

## Supporting Information

### **Coupling amorphous cobalt hydroxide nanoflakes on $\text{Sr}_2\text{Fe}_{1.5}\text{Mo}_{0.5}\text{O}_{5+\delta}$ perovskite nanofibers to induce bifunctionality for water splitting**

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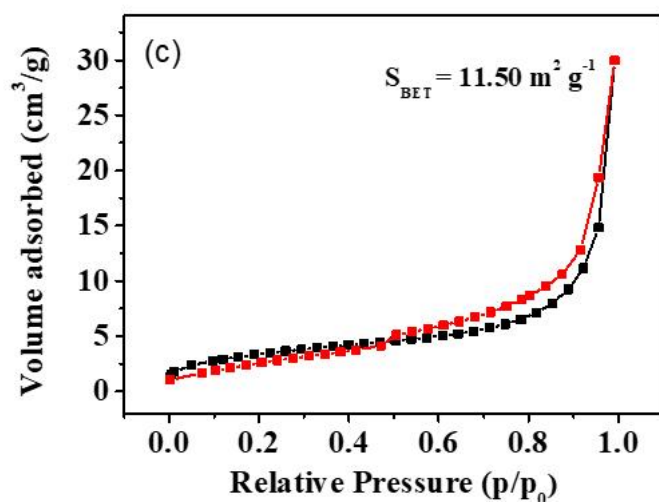
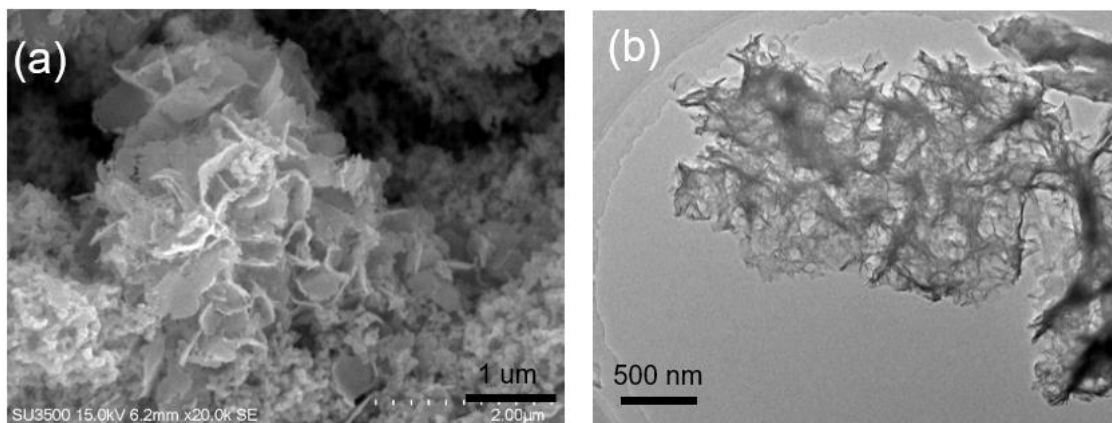
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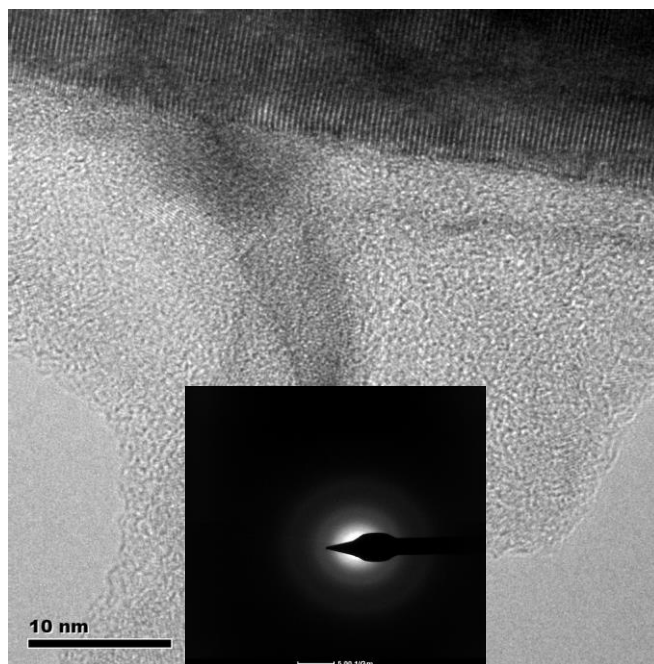
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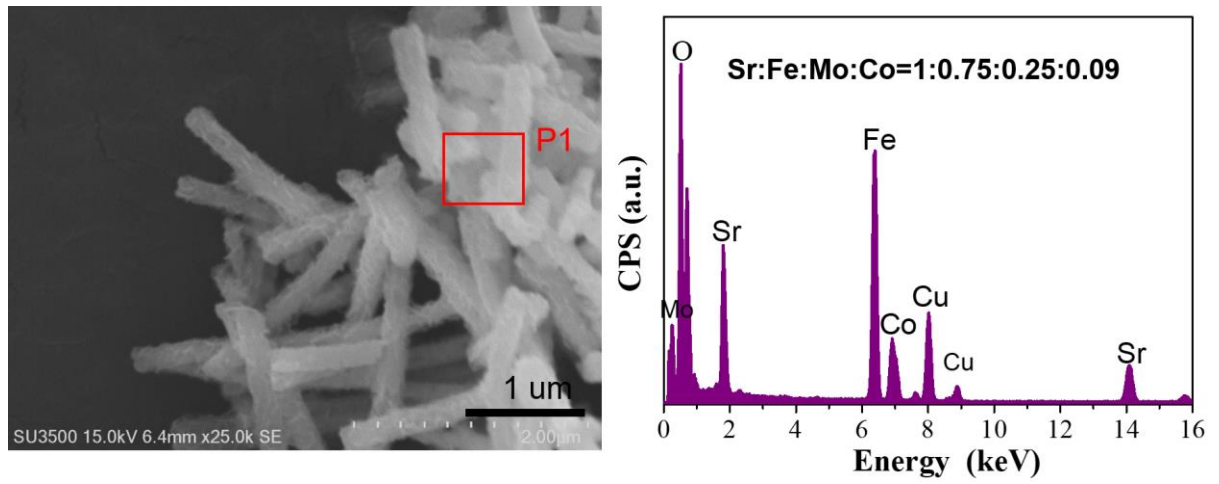
[zhaoling@cug.edu.cn](mailto:zhaoling@cug.edu.cn) (Ling Zhao)



