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Supporting Information

Pd/C-catalyzed Transfer Hydrogenation of *N*-H Indoles with Trifluoroethanol and Tetrahydroxydiboron as Hydrogen Source

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1. General and Materials.

All reagents and solvents were of pure analytical grade. Pd/C (Type 394) is 10 wt% palladium on activated carbon paste with 50 wt% water that purchased from Johnson Matthey (product code: 113390). Thin layer chromatography (TLC) was performed on HSGF254 silica gel, pre-coated on glass-backed plates coated with 0.2 mm silica and revealed with either a UV lamp ($\lambda_{max} = 254$ nm). The products were purified by flash column chromatography on silica gel 200-300 mesh. ¹H and ¹³C NMR spectra were recorded on a 600 MHz spectrometer (¹H 600 MHz, ¹³C 151 MHz) using CDCl₃ or *d*⁶-DMSO as the solvent with tetramethylsilane (TMS) as the internal standard at room temperature. Chemical shifts are in δ (ppm) relative to TMS. The coupling constants (J) are in Hz.

2. The Typical Procedure for Pd/C-catalyzed Transfer Hydrogenation of Indoles



The Typical Procedure for Pd/C Catalyzed Transfer Hydrogenation of Indoles: A mixture of indole 1 (0.50 mmol), Pd/C (10.0 mg, 10 wt% palladium on activated carbon paste and 50% moisture, 0.9 mol% [Pd] based on starting material 1) in 2,2,2-trifluoroethanol (3 mL) was added into a Schlenk flask (25 mL) and stirred at room temperature. Then $B_2(OH)_4$ (90 mg, 1.0 mmol, 2.0 equiv) was added and the mixture was stirred at 40 °C. When the reaction was complete monitored by TLC, the solvent was evaporated under reduced pressure and the residue was purified by column chromatography (petroleum ether/ethyl acetate 50:1 to 20:1) to provide indolines 2.



Indoline (2a):^[1] Yield: 99%, 59.2 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.21 (d, *J* = 7.2 Hz, 1H), 7.11 (t, *J* = 7.6 Hz, 1H), 6.80 (t, *J* = 7.3 Hz, 1H), 6.72 (d, *J* = 7.8 Hz, 1H), 3.78 (s, 1H), 3.61 (t, *J* = 8.4 Hz, 2H), 3.11 (t, *J* = 8.4 Hz, 2H). ¹³C NMR (151 MHz, CDCl₃) δ 151.7, 129.4, 127.3, 124.7, 118.7, 109.5, 47.4, 29.9.



2-methylindoline (2b):^[1] Yield: >99%, 66.5 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.19 (d, J = 7.3 Hz, 1H), 7.13 (t, J = 7.6 Hz, 1H), 6.81 (t, J = 7.4 Hz, 1H), 6.70 (d, J = 7.8 Hz, 1H), 4.11-4.02 (m, 1H), 3.81 (s, 1H), 3.24 (dd, J = 15.4, 8.5 Hz, 1H), 2.74 (dd, J = 15.4, 7.8 Hz, 1H), 1.38 (d, J = 6.3 Hz, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 151.1, 129.0, 127.3, 124.8, 118.6, 109.3, 55.3, 37.9, 22.4.



2,3-dimethylindoline (2c):^[1] Yield: 99%, 73.0 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.13 (dd, *J* = 14.3, 7.4 Hz, 2H), 6.83 (t, *J* = 7.4 Hz, 1H), 6.69 (d, *J* = 7.7 Hz, 1H), 3.81 (s, 1H), 3.54-3.51 (m, 1H), 2.97-2.85 (m, 1H), 1.40 (dd, *J* = 6.6, 3.3 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 150.6, 134.4, 127.4, 123.3, 118.7, 109.2, 64.0, 44.4, 20.6, 17.3.



H₃C

4-methylindoline (2d):^[4] Yield: 96%, 64.1 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.05 (t, *J* = 7.6 Hz, 1H), 6.66 (d, *J* = 7.6 Hz, 1H), 6.59 (d, *J* = 7.7 Hz, 1H), 3.76 (s, 1H), 3.64 (t, *J* = 8.4 Hz, 2H), 3.06 (t, *J* = 8.4 Hz, 2H), 2.34 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 151.5, 134.3, 128.2, 127.4, 119.9, 107.0, 47.1, 28.7, 18.9.

5-methylindoline (2e):^[1] Yield: >99%, 66.4 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.02 (s, 1H), 6.90 (d, J = 7.8 Hz, 1H), 6.63 (d, J = 7.8 Hz, 1H), 3.63 (brs, 1H), 3.58 (t, J = 8.3 Hz, 2H), 3.06 (t, J = 8.3 Hz, 2H), 2.33 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 149.2, 129.8, 128.1, 127.6, 125.5, 109.5, 47.6, 30.0, 20.8.



2,5-dimethylindoline (2f):^[2] Yield: 98%, 72.0 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 6.96 (s, 1H), 6.87 (d, J = 7.8 Hz, 1H), 6.57 (d, J = 7.8 Hz, 1H), 4.03-3.98 (m, 1H), 3.55 (brs, 1H), 3.15 (dd, J = 15.3, 8.4 Hz, 1H), 2.65 (dd, J = 15.3, 7.8 Hz, 1H), 2.29 (s, 3H), 1.33 (d, J = 6.2 Hz, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 148.5, 129.3, 128.0, 127.5, 125.6, 109.3, 55.5, 37.9, 22.3, 20.8.



indolin-5-ol (2g):^[5] Yield: 90%, 60.6 mg, brown solid, m.p. 113-115 °C. ¹H NMR (600 MHz, DMSO) δ 6.60 (dd, J = 1.7, 1.0 Hz, 1H), 6.44 (d, J = 1.8 Hz, 2H), 3.43-3.39 (m, 2H), 2.88 (dd, J = 12.2, 4.4 Hz, 2H). ¹³C NMR (151 MHz, DMSO) δ 150.3, 143.6, 131.0, 113.3, 112.3, 110.4, 47.4, 30.2.



