## **Supporting Information**

## Enhanced stability and ultrahigh activity of amorphous ripple nanostructured Ni-doped Fe oxyhydroxide electrode toward synergetic electrocatalytic water splitting Selvam Mathi and Jayaraman Jayabharathi \*

Department of Chemistry, Material Science Lab, Annamalai University, Annamalainagar, Tamilnadu- 608 002, India: Email- jtchalam2005@yahoo.co.in Figure S1. FESEM –EDS image of Ni@Fe-NP. A) FESEM; B) Fe, C) O; D) Ni; E) overlap of Fe, Ni and OEDS signals and F) EDS spectrum.







FigureS3. A) EDS-HAADF elemental image; B) EDS line scanning analysis spectra and C) Elemental analysis spectra of Ni@Fe-NP.



Figure S4.A) Chronopotentiometric durability test at current density 10 mA cm<sup>-2</sup> for 130 h; B) Morphology of the catalyst after long-term stability test.



Figure S5. Solar panel current-voltage (I-V) curve under the light illumination (1 sum).





Figure S6. Nyquist plots for different catalysts prepared at different ratio of Ni doped FeOH-NP.

Figure S7 Digital image of conventional water electrolyzer during full cell water electrolysis Ni<sub>3</sub>@Fe<sub>7</sub>-NP/Ni F anode and 20%Pt/C/Ni F cathode.



Catalyst	Support	Electrolyte Cell	Cell Voltage @ 10 mA cm <sup>-2</sup> [V]	Durability @ 10 mA cm-2 [h]	Ref
Ni@Fe-NP	GC	0.1 м КОН	1.59	18	This work
	Ni foam	1 M KOH	1.49	150	THIS WOLK
Core-shell Co@NC-600	GC Ni foam	0.1 M КОН 1 м КОН	1.60 1.57	16 350	Adv. Energy Mater. 2018, 1702838
NiCo <sub>2</sub> O <sub>4</sub>	Ni foam	1 м КОН	1.65	15	Angew. Chem. Int. Ed. 2016, 55, 6290.
CoFeLDH-F	Ni foam	1 м КОН	1.63		ACS Appl. Mater. Interfaces 2016, 8, 34474
NiFe/NiCo <sub>2</sub> O <sub>4</sub>	Ni foam	1 м КОН	1.67	10	Adv. Funct. Mater. 2016, 26, 3515–3523
Fe-P nanotube	Carbon cloth	1 M KOH	1.69	14	Chem. Eur. J. 2015, 21, 18062
Co <sub>0.85</sub> Se/NiFe- LDH	Exfoliated graphene foil	1 м КОН	1.67	10	Energy Environ. Sci. 2016, 9, 478.

Table	S1.Overall	water	splitting	cell	voltage	and	durability	comparison	with	recently
reported works.										