

Supplementary Information

MOF-derived Manganese Oxide/Carbon Nanocomposites with Raised Capacitance for Stable Asymmetric Supercapacitor

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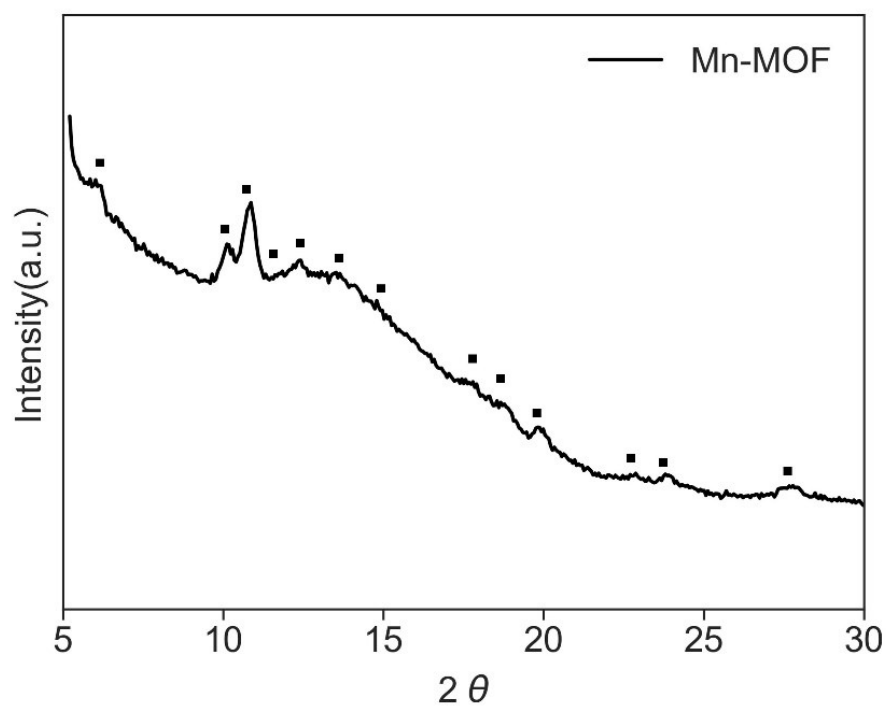


Figure S1. The XRD of as synthesized Mn-MIL-100.

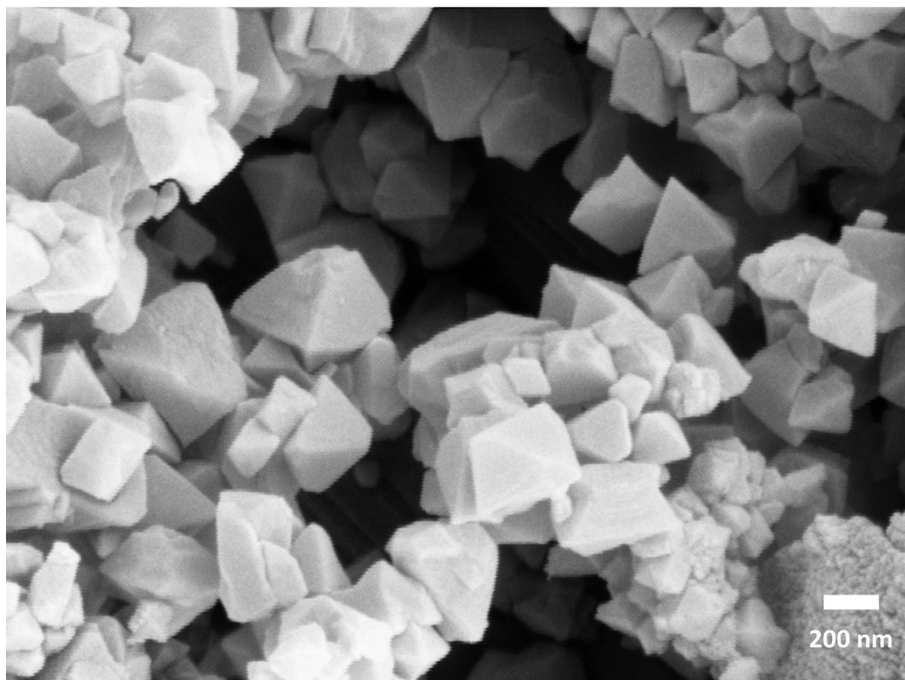


Figure S2. The SEM image of Mn-MIL-100 MOF.

