

Supporting Information  
**Hyper Oxygen Incorporation in CeF<sub>3</sub>: A New Intermediate-Band  
Photocatalyst for Antibiotic Degradation under Visible/NIR Light**

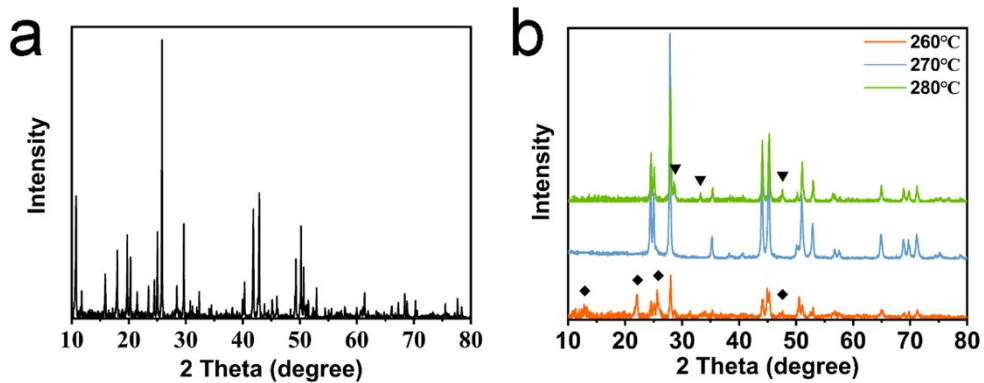
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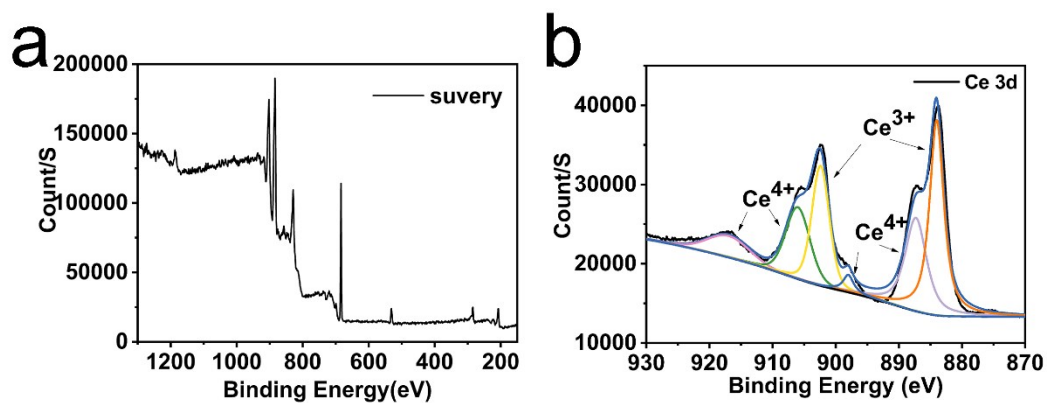
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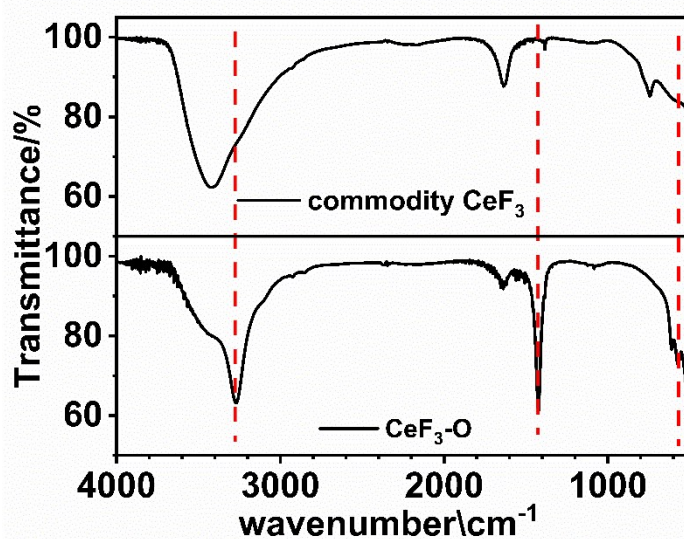
**Figure S1.** the typical XRD patterns of the precursor JCPDS NO.70-4378(a); samples calcined at different temperature(b). ◆ sign is the precursor ; ▼ sign is CeO<sub>2</sub> (JCPDS NO. 08-0045).

