

Electronic Supporting Information

EPR-derived structures of flavin radical and iron-sulfur cluster from
Methylosinus sporium 5 reductase

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Supplementary Schemes

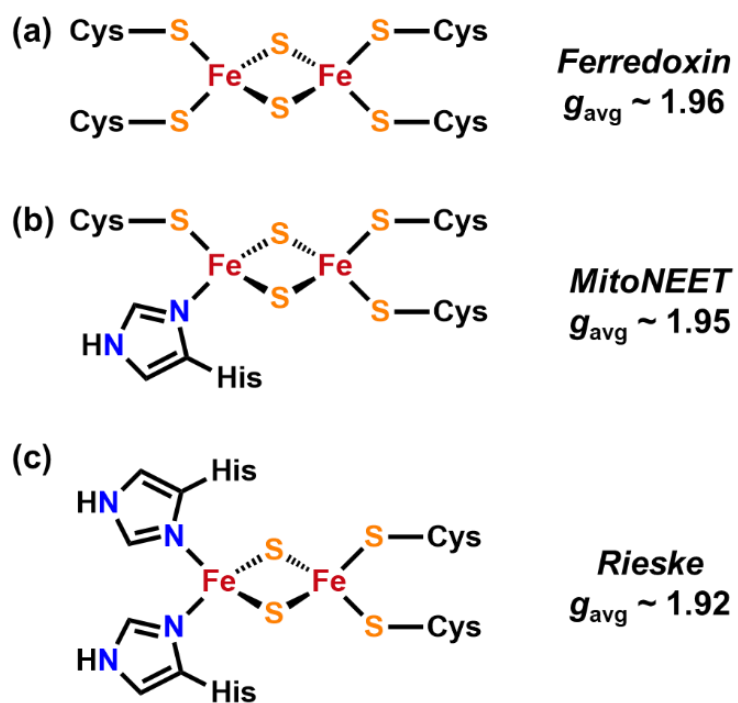
Scheme S1. Alignment of amino acids from ferredoxin domain of sMMO. Gray and red color represent identical and different amino acids, respectively.

	Identity / Similarity	
<i>Methylosinus sporium</i> 5	001	MYQIV ⁰⁸⁰ VIETEDGETCSFECG ⁰⁸¹ PS ⁰⁸² EDVISAGLRQSVILLASCRAGGCATCKADCTDGEYELIDVKVQALPPDEEEDGKVL ⁰⁸³ LCR ⁰⁸⁴
<i>Methylocystis species</i> M	96% / 98%	001 MYQIV ⁰⁸⁰ VIETEDGETCSFECG ⁰⁸¹ PS ⁰⁸² EDVISAGLRQSVILLASCRAGGCATCKADCTDGEYELIDVKVQALPPDEEEDGKVL ⁰⁸³ LCR ⁰⁸⁴
<i>Methylosinus trichosporium</i> OB3b	81% / 85%	001 MYQIV ⁰⁸⁰ VIETEDGETCR-RMRPSEDWISRAEAERN-LLASCRAG-CATCKADCTDGEYELIDVKVQAVPPDEEEDGKVL ⁰⁸³ LCR ⁰⁸⁴
<i>Methylococcus capsulatus</i> Bath	58% / 78%	004 VHTITAVTEDGESLRFECRSDE ⁰⁸⁵ VITAAALRQNI ⁰⁸⁶ FLMSSCREGGCATCKALCSEGDYDLKGC ⁰⁸⁷ SVQALPPEEEEEGLV ⁰⁸⁸ LCR ⁰⁸⁹
<i>Methylovulum miyakonense</i> HT12	53% / 73%	005 -HQV ⁰⁹⁰ TI ⁰⁹¹ V ⁰⁹² TEDHESITFD ⁰⁹³ CRSDE ⁰⁹⁴ VITAAV ⁰⁹⁵ RQDIY ⁰⁹⁶ LMSSCREGGCATCKGVCSEGDYVIGK ⁰⁹⁷ VSAQALPSQEEEE ⁰⁹⁸ GMV ⁰⁹⁹ LCR ¹⁰⁰
		081 TFPRSDLHLIVPYTYD ⁰⁹⁶
		081 TFPRSDLHVVVPYTYD ⁰⁹⁶
		078 TFPRSDLHLLVPYTYD ⁰⁹³
		084 TYPKTDLELELPYTHC ⁰⁹⁹
		084 CYP ¹⁰⁰ TTDIEVEVPYTYE ¹⁰¹