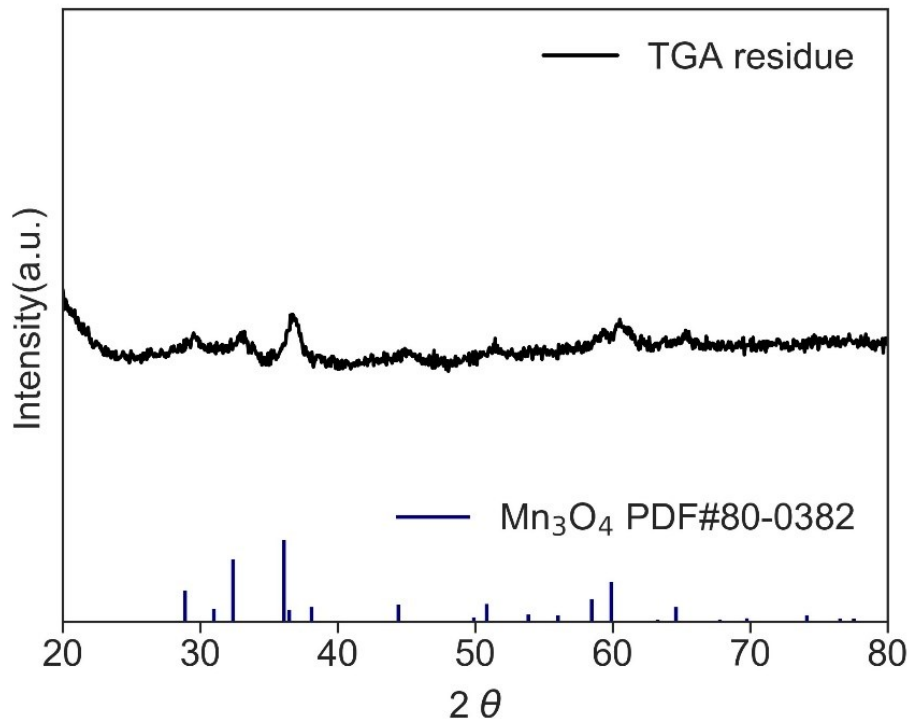


Figure S3. The residue of TGA on MnO@C XRD of is found to be Mn₃O₄.

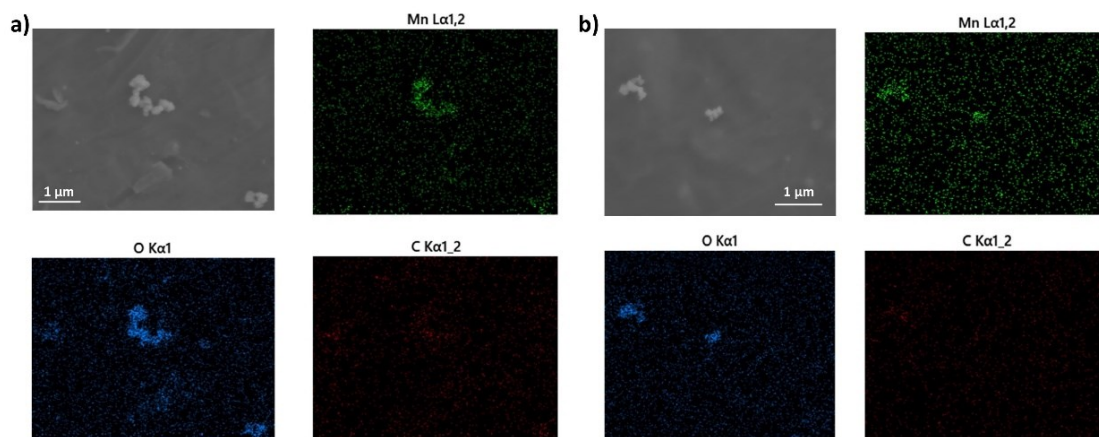


Figure S4. EDS mapping of (a) Mn_3O_4 -300 and (b) Mn_2O_3 -400.

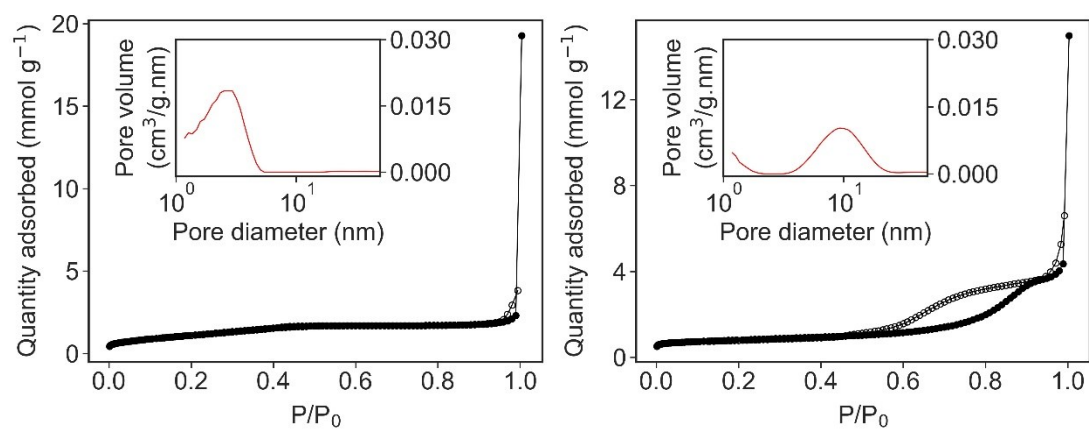


Figure S5. The N₂ isotherm and DFT pore distribution of (a) Mn₃O₄-300 and (b) Mn₂O₃-400, respectively.