**Table S1.** The element content of CeF<sub>3</sub>-O

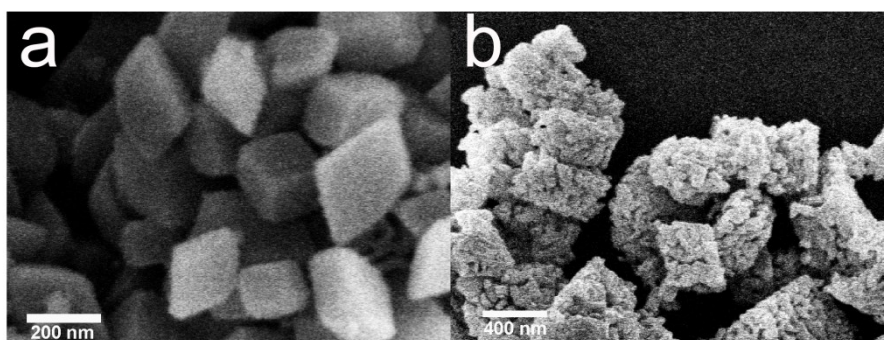
Element	Ce	F	O
Atomic %	21.45	62.82	15.73



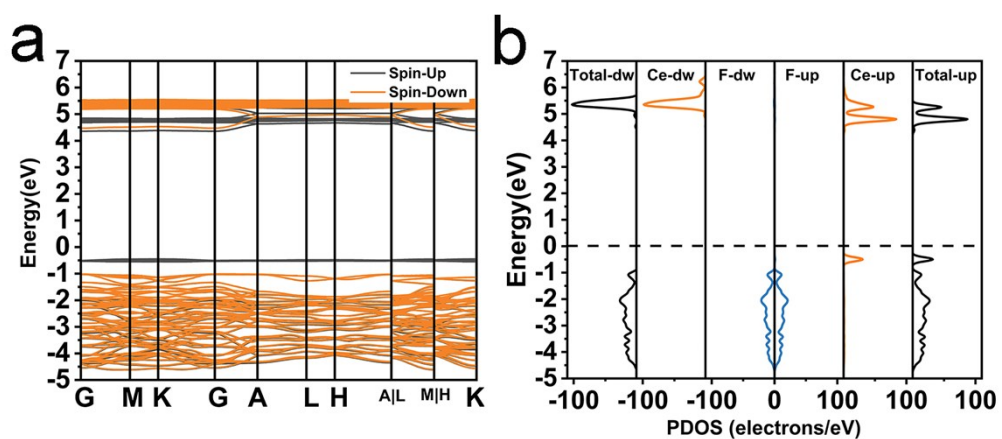
**Figure S2.** The survey spectrum of CeF<sub>3</sub>-O XPS: Ce 3d<sub>5</sub> (884.1 eV), F 1s (684.4 eV), O 1s (531.7 eV), Ce Auger (830.0 eV), and Ce 3d<sub>3</sub> (902.4 eV) (a); The spectrum of Ce 3d (b).



**Figure S3.** FTIR spectra of commercial  $\text{CeF}_3$  and  $\text{CeF}_3\text{-O}$ . The peak at  $750\text{ cm}^{-1}$  of commercial  $\text{CeF}_3$  is attributed to  $\text{-CH}_2\text{-}$ , and it maybe from organ impurity.



