

Supplementary Data

The Role of pH on the corrosion inhibition of tin using proline amino acid: Theoretical and experimental investigations

Brahim EL IBRAHIMI*^a, Lahcen BAZZI^b, Souad EL ISSAMI^a

^a Applied Chemistry-Physic Team, Faculty of Sciences, University of Ibn Zohr, P.O. Box 8106, Cité Dakhla, Agadir, Morocco.

^b Industrial & Logistic Laboratory, Higher School of Management, Telecommunications and Computer Science, SUP MTI, Rabat, Morocco.

***Corresponding author**

E-mail address: brahimhm@gmail.com - Phone number: +212672254020

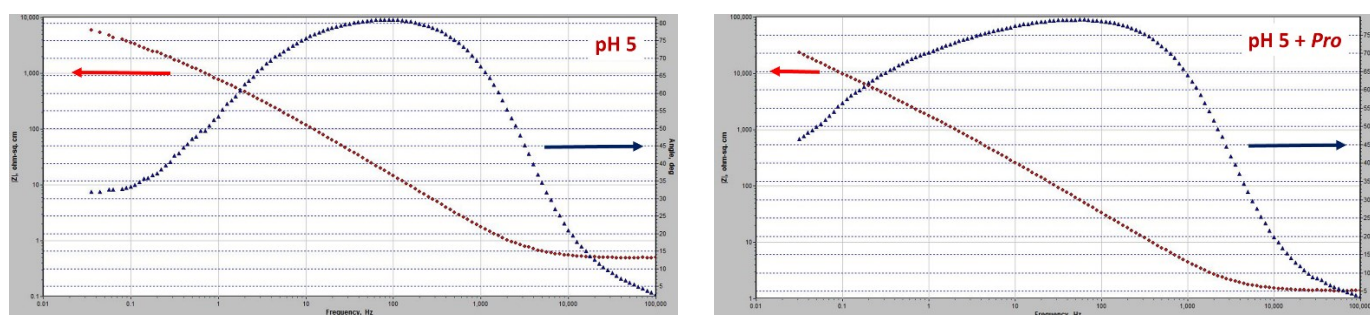


Fig. S1. Bode diagrams of tin immersed in 2% NaCl solution at pH 5 without and with *Pro*.

Table S1. EIS-derived parameters of tin in 2% NaCl solution at pH 2 without and with *Pro*

Medium	R_s (ohm cm ²)	Q ($\mu\text{S s}^n \text{cm}^{-2}$)	n	W ($\mu\text{S s}^5 \text{cm}^{-2}$) 10^{-6}
pH 2	0.355	0.733	0.4452	0.957
pH 2 + <i>Pro</i>	0.427	0.837	0.4107	5.744