

Supporting Information

Strong interfacial coupling for NiS thin layer covered CdS nanorods with highly efficient photocatalytic hydrogen production

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Figures

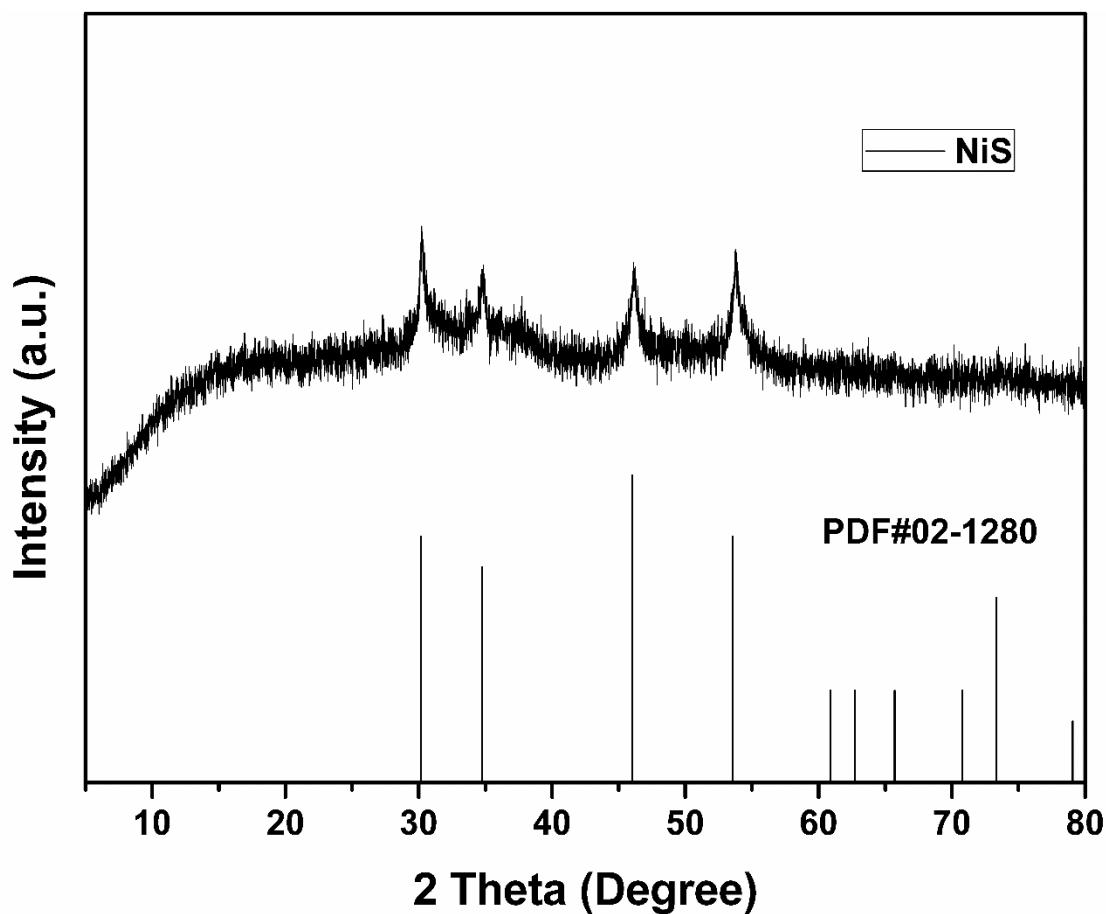


Fig. S1. XRD pattern of pure NiS.

