

(a) 1,4H-AMTN:0



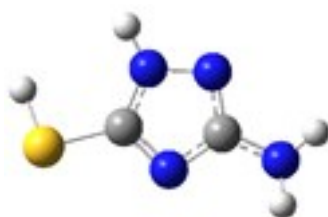
(b) 1,2H-AMTN:12.1



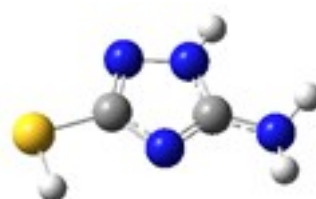
(c) 2,4H-AMTN:19.5



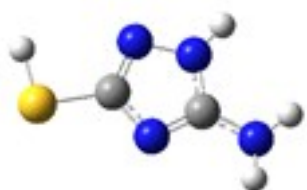
(d) 1H-AMTL:4.66



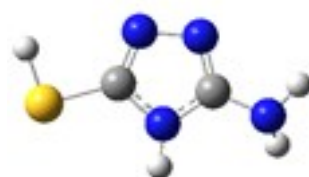
(e) 1H-AMTL-it:6.11



(f) 2H-AMTL:5.51



(g) 2H-AMTL-it:5.35



(h) 4H-AMTL:13.1

Figure S1. Optimized tautomers of AMT and the energies (in kcal/mol) relative to structure (a) at B3LYP-D3(BJ)/6-311++G(d, p) level.

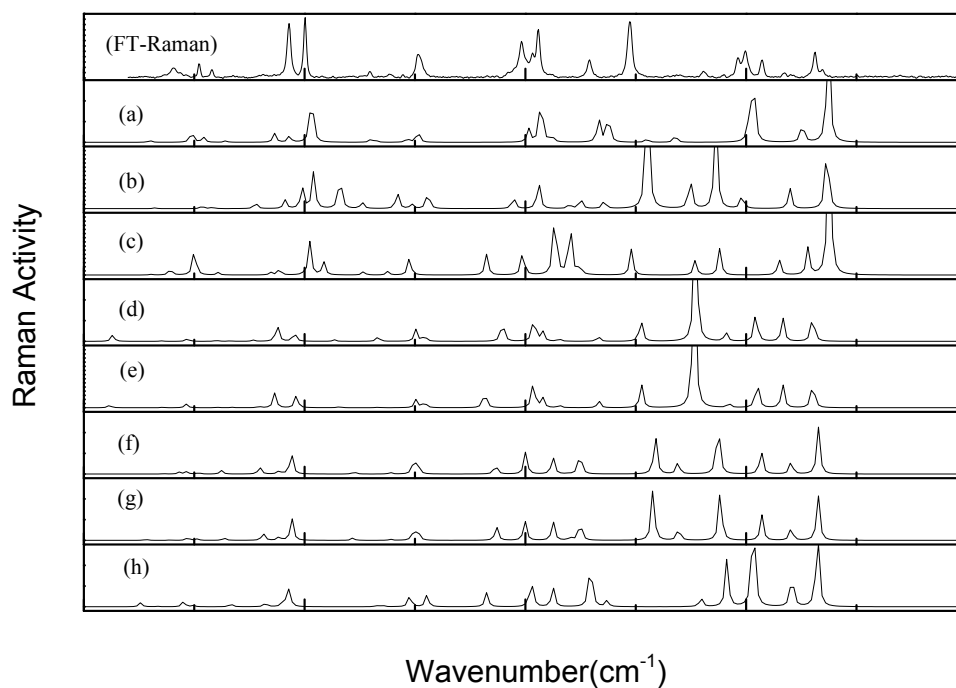


Figure S2. FT-Raman and calculated Raman activities (a), (b), (c), (d),(e), (f), (g) and (h) for the optimized isomers in Figure S1 at B3LYP-D3(BJ)/6-311++G(d, p) level.

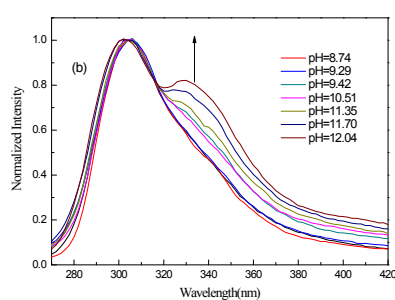


Figure S3. (a) UV absorption and (b) emission ($\lambda_{\text{ex}}=230$ nm) spectra for AMT (4.78×10^{-5} mol/L) in CH_3OH with different volume of $[(\text{Bu})_4 \text{N}]^+\text{OH}^-$ (**1.53** mol/L).

