

Supporting Information

A Novel Photochemical Sensor Based on Quinoline Functionalized Phenazine Derivatives for Multiple Substrates Detection

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Calculation formula of LOD

Linear Equation: $Y = aX + b$

$$\delta = \sqrt{\frac{\sum_{i=1}^N (F_i - \bar{F})^2}{N - 1}} \quad (N = 20)$$

$$\text{LOD} = K \times \delta / S \quad (K = 3, S = a \times 10^6)$$

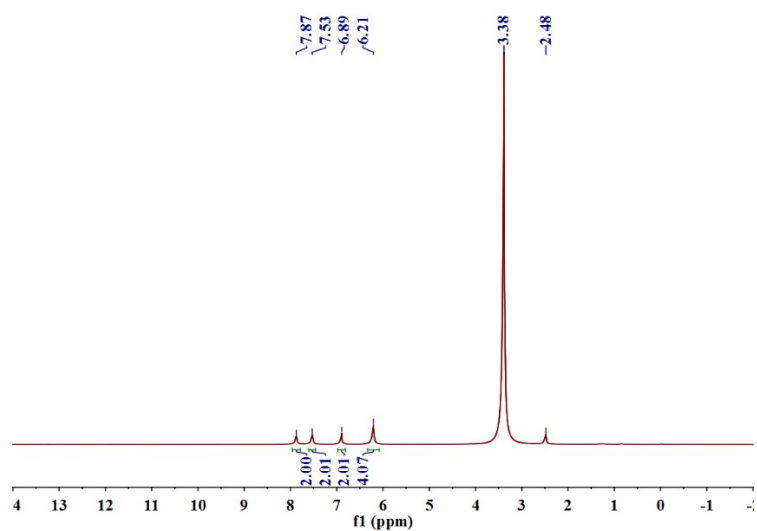


Fig. S1 ^1H NMR spectrum of **1** in $\text{DMSO-}d_6$

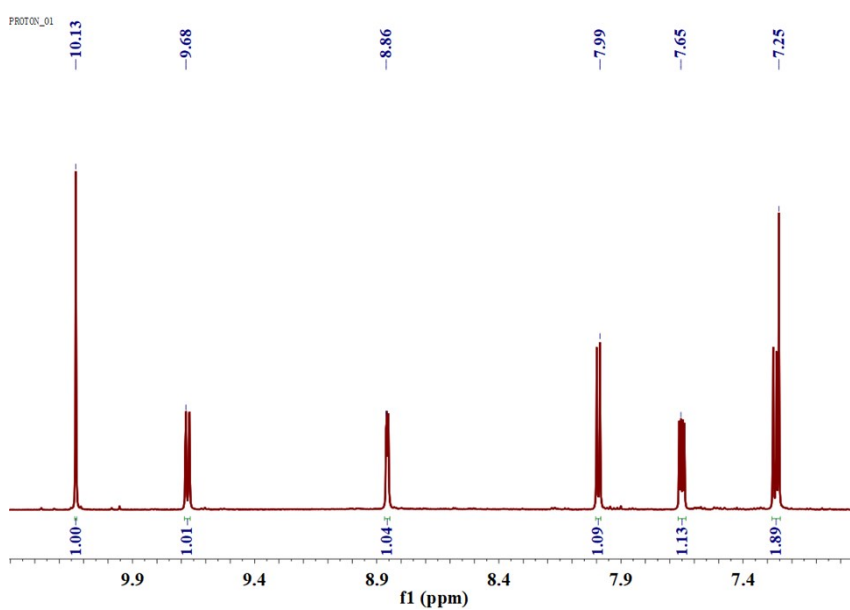


Fig. S2 ^1H NMR spectrum of **2** in CDCl_3

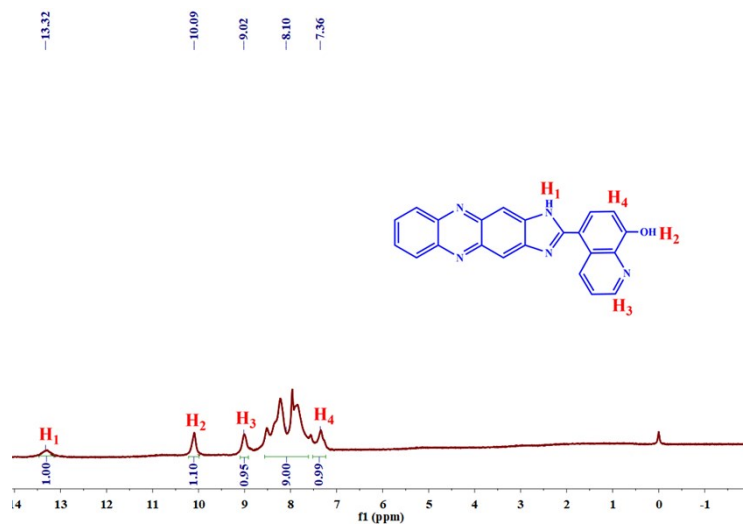


Fig. S3 ^1H NMR spectrum of **FQ-5** in $\text{DMSO}-d_6$

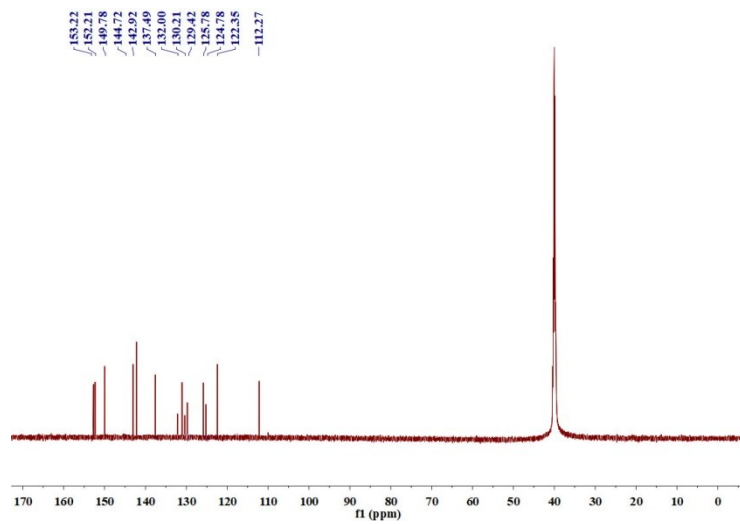


Fig. S4 ^{13}C NMR spectrum of **FQ-5** in $\text{DMSO}-d_6$

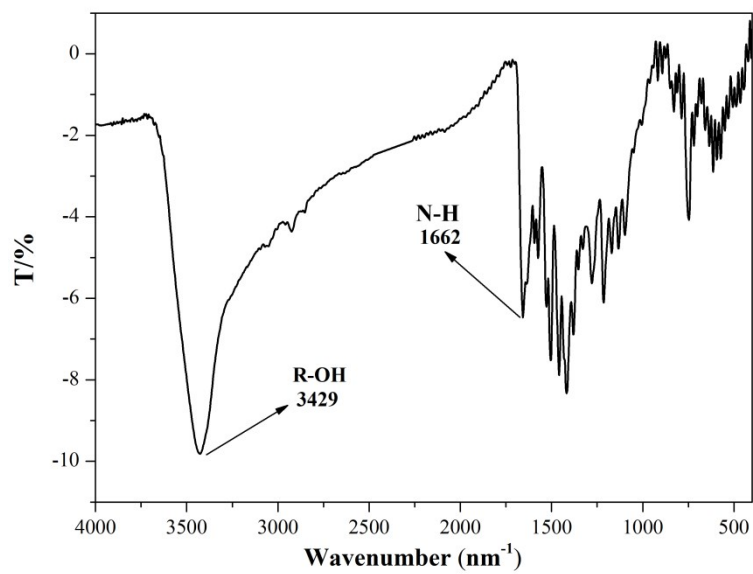


Fig. S5 IR spectrum of FQ-5 in KBr disks

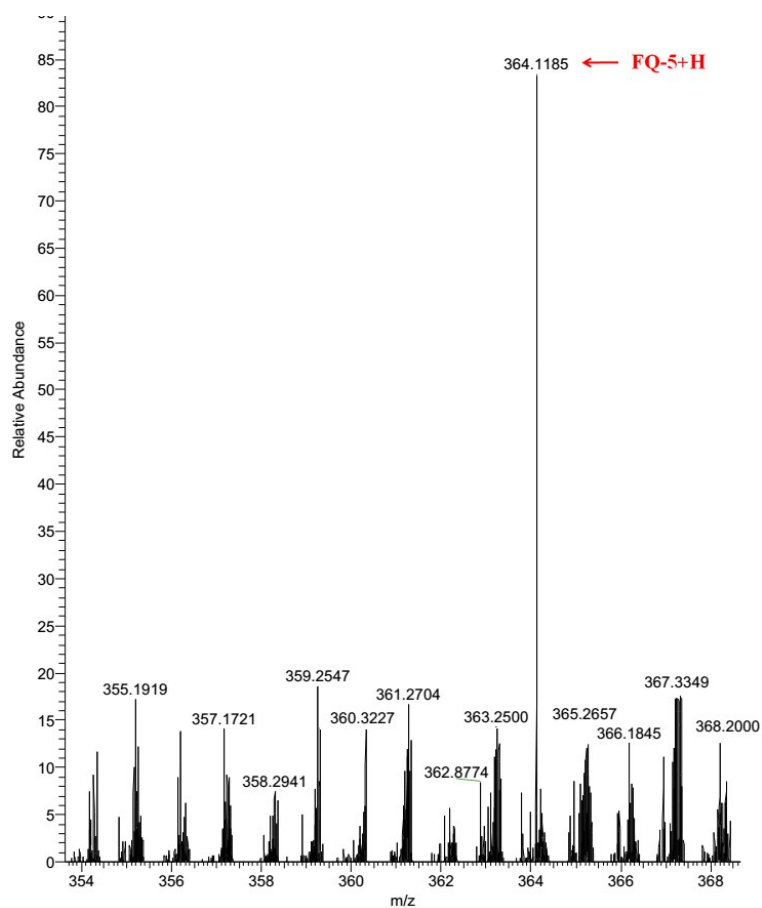