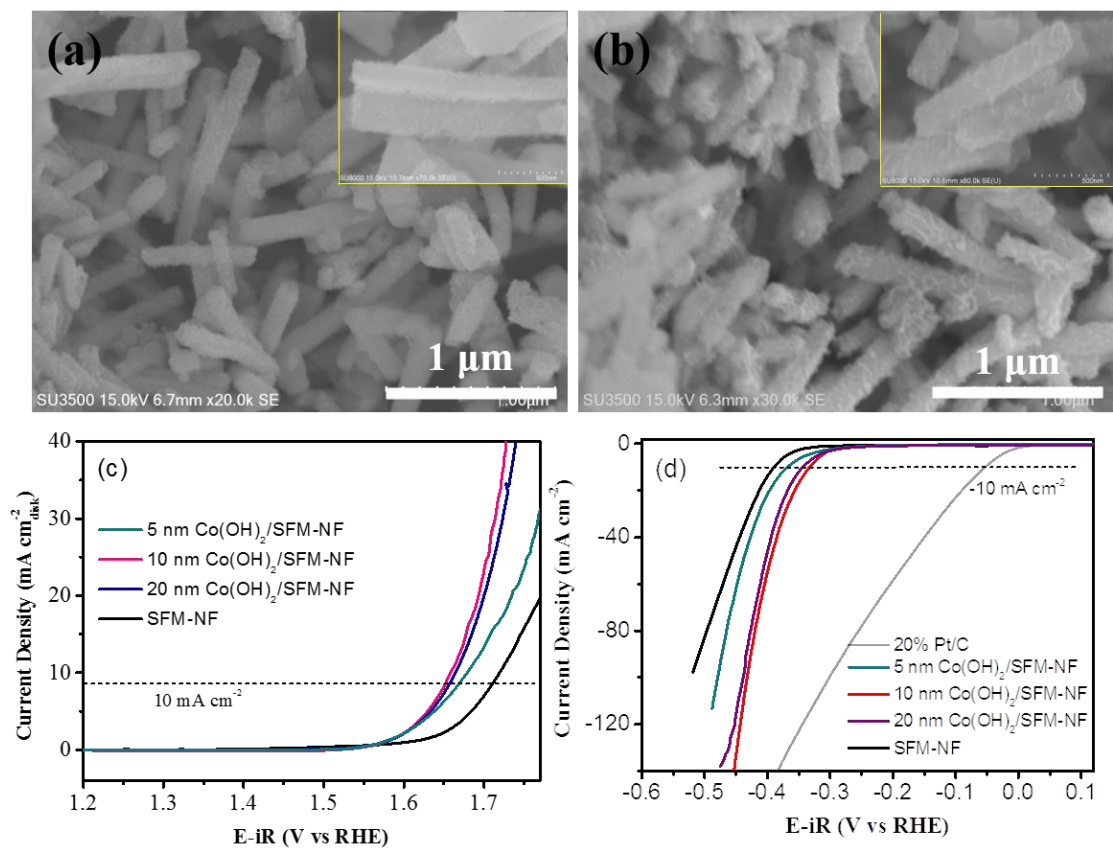
**Figure S1.** (a) SEM and (b) TEM images of Co(OH)<sub>2</sub>, (c) N<sub>2</sub> adsorption-desorption isotherm of Co(OH)<sub>2</sub>.



