

Improving the all-polymer solar cell performance by adding a narrow bandgap polymer as the second donor

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Supporting Information

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1. General materials and characterizations

Materials: PF2 and PM2 were synthesized according to the literature¹⁻². N2200 were purchased from the Sunatech Co. Molecular weight of PF2: $M_n = 20$ kDa, PDI=1.8; molecular weight of PM2: $M_n=42$ kDa, PDI=2.2; molecular weight of N2200: $M_n=42$ kDa, PDI=2.2.

Chemical Properties: Ultraviolet-Visible (UV-Vis) absorption spectra were recorded on a Perkin-Elmer Lambda 900 spectrophotometer. Cyclic voltammetry (CV) measurements were tested on a CHI-800 electrochemistry workstation. Molecular weights of the polymers were determined by Agilent Technologies PL-GPC 220 high-temperature (150 °C) chromatograph using 1,3,4-trichlorobenzene as the eluent and polystyrene as the standards.

Device characteristics: Current density-voltage ($J-V$) curves were recorded by Keithley 2400 source meter. PCEs were measured under the light of 100 mWcm⁻² AM1.5 G from a solar simulator (Enlitech, Taiwan). The light intensity was calibrated by using a China General Certification Centre (CGC) certified reference monocrystal silicon cell (Enlitech). A physical mask containing an aperture with a precise area of 4 mm² was used to define the device area during the $J-V$ measurements. The external quantum efficiency (EQE) spectra were recorded on a commercial EQE measurement system (Enlitech, QE-R3011, Taiwan). Film thickness was determined on KLA-Tencor P7 Profilometer.

2. Cyclic voltammetry measurements

Cyclic voltammetry (CV) measurements were conducted using a standard three-electrode configuration under an argon atmosphere. A three-electrode cell equipped with a glassy carbon working electrode, an Ag/AgCl reference electrode and a Pt wire counter-electrode. The measurements were performed in absolute acetonitrile with tetrabutylammonium hexafluorophosphate (0.1 M) as the supporting electrolyte at a scan rate of 50 mV/s. Polymer films for CV test were drop-casted onto the glassy carbon working electrode from the chlorobenzene solution, and concentrations are 5

mg/mL. The absolute energy level of ferrocene/ferrocenium (Fc/Fc⁺) was set to be 4.8 eV below vacuum. The HOMO level was calculated from the equation:

$$E_{HOMO} = -(4.8 + E_{onset}^{ox}) eV, \text{ and the LUMO level was calculated from:}$$

