

Supplementary Material

Highly efficient, reversible iodine capture and exceptional uptake of amines in one viologen-based porous organic polymers

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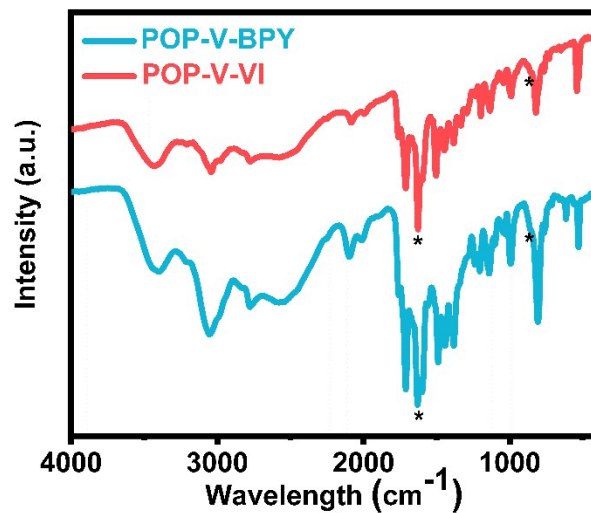


Figure S1. FTIR spectrum of POP-V-VI and POP-V-BPY.

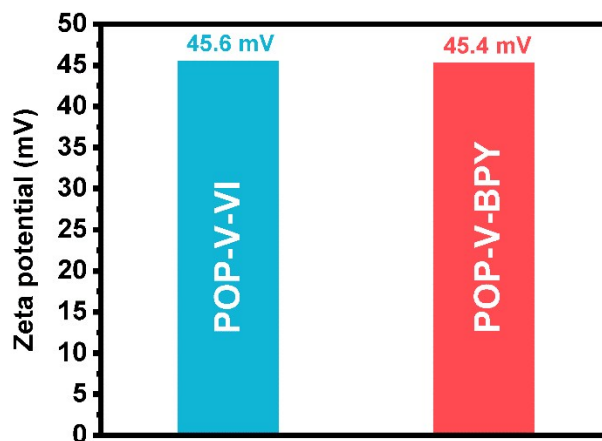


Figure S2. Zeta potential of POP-V-VI and POP-V-BPY.

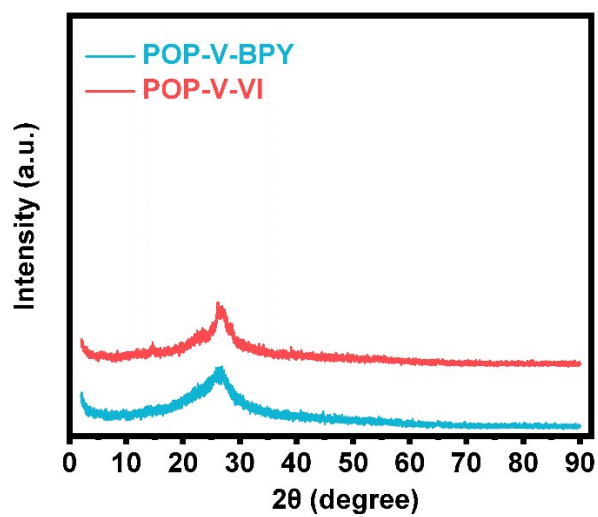


Figure S3. XRD patterns of POP-V-VI and POP-V-BPY.

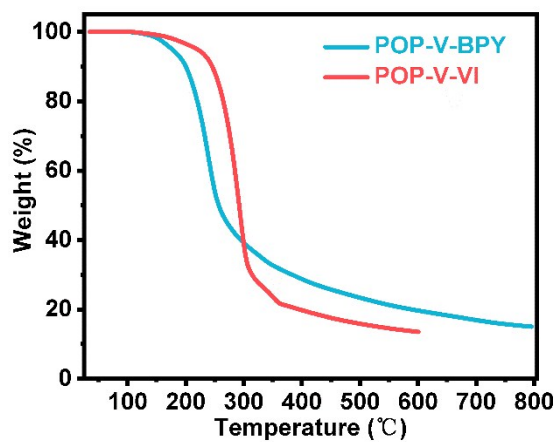


Figure S4. TGA patterns of POP-V-BPY and POP-V-VI under N₂.

