

Electronic Supplementary Information

ZIF-8-Based vs. ZIF-8-Derived Au and Pd Nanoparticles as Efficient Nanocatalysts for the Ullmann Homocoupling Reaction

Wenjuan Wang,^{a,b} Shuang Chen,^c Eduardo Guisasola Cal,^d Marta Martínez Moro,^d Sergio Moya,^d Emerson Coy,^e Changlong Wang,^a Jean-René Hamon,^b and Didier Astruc^{a,*}

^a ISM, UMR CNRS 5255, Université de Bordeaux, Talence 33405 Cedex, France. E-mail: didier.astruc@u-bordeaux.fr

^b Institut des Sciences Chimiques, UMR CNRS 6226, Université de Rennes 1, 35042 Rennes Cedex, France.

^c Department of Chemistry and Center for Atomic Engineering of Advanced Materials, Anhui University, Hefei, Anhui 230601, China.

^d Soft Matter Nanotechnology Lab, CIC biomaGUNE, Paseo Miramón 182, 20014 Donostia-San Sebastián, Gipuzkoa, Spain.

NanoBioMedical Centre, Adam Mickiewicz University in Poznań, Wszechnicy Piastowskiej 3, 61-614 Poznań, Poland

Table of contents

	page
1. General data	S1
2. Procedure for digestion of nanocatalysts with EDTA, and nanoparticles capping with PVP	S1
3. TEM and size distribution of ZIF-8, Au@CN and Pd@CN	S2
4. TEM and size distribution of Pd@ZIF-8 and Au@ZIF-8 after reaction	S3
5. XRD patterns of ZIF-8, Pd@ZIF-8 and Au@ZIF-8	S4
6. XPS spectrum of the Pd@ZIF-8 and Au@ZIF-8	S5
7. XPS spectrum of the Pd@ZIF-8 and Au@ZIF-8 after reaction	S5
8. Physical properties of the nanocatalysts on ZIF-8, Au@ZIF-8 and Pd@ZIF-8	S6
9. Nanocatalysts for recycling experiments	S6
10. ¹ H NMR spectra for various substrates with Pd@ZIF-8 as catalyst	S7
11. ¹ H NMR spectra for various substrates with Au@ZIF-8 as catalyst	S9

1. General data

All the solvents and chemicals were used as received. ^1H NMR spectra were recorded at 25 °C with a Bruker AC 300 MHz. All the chemical shifts are reported in parts per million (δ , ppm) with reference to Me_4Si in CDCl_3 . X-ray diffraction (XRD) patterns were obtained on SmartLab 9KW with $\text{Cu K}\alpha$ radiation. Nanoparticles loaded into the ZIF-8 were determined by Inductively Coupled Plasma Mass Spectrometry (ICP-MS). Transmission Electron Microscopy (TEM): The sizes of the MNPs were determined by TEM using a JEOL JEM 1400 (120 kV) microscope. The TEM samples were prepared by deposition of the nanoparticle suspension (10 μL) onto a carbon-coated microscopy copper grid. X-ray Photoelectron Spectroscopy (XPS): System: SPECS SAGE HR; X-Ray source: $\text{Al K}\alpha$ non-monochromatic; operated at 12.5 kV and 300 W. Take off angle 90°, at $\sim 10^{-8}$ Torr. Pass energy for survey spectra 30 eV, 10 eV for narrow scans analysis; spectra are calibrated to CC carbon 285 eV. Analysis is consisted of Shirley background subtraction. Peaks are fitted with symmetrical Gaussian-Lorentizan (GL) line shapes. Samples are prepared by dehydration on the titania coated glass or silica substrates. Titania is selected as a substrate to avoid the overlap of Si and Au. Flash column chromatography was performed using silica gel (300-400 mesh).

2. Procedure for digestion of nanocatalysts with EDTA, and nanoparticles capping with PVP for TEM characterization.

A schlenk flask is charged with 20 mg of as-prepared Pd@ZIF-8 in 6 mL water, and this mixture is stirred continuously at room temperature. PVP ($M_w=10000$, 10 equiv per NPs) solubilized in 2 mL water is added and allowed to stir for 30 min. $\text{EDTA-2Na}\cdot 2\text{H}_2\text{O}$ (10 equiv) in 2 mL water is then added and the mixture is stirring for 1 h. Several drops of this solution are used for TEM characterization.