$$E_{LUMO} = -(4.8 + E_{onset}^{re}) eV.$$

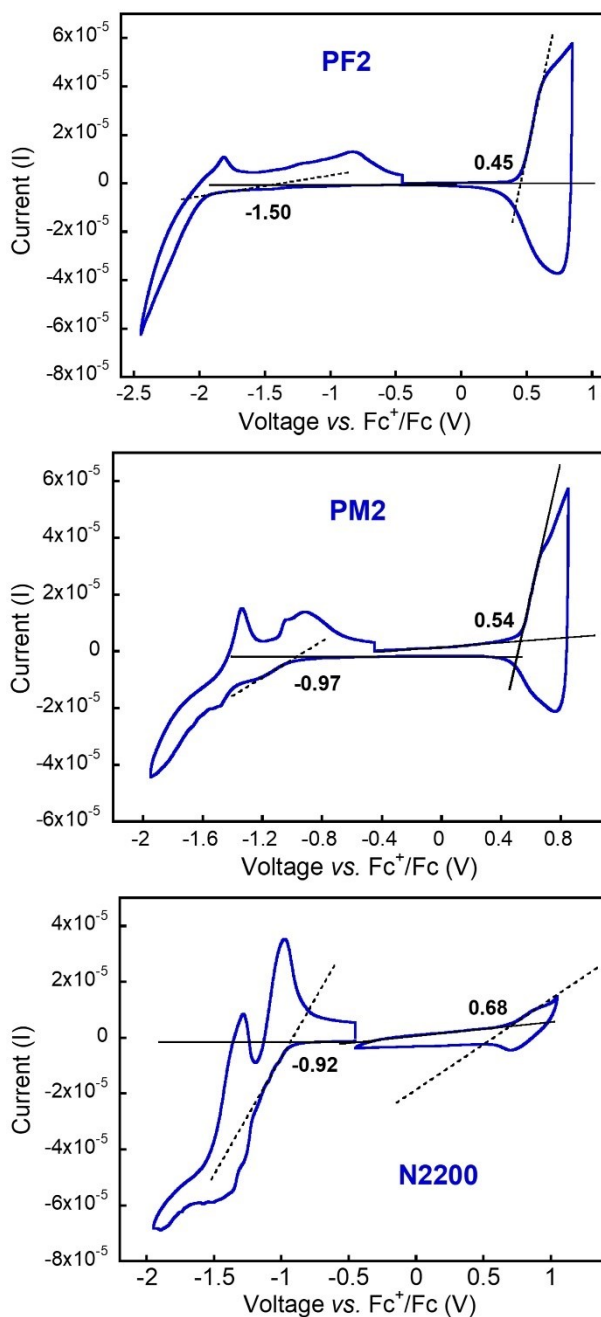


Figure S1. CV measurements of PF2, PM2 and N2200 films.

3. SCLC Mobility measurements

Hole-only devices were fabricated to measure the hole mobility using the space charge limited current (SCLC) method. The diode structure is ITO/PEDOT:PSS/Active Layer/MoO₃/Ag. Polymer films were then spin-coated from the chlorobenzene solutions at the concentration. And the active layers were prepared under the similar condition of those OPV devices. The mobility was determined by fitting the dark current to the model of a single carrier SCLC, which is described by

equation:
$$J = \frac{9}{8} \epsilon_0 \epsilon_r \mu \frac{(V - V_{bi})^2}{d^3}$$
, where J is the current, ϵ_0 is the permittivity of free space, ϵ_r is the relative permittivity of the material, μ is the zero-field mobility, d is the thickness of the polymer layer, V is the applied voltage and the V_{bi} is the built-in voltage. Then hole mobilities were calculated from the slope of the $J^{1/2}$ - V curves in the SCLC region.

Table 1. Hole mobilities of active layers at different polymer ratios.

PF2:PM2:N2200 ratio	Hole mobility (cm ² V ⁻¹ s ⁻¹)
10:0:5	8.3*10 ⁻⁴
9:1:5	2.9*10 ⁻⁴
8:2:5	6.9*10 ⁻⁵
7:3:5	4.6*10 ⁻⁵
0:10:5	1.2*10 ⁻⁶

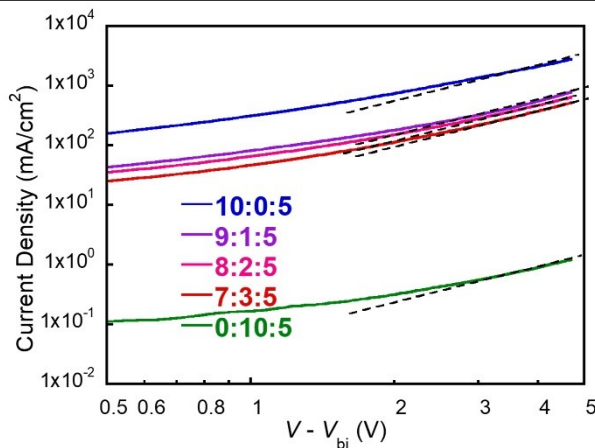


Figure S2. J - V curves of SCLC measurements. Dashed lines indicating the fitting of SCLC region for mobility calculations.

4. Solar cell device characterizations

Table 2. PM2:N2200 devices performance at different D/A ratios. (device structures: ITO/ZnO/PM2:N2200/MoO₃/Ag, films were thermal annealed at 130 °C for 10 minutes.)

	D/A Ratio	J_{sc} (mA/cm ²)	V_{oc} (V)	FF (%)	PCE (%)
PM2:N2200	1:0.5	4.99	0.78	40.09	1.57
	1:0.75	4.24	0.81	38.64	1.33
	1:1	4.12	0.82	40.09	1.35

Table 3. PM2:N2200 devices performance using DIO additive. (device structures: ITO/ZnO/PM2:N2200/MoO₃/Ag, films were thermal annealed at 130 °C for 10 minutes.)

