## **Supplementary Information**

## Superior UV- light photocatalysts of Nano-Crystalline (Ni or Co) FeWO<sub>4</sub>: Structure, Optical Characterization Synthesized by A Microemulsion Method

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**Fig. S1.** Flow chart for the production of FeWO<sub>4</sub>, CoFeWO<sub>4</sub>, and NiFeWO<sub>4</sub> by microemulsion method.



Fig. S2. VSM of FeWO<sub>4</sub>, CoFeWO<sub>4</sub> and NiFeWO<sub>4</sub> thin film.



Fig. S3. MB degradation by generated OH and  $O_2^-$  radicals.



**Fig. S4**. Recycling experiment of  $FeWO_4$ ,  $CoFeWO_4$ , and  $NiFeWO_4$  samples on the photodegradation MB.

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Compound	Assigned as	Metal ion salt (mole)/ Isooctane/water/AO T	Mixed Reacta nt	Molar ratio	Assigned as	Heat treatment temperature (°C)	Heat treatment Time (hr)
$(Na_2WO_4 \cdot 2H_2O)$	Reactant A	0.1/25/25/0.1	A:B	1:1.2	FeWO <sub>4</sub>	600/2°C/min	6
$(Fe(NO_3)_2)$	Reactant B	0.1 or 0.12/25/25/0.1	A:B	1:1	FeWO <sub>4</sub>	600/2°C/min	6
$(Co(NO_3)_2 \cdot 6H_2O)$	Reactant C	0.1:0.1/25/25/0.1	C:A:B	0.5:0.5:1	CoFeWO 4	600/2°C/min	6
$(Ni(NO_3)_2 \cdot 6H_2O)$	Reactant D	0.1/25/25/0.1	D:A:B	0.5:0.5:1	NiFeWO 4	600/2°C/min	6

**Table S1.** Molar ratios of chemicals mandatory for the different metal tungstate precursors to be prepared.

**Table S2**. Comparison of performance characteristics of  $MWO_4$  photocatalysts for the degradation of different pollutants.

Photocatalyst	Target pollutant	Light illumination	Photocatalytic activity	degradation time	Reference
NiFeWO <sub>4</sub>	MB	UV	100%	60 min	Current work
NiWO <sub>4</sub>	MO dye	UV	87%	100 min	1
ZnWO <sub>4</sub>	para-aminobenzoic acid	UV-A	100 %	160 min	2
ZnWO <sub>4</sub>	MB Rh-B	UV	100%	60 min 25 min	3
BaWO4	Methyl thioninium chloride (MTC) dye	UV	75% at pH=10	30 min	4
CuWO <sub>4</sub>	MO	UV	75%	90 min	5
$Pr_2(WO_4)_3$	MB	UV	99.9%	60 min	6
CuWO <sub>4</sub> /ZnO	MB	sunlight	98.9	120 min	7

## **Supplementary references**

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