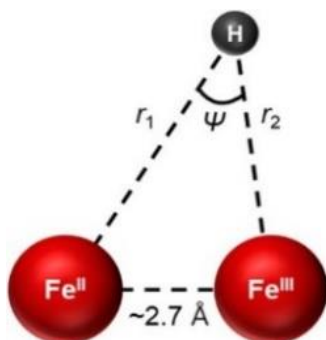
Scheme S2. Alignment of amino acids from FAD binding domain of sMMO. Gray and red color represent identical and different amino acids, respectively.

	Identity / Similarity	
<i>Methylosinus sporium</i> 5	097	RISFEAIQTNWLAEIVECDRVSSNVVRL ¹⁷⁵ LQPLTADGAAPISLNFAPGQFVDIEIPGTHTRRSYSMASVAE-DGRLEFFI ¹⁷⁶
<i>Methylocystis species</i> M	89% / 93%	097 RISFQAIQTNWLAEITECDRVSSNVVRLV ¹⁷⁵ LQPLTADGAAPISLNFAPGQFVDIEIPGTHTRRSYSMASVAE-DGRLEFFI ¹⁷⁶
<i>Methylosinus trichosporium</i> OB3b	78% / 84%	094 RISFEAIQTNWLAEILACDRVSSNVVRLV ¹⁷⁵ LQ- ¹⁷⁶ RSRPM ¹⁷⁷ AARISLNFVPGQFVDIEIPGTHTRRSYSMASVAE-DGQLEFI ¹⁷⁸
<i>Methylococcus capsulatus</i> Bath	47% / 66%	100 RISFGEVGS-FEAEVVLN ¹⁷⁵ WVSSNTVQFLLQK- ¹⁷⁶ RPDECN ¹⁷⁷ RGVKFEPGQFMDLTIPGTDVRSYSANLNP ¹⁷⁸ EGRLEFLI ¹⁷⁹
<i>Methylovulum miyakonense</i> HT12	44% / 66%	100 RISFSP ¹⁷⁵ EGMDFEAEVVGLEQISIN ¹⁷⁶ VVKFQLRR ¹⁷⁷ TGDDK----TIKFEAGQFFDLEIPGTE ¹⁷⁸ TRRSYSPANISNSQGDLEFLI ¹⁷⁹
		176 RLLPDGAFSNYLRTQASV ¹⁷⁵ QQRVALRGPAGSFFLH-KSE-RPRFFVAGGTGLSPVLSMIRQLKKEADPQATLFFGVTN ¹⁷⁶ YE ¹⁷⁷
		176 RLLPDGAFSNYLRTQARV ¹⁷⁵ QQRVALRGPAGSFFLH-KSE-RPRFFVAGGTGLSPVLSMIRQLHKE ¹⁷⁶ SDPQATLFFGVTN ¹⁷⁷ YE ¹⁷⁸
		172 RLLPDGAFSKFLQTEAKV ¹⁷⁵ GM ¹⁷⁶ RV ¹⁷⁷ DLRGPAGSFFLHDHGG-RSRV ¹⁷⁸ FVAGGTGLSPVLSMIRQLGKASDP ¹⁷⁹ SATLFFGVTN ¹⁸⁰ RE ¹⁸¹
		178 RVLPEGRFSDYL ¹⁷⁵ RNDARV ¹⁷⁶ QVLSVKGPLGVFLKERGM-APRY ¹⁷⁷ FVAGGTGLAPV ¹⁷⁸ SMV ¹⁷⁹ RQM ¹⁸⁰ EW ¹⁸¹ TAPNETRIYFGVNT ¹⁸² EP ¹⁸³
		176 RIVDGGK ¹⁷⁵ FSEFLKKEAKV ¹⁷⁶ QRLKAKG ¹⁷⁷ PSGVFGLK-ENGFT ¹⁷⁸ PRY ¹⁷⁹ FVAGGTGLAPILSMV ¹⁸⁰ RHMKEW ¹⁸¹ GE ¹⁸² PQKCVIYFGVNT ¹⁸³ EA ¹⁸⁴
		254 ELFYVEELRALQKAMP ¹⁷⁵ SLDVQVAVVNATEANGVAKGTVIDLMRAELEKLRGAPDIYLCGPPGMI ¹⁷⁶ EAAFDAAATAGVPKEQ ¹⁷⁷
		254 ELFYVDELKALQHAMP ¹⁷⁵ SLDVQIAVNVSENGVAKGTVIDLQDELGRRAEK ¹⁷⁶ PDYIYLCGPPGMI ¹⁷⁷ DAFAAASAGVPKEQ ¹⁷⁸
		251 ELFYVDELKTLAQSMPTL ¹⁷⁵ GVRIAVVNDGGNGVDKGTVIDLRAELEKSDAK ¹⁷⁶ PDYIYLCGPPGMI ¹⁷⁷ EAAFAAATAGVPKEQ ¹⁷⁸
		257 ELFYIDELKSLERSMRNL ¹⁷⁵ TVKACVWHPSGDWEGEGSSPIDALREDESSDAN ¹⁷⁶ PDYIYLCGPPGMI ¹⁷⁷ DAACELVRSRGI ¹⁷⁸ PEQ ¹⁷⁹
		255 EIFHLDLEQLAQMPTLELR ¹⁷⁵ NCVWKCSDDW ¹⁷⁶ HCEKGSVVDILRRDLVETGAK ¹⁷⁷ PDYIYLCGPPGMV ¹⁷⁸ DATFVACADLGI ¹⁷⁹ PKER ¹⁸⁰
		334 VYLEKFLASG ³⁴³
		334 VYLEKFLASG ³⁴³
		331 VYLEKFLASG ³⁴⁰
		337 VFEKFLPSG ³⁴⁶
		335 IYLEKFLPSG ³⁴⁴