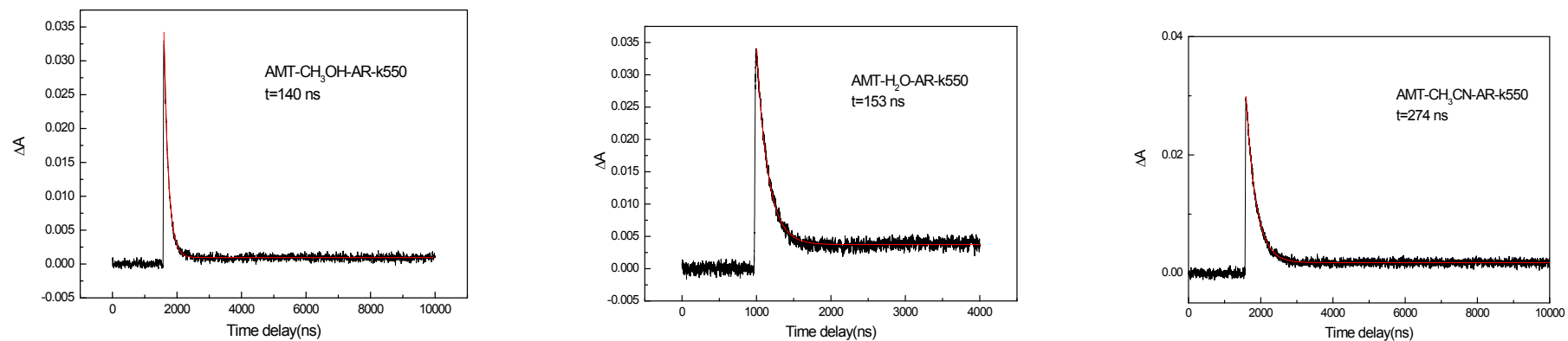


Figure S4. Kinetics of ^3AMT measured at 550 nm in CH_3CN , CH_3OH and H_2O and fitted with a single exponential function, the time constants are listed close to the curves.

Notes: Based on the experimental results, as shown in Figure S4, the solvents induce transient absorption spectra not return to the baseline due to its polarity, especially the hydrogen bonding interaction effect on the excited state structure.