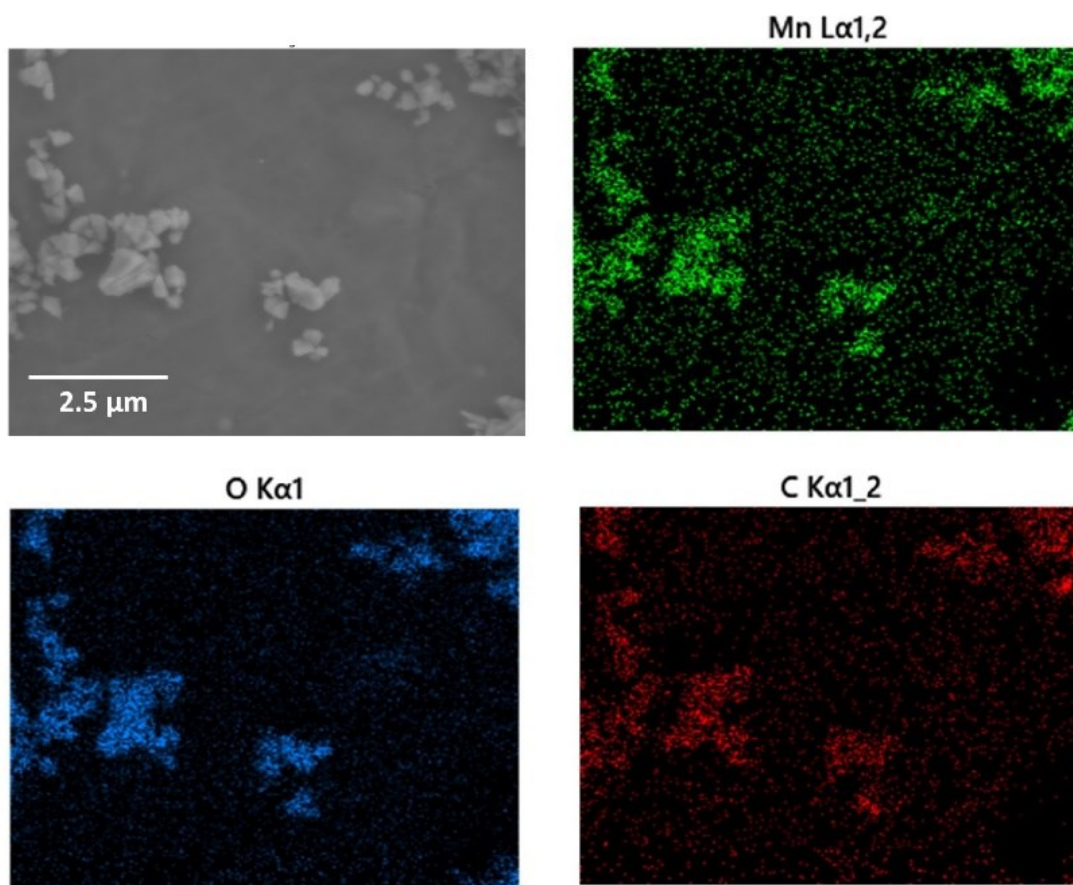


Figure S6. EDS mapping of Mn₃O₄@C-5h.

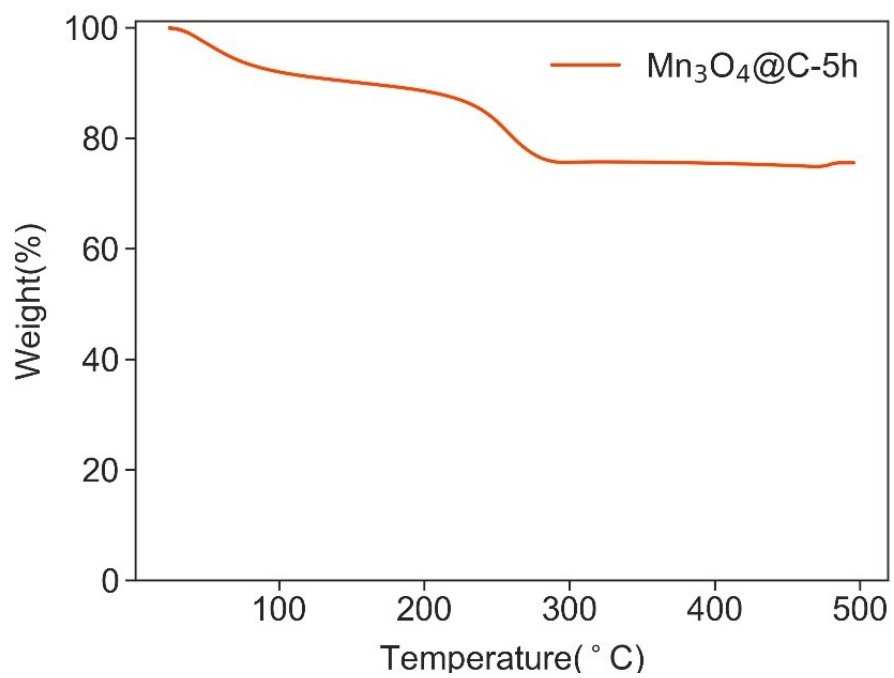


Figure S7. The TGA result of Mn₃O₄@C-5h.

