

## Supporting Information

### Unidirectional growth of organic single crystals of naphthalene, anthracene and pyrene by isothermal expansion of supercritical CO<sub>2</sub>

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## 1. Measurement of depressurized CO<sub>2</sub>.

Depressurized CO<sub>2</sub> was measured through pneumatic trough as described in Fig S2. The outlet of crystallizer vent was connected to the inlet shown by “→”. A flat container of 5 L (1) was taken and filled with 2.5 L water. A graduated cylinder (2) (diameter 20 cm, height 10 cm) was completely filled with water and placed inverted into 1. Incoming CO<sub>2</sub> from crystallizer vessel as shown through arrow (→) was passed through the tube (3) that is inserted under water into 2 as shown in Fig S2. Collected CO<sub>2</sub> (4) was measured with respect to time as per the pre-calibrated water level.

## 2. Determination of rate of crystallization

The rate of crystallization was calculated as described below.

Volume of crystallizer  $V_c = 100$  mL

Working Pressure  $P_w = 120$  bar = 118.43 atm

Depressurization pressure  $P_d = 1$  atm

Volume of CO<sub>2</sub> in 1 atm,  $V_{atm}$  was obtained from eq 1.

Depressurization rate (mL/min) =  $R_d$

Time required for complete depressurization (t) in sec was obtained from eq 2.

Crystal length measured in mm = l

Crystal length in meter  $l_c = \frac{l}{1000}$

Growth rate R in m.s-1 was obtained from eq 3.

$$V_{atm} = \frac{P_w V_c}{P_d} \quad (1)$$

$$t = \frac{V_{atm}}{R_d} \times 60 \quad (2)$$

$$R = \frac{l_c}{t} \quad (3)$$

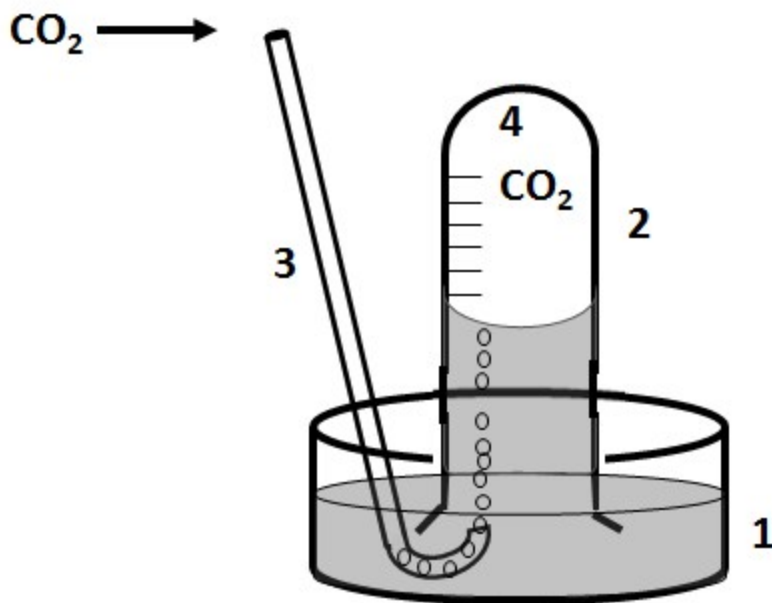
For example, Growth rate of anthracene ((Table 2, in present work) in  $\text{m.s}^{-1}$

$$R_{anthracene} = \frac{l_c}{t} = \frac{l_c R_d}{V_{atm} \cdot 60} = \frac{l_c R_d P_d}{P_w V_c \cdot 60000} = \frac{2 \times 10 \times 1}{118.43 \times 100 \times 60000} = 2.8 \times 10^{-8}$$