**Figure S4.** SEM of the precursor  $\text{H}_{25.5}(\text{NH}_4)_{10.5}\text{Ce}_9\text{O}_{27}\text{F}_{18}$  (a) and  $\text{CeF}_3\text{-O}$ (b)



**Figure S5.** The calculated band structure and DOS of pristine  $\text{CeF}_3$ .

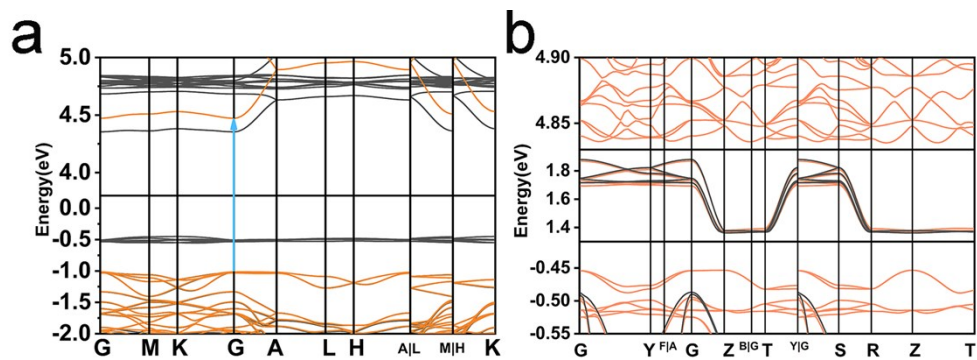


Figure S6. partial enlargement of DFT calculated band structure of pristine  $\text{CeF}_3$  and  $\text{CeF}_3\text{-O}$ .