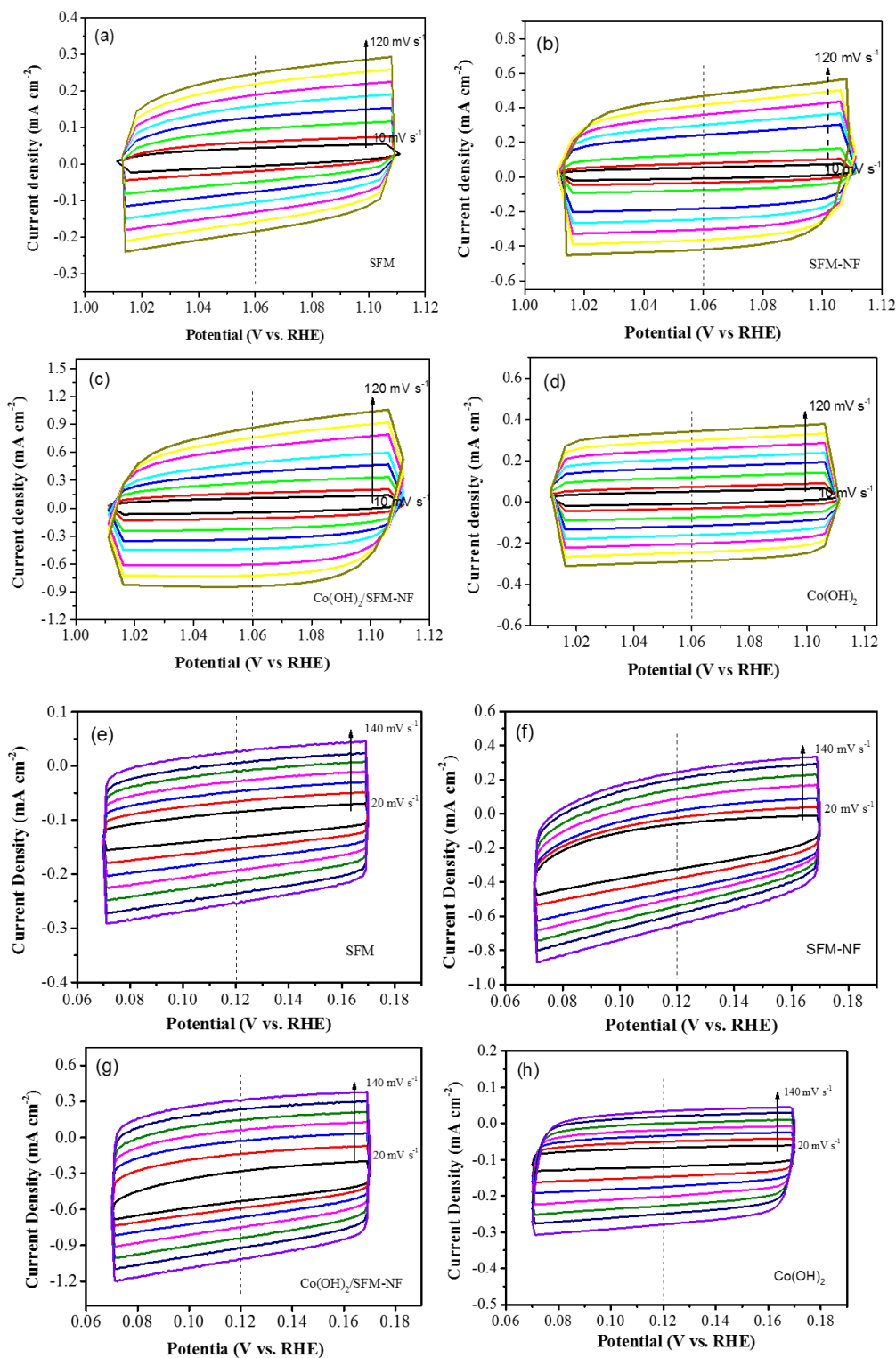
**Figure S2.** TEM images of Co(OH)<sub>2</sub>/SFM-NF sample (inset showing the corresponding SAED pattern).