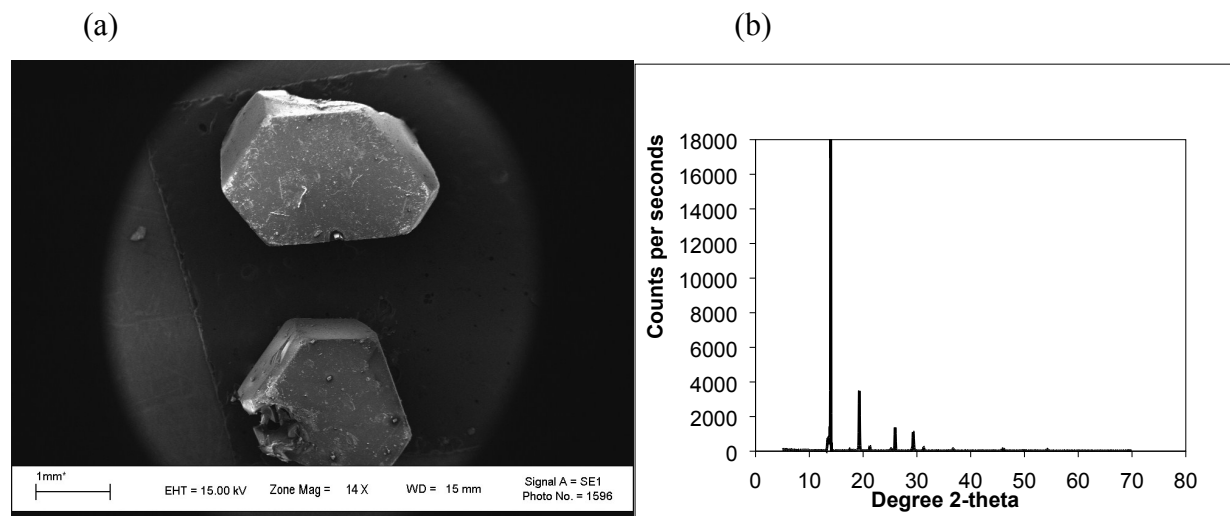
Table S1: Recycle test of anthracene crystallization.<sup>a)</sup>

Recycle Nos.	$W_0$ (g)	$W_c$ (g)	$W_p$ (g)	$W_a$ (g)	% Yield [( $W_c/W_0$ ) x 100]
1	2	1.35	0.52	1.48	67.09
2	2	1.30	0.54	1.46	65.0
3	2	1.33	0.43	--	66.5

a)  $\text{CO}_2$  120 bar; temperature 45 °C; depressurization 10 mL per min.  $W_0$  weight taken;  $W_c$  weight of crystals obtained (g)  $W_p$  weight of powder after cycle.



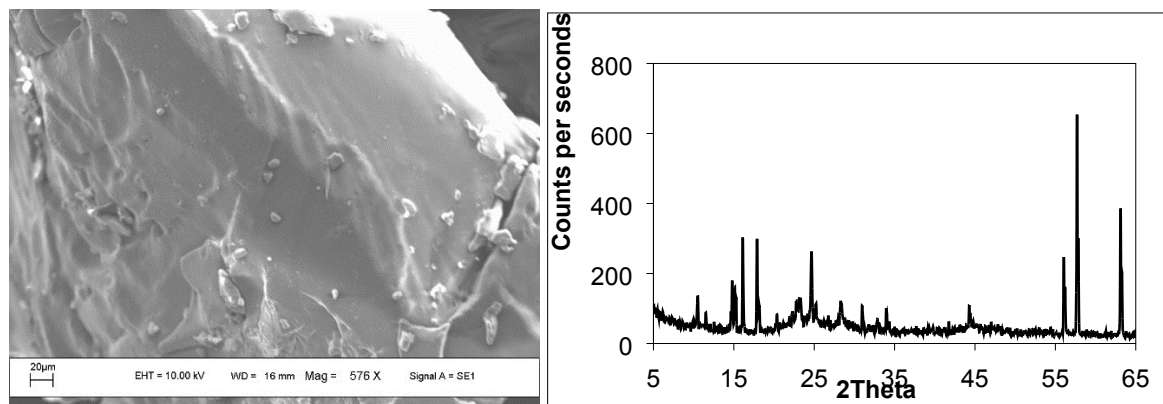
**Fig S1.** Pneumatic trough gas collection for measurement of depressurized CO<sub>2</sub> gas collected to crystallizer vent.



**Fig. S2.** a) SEM Image, b) XRD of naphthalene crystal. CO<sub>2</sub> 120 bar, temp. 45°C, depressurization 10 mL per min, FWHM for 2θ 13.96° [001] 0.0856.

(a)

(b)



**Fig. S3.** a) SEM Image, b) XRD of anthracene crystal. CO<sub>2</sub> 120 bar, temp. 45°C, depressurization rate 50 mL per min.