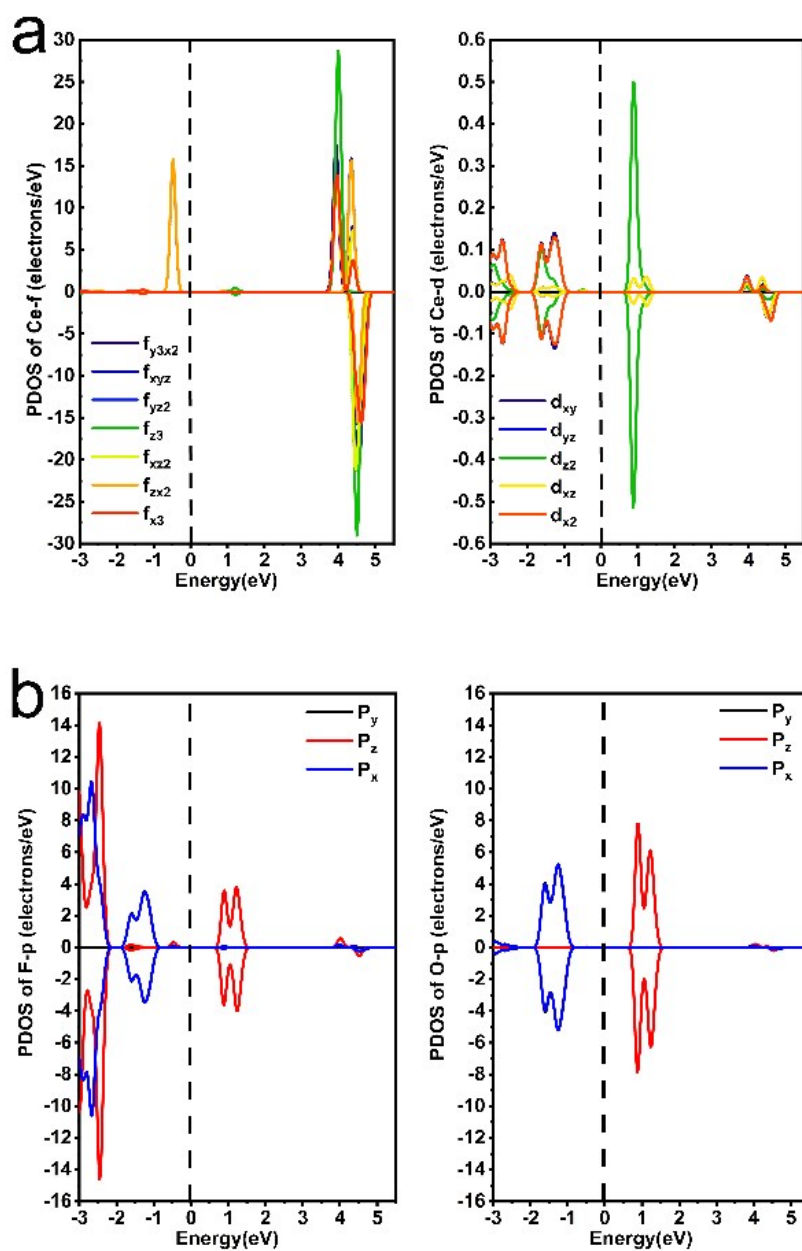
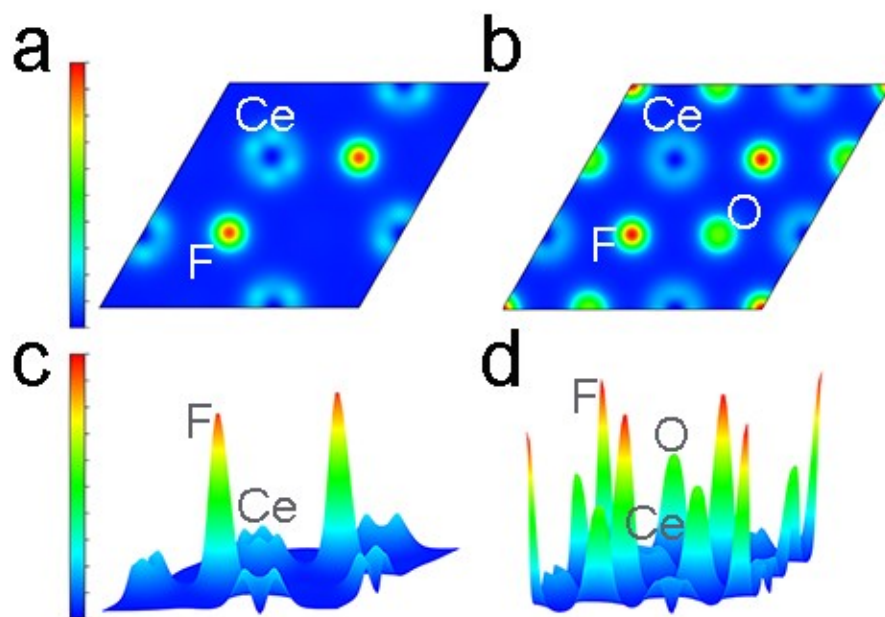
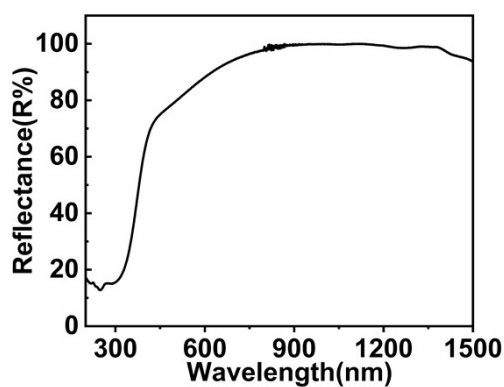