3. TEM and size distribution of ZIF-8, Au@CN and Pd@CN.

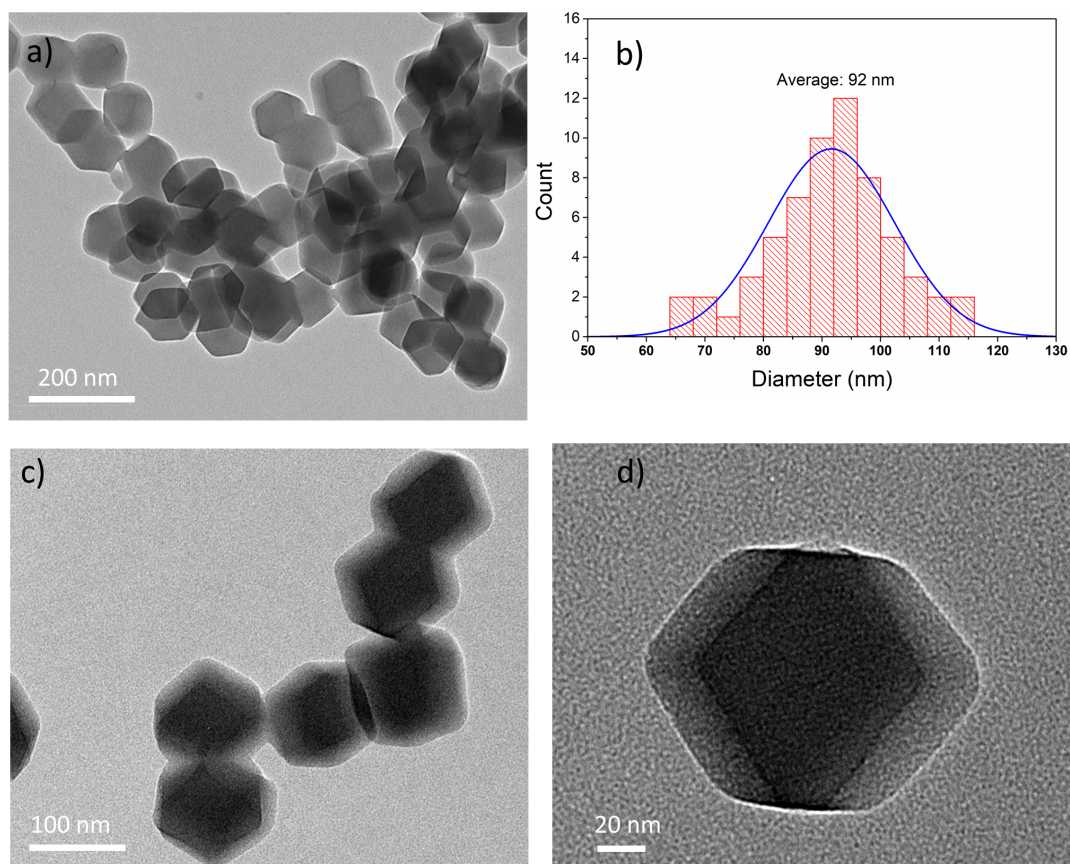


Fig. S1. TEM images and size distribution histogram of the ZIF-8.

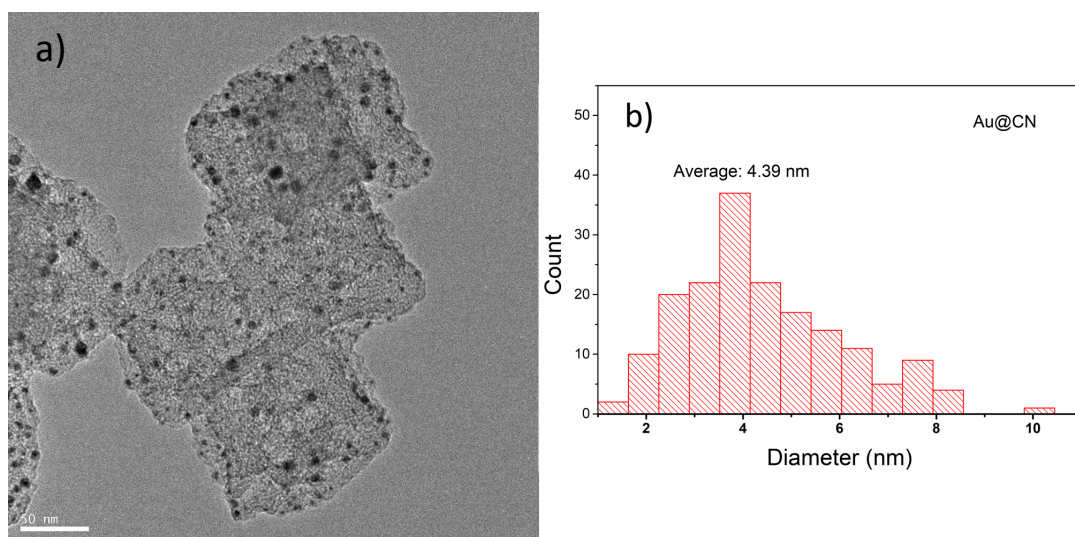


Fig. S2 TEM images and size distribution histogram of as-synthesized Au@CN.

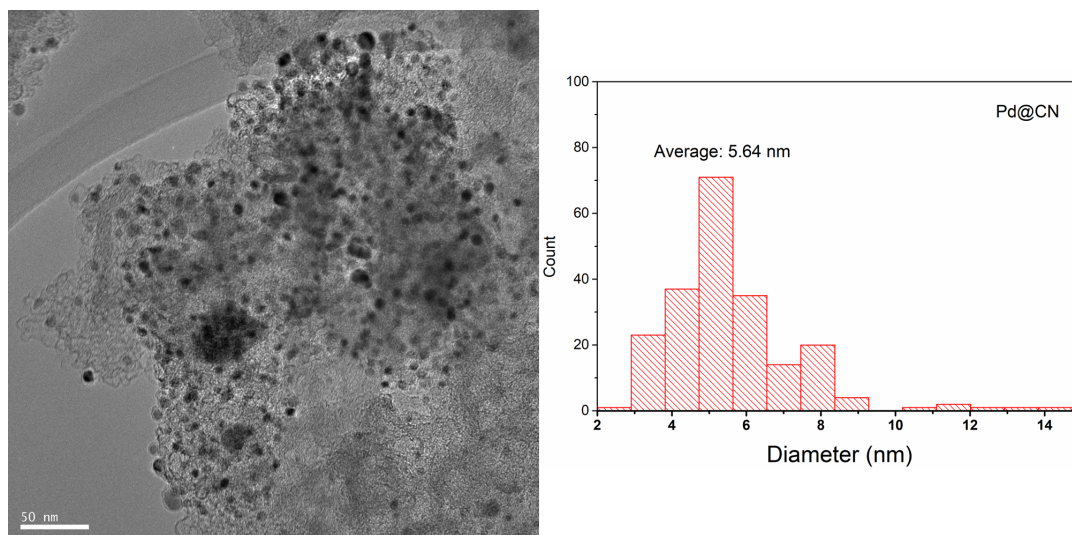


Fig. S3 TEM images and size distribution histogram of as-synthesized Pd@CN.

4. TEM and size distribution of Pd@ ZIF-8 and Au@ZIF-8 after reaction.

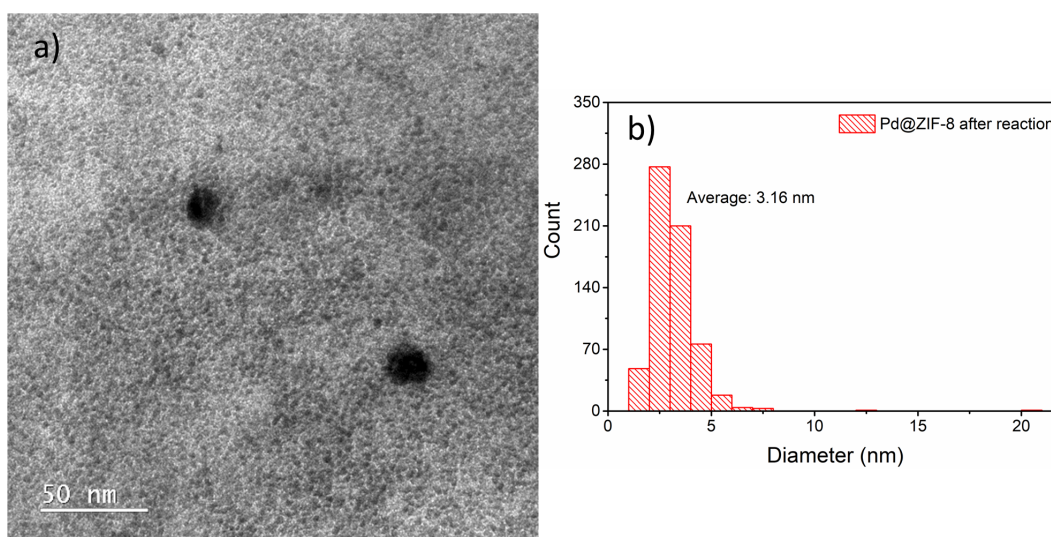


Fig. S4 TEM image (a) and size distribution (b) of Pd@ZIF-8 after reaction. Pd@ZIF-8 after digestion of the nanocatalyst with ethylenediamine tetracetic acid (EDTA) to release nanoparticles, and nanoparticles stabilized by PVP.