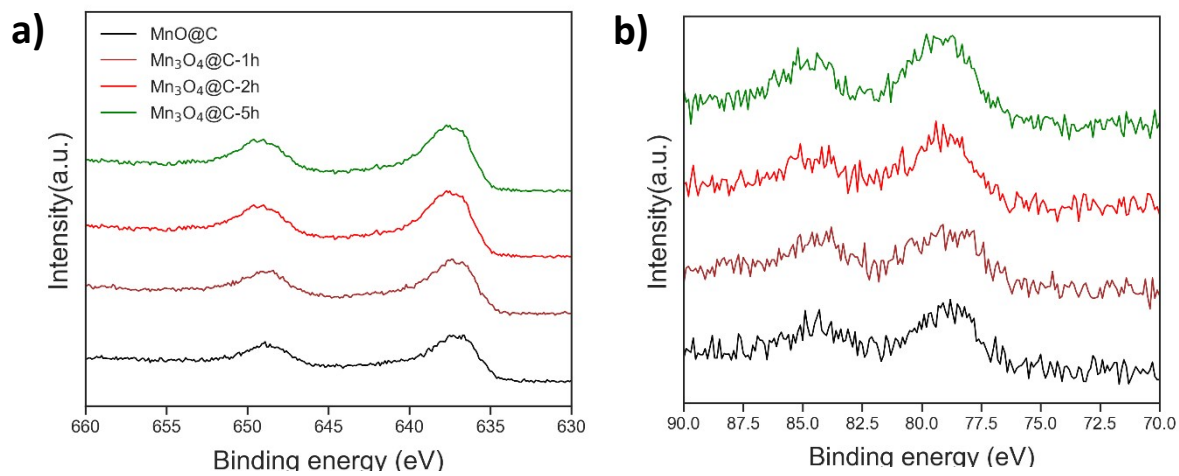


Figure S8. XPS spectroscopy for a) Mn 2p, and b) Mn 3s.

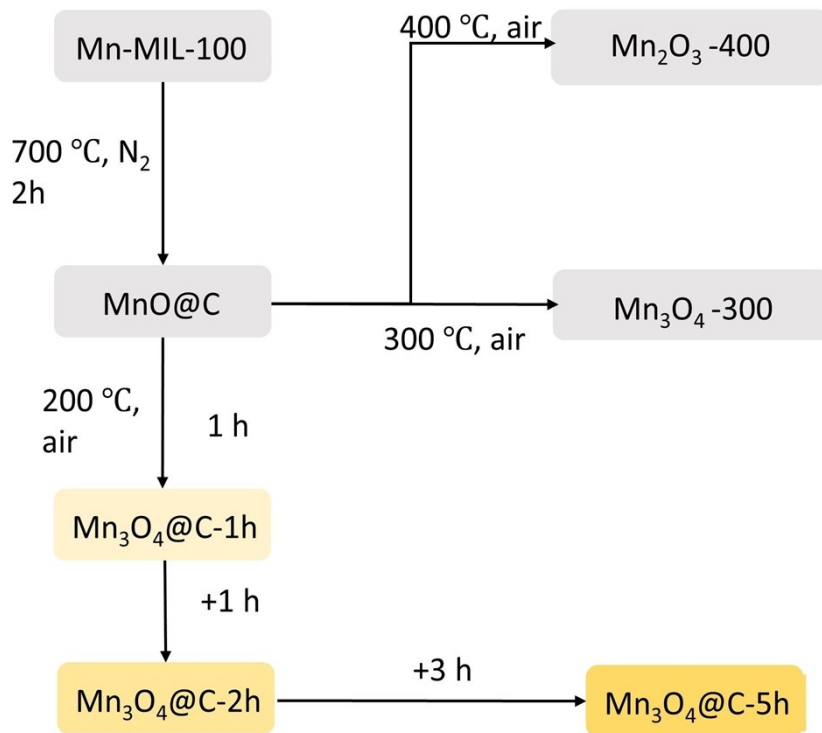


Figure S9. A summary on the thermolysis processes and their respective products involved in this chapter.

