Electronic supplementary information:

A Physical Approach for the Estimation of SERS Enhancement Factor through

Enrichment, Separation of Target Molecules using Magnetic Absorbents

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Figure S1: The size distribution of the (a) Fe₃O₄@1G NPs; (b) Fe₃O₄@3G NPs; and (c)Fe₃O₄@5G NPs.



Figure S2: High-magnification TEM images.



Figure S3: SAED patterns of the (a) Fe₃O₄@1G NPs; (b) Fe₃O₄@3G NPs; and (c)Fe₃O₄@5G NPs.



Figure S4: EDS spectra of (a) Fe₃O₄@1G NPs; (b) Fe₃O₄@3G NPs; and (c)Fe₃O₄@5G NPs.



Fig. S5 Schematic diagram of the formation of core-shell Fe₃O₄@DGL NPs with different core sizes and graphene layers.



Figure S6: XRD pattern of the product prepared using H₂O/ EG ratio of 60 mL:20 mL.



Figure S7: High-resolution TEM image of the product prepared using H₂O/ EG ratio of 60 mL:20 mL.



Figure S8: The calibration curve of RhB working solution.



Fig. S9 The absorption efficiency of RdB using $Fe_3O_4@1G$ NPs as nano-absorbent

Table S1. The intensities and band assignments in SERS spectra of RdB with concentration of 10⁻⁵

Characteristic bands of RdB	Assignment	Fe₃O₄@1G EF
1645 cm ⁻¹	arom C-C stretching	1.64×10 ⁵
1507 cm ⁻¹	arom C-C stretching	2.2 ×10 ⁵
1355 cm ⁻¹	arom C-C stretching	1.75×10 ⁵
1280 cm ⁻¹	C-H in plane bending	1.31×10 ⁵
1192 cm ⁻¹	C-H in plane bending	1.26×10 ⁵





Fig. S10 The recyclability of 200 mg Fe $_3O_4@1G$ NPs for the adsorption of RdB and removal of *E. Coli*

from water (conditions: Room temperature = 25 \pm 1°C; contact time = 4 hr).