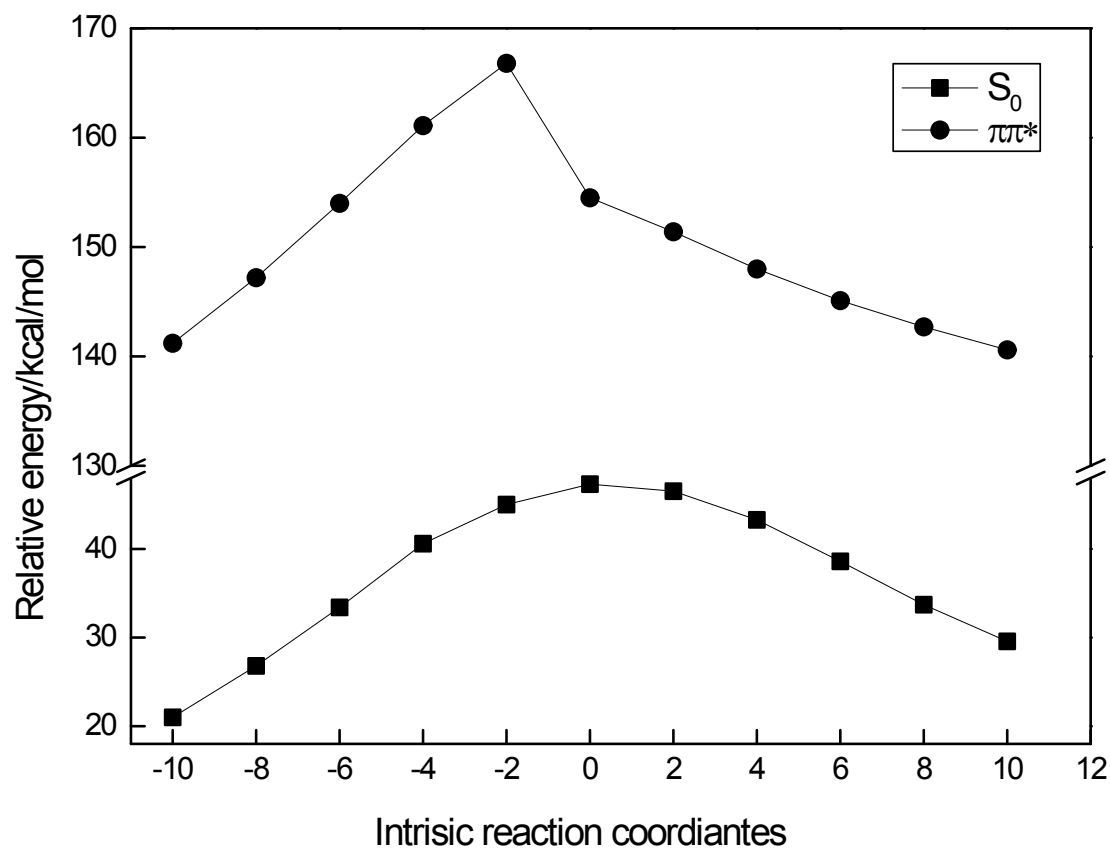


Figure S5. Calculated energies (kcal/mol) of S_0 (circle) and $\pi\pi^*$ (rectangular) states relative to ground state thione AMT on the selected IRC points at B3LYP(TD)-D3(BJ)/6-31G* level.

Table S1. Experimental and calculated vibrational frequencies at B3LYP-D3(BJ)/6-311++G(d, p) level and assignments of thione AMT in gas phase.

Modes	Calc	Exp		Description (PED%)
	(Raman activity /IR activity)	FT-raman	FT-IR	
ν_1	3684(88.1/104)			$N_1H_6(99)$ stretch
ν_2	3662(57.3/65.9)			$N_4H_{11}(99)$ stretch
ν_3	3644(73.1/37.8)			$N_7H_8(58)+N_7H_9(42)$ stretch
ν_4	3549(242/30.9)			$N_7H_8(42)+N_7H_9(58)$ stretch
ν_5	1686(53.0/333)	1655	1647	$N_2C_3(35)$ stretch + $N_7C_3(23)$ stretch + $H_9N_7H_8(19)$ bend
ν_6	1628(9.39/31.8)	1585	1589	$N_2C_3(13)$ stretch + $H_9N_7H_8(59)$ bend
ν_7	1517(26.2/505)	1498	1546	$N_1C_5(33)$ stretch + $N_4C_3(14)$ stretch + $H_6N_1N_2(18)$ bend + $H_{11}N_4C_3(13)$ bend
ν_8	1508(12.9/8.97)		1482	$N_4C_3(19)$ stretch + $N_7C_3(15)$ stretch + $H_6N_1N_2(19)$ bend + $N_1N_2C_3(14)$ bend
ν_9	1340(3.24/21.4)	1402	1391	$H_6N_1N_2(32)$ bend + $H_{11}N_4C_3(30)$ bend + $C_5N_1N_2(12)$ bend
ν_{10}	1275(1.46/2.19)	1344		$N_1C_5(32)$ stretch + $C_5N_1N_2(29)$ bend
ν_{11}	1188(12.2/49.4)	1236	1239	$N_7C_3(11)$ stretch + $N_1N_2(12)$ stretch + $H_{11}N_4C_3(19)$ bend + $H_8N_7C_3(10)$ bend
ν_{12}	1166(9.87/109)	1144	1143	$N_1N_2(21)$ stretch + $S_{10}C_5(18)$ stretch + $H_{11}N_4C_3(13)$ bend + $H_8N_7C_3(25)$ bend
ν_{13}	1060(2.37/45.1)		1068	$N_4C_3(17)$ stretch + $N_1N_2(36)$ stretch + $H_8N_7C_3(23)$ bend
ν_{14}	1035(19.2/11.8)	1028		$N_2C_3(18)$ stretch + $N_4C_3(15)$ stretch + $N_1N_2(10)$ stretch + $N_4C_3N_2(42)$ bend
ν_{15}	1008(4.83/8.83)	991		$N_4C_3(15)$ stretch + $N_1N_2C_3(42)$ bend
ν_{16}	757(4.38/8.96)	758	755	$N_7C_3(27)$ stretch + $N_4C_3N_2(19)$ bend + $N_1N_2C_3(42)$ bend
ν_{17}	732(1.40/83.0)	723	706	$H_9N_7C_3N_4(11)$ tors + $N_4C_3N_2N_1(25)$ tors + $N_7N_4N_2C_3(34)$ out
ν_{18}	660(0.795/154)	692		$H_8N_7C_3N_4(24)$ tors + $H_9N_7C_3N_4(11)$ tors + $C_5N_1N_2C_3(12)$ tors + $S_{10}N_1N_4C_5(29)$ out

