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Electronic Supplementary Information (ESI)

Metal-free polycycloaddition of aldehyde-activated internal diynes and diazides toward post-functionalizable poly(formyl-1,2,3-triazole)s

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Monomer synthesis. The aldehyde-activated internal diynes, named 3,3'-(9,9-dibutyl-9H-fluorene-2,7-diyl)dipropiolaldehyde (**1a**) and 3,3'-(9,9-dimethyl-9H-fluorene- 2,7-diyl)dipropiol-aldehyde (**1b**) were prepared according to the published procedures.^{1,2} The azides monomers, 1,4-bis((6-azidohexyl)oxy)benzene (**2a**), 1,2-bis(4-(azidomethyl)phenyl)-1,2-diphenylethene (**2b**), bis(4-azidophenyl)- methane (**2c**) and 1,2-bis(4-((6-azidohexyl) oxy)phenyl)-1,2-diphenylethene (**2d**) were prepared according to the previous reports.³⁻⁶ The detailed synthetic routes are shown in Schemes S1 and S2.

3,3'-(9,9-Dibutyl-9H-fluorene-2,7-diyl)dipropiolaldehyde (1a): ¹H NMR (500 MHz, CDCl₃), δ (TMS, ppm): 9.46, 7.77, 7.75, 7.64, 7.61, 1.99, 1.10, 0.68, 0.55. ¹³C NMR (125 MHz, CDCl₃), δ (TMS, ppm): 176.77, 151.89, 142.97, 132.92, 128.08, 121.03, 118.98, 96.01, 89.39, 55.62, 39.98, 26.04, 23.05, 13.90.

3,3'-(9,9-Dimethyl-9H-fluorene-2,7-diyl)dipropiolaldehyde (1b): ¹H NMR (500 MHz, CDCl₃), *δ* (TMS, ppm): 9.45, 7.78, 7.76, 7.70, 7.65, 7.63, 1.51. ¹³C NMR (125 MHz, CDCl₃), *δ* (TMS, ppm): 176.74, 154.58, 141.11, 133.06, 128.05, 121.28, 119.16, 95.75, 89.33, 47.34, 26.84.

1,4-Bis((6-azidohexyl)oxy)benzene (2a): ¹H NMR (500 MHz, CDCl₃), *δ* (TMS, ppm): 6.81, 3.90, 3.27, 1.77, 1.62, 1.46. ¹³C NMR (125 MHz, CDCl₃), *δ* (TMS, ppm): 153.23, 115.30, 68.36, 51.37, 29.32, 28.77, 26.63, 25.77.

1,2-Bis(4-(azidomethyl)phenyl)-1,2-diphenylethene (2b): ¹H NMR (500 MHz, CDCl₃), *δ* (TMS, ppm): 7.11, 7.03, 4.26. ¹³C NMR (125 MHz, CDCl₃), *δ* (TMS, ppm): 143.77, 143.26, 140.81, 133.53, 131.83, 131.24, 128.02, 126.83, 54.62.

Bis(4-azidophenyl)methane (2c): ¹H NMR (500 MHz, CDCl₃), *δ* (TMS, ppm): 7.14, 6.96, 3.92. ¹³C NMR (125 MHz, CDCl₃), *δ* (TMS, ppm): 138.26, 137.76, 130.22, 119.47, 40.82.

1,2-Bis(4-((6-azidohexyl)oxy)phenyl)-1,2-diphenylethene (2d): ¹H NMR (500 MHz, CDCl₃), δ (TMS, ppm): 7.12-7.00, 6.94-6.89, 6.65-6.60, 3.90-3.86, 3.29-3.26, 1.78-1.73, 1.66-1.60, 1.50-1.43. ¹³C NMR (125 MHz, CDCl₃), δ (TMS, ppm): 157.58, 144.41, 139.77, 136.43, 132.64, 131.54, 127.68, 126.28, 113.66, 67.63, 51.54, 31.08, 29.31, 28.94, 26.68, 25.87.

Post-modification model reactions. Compounds **6**-A/**6**-B and cis-7/trans-7 were prepared according to the published procedures.^{7,8} The detailed synthetic routes are shown in Scheme S4 and S5.