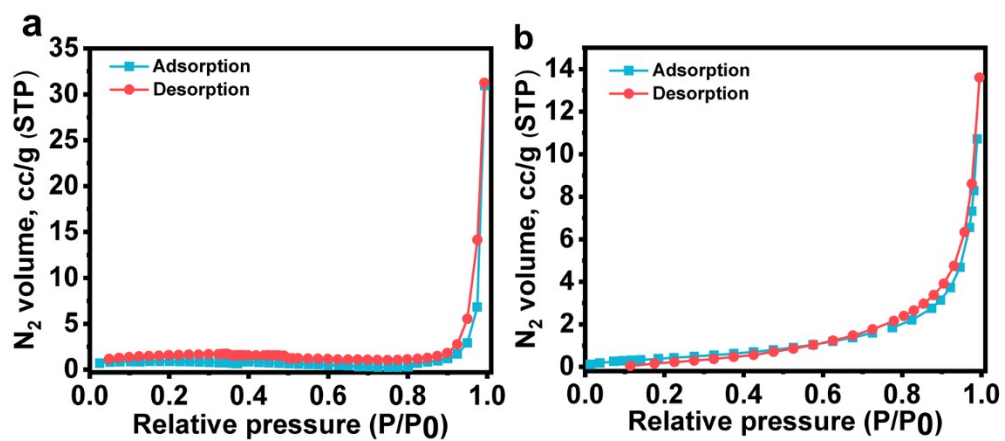


Figure S5. N₂ adsorption-desorption isotherms of (a) POP-V-VI and (b) POP-V-BPY.

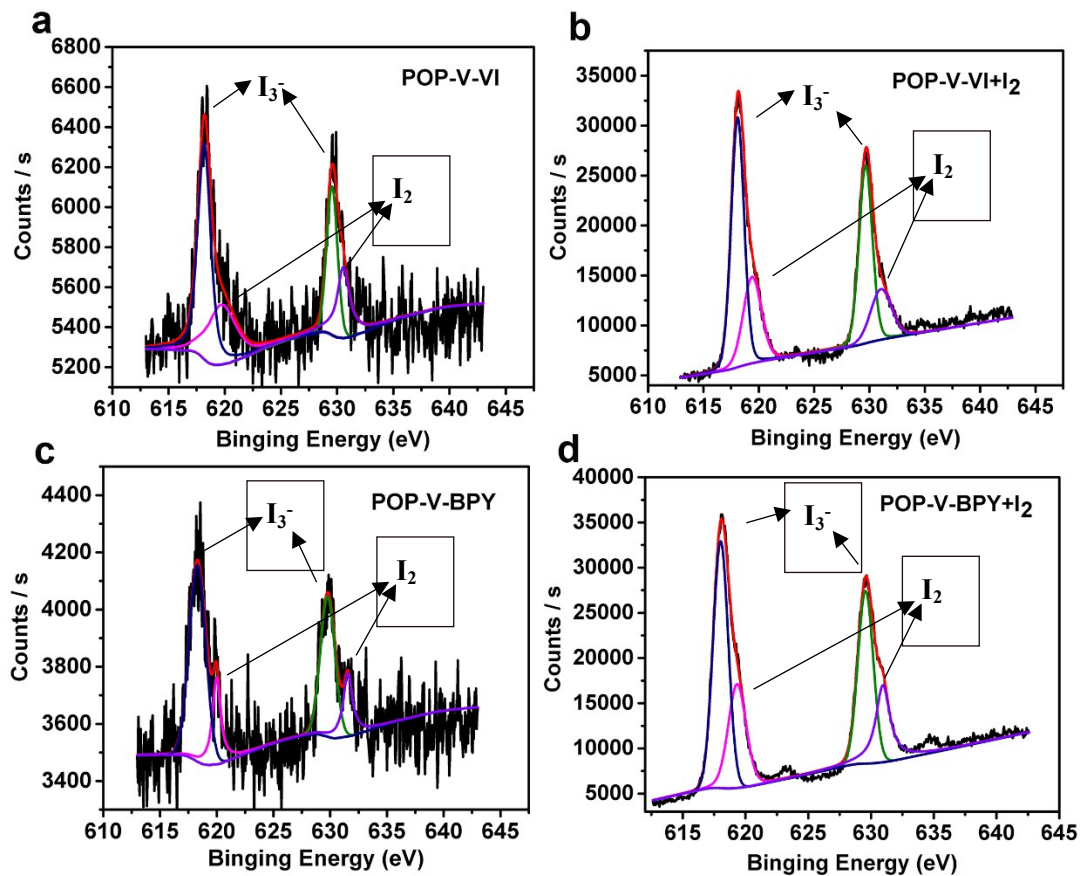


Figure S6. XPS spectrum of (a) POP-V-VI (b) POP-V-VI + I_2 (c) POP-V-BPY and (d) POP-V-BPY + I_2 .