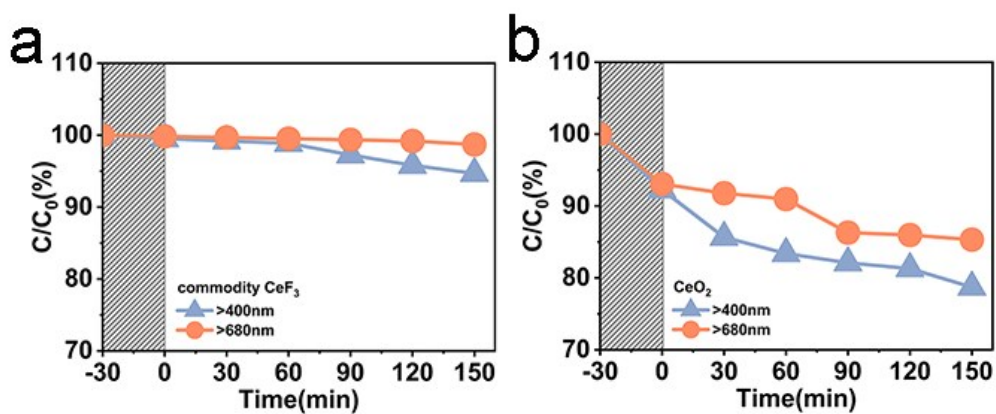
Figure S7. PDOS of Ce(a), O and F(b).



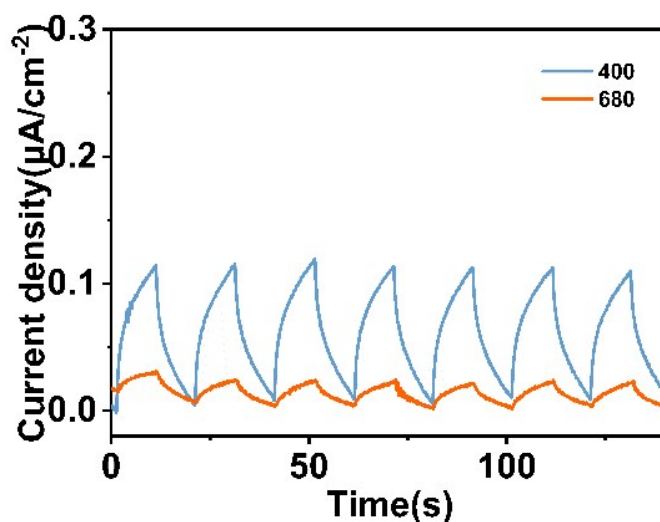
**Figure S8.** Two dimensional charge density distribution plots of pristine  $\text{CeF}_3$  (a,c) and  $\text{CeF}_3\text{-O}$  (b,d) on [001] plane.



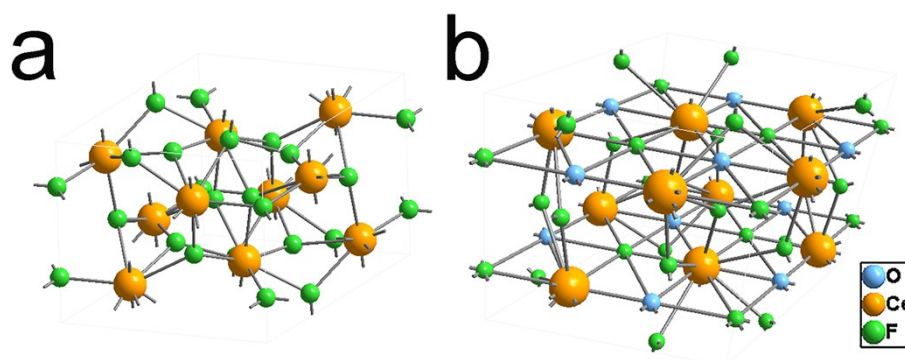
**Figure S9.** UV-Vis-NIR diffuse reflectance absorption (DRS) spectrum of  $\text{CeF}_3\text{-O}$ .