Scheme S3. The schematic drawing of three types of $[2\text{Fe}-2\text{S}]^+$ clusters. (a) Ferredoxin, (b) MitoNEET, and (c) Rieske-type.



Scheme S4. Schematic representation of a proton positioned nearby from the Fe^{II} and Fe^{III} ions of the [2Fe-2S]⁺ cluster.



To estimate the position of the proton from the [2Fe-2S]⁺ cluster, the dipolar hyperfine coupling value of A_{dip} is used by considering the spin projection factor of the respective nucleus. The spin projection factors of two metal atoms, p_A and p_B , are given as follows:

$$p_A = \frac{S(S+1) + S_A(S_A+1) - S_B(S_B+1)}{2S(S+1)}$$

$$p_B = \frac{S(S+1) - S_A(S_A+1) + S_B(S_B+1)}{2S(S+1)}$$

where S is total spin of the cluster, and S_A and S_B are the local spin of each metal atom.^{S1,S2} The spin projection factors of Fe(II) ($S_A = 2$) and Fe(III) ($S_B = 5/2$) of the [2Fe-2S]⁺ cluster are $-4/3$ (p_A) and $+7/3$ (p_B), respectively. Thus, A_{dip} can be calculated by equations as follows:

$$A_{dip} = c \left(-\delta, -\Gamma + \frac{\delta}{2}, \Gamma + \frac{\delta}{2} \right)$$