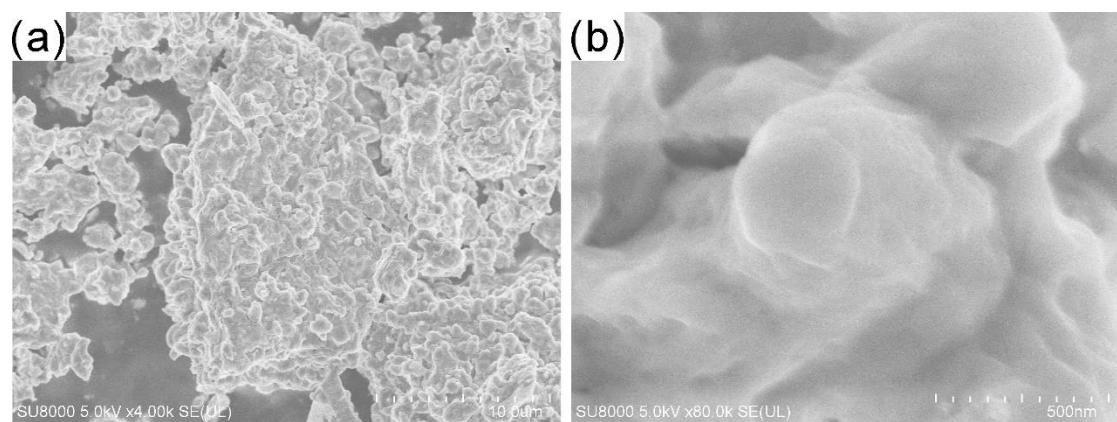


Fig. S2. (a) Low- and (b) high-magnification SEM images of pure NiS.

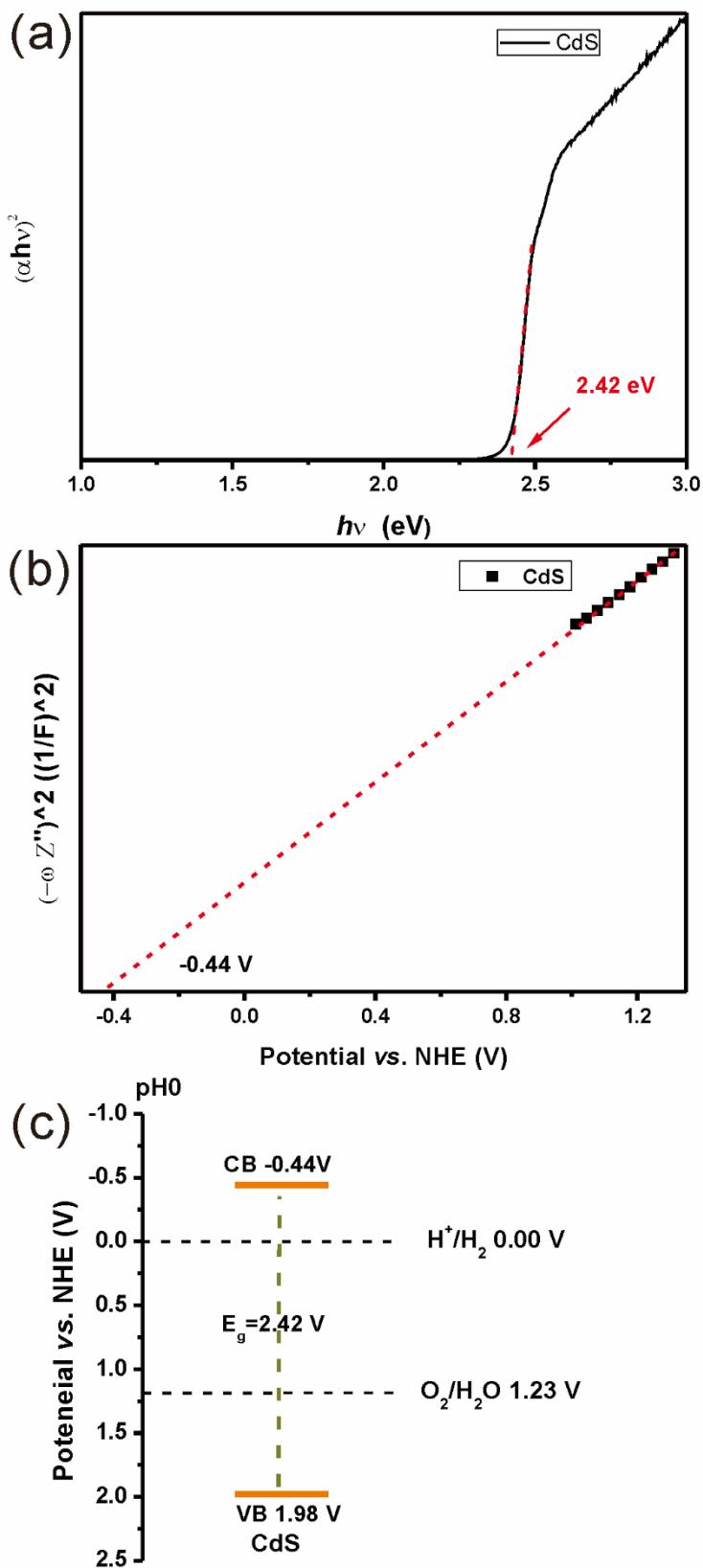


Fig. S3. (a) The plot of $(\alpha h\nu)^2$ versus $h\nu$ of CdS, (b) Mott–Schottky plots of CdS, (c) Energy band diagram of CdS.