Fig. S6 ESI-MS spectrum of FQ-5

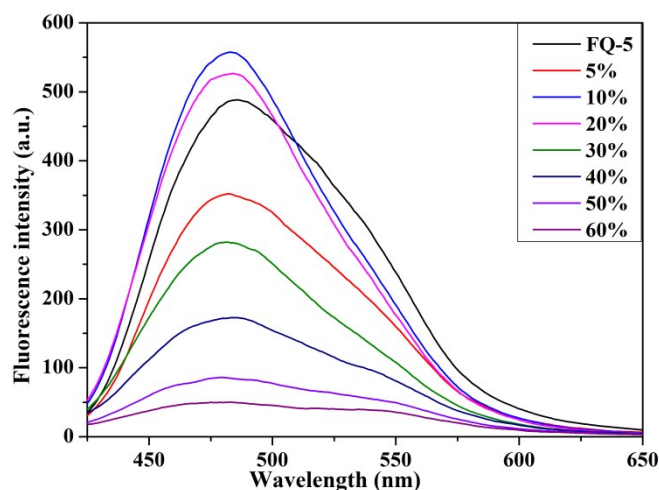


Fig.S7 Fluorescence emission spectra of **FQ-5** at different water ratios. ($\lambda_{\text{ex}} = 400 \text{ nm}$, $\lambda_{\text{em}} = 490 \text{ nm}$; $\text{ex}=5\text{nm}$, $\text{em}=10\text{nm}$)

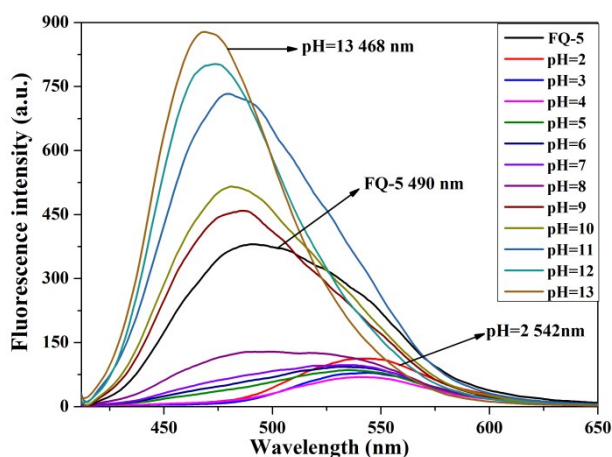


Fig. S8 Fluorescence spectra of **FQ-5** ($1.0 \times 10^{-4} \text{ M}$) in various pH values (pH ranged from 2.0 to 13.0 DMF/H₂O (9:1, v/v) HEPES buffered solution ($\lambda_{\text{ex}} = 400 \text{ nm}$, $\lambda_{\text{em}} = 490 \text{ nm}$, $\text{ex} = 5 \text{ nm}$, $\text{em} = 10 \text{ nm}$))

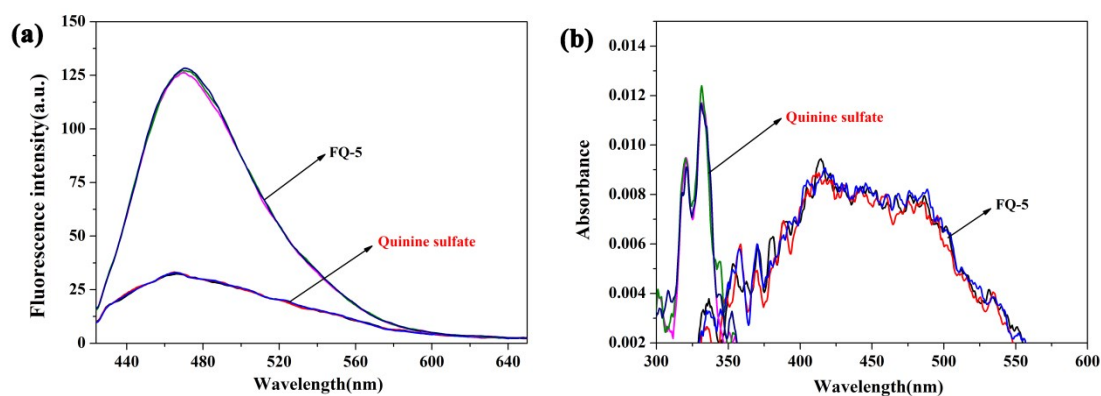


Fig. S9 Fluorescence quantum yield according to the corresponding formula, using quinine sulfate as standard ($\lambda_{\text{ex}} = 400 \text{ nm}$; $\lambda_{\text{em}} = 490 \text{ nm}$; $\text{ex}=5 \text{ nm}$; $\text{em}=10 \text{ nm}$.)