v ₁₉	649(0.804/47.7)	646	639	H ₆ N ₁ N ₂ C ₃ (11)tors+H ₈ N ₇ C ₃ N ₄ (22)tors+C ₅ N ₁ N ₂ C ₃ (24)tors+ S ₁₀ N ₁ N ₄ C ₅ (16)out
v ₂₀	518(9.56/94.7)	501	513	S ₁₀ C ₅ (21)strech+C ₅ N ₁ N ₂ (10)bend+H ₆ N ₁ N ₂ C ₃ (13)tors+H ₁₁ N ₄ C ₃ N ₇ (31)tors
v ₂₁	513(7.47/72.1)		497	S ₁₀ C ₅ (22)stretch+C ₅ N ₁ N ₂ (11)bend+H ₆ N ₁ N ₂ C ₃ (11)tors+H ₁₁ N ₄ C ₃ N ₇ (25)tors
v ₂₂	467(2.15/28.2)	465	467	N ₇ C ₃ N ₂ (12)bend+C ₅ N ₁ N ₂ (45)bend+H ₁₁ N ₄ C ₃ N ₇ (16)tors
v ₂₃	432(2.85/4.25)	399		N ₇ C ₃ N ₂ (37)bend+S ₁₀ C ₅ N ₄ (18)bend+H ₁₁ N ₄ C ₃ N ₇ (16)tors
v ₂₄	322(0.471/5.10)			N ₄ C ₃ N ₂ N ₁ (36)tors+N ₇ N ₄ N ₂ C ₃ (36)out
v ₂₅	271(1.68/38.6)	289		N ₇ C ₃ N ₂ (12)bend+S ₁₀ C ₅ N ₄ (20)bend+H ₈ N ₇ C ₃ N ₄ (16)tors+H ₉ N ₇ C ₃ N ₄ (41)tors
v ₂₆	244(3.78/10.6)	260		N ₇ C ₃ N ₂ (14)bend+S ₁₀ C ₅ N ₄ (44)bend+H ₈ N ₇ C ₃ N ₄ (14)tors+H ₉ N ₇ C ₃ N ₄ (13)tors
v ₂₇	149(0.576/2.28)			C ₅ N ₁ N ₂ C ₃ (43)tors+ N ₄ C ₃ N ₂ N ₁ (17)tors + S ₁₀ N ₁ N ₄ C ₅ (26)out

Table S2. 266 nm resonance Raman spectra and calculated vibrational frequencies at B3LYP-D3(BJ)/6-311++G(d, p) level using H₂O PCM model and assignments of thione AMT in solvents

Modes	Calc.(Raman activity /IR activity) ^a	Exp.		Description (PED%)
		CH ₃ OH	H ₂ O	
v ₅	1669(113/681)	1646	1656	N ₂ C ₃ (25)stretch+N ₇ C ₃ (26)stretch+H ₉ N ₇ H ₈ (27)bend
v ₆	1627(31/112)	1619	1615	N ₂ C ₃ (16)stretch+H ₉ N ₇ H ₈ (55)bend
v ₇	1519(104/359)	1524	1528	N ₁ C ₅ (23)stretch+H ₆ N ₁ N ₂ (39)bend+N ₄ C ₅ N ₁ (10)bend
v ₈	1501 (147/387)	1491	1490	N ₂ C ₃ (16)stretch+N ₄ C ₃ (30)stretch+H ₁₁ N ₄ C ₃ (11)bend+H ₈ N ₇ C ₃ (12)bend
v ₉	1363(25/23)	1378	1386	N ₄ C ₅ (21)stretch+H ₆ N ₁ N ₂ (23)bend+H ₁₁ N ₄ C ₃ (33)bend
v ₁₀	1303(6/3)	1307	1319	N ₁ C ₅ (38)stretch+ H ₆ N ₁ N ₂ (18)bend
v ₁₁	1194(117/117)	1210	1220	N ₁ C ₅ (12)stretch+S ₁₀ C ₅ (10)stretch+H ₆ N ₁ N ₂ (11)bend+