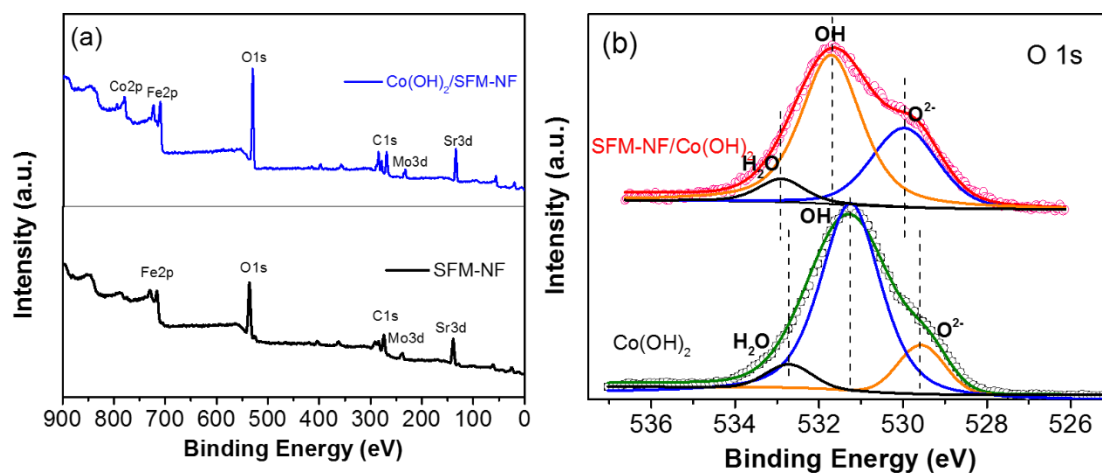
**Figure S3.** EDS spectrum of  $\text{Co(OH)}_2/\text{SFM-NF}$  sample measured at position P1.



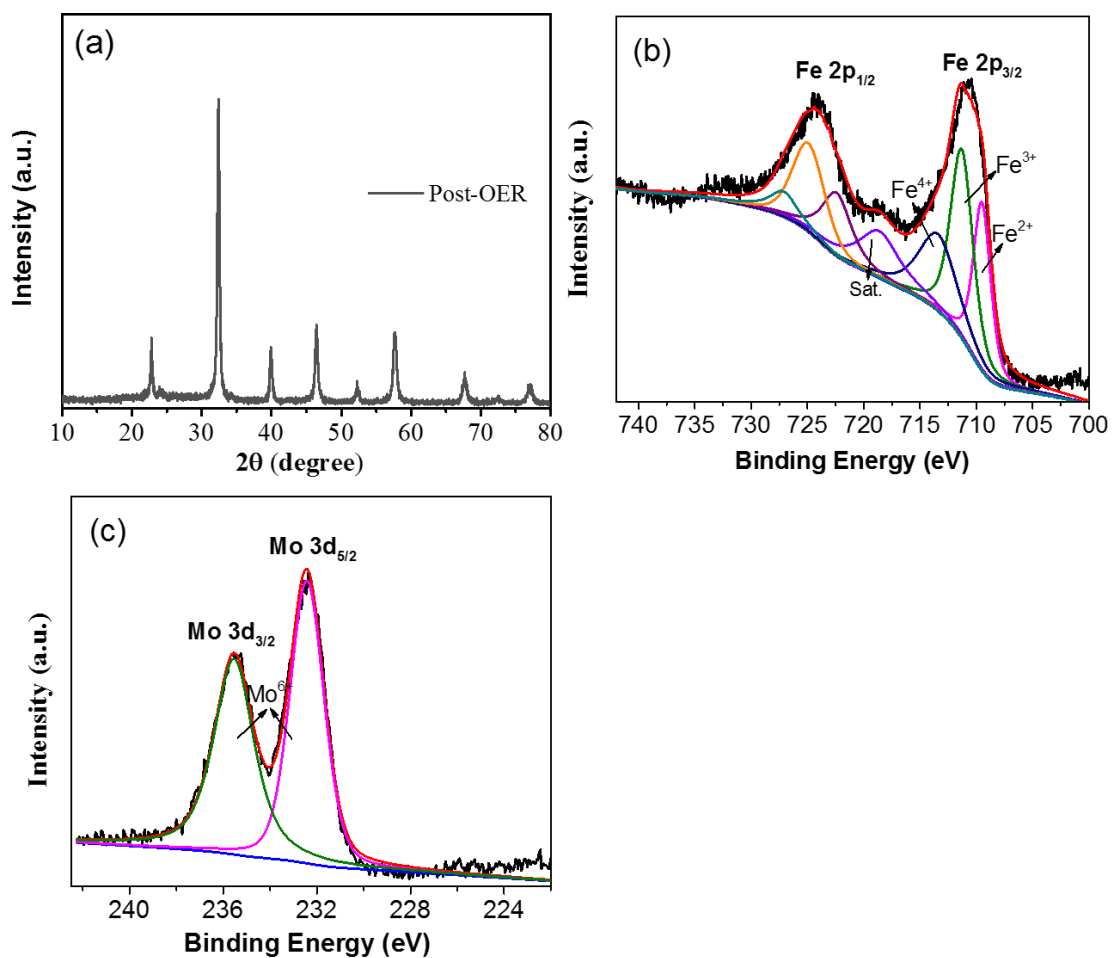
**Figure S4.** SEM images of (a) 5 nm Co(OH)<sub>2</sub>/SFM-NF and (b) 20 nm Co(OH)<sub>2</sub>/SFM-NF; LSV curves of as-prepared samples for (c) OER and (d) HER .



