## Magnetic-responsible Pickering emulsion and its catalytic application at water—oil interface

Shenghua Ma<sup>#1,2</sup>, Wei Zong<sup>#1,3</sup>, Xiaojun Han\*<sup>1</sup>

E-mail addresses: hanxiaojun@hit.edu.cn

# These authors contributed equally to this work

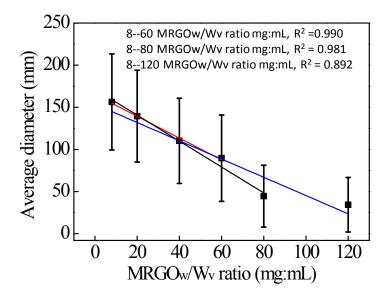


Fig.S1 The size of PE as a function of MRGO<sub>w</sub>/W<sub>v</sub> ratio

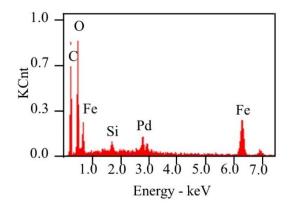


Fig.S2 The EDS spectroscopy spectrum of MRGO-Pd

<sup>&</sup>lt;sup>1</sup> State Key Laboratory of Urban Water Resource and Environment School of Chemistry and Chemical Engineering, Harbin Institute of Technology, Harbin, 150001, China,

<sup>&</sup>lt;sup>2</sup> Institute of Photonics & Photon-Technology, Northwest University, Xi'an, 710069, China

<sup>&</sup>lt;sup>3</sup> College of Chemistry and Chemical Engineering, Qiqihar University, Qiqihar 161006, China

<sup>\*</sup> Corresponding authors.

Table S1 the reaction rate constant at various catalyst dosage

Catalyst dosage	k	$R^2$
(g/L)	(min <sup>-1</sup> )	K²
0.16	0.0057	0.9830
0.4	0.0113	0.9986
0.8	0.0165	0.9755
1.2	0.0542	0.9971
1.6	0.0909	0.9944

Table S2 the reaction rate constant at various H<sub>2</sub>O<sub>2</sub> concentration

H <sub>2</sub> O <sub>2</sub> concentration	k	$R^2$
(g/L)	(min <sup>-1</sup> )	K²
0.5	0.0052	0.9722
0.9	0.0142	0.9882
1.3	0.0238	0.9895
1.9	0.0542	0.9971
2.4	0.0180	0.9961

Table S3 the reaction rate constant at various sudan-4 concentration

Sudan-4 concentration	k	$R^2$
(mM)	(min <sup>-1</sup> )	Λ-
0.25	0.0702	0.9955
0.54	0.0542	0.9971
0.79	0.022	0.9839
1.63	0.0115	0.9780

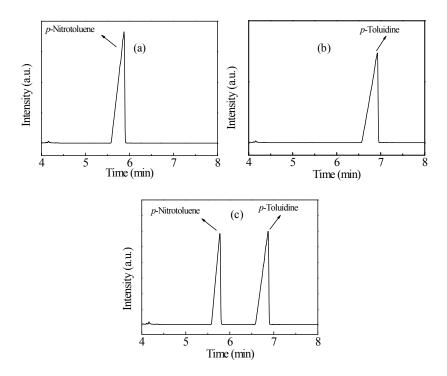


Fig.S3 Representative chromatograms of hydrogenation of *p*-nitrotoluene (a) *p*-nitrotoluene; (b) *p*-toluidine; (c) Products during the hydrogenation of *p*-nitrotoluene catalyzed by MRGO-Pd stabilized Pickering emulsions

Table S4 the reaction rate constant at various catalyst dosage

		· · · · · · · · · · · · · · · · · · ·
catalyst dosage (g/L)	k ( min <sup>-1</sup> )	$R^2$
0.6	0.0249	0.9674
1	0.0386	0.9631
2	0.1963	0.9945
3	0.2668	0.9765

Table S5 the reaction rate constant at various NaBH<sub>4</sub> concentration

NaBH <sub>4</sub> concentration	k	$R^2$
(g/L)	(min <sup>-1</sup> )	K²
5	0.0230	0.9833
10	0.0531	0.9951
20	0.1963	0.9945
40	0.2001	0.9947