The fluorescence quantum yield of the sample was calculated using quinine sulfate as the standard

($\Phi_{\text{std}} = 0.55$). In this equation, Φ_{unk} and Φ_{std} are the fluorescence quantum yields of the sample and the standard, respectively; I_{unk} and I_{std} are the integral areas of the fluorescent spectra, respectively; A_{unk} and A_{std} are the absorbances of the sample and the standard at the excitation wavelength, respectively.

$$\text{pH}=12, \Phi_{\text{unk}} = \Phi_{\text{std}} \times (I_{\text{unk}} / I_{\text{std}}) \times (A_{\text{std}} / A_{\text{unk}}) \quad \Phi_{\text{std}} = 0.55$$

$$I_{\text{unk}}:11.10 \quad I_{\text{std}}: 10.89 \quad A_{\text{std}} :0.013 \quad A_{\text{unk}}: 0.009$$

$$\Phi_{\text{unk}} = 0.55 \times (11.10/10.89) \times (0.013/0.009) = 0.809$$

Fluorescence quantum yield: 80.9 %

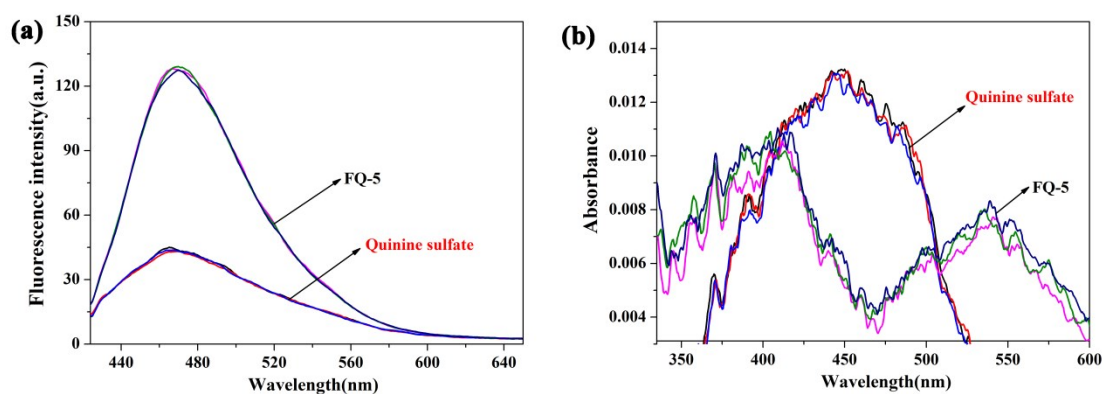


Fig. S10 Fluorescence quantum yield according to the corresponding formula, using quinine sulfate as standard ($\lambda_{\text{ex}} = 400 \text{ nm}$; $\lambda_{\text{em}} = 490 \text{ nm}$; $\text{ex}=5 \text{ nm}$; $\text{em}=10 \text{ nm}$.)

$$\Phi_{\text{unk}} = \Phi_{\text{std}} \times (I_{\text{unk}} / I_{\text{std}}) \times (A_{\text{std}} / A_{\text{unk}}) \quad \Phi_{\text{std}} = 0.55$$

$$I_{\text{unk}}:11.11 \quad I_{\text{std}}:10.87 \quad A_{\text{std}}: 0.014 \quad A_{\text{unk}}: 0.010$$

$$\Phi_{\text{unk}} = 0.55 \times (11.11/10.87) \times (0.014/0.010) = 0.77$$

Fluorescence quantum yield: 77.0 %



Fig.S11 (a) Photograph of **FQ-5** dispersed in different pH values (pH ranged from 2.0 to 13.0 DMF/H₂O (v/v, 9:1) HEPES buffered solution, $C_{\text{FQ-5}} = 1.0 \times 10^{-4}$ M) taken under UV light illumination ($\lambda_{\text{ex}} = 365$ nm); (b) Photograph of **FQ-5** dispersed in different pH values (pH ranged from 2.0 to 13.0 DMF/H₂O (v/v, 9:1) HEPES buffered solution, $C_{\text{FQ-5}} = 1.0 \times 10^{-4}$ M) taken under nature light.

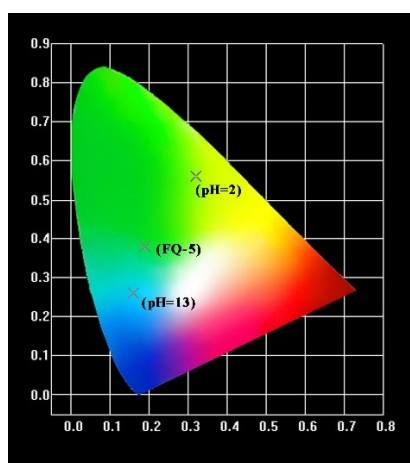


Fig.S12 The Commission Internationale de L'Eclairage (CIE) for **FQ-5**(1.0×10^{-4} M) in pH=2 and pH=13

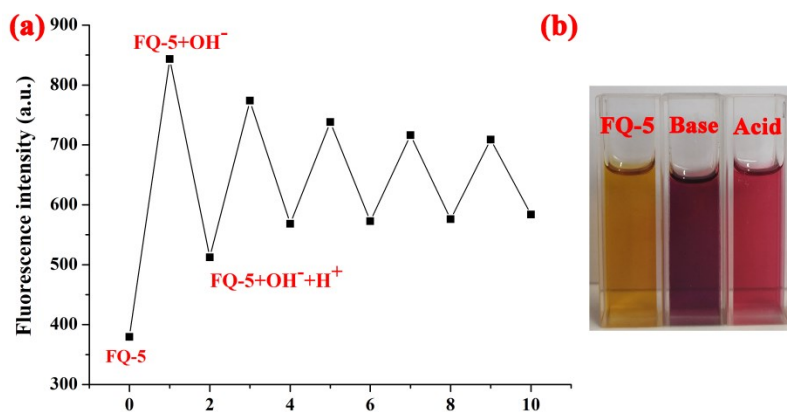


Fig.S13 (a) Fluorescence cycles of **FQ-5** in acid-base transformation; (b) Color changes observed for **FQ-5** upon the addition of acid-base.