$$\delta = p_A r_1^{-3} + p_B r_2^{-3}$$

$$\Gamma = \frac{3}{2} \sqrt{p_A^2 r_1^{-6} + p_A p_B r_1^{-3} r_2^{-3} \cos 2\psi + p_B^2 r_2^{-6}}$$

where c is $g_e \mu_e g_n \mu_n$ and equals $79 \text{ MHz } \text{Å}^3$ for a proton, and geometrical parameters r and ψ are defined in **Scheme S3**.^{S3,S4} From the equation, the distances between the iron ions and the exchangeable proton are estimated to be $r_1 = 4.1\text{-}4.4$ and $r_2 = 3.50\text{-}3.60$ Å with $\psi \sim 40^\circ$. The distance between the Fe^{II} and Fe^{III} centers can also be calculated in accordance with the law of cosines and was estimated to be ~ 2.7 Å.^{S5}

Supplementary Tables

Table S1. The simulation parameters for ^1H Davies ENDOR spectra of the $[\text{2Fe-2S}]^+$ cluster.

	$A(^1\text{H})$ (MHz)	$[\alpha, \beta, \lambda]$ ($^\circ$)
Exchangeable proton	$[6.15 \pm 0.35, -3.50 \pm 0.15, -2.90 \pm 0.15]$	$[30, 90, 50]$
Non-exchangeable proton	$[2.32 \pm 0.12, 11.70 \pm 0.20, 1.58 \pm 0.08]$	$[80, 30, 90]$

Table S2. Simulation parameters for ^{14}N three-pulse ESEEM and HYSCORE spectra of the $[\text{2Fe-2S}]^+$ cluster.

	N_1	N_2
$A(^{14}\text{N})$ (MHz)	[1.2, 1.2, 1.2]	[0.9, 0.6, 0.6]
$A_{\text{iso}}(^{14}\text{N})$ (MHz)	1.2	0.7
e^2qQ/h (MHz)	3.4	3.0
η	0.4	0.9
$[Q_a, Q_b, Q_c]$ ($^\circ$)	[-75, 100, 60]	[0, 90, 0]

Table S3. The reported hyperfine and quadrupole coupling parameters of ^{14}N s coupled to the $[\text{2Fe-2S}]^+$ cluster.

	A_{iso} (MHz)	e^2qQ/h (MHz)	Refs.
Rieske or Rieske-type			
bc1 (bovine)	3.55	2.25	S6
	5.20	2.93	
bc1 (bovine) + UHDBT	3.40	2.30-2.65	S7
	5.30	2.60-3.00	
<i>Cytochrome b6f</i> (spinach)	4.58	2.70	S8
	3.75	2.70	
<i>Burkholderia cepacia</i>	3.70	2.15	S9
	4.70	3.85	
<i>B. cepacia</i> AC 1100	3.87	2.40	S10
	4.90	2.32	
<i>Pseudomonas putida</i>	3.56	2.43	S9
	4.78	2.31	
<i>Pseudomonas cepacia</i>	4.28	2.60	S11
	5.49	2.30	
2,4,5-Trichlorophenoxyacetate Monooxygenase	3.40	2.36	S10
	5.10	2.52	
Ferredoxin or ferredoxin-type			
<i>P. putida</i>	1.11	3.27	S10
<i>Clostridium pasteurianum</i>	0.61	3.29	S12
<i>Arum maculatum</i>	1.10	3.32	S6
<i>Spirulina platensis</i>	1.01	3.52	S6
<i>E. coli</i> (reduced)	1.06	3.41	S6
<i>E. coli</i> (oxidized)	1.10	3.30	S13
<i>Porphyra umbilicalis</i>	1.16	3.24	S14
	0.40	3.04	
<i>Yarrowia lipolytica</i>	0.90	3.10	S15
<i>Adrenodoxin</i>	1.05	3.10	S16
	0.65	2.90	
<i>Arthrospira plantensis</i> ferredoxin	1.16	3.10	S16
	0.72	2.90	

Table S4. Cartesian coordinates for the optimized geometries.**(a) Neutral flavin radical**