6-A: ¹H NMR (500 MHz, CDCl₃), δ (TMS, ppm): 8.43, 7.49-6.61, 5.51, 2.57, 2.49, 1.56, 1.29, 1.25, 0.87.

6-B: ¹H NMR (500 MHz, CDCl₃), δ (TMS, ppm): 8.51, 7.69-6.64, 6.17, 2.62, 2.49, 1.62, 1.26, 0.88.

cis-**7**: ¹H NMR (500 MHz, CDCl₃), δ (TMS, ppm): 7.52-7.45, 7.27, 7.15-7.05, 6.76 (d, *J*=12 Hz, 1H), 5.41 (d, *J*=12 Hz, 1H).

trans-7: ¹H NMR (500 MHz, CDCl₃), δ (TMS, ppm): 7.54-7.49, 7.28, 7.15-7.03, 6.28 (d, *J*=16 Hz, 1H).



Scheme S1 Synthesis of 1a and 1b.



Scheme S2 Synthesis of 2a-2d.



Scheme S3 Synthesis of 3-A and 3-B.



Scheme S4 Synthesis of 6-A and 6-B.



Scheme S5 Synthesis of cis-7 and trans-7.

entry	solvent	yield (%)	$M_{ m w}{}^b$	PDI^b	$F_{\mathrm{A}^{c}}$ (%)
1	DMSO	85	16 360	1.55	71
2	DMF	80	14 580	1.48	68
3	1,4-dioxane	79	12 950	1.41	64
4	toluene	71	15 300	1.47	61

Table S1 Effect of solvent on the polymerization of 1a and $2a^a$

^{*a*} Carried out at 120 °C for 3 h under nitrogen at a monomer concentration of 0.5 M, [1a] = [2a]. ^{*b*} Estimated by advanced polymer chromatography (APC) using THF as an eluent on the basis of a polystyrene (PS) calibration; M_w = weight-average molecular weight; polydispersity index (PDI) = M_w/M_n ; M_n = number-average molecular weight. ^{*c*} Fraction of structure A in the polymers determined by ¹H NMR.

entry	<i>T</i> (°C)	yield (%)	$M_{ m w}{}^b$	PDI^b
1	80	52	4410	1.22
2	100	63	11 090	1.37
3	120	80	15 190	1.48
4	140	87	43 350	2.09
5	150	83	53 110	2.55

Table S2 Effect of reaction temperature on the polymerization of 1a and $2a^a$

^{*a*} Carried out in DMSO for 3 h under nitrogen at a monomer concentration of 0.5 M, [1a] = [2a]. ^{*b*} Estimated by APC using THF as an eluent on the basis of a PS calibration; $M_{\rm w}$ = weight-average molecular weight; PDI = $M_{\rm w}/M_{\rm n}$; $M_{\rm n}$ = number-average molecular weight.

Table S3 Time course of the polymerization of 1a and $2a^a$

entry	<i>t</i> (h)	yield (%)	$M_{ m w}{}^b$	PDI ^b
1	1.0	76	23 440	1.71
2	1.5	88	48 920	2.21
3	2.0	89	54 550	2.29
4	3.0	90	58 690	2.39

^{*a*} Carried out in DMSO at 150 °C under nitrogen at a monomer concentration of 0.5 M, [1a] = [2a]. ^{*b*} Estimated by APC using THF as an eluent on the basis of a PS calibration; M_w = weight-average molecular weight; PDI = M_w/M_n ; M_n = number-average molecular weight.

entry	<i>C</i> (M)	yield (%)	$M_{ m w}{}^b$	PDI^b
1	0.10	75	9730	1.33
2	0.20	80	18 470	1.59
3	0.33	85	37 420	1.94
4^c	0.50	89	54 550	2.29
5	1.00	gelled	-	-

Table S4 Effect of monomer concentration on the polymerization of 1a and $2a^a$

^{*a*} Carried out in DMSO at 150 °C under nitrogen for 2 h, [1a] = [2a]. ^{*b*} Estimated by APC using THF as an eluent on the basis of a PS calibration; M_w = weight-average molecular weight; PDI = M_w/M_n ; M_n = number-average molecular weight. ^{*c*} Data taken from Table S3, entry 3.