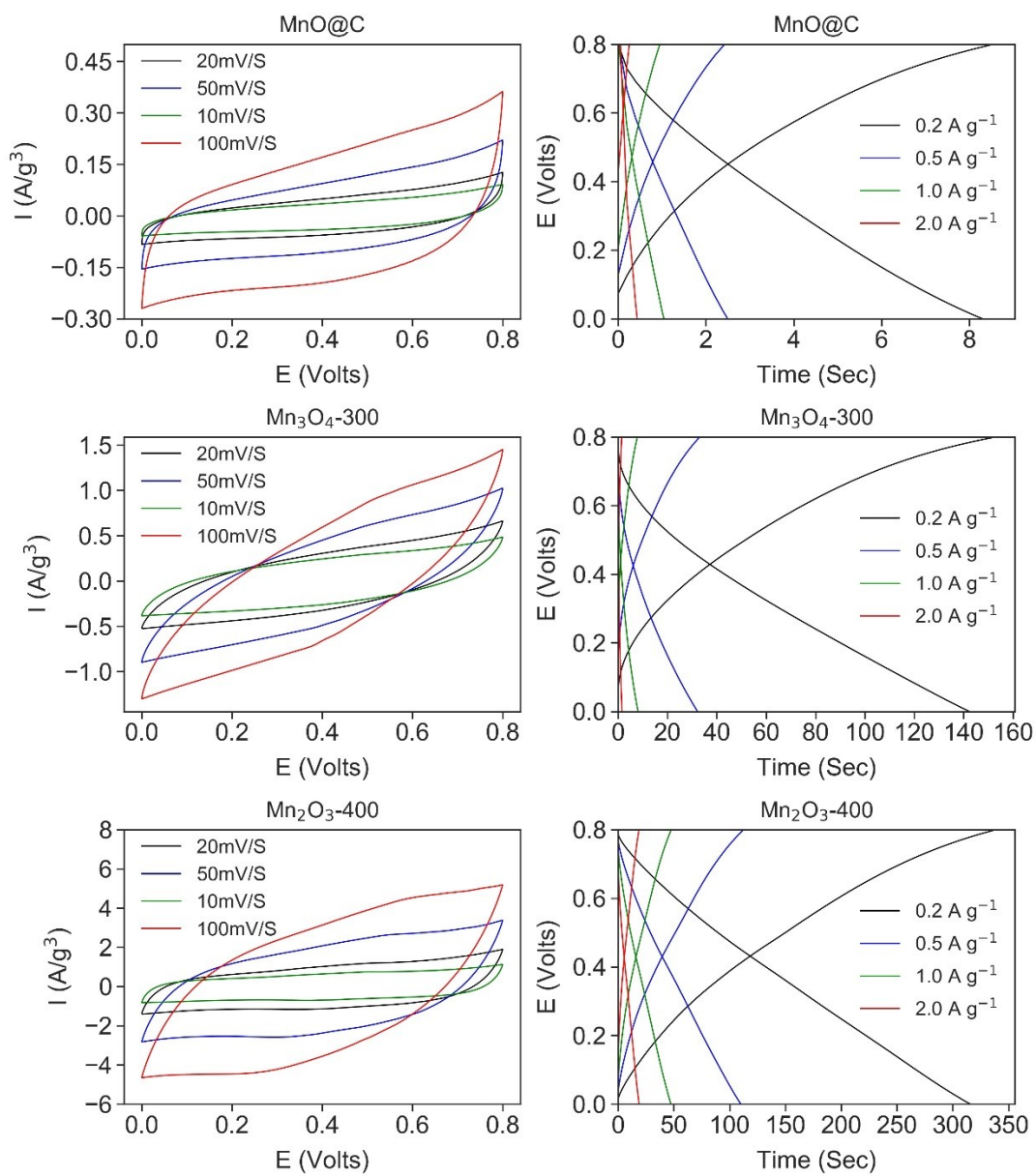


Figure S10. The CV and GCD patterns of MnO@C, Mn₃O₄-300 and Mn₂O₃-400, respectively.

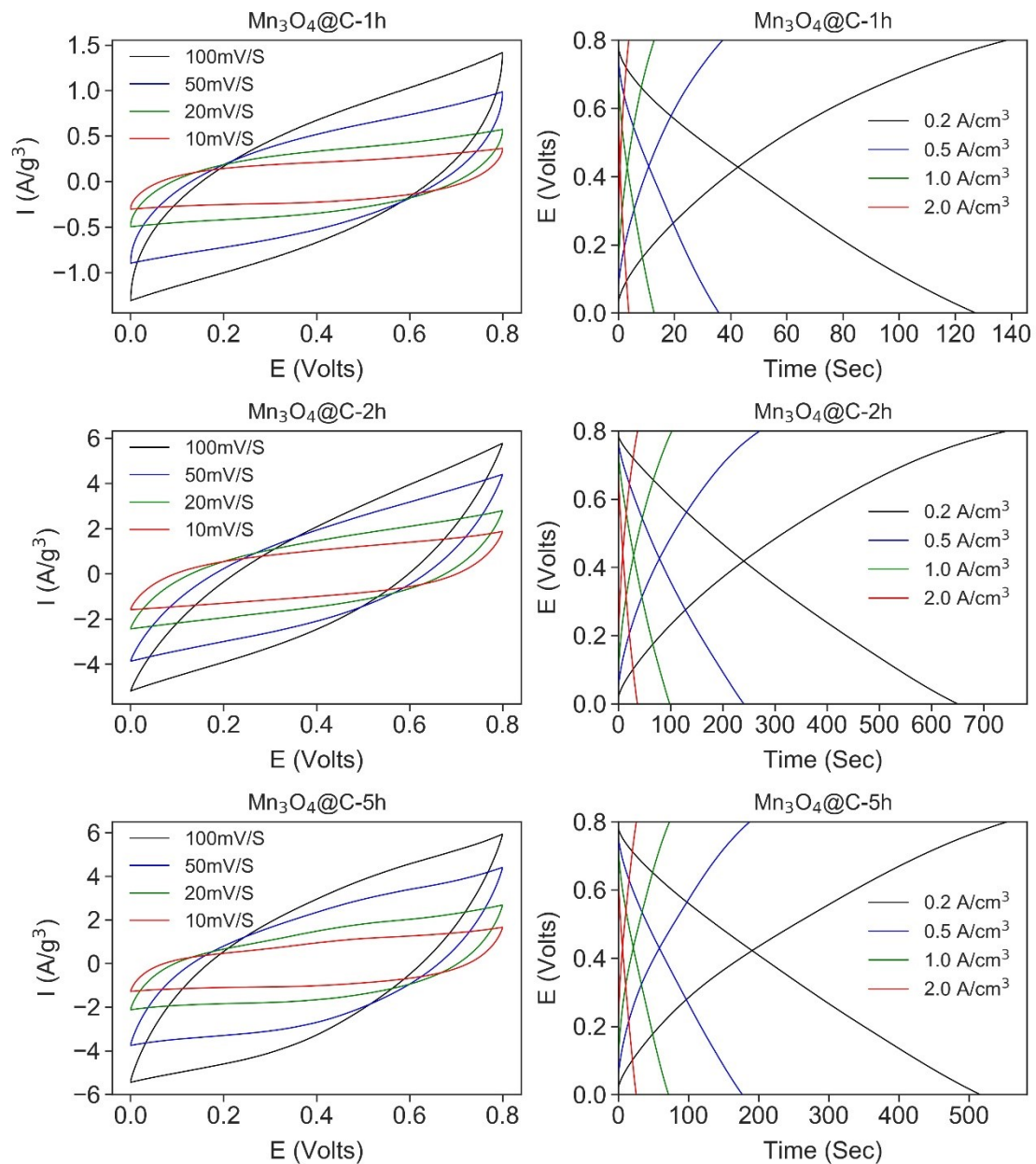


Figure S11. The CV and GCD patterns of $\text{Mn}_3\text{O}_4@\text{C}$ -1h, $\text{Mn}_3\text{O}_4@\text{C}$ -2h and $\text{Mn}_3\text{O}_4@\text{C}$ -5h, respectively.