	DIO	J_{sc} (mA/cm ²)	V_{oc} (V)	FF (%)	PCE (%)
PM2:N2200	0	5.20	0.81	40.63	1.71
	1%	7.35	0.85	44.66	2.80
	2%	7.59	0.84	43.88	2.81
	3%	7.61	0.85	43.94	2.84

Table 4. J_{ph} - V_{eff} parameters of PF2:PM2:N2200 devices.

PF2:PM2:N2200 (wt%:wt%:wt%)	J_{sat} (mA/cm ²)	J_{ph} (mA/cm ²)	J_{ph}/J_{sat} (%)	G_{max} (10 ²⁷ m ⁻³ s ⁻¹)
10:0:5	11.17	10.95	98.1	6.78
9:1:5	13.02	12.04	92.5	7.75
8:2:5	13.70	12.52	91.4	7.44
7:3:5	16.01	14.39	89.9	8.70

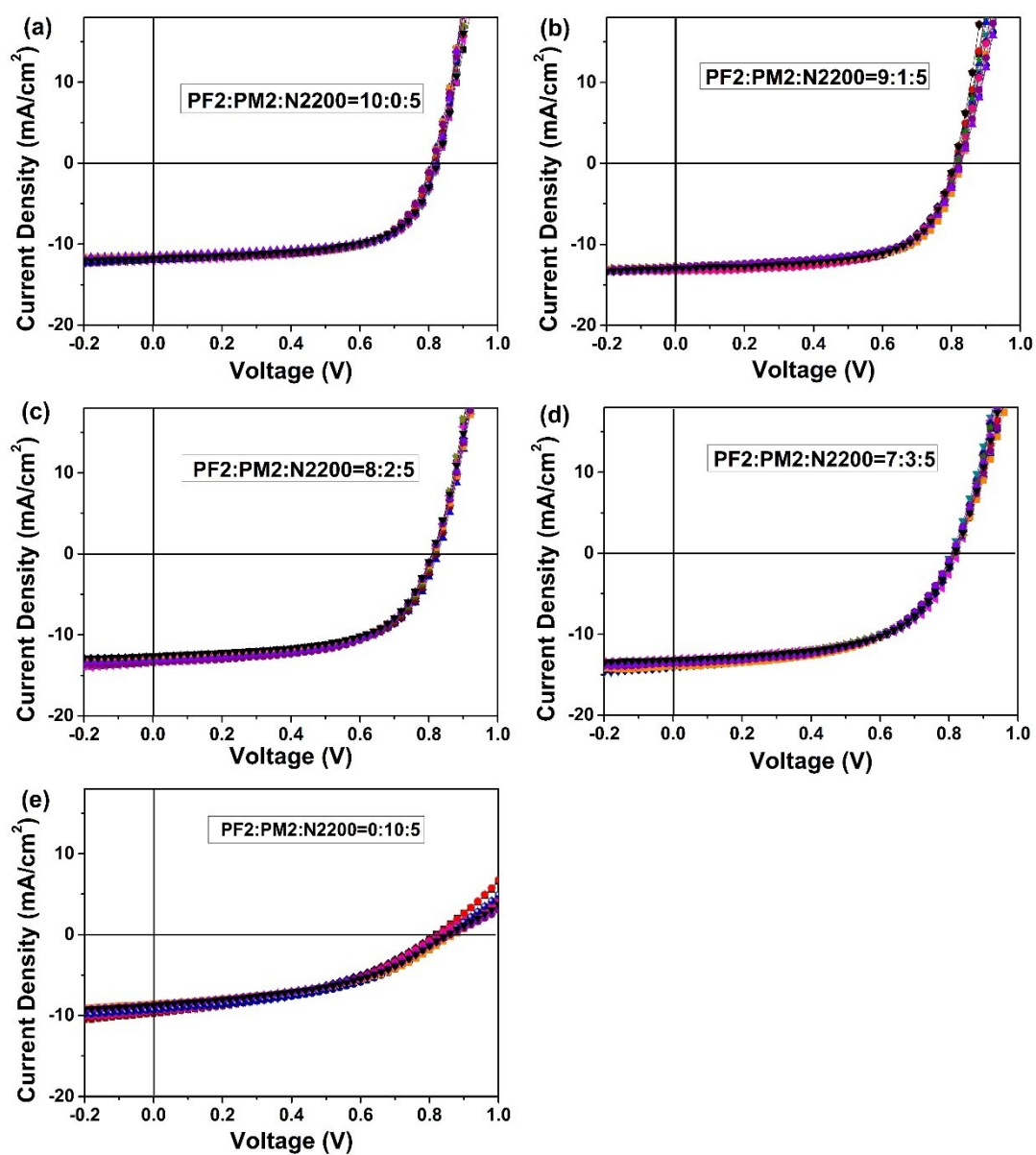


Figure S3. All J - V curves of PF2:PM2:N2200 devices