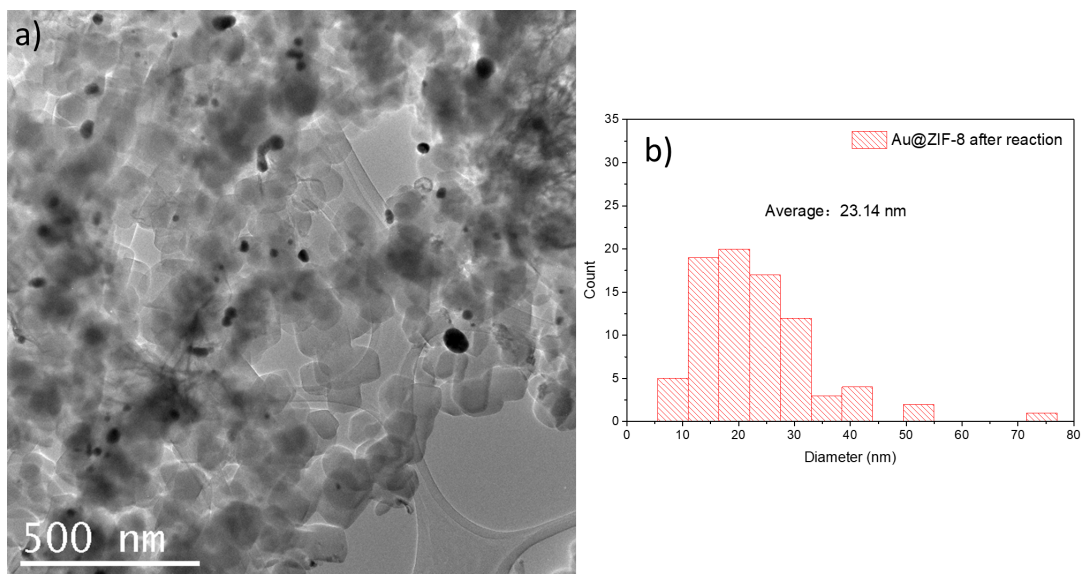


Fig. S5 TEM images and size distribution histogram of as-synthesized Au@ZIF-8 after reaction.

5. XRD patterns of ZIF-8, Pd@ZIF-8 and Au@ZIF-8

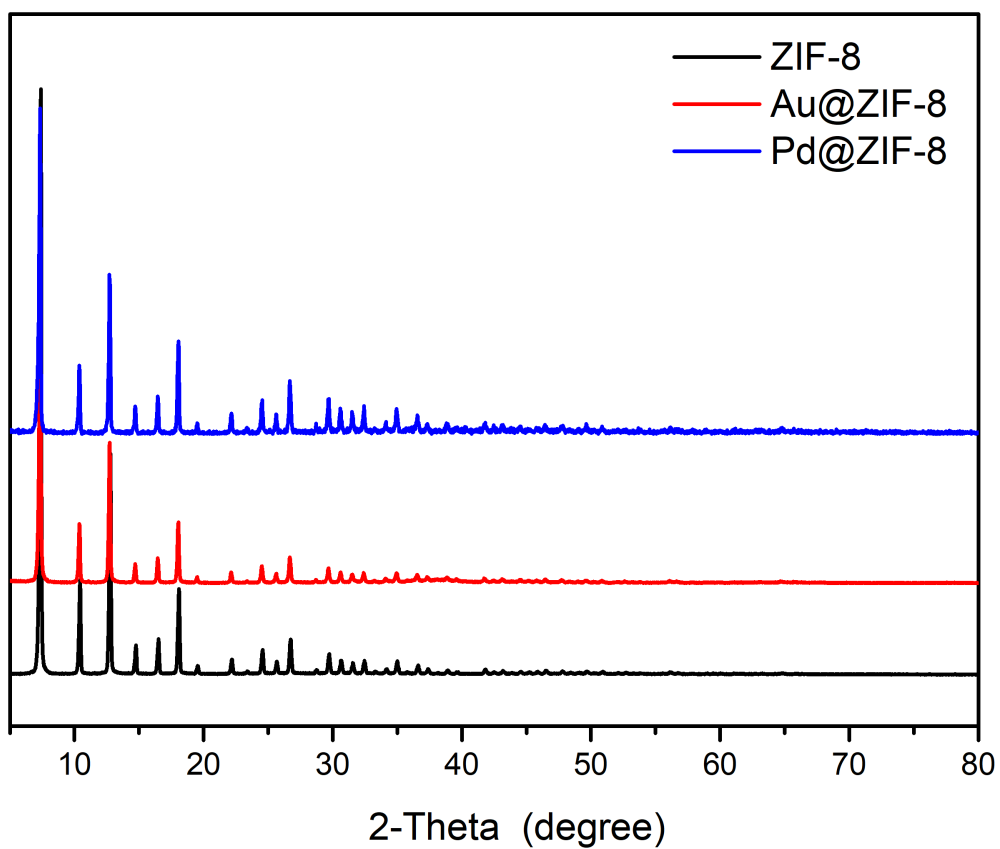


Fig. S6. XRD patterns of ZIF-8, Au@ZIF-8 and Pd@ZIF-8.

