## **Supporting Information**

## **Controllable Polymeric Pseudo-Crown Ether Fluorescent Sensors: Responsiveness and Selective Detection of Metal Ions**

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Figure S1. MALDI-TOF mass spectrum of copolymer PDSM1 (top) and an enlarged zone (bottom)



Figure S2. MALDI-TOF mass spectrum of copolymer PDSM2 (top) and an enlarged zone (bottom)



Figure S3. MALDI-TOF mass spectrum of copolymer PDSM3 (top) and an enlarged zone (bottom)



Figure S4. GPC curve of **PDSM1** 



Figure S5. GPC curve of PDSM2



Figure S7. <sup>1</sup>H NMR spectra of PDSM-PEG-F (DMSO-*d*<sub>6</sub>, 25 <sup>o</sup>C)



**Figure S8.** <sup>1</sup>H NMR spectrum of **PDSM2-PEG1-F1** (CDCl<sub>3</sub>, 25 <sup>o</sup>C)



Figure S9. <sup>1</sup>H NMR spectrum of PDSM3-PEG1-F1 (CDCl<sub>3</sub>, 25 <sup>o</sup>C)



Figure S10. FT-IR spectrum of the grafting cyclopolymer PDSM2-PEG1-F1



Figure S11. FT-IR spectrum of the grafting cyclopolymer PDSM3-PEG1-F1



**Figure S12.** UV-vis spectra of 1-Pyrenemethylamine hydrochloride and **PDSM-PEG-F** (THF :  $H_2O = 1 : 1$ , [1-Pyrenemethylamine hydrochloride] =  $1 \times 10^{-4}$  mol/L, [**PDSM-PEG-F**] =  $2.5 \times 10^{-2}$  g/L)



Figure S13. TGA profiles for the different PDSM-PEG-F copolymers



Figure S14. TGA profiles for the different PDSM2-PEG1-F1 copolymers



Figure S15. TGA profiles for the different PDSM3-PEG1-F1 copolymers



**Figure S16.** Fluorescent spectra of pyrene after adding  $Cu^{2+}$  under room-temperature (pH = 2.0,  $\lambda$  excitation = 345 nm, slit width = 5 nm, [Pyrene] =  $10^{-7}$  M)



**Figure S17.** Interference of Metal ions on the detection of **PDSM1-PEG1-F1** to  $Cu^{2+}$  or  $Ag^+$ , Mixed ions: K<sup>+</sup>, Na<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup> and Fe<sup>3+</sup>, 10  $\mu$ M to each metal ion.



Figure S18. The GPC of polystyrene-*co*-polymaleic anhydride (PSMA).



Figure S19. The <sup>1</sup>H NMR of PSMA-PEG2-F1 (CDCl<sub>3</sub>).



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Figure S20. The IR of PSMA-PEG2-F1.