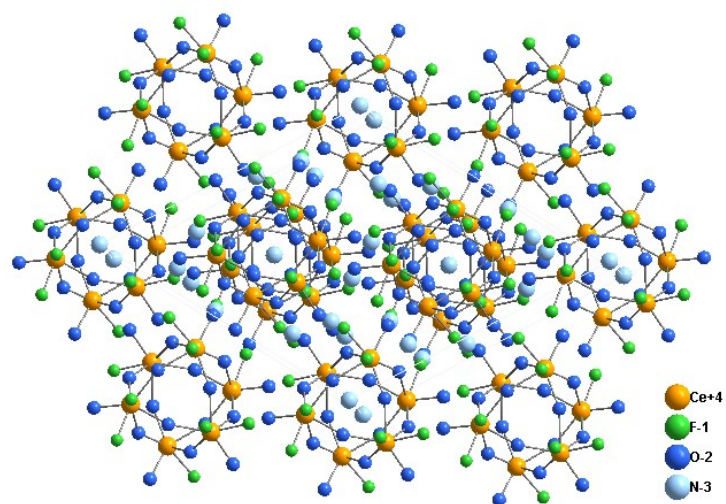
**Figure S10.** Photodegradation of TC-HCl by CeF<sub>3</sub>(a) and CeO<sub>2</sub>(b) under different cutoff filters ( $\lambda > 400$  nm and  $\lambda > 680$  nm). the photodegradation of TC-HCl by CeO<sub>2</sub> is is because the filter (cutoff 400nm) has an optical edge extending to about 380nm. Compared with both of commercial CeF<sub>3</sub> and CeO<sub>2</sub> under visible and NIR light, the CeF<sub>3</sub>-O has excellent photocatalysis activity under visible light. CeO<sub>2</sub> was synthesized by the similar calcination of precursor H<sub>25.5</sub>(NH<sub>4</sub>)<sub>10.5</sub>Ce<sub>9</sub>O<sub>27</sub>F<sub>18</sub> at 600°C for 1h.



**Figure S11.** Photocurrent responses of the CeF<sub>3</sub>-O under visible light irradiation( $\lambda > 400$  nm) and near-infrared light irradiation( $\lambda > 680$  nm). The CeF<sub>3</sub>-O under visible light showed the stronger photocurrent transient response in comparison to that of near-infrared light, indicating the higher charge separation efficiency.



**Figure S12** the structure of pristine CeF<sub>3</sub> (a) and CeF<sub>3</sub>-O (b)



**Figure S13** the structure of precursor  $\text{H}_{25.5}(\text{NH}_4)_{10.5}\text{Ce}_9\text{O}_{27}\text{F}_{18}$

**Table S2.** Details of calculate structure for CeF<sub>3</sub>-O.

<b>Configuration</b>	<b>CeF<sub>3</sub>-O</b>		
Lattice parameter (A°)	a 7.128	b 7.128	c 7.288
Cell angle	$\alpha$ 90°	$\beta$ 90°	$\gamma$ 120°
Ce1	0.33333	0.33333	0.00000
F1	0.38000	0.04600	0.16000
F2	0.33333	0.66667	0.08330
F3	0.00000	0.00000	0.25000
O	0.33333	0.33333	0.25000

**Table S3.** The location and number of k-points

<b>Number</b>	<b>Label</b>			
1	GAMMA	0	0	0
2	Y	0.5	0.5	0
3	T	0.5	0.5	0.5
4	T_2	0.5	0.5	-0.5
5	Z	0	0	0.5
6	Z_2	0	0	-0.5
7	S	0	0.5	0
8	R	0	0.5	0.5
9	R_2	0	0.5	-0.5
10	DELTA_0	-0.33321	0.333207	0
11	F_0	0.333207	0.666793	0
12	B_0	-0.33321	0.333207	0.5
13	B_2	-0.33321	0.333207	-0.5
14	G_0	0.333207	0.666793	0.5
15	G_2	0.333207	0.666793	-0.5