 Table S5 Crystal data and structure refinement of model compound 3-B

Number	CCDC 1958959
Empirical formula	C ₁₆ H ₁₃ N ₃ O
Formula weight	263.29
Temperature	120(10) K
Wavelength	1.54184 Å
Crystal system	monoclinic
Space group	P 1 21/c 1
Unit cell dimensions	$a = 13.1254(7) \text{ Å} \alpha = 90.00 \text{ °}$
	$b = 8.1029(4) \text{ Å}$ $\beta = 109.249 \text{ °}$
	$c = 12.6860(7) \text{ Å} \gamma = 90.00 \text{ °}$
Volume	1273.78(12) Å ³
Ζ	4
Density (calculated)	1.373 mg/m ³
Absorption coefficient	0.713 mm ⁻¹
F(000)	552
Crystal size	$0.07\times0.05\times0.05\ mm^3$
Theta range for data collection	3.567 to 67.097 °
Index ranges	-15<=h<=14, -6<=k<=9, -14<=l<=15
Reflections collections	5898
Independent reflections	2269 [R(int) = 0.0240]
Completeness to theta = 66.97°	99.9%
Absorption correction	multi-scan
Max. and min. transmission	1.00000 and 0.71569
Data / restraints / parameters	2269 / 0 / 181
Goodness-of-fit on F ²	1.077
Final R indices [I>2sigma(I)]	R1 = 0.0437, wR2 = 0.1074
R indices (all data)	R1 = 0.0450, wR2 = 0.1083
Largest diff. peak and hole	0.256 and -0.397 e.Å ⁻³

 Table S6 Crystal data and structure refinement of model compound cis-7

Number	CCDC 1958962
Empirical formula	$C_{18}H_{14}N_4$
Formula weight	286.33
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	triclinic
Space group	P-1
Unit cell dimensions	$a = 9.1045(15) \text{ Å}$ $\alpha = 110.846 \text{ °}$
	$b = 9.4078(14) \text{ Å} \beta = 92.668 \text{ °}$
	$c = 9.9658(17) \text{ Å} \gamma = 107.055 \text{ °}$
Volume	751.7(2) Å ³
Ζ	2
Density (calculated)	1.265 mg/m ³
Absorption coefficient	0.078 mm ⁻¹
F(000)	300
Crystal size	$0.22\times0.19\times0.16\ mm^3$
Theta range for data collection	2.79 to 25.10 °
Index ranges	-10<=h<=10, -11<=k<=11, -11<=l<=11
Reflections collections	10178
Independent reflections	2573 [R(int) = 0.0586]
Completeness to theta = 25.10°	95.9%
Absorption correction	multi-scan
Max. and min. transmission	0.9876 and 0.9830
Data / restraints / parameters	2573 / 0 / 199
Goodness-of-fit on F ²	1.001
Final R indices [I>2sigma(I)]	R1 = 0.0679, WR2 = 0.1817
R indices (all data)	R1 = 0.0944, wR2 = 0.1939

 Table S7 Crystal data and structure refinement of model compound trans-7

Number	CCDC 1958961
Empirical formula	$C_{18}H_{14}N_4$
Formula weight	286.33
Temperature	300(2) K
Wavelength	0.71073 Å
Crystal system	monoclinic
Space group	P2(1)/c
Unit cell dimensions	$a = 11.0650(8) \text{ Å}$ $\alpha = 90.00 \text{ °}$
	b = 15.5926(12) Å β = 108.433 °
	$c = 9.3078(6) \text{ Å} \gamma = 90.00 \text{ °}$
Volume	1523.50(19) Å ³
Ζ	4
Density (calculated)	1.248 mg/m ³
Absorption coefficient	0.077 mm ⁻¹
F(000)	600
Crystal size	$0.24\times0.18\times0.15\ mm^3$
Theta range for data collection	2.61 to 26.00 °
Index ranges	-13<=h<=13, -19<=k<=19, -11<=l<=11
Reflections collections	18156
Independent reflections	2989 [R(int) = 0.0540]
Completeness to theta = 26.00°	99.5%
Absorption correction	multi-scan
Max. and min. transmission	0.9885 and 0.9817
Data / restraints / parameters	2989 / 0 / 199
Goodness-of-fit on F ²	1.005
Final R indices [I>2sigma(I)]	R1 = 0.0537, WR2 = 0.1403
R indices (all data)	R1 = 0.0821, WR2 = 0.1721



Fig. S1 TGA (A) and DSC (B) curves of P**1a2a**-P**1b2a** at a heating rate of 20 °C min⁻¹ under nitrogen. T_d presents temperature of 5% weight loss.