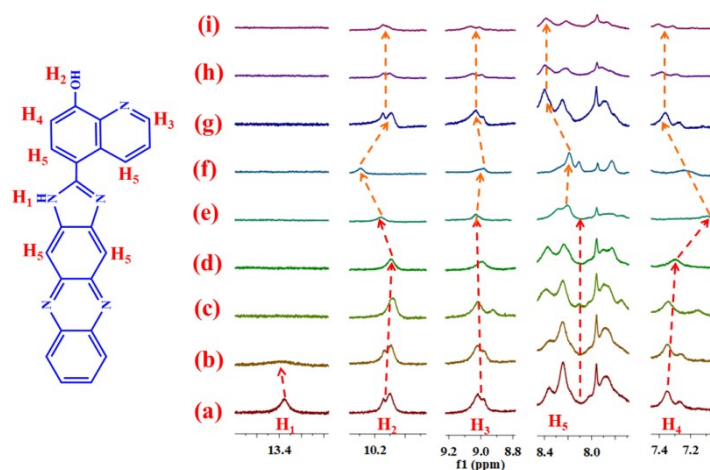


Fig.S14 Partial ^1H NMR spectra of **FQ-5** in $\text{DMSO-}d_6$ with different equiv. of H^+/OH^- (a). **FQ-5**. (b). 0.5 equiv. OH^- . (c). 1.0 equiv. OH^- (d). 2.0 equiv. OH^- (e). 4.0 equiv. OH^- (f). 0.5 equiv. H^+ (g). 1.0 equiv. H^+ (h). 2.0 equiv. H^+ (i). 4.0 equiv. H^+

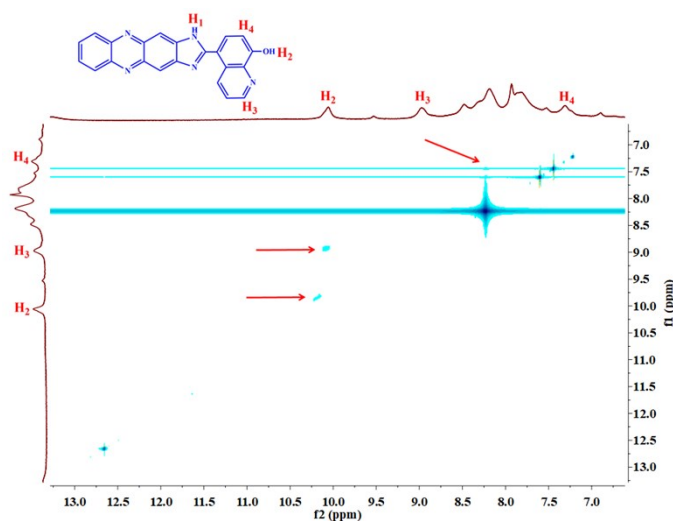


Fig. S15 2D NOESY spectrum of FQ-5 under acidic conditions (5mM) in DMSO-*d*6 solution

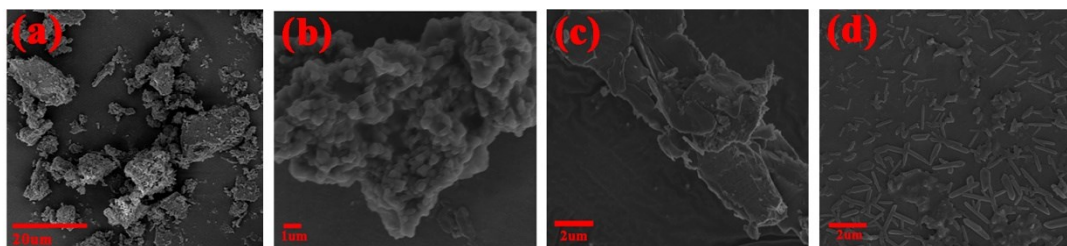


Fig.S16 SEM photos of (a-d) FQ-5; FQ-5 (pH=2); FQ-5 (pH=7); FQ-5 (pH=12).

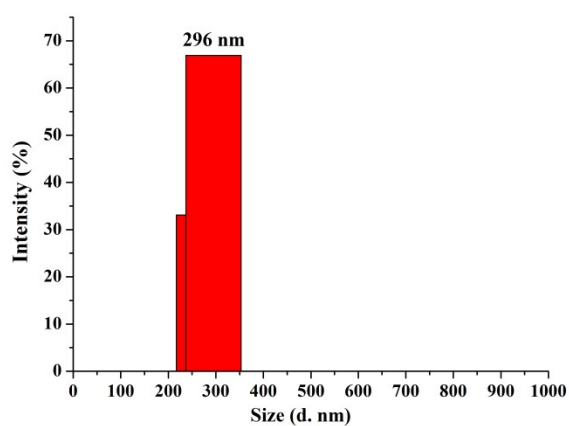


Fig.S17 DLS data of FQ-5-based aggregates in pH=2

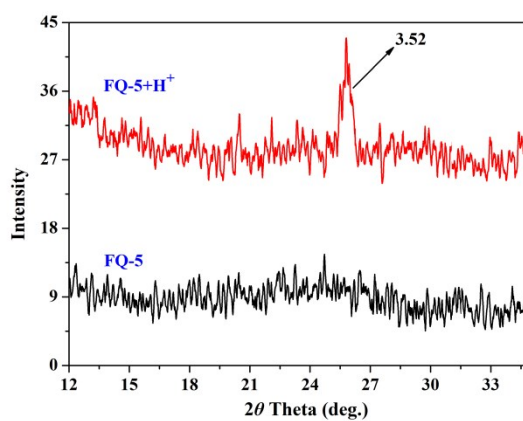


Fig.S18 The powder XRD patterns of the FQ-5 (black), FQ-5 (pH=2) (red)

Symbolic Z-matrix:

Charge = 0 Multiplicity = 1

| | | | |
|---|----------|----------|----------|
| C | -7.76825 | -0.41479 | 0.24252 |
| C | -6.63822 | -1.15009 | 0.50378 |
| C | -5.34592 | -0.59403 | 0.24938 |
| C | -5.24879 | 0.748 | -0.28429 |
| C | -6.44779 | 1.48108 | -0.54333 |
| C | -7.67184 | 0.91313 | -0.28632 |
| C | -2.93113 | 0.58741 | -0.28084 |
| C | -3.02778 | -0.76685 | 0.25613 |
| C | -1.85257 | -1.52223 | 0.52021 |
| H | -1.9435 | -2.52546 | 0.9156 |
| C | -0.6328 | -0.94176 | 0.25567 |
| C | -0.55128 | 0.40603 | -0.267 |
| C | -1.65877 | 1.17121 | -0.53993 |
| H | -8.74949 | -0.83605 | 0.43538 |
| H | -6.67985 | -2.15695 | 0.90347 |
| H | -6.34605 | 2.48365 | -0.94314 |
| H | -8.58233 | 1.46983 | -0.4832 |
| H | -1.62283 | 2.17936 | -0.93444 |
| N | -4.04773 | 1.31515 | -0.53966 |
| N | -4.24012 | -1.32622 | 0.50947 |
| N | 0.81605 | 0.64778 | -0.39152 |
| N | 0.66096 | -1.46403 | 0.41651 |
| C | 1.49772 | -0.50924 | 0.02539 |
| H | 1.23193 | 1.43427 | -0.86159 |
| C | 4.82419 | -2.20464 | -0.44837 |
| C | 5.73124 | -1.16772 | -0.33389 |
| C | 5.27285 | 0.14984 | -0.02924 |
| C | 3.87974 | 0.41748 | 0.14639 |
| C | 2.94976 | -0.66273 | -0.03823 |
| C | 3.45011 | -1.9403 | -0.30111 |

| | | | |
|---|---------|----------|----------|
| H | 5.18021 | -3.20394 | -0.66403 |
| C | 3.55649 | 1.74488 | 0.55578 |
| H | 2.73618 | -2.74865 | -0.40633 |
| C | 4.55275 | 2.69332 | 0.69731 |
| C | 5.89993 | 2.33732 | 0.44478 |
| H | 2.53217 | 2.00338 | 0.79605 |
| H | 4.31517 | 3.70176 | 1.01626 |
| H | 6.69711 | 3.06685 | 0.53876 |
| N | 6.25019 | 1.10001 | 0.10611 |
| O | 7.0745 | -1.36668 | -0.49237 |
| H | 7.52209 | -0.49551 | -0.34899 |