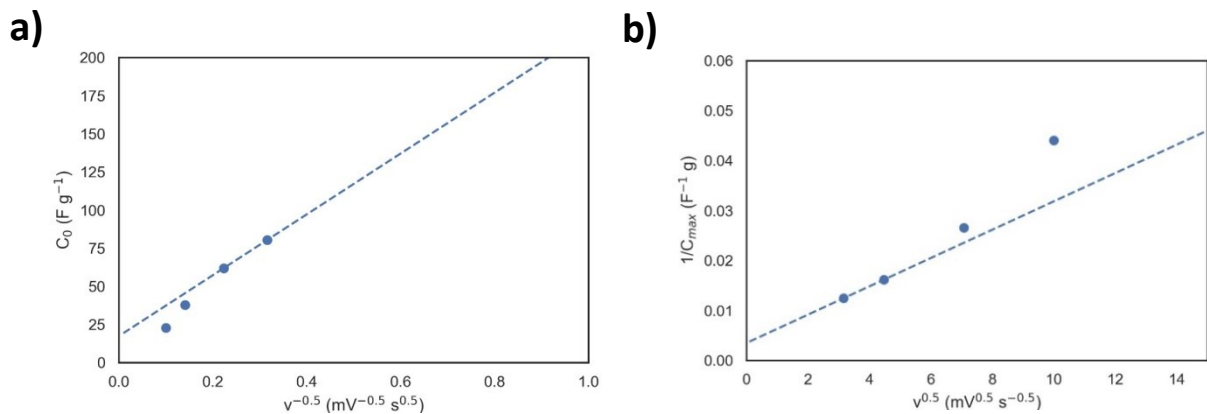


Figure S12. (a) double-layer capacitance C_0 , and (b) maximum capacitance C_{max} of $\text{Mn}_3\text{O}_4@\text{C}-2\text{h}$.

The double-layer capacitance C_0 and maximum capacitance C_{max} of $\text{Mn}_3\text{O}_4@\text{C}-2\text{h}$ were estimated to be 17.3 F g^{-1} and 285.7 F g^{-1} respectively, by the Trassatti's method. The relative contribution of double-layer capacitance was thus within the region of $\sim 10 \%$ (6.1%). It is worth nothing that we only selected the two low frequency (i.e., $v = 10 \text{ mV/s}$ and 20 mV/s) data points from the available data due to the obvious deviation from linear relationship when the scan rate increases as a result of the internal resistance.

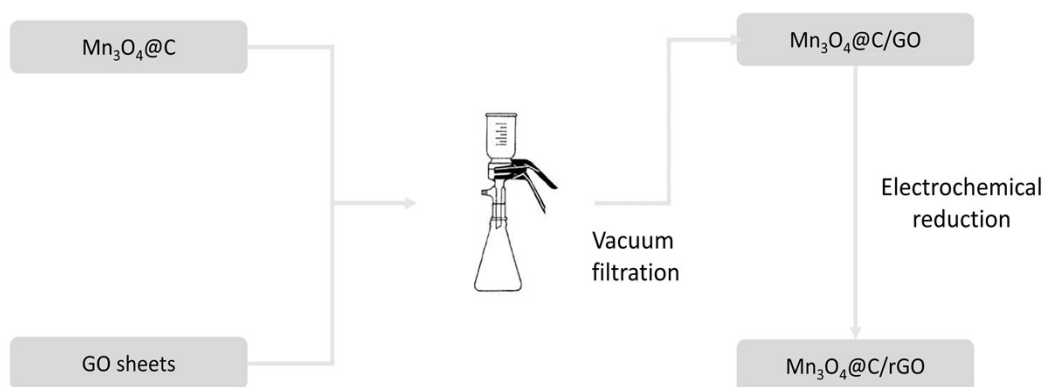


Figure S13. Schematic illustrating the manufacturing process of $\text{Mn}_3\text{O}_4@\text{C}/\text{rGO}$ electrode.

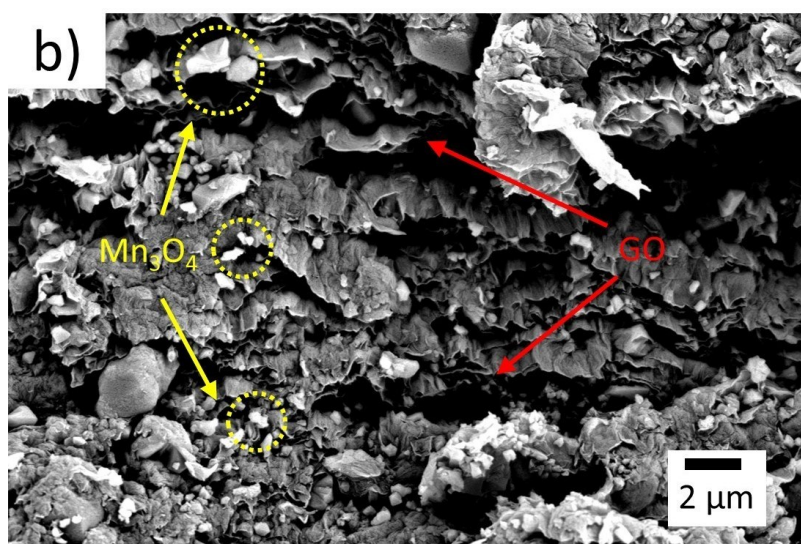


Figure S14. The cross-section SEM of Mn₃O₄@C/GO.