methyl indoline-5-carboxylate (2h):^[1] Yield: 92%, 81.2 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.77 (dd, *J* = 4.0, 2.7 Hz, 2H), 6.55 (d, *J* = 8.6 Hz, 1H), 4.30 (brs, 1H), 3.86 (s, 3H), 3.66 (t, *J* = 8.6 Hz, 2H), 3.07 (t, *J* = 8.5 Hz, 2H). ¹³C NMR (151 MHz, CDCl₃) δ 167.5, 156.0, 130.7, 128.7, 126.1, 119.6, 107.4, 51.5, 47.3, 28.9.



6-methylindoline (2i):^[3] Yield: 98%, 65.2 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.09 (d, J = 7.3 Hz, 1H), 6.62 (d, J = 7.3 Hz, 1H), 6.56 (s, 1H), 3.69 (s, 1H), 3.60 (t, J = 8.3 Hz, 2H), 3.06 (t, J = 8.3 Hz, 2H), 2.36 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 151.9, 137.1, 126.5, 124.3, 119.4, 110.5, 47.6, 29.6, 21.5.



indolin-6-ol (2j):^[5] Yield: 89%, 59.8 mg, brown solid, m.p. 102-104 °C. ¹H NMR (600 MHz, DMSO) δ 8.75 (s, 1H), 6.76 (d, *J* = 7.8 Hz, 1H), 5.9-5.934 (m, 2H), 5.33 (s, 1H), 3.36 (t, *J* = 8.3 Hz, 2H), 2.76 (t, *J* = 8.3 Hz, 2H). ¹³C NMR (151 MHz, DMSO) δ 157.4, 154.2, 124.7, 119.4, 104.0, 97.0, 47.4, 28.9.



7-methylindoline (2k):^[4] Yield: 99%, 65.9 mg, colorless oil. ¹H NMR (600 MHz, CDCl₃) δ 7.06 (d, J = 7.3 Hz, 1H), 6.93 (d, J = 7.5 Hz, 1H), 6.73 (t, J = 7.4 Hz, 1H), 3.62 (t, J = 8.4 Hz, 2H), 3.52 (brs, 1H), 3.12 (t, J = 8.4 Hz, 2H), 2.20 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 150.1, 128.7, 128.2, 122.2, 119.0, 118.9, 47.2, 30.2, 16.9.

3. Mechanism Study

1H NMR Study: 2-methylindole (66 mg, 0.50 mmol) was added to CF_3CH_2OD (1.5 mL) and stirred for 6 h at 40 °C. The solvent was evaporated under reduced pressure at 40 °C and the residue was obtained.



According to ¹H-NMR spectrum, it was indicated that hydrogen-deuterium exchange process existed between CF_3CH_2OD and 2-methylindole to form 2-methyl-1*H*-indole-1-*d* (Figure 1).



Figure 1. ¹H NMR of 2-methylindole in CF₃CH₂OD

Pd/C Catalyzed Transfer Hydrogenation of 2-methylindole in TFE: A mixture of 2-methylindole 1b (66 mg, 0.50 mmol), Pd/C (10.0 mg, 10 wt% palladium on activated carbon paste and 50% moisture, 0.9 mol% [Pd] based on 1b) in TFE (3 mL) was added into a Schlenk flask (25 mL) and stirred at room temperature. Then $B_2(OH)_4$ (90 mg, 1.0 mmol, 2.0 equiv) was added and the mixture was stirred at 40 °C. When the reaction was complete monitored by TLC (24 h), the reaction solution was filtered.

The analysis of ¹⁹F NMR of the filtrate showed that the peak at δ -76.06 was a new fluorine-containing compound. It was suggested that CF₃CH₂OB(OH)₂ might be formed in the reaction system (**Figure 2**).





Pd/C Catalyzed Transfer Hydrogenation of 2-methylindole in 2,2,2-trifluoroethan-1-ol-*d* (CF₃CH₂OD): A mixture of 2-methylindole 1b (66 mg, 0.50 mmol), Pd/C (10.0 mg, 10 wt% palladium on activated carbon paste and 50% moisture, 0.9 mol% [Pd] based on 1b) in CF₃CH₂OD (3 mL) was added into a Schlenk flask (25 mL) and stirred at room temperature. Then $B_2(OH)_4$ (90 mg, 1.0 mmol, 2.0 equiv) was added and the mixture was stirred at 40 °C. When the reaction was complete monitored by TLC (48 h), the solvent was evaporated under reduced pressure and the residue was purified by column chromatography (petroleum ether/ethyl acetate 50:1 to 30:1) to provide 2-methylindoline (colorless oil, 45.2 mg).



The analysis of ¹H NMR of the product showed that three deuterium atoms were imported to the 2- (\sim 86% and \sim 92%) and 3-position (\sim 90%). A hydrogen-deuterium exchange process existed between CF₃CH₂OD and B₂(OH)₄ to form B₂(OD)₄. The two deuterium atoms of 3-position suggested that a reversible process of protonation and deprotonation existed (**Figure 3**).



Figure 3. ¹H NMR of Product 1b in CF₃CH₂OD

4. References

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5. Copy of NMR for the Indolines





¹H-NMR of 2b









¹³C-NMR of 2c

		—123.32 —118.67		63.97		
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6.0

5.5

8.0











¹³C-NMR of 2g

50.34	43.60	31.02	13.25 10.42 10.42	7.37 9.61 9.53 0.23
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¹H-NMR of 2h





























¹³C-NMR of 2k