Fig.S19 Cartesian coordinates of **FQ-5**

Symbolic Z-matrix:

Charge = 0 Multiplicity = 2

| | | | |
|---|----------|----------|----------|
| C | -7.76825 | -0.41479 | 0.24252 |
| C | -6.63822 | -1.15009 | 0.50378 |
| C | -5.34592 | -0.59403 | 0.24938 |
| C | -5.24879 | 0.748 | -0.28429 |
| C | -6.44779 | 1.48108 | -0.54333 |
| C | -7.67184 | 0.91313 | -0.28632 |
| C | -2.93113 | 0.58741 | -0.28084 |
| C | -3.02778 | -0.76685 | 0.25613 |
| C | -1.85257 | -1.52223 | 0.52021 |
| H | -1.9435 | -2.52546 | 0.9156 |
| C | -0.6328 | -0.94176 | 0.25567 |
| C | -0.55128 | 0.40603 | -0.267 |
| C | -1.65877 | 1.17121 | -0.53993 |
| H | -8.74949 | -0.83605 | 0.43538 |
| H | -6.67985 | -2.15695 | 0.90347 |
| H | -6.34605 | 2.48365 | -0.94314 |

| | | | |
|---|----------|----------|----------|
| H | -8.58233 | 1.46983 | -0.4832 |
| H | -1.62283 | 2.17936 | -0.93444 |
| N | -4.04773 | 1.31515 | -0.53966 |
| N | -4.24012 | -1.32622 | 0.50947 |
| N | 0.81605 | 0.64778 | -0.39152 |
| N | 0.66096 | -1.46403 | 0.41651 |
| C | 1.49772 | -0.50924 | 0.02539 |
| C | 4.82419 | -2.20464 | -0.44837 |
| C | 5.73124 | -1.16772 | -0.33389 |
| C | 5.27285 | 0.14984 | -0.02924 |
| C | 3.87974 | 0.41748 | 0.14639 |
| C | 2.94976 | -0.66273 | -0.03823 |
| C | 3.45011 | -1.9403 | -0.30111 |
| H | 5.18021 | -3.20394 | -0.66403 |
| C | 3.55649 | 1.74488 | 0.55578 |
| H | 2.73618 | -2.74865 | -0.40633 |
| C | 4.55275 | 2.69332 | 0.69731 |
| C | 5.89993 | 2.33732 | 0.44478 |
| H | 2.53217 | 2.00338 | 0.79605 |
| H | 4.31517 | 3.70176 | 1.01626 |
| H | 6.69711 | 3.06685 | 0.53876 |
| N | 6.25019 | 1.10001 | 0.10611 |
| O | 7.0745 | -1.36668 | -0.49237 |
| H | 7.52209 | -0.49551 | -0.34899 |

Fig.S20 Cartesian coordinates of **FQ-5** under alkaline conditions

Symbolic Z-matrix:

Charge = 0 Multiplicity = 1

| | | | |
|---|----------|----------|---------|
| C | -6.48188 | -4.99486 | 5.56349 |
| C | -5.17085 | -4.79099 | 5.90756 |

| | | | |
|---|----------|----------|---------|
| C | -4.29903 | -4.08257 | 5.02604 |
| C | -4.81904 | -3.58772 | 3.77259 |
| C | -6.18973 | -3.81973 | 3.44974 |
| C | -6.99678 | -4.50399 | 4.32158 |
| C | -2.73303 | -2.70899 | 3.24675 |
| C | -2.20627 | -3.20854 | 4.51254 |
| C | -0.84578 | -2.99442 | 4.85765 |
| H | -0.47178 | -3.37449 | 5.79548 |
| C | -0.04862 | -2.30929 | 3.97215 |
| C | -0.585 | -1.81307 | 2.72387 |
| C | -1.8911 | -1.99715 | 2.34779 |
| H | -7.14196 | -5.53111 | 6.23205 |
| H | -4.75257 | -5.14859 | 6.83758 |
| H | -6.54954 | -3.43603 | 2.50578 |
| H | -8.03582 | -4.67864 | 4.07642 |
| H | -2.31605 | -1.63811 | 1.4222 |
| N | -4.0305 | -2.91193 | 2.90526 |
| N | -3.00766 | -3.88707 | 5.37519 |
| N | 0.49367 | -1.18028 | 2.11181 |
| N | 1.30993 | -1.97435 | 4.06617 |
| C | 1.6097 | -1.31028 | 2.95066 |
| H | 0.51006 | -0.81125 | 1.17849 |
| C | 5.34268 | -1.28683 | 2.86652 |
| C | 5.61556 | -0.18818 | 2.04893 |
| C | 4.58543 | 0.57018 | 1.49111 |
| C | 3.21494 | 0.28771 | 1.78167 |
| C | 2.95091 | -0.8496 | 2.61393 |
| C | 4.01908 | -1.60293 | 3.14148 |
| H | 6.15001 | -1.87697 | 3.28116 |
| C | 2.2474 | 1.21937 | 1.29361 |

| | | | |
|---|---------|----------|----------|
| H | 3.77602 | -2.44467 | 3.77187 |
| C | 2.62575 | 2.23324 | 0.38051 |
| C | 3.94495 | 2.44439 | 0.07972 |
| H | 1.20983 | 1.1199 | 1.56229 |
| H | 1.8758 | 2.88074 | -0.04741 |
| H | 4.29711 | 3.20232 | -0.59926 |
| N | 4.90838 | 1.64377 | 0.66576 |
| O | 6.91114 | 0.24553 | 1.71953 |
| H | 7.59425 | -0.36781 | 2.03301 |
| H | 5.93866 | 1.81745 | 0.39491 |
| H | 1.72401 | -2.37329 | 4.97962 |

Fig.S21 Cartesian coordinates of FQ-5 under acidic conditions

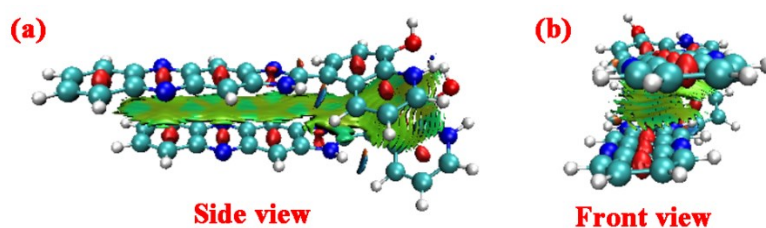


Fig. S22 The NCI graph dimers of FQ-5 under acidic conditions

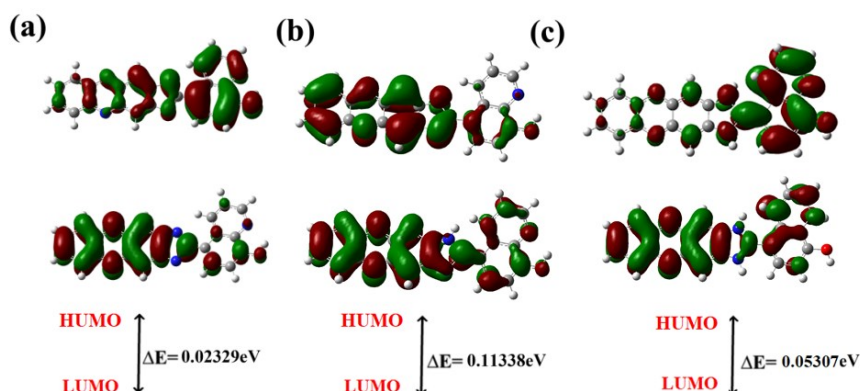


Fig.S23 The HOMO-LUMO energy gaps for (a) Alkaline; (b) FQ-5 (Neutral) and (c) Acidic conditions.