				H ₁₁ N ₄ C ₃ (19)bend+ N ₄ C ₅ N ₁ (19)bend
v ₁₂	1165(24/196)			N ₂ C ₃ (15)stretch+N ₄ C ₅ (24)stretch+H ₈ N ₇ C ₃ (19)bend+ N ₄ C ₅ N ₁ (10)bend
v ₁₃	1056(7.8/73)			N ₂ C ₃ N ₄ (10)bend+H ₈ N ₇ C ₃ (36)bend+C ₃ N ₄ C ₅ (29)bend
v ₁₄	1042(59/1)			N ₄ C ₅ (14)stretch+ N ₄ C ₅ N ₁ (41)bend
v ₁₅	1010(11/47)	989	1004	N ₄ C ₃ (52)stretch+ N ₂ C ₃ N ₄ (16)bend
v ₁₆	759(9.7/10)		804	N ₇ C ₃ (29)stretch+N ₂ C ₃ N ₄ (32)bend+C ₃ N ₄ C ₅ (12)bend
v ₁₇	724(2.8/13.8)	754	724	N ₂ C ₃ N ₄ C ₅ (32)tors+N ₇ N ₄ N ₂ C ₃ (49)out
v ₁₈	659(3/43)	643	658	H ₆ N ₁ N ₂ C ₃ (13)tors+C ₃ N ₄ C ₅ N ₁ (39)tors+S ₁₀ N ₁ N ₄ C ₅ (37)out
v ₁₉	574(1/478)			H ₉ N ₇ H ₈ (11)bend+H ₈ N ₇ C ₃ N ₄ (49)tors+H ₉ N ₇ C ₃ N ₄ (26)tors
v ₂₀	541(0.3/260)			H ₆ N ₁ N ₂ C ₃ (17)tors+H ₁₁ N ₄ C ₃ N ₇ (56)tors+S ₁₀ N ₁ N ₄ C ₅ (10)out
v ₂₁	506(20/38)		509	S ₁₀ C ₅ (50)stretch+ C ₃ N ₄ C ₅ (18)bend
v ₂₂	481(1.6/35)			H ₆ N ₁ N ₂ C ₃ (61)tors+ H ₁₁ N ₄ C ₃ N ₇ (26)tors
v ₂₃	448(14.8/13.5)		457	N ₇ C ₃ N ₂ (51)bend+S ₁₀ C ₅ N ₄ (21)bend
v ₂₄	337(1/20)			N ₂ C ₃ N ₄ C ₅ (36)tors+C ₃ N ₄ C ₅ N ₁ (11)tors+N ₇ N ₄ N ₂ C ₃ (32)out
v ₂₅	279(1.6/52)			H ₈ N ₇ C ₃ N ₄ (31)tors+ H ₉ N ₇ C ₃ N ₄ (51)tors
v ₂₆	249(10/2.5)			N ₇ C ₃ N ₂ (17)+ S ₁₀ C ₅ N ₄ (61)bend

Table S3. Experimental and calculated triplet electronic transition energies, corresponding orbitals and oscillator strengths with the electronic transition character for the optimized T₁ state at B3LYP(TD)-D3(BJ)/6-31G* level.