Figure S15. The demonstration of the flexibility of $\text{Mn}_3\text{O}_4@\text{C}/\text{GO}$.

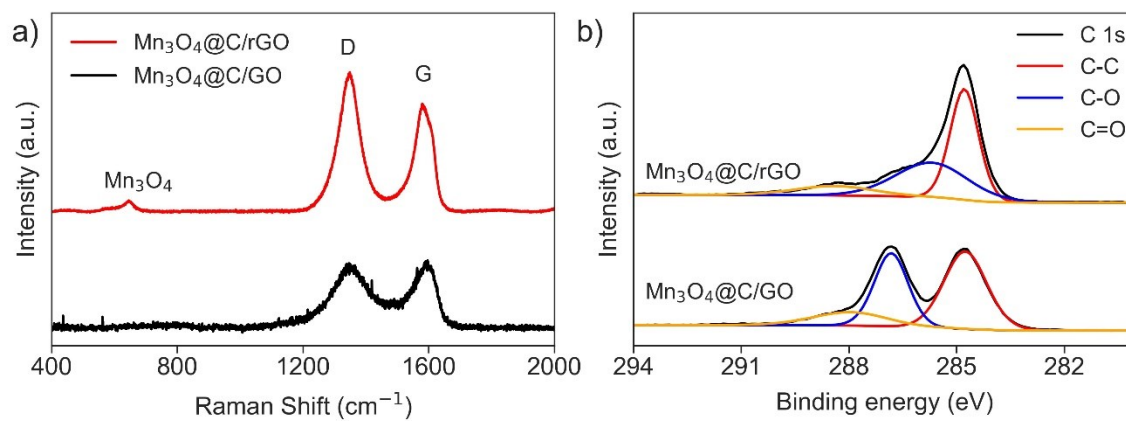


Figure S16. (a) Raman and (b) XPS analysis results of Mn₃O₄@C/GO and Mn₃O₄@C/rGO.

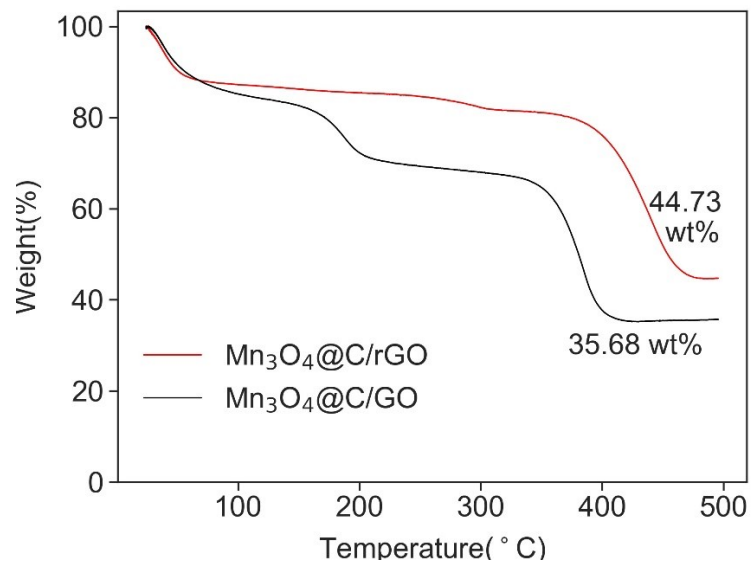


Figure S17. The TGA results of Mn₃O₄@C/GO and Mn₃O₄@C/rGO, respectively.

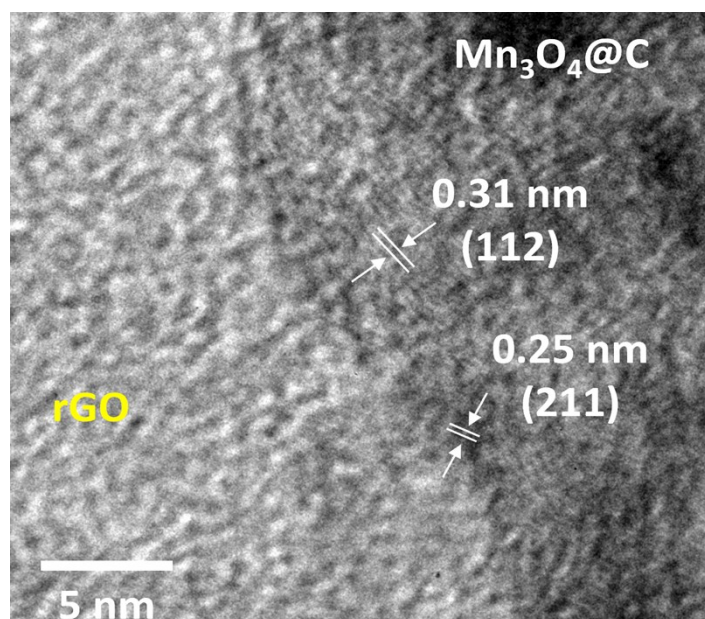


Figure S18. TEM of the Mn₃O₄@C/rGO.