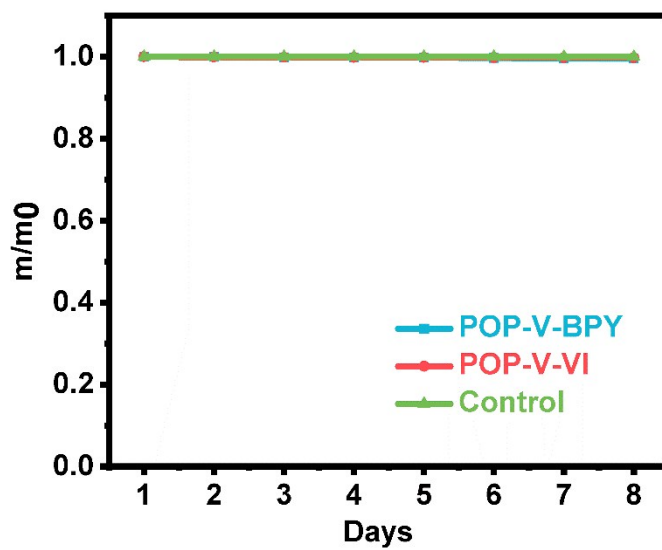


Figure S7. Stability of iodine-loaded POP-V-VI and POP-V-BPY as a function of time at room temperature.

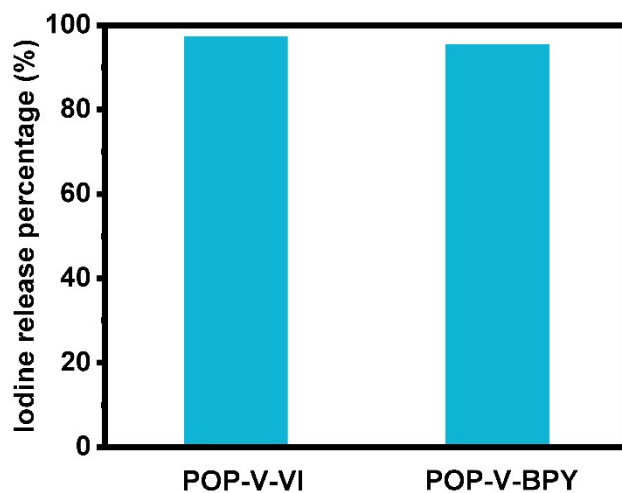


Figure S8. Iodine release from POP-V-VI and POP-V-BPY into ethanol.

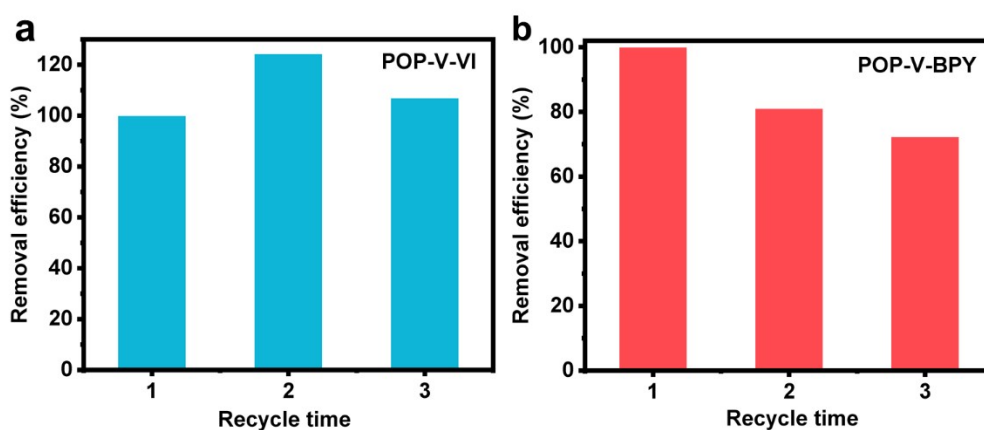


Figure S9. Recyclability of POP-V-VI and POP-V-BPY for the removal of iodine.

