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1 Supporting Information

- 1 Experimental procedures
- 2 Separation, extraction and identification of active natural products.
- 3 The following compounds were isolated and purified according to our previous experience
- 4 accumulation and experimental results of the laboratory¹⁻³.
- 5

Oleanolic acid

Glycyrrhetinic acid

Betulin

6 **Scheme 1.** Structural formula of active natural products

- 7 *Oleanolic acid (OA)* crude extract were obtained by chemical fractionation of 95% (v/v)
- 8 ethanol extract of *Dry papaya powder*. The crude extracts were eluted with petroleum ether
- 9 and carbon tetrachloride. The obtained extracts were then separated and purified by
- 10 macroporous resin (D4020). Then repeated and purification on a reversed-phase C18 column.
- 11 *Betulin (Bet)* crude extracts of *Dry birch bark* were ultrasonically extracted with 95% (v/v)

12 methanol. The crude extracts were eluted with petroleum ether and ethyl acetate. The

- 13 obtained extract was purified by macroporous resin (D101). Then repeated purification on
- 14 reverse-phase C18 column.
- 15 *Glycyrrhetinic acid (GA)* was obtained by repeated decoction. The extraction of the licorice
- 16 powder is made with boiling water. The concentrate is heated under reflux with 5% dilute
- 17 sulfuric acid, washed to neutral, dried and dissolved in hot chloroform, filtered and subjected
- 18 to column chromatography. Recrystallization gave glycyrrhetinic acid crystals.
- 19 Analysis was conducted by high-performance liquid chromatography (HPLC) and
- 20 compounds were identified by comparing with standards. Furthermore, completion of the
- 21 structural characterization was performed by mass spectrometry data and protonic nuclear
- 22 magnetic resonance spectrometric data and comparison with the previously published data.
- 23 Replica-exchange molecular dynamics (REMD) simulation
- 24 To understand the self-assembly process, all-atom replica exchange molecular dynamics
- 25 (REMD) simulations was performed using Gromacs software.⁴ Carriers (OA) and drug (PTX)
- 26 in simulation system, the number is 14 and 1, and the simulation used the Charmm36 force
- 1 field, TIP3P model was used to model aqueous solution sodium ions were added to balance
- 2 the system negative charge. The long-range electrostatic action was calculated by PME
- 3 method, the cut-off radius was set to 10 nm, and the van der Waals force interaction cut-off
-
- 4 radius was 12 nm. The REMD simulation system is divided into two parts, OP and UP. The two systems are first subjected to a 200 ps MVT balance at 300K. In the temperature range of two systems are first subjected to a 200 ps MVT balance at 300K. In the temperature range of
- 6 300K~700K, the online server (http://folding mc uu se/remd/) was used to predict the
- 7 temperature distribution of each copy, and 35 copies were generated.⁵ All replicas were run in
- 8 parallel, each replica was subjected to a 50 ns REMD simulation, the simulated time step was
- 9 set to 2fs, and replica exchange attempts between adjacent replicas were made every 500
- 10 steps. The trajectory was saved every 10 ps, and the simulation process was visually analyzed
- 11 using YASARA and VMD software. All the above calculations are conducted on the
- 12 MolDesigner Molecular Simulation Platform^{6,7}.

- 2 **Figure S1 SEM images of GA-PTX NPs and Bet-PTX NPs.** All GA-PTX NPs are nanosheets morphology and all Bet-PTX NPs are nanofibers morphology
- nanosheets morphology and all Bet-PTX NPs are nanofibers morphology

 $\begin{array}{c} 1 \\ 2 \\ 3 \end{array}$

- Figure S2 SEM images of OA-PTX NPs. The nanofibers morphology is obvious at the
- 3 yellow arrow.

3 encapsulation efficiency of OA-PTX NPs are 17.1% and 62.6%, respectively**.**

4

 $\frac{1}{2}$

- 1 **Figure S5. Wettability measurement of NPs.** Three-phase contact angles of free OA and
- 2 OA NPs and OA-PTX NPs. All NPs prepared by the emulsion solvent evaporation method $\begin{array}{c} 2 \\ 3 \\ 4 \end{array}$
- 3 showed substantially improved hydrophilicity.
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1 **Figure S6. OA-PTX NPs drug release profile.** Drug release profiles of OA-PTX NPs at

three different pH values (7.3, 6.8, and 5.5) at room temperature. Data are expressed as mean

- \pm s.e.m. (n = 3) $\begin{array}{c} 2 \\ 3 \\ 4 \end{array}$
-

Figure S7. Drug release in PBS medium containing 20% lecithin. Data are expressed as mean \pm s.e.m. (n = 3) $\begin{array}{c} 1 \\ 2 \\ 3 \end{array}$

- 1 **Figure S8. Stability of OA-PTX NPs. a**. OA-PTX NPs of freeze-dried powders were redispersion in double-distilled water at different storage times. **b**. Stability in the prese
- 2 redispersion in double-distilled water at different storage times. **b**. Stability in the presence of serum **c**. OA-PTX NPs size after different dilution factors
- serum **c**. OA-PTX NPs size after different dilution factors
- 4

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2 **Figure S9. Cell uptake representative microimages.** 4T1 cells incubated with OA-PTX

3 NPs at different times. Magnification ×400

Figure S10. Fluorescence intensity of fluorescently labeled nanoparticles before and

- **after demulsification.**
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1 **Figure S13. Supplementary data for Haematoxylin and eosin (H&E).**

Table S1. Characterization of OA-PTX NPs by DLS.

Table S2. Characterization of OA-PTX NPs by DLS.

Table S3. Characterization of OA-PTX NPs by DLS.

Table S4. Fluorescence intensity of tumor tissue at different time.

Table S5. The fluorescence intensity of main tissues at different time.

1 References

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