787.
2. M. Yan, Q. Jiang, T. Zhang, J. Wang, L. Yang, Z. Lu, H. He, Y. Fu, X. Wang and H. Huang, *J. Mater. Chem. A*, 2018, **6**, 18165-18172.
3. R. Chang, L. Zheng, C. Wang, D. Yang, G. Zhang and S. Sun, *Appl. Catal. B: Environ.*, 2017, **211**, 205-211.
4. H. H. Li, S. Zhao, M. Gong, C. H. Cui, D. He, H. W. Liang, L. Wu and S. H. Yu, *Angew. Chem., Int. Ed.*, 2013, **52**, 7472-7476.
5. L. Wang and Y. Yamauchi, *J. Am. Chem. Soc.*, 2013, **135**, 16762-16765.
6. H. Meng, F. Xie, J. Chen, S. Sun and P. K. Shen, *Nanoscale*, 2011, **3**, 5041-5048.
7. L. Wang and Y. Yamauchi, *Chem. Mater.*, 2011, **23**, 2457-2465.
8. J. Li, H. Rong, X. Tong, P. Wang, T. Chen and Z. Wang, *J. Colloid Interf. Sci.*, 2018, **513**, 251-257.