**Figure S5.** CV curves of (a) SFM, (b) SFM-NF, (c) Co(OH)<sub>2</sub>/SFM-NF, (d) Co(OH)<sub>2</sub> measured with different scan rates of 10, 20, 40, 60, 80, 100, 120, and 140 mV s<sup>-1</sup> with a potential of 1.01-1.11 V vs. RHE; CV curves of (e) SFM, (f) SFM-NF, (g) Co(OH)<sub>2</sub>/SFM-NF and (h) Co(OH)<sub>2</sub> measured with a potential of 0.07-0.17 V. vs. RHE.

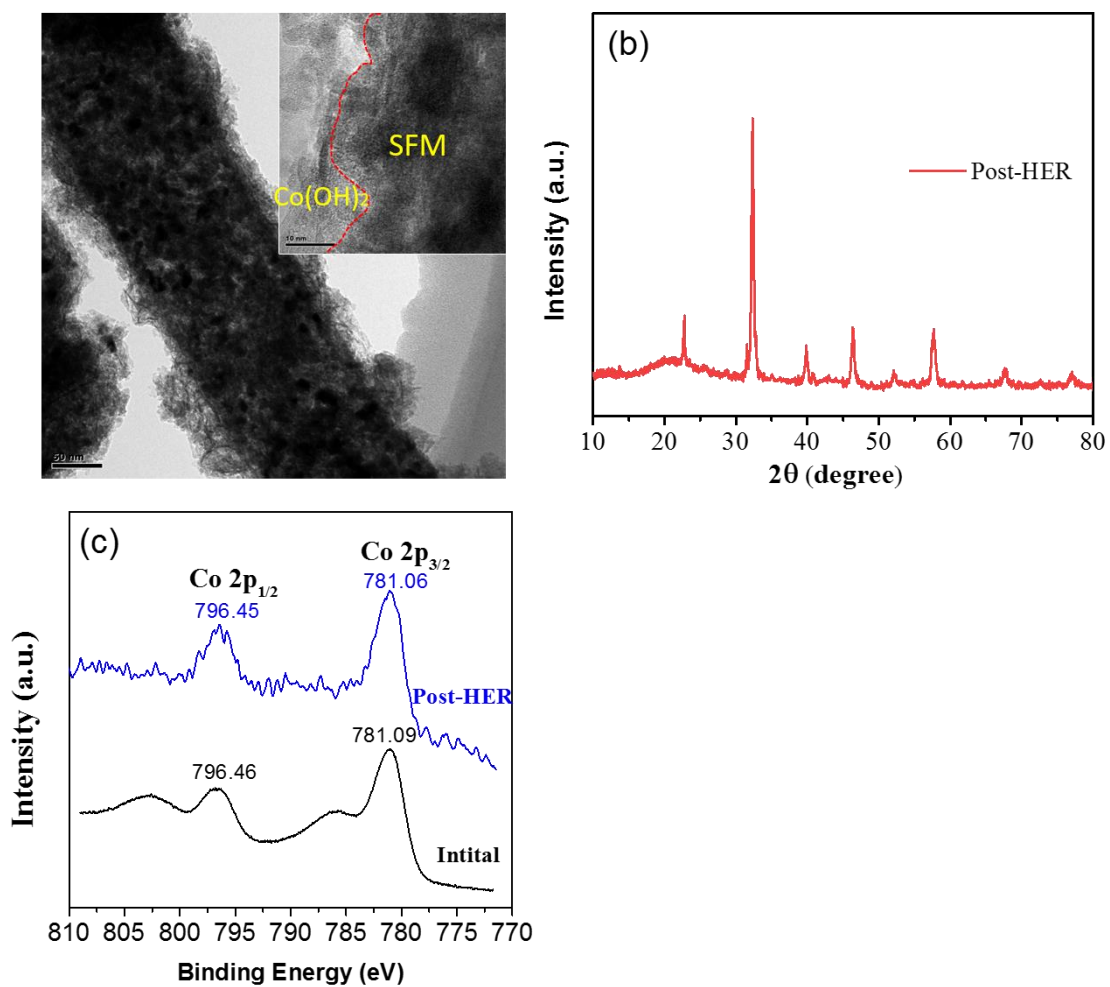


**Figure S6.** (a) Full XPS spectrum of SFM-NF and  $\text{Co(OH)}_2/\text{SFM-NF}$  samples, (b) High resolution XPS spectra of O 1s of  $\text{Co(OH)}_2$  and  $\text{Co(OH)}_2/\text{SFM-NF}$  samples.



**Figure S7.** (a) XRD pattern of  $\text{Co(OH)}_2/\text{SFM-NF}$  electrode after OER durability test; XPS spectra of (b) Fe 2p and (c) Mo 3d for  $\text{Co(OH)}_2/\text{SFM-NF}$  after OER test.





**Figure S8.** (a) TEM images of Co(OH)<sub>2</sub>/SFM-NF electrode after HER durability test, (b) XRD pattern of Co(OH)<sub>2</sub>/SFM-NF electrode after HER durability test, (c) XPS spectra of Co 2p for Co(OH)<sub>2</sub>/SFM-NF before and after HER test.

**Table S1.** Survey of overall water splitting stability with current density and cell voltage of representative bifunctional electrocatalysts in 1 M KOH electrolytes.

Catalyst	Current density (10 mA cm <sup>-2</sup> )	Cell voltage (V)	Stability (h)	Reference