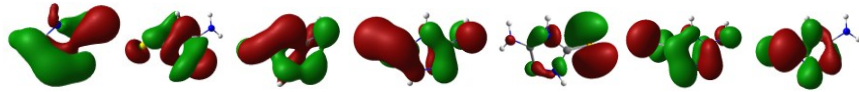
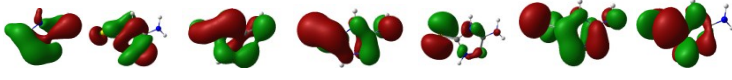
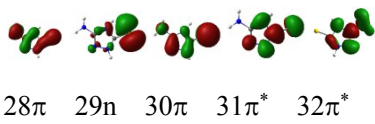
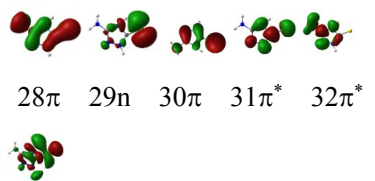
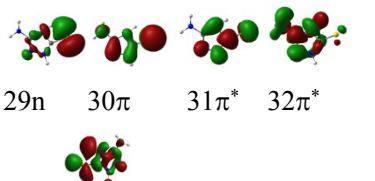
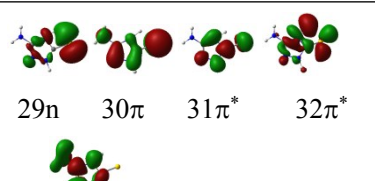
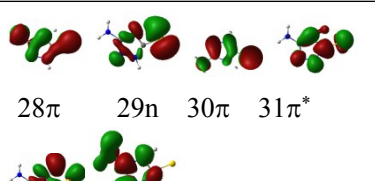

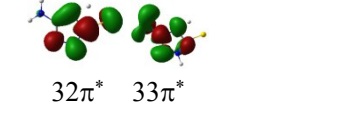
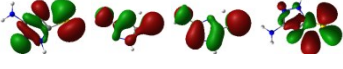
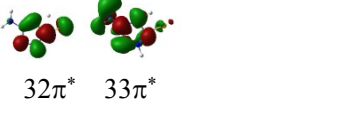


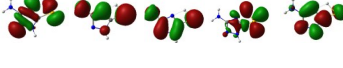

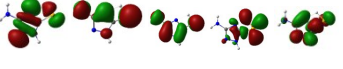

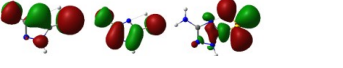

States	Orbitals(coefficient)	Electronic transition	Transition Energy(nm)		Oscillator strength(f)	
			Calc	Expt	Calc	Expt
AMT in CH ₃ CN						
T ₁	29→30(0.99635)	$\pi \rightarrow \pi^*$	1125.57		0.0000	
T₂	28→30(0.97857)	$\pi \rightarrow \pi^*$	565.44	613	0.1494	
T ₃	27→30(0.78374)	$\pi \rightarrow \pi^*$	357.42		0.0028	
T₄	27→30(0.49870)	$\pi \rightarrow \pi^*$	356.35	328.2	0.0238	
T ₅	26→30(0.89785)	$\pi \rightarrow \pi^*$	322.97		0.0063	
T ₆	29→31(0.94498)	$\pi \rightarrow \pi^*$	297.46		0.0064	
T ₇	25→30(0.29788) 28→31(0.32522)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	273.86	295.07	0.0398	
 25π 26π 27π 28π 29π 30π* 31π*						
AMT in H ₂ O						
T ₁	29→30(0.99632)	$\pi \rightarrow \pi^*$	1112.52		0.0000	
T₂	28→30(0.97863)	$\pi \rightarrow \pi^*$	564.82	600	0.1489	
T ₃	27→30(0.87456)	$\pi \rightarrow \pi^*$	357.50		0.0024	
T₄	27→30(0.31013) 28→30(0.15798) 28→31(-0.14434)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	355.57	317.3	0.0243	
T ₅	26→30(0.89856)	$\pi \rightarrow \pi^*$	323.12		0.0062	
T ₆	29→31(0.94623)	$\pi \rightarrow \pi^*$	297.15		0.0064	
T ₇	25→30(0.30532) 28→31(0.33513)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	273.18	292.21	0.0399	
 25π 26π 27π 28π 29π 30π* 31π*						

Table S4. Transition orbital nature and energy (kcal/mol) corresponding to the ground structures in Figure S1