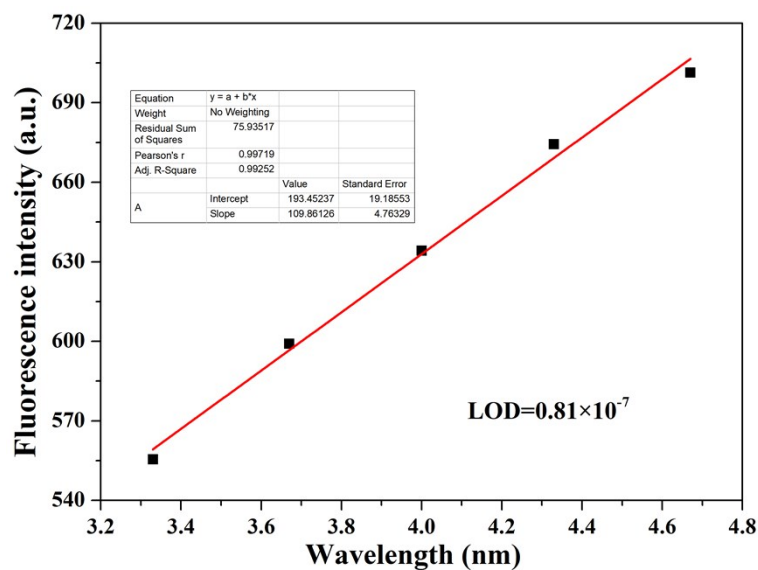


Fig. S24 Photograph of the linear range for L-Arg

The result of the analysis as follows:

$$\text{Linear Equation: } Y = 109.86X + 193.45, \quad R^2 = 0.9925,$$

$$S = 109.86 \times 10^6$$

$$\text{LOD} = K \times \delta/S = 0.81 \times 10^{-7} \text{ M} \quad (K = 3)$$

$$\text{LOD} = 0.81 \times 10^{-7} \text{ M}$$

Table S1

Comparison of detection limit of sensor for L-Arg with previously reported L-Arg sensors

| No | Journal, Year, Volume, Page | LOD (M) | Solvent | Ref. |
|----|--|----------------------|---------------------------------------|-------|
| 1 | <i>Sens. Actuators B: Chem.</i> , 2014, 192, 496-502 | 2.3×10^{-6} | CH ₃ OH/Tris buffer (pH=7) | 57(a) |
| 2 | <i>Chem. Commun.</i> , 2011, 47, 3921-3923 | 2.0×10^{-6} | HEPES (pH=7.4) | 57(b) |
| 3 | <i>Biosens. Bioelectron.</i> , 2017, 87, 772-778 | 3.4×10^{-8} | Phosphate buffer (pH=7.4) | 57(c) |
| 4 | <i>Macromol. Res.</i> , 2012, 20, 344-346 | 1.0×10^{-5} | Water | 57(d) |

| | | | | |
|---|---|-----------------------|--|-------|
| 5 | <i>Talanta</i> , 2012, 97, 16-22 | 2.3×10^{-6} | CH ₃ COOH-CH ₃ COONa buffer(pH=6.0) | 58(a) |
| 6 | <i>New J. Chem.</i> , 2020, 44, 4842-4849 | 1.5×10^{-7} | Tris/HCl buffer (pH=4.0) | 58(b) |
| 7 | <i>New J. Chem.</i> , 2017, 41, 15216-15228 | 2.85×10^{-8} | pH 7.4 adjusted solution | 58(c) |
| 8 | <i>Langmuir</i> , 2014, 30, 15364-1537 | 1.7×10^{-7} | HEPES buffe (pH=7.4) | 58(d) |
| | This work | 8.1×10^{-8} | DMF/Tris HEPES (pH=7.4) | |



Fig. S25 Photograph of **FQ-5** dispersed in different amino acids (pH =7 DMF/H₂O (v/v, 9:1) HEPES buffered solution, $C_{\text{FQ-5}} = 1.0 \times 10^{-4}$ M) taken under UV light illumination ($\lambda_{\text{ex}} = 365$ nm)

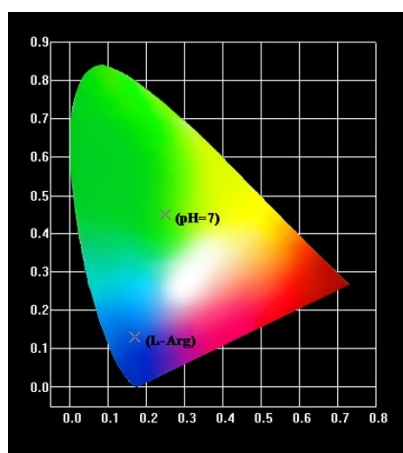


Fig. S26 The Commission Internationale de L'Eclairage (CIE) for **FQ-5** (1.0×10^{-4} M) in DMF/H₂O (9:1, v/v) binary solution

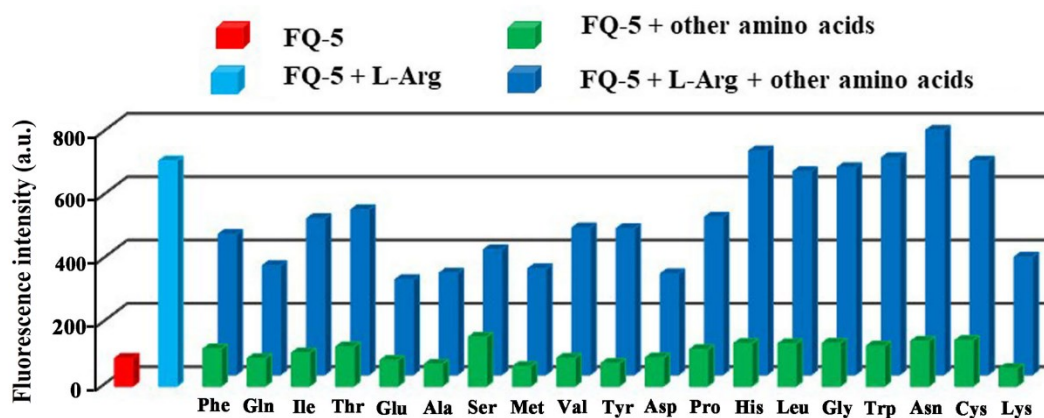


Fig. S27 Fluorescence intensity of **FQ-5** (1×10^{-4} M) and L-Arg (6 equiv.) in the presence of an excess (6 equiv.) of various other amino acids.