N	-2.414300	-0.002840	-0.350910
C	-3.462390	-0.897250	-0.246360
O	-4.627580	-0.565660	-0.307690
N	-3.138700	-2.262240	-0.051520
C	-1.874220	-2.791590	0.086793
O	-1.632360	-3.978330	0.277746
C	-0.837230	-1.785230	-0.022260
N	0.456424	-2.186840	0.095316
C	1.504528	-1.294860	0.027977
C	2.828325	-1.711820	0.191791
C	3.887589	-0.817000	0.124517
C	5.303507	-1.301680	0.305671
C	3.605500	0.547178	-0.111610
C	4.717485	1.562933	-0.185910
C	2.283268	0.960011	-0.273760
C	1.208267	0.066410	-0.215640
N	-0.127170	0.473334	-0.397930
C	-1.171850	-0.432620	-0.255420
C	-0.439600	1.848180	-0.819620
C	-0.571960	2.888741	0.313996
O	-0.743920	4.156021	-0.303730
C	-1.747170	2.618377	1.259150
O	-2.972070	2.721223	0.552885
H	-1.750510	3.397190	2.026123
H	-1.641730	1.648105	1.758025
H	-1.694010	4.220592	-0.481240
H	0.356189	2.936861	0.893044
H	-1.381320	1.807795	-1.363370
H	0.331908	2.182590	-1.512290
H	4.326743	2.563902	-0.372580
H	5.424233	1.324163	-0.987180
H	5.292986	1.596094	0.744891
H	5.331793	-2.378230	0.479685
H	5.916107	-1.088840	-0.576160
H	5.789284	-0.812210	1.155769
H	2.093608	2.010381	-0.442460
H	3.018837	-2.764480	0.374478
H	0.599304	-3.178510	0.258702
H	-3.923420	-2.896660	0.021065
H	-3.120640	1.856585	0.129258

(b) Anionic flavin radical

N	-2.393210	0.005143	-0.330220
C	-3.448690	-0.864050	-0.246410
O	-4.622990	-0.512970	-0.312350
N	-3.136460	-2.216060	-0.069330
C	-1.870250	-2.794930	0.079038
O	-1.761710	-4.001410	0.255585
C	-0.783340	-1.820840	-0.010800
N	0.493493	-2.261470	0.116041
C	1.470644	-1.329430	0.038786
C	2.819761	-1.727500	0.190895
C	3.885654	-0.841370	0.122710
C	5.300808	-1.342940	0.294573
C	3.623364	0.525053	-0.105480
C	4.742748	1.535137	-0.182030
C	2.294739	0.946729	-0.259690
C	1.221315	0.057218	-0.199190
N	-0.113000	0.461644	-0.372360
C	-1.147640	-0.470040	-0.234840
C	-0.425130	1.816773	-0.823720
C	-0.560970	2.884238	0.281675
O	-0.825780	4.136787	-0.353690
C	-1.690840	2.586511	1.271689
O	-2.940450	2.625999	0.602347
H	-1.700620	3.376591	2.029453
H	-1.529180	1.625828	1.773354
H	-1.782440	4.114978	-0.507490
H	0.383766	2.994508	0.823344
H	-1.369170	1.768735	-1.363740
H	0.347130	2.142130	-1.523370
H	4.351766	2.538702	-0.364540
H	5.450330	1.305212	-0.988450
H	5.328536	1.573263	0.744996
H	5.310111	-2.421890	0.461970
H	5.918142	-1.136970	-0.588030
H	5.802444	-0.868950	1.146720
H	2.112575	2.001270	-0.423490
H	2.984989	-2.785440	0.364386
H	-3.921760	-2.848300	-0.004520
H	-3.028010	1.744914	0.171956