Fig. S2 FT-IR spectra of 2b (A), 1a (B) and P1a2b (C).



Fig. S3 FT-IR spectra of 2c (A), 1a (B) and P1a2c (C).



Fig. S4 FT-IR spectra of 2d (A), 1a (B) and P1a2d (C).



Fig. S5 FT-IR spectra of 2a (A), 1b (B) and P1b2a (C).



Fig. S6 ¹³C NMR spectra of **1a** (A), **2a** (B), **3**-A (C), **3**-B (D) and **P1a2a** (E) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S7 ¹H NMR spectra of **1a** (A), **2b** (B) and P**1a2b** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S8 ¹H NMR spectra of **1a** (A), **2c** (B) and P**1a2c** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S9 ¹H NMR spectra of **1a** (A), **2d** (B) and P**1a2d** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S10 ¹H NMR spectra of **1b** (A), **2a** (B) and P**1b2a** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S11 ¹³C NMR spectra of **1a** (A), **2b** (B) and P**1a2b** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S12 ¹³C NMR spectra of **1a** (A), **2c** (B) and P**1a2c** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S13 ¹³C NMR spectra of **1a** (A), **2d** (B) and P**1a2d** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S14 ¹³C NMR spectra of **1b** (A), **2a** (B) and P**1b2a** (C) in CDCl₃. The solvent peaks are marked with asterisks.



Fig. S15 The in situ IR spectra of the polymerization solution of **1a** and **2a** at 150 °C within 3 h.



Fig. S16 PL spectra of P**1a2d** in THF and THF/water mixtures with different water fractions. Concentrations: 10^{-5} M, $\lambda_{ex} = 323$ nm. Inset: photograph taken under illumination of hand-held UV lamp.



Fig. S17 Changes of PL spectrum (A) and the plot of normalized changes in PL intensity $(1-I/I_0)$ of P**1a2a** treated with hydrazine in THF solutions.



Fig. S18 ¹H NMR spectra of P1a2a (A), 6-B (B), 6-A (C) and PPM1 (D) in CDCl₃. The solvent peaks are marked with asterisks.

The grafting degree of **PPM1** is arranged as *x*, then un-reacted part accounts for (1-*x*), $\frac{2x}{1.81} = \frac{1.81}{1.81}$

$$\frac{1}{18x + 10(1 - x)} = \frac{1}{20.89}$$
$$x = 0.657$$



Fig. S19 ¹H NMR spectra of P1a2a (A), 6-B (B), 6-A (C) and PPM2 (D) in $CDCl_3$. The solvent peaks are marked with asterisks.

The grafting degree of **PPM2** is arranged as *x*, then un-reacted part accounts for (1-*x*), $\frac{2x}{1.96} = \frac{1.96}{1.96}$

$$\frac{1}{18x + 10(1 - x)} = \frac{1}{30.94}$$
$$x = 0.424$$



Fig. S20 ¹H NMR spectra of P**1a2a** (A), *trans*-7 (B), *cis*-7 (C) and **PPM3** (D) in CDCl₃. The solvent peaks are marked with asterisks.

The grafting degree of **PPM3** is arranged as *x*, then un-reacted part accounts for (1-*x*), 4x = 3.99

 $\overline{4} = \overline{4.05}$

x = 0.985



Fig. S21 TGA (A) and DSC (B) curves of P**1a2a** and its derivatives at a heating rate of 20 °C min⁻¹ under nitrogen. T_d presents temperature of 5% weight loss.



Fig. S22 UV-vis absorption spectra (A) and PL spectra (B) of P1a2a and PPM3 in THF solutions. Concentrations: 10^{-5} M, $\lambda_{ex} = 323$ nm. Inset: photographs of P1a2a and PPM3 powders (A) and their THF solutions (B) taken under daylight and UV illumination.

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