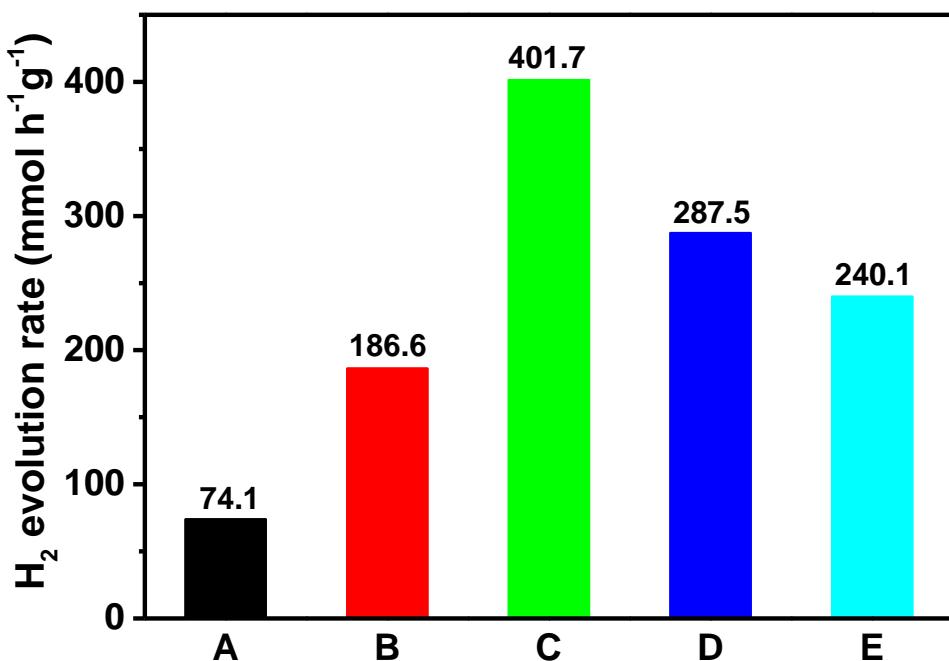


Fig. S4. The H_2 -production rate over 25NS-C photocatalyst using different concentrations of hole scavenger under visible light irradiation. (A) 0.25 M Na_2S , 0.35 M Na_2SO_3 , (B) 0.50 M Na_2S , 0.70 M Na_2SO_3 , (C) 0.75 M Na_2S , 1.05 M Na_2SO_3 , (D) 0.875 M Na_2S , 1.225 M Na_2SO_3 , (E) 1.0 M Na_2S , 1.4 M Na_2SO_3 .