Supplementary Figures

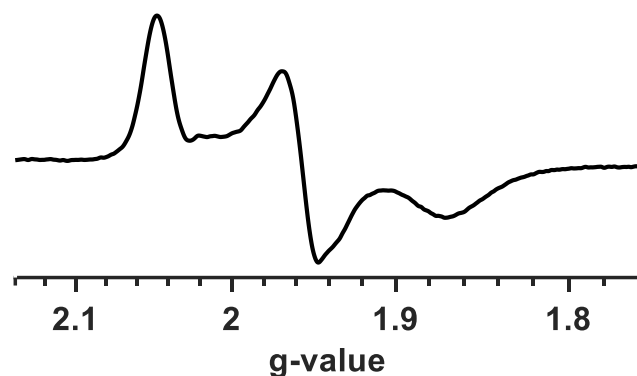


Fig. S1 X-band CW-EPR spectrum of the reduced MMOR after addition of $\text{Na}_2\text{S}_2\text{O}_4$ measured at 35 K. The experimental conditions: microwave frequency = 9.64 GHz, microwave power = 1 mW, modulation amplitude = 10 G, modulation frequency = 100 kHz, time constant = 40.96 ms.

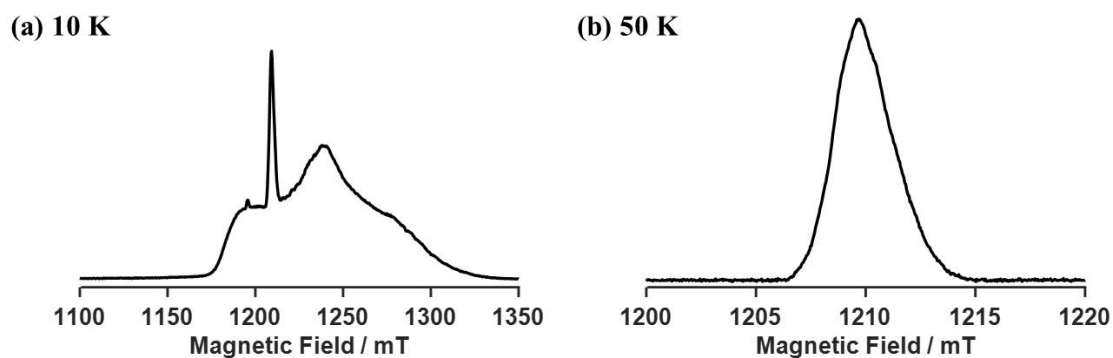


Fig. S2 Q-band ESE spectra of reduced MMOR at different temperatures. The experimental conditions: microwave frequency = 33.9 GHz, microwave power = 10 dB, pulse sequence = $\pi/2(32 \text{ ns}) - \tau - \pi(64 \text{ ns})$, $\tau = 300 \text{ ns}$, temperature = (a) 10 K, (b) 50 K.

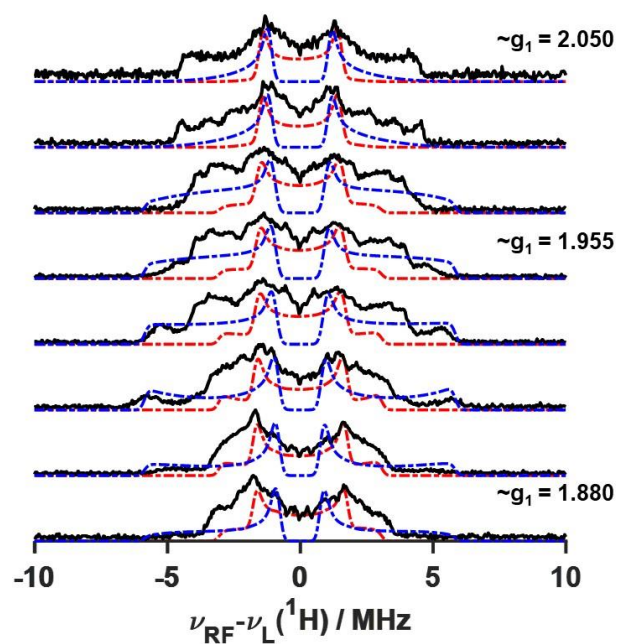
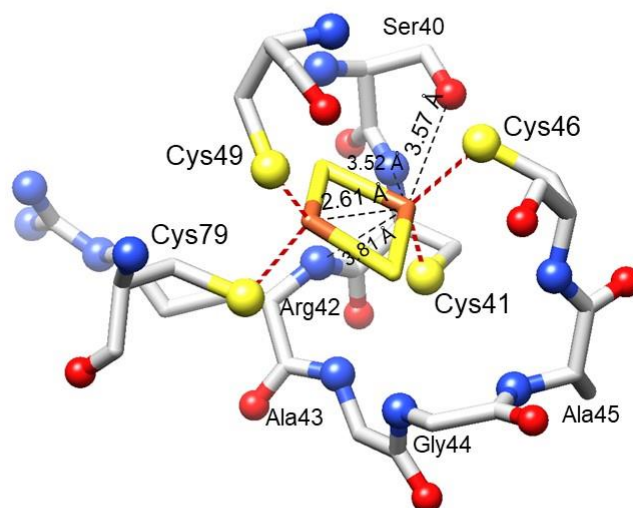