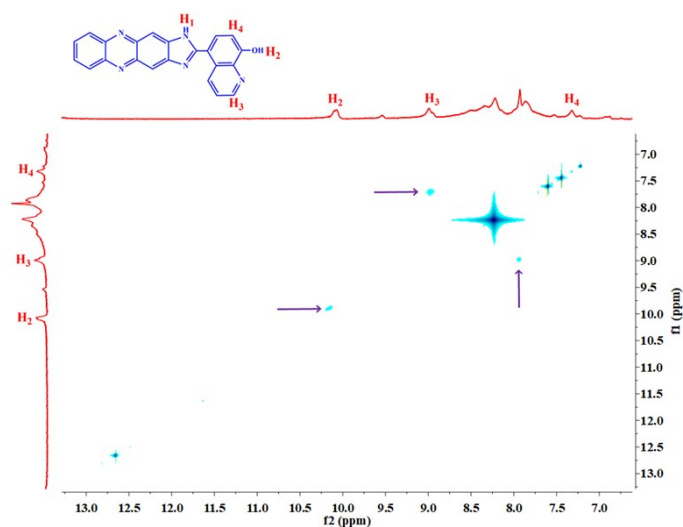


Fig.S28 2D NOESY spectrum of **L-Arg+ FQ-5**

Symbolic Z-matrix:

Charge = 0 Multiplicity = 2

| | | | |
|---|----------|----------|----------|
| C | -7.18342 | 0.16431 | -5.10334 |
| C | -5.89692 | 0.555 | -5.36682 |
| C | -4.80093 | -0.12491 | -4.75364 |

| | | | |
|---|----------|----------|----------|
| C | -5.06959 | -1.2255 | -3.85634 |
| C | -6.42364 | -1.60212 | -3.60674 |
| C | -7.4493 | -0.92538 | -4.21359 |
| C | -2.78901 | -1.51201 | -3.51624 |
| C | -2.51684 | -0.40078 | -4.4201 |
| C | -1.17988 | -0.00136 | -4.68464 |
| H | -0.99578 | 0.82447 | -5.35352 |
| C | -0.16279 | -0.68886 | -4.06925 |
| C | -0.44528 | -1.79473 | -3.18271 |
| C | -1.71902 | -2.21356 | -2.89528 |
| H | -8.01332 | 0.67908 | -5.56777 |
| H | -5.66346 | 1.37318 | -6.0325 |
| H | -6.59365 | -2.42811 | -2.93144 |
| H | -8.4749 | -1.2113 | -4.0238 |
| H | -1.95838 | -3.03399 | -2.2353 |
| N | -4.06358 | -1.89923 | -3.25428 |
| N | -3.53608 | 0.26849 | -5.01984 |
| N | 0.81087 | -2.23784 | -2.77104 |
| N | 1.22633 | -0.49531 | -4.15364 |
| C | 1.77734 | -1.41986 | -3.37822 |
| H | 0.98461 | -2.89811 | -2.03545 |
| C | 5.33551 | -0.3407 | -2.80641 |

| | | | |
|---|----------|----------|----------|
| C | 5.98455 | -1.52281 | -2.51904 |
| C | 5.27121 | -2.75594 | -2.56233 |
| C | 3.88101 | -2.78411 | -2.88543 |
| C | 3.21179 | -1.53576 | -3.12098 |
| C | 3.96114 | -0.3605 | -3.10177 |
| H | 5.88769 | 0.58633 | -2.78756 |
| C | 3.3043 | -4.08155 | -3.00557 |
| H | 3.44827 | 0.56572 | -3.31433 |
| C | 4.06179 | -5.20678 | -2.75406 |
| C | 5.4199 | -5.06902 | -2.38543 |
| H | 2.28045 | -4.18899 | -3.33127 |
| H | 3.63237 | -6.19327 | -2.85269 |
| H | 6.03247 | -5.93348 | -2.17171 |
| N | 6.0109 | -3.8816 | -2.30571 |
| O | 7.31598 | -1.55199 | -2.20198 |
| H | 7.57038 | -2.4908 | -2.05473 |
| N | -2.31987 | -1.14258 | 0.50968 |
| H | -2.96721 | -1.34671 | -0.22468 |
| C | -1.32805 | -0.10119 | 0.33248 |
| H | -1.46594 | 0.66603 | 1.09435 |
| C | 0.08476 | -0.66192 | 0.45566 |
| C | -1.45326 | 0.54709 | -1.03885 |

| | | | |
|---|----------|----------|----------|
| H | 0.2161 | -1.10356 | 1.44348 |
| H | 0.24106 | -1.42513 | -0.30669 |
| C | 1.09656 | 0.46278 | 0.26344 |
| O | -2.31511 | 0.17107 | -1.83022 |
| O | -0.69404 | 1.45449 | -1.3715 |
| H | 0.96614 | 0.90463 | -0.72441 |
| H | 0.94118 | 1.22621 | 1.02576 |
| C | 2.50938 | -0.09795 | 0.38662 |
| H | 2.64071 | -0.53959 | 1.37444 |
| H | 2.66567 | -0.86116 | -0.37573 |
| N | 3.52053 | 0.96746 | 0.20528 |
| H | 3.20176 | 1.90924 | 0.0277 |
| C | 4.82979 | 0.74137 | 0.2655 |
| N | 5.3258 | -0.47109 | 0.49535 |
| N | 5.65037 | 1.77283 | 0.08768 |
| H | 4.69988 | -1.25224 | 0.62995 |
| H | 6.32612 | -0.60509 | 0.53424 |
| H | 5.27043 | 2.69226 | -0.08667 |
| H | 6.6497 | 1.63196 | 0.12782 |
| H | -2.35701 | -1.65715 | 1.36632 |
| H | -1.55229 | -1.62247 | 0.08479 |
| C | -6.51253 | 0.51321 | 3.10079 |

| | | | |
|---|----------|----------|---------|
| C | -5.30286 | 0.14077 | 3.62531 |
| C | -4.14126 | 0.93589 | 3.38466 |
| C | -4.26213 | 2.13288 | 2.58389 |
| C | -5.53958 | 2.48701 | 2.05446 |
| C | -6.63177 | 1.69853 | 2.3067 |
| C | -1.99166 | 2.54776 | 2.85159 |
| C | -1.86823 | 1.3395 | 3.65829 |
| C | -0.60934 | 0.96019 | 4.19523 |
| H | -0.53453 | 0.06194 | 4.78753 |
| C | 0.47586 | 1.76012 | 3.93391 |
| C | 0.33865 | 2.96042 | 3.14071 |
| C | -0.85485 | 3.36434 | 2.59958 |
| H | -7.39215 | -0.08882 | 3.28314 |
| H | -5.18 | -0.74761 | 4.22759 |
| H | -5.59922 | 3.3868 | 1.45946 |
| H | -7.59922 | 1.96749 | 1.90506 |
| H | -0.98495 | 4.25399 | 2.00161 |
| N | -3.18991 | 2.91702 | 2.33107 |
| N | -2.95302 | 0.55959 | 3.90615 |
| N | 1.62337 | 3.5008 | 3.10568 |
| N | 1.81427 | 1.61021 | 4.33444 |
| C | 2.47013 | 2.6481 | 3.83213 |

| | | | |
|---|---------|---------|---------|
| H | 1.92246 | 4.25732 | 2.51792 |
| C | 6.12713 | 1.78867 | 4.0317 |
| C | 6.74339 | 3.02196 | 4.05323 |
| C | 5.95754 | 4.21093 | 4.06347 |
| C | 4.53185 | 4.1439 | 4.0354 |
| C | 3.91301 | 2.85097 | 3.95131 |
| C | 4.724 | 1.71757 | 3.97986 |
| H | 6.73005 | 0.89366 | 4.04108 |
| C | 3.8555 | 5.39179 | 4.16076 |
| H | 4.23997 | 0.75273 | 3.9498 |
| C | 4.57211 | 6.56818 | 4.23682 |
| C | 5.98513 | 6.53132 | 4.20037 |
| H | 2.77856 | 5.41691 | 4.23464 |
| H | 4.0649 | 7.51638 | 4.34138 |
| H | 6.57024 | 7.43891 | 4.24663 |
| N | 6.65809 | 5.38772 | 4.1303 |
| O | 8.10677 | 3.14277 | 4.07934 |
| H | 8.32397 | 4.10175 | 4.11008 |