6. XPS spectrum of the Pd@ZIF-8 and Au@ZIF-8

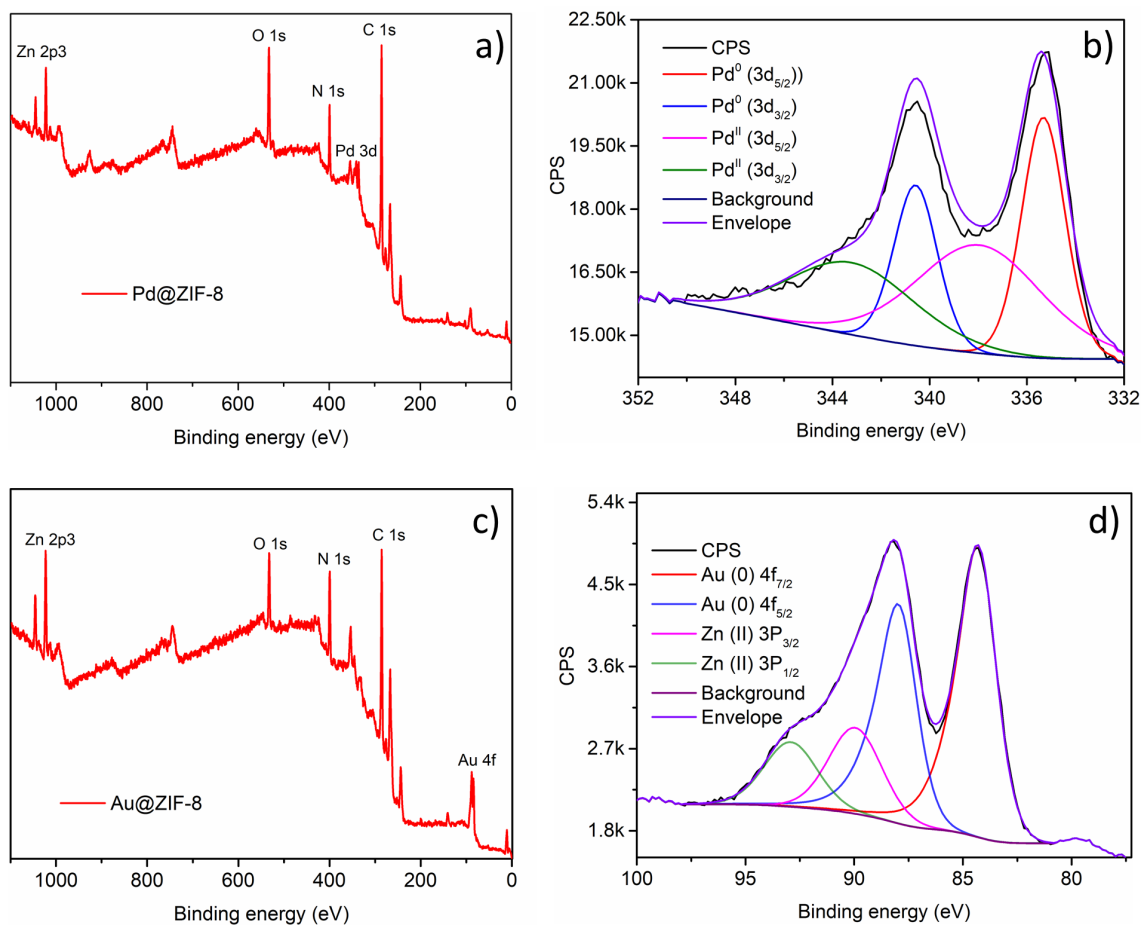


Fig. S7. XPS spectrum of the Pd@ZIF-8(a, b) and Au@ZIF-8 (c, d).

7. XPS spectrum of the Pd@ZIF-8 and Au@ZIF-8 after reaction

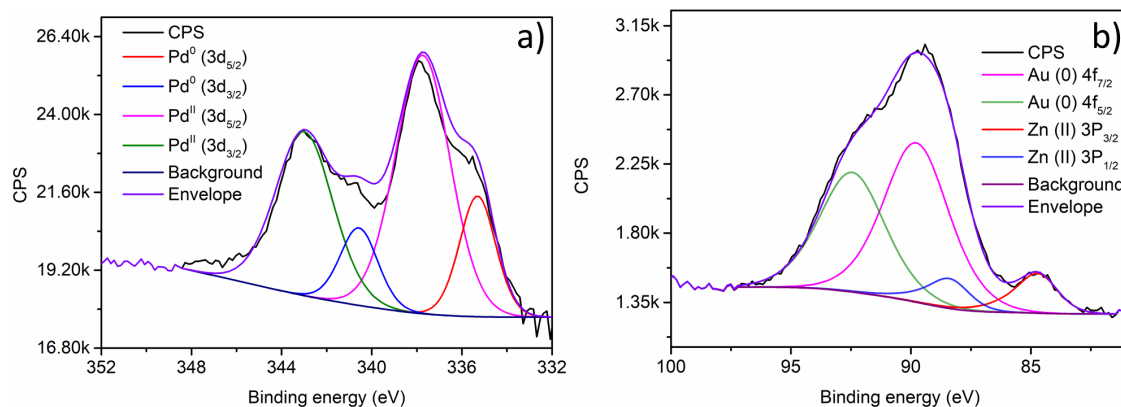


Fig. S8. XPS spectrum of the Pd@ZIF-8(a) and Au@ZIF-8 (b) after the catalytic reaction.

8. Physical properties of the nanocatalysts on ZIF-8, Au@ZIF-8 and Pd@ZIF-8

Table S1. Physical Properties of the Nanocatalysts on ZIF-8, Pd@ZIF-8 and Au@ZIF-8.

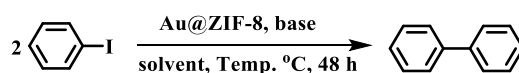
Sample	Surface area [m ² .g ⁻¹]	Total Pore volume [cm ³ .g ⁻¹]	micropore volume [cm ³ .g ⁻¹]	Pore diameter [nm]
ZIF-8	1365.1	0.81	0.651	2.38
Au@ZIF-8	1283.1	0.85	0.595	2.64
Pd@ZIF-8	1247.7	0.66	0.581	2.07

9. Nanocatalysts for recycling experiments

Procedure for recycling the nanoparticle catalyts:

After the reaction, the system was allowed to cool to r.t.. The catalyst was washed with CH₂Cl₂, and with water by centrifugation respectively to remove iodobenzene and K₃PO₄, then dried at 60 °C under vacuum. After charging another run of substrates, DMF and K₃PO₄, the system was purged with N₂ for 10 minutes. The reaction was then allowed to be proceeding. After 48 h, the reaction mixture was filtered. The liquid was extracted several times with CH₂Cl₂ and H₂O. The organic phase was dried over Na₂SO₄. Then CH₂Cl₂ was removed under reduced pressure at the rotary evaporator providing the crude product. ¹H NMR (300 MHz, CDCl₃) was used to calculate the conversion.