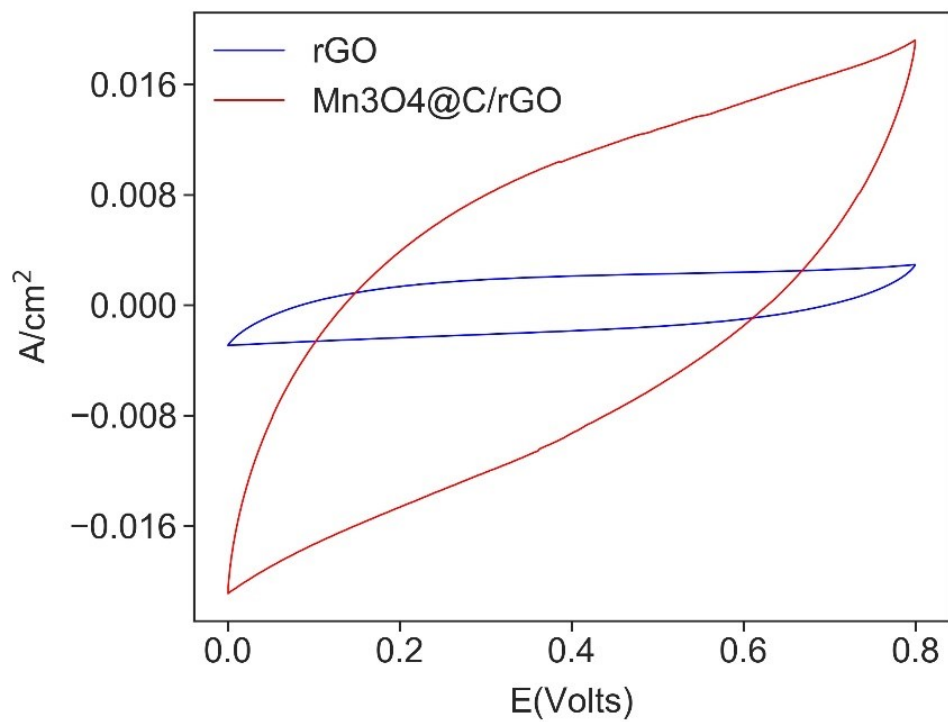


Figure S19. The CV pattern of Mn₃O₄@C/rGO and rGO.

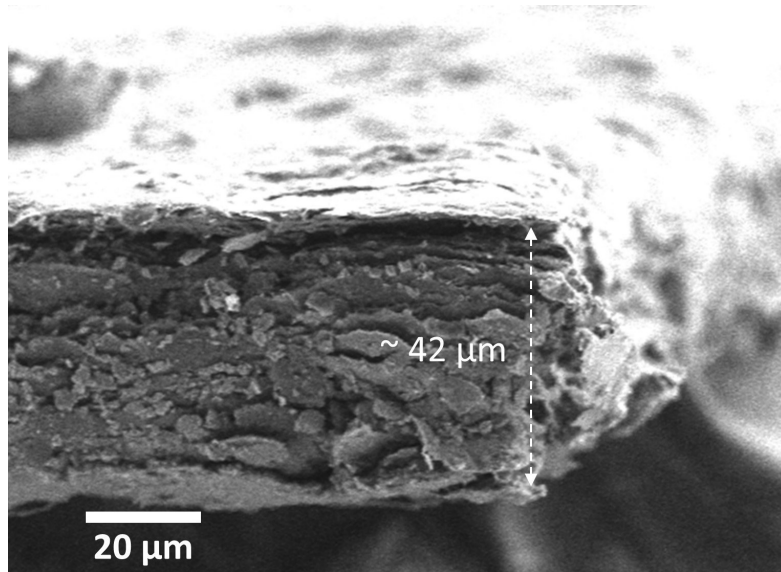


Figure S20. The cross-section SEM of CNT/rGO.

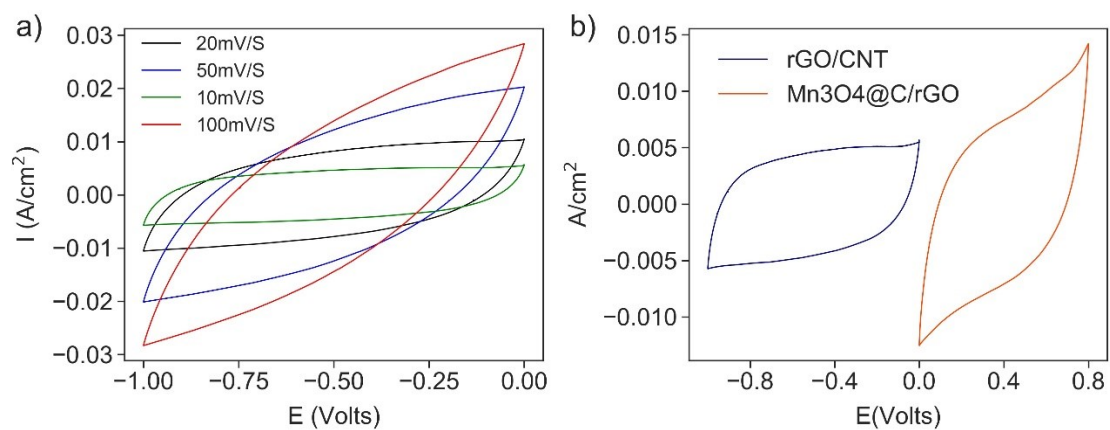


Figure S21. (a) CV patterns of the CNT/rGO paper electrode. (b) CV curves of CNT/rGO and Mn₃O₄@C/rGO, respectively, at 10 mV/s.