Figure S3. Q-band ^1H Davies ENDOR of $[2\text{Fe-2S}]^+$ cluster in MMOR (black line) and simulated spectra (dashed red line: exchangeable proton, dashed blue line: non-exchangeable proton). The simulation parameters are shown in table S1. The experimental conditions: microwave frequency = 33.9 GHz, $\pi/2 = 32$ ns, $\tau = 300$ ns, RF length = 20 μs , temperature = 10 K.

(a) <i>Methylophilus</i> <i>sporium 5</i>	³⁹ SCRAGGCATC...L ⁸⁰ CR
<i>Anthrospira</i> <i>platensis</i>	⁴⁰ SCRAGACSTC...T ⁷⁹ CV

(b)



Anthrospira platensis

PDB ID: 4FXC

Figure S4. (a) Amino acid sequences of residues located around the $[2\text{Fe}-2\text{S}]^+$ cluster of *Methylophilus sporium 5* and *Anthrospira platensis*, and the immediate environment of the $[2\text{Fe}-2\text{S}]^+$ cluster of the ferredoxin of (b) *Anthrospira platensis* with selected bond distances of Fe–Fe and proximal H...Fe. Note that the positions of protons in the ferredoxin domain of *Anthrospira platensis* were not well resolved by X-ray crystallography. Color code: orange, Fe; yellow, S; red, O; blue, N; gray, C; white, H.

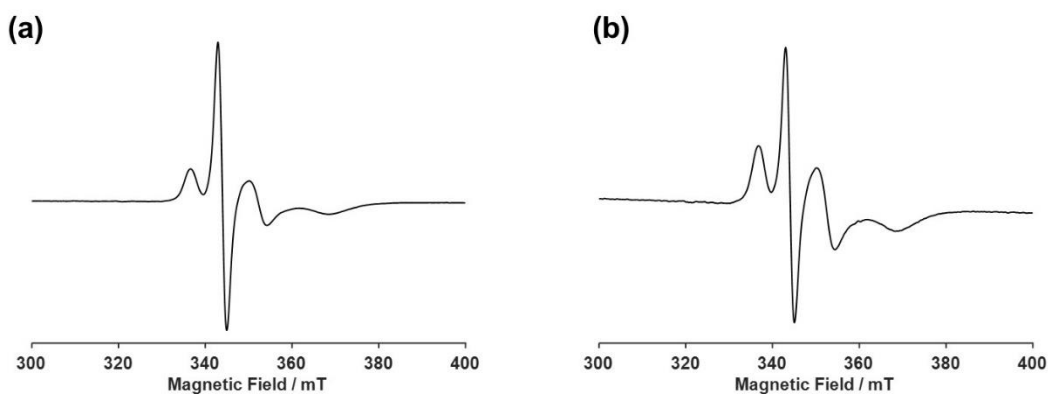


Figure S5. CW-EPR spectra of MMOR (a) and MMOR+MMOB (b). The experimental conditions: microwave frequency = 9.64 GHz, microwave power = 1 mW, modulation amplitude = 10 G, modulation frequency = 100 kHz, time constant = 40.96 ms, temperature = 35 K.