Table S2. Effects of solvent, base and temperature on the homocoupling of iodobenzene catalyzed by Au@ZIF-8.



Entry	Solvent	Base	Temp. (°C)	Conversion (%) ^a
1	DMF	K ₂ CO ₃	130	62
2	DMF	K ₂ CO ₃	110	28
3	DMF	K ₂ CO ₃	140	75
4	Xylene	K ₂ CO ₃	140	11
5	PEG-400	K ₂ CO ₃	140	74
6	DMF	K ₃ PO ₄	140	83

Reaction conditions: 0.2 mmol iodobenzene, 0.6 mmol base, 1 mL solvent, 50 mg Au@ZIF-8 (6 mmol%), 140 °C, N₂, 48 h. ^a ¹H NMR conversion.

10. ^1H NMR spectra of various substrates with Pd@ZIF-8 as catalyst

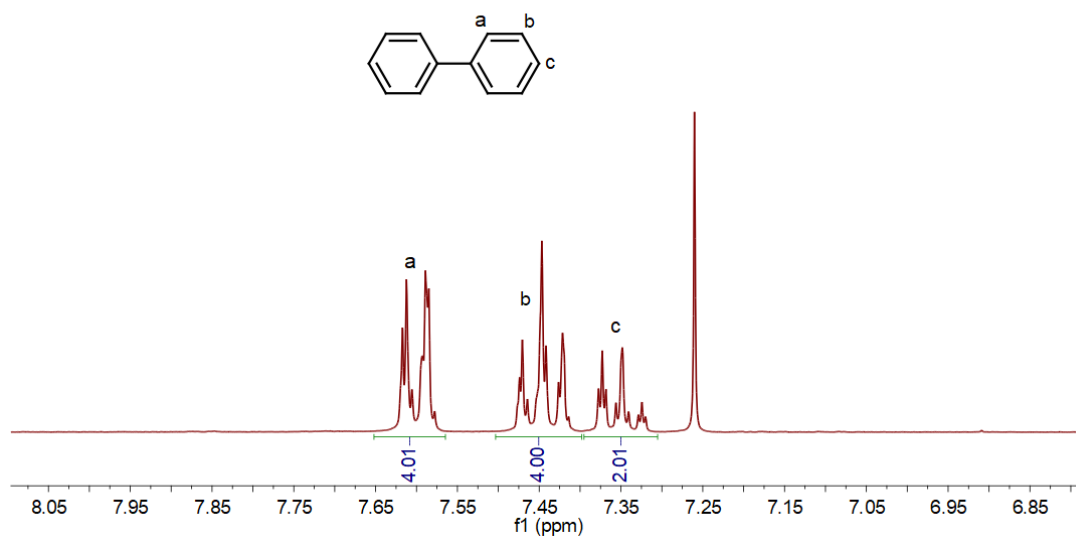


Fig. S9. ^1H NMR (300 MHz, CDCl_3) spectrum of the product of 1,1'-biphenyl (homocoupling of iodobenzene with Pd@ZIF-8).

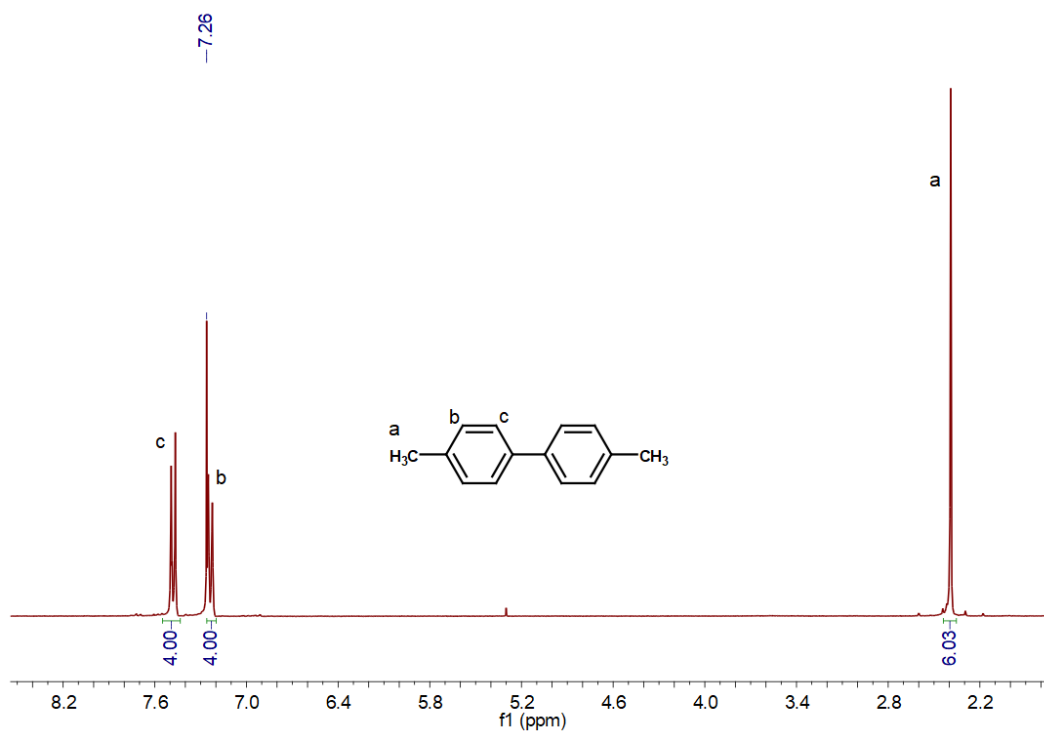


Fig. S10. ^1H NMR (300 MHz, CDCl_3) spectrum of the product of 4,4'-dimethyl-1,1'-biphenyl (Homocoupling of iodotoluene with Pd@ZIF-8).