Fig.S29 Cartesian coordinates of **FQ-5+L-Arg**

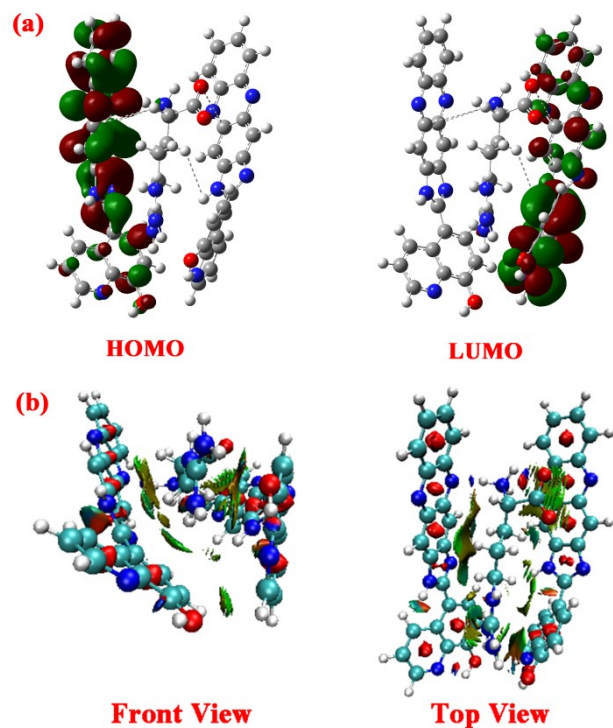


Fig.S30 (a) The HOMO-LUMO energy gaps for **FQ-5+L-Arg**, (b) The NCI graph of **FQ-5+L-Arg**.

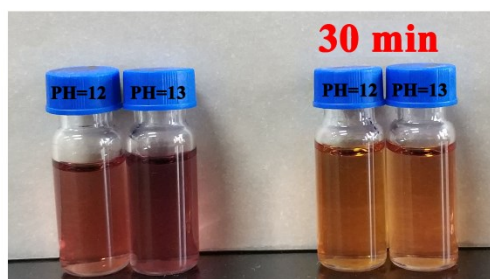


Fig. S31 UV-Vis Photos of = 12/13 solution before and after being placed in the air for 30 min



Fig. S32 Carbon dioxide reactor, please refer to the multimedia file for details

Equipment: Three-necked flask, constant pressure funnel, catheter, HCl (dilute), CaCO₃, beaker, magneton, stirrer, FQ-5(1.0 × 10⁻⁴ M, DMF/H₂O (9:1, v/v)) and HEPES buffered solution (pH=12/13)

Describe: First, we designed a carbon dioxide reactor through chemical reactions: CaCO₃+2HCl (dilute) =CaCl₂+CO₂+H₂O. A small amount of CaCO₃ is placed in a three-necked flask with one end sealed and the other end inserted into a catheter. In addition, pour a little into HCl (dilute) the constant pressure funnel. Then, the funnel pistons and stirrer were opened to allow the reaction to take place, and the catheter was passed into FQ-5 with pH 12/13, and it was obvious that the solution changed from purple to orange.

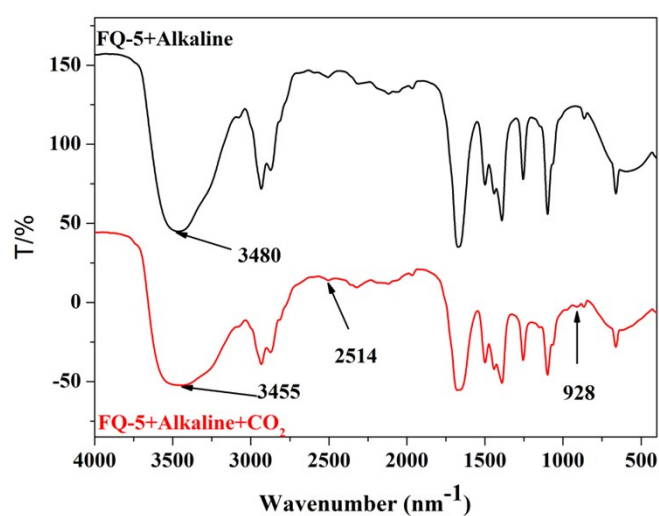


Fig. S33 FT-IR spectroscopy of powder FQ-5-OH and FQ-5-OH +CO₂

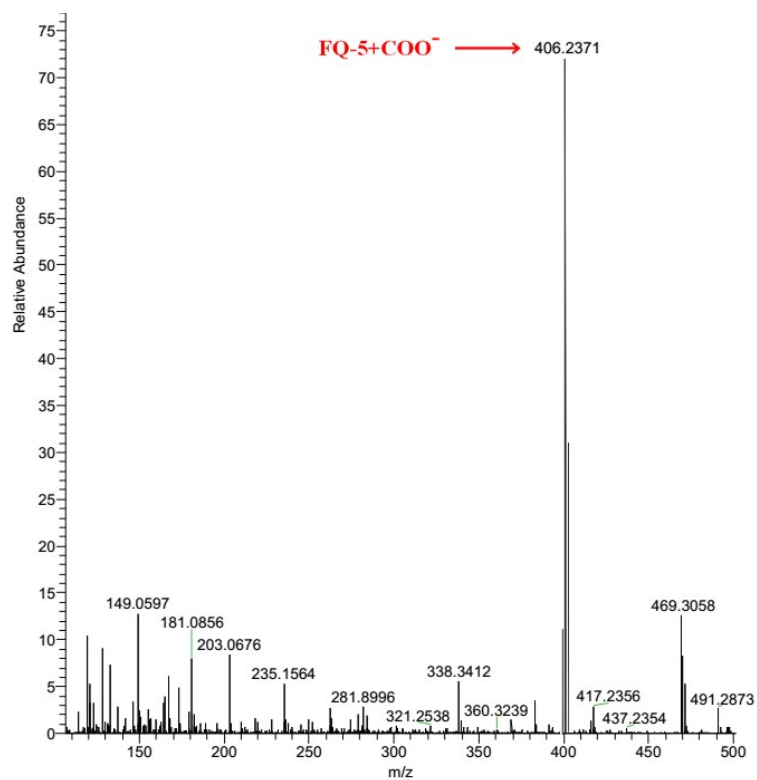


Fig. S34 ESI-MS spectrum of FQ-5+COO⁻