Pt(-)/IrO <sub>2</sub> (+)	10	1.58	30	This work
Co(OH) <sub>2</sub> /SFM-NF	10	1.60	68	This work
A-PBSCF-H	10	1.62	12	1
SrNb <sub>0.1</sub> Co <sub>0.7</sub> Fe <sub>0.2</sub> O <sub>3-δ</sub> -NR	10	1.68	30	2
NiFe LDHs	10	1.70	10	3
NiCo <sub>2</sub> O <sub>4</sub>	10	1.65	20	4
NiFe/NiCo <sub>2</sub> O <sub>4</sub>	10	1.67	10	5
NiCo <sub>2</sub> S <sub>4</sub>	10	1.63	50	6
NdBaMnO <sub>5+δ</sub>	10	1.67	30	7
3DOM-LFC	10	1.75	12	8
La <sub>0.5</sub> Ba <sub>0.25</sub> Sr <sub>0.25</sub> CoO <sub>2.9-δ</sub> F <sub>0.1</sub>	10	1.66	13.3	9
CoNi(OH) <sub>x</sub>	10	1.67	10	10
CoMoV LDH	10	1.61	20	11
Co(OH) <sub>2</sub> -NA	10	1.64	20	12
Co <sub>0.75</sub> Ni <sub>0.25</sub> (OH) <sub>2</sub> nanosheet	10	1.57	15	13
Co <sub>0.9</sub> Fe <sub>0.1</sub> (OH) <sub>x</sub> -NF	10	1.62	30	14
Na <sub>0.08</sub> Ni <sub>0.9</sub> Fe <sub>0.1</sub> O <sub>2</sub>	16	1.60	12	15
Co(OH) <sub>2</sub> @Ni	10	1.64	20	16
Co(OH) <sub>2</sub> /Ag/FeP	10	1.56	50	17

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