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Fig. S1.TOF Mass spectrum of Compound 1.



Fig. S2. ¹H NMR spectrum of Compound 1 (DMSO- d_6).

Supporting



Fig. S3. ¹³C NMR spectrum of Compound $1(DMSO-d_6)$.



















Fig. S9. ¹³C NMR spectrum of CuNI (DMSO- d_6).



Scheme S1. The synthetic route of CuNI.



Fig. S10. Fluorescence changes of **CuNI** (10 μ M) upon addition of Cu²⁺ (0.2~1 equiv.) in HOAc-NaOAc buffer solution (10 mM, pH 5, with 1% DMSO) at 25 °C. (I _{440 nm}, λ ex = 370 nm).



Fig. S11. The fluorescence intensity of CuNI(10 μ M) and CuNI (10 μ M)+ Cu²⁺ (5equiv.) in pH 3-10 solution at 25 °C (I _{440 nm}, λ ex = 370 nm).



Fig. S12. The fluorescence emission intensity of **CuNI** (10 mM)+ Cu²⁺ (5 equiv.) content in HOAc-NaOAc buffer solution (10 mM, pH 5.0, 1% DMSO) at 25°C.



Fig. S13. ¹H NMR spectra of (1) **CuNI** in DMSO-d6, (2) **CuNI** in DMSO-d6 + Cu²⁺ (0.2 equiv.) in D₂O, (3) **CuNI** in DMSO-d6 + Cu²⁺ (0.4 equiv.) in D₂O, (4) **CuNI** in DMSO-d6 + Cu²⁺ (0.6 equiv.) in D₂O, (5) **CuNI** in DMSO-d6 + Cu²⁺ (0.8 equiv.) in

D₂O, (6) **CuNI** in DMSO-d6 + Cu²⁺ (1.0 equiv.) in D₂O, (7) **CuNI** in DMSO-d6 + Cu²⁺ (1.2 equiv.) in D₂O, (8) **CuNI** in DMSO-d6 + Cu²⁺ (1.6 equiv.) in D₂O, (9) **CuNI** in DMSO-d6 + Cu²⁺ (2.0 equiv.) in D₂O.







Fig. S15. (a) TEM of CuNI (10 μ M) added with Cu²⁺ (5 equiv.) in HOAc-NaOAc buffer solution (10 mM, pH 5.0, with 1% DMSO). (b) Expansion of individual

particles



Fig S16. Cytotoxicity data of CuNI (HepG2 cells incubated for 24 h).