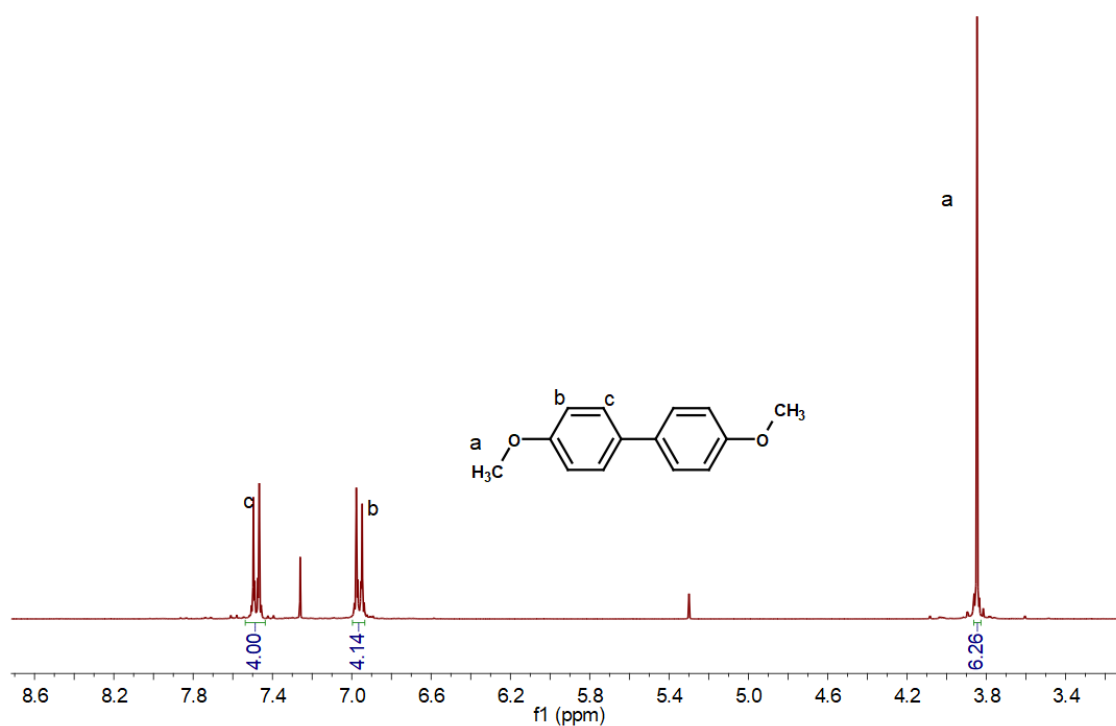


Fig. S11. ^1H NMR (300 MHz, CDCl_3) spectrum of the product of 4,4'-dimethoxy-1,1'-biphenyl (homocoupling of iodoanisole with Pd@ZIF-8).

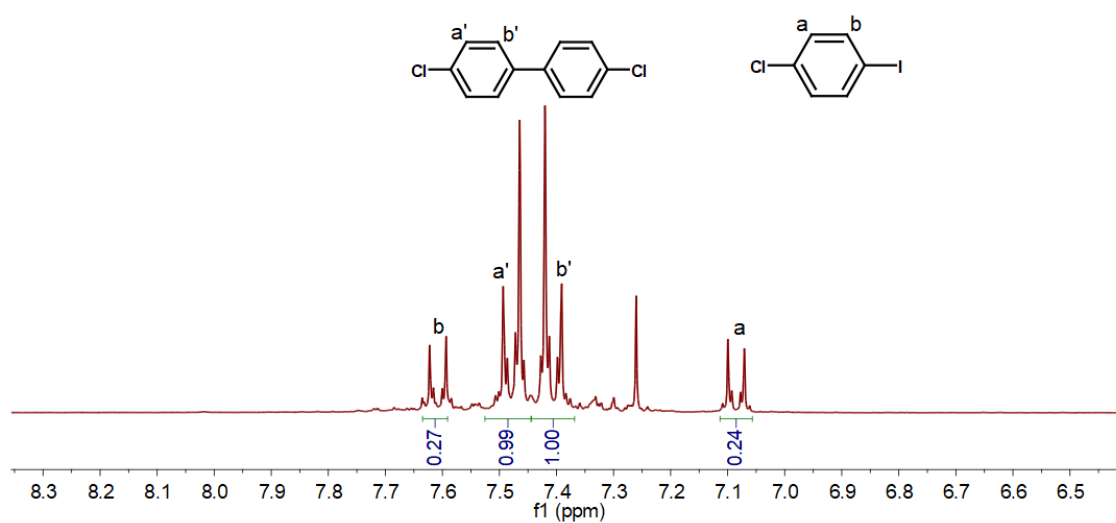


Fig. S12. ^1H NMR (300 MHz, CDCl_3) spectrum of the crude product of 4,4'-dichloro-1,1'-biphenyl (Homocoupling of 1-chloro-4-iodobenzene with Pd@ZIF-8).

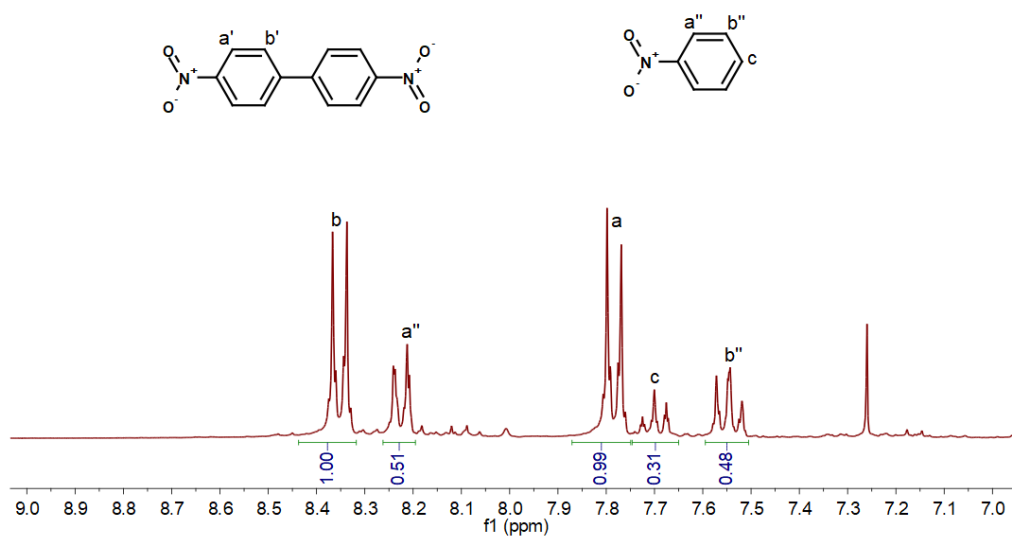


Fig. S13. ^1H NMR (300 MHz, CDCl_3) Spectrum of the crude product of 4,4'-dinitro-1,1'-biphenyl (Homocoupling of 1-iodo-4-nitrobenzene with Pd@ZIF-8).