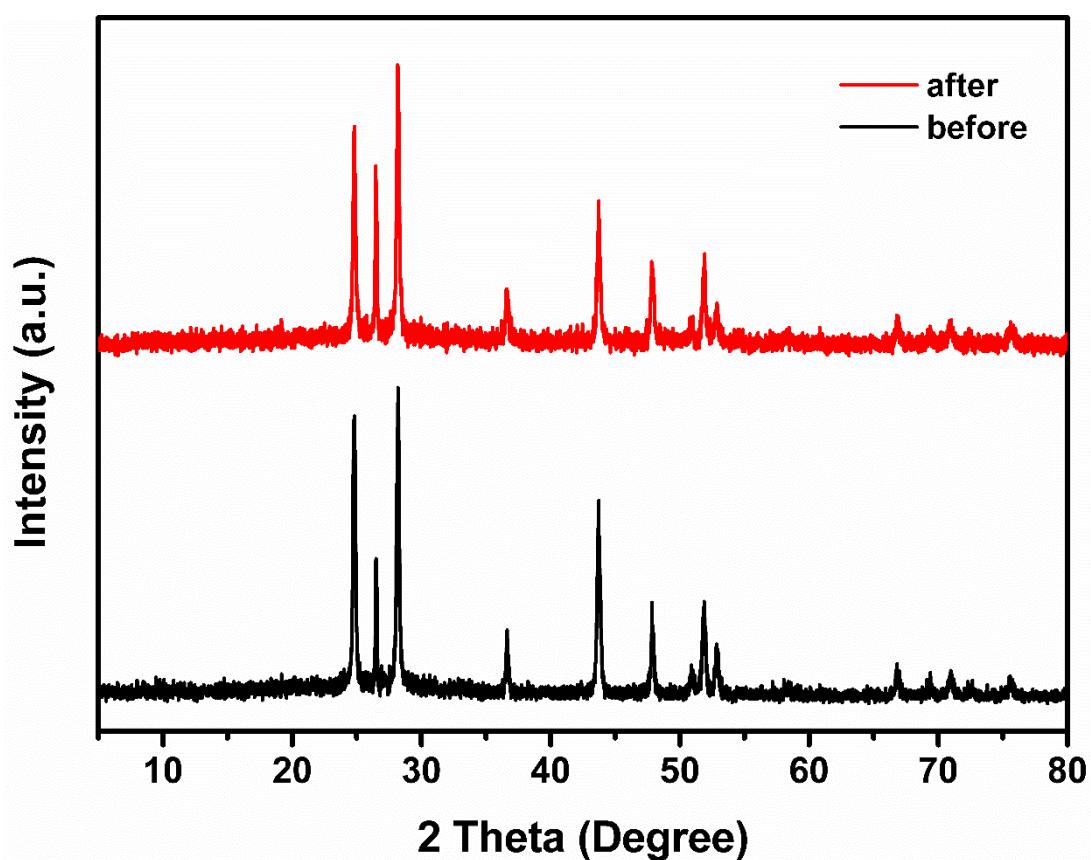


Fig. S5. XRD patterns of 25NS-C before reaction and after 3h of photocatalytic reaction.

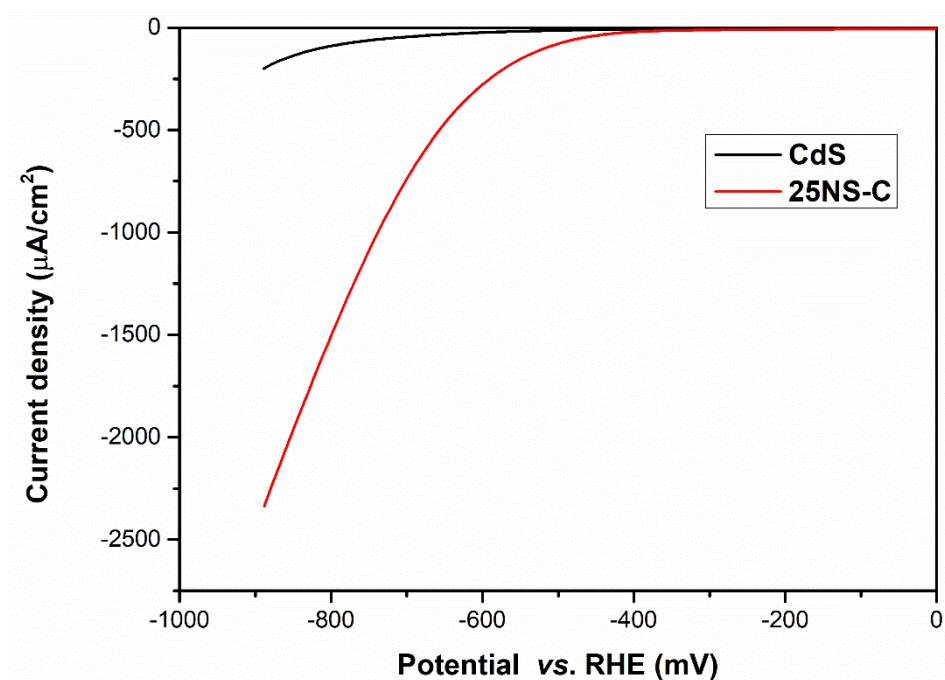


Fig S6. The LSV curves of CdS and 25NS-C samples.