Table S1 Parameter comparison between two different adsorption kinetic models for the adsorption of iodine in POP-V-VI and POP-V-BPY.

Kinetic model		Model parameter			
		Sample name	K_1 (min^{-1})	Q_e (cal) (mg g^{-1})	Q_e (exp) (mg g^{-1})
Pseudo-first-order model	POP-V-VI	0.1067	25.82	40.41	0.97249
	POP-V-BPY	0.2187	62.63	79.37	0.60953
Pseudo-second-order model	Sample name	K_2 ($\text{g} \cdot \text{mg}^{-1} \cdot \text{min}^{-1}$)	Q_e (cal) (mg g^{-1})	Q_e (exp) (mg g^{-1})	R^2
	POP-V-VI	0.0108	42.37	40.41	0.99458
	POP-V-BPY	0.0066	84.39	79.37	0.99448

Table S2 Summary of iodine uptake abilities by porous materials.

Number	Adsorbents	Temperature (K)	Pressure (bar)	S_{BET} (m^2/g)	Iodine uptake (g/g)	Materials type	ref
1	CX4-NS	350	1	468	1.14	Calix[4]arene based 2D polymer	ACS.Appl.Mater.Interfaces.2018.10.17359
2	Por-Py-CMP	350	1	1014	1.3	CMP	RSC.Adv.2016.6.75478
3	Pha-HCOP-1	353	1	217.31	1.31	COP	RSC.Adv.2017.7.54407
4	COP ₁ ++	345	1	17.9	1.6	Covalent Organic nanosheets	Chem.Eur.J.2018.24.8648

5	TPTT	350	1	315.5	1.77	POP	Polym.chem.2018.9.777
6	NTP	350	1	1067	1.8	POP	ACS.Macro.Lett, 2016, 5, 1039
7	NOP-54	348	1	1178	2.02	POP	Chemical.Engineering.Journal.334.2018.900
8	NiP-CMP	350	1	2600	2.02	CMP	Chem.Commun.2014.50.8495
9	CMPN-3	343.3	1	1386	2.08	CMP	J.Mater.Chem.A.2015.3.87
10	NCMP-1	358	1	58	2.15	CMP	ACS.Appl.Mater.Interfaces.2017.9.38390-38400
11	APOP	350	1	490	2.2	POP	Materials.Letters.229.2018.240-243
12	SCMP-2	350	1	855	2.22	CMP	ACS.Appl.Mater.Interfaces.2016.8.21063
13	NRPP-2	360	1	1028	2.22	POP	Acs.Appl.Mater.Interfaces.2018.10.16049-16058
14	Azo-Trip	350	1	510.4	2.38	POP	Polym.chem.2016.7.643
15	PAF-24	350	1	136	2.76	PAF	Angew.Chem.Int.Ed.2015.54.12733
16	AzoPPN	350	1	400	2.9	POP	Chem.Eur.J.2016.22.11863
17	HCMP-3	358.15	1	50	3.36	CMP	Macromolecules.2016.49.6322-6333
18	CTF-CTTD	348	1	1684	3.87	POF	Ind.Eng.Chem.Res.2018.57.15114
19	POP-V-BPY	350	1	10	4.2	POP	In this work
20	ADB-S	350	1	41.53	4.43	POP	Chem.Asian.J.2018.13.2046
21	TTPB	350	1	222	4.43	CMP	J.mater.chem. A. 2017, 5.7612
22	TatPOP-2	350	1	36.5	4.5	CMP	Chem.commun.2018.54.8450
23	COF-DL229	350	1	1762	4.7	3D-COF	Chem.Eur.J, 2018,24,585
24	CalPOF-1	350	1	303	4.77	POF	ACS.Sustainable.Chem.Eng.2018.6.17402
25	SIOC-COF-7	350	1	618	4.81	COF	Chem.Commun.2017.53.7266
26	PSIF-1a	350	1	320	4.85	Porous silsesquiox	Acs.Appl.Mater.Interface.2018.10.19964

						ane-imine Framework ks	
27	POP-V- VI	350	1	10	4.86	POP	In this work
28	TPT-BD COF	350	1		5.43	COF	Chem.Mater.2018.30.2299-2308
29	TPB- DMTP COF	350	1	1927	6.26	COF	Adv.Mater.2018.30.1801991