11. ^1H NMR spectra for various substrates with Au@ZIF-8 as catalyst

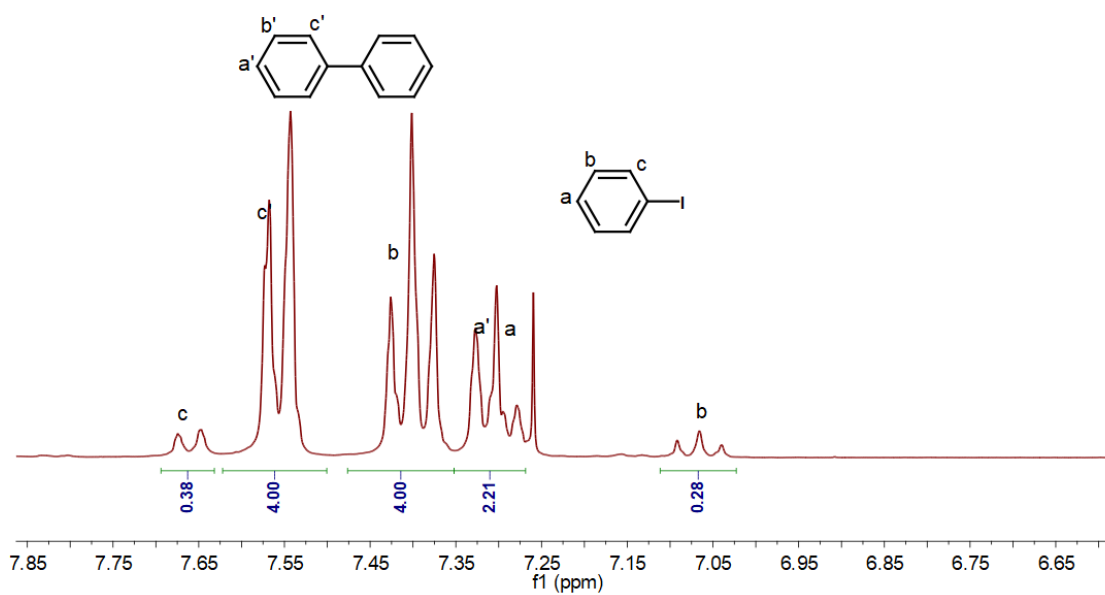


Fig. S14. ^1H NMR (300 MHz, CDCl_3) Spectrum of the crude product of 1,1'-biphenyl (homocoupling of iodobenzene with Au@ZIF-8).

-7.26

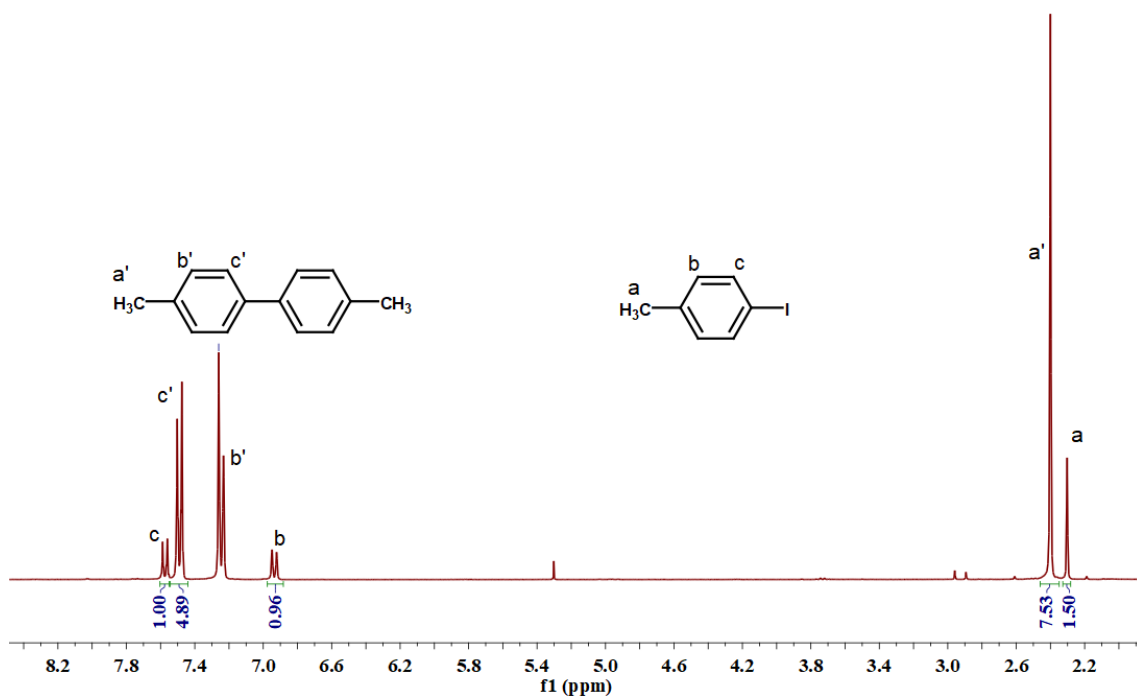


Fig. S15. ¹H NMR (300 MHz, CDCl₃) Spectrum of the crude product of 4,4'-dimethyl-1,1'-biphenyl (Homocoupling of iodotoluene with Au@ZIF-8).