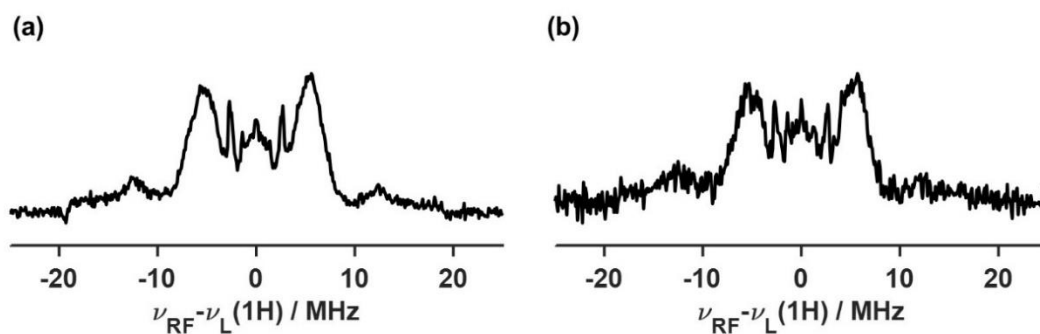


Figure S6. 34 GHz ^1H Davies ENDOR of neutral flavin radical species in (a) MMOR and (b) MMOR+MMOB. The experimental conditions: microwave frequency = 33.9 GHz, Pulse sequence = 64 ns-32 ns-64 ns (π - $\pi/2$ - π), τ = 300 ns, RF length = 20 μs , temperature = 50 K.

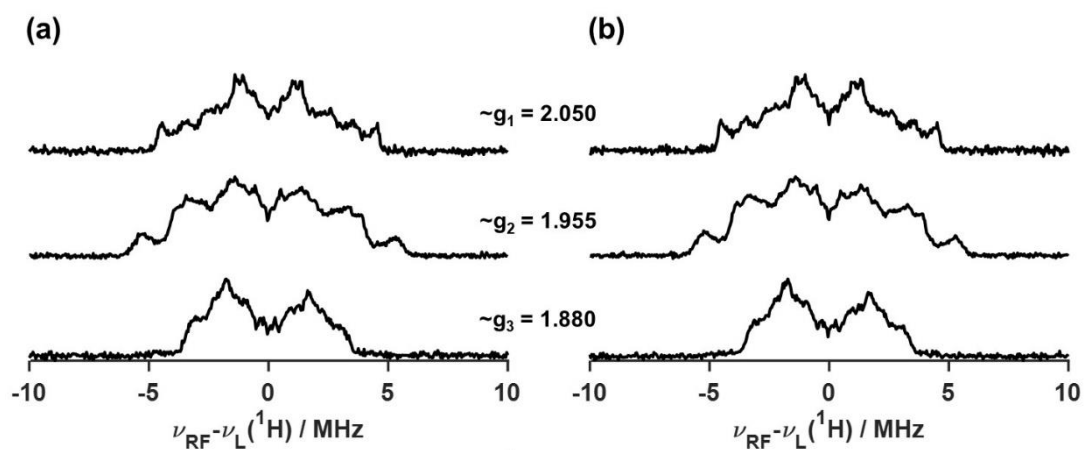


Figure S7. 34 GHz ^1H Davies-ENDOR of $[\text{2Fe-2S}]^+$ cluster in (a) MMOR and (b) MMOR+MMOB. The experimental conditions: microwave frequency = 33.9 GHz, $\pi = 64$ ns, $\pi/2 = 32$ ns, $\tau = 300$ ns, RF length = 20 μs , temperature = 10 K.

References

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