	States	Orbitals(coefficient)	Electronic transition	Transition Energy (kcal/mol)	
-10(thione)	S ₁	29→31(0.70424)	n→π*	133.2	 28π 29n 30π 31π* 32π*
	S ₂	30→31(0.67205) 30→32(0.17648)	π→π* π→π*	141.2	
	S ₃	30→32(0.63838) 28→31(-0.23277)	π→π* π→π*	156.8	
-8	S ₁	29→31(0.70315)	n→π*	141.6	 28π 29n 30π 31π* 32π*
	S ₂	30→31(0.66720) 30→32(0.17646) 29→33(-0.11105)	π→π* π→π* n→π*	147.2	
	S ₃	28→31(-0.19022) 30→31(-0.11702) 30→32(0.50303) 30→33(-0.42931)	π→π* π→π* π→π* π→π*	162.9	
-6	S ₁	29→31(0.68931) 30→31(0.11686)	n→π* π→π*	152.1	 29n 30π 31π* 32π*
	S ₂	30→31(0.64839) 29→31(-0.12649) 29→33(0.15282)	π→π* n→π* n→π*	154.0	
	S ₃	30→32(0.27441) 30→33(0.63972)	π→π* π→π*	162.4	
-4	S ₁	30→32(0.63128) 30→31(0.23986) 29→31(0.16847)	π→π* π→π* n→π*	159.3	 29n 30π 31π* 32π*
	S ₂	30→31(0.59728) 29→32(0.21265) 30→32(-0.24480) 30→33(0.15849)	π→π* n→π* π→π* π→π*	161.1	
	S ₃	29→31(0.68285) 30→32(-0.17108)	n→π* π→π*	164.8	
-2	S ₁	30→32(0.67820) 28→32(0.11368) 30→31(0.15010)	π→π* π→π* π→π*	157.1	 28π 29n 30π 31π*
	S ₂	30→31(0.61272) 29→32(0.25959) 30→32(-0.13736) 30→33(0.14017)	π→π* n→π* π→π* π→π*	166.8	
	S ₃	29→31(0.68760) 29→32(-0.11689)	n→π* n→π*	176.3	

0	S ₁	30 → 31 (0.69063) 29 → 31 (0.14302)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	154.5	
	S ₂	30 → 32 (0.61848) 28 → 31 (0.28304) 29 → 32 (0.11623) 30 → 33 (0.12864)	$\pi \rightarrow \pi^*$ $n \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	170.7	
	S ₃	28 → 32 (0.13753) 29 → 31 (0.67328) 30 → 31 (-0.14716)	$n \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	179.8	
2	S ₁	30 → 31 (0.68505) 29 → 31 (0.16922)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	151.4	
	S ₂	30 → 32 (0.61071) 28 → 31 (0.29529) 29 → 32 (0.12970) 30 → 33 (0.12200)	$\pi \rightarrow \pi^*$ $n \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	172.2	
	S ₃	29 → 31 (0.68087) 30 → 31 (-0.17085)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	174.9	
4	S ₁	30 → 31 (0.67888) 29 → 31 (0.19129)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	148.1	
	S ₂	29 → 31 (0.67540) 30 → 31 (-0.19254)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	170.0	
	S ₃	28 → 31 (0.29526) 29 → 32 (0.12345) 30 → 32 (0.61112) 30 → 33 (0.11399)	$n \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	171.3	
6	S ₁	29 → 31 (0.20580) 30 → 31 (0.67405)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	145.1	
	S ₂	29 → 31 (0.67254) 30 → 31 (-0.20789)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	165.8	
	S ₃	28 → 31 (-0.27753) 30 → 32 (0.62486) 30 → 33 (0.11105)	$n \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \text{Ryd}$	168.8	
8	S ₁	29 → 31 (0.21251) 30 → 31 (0.67141)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	142.7	
	S ₂	29 → 31 (0.67018) 30 → 31 (-0.21527)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	162.6	
	S ₃	28 → 31 (-0.23729) 30 → 32 (0.64449)	$n \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	165.6	
10(thiol)	S ₁	29 → 31 (0.21347) 30 → 31 (0.67073)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	140.6	
	S ₂	29 → 31 (0.66943) 30 → 31 (-0.21673)	$\pi \rightarrow \pi^*$ $\pi \rightarrow \pi^*$	160.4	
	S ₃	28 → 31 (-0.18657) 30 → 32 (0.66030) 30 → 33 (0.10832)	$n \rightarrow \pi^*$ $\pi \rightarrow \pi^*$ $\pi \rightarrow \text{Ryd}$	162.4	