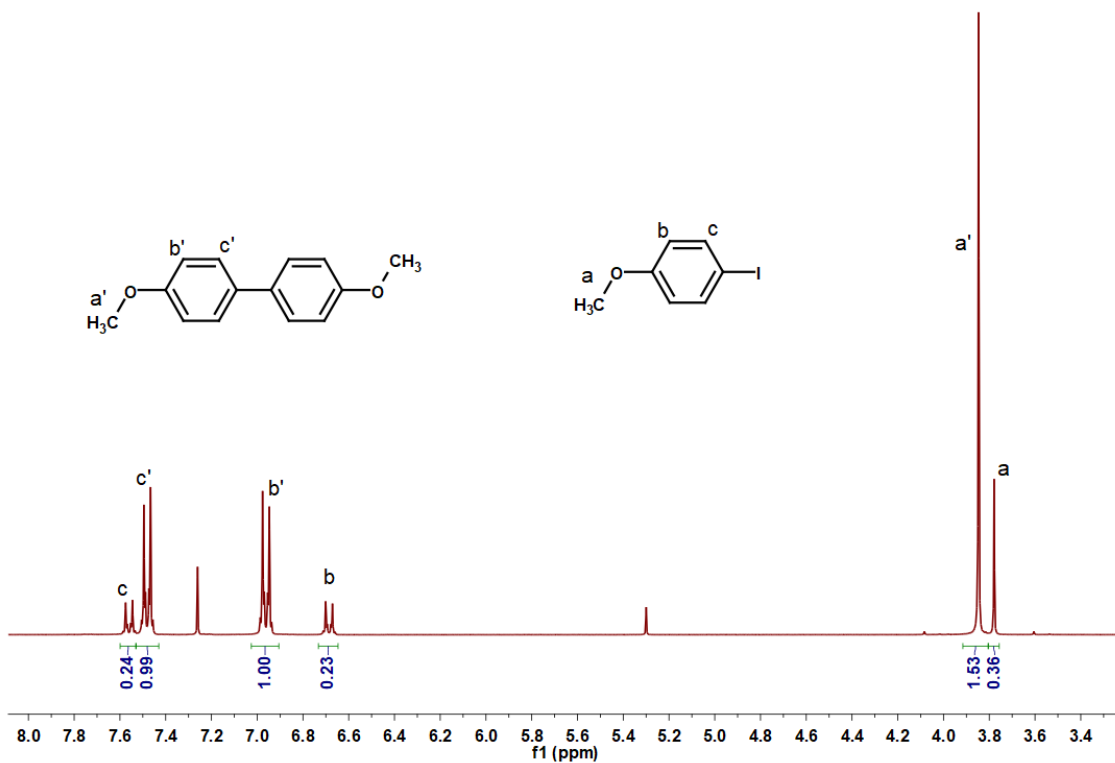


Fig. S16 ¹H NMR (300 MHz, CDCl₃) Spectrum of the crude product of 4,4'-dimethoxy-1,1'-biphenyl (Homocoupling of iodoanisole with Au@ZIF-8).

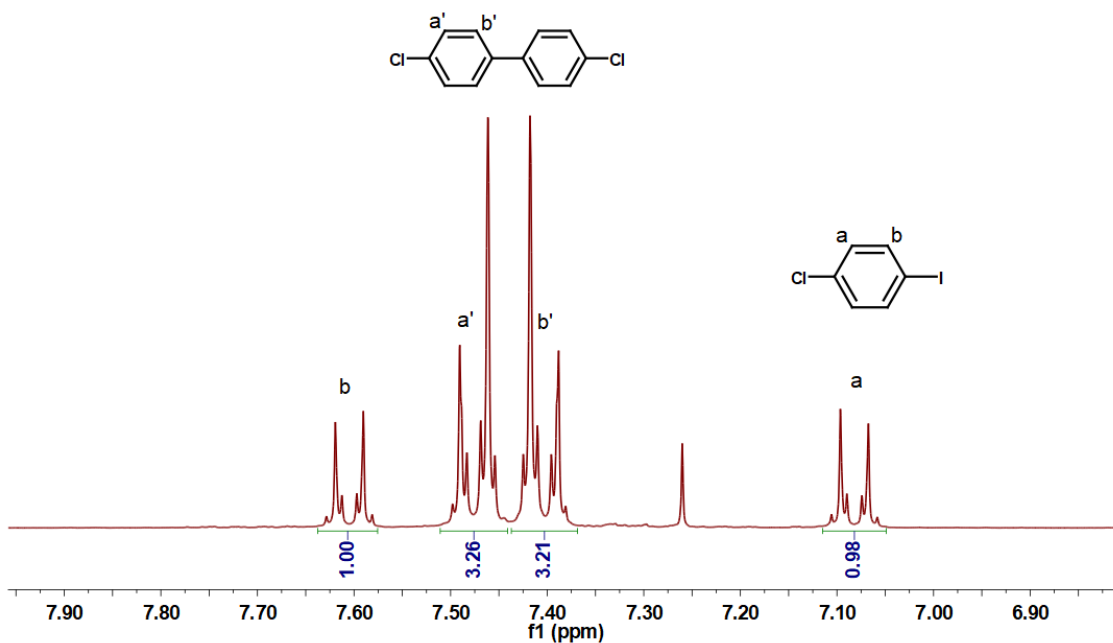


Fig. S17. ¹H NMR (300 MHz, CDCl₃) spectrum of the crude product of 4,4'-dichloro-1,1'-biphenyl (Homocoupling of 1-chloro-4-iodobenzene with Au@ZIF-8).

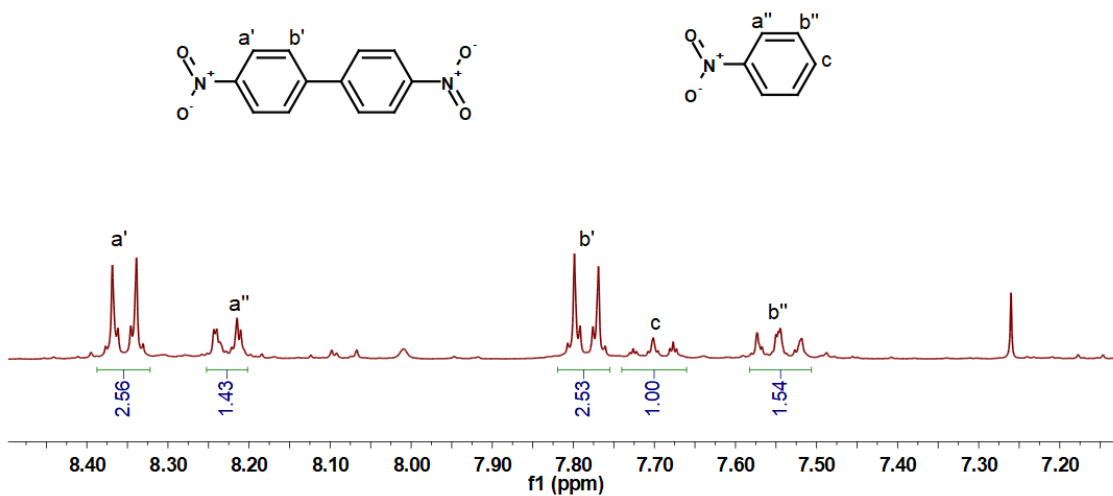


Fig. S18. ¹H NMR (300 MHz, CDCl₃) spectrum of the crude product of 4,4'-dinitro-1,1'-biphenyl (Homocoupling of 1-iodo-4-nitrobenzene with Au@ZIF-8).