(ITO/PEDOT:PSS/PF2:PM2:N2200/PFN-Br/Ag): (a) 10:0:5; (b) 9:1:5; (c) 8:2:5; (d)

7:3:5; (e) 0:10:5.

5. AFM measurements

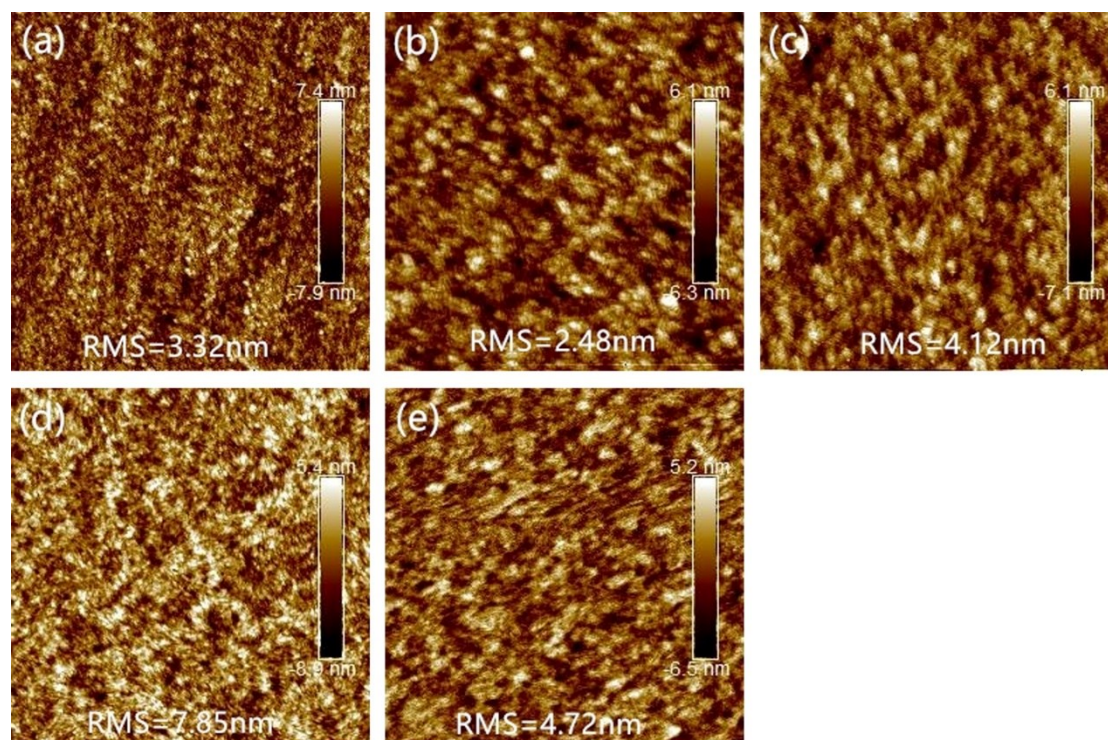


Figure S4. AFM topography measurement of different PF2:PM2:N2200 blend films: (a) 10:0:5; (b) 9:1:5; (c) 8:2:5; (d) 7:3:5; (e) 0:10:5.

References:

1. J. Min, Z.-G. Zhang, S. Zhang, Y. Li, *Chem. Mater.* **2012**, *24*, 3247.
2. M. Wang, H. Wang, M. Ford, J. Yuan, C.-K. Mai, S. Fronk, G. C. Bazan, *J. Mater. Chem. A*, **2016**, *4*, 15232.