Supporting information

One pot fabrication of fluorescein functionalized manganese dioxide

for fluorescence "Turn OFF-ON" sensing of hydrogen peroxide in

water and cosmetic samples

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Figure 1S: FT-IR spectra of (a) MnO₂ NS (b) FLS@MnO₂ NS



Figure 2S: EDX analysis of the synthesized FLS@MnO₂ NS.

Table 1S: EDX analysis of elemental composition of synthesized FLS@MnO₂ NS.

Element	C K	O K	Na K	S K	ΚK	Mn K	Total
Weight %	3.12	31.23	1.97	9.41	7.22	47.05	100.00



Figure 3S: Overlay UV-VIS spectra of: unreacted FLS, untreated FLS / KMnO₄ and FLS@MnO₂ NS. Inset: observable color images of (A) unreacted FLS (B) untreated FLS/KMnO₄ (C) FLS@MnO₂ NS





Figure 4S.The effect of (a) pH of B.R. buffer, (b) concentration of FLS@MnO₂ NS (c) incubation time on fluorescence enhancement efficiency of the experimental procedure containing 2 μ M H₂O₂







sample (a), cosmetic cream sample (b) and natural water sample (C) using the proposed nanosystem at the optimum conditions.





Figure 7S: (a) absorption spectra of the freshly synthesized $FLS@MnO_2$ NS (blue) and the one month stored $FLS@MnO_2$ NS (red), inset: photograph images of color change (b) PXRD charts.



Figure 8S: Plot of the fluorescence intensity at 515 nm against the glucose concentration using the proposed nanoprobe with the intended modification.