Supporting Information for

Aggregation-induced enhanced emission-type cruciform luminophore constructed by carbazole exhibiting mechanical force-induced luminescent enhancement and chromism

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Table S1. Electrochemical data and HOMO/LUMO energy levels of MDCS-BC.

Compound	$E_{\mathrm{ox}}^{\mathrm{a}}(\mathrm{V})$	$HOMO^{b}(eV)$	LUMO ^c (eV)	$Eg^{d}(eV)$
MDCS-BC	1.15	-5.49	-2.57	2.92

^{*a*}Oxidation potential was decided with Ag/Ag⁺ electrode as reference.

^{*b*}HOMO energy levels was obtained by comparing with an external reference, the ferrocene/ferrocenium (Fc/Fc⁺, 4.8 eV relative to vacuum).

^{*c*}The LUMO energy level was estimated by the equation: $E_{\text{LUMO}} = E_{\text{HOMO}} + E_{\text{g}}$.

 ${}^{d}E_{g}$ was estimated from the onset of the absorption spectra ($E_{g} = 1240/\lambda_{onset}$).



Fig. S1 Cyclic voltammogram of **MDCS-BC** measured in DCM with tetrabutylammonium hexafluorophosphate (0.1 mol/L) as electrolyte at a scan rate of 50 mV/s.



Fig. S2 DLS data of MDCS-BC ($1.0 \times 10^{-5} \text{ mol } \text{L}^{-1}$) in THF/water ($f_w = 95\%$).



Fig. S3 Maximum fluorescent emissions of **MDCS-BC** (at 230 °C) upon repeating treated by grinding and heating.



Fig. S4 Maximum fluorescent emissions of **MDCS-BC** upon repeating treated by grinding and fuming with DCM.







Fig. S6 13 C NMR (100 MHz) spectrum of 3



Fig. S7 MALDI/TOF MS spectrum of 3.







Fig. S9 ¹H NMR (600 MHz) spectrum of MDCS-BC.



Fig. S10 ¹³C NMR (150 MHz) spectrum of MDCS-BC.



Fig. S11 MALDI/TOF MS spectrum of MDCS-BC.



Fig. S12 MALDI/TOF MS spectrum of MDCS-BC.