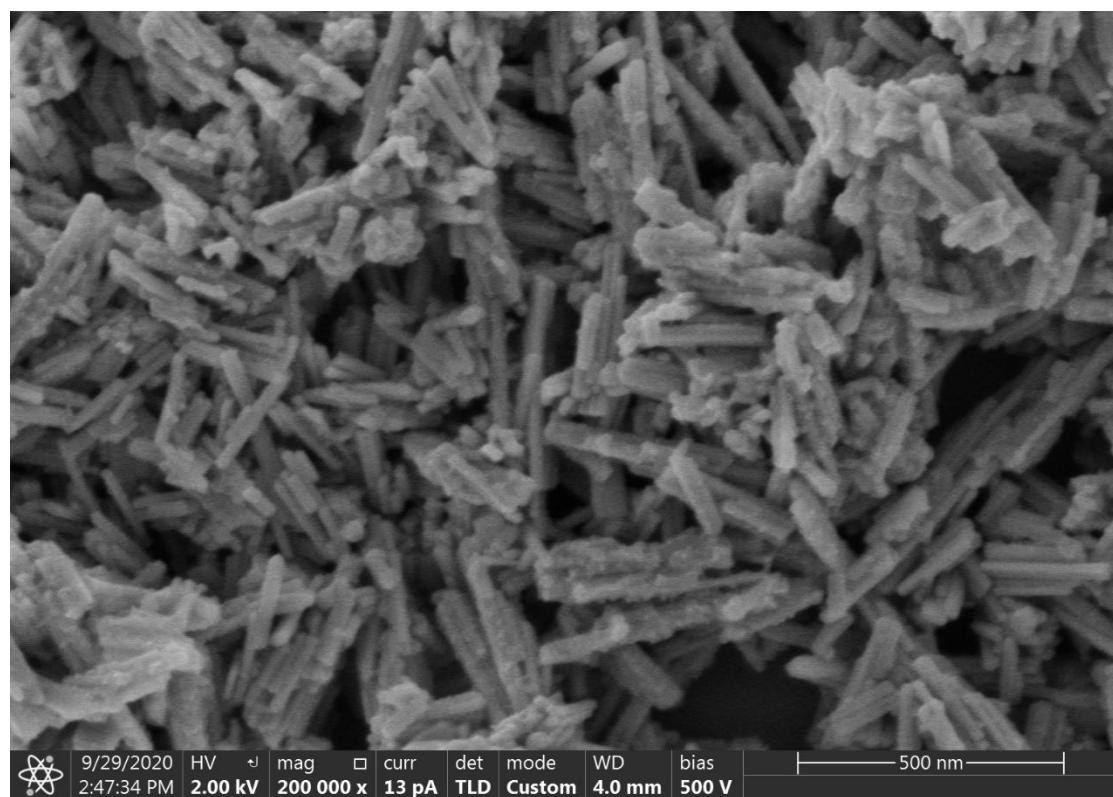


Fig S7. The SEM image of 25NS-C after photocatalytic reaction.

Table S1. Collected data of H₂-evolution activity over CdS-based photocatalysts.

Photocatalyst	Light source	Reactant solution	Activity/ mmol·h ⁻¹ ·g ⁻¹	References
NiS/CdS	Xe-Vis (≥420 nm)	10 vol.% lactic acid	49.2	¹
MoS ₂ /CdS	Xe-Vis (≥420 nm)	10 vol.% lactic acid	49.8	²
NiS/CdS	Xe-Vis (≥420 nm)	20 vol.% lactic acid	30.1	³
BP-Au-CdS	UV-vis-NIR	0.35M Na ₂ S-0.25M Na ₂ SO ₃	10.1	⁴
NiS/CdS	Xe-Vis (≥420 nm)	20 vol.% lactic acid	158.7	⁵
Pd@CdS/PdS	Xe-Vis <td>0.1M Na₂S-0.1M Na₂SO₃</td> <td>144.8</td> <td>⁶</td>	0.1M Na ₂ S-0.1M Na ₂ SO ₃	144.8	⁶
Ni ₂ P/CdS	Xe-Vis (≥420 nm)	0.35M Na ₂ S-0.25M Na ₂ SO ₃	34.9	⁷

Ni(OH) ₂ /CdS	Xe-Vis (≥ 420 nm)	10 vol.% ethanol (pH 14.7)	35	8
Co ₂ P/CdS	Metal halide lamp (sunlight)	10 vol.% lactic acid	262.16	9
Ni/CdS	447 nm laser	10 vol.% ethanol (pH 14.7)	63	10
Ni ₂ P/CdS	Xe-Vis (≥ 420 nm)	1.25M Na ₂ S-1.75M Na ₂ SO ₃	$\sim 1200 \mu\text{mol} \cdot \text{h}^{-1} \cdot \text{g}^{-1}$	11
NiS/CdS	Xe-Vis (≥ 420 nm)	0.75M Na ₂ S-1.25M Na ₂ SO ₃	$401.7 \text{ mmol} \cdot \text{h}^{-1} \cdot \text{g}^{-1}$	This work

References

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